



## Synthesis and antifungal activity of furo[2,3-*f*]quinolin-5-ols

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### ABSTRACT

Furo[2,3-*f*]quinolin-5-ol derivatives were synthesized and tested for in vitro antifungal activity against *Candida*, *Aspergillus* species, and *Cryptococcus neoformans*. Among them tested, many furo[2,3-*f*]quinolin-5-ols showed good antifungal activity. The results suggest that furo[2,3-*f*]quinolin-5-ols would be promising antifungal agents.

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Benzofuran scaffolds **1** have potent biological properties including antifungal activity<sup>1,2</sup> as well as antibacterial<sup>3,4</sup> (Fig. 1). A benzofuran derivative, a novel myristoyltransferase inhibitor, has been reported as antifungal agent.<sup>1,2</sup> *N*-Myristoyltransferase has been proven to be essential for the viability of fungi, including medically important pathogenic fungi, *Candida albicans*<sup>5</sup> and *Cryptococcus neoformans*<sup>6</sup> making it a possible target for the development of antifungal agents with a novel mode of action.

In our previous Letters,<sup>7</sup> benzofuran-5-ol scaffolds **2** have demonstrated potent antifungal activity against pathogenic fungi (Fig. 1). We speculated that incorporation of a heterocyclic ring to the benzofuran-5-ol skeleton would change the physicochemical properties and lead to a new pharmacophore furo[2,3-*f*]quinolin-5-ols **3** with a different biological profile from benzofuran-5-ol scaffolds **2**. Furo[2,3-*f*]quinolin-5-ols **3** could metabolize to 5,8-quinolinedione derivatives with a quinonoid structure in fungi (Fig. 1). Quinonoid compounds display potent biological properties including antifungal, antimalarial, and antibacterial activity.<sup>8</sup> We assumed that furo[2,3-*f*]quinolin-5-ols **3** could have similar biological activities with those of quinonoid compounds. There have been a few reports<sup>9,10</sup> on furo[2,3-*f*]quinoline derivatives that exhibit cytotoxic activity against cancer cell lines. The antifungal activity of compounds **3** has not been reported to the best of our knowledge. The presence of aryl, thio, amino group, or halogen atoms on quinonoid compounds significantly affects their antifungal activity.<sup>11</sup> A variety of furo[2,3-*f*]quinolin-5-ols with different substituents could exhibit the biological activities through different actions and sometimes improve upon the activities.

