



Syntheses of novel 2,3-diaryl-substituted 5-cyano-4-azaindoles exhibiting c-Met inhibition activity

Hannes Koolman^{b,*}, Timo Heinrich^a, Henning Böttcher^a, Wilfried Rautenberg^a, Michael Reggelin^b

^aMerck KGaA, Frankfurter Str. 250, 64293 Darmstadt, Germany

^bClemens-Schöpf Institute of Organic Chemistry and Biochemistry, Technical University of Darmstadt, Petersenstrasse 22, 64287 Darmstadt, Germany

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ABSTRACT

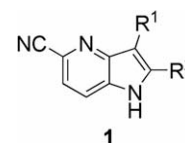
Herein we report the syntheses of 2,3-diaryl-substituted pyrrolo[3,2-*b*]pyridine-5-carbonitriles via a one-pot 5-*endo-dig*-cyclization/protection reaction followed by palladium catalyzed arylation. In addition, a complementary synthesis route employing Larock methodology is applied to efficiently explore further aryl moieties in the 2-position. The novel compounds' expedient c-Met receptor tyrosine kinase inhibition activity is discussed.

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The hepatocyte growth factor (HGF) receptor, also known as mesenchymal–epithelial transition factor (c-Met), is a receptor tyrosine kinase (RTK) present in both normal and malignant cells.¹ The main biological effects upon its signaling pathway activation comprise promotion of tissue regeneration, angiogenesis, and enhanced cell motility.² c-Met is known to be over-expressed and mutated in a variety of human cancer types.³ Thus, signal transduction through the activation of the c-Met receptor is accountable for proliferation, scattering, invasiveness and metastasis of tumor cells. On this account, small molecule inhibitors, preventing receptor autophosphorylation and recruitment of the downstream effectors of c-Met, are of current interest.⁴

Currently, there are several ATP-competitive c-Met kinase inhibitors known, based on different scaffolds.⁵ Recent developments feature 7-azaindoles,⁶ aminopyridines⁶ and triazolopyridazines⁷ with high potency and selectivity for c-Met.

As part of our studies towards aromatically substituted azaindoles as kinase inhibitors, the rarely used core structure of the 4-azaindoles in combination with diaryl substitution in the 2- and 3-position attracted our attention.^{8,9} At present, this structure is unknown as an inhibitor of c-Met and 5-cyano-2,3-diaryl-4-azaindoles (**1**) in particular have not been previously described as kinase inhibitors. By analogy to known compounds,⁸ these pyrrolopyridines are anticipated to offer kinase inhibiting potential.



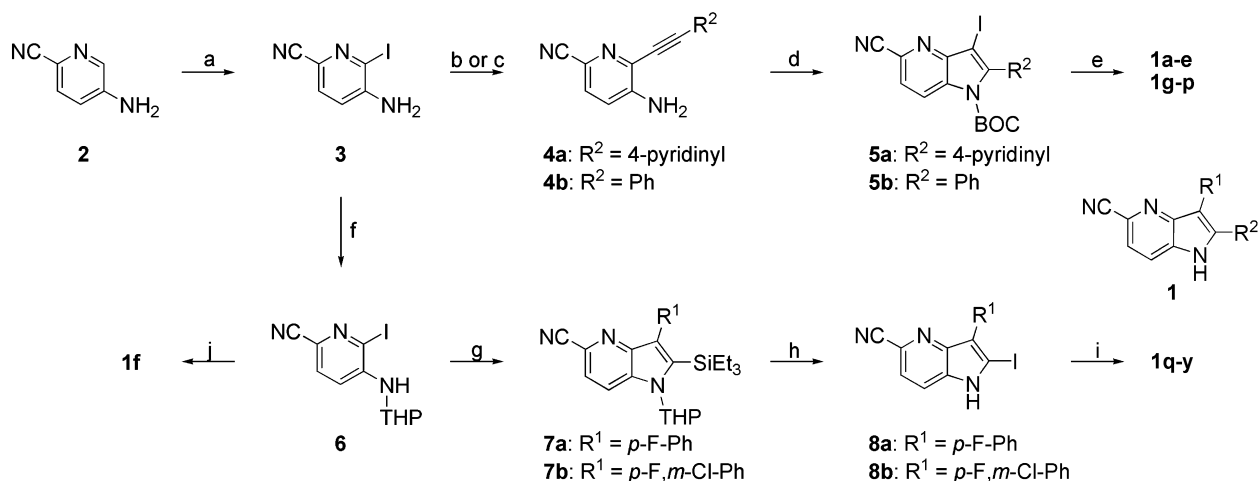
For the full syntheses of azaindoles a growing number of approaches are known, heavily depending on the desired substitution pattern.¹⁰ Recently, Cacchi et al. transferred their highly versatile methodology for the assembly of 2,3-diaryl-substituted indoles¹¹ onto 4- and 7-azaindoles.⁹ This methodology comprises the reaction of 2-alkynyl-3-trifluoroacetamidopyridines with aryl bromides and -triflates in an aminopalladation-reductive elimination procedure. However, in our hands, this approach failed to yield the desired 5-cyano-substituted azaindoles due to loss of the trifluoroacetate protecting group under the reaction conditions.¹² Therefore a strategy based upon base-induced cyclization of *o*-aminoalkynylpyridines was employed (Scheme 1).

