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Synthesis of a Conformationally Constrained Analogue of BW A78U, an Anticonvulsant Adenine Derivative

Laurent Désaubry, Camille Georges Wermuth and Jean-Jacques Bourguignon*

Laboratoire de Pharmacochimie Moléculaire, UPR 421 du CNRS, Centre de Neurochimie, 5, rue Blaise Pascal, 67084 Strasbourg Cedex, France

Abstract : The synthesis of a conformationally constrained analogue of the anticonvulsant BW A78U, a 9-benzyl-adenine derivative, has been devised, using silicon tetrachloride in a new cyclodehydration.

As a part of our program dealing with structural modification of BW A78U, a potent anticonvulsant and anxiolytic drug¹⁻³, we were interested in synthesizing a constrained analogue of this drug. Previous work has shown that conformationally restricted analogues are valuable probes in drug design^{4,5}. We report here the synthesis of the bridged compound 1, that is considered a semi-rigid analogue of BW A78U.



Three different strategies can be formally applied for the preparation of cycloadenine derivatives (scheme 1). The most convenient approach (type I) involves an adenine ring bearing a functionalized residue at the N(9) nitrogen. This method was applied in radical⁷⁻¹¹ or ionic¹² cyclization of adenosine derivatives leading to C(5') -C(8) bridged cycloadenosines. The second method (type II) starts from a functionalized pyrimidine, which is cyclocondensed with a cyclic iminoether, whereas the type III involves a C(2)-N(1) bridged imidazole carboxamide ¹³ or carbonitrile¹⁴. The only attempt to synthesize a C(8)-N(9) bridged adenine derivative according to a type II, which has been described, failed¹³. Preparation of compound 1 was carried out following a modified type II strategy.

The catalytic hydrogenation of the oxime prepared from 4-benzoylbutyric acid 2 afforded the corresponding δ -aminoacid 3 (scheme 2). This compound did not react with 4,6-dichloro-5-aminopyrimidine 4 to yield 5, but easily cyclized into lactam 6. However, the treatment of 3 with the more electrophilic 4-methylamino-5-nitro-6-chloropyrimidine 7¹⁵ afforded the amidine 8, which was catalytically hydrogenated to

give the triaminoacid 9. Different attempts to cyclize 9 with classical dehydration agents failed (phosphorus oxychloride, triphenyl or tributylphosphine in CCl4, anhydrous phosphoric acid, hydrochloric acid in ethanol). In each case either no reaction was observed, or degradation occurred.



Scheme 1: Strategies for the synthesis of cycloadenine derivatives.

The general problem to overcome is efficient activation of the tertiary amide 10 (scheme2) without blocking the nucleophilic character of the vicinal 5-amino function. As far as we know, among the tertiary amides, only formamides I ($R \neq H$, R' = H) have been cyclized to provide fused imidazolic heterocycles II, because of their satisfactory electrophilic character¹⁶. DMF activation has been performed using silicium derivatives such as tertiobutyl dimethyl silyl chloride¹⁷, or TMSCl¹⁸. The reaction has been extended by us to other tertiary amides ($R' \neq H$) by means of SiCl4 as dehydrating agent.

Thus, treatment of orthophenylenediamine 11 with DMF and 2.5 equivalents of SiCl4 in refluxing CH₂Cl₂ led to benzimidazole 12 (scheme 3). Moreover, the N-aryl lactame 13 considered as a model compound of our intermediate 10, afforded the awaited bridged benzimidazole 14.

When considering the amino lactame 10, the formation of N-Si bonds with the amino groups is reversible and the reaction is driven by the higher oxygen affinity for silicium leading to the cyclized compound 1 and silica. SiCl4 is known as an efficient dehydrating $agent^{19-21}$. However, only one example of activation of a tertiary amide using SiCl4 is described in the literature. It concerns the dehydration of N-alkyl N-acyl alanines leading to mesoionic oxazolones ²². The same experimental procedure let us to achieve in one step the cyclization of 9 into 1 (mp 170-171°C) with a satisfactory overall yield (55 %).

The application of activation of tertiary amides by SiCl4 for the synthesis of the cycloadenine derivative 1 and benzimidazoles 12 et 14 demonstrates the potential of this methodology in the synthesis of imidazoheterocycles. The anticonvulsant properties of compound 1 are under investigation, and will be reported elsewhere.



a) NH₂OH.HCl, 65[°]C, 12 h; b) H₂, Pd/C, EtOH; c) 4, Et₃N, nBuOH, reflux; d) 7, KHCO₃, MeOH, 50[°]C, 2 days; e) H₂, Raney Ni, H₂O ; f) SiCl₄, Et₃N, CH₂Cl₂, reflux, 4 days

Scheme 2





a : SiCl₄ , Et₃N, CH₂Cl₂, reflux

Scheme 3

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