

We began by examining the reactivity of cyclopropene **1a** under Cu-catalyzed borylation conditions, using achiral phosphine ligands (Table 1). In the presence of CuCl (10 mol

Table 1. Effect of the Ligand and Temperature on the Diastereoselective Hydroboration of Cyclopropenes

entry ^a	L	T (°C)	dr ^b	yield (%) ^c
1	xantphos	23	95:5	40
2	xantphos	-20	≥98:2	60
3 ^d	xantphos	-20	≥98:2	90

^aReaction conditions: **1a** (0.2 mmol), B₂pin₂ (0.22 mmol), NaOt-Bu (0.1 mmol), CuCl (10 mol %), xantphos (11 mol %), MeOH (0.8 mmol), THF (0.33 M). ^bDetermined by ¹H NMR analysis. ^cYield of isolated (±)-**2a**. ^dCyclopropene **1a** and MeOH were added at -78 °C; the reaction mixture was then warmed up to -20 °C.

%), xantphos (11 mol %), B₂pin₂ (1.1 equiv), NaOt-Bu (0.5 equiv), and MeOH¹⁵ (4 equiv) in THF, we observed the formation of cyclopropylboronate (±)-**2a** with excellent diastereoselectivity (entry 1, Table 1). However, the yields were consistently low due to formation of variable quantities of an inseparable mixture of dimers **B** and **C**. This problem was not completely unexpected since dimerization is one of the most common undesired pathways in transition-metal-catalyzed reactions with cyclopropenes.^{13b} Trying to minimize the formation of these dimeric structures therefore became one of the major challenges of this project.¹⁶ The yield of (±)-**2a** increased when we carried out the reaction at lower temperature (entry 2, Table 1), but significant amounts of **B** and **C** were still produced. After extensive experimentation,¹⁷ we observed that addition of cyclopropene **1a** and MeOH to a -78 °C solution of the preformed xantphos-copper-boryl complex, followed by warming to -20 °C, afforded (±)-**2a**⁷ in excellent yield as a single diastereomer.

Once we minimized the dimerization pathway for the diastereoselective Cu-catalyzed hydroboration, we looked at the possibility of developing an asymmetric variant (Table 2). We started testing several commercially available phosphines with different steric and electronic properties using the conditions previously optimized for xantphos (entries 1–6, Table 2).¹⁷ We soon realized that the yields and stereoselectivities were highly dependent on the ligand. (*R*)-DTBM-Segphos **L6** was superior to other chiral ligands affording cyclopropane (*R,R*)-**2a** in 71% yield and 92:8 enantiomeric ratio. However, we found that these values were poorly reproducible with enantiomeric ratios varying inconsistently from 85:15 to 92:8. Trying to solve this problem, we searched for a different Cu source.¹⁷ A first attempt using [Cu(CH₃CN)₄]PF₆ (entry 7, Table 2) afforded (*R,R*)-**2a** with high diastereoselectivity but only moderate yield and enantioselectivity. Gratifyingly, when acetonitrile was removed *in vacuo* after phosphine-copper complex formation (entry 8, Table 2), the desired compound was consistently obtained in high yield with excellent diastereo- and enantioselectivity (dr = 97:3, er = 95:5). This result significantly improves upon the 58% ee found in the Rh-catalyzed hydroboration of **1a**.⁷ The use of 5 mol % of [Cu(CH₃CN)₄]PF₆ resulted in a lower yield and enantioselectivity (entry 9, Table 2).

