CHEMISTRY AN ASIAN JOURNAL

www.chemasianj.org

Accepted Article

Title: Synthesis and Configuration of Neomaclafungin A

Authors: Shijun Zhu and Yikang Wu

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: Chem. Asian J. 10.1002/asia.201700950

Link to VoR: http://dx.doi.org/10.1002/asia.201700950

A Journal of

ACES Asian Chemical Editorial Society A sister journal of Angewandte Chemie and Chemistry – A European Journal



WILEY-VCH

Synthesis and Configuration of Neomaclafungin A

Shijun Zhu^[a] and Yikang Wu^{*[a]}

Dedication ((optional))

Abstract: The relative and absolute configuration of neomaclafungins were impossible to establish by spectroscopic analyses alone because of the lack of exploitable ¹H-¹H couplings and nOe's between the upper and the lower subunits. This very difficult task now is finally completed by an enantioselective total synthesis of neomaclafungin A (revised) and its diastereomer (reported). The results also provided a key reference for the complete structures for other neomaclafungins and the long-known closely related natural product maclafungin.

Neomaclafungins (**1a-i**, Figure 1) were isolated in 2012 by Sato and coworkers^[1] from *Actinoalloteichus* sp. NPS702 (found in marine sediment). In the preliminary testing, **1a-i** showed significant antifungal activity (MIC 1-3 μ g/mL) against *Trichophyto*n mentagrophytes (ATCC9533). The gross structures of



Figure 1. The structures of neomaclafungins A-I (1a-i, deduced from ref 1 with correction at C-5; disproved in this work), maclafungin (2) and oligomycin A (3).

 S. Zhu, Prof. Dr. Y. Wu
 State Key Laboratory of Bioorganic and Natural Products Chemistry, Collaborative Innovative Center for Chemistry and Life Sciences, Shanghai Institute of Organic Chemistry and the University of Chinese Academy of Sciences, Chinese Academy of Sciences
 345 Lingling Road, Shanghai 200032, China
 E-mail: yikangwu@sioc.ac.cn

Supporting information for this article is given via a link at the end of the document.((Please delete this text if not appropriate))

neomaclafungins were assigned on the basis of spectroscopic analyses of neomaclafungin A (the most abundant component), which differ from maclafungin^[2] (2) only at the C-4 and C-33, with the lower (spiroketal) subunit very similar to that in oligomycin A^[3] (3). The relative configurations for the upper (C-1 to C-17) and the lower fragments of neomaclafungin A were assigned separately by analysis of the ¹H-¹H coupling constants. However, lack of any exploitable ¹H-¹H couplings between the two arrays of stereocenters made it impossible to establish a relative configuration for the whole molecule (cf. configurations of oligomycins were determined by X-ray). To overcome this difficulty, an enantioselective synthesis appeared to be essential



Scheme 1.

Our initial retrosynthetic analysis is shown in Scheme 1. The target **1a** was planned to be disconnected first into the lower fragment **4** and the upper fragment **5**, each of which was then further disconnected into readily accessible fragments **6-13**.



Scheme 2. Reagents and conditions: a) (i) Brown asy. crotylation, (ii) TBSCI, DMAP, imidazole, 70% from **6**; b) (i) 9-BBN, THF, 50 °C, (ii) **7**, DMF, Cs_2CO_3 . Pd(PPh₃)₄, 50 °C, 90% from **14**; c) NBS, CH₂Cl₂, DMF, propylene oxide, -72 °C, 98%. TBS = *t*-Butyldimethylsilyl, DMAP = 4-(dimethylamino)pyridine, 9-BBN = 9-borabicyclo[3.3.1]nonane, NBS = *N*-Bromosuccinimide.