Silver(I)-assisted electrophilic aromatic iodination of commercially available 5-aminopicolonitrile **2** gave the mono-iodinated intermediate **3** in 79% yield. The cyclization precursor **4a** could be obtained under Sonogashira-like alkynylation conditions in moderate yield (66%).¹³

Azaindoles **5** were synthesized by subsequent base-induced 5-*endo-dig* cyclization in NMP,¹⁴ direct addition of *N*-iodosuccinimide and BOC-anhydride, leading to the protected 3-iodo-4-azaindoles in good yields of 73–83% (see Ref. 15 for an exemplary procedure). These compounds are stable if directly purified via chromatography over neutral aluminum oxide and stored at –20 °C.¹⁶ Arylation in the 3-position was achieved via Suzuki–

* Corresponding author. Tel.: +49 6151 72 5881.

E-mail address: hannes.koolman@merck.de (H. Koolman).



Scheme 1. Reagents and conditions: (a) I_2 , Ag_2SO_4 , EtOH, rt, 79%; (b) 4-ethynylpyridine, Cs_2CO_3 , $\text{Pd}(\text{dppf})\text{Cl}_2$, THF, 50 °C, 66%; (c) 3-phenylpropionic acid, Na_2CO_3 , LiCl, $\text{Pd}(\text{dppf})\text{Cl}_2$, DMF, 110 °C, 20%; (d) KO^tBu , NMP, 90 °C, then NIS, DCM, 0 °C to RT, then $(\text{BOC})_2\text{O}$, DMAP, DCM, 0 °C to RT, 73–83%; (e) $\text{R}^1\text{B}(\text{OR})_2$, K_2CO_3 , $\text{Pd}(\text{dppf})\text{Cl}_2$, DME/ H_2O 2:1, 85 °C then TFA or ethan. HCl, 60 °C, 25–86%; (f) DHP, MgBr_2 , THF, 65 °C, quant.; (g) **11a** or **11b**, Na_2CO_3 , LiCl, $\text{Pd}(\text{dppf})\text{Cl}_2$, DMF, 110 °C (h) IPy_2BF_4 , TfOH, DCE, 83 °C, 48–66% over two steps (i) $\text{R}^2\text{B}(\text{OR})_2$, K_2CO_3 , $\text{Pd}(\text{dppf})\text{Cl}_2$, DME/ H_2O 2:1, 85 °C, 9–77% (j) **13**, Na_2CO_3 , LiCl, $\text{Pd}(\text{dppf})\text{Cl}_2$, DMF, 110 °C then 1 N HCl, 110 °C, 39%.

Miyaura-coupling with various boronic acids or -esters. The protecting group was quantitatively removed by subsequent acidification of the reaction mixture with TFA or hydrochloric acid to yield the desired 2,3-diaryl-substituted pyrrolo[3,2-*b*]pyridine-5-carbonitriles **1a–e** and **1g–p**. The BOC-protecting group was indispensable in this step, as unprotected 3-iodo-4-azaindoles failed to undergo efficient Suzuki-type cross-coupling, showing dehalogenation as major side reaction.¹⁷ Thus, employing two one-pot routines we were able to prepare highly decorated 4-azaindoles from quite simple starting materials.

