

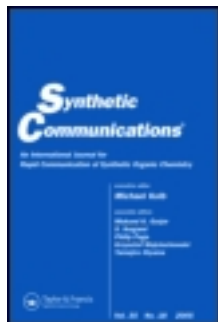
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**O-DEMETHYLATION OF OPIOID DERIVATIVES WITH METHANE
SULFONIC ACID / METHIONINE : APPLICATION TO THE
SYNTHESIS OF NALOXONE AND ANALOGUES.**

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cedex, France.**

Abstract : *Naloxone 2 was obtained by demethylation of N-allylnoroxycodone 1 with methane sulfonic acid / methionine. This reagent is an excellent substitute for boron tribromide. It was used for the synthesis of analogous derivatives with variable results.*

Many active opioid derivatives¹ have an hydroxyl group at the 3 position and their synthesis requires O-demethylation of the intermediate methyl ether. Cleavage of aryl methyl ethers to give phenols is well

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described²; many reagents have been used for the hemisynthesis of morphinans but the results depend on the structure.

Boron tribromide³ is the reagent most frequently employed as a result of mild conditions together with high yield and selectivity. Potassium hydroxide⁴, used for the Diels-Alder adducts of thebaine, and pyridine hydrochloride⁵ require more drastic conditions. Diphenyllithium phosphide⁶ and, recently, thiolate anions⁷ have also been used. In spite of certain advantages, boron tribromide is not convenient at industrial scale owing to toxicity and difficult handling.

We have found a reagent to replace it, particularly for the synthesis of naloxone 2. This is the methane sulfonic acid - methionine system, known as hard acid - soft nucleophile reacting by a push-pull mechanism⁸. This type of reagent, developed by Yajima⁹ and Kiso¹⁰, has been applied to deprotect peptides. Only a few examples of cleavage of aryl methyl ethers have been mentioned and the yields are poor¹¹.

We report herein a new preparation of naloxone 2 from N-allyl noroxycodone 1 with 30 equivalents of methane sulfonic acid and 1.5 equivalents of methionine over 15 hours at 40°C (scheme 1).

A large excess of methane sulfonic acid is needed whereas just a slight excess of methionine is

Scheme 1

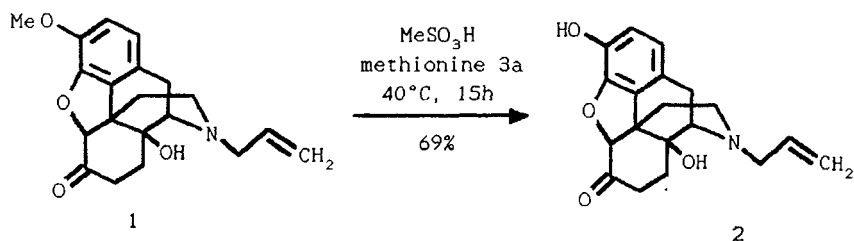


Table 1. Demethylation of N-allyl noroxycodone **1** with MeSO_3H / methionine^a

Reaction Conditions		Yield (%) ^b	
Time (h)	Temp. ($^\circ\text{C}$)	2	1
55	20	66	23
15	30	67	17
15	40	69	12
40	40	58	0
14	60	58	1
6	80	46	0

^a 30 equivalents MeSO_3H / 1.5 equivalents methionine.

^b Determined by HPLC (external standard).

Table 2. Demethylation of N-allyl noroxycodone **1** with strong acid / methionine^a

Reaction Conditions				Yield (%) ^b	
Acid(equiv)	Solvent	Time(h)	Temp.(°C)	2	1
MeSO ₃ H (30)	-	15	40	69	12
CF ₃ SO ₃ H (15)	-	6.5	20	59	3
CF ₃ SO ₃ H (5)	CF ₃ CO ₂ H	29	20	69	5
ClSO ₃ H (15)	-	1	20	0	0
H ₂ SO ₄ , SO ₃ ^c (15)	-	8	20	0	0
Nafion ^d (30)	CHCl ₃	30	25-35	0	65

a 1.5 equivalents methionine.

b Determined by HPLC (external standard).

c 20% of SO₃.

d Nafion 117, H⁺ form, contains 0.9 mmol RSO₃H/g.

