

Isocyanates, Part 5.¹
Synthesis of Chiral Oxazolidin-2-ones and Imidazolidin-2-ones via DMAP-Catalyzed Isocyanation of Amines with Di-*tert*-butyl Dicarbonate
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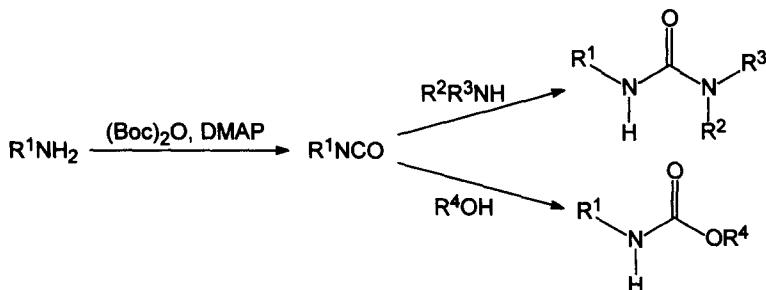
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Abstract: Oxazolidin-2-ones and imidazolidin-2-ones are prepared under mild reaction conditions by DMAP-catalyzed isocyanation of 1,2-aminoalcohols and 1,2-diamines with di-*tert*-butyl dicarbonate and subsequent cyclization.

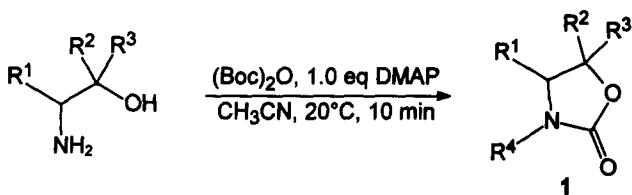
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Optically active oxazolidin-2-ones derived from α -amino acid esters are an important class of chiral auxiliaries for asymmetric synthesis.² They were applied for example to the total syntheses of the macrolide antibiotic rutamycin B³ and the immunosuppressant (−)-FK-506.⁴ More recently, enantiopure imidazolidin-2-ones were used as chiral auxiliaries for dynamic kinetic resolution.⁵ The most common method for the preparation of oxazolidin-2-ones and imidazolidin-2-ones is the reaction of 1,2-aminoalcohols and 1,2-diamines with diethyl carbonate,⁶ phosgene,⁷ triphosgene,⁸ or 1,1'-caronyldiimidazole.⁹

We recently reported a novel procedure for the synthesis of isocyanates under mild conditions (10 min at room temperature) by a DMAP-catalyzed¹⁰ reaction of amines¹¹ and α -amino acid esters¹ with di-*tert*-butyl dicarbonate, $(\text{Boc})_2\text{O}$. *In situ* derivatization of the isocyanates by addition of amines and alcohols affords the corresponding ureas¹² and carbamates¹³ (Scheme 1).

**Scheme 1**

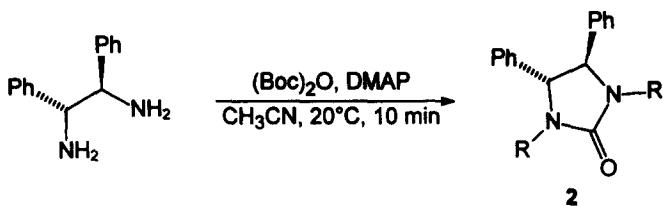
An intramolecular version of the above-mentioned *in situ* addition of amines and alcohols should provide an easy access to a variety of cyclic ureas and cyclic carbamates. Thus, using the standard set of reaction conditions for the isocyanation of amines, $(\text{Boc})_2\text{O}$, DMAP, 10 min, 20°C, a broad range of enantiopure 1,2-aminoalcohols were converted into the corresponding oxazolidin-2-ones **1** (Table 1).

