

SYNTHESIS OF [ $^3\text{H}$ ]CI-980, ETHYL[5-AMINO-1,2-DIHYDRO-2(*S*)-METHYL-3-(3- $^3\text{H}$ PHENYL)PYRIDO[3,4-*b*]PYRAZIN-7-YL]CARBAMATE ISETHIONATE SALT, A TUBULIN-BINDING, ANTIMITOTIC, BROAD-SPECTRUM ANTITUMOR AGENT

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SUMMARY

[ $^3\text{H}$ ]CI-980 (10b) was synthesized in an eight-step sequence with an overall yield of 2.4%. Reaction of *m*-bromophenyl lithium (2) with *N*-ethoxycarbonyl-L-alanine gave the chiral ketone 3. Reduction of 3 with sodium borohydride followed by alkaline *N*-deprotection and condensation with ethyl 6-amino-4-chloro-5-nitro-2-pyridine carbamate (6) gave 7. Chromium trioxide oxidation of 7 followed by reductive cyclization with iron-acetic acid gave the key bromo intermediate 9. Palladium catalyzed  $^3\text{H}$ -hydrogenolysis of 9 gave the free base form of [ $^3\text{H}$ ]CI-980 (10a), which was converted to the isethionate salt (10b) before use.

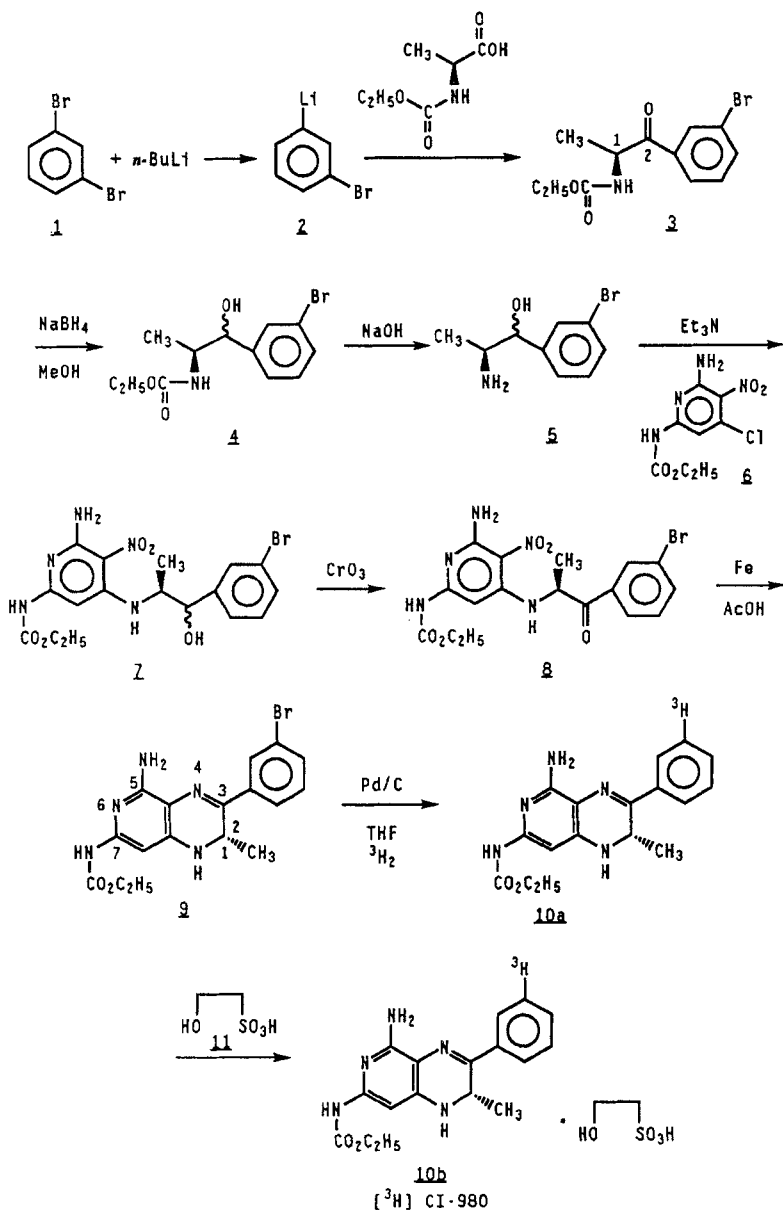
**Key Words:** [ $^3\text{H}$ ]CI-980, antitumor agent, catalytic  $^3\text{H}$ -hydrogenolysis

