Synthesis and Comparative Glycosidase Inhibitory Properties of Reducing Castanospermine Analogues

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Keywords: Carbamates / Castanospermine / Inhibitors / Thiocarbamates / Indolizidines / Enzymes

The feasibility of the intramolecular nucleophilic addition of the nitrogen atom in cyclic (thio)carbamates with a pseudo-*C*-nucleoside structure to the masked carbonyl group in aldose precursors in the synthesis of reducing (i.e., 5-hydroxy)-6-oxaindolizidine frameworks is illustrated by the preparation of the 6-*epi*, 7-*epi*, 8-*epi* and 6,8a-di-*epi* diastereomers of the potent glycosidase inhibitor (+)-castanospermine. In all cases, the increased anomeric effect caused by the high sp² character of the pseudoamide-type nitrogen atom resulted in the pseudoanomeric hydroxy group being anchored in an axial orientation in aqueous solution, as in the aglycons in α glycosides. These analogs of the natural alkaloid showed a

Introduction

Castanospermine (1) is one of a number of plant-derived polyhydroxyindolizidine alkaloids that exhibit glycosidase inhibition activity (Figure 1).^[1–7] Thus, **1** is a potent, competitive and specific inhibitor of several glucosidases, including lysosomal glycoprotein processing enzymes, which gives it potential therapeutic value in the treatment of ailments as varied as cancer, viral infections and diabetes. The pronounced biological activity of **1** stems from its ability to mimic the transition state involved in enzymatic glycoside hydrolysis. However, contrary to the natural glucosides, neither **1** nor most of the analogues reported so far possess a defined configuration at the pseudoanomeric centre.^[8–16] It is therefore not surprising that several α - and β -glucosidases can be simultaneously inhibited, which may be particularly problematic in clinical trials.

We have recently found that a subtle change in the structure of nitrogen-in-the-ring carbohydrate mimics (iminosugars, azasugars^[17]), consisting of the replacement of the sp³ imino nitrogen by a pseudoamide-type nitrogen atom with substantial sp² character, leads to a dramatic in-

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[b] Instituto de Investigaciones Químicas, CSIC, Américo Vespucio 49, Isla de la Cartuja, Sevilla, Spain Fax: +34-954460565 higher selectivity in the inhibition of α -glucosidases. Structure/glycosidase inhibitory activity studies indicated that inversion of any hydroxy group resulted in a dramatic decrease in the inhibition potency, confirming the critical importance of a correct hydroxylation profile. In the case of (+)-8-*epi*-6-oxacastanospermine derivatives, with a hydroxylation profile with a structural complementarity to that of D-galactose, a moderate but very selective inhibition of α -galactosidase was observed, supporting the importance of a defined configuration at pseudoanomeric centres for anomeric specificity. (© Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, Germany, 2005)



Figure 1. Structures of (+)-castanospermine and of the reducing thiocarbamic and carbamic analogues.

crease in the orbital contribution to the anomeric effect at amino acetal pseudoanomeric centres, resulting in a total preference for the axial orientation of the oxygen substituent and a notably enhanced stability,^[18-20] even in the case of hemiaminal derivatives.^[21-24] Thus, the reducing cyclic thiocarbamate and carbamate castanospermine analogues 2 and 3 (Figure 1) showed configurational integrity in water, with the C-5 hydroxy substituent anchored in the axial position, as in the natural aglycons in α -glucopyranosides. Noteworthy, these sp²-azasugar-type glucomimetics exhibited a dramatic increase in the α -anomeric selectivity (up to 10000 times that of the parent compound 1) when assayed against a panel of glycosidases, as well as a remarkable specificity within the α -glucosidase isoenzymes.^[25,26] This marked specificity has been ascribed to the interplay of topographical and electronic characteristics, the main factors being: (i) a configurational pattern analogous to that of α -D-glucopyranosides; (ii) a locked conformation about the bond homologous to C-5-C-6 in the parent

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sugar; and (iii) a partial positive charge at the region of the endocyclic heteroatom in the six-membered ring as a consequence of electron delocalization in the pseudoamide segment.^[27] In order to further investigate the critical features of the specificity and potency of the inhibition, we have now prepared a series of reducing 3-oxo(thioxo)-2oxaindolizidine diastereomers in which the four chiral centers at C-6, C-7, C-8 and C-8a in **2** and **3** have been systematically modified by inversion. This series has been assayed against several α - and β -glycosidases to map the geometrical requirements for biological activity in this family of compounds.

Results and Discussion

Synthesis and Structure of Reducing 2-Oxacastanospermine Diastereomers

The general synthetic strategy developed for the assembly of the bicyclic skeleton of 2-oxaindolizidines starts from hexofuranose precursors and is depicted in Figure 2. It involves the intramolecular nucleophilic addition of the nitrogen atom of five-membered cyclic (thio)carbamate derivatives with pseudo-C-nucleoside structure (II) to the masked carbonyl group of the aldose through the open chain aldehydo form (III). The key 5,6-(cyclic thiocarbamate) or (cyclic carbamate) intermediates are accessible from 5amino-5-deoxyaldohexofuranose derivatives (I) by thiocarbonylation or carbonylation reactions, respectively. The stereochemical complementarity of the hydroxylation profiles of the piperidine ring of the natural (+)-castanospermine (2, 3) and its 6-epi (4, 5), 7-epi (6, 7), 8-epi (8, 9) and 6,8adi-epi diastereomers (10, 11) implies that the starting monosaccharide precursors have the D-gluco, D-manno, D-allo, Dgalacto and L-gulo configurations, respectively (Figure 2).

The 6-epi-castanospermine analogues 4 and 5 were synthesized from the previously reported 1-O-acetyl-5-azido-5-deoxy-2,3-*O*-isopropylidene- α -D-mannofuranose **12**^[21] by reduction of the azido group. Thiocarbonylation of the resulting vic amino alcohol 13 with the carbon disulfide/dicyclohexylcarbodiimide system afforded the oxazolidine-2thione derivative 14. Similarly, carbonylation using triphosgene/diisopropylethylamine provided the corresponding cyclic carbamate 15. Sequential removal of the anomeric acetyl and the cyclic acetal groups by treatment with methanolic sodium methoxide and aqueous trifluoroacetic acid provided, after neutralization with a basic ion-exchange resin, the target 2-oxaindolizidine derivatives 4 and 5, whose structures were further confirmed by their transformation into the corresponding tetra-O-acetates 16 and 17 (Scheme 1).

To access the 7-*epi*-2-oxacastanospermine derivatives **6** and **7**, a synthetic scheme analogous to that previously reported for the preparation of **2** and **3**,^[24] involving the D-*allo*-configured amino alcohol **23** as pivotal intermediate, was considered. Compound **23** was obtained from the known selectively protected 1,2-*O*-isopropylidene- α -D-allo-furanose derivative **18**^[17] by double inversion at C-5, as de-



Figure 2. Synthetic strategy for the preparation of the reducing (+)-2-oxacastanospermine diastereomers from 5,6-cyclic (thio)carbamate hexofuranoses (II), obtained from 5-amino-5-deoxysugar precursors (I), via the open-chain *aldehydo* form (III).



Scheme 1. Reagents and conditions: a, H₂, Pd/C, MeOH, room temp., 2 h, quant.; b, for 14: CS₂, DCC, CH₂Cl₂, $-10 \,^{\circ}\text{C} \rightarrow$ room temp., 2.5 h, 78%; for 15: triphosgene, DIPEA, CH₂Cl₂, $0 \,^{\circ}\text{C} \rightarrow$ room temp., 15 min, 65%; c, 1. NaOMe, MeOH; 2. 90% TFA/ water, 0 °C, 15 min, 74–82%; d, Ac₂O/pyridine, 95–98%.

picted in Scheme 2. Trifluoromethanesulfonation at O-5 followed by $S_N 2$ nucleophilic displacement using nitrite anion afforded, after work up, the corresponding L-talofuranose alcohol **19**. A second triflation/inversion process using the azide anion as the nucleophile provided the 5-azido-5-deoxy-D-allofuranose derivative **20**. Selective removal of the trityl group with boron trifluoride–diethyl ether complex

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 $(\rightarrow 21)$, deacetylation $(\rightarrow 22)$ and catalytic hydrogenation of the azido group led to 23. The subsequent (thio)carbonylation reactions, using the above conditions, proceeded with total chemoselectivity to give the desired cyclic (thio)carbamates 24 and 25. Trifluoroacetic acid catalyzed hydrolysis of the isopropylidene group did not result in the expected furanose \rightarrow piperidine rearrangement, however, even after prolonged treatment with Amberlite IRA-68 (OH⁻) anion-exchange resin, probably because of the higher basicity of the endocyclic nitrogen in these compounds. Acetvlation of the reaction mixture with acetic anhydride/ pyridine afforded a mixture in which D-allofuranose derivatives predominated, as seen from the NMR spectra of the crude mixture, from which the β -peracetate 26 could be isolated. Deacetylation of the acetylated mixture with methanolic sodium methoxide proceeded with concomitant isomerization of the furanose thiocarbamates to give the desired 2-oxaindolizidine 6 as the only product.