Synthesis of fragment **4** began as shown in Scheme 2. Aldehyde **6**^[4] was subjected to Brown^[5] crotylation and TBS protection to afford **14**, which on hydroboration with 9-BBN at 50 °C^[6] followed by a Suzuki coupling with **7** gave **15**.^[7] Further treatment with NBS delivered the corresponding bromide **16**.^[8]



Scheme 3. Reagents and conditions: a) TiCl₄, iPr_2NEt , NMP, CH₂Cl₂, -98 °C to 0 °C, 80%; b) MeNH(OMe)·HCl, AlMe₃, THF, 0 °C, 86%; c) TBSOTf, 2,6-lutidine, CH₂Cl₂, 88%; d) DIBAL-H, THF, 0 °C, 82%; e) **15**, EtAlCl₂, *n*Bu₄NSiPh₃F₂, CH₂Cl₂, -78 °C, 47%; f) (i) OsO₄, NMO, PhI(OAc)₂, 76%, (ii) aq. HClO₄, MeCN-CH₂Cl₂, 0 °C, 58%. NMP = *N*-methylpyrrolidinone, DIBAL-H = diisobutylaluminum hydride.

The C-19 to C-25 moiety of fragment **4** was achieved as shown in Scheme 3. Evans aldol condensation of **9**^[9] with **17** under the Crimmins^[10a] conditions (TiCl₄//Pr₂NEt/NMP^[10b,11]) at – 98 °C gave **18** in 80% yield. Removal of the chiral auxiliary in **18** with MeN(OMe)·HCl/AIMe₃^[12] afforded **19**, which on TBS protection and DIBAL-H reduction gave **21**.^[13] Subsequent coupling with **15** in the presence of EtAICl₂^[14a]/*n*Bu₄NSiPh₃F₂^[14b] afforded **22** (47%), the C-25 configuration of which was determined by the *J*_{H-25/H-26} (< 8 Hz) in **23** (by treatment with first OsO₄ and then HClO₄)^[15] and opposite to Evans^{*[14a]} results.

Reaction of **16** with **21** mediated by $CrCl_2/NiCl_2^{[16]}$ (Scheme 4) led to **22** (59%) and *epi-***22** (19%). The undesired **22** was oxidized and stereoselectively reduced with DIBAL-H (cf LiAlH₄ gave a 1:1 mixture) to afford *epi-***22**, which on sequential exposure to OsO₄/NMO/PhI(OAc)₂^[17] and CSA^[18] gave ketal **25**. It is noteworthy that either HClO₄ (worked well for **22**) or HF (worked well^[14c] for oligomycin C) resulted in only traces of **25**.

Diol **25**, with its C-25 OH configuration proven by the nOe and $J_{H-25/H-26a}$ (11.8 Hz), was then selectively oxidized at C-19 with RuCl₂(PPh₃)₃^[19a]. The resulting aldehyde reacted with CrCl₂/ CHI₃^[19b,20] to afford **26**. Finally, a Yamaguchi^[21] esterification and removal^[22] of the PMB furnished the desired lower fragment **4**.



Scheme 4. Reagents and conditions: a) $CrCl_2$, $NiCl_2$, THF, rt, 12 h, 19% for *epi-*22, 59% for 22; b) Dess-Martin oxidation, 89%; c) DIBAL-H, –78 °C, 84%; d) PhI(OAc)₂, OsO₄, NMO, acetone-H₂O, 96%; e) CSA, MeCN-H₂O, 58%; f) (i) RuCl₂(PPh₃)₃, toluene, rt, 77%, (ii) $CrCl_2$, CHI_3 , THF, dioxane, rt, 56%; g) Yamaguchi esterification; h) DDQ, rt, $CH_2Cl_2-H_2O$, 100%. CSA = D-camphorsulfonic acid, DDQ = 2,3-dicyano-5,6-dichlorobenzoquinone.