INTRODUCTION

As indicated in the previous communication, the biologically more active 2(*S*) enantiomeric component of ethyl[5-amino-1,2-dihydro-2-methyl-3-phenylpyrido[3,4-*b*]pyrazin-7-yl]carbamate (*R,S*-10), a broad spectrum antitumor agent,<sup>1</sup> is being developed clinically as CI-980. During the process, the labeled forms of CI-980 were needed for pharmacokinetics and drug metabolism studies. This paper presents the synthesis of [ $^3\text{H}$ ]CI-980.

RESULTS AND DISCUSSION

[ $^3\text{H}$ ]CI-980 was synthesized in eight steps (Scheme I). Lithiation of 1 with *n*-butyl lithium at  $-110^\circ\text{C}$ , followed by reaction with *N*-ethoxycarbonyl-L-alanine<sup>2</sup>

Scheme I. Synthesis of [ $^3\text{H}$ ]CI-980

gave, after chromatography, 21% of pure 3. Reduction of 3 with sodium borohydride in methanol at room temperature, followed by alkaline hydrolysis with

potassium hydroxide in aqueous methanol (2 h, reflux), gave a crude product which was purified by trituration with ethanol; pure **5** (77%) was isolated in the mother liquor. Condensation of **5** with **6** in the presence of triethylamine in ethanol (24 h, reflux) gave, after chromatography, 30% of **7**. Oxidation of **7** with chromium trioxide in pyridine-dichloromethane (2 h, room temperature) gave 66% of **8**. Reductive cyclization of **8** with iron powder in acetic acid-ethanol (3.5 h, reflux) gave 75% of **9**.

Replacement of the bromine in **9** was readily accomplished in 1 h with either hydrogen or deuterium, using Pd/C as catalyst and THF as solvent. Tritiation was then carried out under similar conditions but in the presence of 15 equivalents of unlabeled CI-980 free base as carrier, as no product could be isolated in its absence. The product, after purification by silica gel chromatography, was characterized and stored as the free base **10a**. The free base was converted to [ $^3\text{H}$ ]CI-981 (**10b**), the isethionate salt, for pharmacokinetic and metabolic studies immediately before use, as the latter was found to be unstable.

## EXPERIMENTAL

Tritiation was performed by Amersham Co. Radioactivity was determined with a Packard Tri-Carb 4530 liquid scintillation counter, using Beckman Ready-Gel as the counting medium. TLC plates, E. Merck silica gel 60 F-254, were scanned on a Berthold LB2832 automatic TLC linear analyzer. Column chromatography was performed using E. Merck silica gel, 230-400 mesh. HPLC was performed using a Spectra Physics SP8700 solvent delivery system, Kratos Spectroflow 773 variable wavelength UV detector, and Radiomatic Beta Flow 1 radioactivity flow detector. Unless otherwise specified, Alltech Econosil Columns, C18 4.7 mm x 20 cm were used. MS data were obtained by electron impact at 70 eV using a V G Analytical 7070E/HF mass spectrometer.