Table 2. Effect of the Ligand and Copper Source on the Enantioselective Hydroboration of Cyclopropenes

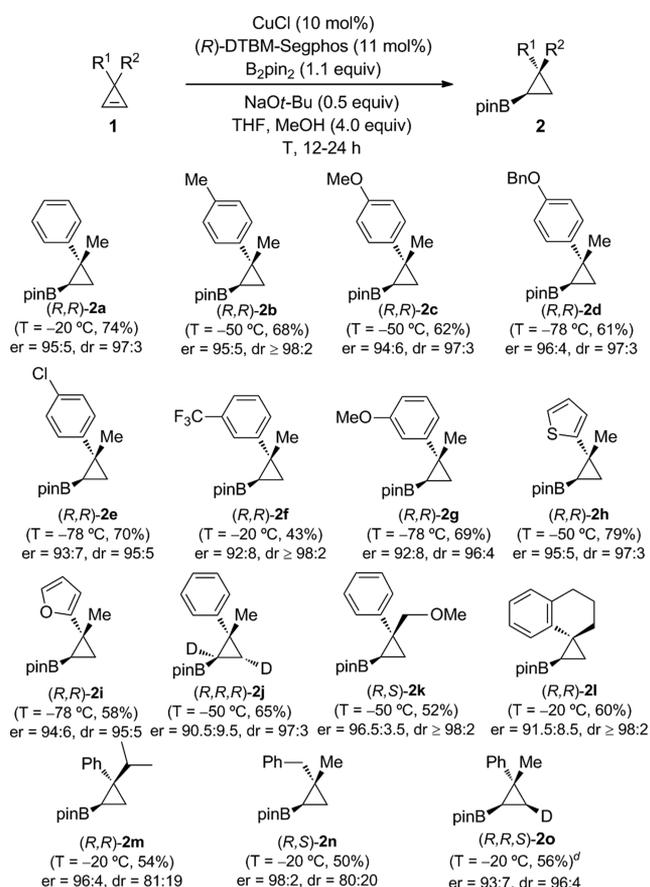
entry ^{a,b}	Cu(I)	L*	dr ^c	er ^d	yield ^e (%)
1	CuCl	L1	96:4	78:22	49
2	CuCl	L2	70:30	77:23	70
3	CuCl	L3	91:9	42:58	65
4	CuCl	L4	95:5	60:40	15
5	CuCl	L5	94:6	82:18	70
6	CuCl	L6	≥98:2	92:8	71
7	[Cu(CH ₃ CN) ₄]PF ₆	L6	96:4	82:18	58
8 ^f	[Cu(CH ₃ CN) ₄]PF ₆	L6	97:3	95:5	74
9 ^g	[Cu(CH ₃ CN) ₄]PF ₆	L6	95:5	90:10	50
10	–	L6	–	–	0

^aReaction conditions: **1a** (0.2 mmol), B₂pin₂ (0.22 mmol), NaOt-Bu (0.1 mmol), Cu(I) (10 mol %), L (11 mol %), MeOH (0.8 mmol), THF (0.33 M). ^bCyclopropene **1a** and MeOH were added at -78 °C; the reaction mixture was then warmed up to -20 °C. ^cDetermined by ¹H NMR analysis. ^der determined by chiral SFC. ^eYield of isolated (*R,R*)-**2a**. ^fCH₃CN was removed *in vacuo* after phosphine-copper complex formation. ^g5% of [Cu(CH₃CN)₄]PF₆ was used.

To rule out a possible organocatalytic activation of B₂pin₂,¹⁸ we carried out the reaction in the absence of a Cu salt (entry 10, Table 2). Under these conditions, formation of (*R,R*)-**2a** was not observed.

Next, we applied the Cu-catalyzed hydroboration conditions to different (3,3-disubstituted)-cyclopropenes (Table 3). In some cases, the reaction was carried out at either -50 or -78 °C to optimize the enantiomeric ratio. Compounds bearing an electron-rich aromatic substituent afforded the corresponding cyclopropylboronates [(*R,R*)-**2b–2d**] with similar efficiency to model (*R,R*)-**2a** (dr up to ≥98:2, er up to 96:4). Cyclopropenes with electron-deficient aryl groups underwent hydroboration with good yields and high diastereoselectivities although slightly lower enantiomeric ratios [(*R,R*)-**2e–2f**]. Moreover, substitution at the *meta* position of the aryl group also seemed to affect the enantioselectivity [(*R,R*)-**2g**]. Compounds (*R,R*)-**2h** and (*R,R*)-**2i**, with a thiophene and a furan ring, respectively, were also prepared with good yields and high enantioselectivities. Starting from dideuterated **1a** (**1a-d2**), compound (*R,R,R*)-**2j** with three contiguous stereocenters was successfully obtained. Additionally, coordinating groups (CH₂OMe) were also compatible with the hydroboration conditions and compound (*R,S*)-**2k** was obtained with excellent results (er = 96.5:3.5, dr ≥ 98:2).¹⁹