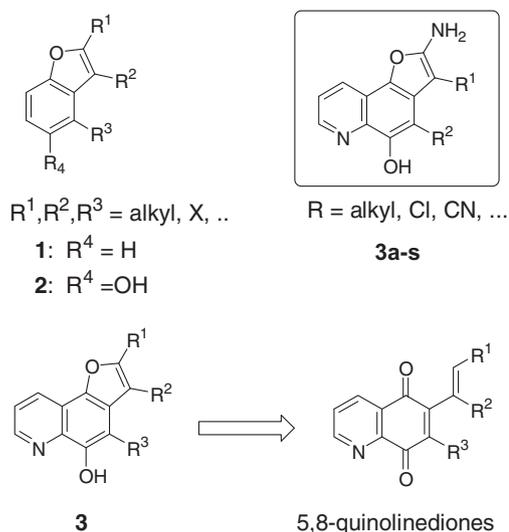
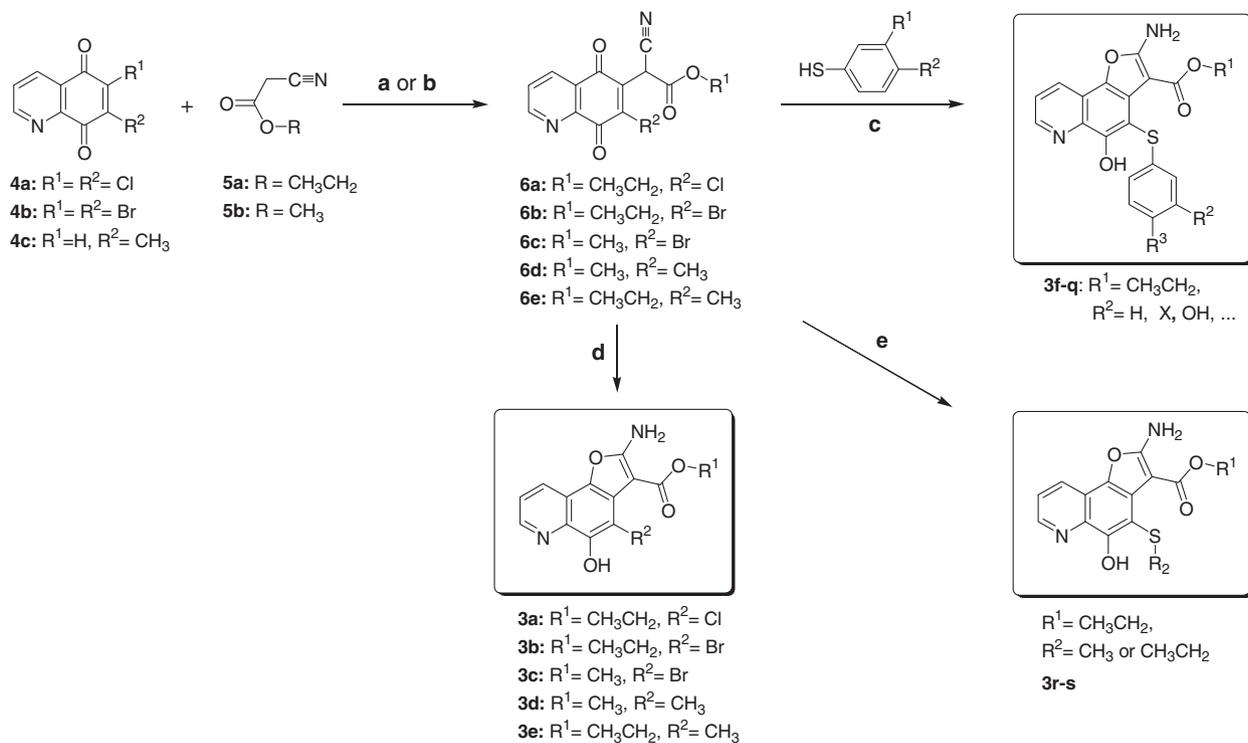


Figure 1. Benzofuran scaffolds and furo[2,3-*f*]quinolin-5-ol derivatives.

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Based on this speculation, furo[2,3-*f*]quinolin-5-ols **3a-s** with various substituents were designed and synthesized to elucidate their contribution to the antifungal activity (Scheme 1). The in vitro antifungal activity of compounds **3a-s** against pathogenic fungi was determined by the twofold broth dilution method. Additional data for antifungal activity of 5-aminoquinolin-8-ol (**7**) and 7-methyl-8-nitroquinoline (**8**) are provided.



**Scheme 1.** Synthesis of furo[2,3-*f*]quinolin-5-ol derivatives. Reagents and conditions: (a) **5a** for **6a**, **6b**, or **6e**/NH<sub>4</sub>OH/CeCl<sub>3</sub> (0.1 equiv)/EtOH/rt/10 min/47–61%; (b) **5b** for **6c** or **6d**/NH<sub>4</sub>OH/CeCl<sub>3</sub> (0.1 equiv)/EtOH/rt/10 min/40–55%; (c) **6a**/ arylthiol **9** (1 equiv)/EtOH/reflux/5 h/60–86%; (d) **6a–e**/ hydrazine hydrate (1 equiv)/EtOH/reflux/2 h/42–72%; (e) **6a**/alkylthiol (1 equiv)/EtOH/reflux/5 h/63–81%.

Methods for the synthesis of furo[2,3-*f*]quinolin-5-ols **3a–s** are shown in Scheme 1 and Table 1. The 6,7-dichloro-5,8-quinolinedione (**4a**),<sup>12</sup> 6,7-dibromo-5,8-quinolinedione (**4b**),<sup>13</sup> and 7-methyl-5,8-quinolinedione (**4c**) were prepared from commercially available 5-aminoquinolin-8-ol (**7**) and 7-methyl-8-nitroquinoline (**8**) according to the known method.

