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CONCISE ASYMMETRIC ROUTES TO 2,2,4-TRISUBSTITUTED TETRAHYDROFURANS VIA CHIRAL TITANIUM IMIDE ENOLATES: KEY INTERMEDIATES TOWARDS SYNTHESIS OF HIGHLY ACTIVE AZOLE ANTIFUNGALS SCH 51048 AND SCH 56592

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Abstract: Two complimentary approaches to the key (\cdot) -(2R)-cis-tosylate 1 and its (+)-(2S)-enantiomer 15 via generation of chiral imide enolates having a 2,2-disubstituted olefin functionality in the β -position, are described. In a "protecting group free" sequence, reaction of the titanium enolate generated from (4R)-benzyl-2-oxazolidinone derived imide 5b with s-trioxane provided a convenient intermediate 19 which could be directly subjected to 2,4-diastereoselective iodocyclization. Copyright © 1996 Elsevier Science Ltd

As part of an extensive search for orally effective antifungal agents we recently described a practical synthesis of **Sch 56592** which posesses improved therapeutic potential over **Sch 51048** and other clinically useful agents against a variety of systemic fungal infections in normal and immunocompromised infection models. ^{1,2} Both compounds belong to an uncommon 2,2,4-trisubstituted *cis*-tetrahydrofuran family of orally active broad-spectrum antifungals. In our earlier chemoenzymatic route to these compounds, the key (-)-(2R)-*cis*-tosylate **1** was secured *via* an unprecedented and remarkably efficient 2,4-diastereoselective 5-exo-halocyclization process. The (4S)-diol monoester synthons **3a** and **3b** of high optical purity (~99% ee) were obtained in this case by organic-phase enzymatic acylation of **2** in the presence of Novo SP 435 (Novozyme 435 from Novo Nordisk).²

Enantioselective chemical routes to 1 and its corresponding (+)-(2S)-enantiomer 15 were viewed as being directly accessible via the chiral imide enolate technology developed independently by the Evans³ and Oppolzer⁴ groups. At the outset it was not clear in what manner the unsaturation (arrow) β to the site of enolate derived from 5 would influence the efficiency of electrophilic substitutions; to the best of our knowledge these type of substrates have not been studied earlier. The wide choice of electrophiles, excellent diastereoselectivities for both alkylations and aldol condensations, and easy access to chiral auxiliaries led us to choose the Evans methodology.⁵ We now describe two complimentary routes to 1 using chiral oxazolidinones derived from (S)-valinol and (R)-phenylalaninol respectively.

Synthesis of the desired olefinic acid **7** was accomplished quite simply and in high yields from the allyl alcohol **6** available in bulk from our original chiral epoxide route to **1**.⁶ In a one-pot sequence **6** was subjected to Claisen-Johnson orthoester rearrangement⁷ with triethyl orthoacetate followed by basic hydrolysis of the resulting ethyl ester to provide **7**

as a crystalline solid, m.p. 65° C in 86% yield.⁸ Treatment of the acid chloride **8** with the lithium salt of (4S)-(-)-4-isopropyl-2-oxazolidinone according to standard conditions⁵ readily provided the chiral imide **5a** in over 80% yield.^{8,9} Our first experiments to introduce the benzyloxmethyl funtionality were conducted with the lithium enolate of **5a** using benzyloxymethyl bromide as the alkylating reagent. Although excellent diastereofacial selectivity for the desired benzyl

ether **9** ¹⁰ (over 98:2) was realized, the chemical yields were unacceptably low (~30%). Under these conditions ~40% unchanged **5a** was recovered, plus intractable products. Under the same conditions, saturated substrates have been reported to provide the corresponding benzyl ethers in over 70% yields.⁵

Titanium enolates of N-acyloxazolidinones have been utilized with greatly improved results over Li-enolates in terms of operational simplicity and high diastereofacial selectivity. 11,12 Thus in a marked improvement, alkylation of **5a** with benzyloxymethyl chloride via Evans' titanium enolate protocol 2 gave the benzyl ether **9** in 84% yields (>98% de). No isomerization of the 2,2-disubstituted olefin was detected during either lithium or titanium enolate formation, a requirement crucial to success in this chiral imide route to **1**. Reduction of **9** with LAH in tetrahydrofuran gave the desired (-)-(2S)-diol monobenzyl ether **10**¹³ (85% yield) with ~80% recovery of the chiral auxiliary. The remaining steps from iodocyclization onwards proceeded as expected to provide *cis*-1-N-triazolyl benzyl ether **13** and the corresponding *trans*-isomer **13a** (*cis:trans*; >90:10). 14 The undesired **13a** was readily separated from **13** by chromatography. Hydrogenolytic debenzylation of **13** followed by tosylation of the resulting alcohol **14** provided the desired (-)-(2R)-*cis*-tosylate **1**⁶ in overall 90% yield and excellent optical purity (>99% e.e.). (Scheme I)