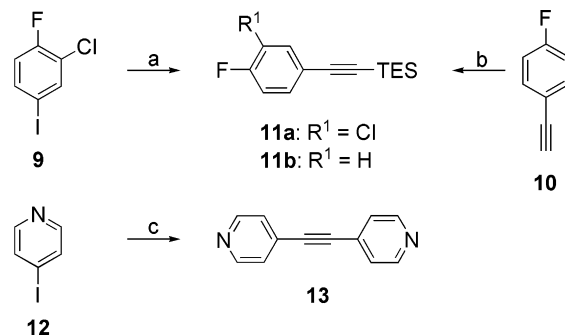
An alternative approach was based upon Larock's indole synthesis. Via this route variable aryl substitution in the 2-position could be efficiently explored. For the key cycloaddition step introduction of an electron donating protecting group was indispensable.¹⁸ Therefore, iodo-aminopyridine **3** was reacted with dihydropyran under Lewis-acid catalysis, yielding **6** quantitatively. The 4-azaindoles **7** were prepared under Larock conditions with triethylsilyl-protected alkynes **11** (vide infra).¹⁹ Thereby, traces of N-deprotected products were generated. This crude product mixture was treated with IPy_2BF_4 in the presence of an excess of triflic acid to convert the TES-group to the corresponding iodides **8**.²⁰ The THP-group was concurrently removed during this halogenation. Derivatives **1q–y** were achieved under standard Suzuki-coupling conditions. It is noteworthy that this coupling works uneventfully without N-protection on the indole. The 2,3-dipyridinyl-derivative **1f** was yielded by Larock-reaction of **6** with symmetrically substituted ethyne **13** and subsequent acidic THP-group removal.

The alkynes **11a**, **11b** and **13**, employed in the Larock reactions with pyridine **6** were prepared according to known procedures (Scheme 2).²¹

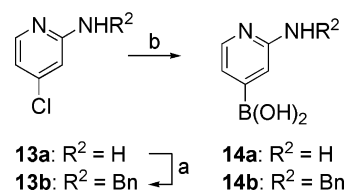
Iodobenzene **9** was converted to **11a** via Sonogashira-coupling with TES-ethyne, whereas 4-ethynylpyridine and 4-iodopyridine **12** were coupled to ethyne **13**, respectively. Ethynyl-benzene **10** was silylated with TES-chloride after deprotonation with *n*-BuLi, to yield **11b**.

Synthesis of the non-commercially available boronic acids **14a** and **14b** employed in the final synthesis step for the derivatives **1u** and **1v**, respectively, were performed following a procedure published by Buchwald et al. (Scheme 3).²²

First, the structure activity relation of 4-azaindoles **1a–o** with different aryl-substituents in the 3-position is discussed, and their c-Met-kinase inhibition activity is listed in Table 1. The phenyl-



Scheme 2. Reagents and conditions: (a) triethylsilylacetylene, $\text{Pd}(\text{PPh}_3)_4$, CuI, Et_2NH , 75%; (b) *n*-BuLi, TESCl, THF, –78 °C to rt, 86%; (c) 4-ethynylpyridine-hydrochloride, $\text{Pd}(\text{PPh}_3)_4$, CuI, Et_2NH , 53%.

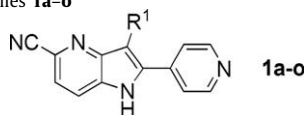


Scheme 3. Reagents and conditions: (a) NaH, BnCl, DMF, 0 °C to rt, 23%; (b) Bis(pincolato)diboron, KOAc, $\text{Pd}(\text{dba})_2$, XPhos, dioxane, 110 °C, 95% -quant.

derivative **1a** shows an IC_{50} -value of 2.09 μM , suggesting that lipophilic moieties are preferred as substituents in the 3-position of the 4-azaindole. The activity is maintained with the introduction of a *p*-fluoro substitution (**1b**), but steadily decreases with increasing size of the halogen (**1a–d**). Similarly, derivative **1e**, bearing a pseudohalogenic cyano substitution, is significantly less active. The *m*-chloro-derivative **1g**, exhibiting a 50-fold activity increase compared to the phenyl derivative **1a**, indicates that additional halogen substituents are beneficial in this position. In case of the *m,m'*-disubstituted derivatives (**1i**, **1j**), the *m*-chloro, *m'*-fluoro-derivative (**1j**) is about 7.5-fold more potent than **1i**. But remarkably, when combining *meta*- and *para*-substitution (**1k**, **1l**) the chloro-substituent (**1k**) is about ninefold superior to the fluoro-derivative (**1l**). However, the *m,p*-dichloro-derivative **1h** is