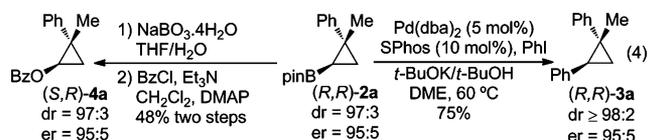
Cyclopropane (*R,R*)-**2l**, bearing a spiro-quaternary stereocenter, was also prepared with high levels of stereocontrol. Gratifyingly, bulkier groups on the cyclopropene (R² = *i*-Pr, (*R,R*)-**2m**) maintained high levels of enantiocontrol although the diastereoselectivity was moderately decreased (80:20 vs 97:3). Similarly, cyclopropane (*R,S*)-**2n** bearing two alkyl groups (R¹ =

Table 3. Substrate Scope^{a,b,c}

^aReaction conditions: **1** (0.2 mmol), B₂pin₂ (0.22 mmol), NaOt-Bu (0.1 mmol), CuCl (10 mol %), (R)-DTBM-Segphos (11 mol %), MeOH (0.8 mmol), THF (0.33 M). ^bYield of isolated **2**. ^cer determined by chiral SFC; dr determined by ¹H NMR analysis. ^dMeOD (0.8 mmol) instead of MeOH was used.

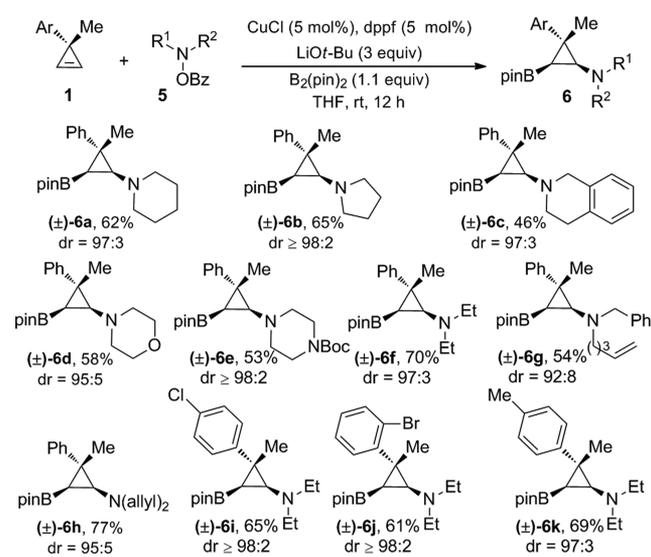
Bn, R² = Me) was successfully obtained with somewhat lower diastereoselectivity than (R,R)-**2a** but with excellent enantioselectivity (er = 98:2).²⁰ Finally, in the presence of MeOD, compound (R,R,S)-**2o** was obtained in good yield and high stereoselectivity (er = 93:7, dr = 96:4, >98% D incorporation). This experiment demonstrates the *syn* insertion of the cyclopropene in the copper–boryl complex and reveals the configurational stability of the cyclopropylcopper intermediate **A** (Scheme 1, eq 3). Overall, the results described in Table 3 suggest that the stereoselectivity is mostly controlled by steric factors.²¹

To demonstrate the versatility of the cyclopropylboronate species, (R,R)-**2a** was easily transformed into cyclopropane (R,R)-**3a** through a Suzuki–Miyaura coupling with iodobenzene (eq 4).²² Additionally, cyclopropanol derivative (S,R)-**4a** was prepared through an oxidation–benzoylation sequence.