In principle, synthesis of the enantiomeric (+)-(2S)-cis-tosylate **15** could be easily carried out by following the above sequence using an appropriate chiral auxiliary such as (4R)-(+)-4-isopropyl-2-oxazolidinone. This was found quite unnecessary as iodocyclization of **11** under the equilibrating conditions according to Rychnovsky and Bartlett¹⁵ provided the iodoacetate of opposite configuration. As we have reported earlier,² under these conditions the 'cis:trans' diastereoselectivity was relatively poor (70:30). However following established conditions, the (+)-(2S)-cis-tosylate **15** could be prepared from the major iodocyclization product after chromatographic separation from the minor *trans*-isomer.

The following alternative approach to 1 was undertaken as we were keen to avoid yield limiting protection-deprotection operations. The allyl alcohol 6 we had been using so far was prepared in 4-steps from 1,3-

difluorobenzene.⁶ In this manner synthesis of the olefinic acid **7** used above required a total of 6-steps. We were able to synthesize **7** in two simple steps by Friedel-Crafts reaction of m-difluorobenzene with succinic anhydride to provide the crystalline keto acid **18**, m.p. 110°C in over 90% yield. Wittig reaction of **18** in tetrahydrofuran with two equivalents of methylene triphenyl phosphorane then gave the olefinic acid **7** in ~60% yield over 2 steps.

The (R)-phenylalaninol derived chiral imide **5b** was produced in 90% yield by activation of **7** with pivaloyl chloride followed by in-situ reaction of the resulting anhydride with the Li-salt of (4R)-(+)-4-benzyl-2-oxazolidinone. Diastereofacially selective hydroxymethylation of **5b** was best carried out with sym-trioxane using titanium enolate chemistry 12 to provide the aldol product **19** in ~60% yield. The direct iodocyclization of **19** at room temperature with a high degree of diastereoselectivity in favor of the desired *cis*-iodo imide **20** (*cis:trans*, >90:10; 90% yield) is noteworthy. The contrast iodocyclization of **3a** or **10** type substrates required lower temperatures (0-5°C) to achieve the same degree of diastereoselectivity. Lithium borohydride reduction of **20** under controlled conditions followed by chromatography provided the *cis*-iodoalcohol **21** and recovered (4R)-benzyl-2-oxazolidinone in ~90% and 71% yields respectively. As we have noted earlier, displacement of iodine in a neopentyl-like system by the highly nucleophilic triazolyl anion posed no problems. Hydroxyl protection used in our earlier study, proved unnecessary. Indeed direct displacement of iodine in **21** with sodium-triazole according to conditions described earlier, provided the alcohol **22** in 75% yield; tosylation then efficiently provided the (-)-(2R)-*cis*-tosylate **1**, thus saving two steps. (Scheme II)

Reagents: (a) Friedel-Craft; (b) Ph₃P=CH₂ (2 equiv.), THF; (c) pivaloyl chloride activation of **6**, oxazolidinone Li-salt, THF, -78°C; (d) **18**, CH₂Cl₂, TiCl₄, 0°C, then s-Trioxane, 0°C; (e) I₂, MeCN, pyridine, 0-5°C; (f) LiBH₄, THF, -78°C to R.T., 2h; (g) Na-triazole, DMF, DMPU, 100°C, 24 h; (h) TsCl, pyridine.

Notably, no protecting groups were required throughout this entire sequence. This concise new route to 1 offers excellent opportunity for large scale operations. These examples further enhance the scope of Evans' imide enolates as well as the unprecedented 2,4-diastereoselective halocyclizations² reported recently.

Based on its superior efficacy and pharmacokinetic profile, **Sch 56592** is in Phase I clinical trials at the present time. ^{1, 19} We shall report on additional examples of diastereoselective bond constructions with chiral imides of the type **5a** and **5b** in future communications. Further chemistry relating to **Sch 56592** and structure-activity relationships in this series of novel orally active broad-spectrum antifungals will also be reported elsewhere.