Scheme 2. Reagents and conditions: a, 1. Tf₂O, CH₂Cl₂, pyridine, -25 °C \rightarrow room temp., 30 min; 2. NaNO₂, DMF, room temp., 18 h, 51%; b, 1. Tf₂O, CH₂Cl₂, pyridine, -25 °C \rightarrow room temp., 30 min; 2. NaN₃, DMF, room temp., 18 h, 85%; c, BF₃-Et₂O, MeOH, 0 °C \rightarrow room temp., 2 h, 80%; d, NaOMe, MeOH, 85%; e, H₂, Pd/C, MeOH, room temp., 1 h, quant.; f, for **24**: CS₂, DCC, CH₂Cl₂, -10 °C \rightarrow room temp., 5 h, 89%; for **25**: triphosgene, DIPEA, CH₂Cl₂, 0 °C \rightarrow room temp., 20 min, 89%; g, 1.90% TFA/water, 0 °C, 15 min; 2. Ac₂O/pyridine; h, 1. 90% TFA/water, 0 °C, 15 min; 2. Ac₂O/pyridine; 3. NaOMe, MeOH, 50–64%; i, Ac₂O/pyridine, 98%.

The above acetylation/deacetylation procedure proved to be very efficient in promoting the intramolecular nucleophilic addition of the nitrogen atom in pseudo-*C*-nucleoside (thio)carbamates to the aldehyde functionality in fully unprotected hexofuranose derivatives. Interestingly, the mixture of peracetates can easily be purified by silica gel column chromatography so that the subsequent deacetylation step provides the target bicyclic derivative in high purity, thus avoiding the need of a final purification by gel-permeation chromatography. In this way, the 3-oxo-2-oxaindolizidine glycomimetic 7 was obtained from carbamate 25. Both 6 and 7 afforded the corresponding tetraacetates 27 and 28 upon conventional acetylation. No reversion into the furanose isomers was observed under these conditions (Scheme 2).

To access 8-epi-castanospermine analogues of the sp² azasugar type using the above strategy, D-galacto-configured precursors were required. The synthetic scheme parallels that already discussed above for the D-allose series but starts from the known 3-O-acetyl-1,2-O-isopropylidene-6-O-trityl-α-D-galactofuranose 29.^[17] Thus, double inversion at C-5 by triflation/nitrite addition (\rightarrow 30) followed by triflation/azide substitution (\rightarrow 31), detritylation (\rightarrow 32), deacetylation (\rightarrow 33) and reduction of the azide group provided the key D-galacto amino alcohol 34. Thiocarbonylation with carbon disulfide/dicyclohexylcarbodiimide or carbonylation with triphosgene/diisopropylethylamine afforded the corresponding five-membered (thio)carbamate (35 and 36, respectively), which after deacetalation, acetylation and deacetylation as above led to the target reducing tetrahydroxyindolizidines 8 and 9, respectively, with a substitution pattern at the six-membered ring analogous to that of α-Dgalactopyranosides. Small samples of 8 and 9 were acetylated (\rightarrow 37 and 38, respectively) to confirm their structures (Scheme 3).

Inverting the configuration at position C-8a in any of the 2-oxaindolizidine derivatives **4–9** implies the need of a (thio)carbamate monosaccharide precursor of the non-natural L-series. Since the L-gulo vicinal amino alcohol **39**, the C-5 epimer of the D-mannofuranose derivative **13**, had been prepared previously by our group,^[17] an investigation of its suitability as a precursor for 6,8-di-*epi*-castanospermine analogues seemed intriguing. After removal of the protecting groups in the corresponding cyclic (thio)carbamate (**40** and **41**) and subsequent acetylation/deacetylation, the furanose \rightarrow pyranose rearrangement took place to give **10** and **11**, respectively, as the only detectable diastereomers. In accord with earlier reactions, further acetylation afforded exclusively the tetraacetoxyindolizidines **42** and **43**, respectively (Scheme 4).

All the new tetrahydroxy-2-oxaindolizidine analogues **4**– **11** existed in D₂O solution as single diastereomers. The high-field pseudoanomeric C-5 resonance (82.7–75.7 ppm) confirmed the amino acetal bicyclic structure, while the vicinal ${}^{3}J(H,H)$ values of the piperidine ring were in agreement with a configurational pattern that matched that of the related hexopyranoses. The configuration at the hemiaminal centre C-5 was inferred directly from the corresponding $J_{5,6}$ value, indicative of a 5*R* configuration in the cases of the 7- and 8-*epi*-castanospermine glycomimetics (**6**, **7** and **8**, **9**, respectively) and of a 5*S* configuration in the case of **10** and **11** ($J_{5,6} = 3.7$ –4.5 Hz for *cis* axial/equatorial relative dispositions). The 5*R* configuration of the 6-*epi* dia-



Scheme 3. Reagents and conditions: a, 1. Tf₂O, CH₂Cl₂, pyridine, -25 °C \rightarrow room temp., 30 min; 2. NaNO₂, DMF, room temp., 24 h, 51%; b, 1. Tf₂O, CH₂Cl₂, pyridine, -25 °C \rightarrow room temp., 30 min; 2. NaN₃, DMF, room temp., 24 h, 51%; c, BF₃-Et₂O, MeOH, 0 °C \rightarrow room temp., 2 h, 75%; d, NaOMe, MeOH, 80%; e, H₂, Pd/C, MeOH, room temp., 1 h, 96%; f, for **35**: CS₂, DCC, CH₂Cl₂, -10 °C \rightarrow room temp., 6 h, 86%; for **36**: triphosgene, DIPEA, CH₂Cl₂, 0 °C \rightarrow room temp., 10 min, 90%; g, 1. 90% TFA/water, 0 °C, 15 min; 2. Ac₂O/pyridine; 3. NaOMe, MeOH, 65-89%; h, Ac₂O/pyridine, 82-85%.

stereomers 4 and 5 ($J_{5,6} = 2.9$ and 2.8 Hz, respectively, for *trans* diequatorial relative dispositions of the corresponding protons) was confirmed by the absence of NOE contacts between H-5 and H-8a. In all cases, a single compound, with a pseudoanomeric configuration identical to that of its unprotected precursor, was detected after conventional acetylation, thus confirming the high configurational integ-



Scheme 4. Reagents and conditions: a, for 40: CS_2 , DCC, CH_2Cl_2 , $-10 \,^{\circ}C \rightarrow$ room temp., 8 h, 70%; for 41: triphosgene, DI-PEA, CH_2Cl_2 , 0 °C \rightarrow room temp., 45 min, 43%; b, 1. 90% TFA/ water, 0 °C, 15 min; 2. Ac₂O/pyridine; 3. NaOMe, MeOH, 50%; c, Ac₂O/pyridine, 87–90%.

rity imparted by the pseudoamide-type endocyclic nitrogen atom.

Mapping the Structure/Glycosidase Inhibitory Properties of Reducing 2-Oxacastanospermine Diastereomers

The inhibitory activities of the 2-oxacastanospermine diastereomers 4–11 were first screened against yeast α -glucosidase and almond β -glucosidase (Table 1). The (+)-castanospermine analogues 2 and 3 exhibited total anomeric selectivity with these two glycosidases, behaving as potent competitive inhibitors of the enzyme with α specificity (K_i = 40 and 2.2 μ M, respectively).^[24] Inverting the configuration at C-6 (4, 5 or 10, 11) or C-8 (8, 9) was found to have a deleterious effect on the glucosidase inhibition ability, while the C-7 epimer 7 retained a certain amount of activity, although two orders of magnitude weaker than that of the *D*-gluco analogue 3 ($K_i = 540 \mu$ M). These results are consistent with those obtained with the parent tetrahy-

Table 1. Comparison of inhibitory activities^[a] (K_i , μ M) for reducing (+)-2-oxacastanospermine derivatives (**2**, **3**),^[b] and the corresponding 6-*epi* (**4**, **5**), 7-*epi* (**6**, **7**), 8-*epi* (**8**, **9**) and 6,8a-di-*epi* (**10**, **11**) diastereomers.^[c,d]

Compound	α-Glucosidase (yeast)	α-Mannnosidase (Jack bean)	α-Galactosidase (coffee beans)	β-Galactosidase (bovine liver, cytosolic)
2	40	n.i.	n.i.	n.i.
3	2.2	n.i.	n.i.	n.i.
4	n.i.	n.i.	n.i.	n.i.
5	n.i.	n.i.	n.i.	n.i.
6	n.i.	n.i.	n.i.	n.i
7	540	n.i.	n.i.	n.i.
8	n.i.	n.i.	137	267
9	n.i.	n.i.	259	n.i.
10	n.i.	n.i.	n.i.	n.i.
11	ni	ni	ni	ni

[a] Inhibition, when detected, was always of the competitive type. [b] See ref.^[24]. [c] The new compounds **4–11** showed no inhibition when assayed against β -glucosidase (almonds), trehalase (pig kidney), isomaltase (yeast), invertase (yeast), amyloglucosidase (*Aspergillus niger*) and α -L-fucosidase (pig kidney). [c] n.i. = no inhibition detected.

droxyindolizidine alkaloid series^[28] and show the importance of the correct number and stereochemistry of the hydroxy groups for the specific inhibition of glycosidases.