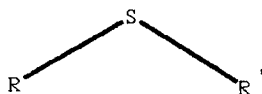
sufficient. Time and temperature have an influence on kinetics and also on degradation which can attain 40% above 60°C. Also, prolonged reaction leads to much degradation, even at 30-40°C (Table 1).

The optimal temperature seems to be 30-40°C and the time should not exceed 15 hours.

A cosolvent such as dichloromethane, trifluoroacetic acid, trifluoromethane sulfonic acid or

pyridine does not improve the results. Methane sulfonic acid can be replaced by trifluoromethane sulfonic acid. The other acids tested are less effective (Table 2).

Among various sulfides used, tetrahydrothiophene and dibutyl sulfide give similar results to those obtained with methionine (Table 3).



3

3	R	R'
a	Me	CH ₂ CH ₂ CH(NH ₂)CO ₂ H
b	Me	Ph
c	n-Bu	n-Bu
d	Me	t-Bu
e	-(CH ₂) ₄ -	
f	H	CH ₂ CH ₂ CH(NH ₂)CO ₂ H, HCl

Possible side-reactions are :

- Alkylation of tertiary amine 1 or 2 by sulfonium ion resulting from demethylation reaction and formation of quaternary ammonium compound,
- Cleavage of cyclic arylalkyl ether 1 or 2 and formation of sulfonium ion,

Table 3. Demethylation of N-allyl noroxycodone 1 with MeSO₃H / dialkyl sulfide^a

Reaction conditions			Yield (%) ^b	
Dialkyl sulfide (equiv)	Time(h)	Temp.(°C)	2	1
Methionine 3a (1.5)	15	40	69	12
3b (3)	32	20-60	12	3
3c (1.5)	27	20-60	59	10
3d (1.5)	16	20-60	0.5	5
3e (1.5)	20	40	62	16
3f (1.5)	6.5	20	0	0

a 30 equivalents MeSO₃H.

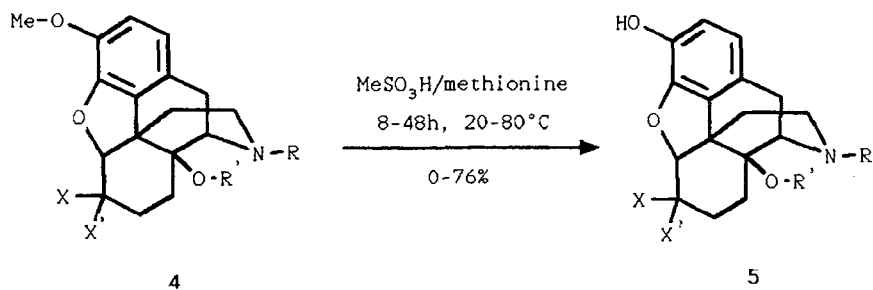
b Determined by HPLC (external standard).

- Alkylation of sulfide 3a with tertiary alcohol 1 or 2 and formation of sulfonium ion¹².

All these by-products are soluble in water and easily eliminated. Not all have been fully characterized.

Our demethylation conditions, methane sulfonic acid / methionine, have been applied to various intermediates in the synthesis of naloxone (Scheme 2 and Table 4). Oxymorphone 5a is obtained in 76% yield. In the case of oxycodone acetate 4b, hydrolysis of acetate leads to oxymorphone 5a.