Finally, we wanted to explore the trapping of **A** with electrophiles other than a proton. We focused on the use of *O*-benzoyl-*N,N*-dialkylhydroxylamines²³ due to the importance of cyclopropylamines in biologically active compounds. The

products would be cyclopropylaminoboronates (**6**) with three contiguous stereocenters, which would be difficult to obtain by known methods. We started our study using achiral phosphines. Unfortunately, the conditions found for the diastereoselective hydroboration of cyclopropene **1a**¹⁷ (entry 3, Table 1) were not optimal for the aminoboration reaction.²⁴ After significant optimization, we were pleased to find that a CuCl/dppf catalyst system (5 mol %) and LiOt-Bu in THF afforded cyclopropylaminoboronates (±)-**6** in good yields and excellent diastereomeric ratios (Table 4). Interestingly, the reactions were carried

Table 4. Copper-Catalyzed Aminoboration^{a,b,c}

^aReaction conditions: **1** (0.4 mmol), B₂pin₂ (0.44 mmol), R₂NOBz (0.6 mmol), LiOt-Bu (1.2 mmol), CuCl (5 mol %), dppf (5 mol %), THF (0.2 M). ^bDetermined by ¹H NMR analysis. ^cYield of isolated **6**.

out at rt without observing significant amounts of dimerization products **B** and **C**. These conditions worked well for a variety of *O*-benzoyl-*N,N*-dialkylhydroxylamines. Piperidine, pyrrolidine, tetrahydroisoquinoline, morpholine, and piperazine derivatives ((±)-**6a–6e**, Table 4) were easily prepared using the conditions described above. In all cases, a cyclopropane with the methyl, nitrogen, and boron substituents in a *syn* orientation was obtained with high diastereoselectivity.²⁵ Cyclopropylaminoboronates bearing an acyclic *N,N*-dialkyl amine moiety were also successfully obtained through this method ((±)-**6f–6h**, Table 4). Additionally, we performed the reaction with cyclopropenes bearing different substituents on the aromatic ring, obtaining the desired compounds in good yields as nearly single diastereomers ((±)-**6i–6k**, Table 4). Unfortunately, none of the chiral phosphines used in the optimization of (R,R)-**2a** were compatible with the aminoboration conditions. In all cases, we obtained inseparable mixtures of the desired compounds and unknown byproducts with diastereomeric ratios significantly lower than in the case with dppf.¹⁷

In summary, we describe here the first diastereo- and enantioselective Cu-catalyzed hydroboration of cyclopropenes. This method allows for the synthesis of enantiomerically enriched cyclopropylboronates with a quaternary stereocenter and represents the first enantioselective Cu-catalyzed desymmetrization of cyclopropenes. Our approach nicely complements the few existing methods to synthesize nonracemic cyclopropylboronates and gives new insights into the enantioselective metal-catalyzed desymmetrization of cyclopropenes. Additionally, the

capture of the cyclopropylcopper intermediate with electrophilic amines highlights the synthetic potential of this approach and opens a new way to synthesize functionalized cyclopropanes. Further applications toward the enantioselective synthesis of cyclopropylaminoboronates as well as functionalization of cyclopropenes with different electrophiles are underway.

■ ASSOCIATED CONTENT

■ Supporting Information

Experimental procedures, spectral data, and crystallographic CIF files. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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(16) We believe **B** is formed through a Cu-catalyzed formal [2 + 2] cycloaddition. **C** could be formed from **B** through an electrocyclic ring opening.

(17) For a full account on all the ligands used and other parameters see the Supporting Information.

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(19) Unfortunately, cyclopropenes bearing a carbomethoxy group ($R^2 = CO_2Me$ instead of Me) afforded a complex mixture of compounds.

(20) The absolute configuration of (*R,R*)-**2e** was established from single crystal X-ray crystallography of a *p*-nitrobenzoate derived by oxidation of the C–B bond followed by benzoylation. The absolute configuration of the other cyclopropylboronates was assigned by analogy. The relative stereochemistry of compound (*R,R*)-**2l** was assigned by single crystal X-ray crystallography.

(21) Proposed transition-state models to explain the observed stereoselectivity are included in the Supporting Information.

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(25) The relative configuration of (\pm)-**6a** was established by single crystal X-ray crystallography.