Scheme 5. Reagents and conditions: a) CuBr·Me₂S, THF, -20 °C to 0 °C, 98%; b) VO(acac)₂, *t*BuOOH, CH₂Cl₂, 0 °C, 24 h, 87% (dr 20:1); c) NaH, Mel, DMF, 80%; d) *n*BuLi, THF, rt, 40% for **31a**, 20% for **31b**; e) TBSCI, Et₃N, DMAP, 93%; f) *n*BuLi, THF, rt, 30% for **33a**, 68% for **33b**; g) Hg(ClO₄)₂, CaCO₃, THF-H₂O, 58%; h) (i) *c*Hex₂BCI, Et₃N, Et₂O, -30 °C (dr 20:1), (ii) LiBH₄, -78 °C; i) PPTS, Me₂C(OMe)₂, 75% from **34**. PPTS = Pyridium *p*-toluenesulfonic acid.

To access fragment **5** (Schemes 5-6), **10**^[23] was treated with **11**^[24] to afford **28**. Epoxidation of **28** using VO(acac)₂/TBHP^[25] gave **29** (dr 20:1). The OH in **29** was masked as OTBS (to revert the abnormal regioselectivity observed in reaction of **12**^[26] with **30** caused by the OMe, cf the inset in Scheme 5) to provide **33b** in 68%.^[27] Treatment of **33b** with Hg(ClO₄)₂/ CaCO₃^[28] furnished **34**. Finally, reduction of **34** with *c*-hex₂BCl/LiBH₄^[29] and the following acetonization of the diol afforded **36** (the hydroxyl configuration of which was confirmed by the nOe between H-11/H-13).



Scheme 6. Reagents and conditions: a) nBu_4NF , THF, rt, 88%; b) (i) NaH, MeI, DMF, (ii) DDQ, CH₂Cl₂, pH 7 buffer, 91% from **37**; c) Dess-Martin oxidation, 90%; d) (i) **40**, Et₃N, Sn(OTf)₂, CH₂Cl₂, -20 °C, (ii) **39**, -78 °C, 88% from **40**, dr 18:1; e) (i) DIBAL-H, Et₂O, -98 °C, (ii) CSA, Me₂C(OMe)₂, 74% from **41**; f) LiAlH₄, THF, -78 °C, 7 h, 90%; g) (i) **44**, Grubbs I catalyst, CH₂Cl₂, 51%, (ii) Dess-Martin oxidation, 30%; h) (i) Dess-Martin oxidation, 69%, (ii) **44**, Grubbs I catalyst, CH₂Cl₂, 71%. DMF = *N*,*N*-dimethylformamide.



Scheme 7. Reagents and conditions: a) Pd(PPh₃)₄, Ag₂O, THF-H₂O, rt, 12 h, 70%; b) K₂CO₃, 18-crown-6, toluene, 48 h, 33%; c) (i) 65% AcOH, rt, 4 h, (ii) 80% AcOH, 80 °C, 30 min, 51% from **46**.

Conversion of **36** into **41** was achieved by removal of the TBS, methylation of the hydroxyl group with Mel/NaH, cleavage of PMB, Dess-Martin oxidation^[30] and a tin-mediated^[31] Evans aldol condensation (Scheme 6). Reduction of **41** with DIBAL-H^[31] at –98 °C (dr > 20:1) followed by acetonide protection gave **42**, which on removal^[31] of the chiral auxiliary, oxidation and coupling^[32] with **44** afforded the upper fragment **5**.

A Suzuki coupling^[33] of **4** with **5** (Scheme 7) was then performed to afford **45**.^[34] Subsequent exposure of **45** to $K_2CO_3/18$ -c-6/PhMe^[35] afforded **46** as a 2:1 mixture of the (*E*)/(*Z*) isomers, which was converted into the end product **1a** using a two-stage^[36] hydrolysis strategy (also a 2:1 (*E*)/(*Z*) mixture) after repeated failures^[37] using conventional protocols.