Scheme 2



4, 5	R	R'	X, X'
a	Me	H	= O
b	Me	Ac	= O
c	H	H	= O
d	CO ₂ Et	H	= O
e	CH ₂ - 	H	= O
f	CH ₂ - 	H	H, OH
g	CH ₂ - 	H	= CH ₂

Table 4. Demethylation of **4a-g** with MeSO₃H/methionine^a

4 Reaction Conditions			5 Yield ^b		Litterature		
			(%)				
Time(h)	Temp.(°C)				Ref	Yield(%)	Method ^c
4a	12	40	5a	76	13	27	A
					5	69	B
					14	76	C
4b	19	20-40	5a	59	13	d	
4c	14	40-80	5c	0	15	e	C
4d	48	20	5d	18	16	73	C
4e	8	40	5e	65	17	73	B
4f	8	40	5f	47	18	69	C
4g	24	40	5g	mixt.	19	f	

a 30 equivalents MeSO₃H / 1,5 equivalents methionine.

b Determined by HPLC (external standard).

c A: HBr/reflux; B: Pyridine,HCl/200°C; C: BBr₃.

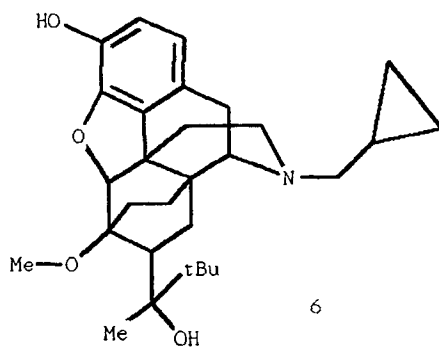
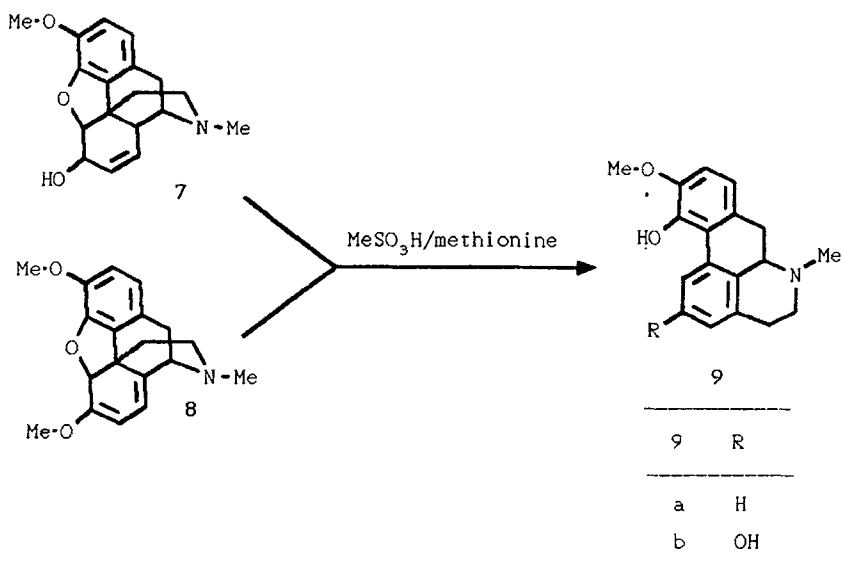
d Obtained by demethylation of 5a.

e Not indicated.

f Obtained by a Wittig reaction with naltrexone.

Demethylation of analogous derivatives to obtain naltrexone **5e**, nalbuphine **5f** and nalmetrene **5g** has been

Scheme 3



carried out with varying results (Table 4). Buprenorphine 6 was not obtained under these conditions, the tertibutyl group being acid-labile.

Codeine **7** and thebaine **8** undergo rearrangement to **9a** and **9b** respectively (Scheme 3). This reaction is known with methane sulfonic acid alone²⁰.

Until now, this new method of demethylation of ethers was essentially used to deprotect peptides. We have advantageously applied it to the preparation of opioid alkaloids²¹. Reagents are easily handled and can be used at industrial scale.

EXPERIMENTAL

98% MeSO₃H and DL-methionine were purchased from Janssen Chimica. Ammonium hydroxyde contains 21% of NH₃. Alumina CBT2 was purchased from Rhone Poulenc. Reagent grade solvents were used without further purification. Melting points were taken using a Tottoli apparatus (Buchi) and are uncorrected. Optical rotations at the Na-D line were obtained at ambient temperature using a Perkin-Elmer 241 polarimeter. ¹H-NMR spectra were obtained using a Bruker 300 MHz spectrometer. HPLC were performed on a C₁₈ μbondapack column; mobile phase: MeOH 50 / 0.1% aqueous (NH₄)₂CO₃·7H₂O 50; detector UV at 240 nm; flow rate: 2mL/min.

