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### 4-THIOPHENYL-2-AZETIDINONE AS CHIRON : ENANTIOSPECIFIC SYNTHESES OF 3R AND 3S DEUTERIATED β-ALANINES

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Abstract: Stereospecific syntheses of chirally C-3 deuteriated  $\beta$ -alanines 6 and 7 from optically pure 4 - thiophenyl azetidinones 13 and 14 are described.

Chirally deuteriated molecules are important because of their applications in biosynthetic studies<sup>1,2</sup> especially for determination of stereochemistry and kinetic isotope effects. Aminoacids, which are often involved in natural product biosyntheses<sup>3</sup>, are useful for the same reason in chirally deuteriated forms. Their simplest member  $\beta$ - alanine, besides being a constituent of vitamin pantothenic acid / coenzyme A, is also a synthon for biologically important molecules like 2 - azetidinone<sup>4</sup>. Recently<sup>5</sup>, it has been utilized in the synthesis of isotopically labelled proclavaminic acid 1, the monocyclic intermediate for clavulanic acid biosynthesis (Scheme 1). In this article we describe a simple synthesiz of contrally 3 - deuteriated  $\beta$ - alanine, stereospecifically substituted at C - 3 with allyl, aryl or alkynyl groups<sup>6</sup>.

#### Scheme 1



Recently, 2 - azetidinones have been utilized in the synthesis of dealanylalahopcin<sup>7</sup> and  $\infty$ - amino phosphonic acid<sup>8</sup> by ring opening in presence of base or acid. We envisioned that our problem will be solved if we can achieve a synthesis of chiral 4 - deuteroazetidinones 4 and 5 which can then be opened up to 6 and 7 (Scheme II).

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#### Scheme II



Chiral deuteriated molecules are usually prepared by asymmetric reduction of deuteriated aldehydes with Alpine boranes<sup>9</sup>. Departing from this conventional methodology, we have earlier prepared<sup>2</sup> chiral 4/ - deuteriated benzylazetidinyl acetates 10 and 11 by tin deuteride reduction of <u>cis</u> and <u>trans</u> benzyl 3' - trimethylsilyl - 4/-thiophenyl azetidinyl acetates 8 and 9, respectively, exploiting the steric effect operative during deuterium abstraction (Scheme III). We have utilized this stereospecific reduction in the synthesis of chiral  $\beta$ - alanines.

Scheme III



One problem in the above method is the relatively low yield for the reduction of the cis isomer 8 in which case the reaction is also sluggish (4 eq of tin deuteride, 24 h, 50% yield). To circumvent this problem both enantiomers of trans 3-trimethylsilyl - 4 thiophenylazetidinones 15 and 16 were prepared from 4S and 4R thiophenyl azetidinones 13 and 14, respectively, which in turn were made from racemic 4 phenylsulphonylazetidinone 12 via displacement with thiophenol using cinchonine or cinchonidine<sup>10</sup> as chiral base (SchemelV). The <u>3R</u>, <u>4S</u> - isomer 15 upon tin deuteride reduction gave <u>3R</u> - trimethylsilyl - <u>4S</u> - <sup>2</sup> H - azetidinone 17 in 82% yield. That the deuterium was stereospecifically delivered from the opposite face of the TMS was evident in the <sup>1</sup> H NMR spectrum. The C - 3 hydrogen appeared as a doublet at 3.08 with a coupling constant of 2.6 Hz (characteristic of trans coupling<sup>11</sup>). Moreover the 3.41 for the C - 4 - hydrogen was virtually absent. Careful integration in the signal at high field NMR showed the extent of stereospecificity to be 18:1, while mass spectrum showed 98% deuterium incorporation. The TMS group was removed<sup>12</sup> (KF/acetonitrile) and the resulting 4S-2 H - azetidinone 18 was N - protected as I-butyloxycarbonyl (for easy isolation and for promoting the rate of ring opening). The resulting N - t - Boc - 4S - 2H - azetidenone 19 was hydrolysed (powdered KOH, MeOH) and esterified with diphenyldiazomethane to give the fully protected



Scheme IV

<u>3S</u> - deuteriated  $\beta$ - alanine **20** in essentially quantitative yield. TFA deprotection finally afforded <u>3S</u> -<sup>2</sup> H -  $\beta$  - alanine **6**, crystallised from MeOH. The <sup>1</sup>H NMR showed signals for C - 3 and C - 4 hydrogens at 2.45 and 3.06 integrating in a 2 : 1 ratio, suggesting no loss of deuterium during the chemical manipulations. Definitive evidence was obtained from mass spectral analysis of **6** which showed 98% d<sub>1</sub> species, same as that in the tin deuteride reduction product **17**.

