



## New application of the Julia olefination for the synthesis of Tyr-Gly *E*-alkene and carba isostere pseudopeptides

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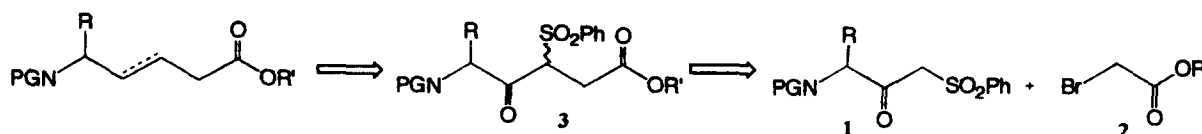
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### Abstract

A new application of the Julia olefination to the synthesis of TyrΨ[*E*-CH=CH]Gly and TyrΨ[CH<sub>2</sub>CH<sub>2</sub>]Gly pseudopeptides is described via condensation of tertibutyl bromoacetate on tyrosine-derived β-ketosulfone and subsequent reductive desulfonation. © 1999 Elsevier Science Ltd. All rights reserved.

The replacement of the amide bond linkage within peptides is a classical strategy for the study of biologically active peptides and for the preparation of peptide bonds resistant to proteolysis.<sup>1</sup> Among the different described amide bond surrogates, the non-hydrolysable *E*-ethylenic isosteres mimic the three-dimensional structure of the amide bond.<sup>2</sup> Several synthetic routes of *E*-olefin pseudopeptides are described in the literature.<sup>3</sup> Among them, the Julia olefination has been applied in this context.<sup>4</sup> We would like to disclose herein our results on a new application of the Julia olefination to the synthesis of TyrΨ[*E*-CH=CH]Gly and TyrΨ[CH<sub>2</sub>CH<sub>2</sub>]Gly dipeptides according to Scheme 1.

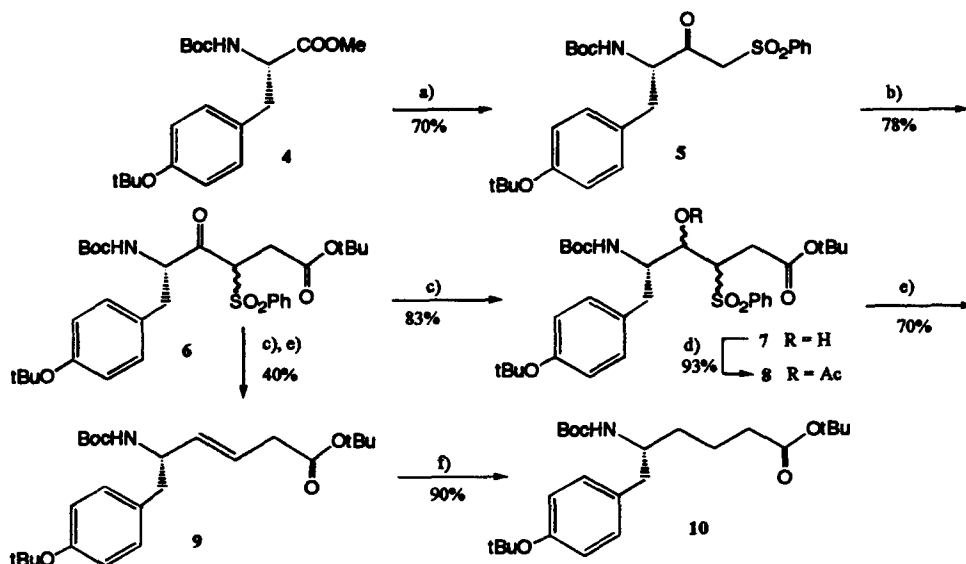


Scheme 1.