Table 1

c-Met inhibition activity of the 3-aryl-substituted 2-(pyridin-4-yl)-1*H*-pyrrolo[3,2-*b*]pyridine-5-carbonitriles **1a–o**



Compounds	R ¹	IC ₅₀ ^a
1a		2.09
1b		1.95 ^b
1c		10
1d		>10 ^c
1e		>10 ^c
1f		na ^d
1g		0.04
1h		>10 ^c
1i		2.55
1j		0.34
1k		0.18
1l		1.59
1m		>10 ^c
1n		na ^d
1o		na ^d

^a IC₅₀ values represent average values from 2 to 4 measurements, and are defined as the concentration (μM) resulting in 50% inhibition of activity, for assay conditions see Ref. 23.

^b Exemplary solubility data: (in water at pH 7.4): 1 μg/ml.

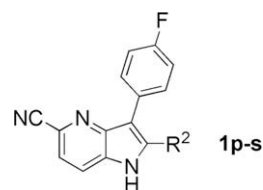
^c Remaining activity >60% at 10 μM

^d Inactive at 10 μM.

significantly less active. The compounds **1n** and **1o**, bearing an amino-pyrimidinyl and a *p*-methylsulfonyl-phenyl substitution, respectively, are inactive.

Table 2

c-Met inhibition activity of the 2-aryl-substituted 3-(4-fluorophenyl)-1*H*-pyrrolo[3,2-*b*]pyridine-5-carbonitriles **1p–s**



Compounds	R ²	IC ₅₀ ^a
1p		na ^b
1q		>10 ^c
1r		>10 ^c
1s		0.33

^a IC₅₀ in μM.

^b Inactive at 10 μM.

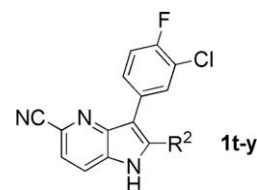
^c Remaining activity >60% at 10 μM.

The potency of **1g** became apparent when the optimization of the 2-aryl-moiety with derivatives of **1b** (see Table 2) and **1k** (see Table 3) was already ongoing.

Comparing the inactive phenyl-derivative **1p** with the above-mentioned pyridine-analog **1b** the pyridine nitrogen is clearly accountable for the μM activity. In case of an influencing *o*-chloro-substituent (**1q**) the activity gain of the pyridine nitrogen

Table 3

c-Met inhibition activity of the 2-aryl-substituted 3-(3-chloro-4-fluorophenyl)-1*H*-pyrrolo[3,2-*b*]pyridine-5-carbonitriles **1t–y**



Compounds	R ²	IC ₅₀ ^a
1t		>10 ^b
1u		0.13
1v		5.57
1w		>10 ^b
1x		>10 ^b
1y		>10 ^b

^a IC₅₀ in μM.

^b Remaining activity >60% at 10 μM.

is reduced. Derivative **1r**, the pyridine-2-yl-isomer of **1b**,²⁴ cannot sustain the activity, whereas introduction of a second *meta*-nitrogen (**1s**) obviously compensates this effect. Thus, **1s** is about sixfold more active than the pyridine-4-yl-analog **1b**. In light of published structures⁸ this effect is surprising. Obviously, this result cannot be generally extrapolated, because the pyrimidinyl in position 2 together with a *m*-chloro,*p*-fluoro-phenyl-moiety in the 3-position (**1t**, Table 3) reduces activity drastically, revealing subtle structure activity relations.

Compounds **1u** and **1v** are based on **1k**. Comparable derivatization of the pyridine-moiety is known to be beneficial for similar p38-kinase inhibitors.⁸ Introduction of a 2-aminopyridin-4-yl-moiety in the 2-position of the azaindole has no effect on activity (**1u**) evidencing that an *o*-amino-substituent is tolerated. Benzyl-substitution in **1v**, however, results in a 30-fold loss of activity compared to **1k**, showing that large lipophilic moieties are not tolerated in this position.