### Ethyl [2-(3-Bromophenyl)-1(*S*)-methyl-2-oxoethyl]carbamate (**3**).

*n*-Butyl lithium (30 mL, 75 mmol, 2.5 M) was added dropwise to an ether solution (125 mL) of *m*-dibromobenzene (17.7 g, 75 mmol) at  $-110^\circ\text{C}$  and stirred at this temperature for 0.5 h. *N*-(Ethoxycarbonyl)-L-alanine<sup>2</sup> (0.4 g, 24.8 mmol) in

ether (50 mL) was then added slowly. This reaction mixture was allowed to warm up from  $-110^{\circ}\text{C}$  to  $-10^{\circ}\text{C}$ . 1 M  $\text{H}_3\text{PO}_4$  (100 mL) was added all at once, and the temperature was kept below  $5^{\circ}\text{C}$ . The residue from evaporation of the ether layer was purified by chromatography over silica gel (3:2,  $\text{CHCl}_3$ :hexane) to give product 3 (1.6 g, 21%).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.05–1.5 (m, 6H,  $\text{CH}_3$ ), 4.05 (q, 2H,  $\text{CH}_2$ ), 5.0–5.36 (m, 1H, CH), 5.5 (b, 1H, NH), 7.25–8.0 (m, 4H, Ar). Elemental analysis: calc. for  $\text{C}_{12}\text{H}_{14}\text{N}_3\text{O}_3\text{Br}$ : C 48.02; H 4.70; N 4.67. Found: C 48.13; H 4.82; N 4.83.

$\alpha$ -(1-Aminoethyl)-3-bromobenzenemethanol (5).

Sodium borohydride (361 mg, 9.54 mmol) was added in two portions over 0.5 h to a stirred solution of 3 (1.9 g, 6.33 mmol) in  $\text{CH}_3\text{OH}$  (25 mL) at room temperature. After 0.5 h, acetic acid (2 mL) was added and the reaction was evaporated to dryness. The residue was distributed between  $\text{CH}_2\text{Cl}_2$  and saturated  $\text{NaHCO}_3$ . The aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$  (2 x 20 mL), and the combined  $\text{CH}_2\text{Cl}_2$  portions were evaporated to an oil. Aqueous MeOH (25 mL, 3:1, MeOH:H<sub>2</sub>O) containing 0.9 g of KOH was added and the solution was refluxed for 2 h, cooled, and evaporated to dryness. An insoluble impurity was removed by recrystallization from EtOH. Evaporation of the mother liquor gave pure 5, 1.12 g (77%).

Ethyl [6-Amino-4-[[2-(3-bromophenyl)-2-hydroxy-1-methylethyl]amino]-5-nitro-2-pyridinyl]carbamate (7).

A mixture of 6 (1.017 g, 4.45 mmol), 5 (1.12 g, 4.87 mmol), and  $\text{Et}_3\text{N}$  (0.65 mL) in EtOH (17 mL) was heated under reflux for 24 h, cooled, and evaporated to dryness. The residue was purified by silica gel column chromatography (1% EtOH in  $\text{CH}_2\text{Cl}_2$ ) to give 670 mg of 7 (30%).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.18 (d, 3H,  $\text{CH}_3$ ), 1.3 (t, 3H,  $\text{CH}_3$ ), 3.8–4.0 (m, 1H, CH), 4.25 (q, 2H,  $\text{CH}_2$ ), 5.05 (m, 1H, CH), 6.95–7.65 (m, 5H, Ar).

Ethyl [6-Amino-4-[[2-(3-bromophenyl)-1-methyl-2-oxoethyl]amino]-5-nitro-2-pyridinyl]carbamate (8).

A solution of 7 (640 mg, 1.41 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) was added to a stirred solution of pyridine (1.77 mL in 25 mL CH<sub>2</sub>Cl<sub>2</sub>), followed by CrO<sub>3</sub> (1.078 g, 10.78 mmol). After stirring at room temperature for 2 h, the mixture was filtered through celite. The filtrate was evaporated, and the product (8) was obtained after purified by silica gel column chromatography (3% EtOH in CHCl<sub>3</sub>) as yellow powder (420 mg, 66%). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 1.3 (t, 3H, CH<sub>3</sub>), 1.6 (d, 3H, CH<sub>3</sub>) 4.22 (q, 2H, CH<sub>2</sub>), 5.16 (t, 1H, CH), 6.8 (s, H, NH), 7.1-7.98 (m, 4H, Ar), 8.05 (s, 1H, Ar). Elemental analysis: calc. for C<sub>17</sub>H<sub>18</sub>N<sub>5</sub>O<sub>5</sub>Br·0.15CHCH<sub>3</sub>: C 43.81; H 3.89; N 14.90. Found: C 43.75; H 3.60; N 15.14. MS: m/z 452, 454 [(M+H)<sup>+</sup>; <sup>79</sup>Br, <sup>81</sup>Br].