Compounds 4 and 5 are related to α -D-mannopyranosides in the same way as 2 and 3 are related to α -D-glucopyranosides. Yet, neither 4 nor 5 inhibited Jack bean α -mannosidase, which may be due to an unfavourable disposition about the bond homologous to C-5-C-6 (i.e. C-8-C-8a) as compared with that in the bound substrates. The D-galactopyranoside analogues 8 and 9 behaved as moderate competitive inhibitors of green coffee α -galactosidase ($K_i = 137$ and 259 μ M, respectively). Compound 9 exhibited total α anomeric selectivity with bovine liver β -galactosidase, in agreement with the α -pseudoanomeric configuration at C-5, while the thioxo analogue **8** inhibited the β -enzyme with half the activity of the inhibition of the α form (K_i = 267 μM, respectively). No cross-configurational activity was observed in any case when a larger glycosidase panel, which included trehalase (pig kidney), isomaltase (yeast), invertase (yeast), amyloglucosidase (Aspergillus niger) and α -L-fucosidase (pig kidney), was considered.

Conclusions

The results described herein demonstrate the concept of stereoelectronic control of the stereochemistry at amino acetal centres which allows access to reducing glycomimetics of the sp² azasugar type that retain a defined pseudoanomeric configuration in aqueous solution. By exploiting the ability of the nitrogen atom of oxazolidine-2-(thi)one to act as a nucleophile in intramolecular addition to the masked aldehyde group in hexose precursors, a series of epimeric reducing castanospermine analogues has been prepared which has allowed structure/glycosidase inhibition specificity and potency relationships to be evaluated. The data obtained underline the critical influence of the hydroxylation profile on the inhibitory activity and support the importance of the configuration at pseudoanomeric centres in anomeric specificity.

Experimental Section

General Remarks: Optical rotations were measured at room temperature in 1-cm or 1-dm tubes. Unit for ε values: mm⁻¹ cm⁻¹. IR spectra were recorded on a FTIR instrument. ¹H (and ¹³C) NMR spectra were recorded at 500 (125.7) and 300 (75.5) MHz. 1D TOCSY as well as 2D COSY and HMQC experiments were carried out to assist in signal assignment. The FABMS spectra were recorded by using a primary beam consisting of Xe atoms with a maximum energy of 8 keV. The samples were dissolved in m-nitrobenzyl alcohol or thioglycerol as the matrices and the positive ions were separated and accelerated over a potential of 7 keV. NaI was added as a cationizing agent. TLC was performed with E. Merck precoated TLC plates, silica gel 30F-245, with visualization by UV light and by charring with 10% H₂SO₄ or 0.2% w/v cerium(IV) sulphate/5% ammonium molybdate in 2 м H₂SO₄. Column chromatography was carried out with silica gel 60 (E. Merck, 230–400 mesh). Fully deprotected compounds were purified by

GPC (Sephadex G-10, 1:1 MeOH/H₂O). Acetylations were effected conventionally with 1:1 pyridine/Ac₂O (10 mL per 1 g of sample). Deacetylations were effected following the method of Zemplén by treatment with methanolic NaOMe (0.1 equiv. per mol of acetate) at room temperature for 1 h followed by neutralization with Amberlite IR-120 (H⁺) ion-exchange resin or solid CO₂. Microanalyses were performed by the Instituto de Investigaciones Químicas (Sevilla, Spain).

Materials: 1-O-Acetyl-5-azido-5-deoxy-2,3-O-isopropylidene-α-Dmannofuranose (12) was obtained from 2,3:5,6-di-O-isopropylidene-L-gulonolactone as reported previously.^[21] 3-O-Acetyl-1,2-Oisopropylidene-6-O-trityl- α -D-allofuranose (18) was prepared from 3-*O*-acetyl-1,2:5,6-di-*O*-isopropylidene-α-D-allofuranose.^[17] 3-0-Acetyl-1,2-O-isopropylidene-6-O-trityl-α-D-galactofuranose (29) was prepared from 3-O-acetyl-1,2-O-isopropylidene-a-D-galactofuranose.^[17] 1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-β-Lgulofuranose (39) was prepared from 1-O-acetyl-2,3:5,6-di-O-isopropylidene-a-D-mannofuranose following the previously reported procedure.^[17] The glycosidases α -glucosidase (from yeast), β -glucosidase (from almonds), β -glucosidase (from bovine liver, cytosolic), trehalase (from pig kidney), α-galactosidase (from green coffee beans), α -mannosidase (from Jack beans), α -fucosidase (from bovine kidney) and amyloglucosidase (from Aspergillus Niger), used in the inhibition studies, as well as α, α' -trehalose and the corresponding o- and p-nitrophenyl glycoside substrates were purchased from Sigma Chemical Co.

1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-a-D-mannofuranose (13): A solution of 1-O-acetyl-5-azido-5-deoxy-2,3-O-isopropylidene-a-D-mannofuranose 12 (313 mg, 1.20 mmol) and 10% Pd/ C (109 mg) in MeOH (10 mL) was hydrogenated under an atmospheric pressure of hydrogen using a balloon. The suspension was stirred for 2 h, filtered through Celite and concentrated to give 13 in quantitative yield, as seen by NMR, as a hygroscopic solid that was used in the next step without further purification. $R_{\rm f} = 0.35$ (10:1:1 MeCN/H₂O/NH₄OH). $[a]_{D}^{22} = +55.3$ (c = 1.0, MeOH). ¹H NMR (300 MHz, CD₃OD): δ = 1.51, 1.63 (2 s, 6 H, CMe₂), 2.23 (s, 3 H, MeCO), 3.10 (ddd, $J_{5,6a}$ = 3.3, $J_{5,6b}$ = 6.2, $J_{4,5}$ = 9.1 Hz, 1 H, H-5), 3.72 (dd, $J_{6a,6b}$ = 11.0 Hz, 1 H, H-6b), 3.92 (dd, 1 H, H-6a), 4.13 (dd, $J_{3,4}$ = 3.7 Hz, 1 H, H-4), 4.91 (d, $J_{2,3}$ = 5.9 Hz, 1 H, H-2), 5.19 (dd, 1 H, H-3), 6.04 (s, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CD₃OD): δ = 20.9 (MeCO), 24.7, 26.3 (CMe₂), 52.9 (C-5), 64.2 (C-6), 80.8 (C-3), 83.3 (C-4), 86.3 (C-2), 101.8 (C-1), 114.0 (*C*Me₂), 171.1 (CO) ppm. FABMS: m/z (%) = 262 (100) [M + H]⁺. HRFABMS: m/z = 262.128786; calcd. for C₁₁H₂₀NO₆: 262.129063.

1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-a-D-mannofuranose 5,6-Cyclic Thiocarbamate (14): CS2 (6.2 mmol) and DCC (0.62 mmol) were added to a stirred solution of 13 (161 mg, 0.62 mmol) in CH₂Cl₂ (2.6 mL) at -10 °C. The reaction mixture was allowed to reach room temperature and stirred for 2.5 h. The solvent was removed under reduced pressure and the residue was purified by column chromatography (1:3 \rightarrow 1:1 EtOAc/petroleum ether) to give 14. Yield: 145 mg (78%). $R_{\rm f} = 0.32$ (1:1 EtOAc/petroleum ether). $[a]_{D}^{22} = +39.0$ (c = 1.0, CH₂Cl₂). UV (CH₂Cl₂): λ = 248 nm (ε = 22.1). IR (KBr): \tilde{v}_{max} = 3325, 2999, 1738, 1521, 1442, 1386, 1220, 1100 cm⁻¹. ¹H NMR (300 MHz, CDCl₃, 313 K): δ = 1.31, 1.46 (2s, 6 H, CMe₂), 2.07 (s, 3 H, MeCO), 4.21 (dd, $J_{3,4}$ = 3.8, $J_{4,5} = 7.4$ Hz, 1 H, H-4), 4.29 (m, 1 H, H-5), 4.67 (dd, $J_{5,6b} =$ 5.4, $J_{6a,6b}$ = 9.3 Hz, 1 H, H-6b), 4.71 (d, $J_{2,3}$ = 5.9 Hz, 1 H, H-2), 4.74 (t, $J_{5,6a}$ = 9.3 Hz, 1 H, H-6a), 4.86 (dd, 1 H, H-3), 6.15 (s, 1 H, H-1), 8.00 (br. s, 1 H, NH) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 20.8$ (MeCO), 24.1, 25.5 (CMe₂), 55.4 (C-5), 73.2 (C-6), 77.3

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(C-3), 81.7 (C-4), 84.6 (C-2), 100.1 (C-1), 113.6 (*C*Me₂), 169.1 (CO), 189.8 (CS) ppm. FABMS: m/z (%) = 326 (100) [M + Na]⁺, 304 (80) [M + H]⁺. C₁₂H₁₇O₆NS (303.34): C 47.51, H 5.65, N 4.62; found C 47.45, H 5.67, N 4.64.