The <u>3R</u> - <sup>2</sup>H -  $\beta$  - alanine 7 was similarly prepared from <u>4R</u> - thiophenylazetidinone 14.

# **Experimental section**

Melting points were determined in open capillaries using a Thomas - Hoover apparatus and are uncorrected. <sup>1</sup>H and <sup>13</sup>C spectra were recorded in XL - 400 instrument. Mass spectra were recorded on a VG 70 - S mass spectrometer, IR spectra were obtained using Perkin - Elmer 1600 series FT instrument. Experimental procedure is given for the synthesis of <u>3S</u> - <sup>2</sup>H -  $\beta$  - alanine 6.

# (3R) - Trimethyisiiyi - (4S) -2 H - 2 - azetidinone (17)

A dry degassed solution of  $15^2$  (502 mg, 2 mmol), <u>tri</u> - n - butyltin deuteride (1.1 mL, 4 mmol) and AlBN (66 mg, 0.4 mmol) in benzene (15 mL) was refluxed under nitrogen for 6 h. After cooling to room temperature, the solution was evaporated, the oily residue was partitioned between acetonitrile and hexane (50 mL each) and the acetonitrile layer was evaporated. The residue was chromatographed using Si - gel. Hexane - EtOAC (2:1) elution furnished the title compound 17 as a white solid, which crystallized

(hexane-CH<sub>2</sub>Cl<sub>2</sub>) as needless (236 mg, 82%); m.p. 71-72°C; [ $\infty$ ] = 158° (CHCl<sub>3</sub>, CI); IR (CHCl<sub>3</sub>) 3412, 3009, 2956, 1737, 1253, 1138, 1110, 1031, 863, 845 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) 5.57 (1H, br, NH), 3.05 (1H, d, J= 2.6Hz, H-3), 2.83 (1H, brS, H-4), 0.14 (9H,S,TMS); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  171.0, 43.0, 36.24(t), -3.11; MS : m/s 129 (M-CH<sub>3</sub>, 23.1%), 116, 101, 99, 86 (100%); accurate mass : Calcd. for C<sub>6</sub><sup>-1</sup>H<sub>12</sub><sup>2</sup>HNOS -CH<sub>3</sub> 129.0594 found 129.0592.

# (4<u>S</u>) - <sup>2</sup>H -2- azetidinone (18)

To a solution of 17 (200 mg, 1.4 mmol) in acetonitrile (10 mL), KF (568 mg, 7 eq) was added and the mixture was stirred at room temperature for 7 hours. It was filtered through a small bed of Si - gel which was thoroughly washed with EtOAc. The combined filtrate and washings were evaporated and the oily residue was chromatographed over Si - gel. The title compound 18 was isolated from the hexane : EtOAc eluates as a white solid crystallized from hexane -CH<sub>2</sub>Cl<sub>2</sub> (98 mg, 97%); m.p. 73-75° C; lit<sup>13</sup>. 73.5-74.5° C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) $\delta$  5.98(1H, br, NH), 3.28(1H, m, H-4), 3.0 (2H, m, H-3); accurate mass : Calcd. for C<sub>3</sub><sup>1</sup>H<sub>4</sub> <sup>2</sup>HNO 72.0433 found 72.0431.