**Table 1.** Synthesis of enantiomerically pure oxazolidin-2-ones **1**.

	R ¹	R ²	R ³	R ⁴	(Boc) ₂ O [eq]	1 , Yield [%]	[α] _D ²⁰ , c=1, solvent
a	(<i>R</i>)-Et	H	H	H	1.1	85	+5.9°, CHCl ₃
b	(<i>S</i>)- <i>i</i> Pr	H	H	H	1.1	71 ^a	-20.0°, EtOH
c	(<i>S</i>)- <i>t</i> Bu	H	H	H	1.1	90	-19.6°, EtOH
d	(<i>R</i>)-Ph	H	H	H	1.1	66	-60.4°, CHCl ₃
e	(<i>R</i>)-CH ₂ Ph	H	H	H	1.1	63	+64.9°, CHCl ₃
f	(<i>R</i>)-Ph	(<i>S</i>)-Ph	H	H	1.1	80	+80.9°, CHCl ₃
g	(<i>S</i>)-Me	Ph	Ph	H	1.1	65	-298.9°, DMF
h	(<i>R</i>)-Et	H	H	Boc	2.1	81 ^b	-36.4°, CHCl ₃
i	(<i>S</i>)- <i>i</i> Pr	H	H	Boc	2.1	85 ^b	+49.9°, CHCl ₃
j	(<i>S</i>)- <i>t</i> Bu	H	H	Boc	2.1	95 (72) ^c	+41.9°, EtOH
k	(<i>R</i>)-Ph	H	H	Boc	2.1	85	-68.9°, CHCl ₃
l	(<i>R</i>)-CH ₂ Ph	H	H	Boc	2.1	84	-19.4°, CHCl ₃
m	(<i>R</i>)-Me	(<i>R</i>)-Ph	H	Boc	2.1	82	-70.4°, CHCl ₃
n	(<i>R</i>)-Ph	(<i>S</i>)-Ph	H	Boc	2.1	92 (78) ^c	+74.0°, CHCl ₃
o	(<i>S</i>)-Me	Ph	Ph	Boc	2.1	81	-252.3°, CHCl ₃

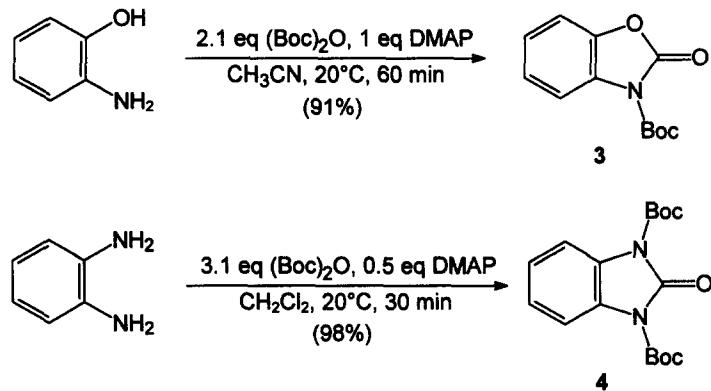
^a Solvent: CH₂Cl₂. ^b Solvent: CH₂Cl₂, reaction time: 60 min. ^c Catalytic reaction with 0.1 eq DMAP.

Dependent on the amount of di-*tert*-butyl dicarbonate (1.1 or 2.1 eq) either the *N*-unsubstituted oxazolidin-2-ones **1a-g** or the *N*-Boc-oxazolidin-2-ones **1h-o** were obtained as products. In the synthesis of the *N*-unsubstituted heterocycles the *N*-Boc- and *N,O*-bis-Boc-1,2-aminoalcohols were formed as by-products due to competing *tert*-butoxycarbonylation of starting material. Again, we noted that high sterical demand of substituents in the α- or β-position of the amino group leads to higher yields of the isocyanates and their consecutive products. Therefore, (*S*)-2-amino-3,3-dimethyl-1-butanol (*tert*-valinol) and (1*S*,2*R*)-2-amino-1,2-diphenylethanol gave the best results. The transformation to the oxazolidin-2-ones **1** was also achieved by using catalytic amounts of DMAP. The values for the optical rotation of **1a-e** and **1l** were in agreement with those reported in the literature,¹⁴ indicating that the cyclization to the oxazolidin-2-ones occurred without partial racemization. The method was also used for the transformation of 1,2-diamines to the imidazolidin-2-ones **2** (Table 2). In this case the intramolecular trapping is much more efficient because of the higher nucleophilicity of the amine. Thus, (1*R*,2*R*)-1,2-diamino-1,2-diphenylethane was quantitatively converted to the *N,N'*-bis-Boc-substituted cyclic urea **2a** with 3.1 eq of di-*tert*-butyl dicarbonate using stoichiometric or catalytic amounts of DMAP. The value for the optical rotation of the *N,N'*-unsubstituted imidazolidin-2-one **2b** was in agreement with that reported in the literature.¹⁵