Ethyl [5-Amino-3-(3-bromophenyl)-1,2-dihydro-2(S)-methylpyrido[3,4-b]pyrazin-7-yl]carbamate (9).

Iron powder (39.6 mg, 0.71 mmol) was added to a solution of 8 (80 mg, 0.18 mmol) in 6.2 mL of glacial acetic acid and 6.2 mL of abs EtOH. The mixture was heated under reflux for 3.5 h. After addition of H<sub>2</sub>O (1 mL) and neutralization with solid Na<sub>2</sub>CO<sub>3</sub>, the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 20 mL), and the combined extracts were dried over MgSO<sub>4</sub> and concentrated under vacuum. Pure 9 was isolated by column chromatography on silica gel using Et<sub>3</sub>N:EtOH:CH<sub>2</sub>Cl<sub>2</sub> (3:3:94) as eluent. The yellow powder (9) weighed 70 mg (98%). MS: m/z 404, 406 [(M+H)<sup>+</sup>; <sup>79</sup>Br, <sup>81</sup>Br].

[<sup>3</sup>H]CI-980 (10b)

The <sup>2</sup>H analog of 10a was first prepared in order to optimize the reaction conditions and isolation procedures. Compound 9 (10 mg, 0.025 mmol) was subjected to <sup>2</sup>H-hydrogenolysis in the presence of 5% Pd/C (10 mg) in 4 mL THF under deuterium for 1 h at 1 atm and room temperature. The mixture was filtered and washed with methanol, and the filtrate was neutralized with K<sub>2</sub>CO<sub>3</sub> (0.1 g) and

evaporated to dryness. The pure  $^2\text{H}$  analog (6 mg, 75%) was isolated by column chromatography using 2% EtOH in  $\text{CHCl}_3$  as eluant. MS:  $m/z$  327  $[(\text{M}+\text{H})^+]$ ; 326 for the reference  $^1\text{H}$  analog).

Using similar procedure as for the  $^2\text{H}$  analog above but in the presence of unlabeled CI-980 free base as carrier,  $^3\text{H}$ -hydrogenolysis of **9** was carried out (8 mg of **9**, 100 mg of unlabeled **10a**, 20 mg Pd/C, 10 mL THF, 5 Ci of tritium gas), and the crude product was isolated as the free base (5 mL methanol rinse, 100 mg  $\text{K}_2\text{CO}_3$ ). The crude product was coevaporated three times with ethanol to remove tritium gas, purified by chromatography, and further diluted with unlabeled **10a** to give the tritiated free base product (210 mg; 113 mCi 19.5% of theoretical). The purity of **10a**, according to TLC, was 98.5% (2:98 EtOH: $\text{CHCl}_3$ ;  $R_f$  0.22), and according to HPLC (55% 0.05 M triethylamine adjust to pH 3.1 with formic acid, 45% acetonitrile, 1 mL/min, retention time 5.1 min), was 98.5% radiochemical and 100% chemical (270 nm).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  1.1-1.35 (m, 6H,  $\text{CH}_3$ ), 4.1-4.35 (q, 2H,  $\text{CH}_2$ ), 4.45 (b, 1H, NH), 4.8-4.9 (m, 1H, CH), 5.0 (b, 2H,  $\text{NH}_2$ ), 7.35-7.05 (s, 1H, Ar), 7.75 (s, 1H, Ar), 7.9-8.05 (m, 2H, Ar). The specific activity was 175 mCi/mmol. Compound **10a** was treated with 1 eq of isethionic acid to give the salt **10b** immediately before pharmacokinetic and metabolic studies.

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#### REFERENCES

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