# N-t-Butyloxycarbonyl -(4S)-2H-2-azetidinone (19)

A solution of **18** (72 mg, 1 mmol), di - t - butyldicarbonate (327 mg, 1.5 mmol) and DMAP (15 mg, 0.1 mmol) in acetonitrile (10 mL) was stirred at room temperature for 10h. The solution was then partitioned between EtOAc and water (50 mL each). The EtOAc layer was washed with 0.01 N HCl, 5% NaHCO<sub>3</sub>, brine and dried (Na<sub>2</sub>SO<sub>4</sub>). Filtration followed by evaporation gave an oil from which the title compound **19** was isolated as a colourless oil (165 mg, 96%); IR (CHCl<sub>3</sub>) 2977, 2922, 1733, 1716, 1706, 1602, 1558, 1540, 1506, 1498, 1456, 1368, 1164, 1145 cm<sup>-1</sup>,<sup>1</sup>H NMR (CDCl<sub>3</sub>) 3.53 (1H, t, J = 5.2 Hz, H-4), 2.96 (2H, <u>d</u>, J = 5.2 Hz, H-3), 1.51 (9H, <u>S</u>, COOC(CH<sub>3</sub>)<sub>3</sub>) accurate mass : Calcd. for C<sub>8</sub>H<sub>12</sub><sup>2</sup>HNO<sub>3</sub> 172.1092 found 172.1094.

# N-1-Butyloxycarbonyl-diphenylmethyl-3S-2H-3-amino propanoate(20)

To a solution of 19 (172 mg, 1 mmol) in aqueous MeOH (1:1, 5 mL), powdered KOH (112 mg, 2 mmol) was added and stirred at room temperature for 2h. It was diluted with water (20 mL), the pH was adjusted to 2.3, then extracted with EtOAc (2 x 25 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). Filtration followed by removal of solvent furnished an oily residue which was dissolved in EtOAc(10 mL). Diphenyl diazomethane (291 mg, 1.5 mmol) was added and stirred at room temperature for 5h. The solution was evaporated and the residue was chromatographed over Si-gel. The title compound 20 was isolated as a white solid crystallized from hexane - CH<sub>2</sub>Cl<sub>2</sub> as needles (324 mg, 91%); m.p. 63-64° C; IR (CHCl<sub>3</sub>) 3457,2924,1732,1714,1706,1601,1506,1455,1368,1164,1146 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub> ) 7.38 - 7.23 (10H,m, 2 x Ph), 4.95 (1H, br, NH), 3.40 (1H, m, H-3), 2.65 (2H, d, J = 6.0 Hz, H-2), 1.43 ((OH, S, COOC(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup> C NMR(CDCl<sub>3</sub>) 171.57, 155.76, 139.97, 128.01, 127.01, 79.35, 77.22, 35.87(t), 34.72, 28.36; MS : m/z 300 (M - C<sub>4</sub>H<sub>8</sub>, 16.7%), 282, 272, 256, 185, 184, 183, 168, 167 (100%); accurate mass : Calcd. for C<sub>21</sub><sup>1</sup>H<sub>24</sub><sup>2</sup>H NO<sub>4</sub> - C<sub>4</sub><sup>1</sup>H<sub>8</sub> 300.1220 found 300.1224.

#### (3<u>S</u>)-<sup>2</sup>H - β -alanine (6)

Fully protected  $\beta$  - alanine **20** (50 mg, 0.14 mmol) was dissolved in ice - cold TFA (3 mL) and stirred under nitrogen at 0° C for 15 min. and then at room temperature for another 15 min. The pink solution was evaporated to dryness and the process of evaporation was repeated twice with toluene. The residue was triturated with ether, then taken up in water (5 mL) and lyophilized to give the title compound **8** as a white solid, crystallized from water - EtOH as fine needles (12mg, 96%); m.p. 200 - 202° C (dec), lit<sup>14</sup>. 195 - 200°C (dec); <sup>1</sup>H NMR (D<sub>2</sub>d) 3.06 (t, 1H, J = 6.4 Hz, NCHD), 2.45 (d, 2H, J = 6.4 Hz, CH<sub>2</sub>CO); MS (CI, NH<sub>3</sub>) : m/z 108 (MNH<sub>4</sub><sup>+</sup>, 0.21%), 91 (MH<sup>+</sup>, 100%); accurate mass : Calcd. for C<sub>3</sub><sup>1</sup>H<sub>6</sub><sup>2</sup>HNO<sub>2</sub> 90.0540, found 90.0537.

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