N-allyl-7,8-dihydro-14-hydroxynormorphinone or naloxone (2):

DL-methionine (**3a**; 22.4g, 0.15mol) is added over 15 min at 40°C to a stirred solution of N-allyl-noroxycodone (**1**; 34.12g, 0.1mol) in MeSO₃H (288g, 3mol) and stirring

is continued at 40°C for 15h. After cooling to room temperature, the mixture is slowly poured into NH₄OH (300mL) and ice (300g) and extracted with AcOEt (5 X 400mL). The organic layers are washed with water (400mL), dried (Na₂SO₄) and the solvent removed under reduced pressure. The residue (30.7g) is taken up in chloroform (600mL) and stirred with alumina CBT2 (86g) during 10 min. The suspension is filtered and the filtrate evaporated. The crude product is crystallized from toluene (600mL) to give naloxone 2; yield: 19.8g (60%); mp 179.5°C; $[\alpha]_D^{22}$ -215° (c=1, MeOH) (Lit.²² mp 177-178°C). ¹H-NMR (CDCl₃-DMSO): δ= 1.5-3.3 (m, 13H), 4.7 (s, 1H, CHO), 5.2 (m, 2H allyl), 5.5 (s, 2H, 2OH), 5.8 (m, 1H allyl), 6.6 (d, 1H, J=8, H arom), 6.7 (d, 1H, J=8, H arom).

Hydrochloride: a solution of purified product (2; 5g, 15.2mmol) in acetone (30mL) is concentrated to 10mL and HCl 6N (5mL, 30mmol) added at 50°C. The hydrochloride crystallizes at -10°C. Yield: 4.83g (87%); assay 99.9% (HPLC); $[\alpha]_D^{22}$ -179° (c=2.5, H₂O).

O-Demethylation of 4a-g with MeSO₃H / methionine:

General Procedure:

DL-methionine (3a; 0.75g, 5 mmol) is added at 40°C to a solution of 4a-g (3.3 mmol) in MeSO₃H (9.6g, 100 mmol) and stirring continued. Time and temperature are given in Table 4. The mixture is cooled to r.t. and poured into NH₄OH (10mL) and ice (10g). The crude product 5a-g

is obtained after extraction with AcOEt (5 X 20mL). Yields are determined by HPLC (Table 4).

$^1\text{H-NMR}$ (CDCl_3 -DMSO) δ /TMS, J(Hz):

5a: 1.5-3.25(m,11H), 2.4(s,3H,NCH₃), 4.7(s,1H,CHO), 4.8(s,2H,2OH), 6.6(d,1H,,J=8,H arom), 6.7(d,1H,J=8,H arom).

5d: 0.9-3.3(m,11H), 1.25(t,3H,J=6,CO₂Et), 4.1(q,2H,J=6,CO₂Et), 4.7(s,1H,CHO), 5-6(m,2H,2OH), 6.65(d,1H,J=8,H arom), 6.75(d,1H,J=8,H arom).

5e hydrochloride: 0.35-1.2(m,5H,cycloPr), 1.4-3.5(m, 12H), 4(m,1H,CHN), 5(s,1H,CHO), 6.6(d,1H,J=8,H arom), 6.7(d,1H,J=8,H arom), 6.95, 9.05 and 9.5(3s,3H,2OH, NH⁺).

5f hydrochloride: 0.9-3.5(m,2OH), 4.1(m,1H,CHOH), 4.5 (d,1H,J=4,CHO), 4.65(d,1H,J=5,CHOH), 6.25(s,1H,14-OH), 6.5(d,1H,J=8,H arom), 6.65(d,1H,J=8,H arom), 8.7(s,1H, NH⁺), 9.15(s,1H,OH arom).

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