Alkyl 2-(7-halo-5,8-dioxo-5,8-dihydroquinolin-6-yl)-2-cyanoacetates **6a–e** were synthesized by regioselective nucleophilic substitution or addition of compound **4a**, **4b**, or **4c** with 1 equiv of methyl- or ethylcyanoacetate (**5a** or **5b**) and 0.1 equiv of CeCl<sub>3</sub> in the presence of NH<sub>4</sub>OH according to the reported method<sup>14,15</sup> with minor modification. The mechanism of formation of compounds **6** was cited in Ref. 16.

When the equivalent amount of each compound **6a–e** and hydrazine hydrate were mixed in EtOH and refluxed for 2 h, furo[2,3-*f*]quinolin-5-ols **3a–e** were formed. 2-Amino-4-arylthio-5-hydroxyfuro[2,3-*f*]quinolines **3f–q** were synthesized by nucleophilic substitution and cyclization of compound **6a** with appropriate arylthiols **9** in EtOH. To a solution of compound **6a** in EtOH, 1 equiv of arylthiol was added. The mixture was refluxed for 5 h and concentrated in vacuo. Purification of residual crude product by column chromatography yielded compounds **3f–q**. Most of these reactions went as expected and had overall high yields of 60–86%.

In a similar manner, 4-alkylthio-5-hydroxyfuro[2,3-*f*]quinolines **3r–s** were synthesized by cyclization of compound **6a** with 1 equiv of methyl- or ethylthiol in EtOH.

The mechanism for the formation of compounds **6** involves Michael-type addition of the anion of compounds **5** to quinones **4** followed by subsequent dechlorination.<sup>7</sup> The substitution of compounds **6** with nucleophilic thiols resulted in the formation of aromatic hydroquinone system as intermediates and subsequent cyclization to compounds **3**. The substitution was similar to the formation of stable aromatic hydroquinone system by the

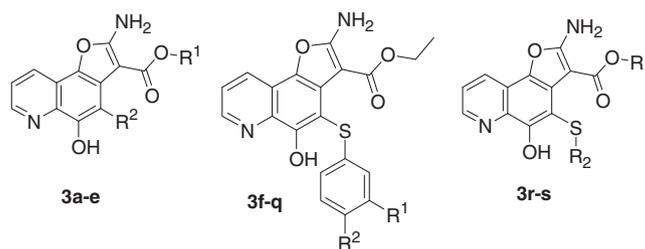
substitution of thiols on quinones.<sup>17</sup> The mechanism of formation of compounds **3** by the cyclization was cited in Ref. 16.

The synthesized furo[2,3-*f*]quinolin-5-ols **3a–s**, compound **7**, and **8** were tested in vitro for their growth inhibitory activity against pathogenic fungi using the standard method.<sup>18</sup> The MIC (minimum inhibitory concentration) values were determined by comparison with fluconazole and 5-fluorocytosine as standard agents.

As indicated in Table 1, many of furo[2,3-*f*]quinolin-5-ols **3a–s** showed potent antifungal activity against all tested fungi. Actually, the activity of compounds **3h**, **3i**, and **3k** was superior or comparable to those of 5-fluorocytosine against fungi. The compounds **3h**, **3i**, and **3k** completely inhibited the growth of all against *Candida* and *Aspergillus* species tested at the MIC level of 0.8–3.2 µg/mL. Many of furo[2,3-*f*]quinolin-5-ols **3a–s** also showed potent antifungal activity against *Candida krusei*, *C. neoformans*, and *Aspergillus* species. Actually, the activity of compounds **3j** or **3q** were superior or comparable to those of 5-fluorocytosine against *C. krusei* or *C. neoformans*.

Generally, the 4-arylthio-furo[2,3-*f*]quinolin-5-ols scaffolds **3f–q** exhibited the greatest activity, indicating a correlation that may offer insight into the mode of action of these compounds. In contrast, furo[2,3-*f*]quinolin-5-ols **3a–e** and 4-alkylthio-moieties of compounds **3r–s** did not show significant antifungal activity against tested fungi, although they exhibited good activity against *C. neoformans* and *Aspergillus flavus*. 4-Alkylthio-moieties of compounds **3r–s** did not improve their antifungal activity in comparison to 4-arylthio-compounds **3f–q**. The structure–activity relationship may not exist between properties of substituents (R<sup>1</sup>, R<sup>2</sup>: H, Me, Et, X, ...) for the 4-arylthio moieties of compounds **3f–q** and alkyl substituents for the compounds **3a–e** and **3r–s**. In addition, the 5-aminoquinolin-8-ol (**7**) and 7-methyl-8-nitroquinoline (**8**) exhibited no or poor, if any, antifungal activity. Thus, furo[2,3-*f*]quinolin-5-ols moiety is important for the antifungal activity.