1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-a-D-mannofuranose 5,6-Cyclic Carbamate (15): Diisopropylethylamine (11.4 mmol) and triphosgene (1.71 mmol) were added to a stirred solution of 13 (299 mg, 1.14 mmol) in CH₂Cl₂ (5.9 mL) at 0 °C. The reaction mixture was allowed to reach room temperature and stirred for 15 min. The solvent was removed under reduced pressure and the residue was purified by column chromatography (2:1 EtOAc/petroleum ether) to give 15. Yield: 210 mg (65%). $R_{\rm f} = 0.23$ (3:1 EtOAc/petroleum ether). $[a]_{D}^{22} = +28.0$ (c = 1.0, CH₂Cl₂). IR (KBr): $\tilde{v}_{max} = 2990, 1752, 1657, 1426, 1259, 1093 \text{ cm}^{-1}$. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.27$, 1.42 (2s, 6 H, CMe₂), 2.02 (s, 3 H, MeCO), 4.07 (ddd, $J_{5.6b} = 4.8$, $J_{4.5} = 7.2$, $J_{5.6a} = 8.1$ Hz, 1 H, H-5), 4.12 (dd, $J_{3,4}$ = 3.7 Hz, 1 H, H-4), 4.38 (dd, $J_{6a,6b}$ = 9.2 Hz, 1 H, H-6b), 4.45 (dd, 1 H, H-6a), 4.67 (d, $J_{2,3} = 5.8$ Hz, 1 H, H-2), 4.79 (dd, 1 H, H-3), 6.11 (s, 1 H, H-1), 6.11 (s, 1 H, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 21.0$ (*Me*CO), 24.4, 25.7 (C*Me*₂), 51.6 (C-5), 67.8 (C-6), 78.9 (C-3), 82.8 (C-4), 84.8 (C-2), 100.4 (C-1), 113.5 (CMe₂), 159.5 (CO carbamate), 169.3 (CO ester) ppm. FABMS: m/z (%) = 310 (100) [M + Na]⁺. C₁₂H₁₇O₇N (287.28): C 50.17, H 5.96, N 4.87; found C 50.05, H 5.89, N 4.76.

(5R,6S,7S,8R,8aR)-5,6,7,8-Tetrahydroxy-3-thioxo-2-oxaindolizidine (4): Compound 14 (121 mg, 0.40 mmol) was subjected to conventional Zemplén O-deacetylation, followed by deacetalation with 90% TFA/water (5 mL). The reaction mixture was concentrated, and the residue was coevaporated several times with water to eliminate traces of acid and then purified by GPC (Sephadex G-10, 1:1 MeOH/H₂O) to give 4. Yield: 72.5 mg (82%). $R_{\rm f} = 0.41$ (45:5:3 EtOAc/EtOH/H₂O). $[a]_{D}^{22}$ = +8.0 (c = 1.0, MeOH). UV (MeOH): λ = 248.0 nm (ε = 15.0). ¹H NMR (500 MHz, D₂O): δ = 3.81 (t, $J_{7,8} = J_{8,8a} = 9.7$ Hz, 1 H, H-8), 3.90 (dd, $J_{6,7} = 2.9$ Hz, 1 H, H-7), 4.13 (t, $J_{5,6}$ = 2.9 Hz, 1 H, H-6), 4.17 (ddd, $J_{8a,1b}$ = 8.1, $J_{8a,1a}$ = 9.2 Hz, 1 H, H-8a), 4.55 (dd, $J_{1a,1b}$ = 9.2 Hz, 1 H, H-1b), 4.88 (t, 1 H, H-1a), 5.80 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D₂O): δ = 59.9 (C-8a), 72.4 (C-1), 72.8 (C-7), 73.7 (C-8), 74.8 (C-6), 82.7 (C-5), 190.5 (CS) ppm. FABMS: m/z (%) = 244 (60) [M + Na]⁺. C₇H₁₁O₅NS (221.24): C 38.00, H 5.01, N 6.33; found C 37.98, H 5.10, N 6.38.

(5*R*,6*S*,7*S*,8*R*,8a*R*)-5,6,7,8-Tetrahydroxy-3-oxo-2-oxaindolizidine (5): Zemplén *O*-deacetylation of 15 (115 mg, 0.40 mmol), followed by deacetalation with 90% TFA/water (5 mL), elimination of the traces of acid by coevaporation with water and purification by GPC (Sephadex G-10, 1:1 MeOH/H₂O) afforded 5. Yield: 60.7 mg (74%). *R*_f = 0.16 (45:5:3 EtOAc/EtOH/H₂O). $[a]_{D^2}^{22}$ = +4.6 (*c* = 1.0, MeOH). ¹H NMR (500 MHz, D₂O): δ = 3.68 (t, *J*_{7,8} = *J*_{8,8a} = 9.5 Hz, 1 H, H-8), 3.75 (dd, *J*_{6,7} = 2.8 Hz, 1 H, H-7), 3.87 (ddd, *J*_{8a,1b} = 6.5, *J*_{8a,1a} = 9.0 Hz, 1 H, H-8a), 4.00 (t, *J*_{5,6} = 2.8 Hz, 1 H, H-6), 4.26 (dd, *J*_{1a,1b} = 9.0 Hz, 1 H, H-1b), 4.56 (t, 1 H, H-1a), 5.20 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D₂O): δ = 54.0 (C-8a), 67.6 (C-1), 70.1 (C-7), 70.3 (C-8), 71.0 (C-6), 77.2 (C-5), 158.6 (CO carbamate) ppm. FABMS: *m/z* (%) = 228 (10) [M + Na]⁺. C₇H₁₁O₆N (205.17): C 40.98, H 5.40, N 6.83; found C 41.06, H 5.42, N 6.87.

(5*R*,6*S*,7*S*,8*R*,8a*R*)-5,6,7,8-Tetraacetoxy-3-thioxo-2-oxaindolizidine (16): Conventional acetylation of 4 (30 mg, 0.13 mmol) afforded 16. Yield: 53 mg (98%). $R_f = 0.51$ (3:1 EtOAc/petroleum ether). $[a]_D^{22} = +5.0$ (c = 1.0, CH₂Cl₂). ¹H NMR (500 MHz, CDCl₃): $\delta = 2.00-2.20$ (4 s, 12 H, 4 MeCO), 4.16 (ddd, $J_{8a,1b} = 6.8$, $J_{8a,1a} = 9.5$, $J_{8,8a} = 10.0$ Hz, 1 H, H-8a), 4.55 (dd, $J_{1a,1b} = 9.5$ Hz, 1 H, H-1b), 4.65 (t, 1 H, H-1a), 5.25 (t, $J_{7,8} = 10.0$ Hz, 1 H, H-8), 5.35 (dd, 1 H, H-7), 5.40 (t, $J_{5,6} = J_{6,7} = 2.8$ Hz, 1 H, H-6), 6.89 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 20.6$ (*Me*CO), 56.7 (C-8a), 67.8 (C-6), 68.2 (C-7), 69.0 (C-8), 71.0 (C-1), 77.0 (C-5), 167.6, 169.4, 169.7, 170.0, (CO), 187.6 (CS) ppm. FABMS: *m/z* (%) = 412 (60) [M + Na]⁺. C₁₅H₁₉O₉NS (389.39): C 46.27, H 4.92, N 3.60, S 8.23; found C 46.25, H 4.99, N 3.57, S 8.42.

(5*R*,6*S*,7*R*,8*R*,8a*R*)-5,6,7,8-Tetraacetoxy-3-oxo-2-oxaindolizidine (17): Conventional acetylation of 5 (30 mg, 0.15 mmol) afforded 17. Yield: 52 mg (95%). $R_f = 0.24$ (1:1 EtOAc/petroleum ether). $[a]_{22}^{22} = +7.1$ (c = 1.0, CH₂Cl₂). ¹H NMR (300 MHz, CDCl₃): $\delta =$ 2.07–2.09 (4 s, 12 H, 4 MeCO), 3.96 (ddd, $J_{8a,1b} = 5.7$, $J_{8a,1a} = 8.3$, $J_{8,8a} = 9.2$ Hz, 1 H, H-8a), 4.32 (dd, $J_{1a,1b} = 9.2$ Hz, 1 H, H-1b), 4.42 (dd, 1 H, H-1a), 5.21 (t, $J_{7,8} = 9.2$ Hz, 1 H, H-8), 5.30 (dd, $J_{6,7} = 2.4$ Hz, 1 H, H-7), 5.34 (t, $J_{5,6} = 2.4$ Hz, 1 H, H-6), 6.34 (d, 1 H, H-5) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 20.5$ (*Me*CO), 53.1 (C-8a), 66.1 (C-1), 67.7 (C-7), 68.3 (C-8), 68.9 (C-6), 74.4 (C-5), 154.8 (CO carbamate), 167.7, 169.1, 169.6, 169.9 (CO ester) ppm. FABMS: *m/z* (%) = 396 (100) [M + Na]⁺. C₁₅H₁₉O₁₀N (373.32): C 48.26, H 5.13, N 3.75; found C 48.16, H 5.17, N 3.70.