The 2-aminopyridin-3-yl isomer **1x** cannot maintain the activity (compared to **1u**). This observation is consistent with the loss of activity already noticed for the pyridine-3-yl-isomer **1r**. Similarly, the analogs **1w** and **1y** exhibit only weak inhibition activity.

Employing two different protocols based on identical starting material various 2,3-diaryl-substituted 5-cyano-4-azaindoles could be efficiently synthesized. The application of optimized protocols in one-pot procedures enabled a convenient synthesis of highly decorated heterocycles. The compounds showed promising inhibition activity of the c-Met RTK and led to the identification of an inhibitor with an IC₅₀ of 40 nM (**1g**).

Elucidation of the compounds' selectivity profiles, particularly discrimination of p38-kinases, is currently under investigation: Preliminary screening of single compounds against a panel of 80 kinases revealed promising selectivities. Further studies comprising improvement of activity and solubility enhancement are currently ongoing in our group.

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- Exemplary synthesis procedure (1g)**: 331 mg (1.50 mmol) of **4a** were Schlenk tube-dried and were dissolved in NMP (15 ml) under nitrogen. 287 mg (2.55 mmol) KO^tBu were added and the mixture was stirred for 4 h at 90 °C, then cooled to 0 °C and a solution of 507 mg (2.25 mmol) NIS in NMP (10 ml) added drop wise. After 1 h at room temperature, a solution of 1.64 g (7.52 mmol) (BOC)₂O and 183 mg (1.50 mmol) DMAP in NMP (5 ml) was added at 0 °C. After 1 h at 0 °C the reaction mixture was diluted with water (250 ml) and extracted with CH₂Cl₂. The combined organic phases were washed with brine, dried and evaporated. After flash-chromatography over neutral Al₂O₃ (eluent: ethyl acetate/cyclohexane 9:1) 563 mg (1.26 mmol, 83%) tert-butyl 5-cyano-3-iodo-2-(pyridin-4-yl)-1H-pyrrolo[3,2-b]pyridin-1-carboxylate **5a** was yielded as white solid. Mp 126–129 °C. ¹H NMR (400 MHz, DMSO-*d*₆): δ 8.75 (m, 2H), 8.60 (d, *J* = 8 Hz, 1H), 8.07 (d, *J* = 8 Hz, 1H), 7.53 (m, 2H), 1.25 (s, 9H). ESI-MS: *m/z*: 447 ([M+H]⁺), 347 ([M-BOC]⁺). Anal. Calcd for C₁₈H₁₅IN₄O₂: C, 48.45; H, 3.39; N, 12.56. Found: C, 49.00; H, 3.80; N, 12.00.
- 223 mg (0.50 mmol) of **5a**, 117 mg (0.75 mmol) of 3-chlorophenylboronic acid and 207 mg (1.50 mmol) of K₂CO₃ were dissolved in 7.5 ml of DME/H₂O 2:1 in a capped vial. After degassing employing ultrasound under nitrogen 20.4 mg (0.025 mmol) of Pd(dppf)Cl₂·CH₂Cl₂ were added and the reaction mixture stirred at 80 °C for 6 h. Then 5 ml of TFA were added and the mixture stirred at 60 °C for 15 h, set to pH 12 using dil NaOH and extracted with ethylacetate. The combined organic phases were dried, evaporated and purified via flash-chromatography (eluent: ethyl acetate to ethyl acetate/methanol 95:5) to yield 95.0 mg (0.287 mmol) of 3-(3-chlorophenyl)-2-(pyridin-4-yl)-1H-pyrrolo[3,2-b]pyridine-5-carbonitrile **1g** as a beige solid. Mp 267 °C. ¹H NMR (300 MHz, DMSO-*d*₆): δ 12.71 (s, 1H), 8.67 (d, *J* = 5.6 Hz, 2H), 8.00 (d, *J* = 8.2 Hz, 1H), 7.82 (d, *J* = 8.2 Hz, 1H), 7.55 (m, 1H), 7.51–7.36 (m, 5H). APCI-MS: *m/z* (%): 331 (100, [M+H]⁺).
- At room temperature under atmosphere compounds **5** show dehalogenation as major decomposition reaction.
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