Unexpectedly, the ¹H NMR of **1a** was incompatible with that for natural neomaclafungin A. Therefore, *ent-***5** was next synthesized (cf. Supporting Information) and coupled with **4** in the same manner (Scheme 8). Under the same conditions, **48** formed as a single isomer (rather than an (*E*)/(*Z*) mixture). However, the final hydrolysis became much more difficult (with unidentified side products predominated) than hydrolysis of **46**. Despite our exhaustive efforts, the yield for **49** could not exceed 10%, illustrating a previously unknown differential reactivity caused by the relative configuration between two remotely located subunits. Nevertheless, the ¹H and ¹³C NMR as well as [α]_D for **49** were consistent with those for natural neomaclafungin A, proving that these two compounds have the same configuration.



Scheme 8. Reagents and conditions: a) Pd(PPh₃)₄, Ag₂O, THF-H₂O, rt, 12 h, 77%; b) K₂CO₃, 18-crown-6, toluene, rt, 48 h, 38%; c) (i) 65% AcOH, rt, 4 h, (ii) 80% AcOH, 80 °C, 30 min, 10% from **48**.

In summary, the first synthesis of neomaclafungin A was completed. The relative and absolute configurations for neomaclafungins, which previously were impossible to assign due to lack of exploitable ¹H-¹H couplings/nOe's between the two major fragments, were eventually established. The results also provided direct configuration reference for other neomaclafungins and maclafungin and offered valuable opportunities to observe the interesting reactivity differences caused by the relative configurations (e. g., between **45** and **47**, also **46** and **48**). The regioselectivity of the opening of epoxides **30** and **32**, the reverted stereoselectivity of addition of **15** to **21**, the various problems caused by a substituent larger than methyl group at the C-24 (compared with oligomycins) and the tricky hydrolysis of the acetonides in **46** and **48** are also noteworthy.

Acknowledgments: Prof. Ang Li (SIOC) and Prof. Zhujun Yao/Dr. Xiaoliang Yang (Nanjing University) are thanked for access to HPLC facility and recording the NMR of **49**, respectively. This work was supported by the NSFC (21372248, 21532002, 21672244) and the Strategic Priority Research Program of the Chinese Academy of Sciences (XDB20020200).

Keywords: Configuration• macrolides • condensation• total synthesis •Natural products

- [1] a) S. Sato, F. Iwata, S. Yamada, M. Katayama, J. Nat. Prod. 2012, 75, 1974-1982; b) Two full structures were given for neomaclafungins in ref 1a (both were incorrect); the correct relative configuration for the upper fragment (with absolute configuration opposite to that proven in this work) was only found in the truncated structure in Figure 2 therein.
- [2] T. Mukhopadhyay, S. R. Nadkarni, M. V. Patel, R. G. Bhat, K. R. Desikan, B. N. Ganguli, R. H. Rupp, H.-W. Fehlhaber, H. Kogler, *Tetrahedron* **1998**, *54*, 13621-13628.
- [3] a) G. T. Carter, J. Org. Chem. 1986, 51, 4264-4271; b) J. S. Panek, N.
 F. Jain, J. Org. Chem. 2001, 66, 2747-2756; c) J. S. Panek, N. F. Jain,
 J. Org. Chem. 1998, 63, 4572-4573.
- [4] P.-Y. Dakas, R. Jogireddy, G. Valot, S. Barluenga, N. Winssinger, *Chem. Eur. J.* 2009, 15, 11490-11497.
- [5] H. C. Brown, K. S. Bhat, J. Am. Chem. Soc. 1986, 108, 5919-5923.
- [6] At ambient temperature the reaction never went to completion.
- [7] D. Lapinsky, S. C. Bergmeier, Tetrahedron 2002, 58, 7109-7117.
- [8] M. Xuan, I. Paterson, S. M. Dalby, Org. Lett. 2012, 14, 5492-5495.
- [9] M. E. Kuehne, W. G. Bornmann, J. Org. Chem. 1989, 54, 3407-3420.
- [10] a) M. T. Crimmins, B. W. King, E. A. Tabet, *J. Am. Chem. Soc.* **1997**, *119*, 7883-7884; b) M. T. Crimmins, J. She, *Synlett* **2004**, 1371-1374; c) The yield for **18** under the *n*Bu₂OTf/Et₃N/–78 °C or TiCl₄/*i*Pr₂NEt/–78 °C conditions was 32% and 62%, respectively.
- [11] T. Andreou, A. M. Costa, L. Esteban, L. Gonzàlez, G. Mas, J. Vilarrasa, Org. Lett. 2005, 7, 4083-4086 (a more or less similar case).
- [12] D. A. Evans, S. L. Bender, J. Morris, J. Am. Chem. Soc. 1988, 110, 2506-2526.
- [13] Addition of the Grignard reagent of **16** to **20** failed.
- [14] a) D. A. Evans, B. D. Allison, M. G. Yang, C. E. Masse, J. Am. Chem. Soc. 2001, 123, 10840-10852; b) The yield for 22 was 20% in the absence of nBu₄NSiPh₃F₂ (which has never been reported for this reaction before); c) It is noteworthy that the -(CH₂)₃OPMB at the C-24 instead of a Me as in the Evans' case reverted the stereoselectivity. The spiroketal formation was also retarded by the C-24 -(CH₂)₃OPMB (compared with the facile synth. of the spiroketal in oligomycins).
- [15] Use of F₃B·Et₂O to replace EtAlCl₂ led to a 1:1 mixture of C-25 epimers in 40% yield. Under the Yamamoto's conditions (see, K. Ishihara, M.