3-*O*-Acetyl-1,2-*O*-isopropylidene-6-*O*-trityl-β-L-talofuranose (19): Pyridine (0.92 mL) and trifluoromethanesulfonic anhydride (1.22 mL, 7.38 mmol) were added under nitrogen to a solution of 3-O-acetyl-1,2-O-isopropylidene-6-O-trityl-α-D-allofuranose (18) (2.74 g, 5.43 mmol) in CH₂Cl₂ (15 mL) at -25 °C. The reaction mixture was allowed to reach room temperature and after stirring for 30 min, the mixture was diluted with CH₂Cl₂ (15 mL), washed with iced saturated aqueous NaHCO3 (15 mL), dried (MgSO4) and concentrated. The resulting triflate ester was dissolved in DMF (14 mL), NaNO₂ (1.75 g, 25.4 mmol) was added and the reaction mixture was stirred at room temperature for 18 h. The solvent was removed under reduced pressure and the resulting residue was dissolved in CH_2Cl_2 (15 mL) and washed with water (2×10 mL). The organic extract was dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography using $1:6 \rightarrow 1:4$ EtOAc/petroleum ether as eluent to give 19. Yield: 1.40 g (51%). $R_{\rm f} = 0.30$ (1:2 EtOAc/petroleum ether). $[a]_{\rm D}^{22} = +37.6$ (c = 1.0, CH₂Cl₂). IR (KBr): \tilde{v}_{max} = 3436, 3047, 2991, 1736, 1363, 1236, 1061 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 1.34, 1.54 (2 s, 6 H, CMe₂), 2.10 (s, 3 H, MeCO), 2.27 (d, $J_{OH,5}$ = 7.0 Hz, 1 H, OH), 3.25 (dd, $J_{5,6b}$ = 5.6, $J_{6a,6b}$ = 9.5 Hz, 1 H, H-6b), 3.34 (dd, $J_{5,6a}$ = 7.3 Hz, 1 H, H-6a), 3.75 (m, 1 H, H-5), 4.23 (dd, $J_{4,5} = 2.0$, $J_{3,4} =$ 8.9 Hz, 1 H, H-4), 4.81 (dd, $J_{1,2} = 3.7$, $J_{2,3} = 4.7$ Hz, 1 H, H-2), 4.88 (dd, 1 H, H-3), 5.80 (d, 1 H, H-1), 7.23-7.54 (m, 15 H, Ph) ppm. ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 20.7$ (*Me*CO), 26.4, 26.6 (CMe2), 60.4 (C-6), 64.9 (C-5), 71.9 (C-3), 77.3 (C-4), 77.4 (C-2), 86.8 (CPh₃), 104.3 (C-1), 113.1 (CMe₂), 127.0-143.7 (Ph), 170.3 (CO) ppm. FABMS: m/z (%) = 527 (100) [M + Na]⁺. C₃₀H₃₂O₇ (504.59): C 71.41, H 6.39; found C 71.36, H 6.24.

3-O-Acetyl-5-azido-5-deoxy-1,2-O-isopropylidene-6-O-trityl-a-D-allofuranose (20): Pyridine (0.47 mL) and trifluoromethanesulfonic anhydride (0.62 mL, 3.75 mmol) were added under nitrogen to a solution of **19** (1.40 g, 2.77 mmol) in CH₂Cl₂ (7.5 mL) at -25 °C. The reaction mixture was allowed to reach room temperature and, after stirring for 30 min, the mixture was diluted with CH₂Cl₂ (15 mL), washed with iced saturated aqueous NaHCO₃ (15 mL), dried (MgSO₄) and concentrated. The resulting triflate ester was dissolved in DMF (11 mL), NaN₃ (1.26 g, 19.38 mmol) was added and the reaction mixture was stirred at room temperature for 18 h. The solvent was removed under reduced pressure and the resulting residue was dissolved in CH₂Cl₂ (10 mL) and washed with water (2×5 mL). The organic extract was dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography to give **20**. Yield: 1.24 g (85%). $[a]_{D}^{22}$ = +49.2 (c = 1.0, CH₂Cl₂). IR (KBr): \hat{v}_{max} = 3150, 2101, 1744, 1633, 1386, 1236, 1093 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 1.32, 1.52 (2 s, 6 H, CMe₂), 1.92 (s, 3 H, MeCO), 3.18 (dd, $J_{5,6b}$ = 8.5, $J_{6a,6b}$ = 9.9 Hz, 1 H, H-6b), 3.28 (dd, $J_{5,6a}$ = 4.1 Hz, 1 H, H-6a), 3.89 (dt, $J_{4,5}$ = 4.1 Hz, 1 H, H-5), 4.21 (dd, $J_{3,4}$ = 8.5 Hz, 1 H, H-4), 4.74 (dd, $J_{2,3}$ = 4.1 Hz, 1 H, H-3), 4.78 (dd, $J_{1,2}$ = 3.6 Hz, 1 H, H-2), 5.78 (d, 1 H, H-1), 7.25–7.47 (m, 15 H, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.4 (*Me*CO), 26.5 (*CMe*₂), 62.4 (C-6), 63.1 (C-5), 72.3 (C-3), 76.4 (C-4), 77.3 (C-2), 87.2 (CPh₃), 103.9 (C-1), 113.0 (*CMe*₂), 127.1–143.3 (Ph), 169.6 (CO) ppm. FABMS: *m/z* (%) = 552 (100) [M + Na]⁺. C₃₀H₃₁N₃O₆ (529.60): C 68.04, H 5.90, N 7.94; found C 68.14, H 5.81, N 7.84.

3-O-Acetyl-5-azido-5-deoxy-1,2-O-isopropylidene-a-D-allofuranose (21): BF₃·Et₂O complex (0.84 mL) and MeOH (5 mL) were added to a solution of the tritylated azido derivative 20 (1.85 g, 3.5 mmol) in CH₂Cl₂ (25 mL) at 0 °C under argon. The reaction mixture was allowed to reach room temperature and stirred for 2 h, diluted with CH_2Cl_2 (15 mL), then washed with saturated aqueous NaHCO₃ $(2 \times 15 \text{ mL})$, dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography $(1:2 \rightarrow 1:1 \text{ EtOAc/petro-}$ leum ether) to give 21 (804 mg, 80%) as an amorphous solid. $R_{\rm f}$ = 0.27 (1:2 EtOAc/petroleum ether). $[a]_{D}^{22} = +165.8 (c = 1.0, CH_2Cl_2).$ IR (KBr): $\tilde{v}_{max} = 3443$, 2990, 2114, 1740, 1374, 1231, 1064 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 1.32, 1.53 (2 s, 6 H, CMe₂), 2.12 (s, 3 H, MeCO), 3.60 (dd, $J_{5,6b}$ = 7.0, $J_{6a,6b}$ = 11.5 Hz, 1 H, H-6b), 3.75 (dd, $J_{5.6a}$ = 4.5 Hz, 1 H, H-6a), 3.79 (dt, $J_{4.5}$ = 4.5 Hz, 1 H, H-5), 4.25 (dd, 1 H, H-4), 4.84 (m, 2 H, H-2, H-3), 5.82 (d, $J_{1,2}$ = 3.7 Hz, 1 H, H-1) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 20.8 (MeCO), 26.8 (CMe₂), 62.1 (C-6), 64.3 (C-5), 72.9 (C-3), 77.4 (C-4), 77.7 (C-2), 104.3 (C-1), 113.6 (CMe₂), 170.3 (CO) ppm. FABMS: m/z (%) = 310 (100) [M + Na]⁺. C₁₁H₁₇N₃O₆ (287.28): C 45.99, H 5.96, N 14.63; found C 45.97, H 5.90, N 14.58.

5-Azido-5-deoxy-1,2-O-isopropylidene-a-D-allofuranose (22): Conventional deacetylation of 21 (2.8 mmol) and purification by column chromatography (1:1 EtOAc/petroleum ether) afforded 22. Yield: 584 mg (85%). $R_{\rm f} = 0.24$ (1:1 EtOAc/petroleum ether). $[a]_{\rm D}^{22}$ = +48.0 (c = 1.0, CH₂Cl₂). IR (KBr): \tilde{v}_{max} = 3420, 2125, 1641, 1386, 1093 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 1.36, 1.57 (2 s, 6 H, CMe₂), 2.81 (t, $J_{OH,6a}$ = 5.0 Hz, 1 H, OH), 3.25 (d, $J_{OH,3}$ = 8.2 Hz, 1 H, OH), 3.77 (dd, $J_{5,6b} = 4.4$, $J_{6a,6b} = 12.2$ Hz, 1 H, H-6b), 3.85 (ddd, $J_{5.6a}$ = 6.2 Hz, 1 H, H-6a), 3.96 (ddd, $J_{4.5}$ = 3.1 Hz, 1 H, H-5), 4.00 (dd, $J_{3,4}$ = 8.2 Hz, 1 H, H-4), 4.18 (td, $J_{2,3}$ = 5.0 Hz, 1 H, H-3), 4.65 (dd, $J_{1,2}$ = 3.7 Hz, 1 H, H-2), 5.84 (d, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 26.3, 26.4 (CMe₂), 61.6 (C-6), 62.8 (C-5), 70.4 (C-3), 78.8 (C-2), 81.1 (C-4), 103.6 (C-1), 113.0 (*C*Me₂) ppm. FABMS: m/z (%) = 268 (100) [M + Na]⁺. C₉H₁₅N₃O₅ (245.24): C 44.08, H 6.16, N 17.13; found C 43.90, H 6.19, N 16.88.

5-Amino-5-deoxy-1,2-*O***-isopropylidene-***a***-D-allofuranose (23):** A solution of azido sugar **22** (235 mg, 0.96 mmol) and 10% Pd/C (95 mg) in MeOH (8.8 mL) was hydrogenated under an atmospheric pressure of hydrogen using a balloon. The mixture was stirred for 1 h, filtered through Celite and concentrated to give **23** (210 mg, 99%) as a hygroscopic solid that was used in the next step without further purification. $[a]_{D}^{22} = +44.8$ (c = 1.0, MeOH). IR (KBr): $\tilde{v}_{max} = 3396$, 1641, 1450, 1386, 1093 cm⁻¹. ¹H NMR (300 MHz, CD₃OD): $\delta = 1.33$, 1.53 (2 s, 6 H, CMe₂), 3.58 (dd, $J_{5,6b} = 5.5$, $J_{6a,6b} = 11.2$ Hz, 1 H, H-6b), 3.70 (dd, $J_{5,6a} = 4.0$ Hz, 1 H, H-6a), 3.00 (td, $J_{4,5} = 5.5$ Hz, 1 H, H-5), 3.83 (dd, $J_{3,4} = 9.1$ Hz, 1 H, H-4), 3.95 (dd, $J_{2,3} = 3.7$ Hz, 1 H, H-3), 4.56 (t, $J_{1,2} = 3.7$ Hz,

1 H, H-2), 5.72 (d, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CD₃OD): δ = 26.6, 26.8 (*CMe*₂), 55.1 (C-5), 63.9 (C-6), 73.6 (C-3), 80.7 (C-4), 81.0 (C-2), 105.1 (C-1), 113.7 (*CMe*₂) ppm. FABMS: *m/z* (%) = 242 (100) [M + Na]⁺, 220 (60) [M + H]⁺. HRFABMS: *m/z* = 220.118101; calcd. for C₉H₁₈NO₅: 220.118498.