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- 8. Yields refer to isolated products and have not been optimised. All new compounds were characterised by PMR, CMR and high resolution mass spectra. When necessary diff NOE, COSY and NOESY spectra were obtained. Cis:trans ratios of iodocyclization products were determined by NMR as fully described in ref. 2. Where applicable, optical purity of key intermediates was determined by chiral shift NMR spectroscopy. Chiralcel® OD Analytical Columns (Chiral Technologies Inc.) were also used to determine optical purity of a number of final compounds. Selected spectral data is given here.
- 7: MS = 213 (M+H)+; 1 HNMR [CDCl3] δ 2.48 (t, J = 6.0 Hz, 2H), 2.80 (t, J = 6.0 Hz, 2H), 5.19 (s, 1H), 5.28 (s,1H),6.78 ~ 6.90 (m , 2H), 7.18 ~ 7.30 (m , 1H). 5a: $MS = 324 (M+H)^+$. $^1HNMR [CDCl3] \delta 0.90 (dd, J = 6.7 Hz, 18.0 Hz, 6H), 2.30 ~$ 2.43 (m, 1H), 2.74 ~ 2.90 (m, 2H), 2.95 ~ 3.20 (m, 2H), 4.15 ~ 4.30 (m, 2H), 4.40 ~ 4.47 (m, 1H), 5.20 (s, 1H), 5.30 (s, 1H), 6.77 ~ 6.92 (m, 2H), 7.20 ~ 7.30 (m, 1H).
- 9: $MS = 444 \ (M + H)^{+}$. ¹HNMR [CDCl₃] δ 0.75 (dd, $J = 5.3 \ Hz$, 33.8 Hz, 6H), 2.19 ~ 2.30 (m, 1H), 2.62 (dd, $J \approx 6.0 \ Hz$, 15.0 10. Hz, 1H), 2.91 (dd, J = 7.5 Hz, 12.8 Hz, 1H), 3.60 (dd, J = 6.0 Hz, 10.5 Hz, 1H), 3.66 (t, J = 7.5 Hz, 1H), 4.12 (m, 2H), 4.31 (m, 2H); 4.43 (dd, J = 12.0 Hz, 18.8 Hz, 2 H), 5.10 <math>(s, 1H); 5.23 (s, 1H), 6.70 \sim 6.85 (m, 2H), 7.09 \sim 7.34 (m, 6H).
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- **10**: $\left[\alpha\right]_{D}^{25}$ -28.4° (c = 1.18, CHCl₃). MS = 341 (M+Na)⁺. ¹HNMR [CDCl₃] δ 1.77 ~ 1.91 (m, 1H), 2.42 (t, J = 4.5 Hz, 1H, exchangeable with D_2O_1 , 2.50 (d, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.45 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 ~ 3.75 (m, 3H), 4.46 (dd, J = 7.5 Hz, 2H), 3.52 (dd, J = 6.8 Hz, 9.0 Hz, 1H), 3.52 (dd, J = 6.8 Hz, 1H), 13.5 Hz, 2H); 5.17 (s, 1H), 5.23 (s, 1H), 6.73 ~ 6.86 (m, 2H), 7.12 ~ 7.38 (m, 6H).
- 13 (cis+isomer): $\left[\alpha\right]_{D}^{25} = -42.1^{\circ}$ (c = 1.17, CHCl₃), MS = 386 (M+H)⁺. ¹HNMR [CDCl₃] δ 1.90 (m, 1H), 2.42 (m, 2H), 3.17
 - 13a (trans-isomer): $\left[\alpha\right]_{D}^{25} = +10.6^{\circ}$ (c = 1.12, CHCl₃), MS = 386 (M+H)⁺. ¹HNMR [CDCl₃] δ 1.89 ~ 2.0 (m, 1H); 2.06 ~ 2.21 (m, 1H); 2.55 ~ 2.68 (m, 1H); 3.12 (dd, J = 7.5 Hz, 8.2 Hz, 1H); 3.22 (dd, J = 7.5 Hz, 8.2 Hz, 1H); 3.59 (t, J = 7.9 Hz, 1H); 4.06 (dd, J = 7.5 Hz, 8.2 Hz, 1H); 4.34 (s, 2H); 4.50 (d, J = 15.0 Hz, 2H); 6.70 ~ 6.85 (m, 2H); 7.13 ~ 7.40 (m, 6H); 7.80
 - (s, 1H); 8.05 (s, 1H). 1: $[\alpha]_D^{25} = -39.4^{\circ}$ (c = 1.16, CHCl₃).² 15: $[\alpha]_D^{25} = +37.4^{\circ} = 1.12$, CHCl₃).. S. D. Rychnovsky and P. A. Bartlett, *J. Am. Chem. Soc.* 1981, 103, 3963.