Mouri, Q. Gao, T. Maruyama, K. Furuta, H. Yamamoto, *J. Am. Chem.* Soc. **1993**, *115*, 11490-11495), neither **22** nor *epi-***22** was detected.

- [16] a) H. Jin, J. Uenishi, W. J. Christ, Y. Kishi, J. Am. Chem. Soc. 1986, 108, 5644-5646; b) R. D. Cink, C. J. Forsyth, J. Org. Chem. 1997, 62, 5672-5673.
- [17] K. C. Nicolaou, V. A. Adsool, C. R. H. Hale, Org. Lett. 2010, 12, 1552-1555.
- [18] I. Paterson, A. D. Findlay, E. A. Anderson, Angew. Chem. Int. Ed. 2007, 46, 6699-6702
- [19] a) H. Tomioka, K. Takai, K. Oshima, H. Nozaki, *Tetrahedron Lett.* 1981, 22, 1605-1608; b) J. S. Panek, N. F. Jain, *J. Org. Chem.* 1998, 63, 4572-4573; c) TEMPO oxidation of 25 failed.
- [20] K. Takai, K. Nitta, K. Utimoto, J. Am. Chem. Soc. 1986, 108, 7408-7410.
- [21] M. Yamaguchi, J. Innaga, K. Hirata, H. Saeki, T. Katsuki, Bull. Chem. Soc. Jpn. 1979, 52, 1989-1993.
- [22] K. Horita, T. Yoshioka, T. Tanaka, Y. Oikawa, O. Yonemitsu, *Tetrahedron* **1986**, *42*, 3021-3028.
- [23] K. Mori, Y. Shikichi, S. Shankar, J. Y. Yew, *Tetrahedron* 2010, 66, 7161-7168.
- [24] a) B. M. Trost, J. P. N. Papillon, T. Nussbaumer, J. Am. Chem. Soc. 2005, 127, 17921-17937; b) J. Kant, J. Org. Chem. 1993, 58, 2296-2301.
- [25] a) B. H. Lipshutz, J. Varton, *J. Org. Chem.* **1988**, *53*, 4495-4499; b) R.
 W. Bates, R. Fernandez-Moro, S. V. Ley, *Tetrahedron* **1991**, *47*, 9929-9938.
- [26] N. Proisy, S. Y. Sharp, K. Boxall, S. Connelly, S. M. Roe, C. Prodromou, A. M. Z. Slawin, L. H. Pearl, P. Workman, C. J. Moody, *Chemistry & Biology* **2006**, *13*, 1203-1215.,
- [27] M. Ide, K. Tsunashima, M. Nakata, *Bull. Chem. Soc. Jpn.* **1999**, 72, 2501-2507. The regioselectivity in the absence of large difference in steric crowding/reactivity (caused by e. g. allylic activation) between the two epoxy sites observed with **30** is noteworthy.
- [28] H. Watanabe, H. Watanabe, M. Bando, M. Kido, T. Kitahara, *Tetrahedron* **1999**, 55, 9755-9776; Mel/CaCO₃, CuCl₂/CuO, Phl(OAc)₂ and Phl(OCOCF₃)₂ all failed.
- [29] a) P. Li, J. Li, F. Arikan, W. Ahlbrecht, M. Dieckmann, D. Menche, J. Am. Chem. Soc. 2009, 131, 11678-11679; b) Reduction using Et₂BOMe/Li(or Na)BH₄/–78°C on 10 mg scale led to a 2:1 mixture of 35 and *epi-*35. The ratio was reverted to 1:2 on 100 mg scale. DIBAL-H/– 98°C gave a syn diol with the TBS migrated.