**Table 1**  
Structures and antifungal activity for furo[2,3-*f*]quinolin-5-ols



Compound	R <sup>1</sup>	R <sup>2</sup>	MIC <sup>a</sup> (μg/mL)					
			<i>C. albicans</i> <sup>b</sup>	<i>C. tropicalis</i>	<i>C. krusei</i>	<i>C. neoformans</i>	<i>A. niger</i>	<i>A. flavus</i>
<b>3a</b>	CH <sub>3</sub> CH <sub>2</sub>	Cl	>50.0	50.0	6.3	25.0	>50.0	>50.0
<b>3b</b>	CH <sub>3</sub> CH <sub>2</sub>	Br	12.5	6.3	6.3	25.0	6.3	6.3
<b>3c</b>	CH <sub>3</sub>	Br	50.0	6.3	25.0	12.5	25.0	25.0
<b>3d</b>	CH <sub>3</sub>	CH <sub>3</sub>	0.8	6.3	25.0	25.0	12.5	3.2
<b>3e</b>	CH <sub>3</sub> CH <sub>2</sub>	CH <sub>3</sub>	6.3	25.0	25.0	12.5	>50.0	12.5
<b>3f</b>	F	H	50.0	50.0	50.0	50.0	25.0	12.5
<b>3g</b>	H	F	50.0	50.0	50.0	50.0	50.0	6.3
<b>3h</b>	H	Br	1.6	0.8	1.6	25.0	1.6	0.8
<b>3i</b>	H	Cl	0.8	0.8	1.6	25.0	3.2	3.2
<b>3j</b>	H	H	3.2	1.6	3.2	25.0	3.2	3.2
<b>3k</b>	F	F	0.8	1.6	1.6	50.0	1.6	1.6
<b>3l</b>	CH <sub>3</sub>	H	12.5	25.0	25.0	3.2	25.0	12.5
<b>3m</b>	CH <sub>3</sub>	CH <sub>3</sub>	25.0	12.5	25.0	12.5	12.5	6.3
<b>3n</b>	H	OH	12.5	6.3	25.0	12.5	3.2	25.0
<b>3o</b>	H	CH <sub>3</sub> O	6.3	6.3	6.3	25.0	3.2	12.5
<b>3p</b>	H	CH <sub>3</sub>	6.3	12.5	50.0	12.5	3.2	50.0
<b>3q</b>	Cl	H	6.3	12.5	6.3	1.6	3.2	12.5
<b>3r</b>	CH <sub>3</sub> CH <sub>2</sub>	CH <sub>3</sub>	12.5	12.5	50.0	6.3	50.0	50.0
<b>3s</b>	CH <sub>3</sub> CH <sub>2</sub>	CH <sub>3</sub> CH <sub>2</sub>	50.0	25.0	>50.0	>50.0	25.0	6.3
<b>7</b>	—	—	50.0	50.0	>50.0	>50.0	50.0	>50.0
<b>8</b>	—	—	>50.0	>50.0	>50.0	50.0	>50.0	>50.0
Fluconazole	—	—	50.0	6.3	12.5	6.3	12.5	25.0
5-Fluorocytosine	—	—	3.2	3.2	3.2	6.3	3.2	1.6

<sup>a</sup> The MIC value was defined as the lowest concentration of the antifungal agent. MIC values were read after one day for *Candida* species and *Cryptococcus neoformans*, and two days for *Aspergillus* species in 37 °C. The inoculum sizes contained approximately  $1 \times 10^5$  cells/mL. Culture media tested were the modified Sabouraud dextrose broth (Difco Lab.). The final concentration of antifungal agents was between 0.2 and 50.0 μg/mL.