- Procedure: A stirred and cooled (0°C) solution of the chiral imide 5b (2.18 g, 5.88 mmol) in CH2Cl2 (25 ml) was treated with1M TiCl₄ (6.5 ml) in CH₂Cl₂. After 5 minutes Hunig's base (1.12 ml) was added and the solution stirred for an additional 30 minutes. A solution of 1,3,5-trioxane (0.67 g, 7.44 mmol) in CH₂Cl₂ (5 ml) was added followed by another 6.5 ml 1M TiCl₄ in CH₂Cl₂. The reaction mixture was stirred for 2.5 h at 0^o-3^oC. Addition of saturated NH₄Cl and extractive work-up gave the crude product. Chromatography on silica gel using 15-25% EtOAc in n-hexane as eluent provided pure 19 (1.33 g).
 - **5b**: $\left[\alpha\right]_{D}^{25} = -44.4^{\circ} \text{ (c = 1.67, CHCl_3); MS = 371 (M+H)+}. \ ^{1}\text{HNMR [CDCl_3]} \ \delta \ 2.74 (dd, J = 9.6 Hz, 13.3 Hz, 1H), 2.82 \sim 10^{-2} \text{ m}^{-2}$ $3.12 \text{ (m, $\overline{4}$H)}, 3.28 \text{ (dd, J} \approx 3.4 \text{ Hz, } 13.3 \text{ Hz, } 1\text{H)}, 4.13 \sim 4.19 \text{ (m, } 2\text{H)}, 4.58 \sim 4.71 \text{ (m, } 1\text{H)}, 5.20 \text{ (s, } 1\text{H)}, 5.31 \text{ (m, } 1\text{H)}, 6.74 \sim 1.00 \text{ (m, } 1\text{H)}, 5.20 \text{ (m, } 1\text{H)}, 5.20 \text{ (m, } 1\text{H)}, 5.31 \text{ (m, } 1\text{H)}, 6.74 \sim 1.00 \text{ (m, } 1\text{H)}, 6.$ 7.34 (m, 8H).
 - 19: $[\alpha]D = -62.9^{\circ}$ (c = 1.7, CHCl₃); MS = 402 (M+H)⁺. HNMR [CDCl₃] δ 2.64 ~ 3.01 (m, 3H), 3.24 (dd, J = 3.4 Hz, 13.5 Hz, 1H), 3.77 ~ 3.91 (bs, 2H), 4.0 ~ 4.18 (m, 3H), 4.49 ~ 4.61 (m, 1H), 5.19 (s, 1H), 5.33 (s, 1H); 6.73 ~ 7.4 (m, 8H).
- Procedure: A stirred solution of 19 (1 g, 2.5 mmol) in CH3CN (20 ml) was treated with pyridine (0.45 ml). After cooling to 0°C iodine (1.78 g) was added allowing the reaction mixture to warm to room temperature. After ~20 h the reaction was quenched with dilute aq. sod sulfite and extracted with Et₂O (2 x 20 ml); the ether extract was dried over MgSO₄ and concentrated to dryness in vacuo to provide a residue. Chromatography over silica gel using 50-75% EtOAc in hexane as eluent gave pure iodo-imide **20** (1.2 g). MS = 528 (M+H) $^+$. HNMR [CDCl₃] δ 2.61 ~ 2.95 (m, 3H), 3.28 (dd, J = 3.4 Hz, 13.3 Hz, 1H), 3.79 (AB quartet, J = 16.7 Hz, 2H), $4.05 \sim 4.33$ (m, 5H), $4.65 \sim 4.76$ (m, 1H), $6.75 \sim 7.61$ (m, 8H).
- Procedure: The iodo-imide 20 (0.9 g) from above was dissolved in THF and the solution cooled to -780C. A 2M LiBH4 solution in THF (0.85 ml) was then added with stirring while warming to room temperature. After stirring at room temperature for 2 h, the reaction mixture was cooled to -100C and quenched with saturated NH4Cl. Evaporation of THF in vacuo followed by extractive isolation with CH₂Cl₂ gave the iodo alcohol 21 (0.45 g). Its acetate was identical to the authentic material.²
- 19. 35th Interscience Conference on Antimicrobial Agents and Chemotherapy (ICAAC), San Francisco, 17-20 September1995 (abstract nos. F 61, F 62, F63, F 64, F 65, F 66, F 67, F 68, and F 83).