5-Amino-5-deoxy-1,2-O-isopropylidene-α-D-allofuranose 5,6-Cyclic Thiocarbamate (24): Thiocarbonylation of 23 (193 mg, 0.88 mmol) by treatment with CS₂/DCC for 5 h as above described for the preparation of 14, followed by purification by column chromatography $(2:1 \rightarrow 3:1 \text{ EtOAc/petroleum ether})$ yielded 24. Yield: 204 mg (89%). $R_{\rm f} = 0.36$ (3:1 EtOAc/petroleum ether). $[a]_{\rm D}^{22} = +5.6$ (c = 0.5, MeOH). UV (MeOH): $\lambda = 243$ nm ($\varepsilon = 15.9$). IR (KBr): \tilde{v}_{max} = 3364, 2983, 1641, 1514, 1386, 1093 cm⁻¹. ¹H NMR (300 MHz, CD₃OD): δ = 1.33, 1.52 (2 s, 6 H, CMe₂), 3.80 (dd, $J_{2,3}$ = 4.6, $J_{3,4}$ = 9.2 Hz, 1 H, H-3), 3.94 (dd, $J_{4,5}$ = 3.1 Hz, 1 H, H-4), 4.33 (ddd, $J_{5,6a} = 6.3, J_{5,6b} = 9.1$ Hz, 1 H, H-5), 4.56 (dd, $J_{1,2} = 3.6$ Hz, 1 H, H-2), 4.63 (t, $J_{6a.6b} = 9.1$ Hz, 1 H, H-6b), 4.68 (dd, 1 H, H-6a), 5.74 (d, 1 H, H-1) ppm. ¹³C NMR (125.7 MHz, CD₃OD): δ = 26.6, 26.9 (CMe2), 58.2 (C-5), 71.5 (C-6), 72.9 (C-3), 79.9 (C-4), 81.1 (C-2), 105.3 (C-1), 114.1 (CMe2), 191.7 (CS) ppm. FABMS: m/z (%) = 284 (70) $[M + Na]^+$, 262 (100) $[M + H]^+$. $C_{10}H_{15}NO_5S$ (261.30): C 45.96, H 5.79, N 5.36; found C 46.06, H 5.72, N 5.43.

5-Amino-5-deoxy-1,2-O-isopropylidene-α-D-allofuranose 5,6-Cyclic Carbamate (25): Carbonylation of 23 (210 mg, 0.96 mmol) by treatment with triphosgene/diisopropyl(ethyl)amine for 20 min, as described for the preparation of 15, and purification by column chromatography (EtOAc \rightarrow 45:5:3 EtOAc/EtOH/H₂O) afforded 25. Yield: 210 mg (89%). $R_{\rm f}$ = 0.55 (45:5:3 EtOAc/EtOH/H₂O). $[a]_{\rm D}^{22}$ = +22.5 (c = 0.52, MeOH). IR (KBr): $\tilde{v}_{max} = 3396$, 2985, 1744, 1633, 1402, 1093 cm⁻¹. ¹H NMR (500 MHz, CD₃OD): δ = 1.33, 1.52 (2 s, 6 H, CMe₂), 3.85 (dd, $J_{2,3}$ = 3.6, $J_{3,4}$ = 9.1 Hz, 1 H, H-3), 3.90 $(dd, J_{4,5} = 3.4 \text{ Hz}, 1 \text{ H}, \text{H-4}), 4.13 (ddd, J_{5,6a} = 6.2, J_{5,6b} = 9.6 \text{ Hz},$ 1 H, H-5), 4.43 (m, 2 H, H-6a, H-6b), 4.57 (t, $J_{1,2} = 3.6$ Hz, 1 H, H-2), 5.73 (d, 1 H, H-1) ppm. $^{13}\mathrm{C}$ NMR (125.7 MHz, CD₃OD): δ $= 26.7, 26.9 (CMe_2), 54.3 (C-5), 66.7 (C-6), 73.0 (C-3), 80.5 (C-4),$ 81.1 (C-2), 105.3 (C-1), 114.0 (CMe2), 162.4 (CO carbamate) ppm. FABMS: m/z (%) = 268 (100) [M + Na]⁺. C₁₀H₁₅O₆N (245.24): C 48.98, H 6.16, N 5.71; found C 49.00, H 6.17, N 5.64.

N-Acetyl-1,2,3-tri-O-acetyl-5-amino-5-deoxy-α-D-allofuranose 5,6-Cyclic Thiocarbamate (26): Compound 24 (104 mg, 0.40 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated and the residue was coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 5.0 mL) and the peracetylated mixture was purified by column chromatography using 1:1 EtOAc/petroleum ether as eluent. A mixture of furanose and indolizidine isomers was obtained according to NMR spectroscopy. A second column chromatography using 4:1 Et₂O/petroleum ether as eluent allowed a pure sample of **26** to be isolated. Yield: 21 mg (14%). $R_{\rm f} = 0.26$ (4:1 Et₂O/ petroleum ether). $[a]_{D}^{22}$ = +53.7 (c = 1.0, CH₂Cl₂). UV (CH₂Cl₂): λ = 266 nm (ε = 12.5). IR (KBr): \tilde{v}_{max} = 3301, 2920, 1752, 1698, 1426, 1371, 1220, 1088 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.07, 2.12, 2.13, (3 s, 9 H, MeCO ester), 2.82 (s, 3 H, MeCO amide), 4.49 (dd, $J_{5,6b}$ = 8.4, $J_{6a,6b}$ = 9.0 Hz, 1 H, H-6b), 4.54 (dd, $J_{4,5}$ = 3.9, $J_{3,4} = 8.9$ Hz, 1 H, H-4), 4.65 (dd, $J_{5,6a} = 2.7$ Hz, 1 H, H-6a), 4.95 (ddd, 1 H, H-5), 5.24 (dd, J_{2,3} = 4.5 Hz, 1 H, H-3), 5.34 (d, 1 H, H-2), 6.16 (s, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.3-20.8 (MeCO ester), 26.0 (MeCO amide), 59.0 (C-5), 67.3 (C-6), 70.3 (C-3), 73.9 (C-2), 78.4 (C-4), 97.4 (C-1), 168.1-171.2 (CO), 185.5 (CS) ppm. FABMS: m/z (%) = 412 (80) [M + Na]⁺. C₁₅H₁₉NO₉S (389.39): C 46.27, H 4.92, N 3.60; found C 46.12, H 4.71, N 3.51

(5R,6R,7R,8R,8aR)-5,6,7,8-Tetrahydroxy-3-thioxo-2-oxaindolizidine (6): Compound 24 (104 mg, 0.40 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated, and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 5.0 mL) and the peracetylated mixture was purified by column chromatography using 1:1 EtOAc/petroleum ether as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 6 as the only reaction product. Yield: 44 mg (50%). $R_{\rm f} = 0.41$ $(45:5:3 \text{ EtOAc/EtOH/H}_2\text{O})$. $[a]_{D}^{22} = -27.5$ (c = 1.0, H₂O). UV (H₂O): $\lambda = 243$ nm ($\varepsilon = 16.0$). ¹H NMR (500 MHz, D₂O): $\delta = 3.63$ (dd, $J_{7,8} = 2.7$, $J_{8,8a} = 9.1$ Hz, 1 H, H-8), 3.67 (dd, $J_{6,7} = 2.7$, $J_{5,6}$ = 4.4 Hz, 1 H, H-6), 4.03 (t, 1 H, H-7), 4.24 (td, $J_{8a,1b}$ = 7.8, $J_{8a,6a}$ = 9.1 Hz, 1 H, H-8a), 4.41 (dd, $J_{1a,1b}$ = 9.1 Hz, 1 H, H-1b), 4.66 (t, 1 H, H-1a), 5.64 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D_2O): $\delta = 52.6$ (C-8a), 66.7 (C-6), 70.2 (C-8), 72.1 (C-7, C-1), 78.1 (C-5), 187.4 (CS) ppm. FABMS: m/z (%) = 244 (60) [M + Na]⁺. C₇H₁₁NO₅S (221.24): C 38.00, H 5.01, N 6.33; found C 37.92, H 4.93, N 6.31.