<sup>b</sup> Fungi tested: *Candida albicans* Berkout KCCM 50235, *C. tropicalis* Berkout KCCM 50662, *C. krusei* Berkout KCCM 11655, *Cryptococcus neoformans* KCCM 50564, *Aspergillus niger* KCTC 1231, and *Aspergillus flavus* KCCM 11899.

In conclusion, 4-arylthio-furo[2,3-*f*]quinolin-5-ol scaffolds **3f–q** were synthesized by cyclization of compounds **4** with 1 equiv of appropriate arylthiols **9**. Furo[2,3-*f*]quinolin-5-ols **3a–e** were synthesized by cyclization of compounds **4** with hydrazine. 4-Alkylthio-furo[2,3-*f*]quinolin-5-ols **3r–s** were synthesized by cyclization of compound **6a** with 1 equiv of alkylthiols in EtOH. Most of these reactions went as expected and had overall high yields. We have identified a lead compound that has antifungal activity by screening of our furo[2,3-*f*]quinolin-5-ols **3a–s**. Among them tested, many of furo[2,3-*f*]quinolin-5-ols showed potent antifungal activity. The results suggest that furo[2,3-*f*]quinolin-5-ol scaffolds would be promising leads for the development of antifungal agents. Moreover, the results should encourage the synthesis of furo[2,3-*f*]quinolin-5-ol analogs for improving antifungal properties.

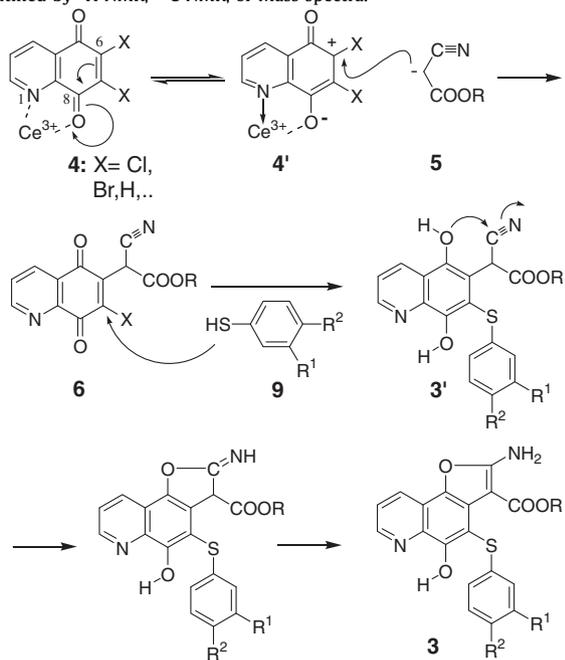
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## References and notes