**Table 2.** Synthesis of (4*R*,5*R*)-4,5-diphenylimidazolidin-2-ones **2**.

	R	(Boc) ₂ O [eq]	DMAP [eq]	2 , Yield [%]	[α] _D ²⁰ , c=1, solvent
a	Boc	3.1	1.0	94	+30.1°, CHCl ₃
a	Boc	3.1	0.1	96	+30.1°, CHCl ₃
b	H	1.1	1.0	81	+53.3°, CHCl ₃

For **1a** and **2b** the formation of an intermediate β -hydroxy and β -amino isocyanate was confirmed by following the reaction using FT-IR spectroscopy. A band for the intermediate isocyanate occurred at 2273 and 2274 cm⁻¹ respectively, increased to a maximum (after a reaction time of 29 s in case of **1a**), and decreased again due to cyclization to the heterocycles. Subsequent to the cyclization an additional *N*-*tert*-butoxycarbonylation of the cyclic carbamates and ureas occurred if an excess of di-*tert*-butyl dicarbonate was applied. The efficiency of the DMAP/(Boc)₂O reagent for the *N*-Boc protection of amides and carbamates is well-known.¹⁶

**Scheme 2**

The corresponding benzo-annulated heterocycles are also available by the present method. Using 2.1 eq di-*tert*-butyl dicarbonate and stoichiometric amounts of DMAP *o*-aminophenol afforded the *N*-Boc-protected benzoxazolin-2-one **3** in 91% yield.¹⁷ Reaction of *o*-phenylenediamine with 3.1 eq (Boc)₂O and substoichiometric amounts of DMAP provided quantitatively the *N,N'*-bis-Boc-benzimidazolin-2-one **4** (Scheme 2).¹⁷

In conclusion, a novel method for the synthesis of oxazolidin-2-ones and imidazolidin-2-ones under mild reaction conditions was developed using the DMAP/(Boc)₂O reagent system. Either the *N*-unsubstituted or the *N*-Boc-protected heterocycles were obtained dependent on the amount of (Boc)₂O. The present one-pot procedure provides an improved route to Boc-protected chiral oxazolidin-2-one auxiliaries.¹⁸

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17. **3:** colorless crystals, mp 78°C; ^1H NMR (500 MHz, CDCl_3): δ = 1.65 (s, 9 H), 7.13-7.19 (m, 3 H), 7.65 (m, 1 H); ^{13}C NMR and DEPT (125 MHz, CDCl_3): δ = 27.87 (3 CH_3), 86.03 (C), 109.82 (CH), 114.50 (CH), 124.25 (CH), 124.62 (CH), 127.35 (C), 141.56 (C), 147.43 ($\text{C}=\text{O}$), 149.26 ($\text{C}=\text{O}$). Analysis calcd. for $\text{C}_{12}\text{H}_{13}\text{NO}_4$: C 61.27, H 5.57, N 5.95; found: C 61.51, H 5.71, N 5.92.
18. **4:** colorless crystals, mp 144°C; ^1H NMR (400 MHz, CDCl_3): δ = 1.66 (s, 18 H), 7.20 (m, 2 H), 7.85 (m, 2 H); ^{13}C NMR and DEPT (100 MHz, CDCl_3): δ = 28.02 (6 CH_3), 85.30 (2 C), 113.99 (2 CH), 124.29 (2 CH), 126.10 (2 C), 147.32 ($\text{C}=\text{O}$), 148.44 (2 $\text{C}=\text{O}$). Analysis calcd. for $\text{C}_{17}\text{H}_{22}\text{N}_2\text{O}_5$: C 61.07, H 6.63, N 8.38; found: C 61.07, H 6.70, N 8.29.
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