(5R,6R,7R,8R,8aR)-5,6,7,8-Tetrahydroxy-3-oxo-2-oxaindolizidine (7): Compound 25 (99 mg, 0.40 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated, and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 7.3 mL) and the peracetylated mixture was purified by column chromatography using 5:2 EtOAc/petroleum ether as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 7 as the only reaction product. Yield: 44.7 mg (64%). $R_{\rm f} = 0.61$ (6:3:1 MeCN/H₂O/NH₄OH). $[a]_{D}^{22} = +33.0$ (c = 1.0, H₂O). ¹H NMR (500 MHz, D₂O): δ = 3.67 (t, $J_{7,8}$ = 2.8 Hz, $J_{8,8a}$ = 10.0 Hz, 1 H, H-8), 3.69 (dd, $J_{6,7}$ = 2.8, $J_{5,6}$ = 4.4 Hz, 1 H, H-6), 4.07 (t, 1 H, H-7), 4.11 (m, 1 H, H-8a), 4.25 (dd, $J_{8a,1b} = 7.2$, $J_{1a,1b} = 8.7$ Hz, 1 H, H-1b), 4.61 (t, *J*_{8a,1a} = 8.7 Hz, 1 H, H-1a), 5.22 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D₂O): δ = 50.1 (C-8a), 68.5 (C-8), 68.8 (C-1), 71.8 (C-6), 73.5 (C-7), 75.7 (C-5), 159.1 (CO) ppm. FABMS: m/z (%) = 228 (100) [M + Na]⁺. HRFABMS: m/z = 206.066916; calcd. for C₇H₁₂O₆N: 206.066462. C₇H₁₁O₆N (205.17): C 40.98, H 5.40, N 6.83; found C 40.90, H 5.47, N 6.78.

(5*R*,6*R*,7*R*,8*R*,8a*R*)-5,6,7,8-Tetraacetoxy-3-thioxo-2-oxaindolizidine (27): Conventional acetylation of **6** (20.4 mg, 0.09 mmol) afforded 27. Yield: 34 mg (98%). $R_{\rm f} = 0.55$ (2:1 EtOAc/petroleum ether). $[a]_{22}^{22} = +31.0$ (c = 1.0, CH₂Cl₂). UV (CH₂Cl₂): $\lambda = 249.6$ nm ($\varepsilon =$ 12.2). IR (KBr): $\tilde{v}_{\rm max} = 2983$, 1752, 1434, 1371, 1220, 1053 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 2.00-2.23$ (4 s, 12 H, 4 MeCO), 4.49 (t, $J_{8a,1b} = J_{1a,1b} = 8.5$ Hz, 1 H, H-1b), 4.46 (td, $J_{1a,8a} = 8.5$, $J_{8,8a} = 10.0$ Hz, 1 H, H-8a), 4.71 (t, 1 H, H-1a), 4.95 (dd, $J_{3,4} =$ 2.9 Hz, 1 H, H-8), 5.18 (dd, $J_{6,7} = 2.9$, $J_{5,6} = 4.5$ Hz, 1 H, H-6), 5.65 (t, 1 H, H-7), 7.11 (d, 1 H, H-5) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 20.2-20.4$ (*Me*CO), 52.2 (C-8a), 65.7 (C-6), 67.0 (C-7), 69.3 (C-8), 70.9 (C-1), 74.0 (C-5), 168.3-169.5 (CO), 187.1 (CS) ppm. FABMS: *m/z* (%) = 412 (100) [M + Na]⁺. C₁₅H₁₉NO₉S (389.39): C 46.27, H 4.92, N 3.60; found C 46.39, H 4.78, N 3.54.

(5*R*,6*R*,7*R*,8*R*,8a*R*)-5,6,7,8-Tetracetoxy-3-oxo-2-oxaindolizidine (28): Conventional acetylation of 7 (60 mg, 0.29 mmol) afforded 28. Yield: 106 mg (98%). *R*_f = 0.50 (2:1 EtOAc/petroleum ether). [*a*]_D²² = +38.6 (*c* = 1.0, CH₂Cl₂). IR (NaCl): \tilde{v}_{max} = 2991, 2928, 1776, 1752, 1418, 1371, 1228, 1053 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.00–2.17 (4 s, 12 H, 4 MeCO), 4.16 (dd, *J*_{8a,1b} = 8.0, $J_{1a,1b} = 8.6 \text{ Hz}, 1 \text{ H}, \text{H-1b}, 4.30 \text{ (dt}, J_{8a,1a} = 8.0, J_{8,8a} = 10.0 \text{ Hz}, 1 \text{ H}, \text{H-8a}), 4.49 \text{ (dd}, 1 \text{ H}, \text{H-1a}), 4.93 \text{ (dd}, J_{7,8} = 3.0 \text{ Hz}, 1 \text{ H}, \text{H-8}), 5.13 \text{ (dd}, J_{6,7} = 3.0, J_{5,6} = 4.6 \text{ Hz}, 1 \text{ H}, \text{H-6}), 5.63 \text{ (t}, 1 \text{ H}, \text{H-7}), 6.65 \text{ (d}, 1 \text{ H}, \text{H-5}) \text{ ppm}. ^{13}\text{C} \text{ NMR} (75.5 \text{ MHz}, \text{CDCl}_3): \delta = 20.2-20.5 (MeCO), 48.5 (C-8a), 66.1 (C-6), 66.2 (C-1), 67.4 (C-7), 69.8 (C-8), 71.4 (C-5), 154.5 (CO carbamate), 168.6-169.7 (CO ester) ppm. FABMS: <math>m/z$ (%) = 396 (100) [M + Na]⁺. C_{15}H_{19}O_{10}N (373.32): C 48.26, H 5.13, N 3.75; found C 48.44, H 5.47, N 3.52.

3-O-Acetyl-1,2-O-isopropylidene-6-O-trityl-β-L-altrofuranose (30): Pyridine (0.67 mL) and trifluoromethanesulfonic anhydride (0.88 mL, 5.3 mmol) were added under nitrogen to a solution of 3-O-acetyl-1,2-O-isopropylidene-6-O-trityl-α-D-galactofuranose 29 (1.99 g, 3.95 mmol) in CH₂Cl₂ (10.6 mL) at -25 °C. The reaction mixture was allowed to reach room temperature and after stirring for 30 min, the mixture was diluted with CH₂Cl₂ (15 mL), washed with iced saturated aqueous NaHCO₃ (15 mL), dried (MgSO₄) and concentrated. The resulting triflate ester was dissolved in DMF (10 mL), NaNO₂ (1.27 g, 18.4 mmol) was added and the reaction mixture was stirred at room temperature for 24 h. The solvent was removed under reduced pressure and the resulting residue was dissolved in CH_2Cl_2 (20 mL) and washed with water (2×10 mL). The organic extract was dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography using 1:6 EtOAc/ petroleum ether as eluent to give **30**. Yield: 1.0 g (51%). $R_{\rm f} = 0.49$ (1:2 EtOAc/petroleum ether). $[a]_{D}^{22} = -16.1$ (c = 1.0, CH₂Cl₂). IR (KBr): $\tilde{v}_{max} = 3476, 1744, 1386, 1220, 1093 \text{ cm}^{-1}$. ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.29, 1.48 (2 \text{ s}, 6 \text{ H}, \text{CMe}_2), 2.07 (\text{s}, 3 \text{ H}, \text{CDCl}_3)$ MeCO), 2.58 (d, $J_{OH,5}$ = 5.4 Hz, 1 H, OH), 3.32 (dd, $J_{5.6b}$ = 5.4, $J_{6a,6b} = 9.6$ Hz, 1 H, H-6b), 3.38 (dd, $J_{5,6a} = 3.6$ Hz, 1 H, H-6a), 4.01 (m, 1 H, H-5), 4.18 (dd, $J_{3,4} = 1.2$, $J_{4,5} = 9.6$ Hz, 1 H, H-4), 4.56 (d, $J_{1,2}$ = 4.1 Hz, 1 H, H-2), 5.39 (d, 1 H, H-3), 5.89 (d, 1 H, H-1), 7.21–7.37 (m, 15 H, Ph) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 20.8$ (MeCO), 25.6, 26.5 (CMe₂), 63.1 (C-6), 70.1 (C-5), 77.6 (C-3), 86.8 (CPh₃), 84.6 (C-2), 85.6 (C-4), 105.7 (C-1), 112.4 (CMe_2) , 127.0–143.7 (Ph), 169.8 (CO) ppm. FABMS: m/z (%) = 527 (100) $[M + Na]^+$. $C_{30}H_{32}O_7$ (504.59): C 71.41, H 6.39; found C 71.29. H 6.36.