Lygo has been the first person to investigate this strategy but has failed to obtain the *E*-olefin isostere of Ala-Gly dipeptide.<sup>5</sup> *N*<sup>α</sup>-Boc-Tyr(*t*Bu)-OMe **4**<sup>6</sup> was first transformed into the β-ketosulfone **5**<sup>7</sup> by alkylation with two equivalents of the di-lithio anion of methyl phenyl sulfone at –78°C as described by Lygo (Scheme 2). It was necessary to use an excess of the anion to obtain complete conversion of the starting compound **4**. Condensation of tertibutyl bromoacetate onto β-ketosulfone **5** was achieved with potassium carbonate in DMF for 6 h in 78% yield. Next we turned our attention to the conversion of the β-ketosulfone **6** into the corresponding *E*-alkene isostere **9**. Lygo did not succeed in the transformation of

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the corresponding Ala-Gly ketosulfone ethyl ester into the *E*-alkene product using sodium borohydride reduction followed by reductive desulfonation with sodium amalgam at  $-10^{\circ}\text{C}$ , but instead obtained the fully reduced methyl ester. However, we found that a *t*Bu protecting group for the ester **2** is required for the efficient reductive desulfonation of **3**. In this way, the  $\beta$ -ketosulfone tertibutyl ester **6** was transformed in the same conditions into the *E*-alkene isostere **9** in 40% yield. However, we improved the yield by isolating the intermediate alcohols **7** and then reducing the acetylated diastereomers **8** to obtain the dipeptide isostere **9** in 54% overall yield from **6**.



Scheme 2. Reagents: (a)  $\text{PhSO}_2\text{CHLi}_2$ , THF,  $0^{\circ}\text{C}$  30 min then  $-30^{\circ}\text{C}$  3 h; (b)  $\text{BrCH}_2\text{COOtBu}$ ,  $\text{K}_2\text{CO}_3$ , DMF, rt, 6 h; (c)  $\text{NaBH}_4$ , MeOH,  $-10^{\circ}\text{C}$  3 h; (d)  $\text{Ac}_2\text{O}$ , DMAP, pyridine, rt, 3 h; (e)  $\text{Na/Hg}$ , MeOH,  $\text{Na}_2\text{HPO}_4$ ,  $-10^{\circ}\text{C}$ , 2 h; (f)  $\text{H}_2$ , Pd/C, EtOH  $95^{\circ}$ , rt, 20 h.

Finally, catalytic hydrogenation of *E*-alkene **9** gave the fully protected carbapeptide **10** in 90% yield.<sup>8</sup>

In conclusion, we have shown that a new application of Julia olefination methodology allows for the efficient synthesis of  $\text{Tyr}\Psi[\text{E-CH=CH}]\text{Gly}$  and  $\text{Tyr}\Psi[\text{CH}_2\text{CH}_2]\text{Gly}$  pseudopeptide in five (29.5% overall yield) and six steps, respectively, from Boc-protected tyrosine methyl ester. Furthermore, synthesis of  $\text{Xaa}\Psi[\text{CH=CH}]\text{Xbb}$  pseudopeptides would be feasible through condensation of  $\beta$ -ketosulfone **1** with  $\alpha$ -substituted bromoester **2**.<sup>9</sup>

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6. Compound **4** was prepared by standard procedures from either the corresponding commercial *N*<sup>α</sup>-Boc-protected acid or the free amine methyl ester.
7. All compounds gave satisfactory <sup>1</sup>H NMR and FTIR data. Compound **9**: [ $\alpha$ ]<sub>D</sub> +2.6 (c 0.57, CHCl<sub>3</sub>), glassy solid. IR (film,  $\nu$  cm<sup>-1</sup>): 3360, 2932, 1713, 1503, 1391, 1237, 1162. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>,  $\delta$  ppm): 1.32 (s, 9H, *t*Bu), 1.39 (s, 9H, *t*Bu), 1.43 (s, 9H, *t*Bu), 2.78 (d, *J*=6.2 Hz, 2H, CH<sub>2</sub>), 2.94 (d, *J*=6.6 Hz, 2H, CH<sub>2</sub>), 4.37 (m, 1H, CH  $\alpha$  or NH), 4.46 (m, 1H, NH or CH  $\alpha$ ), 5.50 (dd, *J*=15.8 Hz and 5.1 Hz, 1H, =CH), 5.62 (m, 1H, =CH), 6.90 (d, *J*=8.3 Hz, 2H, 2CH ar), 7.05 (d, *J*=8.3 Hz, 2H, 2CH ar).
8. Compound **10** was found enantiomerically pure by C18 RP-HPLC analysis after complete deprotection (TFA/CH<sub>2</sub>Cl<sub>2</sub>) and derivatization with GITC (see: Nimura, N.; Ogura, H.; Kinoshita, T. *J. Chromatography* **1980**, *202*, 375–379).
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