- [30] D. B. Dess, J. C. Martin, J. Org. Chem. 1983, 48, 4155-4156.
- [31] a) D. A. Evans, J. S. Clark, R. Metternich, V. J. Novack, G. S. Sheppard, J. Am. Chem. Soc. 1990, 112, 866-868; b) Use of Na(or Li)BH₄/THF-MeOH (or H₂O) led to partial racemization at the C-4.
- [32] a) P. Schwab, M. B. France, J. W. Ziller, R. H. Grubbs, *Angew. Chem.*, *Int. Ed.* **1995**, *34*, 2039-2041; b) R. H. Grubbs, S. J. Miller, G. C. Fu, *Acc. Chem. Res.* **1995**, *28*, 446-452.
- [33] a) M. Tortosa, N. A. Yakelis, W. R. Roush, *J. Org. Chem.* 2008, *73*, 9657-9667; b) H. Lei, J. Yan, J. Yu, Y. Liu, Z. Wang, Z. Xu, T. Ye, *Angew. Chem. Int. Ed.* 2014, *53*, 6533-6537.
- [34] Use of 4' in Suzuki coupling with 5 led to <10% yields and removal of PMB at a later stage was unfeasible. We are not aware of any precedents involving substrates with an aliphatic aldehyde group.
- [35] a) N. Hayashi, T. Suzuki, K. Usui, M. Nakada, *Tetrahedron* 2009, *65*, 888-895; b) M. A. Blanchette, W. Choy, J. T. Davis, A. P. Essenfeld, S. Masamune, W. R. Roush, T. Sakai, *Tetrahedron Lett.* 1984, *25*, 2183-2186 (the LiCl/iPrNEt/MeCN conditions, failed in this work).
- [36] Use of PPTS/MeOH-H₂O/rt resulted in a side product (with one less C-C double bond and an extra THP ring). HS(CH₂)₃SH/p-TsOH/CHCl₃ or 60% aq. AcOH/rt gave a mixture (one acetonide remained). 80% aq AcOH/80 °C/30 min led to a side product (same as that obtained with PPTS) but no **1a**. Using the two-stage hydrolysis prevented the formation of the undesired THP ring. Storage of **1a** in CDCl₃ at rt for more than 24 h led to significant deterioration.
- [37] Either no reaction occurred or a complex mixture was resulted.

WILEY-VCH

Entry for the Table of Contents (Please choose one layout)

Layout 1:

COMMUNICATION

Although the relative configurations of the upper and lower fragments of neomaclafungin A were assigned, establishment of the configuration for the whole molecule remained impossible because of the lack of exploitable ¹H-¹H couplings between the two fragments. Now the problem is solved by an enantioselective total synthesis. Some interesting relative configuration dependent reactivity differences were also observed en route to the total synthesis...



Shijun Zhu, Yikang Wu*

Synthesis and Configuration of