- Masubuchi, M.; Kawasaki, K.; Ebiike, H.; Ikeda, Y.; Tsujii, S.; Sogabe, S.; Fujii, T.; Sakata, K.; Shiratori, Y.; Aoki, Y.; Ohtsuka, T.; Shimma, N. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 1833.
- Masubuchi, M.; Ebiike, H.; Kawasaki, K.; Sogabe, S.; Morikami, K.; Shiratori, Y.; Tsujii, S.; Fujii, T.; Sakata, K.; Hayase, M.; Shindoh, H.; Aoki, Y.; Ohtsuka, T.; Shimma, N. *Bioorg. Med. Chem.* **2003**, *11*, 4463.
- Abdel-Aziz, H. A.; Mekawey, A. A. I.; Dawood, K. M. *Eur. J. Med. Chem.* **2009**, *44*, 3637.
- Chen, Y.; Chen, S.; Lu, X.; Cheng, H.; Ou, Y.; Cheng, H.; Zhou, G.-C. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 1851.
- Weinberg, R. A.; McWherter, C. A.; Freeman, S. K.; Wood, D. C.; Gordon, J. I.; Lee, S. C. *Mol. Microbiol.* **1995**, *16*, 241.
- (a) Lodge, J. K.; Jackson-Machelski, E.; Toffaletti, D. L.; Perfect, J. R.; Gordon, J. I. *Proc. Natl. Acad. Sci. U.S.A.* **1994**, *91*, 12008; (b) Lodge, J. K.; Jackson-Machelski, E.; Higgins, M.; McWherter, C. A.; Sikorski, J. A.; Devadas, B.; Gordon, J. I. *J. Biol. Chem.* **1998**, *273*, 12482.
- Ryu, C.-K.; Song, A. L.; Lee, J. Y.; Hong, J. A.; Yoon, J. H.; Kim, A. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 6777.
- Middleton, R. W.; Parrick, J. In *The Chemistry of The Quinonoid Compounds*; Pataki, S., Rappoport, Z., Eds.; John Wiley & Sons: London, 1988; pp 1019–1066.
- Nebois, P.; Cherkaoui, O.; Benameur, L.; Boitard, M.; Bartoli, M.-H.; Fillion, H. *Pharmazie* **1999**, *54*, 215.
- Benameur, L.; Bouaziz, Z.; Nebois, P.; Bartoli, M.-H.; Boitard, M.; Fillion, H. *Chem. Pharm. Bull.* **1996**, *44*, 605.
- Ryu, C.-K.; Choi, K. U.; Shim, J.-Y.; You, H.-J.; Choi, I. H.; Chae, M. J. *Bioorg. Med. Chem.* **2003**, *11*, 4003.
- (a) Pratt, Y. T.; Drake, N. L. *J. Am. Chem. Soc.* **1960**, *82*, 1155; (b) Yasuda, M.; Boger, D. L. *J. Med. Chem.* **1987**, *30*, 1918.
- Schellhammer, C. W.; Petersen, S. *Ann. der Chem.* **1959**, *624*, 108.
- Ryu, C.-K.; Lee, J. Y.; Jeong, S. H.; Nho, J.-H. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 146.
- (a) Pratt, Y. T. *J. Org. Chem.* **1962**, *27*, 3905; (b) Yoshida, K.; Ishiguro, M.; Honda, H.; Yamamoto, M.; Kubo, Y. *Bull. Chem. Soc. Jpn.* **1988**, *61*, 4335.
- Compounds **6** were formed by regioselective nucleophilic substitution or addition of compounds **4** with 1 equiv of alkylcyanoacetates **5** and 0.1 equiv of CeCl<sub>3</sub> in the presence of NH<sub>4</sub>OH. As results of catalytic action of Ce<sup>3+</sup> ions, the substitution or addition in compounds **4** gave mainly 6-substituted products **6**.

The regioselectivity should be originated from the selective increment of the electrophilicity of 6-position in compounds **4** by the formation of Ce(III) chelate between carbonyl oxygen at 8-position and nitrogen at 1-position. The catalysis by Ce<sup>3+</sup> ions is understood from the intermediate **4'**. The 6-substituted products **6** were formed by a Michael-type addition of the anion of alkylcyanoacetates **5** to **4'**. These substitutions or additions were similar to the regioselective reaction of arylamines on 5,8-quinolinedione in the presence of Ce(III).<sup>15</sup> The substitution of **6** with nucleophilic thiols **9** resulted in the formation of aromatic hydroquinone system<sup>7,17</sup> as intermediates **3'** and subsequent cyclization to compounds **3**. The mechanism of cyclization was similar to the formation of furo[2,3-*f*]quinolin-5-ols<sup>9,10</sup> by cyclization of *N,N'*-dimethylhydrazone on 5,8-quinolinedione and naphthofuranol<sup>19</sup> by cyclization of acetylacetone on 1,4-naphthoquinone. Most of these reactions went as expected. The products **3** were separated by silica gel column chromatography. Purity of products **3** was determined both by TLC and GC. The results showed that a single compound was contained in each product. TLC was performed on precoated silica gel (60G 254, Merck) using CHCl<sub>3</sub> for solvent. The compounds were detected under UV light (254 nm). The purity of products was also verified by GC (Hewlett Packard 5890A, HP-5 capillary column at 260 °C, N<sub>2</sub>, 17 mL/min as carrier gas, FID). The compounds **3** were identified by <sup>1</sup>H NMR, <sup>13</sup>C NMR, or Mass spectra.



17. Kutyrev, A. A. *Tetrahedron* **1991**, 47, 8043.

18. McGinnis, M. R.; Rindali, M. G. In *Antibiotics in Laboratory Medicine*; Lorian, V., Ed., 4th ed.; Williams and Wilkins: Baltimore, 1996; pp 176–211.

19. Bernatek, E. *Acta Chem. Scand.* **1956**, 10, 273.