3-O-Acetyl-5-azido-5-deoxy-1,2-O-isopropylidene-6-O-trityl-a-D-galactofuranose (31): Pyridine (0.31 mL) and trifluoromethanesulfonic anhydride (0.38 mL, 2.3 mmol) were added under nitrogen to a solution of 30 (878 mg, 1.74 mmol) in CH₂Cl₂ (4.8 mL) at -25 °C. The reaction mixture was allowed to reach room temperature and, after stirring for 30 min, the mixture was diluted with CH₂Cl₂ (15 mL), washed with iced saturated aqueous NaHCO₃ (15 mL), dried (MgSO₄) and concentrated. The resulting triflate ester was dissolved in DMF (7 mL), NaN₃ (1.15 g, 17.7 mmol) was added and the reaction mixture was stirred at room temperature for 24 h. The solvent was removed under reduced pressure and the resulting residue was dissolved in CH₂Cl₂ (15 mL) and washed with water $(2 \times 10 \text{ mL})$. The organic extract was dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography using $1:8 \rightarrow 1:6$ EtOAc/petroleum ether as eluent to give 31. Yield: 475 mg (51%). $R_{\rm f} = 0.18$ (1:6 EtOAc/petroleum ether). $[a]_{\rm D}^{22}$ = -22.7 (c = 1.0, CH₂Cl₂). IR (KBr): \tilde{v}_{max} = 3420, 2101, 1744, 1633, 1379, 1220, 1093 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 1.32, 1.57 (2 s, 6 H, CMe₂), 2.02 (s, 3 H, MeCO), 3.30 (dd, J_{5,6b} = 6.5, $J_{6a,6b}$ = 10.3 Hz, 1 H, H-6b), 3.38 (dd, $J_{5,6a}$ = 3.8 Hz, 1 H, H-6a), 3.80 (ddd, $J_{4,5}$ = 7.8 Hz, 1 H, H-5), 4.08 (dd, $J_{3,4}$ = 3.0 Hz, 1 H, H-4), 4.54 (d, $J_{1,2}$ = 4.0 Hz, 1 H, H-2), 4.97 (d, 1 H, H-3), 5.87 (d, 1 H, H-1), 7.25–7.46 (m, 15 H, Ph) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): $\delta = 20.6 (MeCO), 26.8, 26.0 (CMe_2), 62.3 (C-5), 63.7 (C-$ 6), 77.1 (C-3), 83.9 (C-4), 84.6 (C-2), 87.2 (CPh₃), 105.2 (C-1), 113.3 (CMe₂), 127.0–143.6 (Ph), 169.5 (CO) ppm. FABMS: m/z (%) =

552 (10) [M + Na]⁺. C₃₀H₃₁N₃O₆ (529.60): C 68.04, H 5.90, N 7.94; found C 68.31, H 5.81, N 7.80.

3-O-Acetyl-5-azido-5-deoxy-1,2-O-isopropylidene-a-D-galactofuranose (32): BF₃-Et₂O complex (318 mL) and MeOH (1.6 mL) were added to a solution of the tritylated azido derivative 31 (572 mg, 1.08 mmol) in CH₂Cl₂ (8.5 mL) at 0 °C under argon. The reaction mixture was allowed to reach room temperature and stirred for 2 h, diluted with CH₂Cl₂ (10 mL), then washed with saturated aqueous NaHCO₃ (2×10 mL), dried (MgSO₄) and concentrated. The resulting residue was purified by column chromatography $(1:3 \rightarrow 1:1)$ EtOAc/petroleum ether) to give 32. Yield: 233 mg (75%). $R_{\rm f}$ 0.22 (1:2 EtOAc/petroleum ether). $[a]_{D}^{22} = -57.1$ (c = 0.98, CH₂Cl₂). IR (KBr): $\tilde{v}_{max} = 3435$, 2991, 2101, 1746, 1375, 1233, 1053 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 1.32, 1.59 (2 s, 6 H, CMe₂), 2.11 (s, 3 H, MeCO), 3.75 (dd, $J_{5,6b}$ = 5.6, $J_{6a,6b}$ = 12.1 Hz, 1 H, H-6b), 3.79 (td, J_{4.5} = J_{5.6a} = 9.3 Hz, 1 H, H-5), 3.84 (dd, 1 H, H-6a), 4.15 $(dd, J_{3,4} = 1.4 Hz, 1 H, H-4), 4.69 (d, J_{1,2} = 4.0 Hz, 1 H, H-2), 5.12$ (d, 1 H, H-3), 6.00 (d, 1 H, H-1) ppm. ¹³C NMR (125.7 MHz, $CDCl_3$): $\delta = 20.9 (MeCO), 25.9, 26.8 (CMe_2), 62.7 (C-6), 63.6 (C-6))$ 5), 77.8 (C-3), 84.4 (C-4), 86.4 (C-2), 106.0 (C-1), 113.2 (CMe₂), 170.9 (CO) ppm. FABMS: m/z (%) = 310 (100) [M + Na]⁺. C₁₁H₁₇N₃O₆ (287.28): C 45.99, H 5.96, N 14.63; found C 45.61, H 5.78, N 14.47.

5-Azido-5-deoxy-1,2-*O*-isopropylidene-*α*-D-galactofuranose (33): Conventional deacetylation of **32** (0.8 mmol) and purification by column chromatography (1:1 → 2:1 EtOAc/petroleum ether) afforded **33**. Yield: 171 mg (80%). $R_{\rm f} = 0.41$ (2:1 EtOAc/petroleum ether). $[a]_{\rm D}^{22} = -113.8$ (c = 1.0, CH₂Cl₂). IR (KBr): $\tilde{v}_{\rm max} = 3412$, 2991, 2109, 1641, 1386, 1093 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.34$, 1.55 (2 s, 6 H, CMe₂), 2.65 (br. s, 1 H, OH), 3.33 (br. s, 1 H, OH), 3.75 (dd, $J_{5,6b} = 4.3$, $J_{6a,6b} = 10.9$ Hz, 1 H, H-6b), 3.81 (m, 1 H, H-6a), 3.94 (m, 1 H, H-5), 4.04 (dd, $J_{3,4} = 3.3$, $J_{4,5} =$ 5.6 Hz, 1 H, H-4), 4.27 (br. d, 1 H, H-3), 4.59 (dd, $J_{2,3} = 0.9$, $J_{1,2} =$ 4.0 Hz, 1 H, H-2), 5.92 (d, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 26.0$, 26.8 (CMe₂), 62.2 (C-6), 63.2 (C-5), 75.9 (C-3), 86.5 (C-2), 86.8 (C-2), 105.1 (C-1), 113.2 (CMe₂) ppm. FABMS: m/z (%) = 268 (100) [M + Na]⁺. HRFABMS: m/z = 246.109779; calcd. for C₉H₁₆N₃O₅: 246.108996.

5-Amino-5-deoxy-1,2-O-isopropylidene- α -D-galactofuranose (34): A solution of azido sugar 33 (69.8 mg, 0.29 mmol) and 10% Pd/C (28.2 mg) in MeOH (2.6 mL) was hydrogenated under an atmospheric pressure of hydrogen using a balloon. The mixture was stirred for 1 h, filtered through Celite and concentrated to give 34 as a hygroscopic solid that was used in the next step without further purification. Yield: 59 mg (96%). $[a]_{D}^{22} = -25.3$ (c = 1.0, MeOH). IR (KBr): \tilde{v}_{max} = 3357, 2991, 1379, 1212, 1013 cm⁻¹. ¹H NMR (500 MHz, CD₃OD): δ = 1.29, 1.47 (2 s, 6 H, CMe₂), 3.08 (m, 1 H, H-5), 3.49 (dd, $J_{5,6b} = 6.0$, $J_{6a,6b} = 11.1$ Hz, 1 H, H-6b), 3.63 (dd, $J_{5,6a} = 4.4$ Hz, 1 H, H-6a), 3.77 (dd, $J_{3,4} = 1.7$, $J_{4,5} = 9.2$ Hz, 1 H, H-4), 4.17 (d, 1 H, H-3), 4.52 (d, $J_{1,2} = 4.0$ Hz, 1 H, H-2), 5.82 (d, 1 H, H-1) ppm. ¹³C NMR (125.7 MHz, CD₃OD): δ = 26.1, 27.1 (CMe2), 58.4 (C-5), 64.2 (C-6), 77.2 (C-3), 88.6 (C-2), 89.8 (C-4), 106.8 (C-1), 113.4 (CMe₂) ppm. FABMS: m/z (%) = 242 (100) $[M + Na]^+$, 220 (40) $[M + H]^+$. HRFABMS: m/z = 220.118354; calcd. for C₉H₁₈NO₅: 220.118498.

5-Amino-5-deoxy-1,2-*O*-isopropylidene-*a*-D-galactofuranose 5,6-Cyclic Thiocarbamate (35): Thiocarbonylation of 34 (57 mg, 0.26 mmol) by treatment with CS₂/DCC for 6 h, as above described for the preparation of 14, followed by purification by column chromatography (2:1 EtOAc/petroleum ether) gave 35. Yield: 58 mg (86%). $R_{\rm f} = 0.40$ (3:1 EtOAc/petroleum ether). $[a]_{\rm D}^{22} = -142.6$ (c = 0.54, MeOH). UV (MeOH): $\lambda = 244$ nm ($\varepsilon = 15.5$). IR (KBr): $\tilde{v}_{\rm max}$ = 3237, 2928, 1514, 1386, 1283, 1090 cm⁻¹. ¹H NMR (500 MHz, CD₃OD): δ = 1.30, 1.50 (2 s, 6 H, CMe₂), 3.84 (dd, $J_{3,4}$ = 2.7, $J_{4,5}$ = 8.5 Hz, 1 H, H-4), 4.02 (d, 1 H, H-3), 4.31 (ddd, $J_{6b,5}$ = 5.8, $J_{6a,5}$ = 9.2, H-5), 4.44 (dd, $J_{6a,6b}$ = Hz, 1 H, H-6b), 4.51 (d, $J_{1,2}$ = 3.7 Hz, 1 H, H-2), 4.71 (t, 1 H, H-6a), 5.82 (s, 1 H, H-1) ppm. ¹³C NMR (125.7 MHz, CD₃OD): δ = 26.3, 27.5 (*CMe*₂), 59.4 (C-5), 72.8 (C-6), 76.2 (C-3), 88.5 (C-2), 89.4 (C-4), 107.0 (C-1), 114.2 (*CMe*₂), 191.3 (CS) ppm. FABMS: m/z (%) = 284 (100) [M + Na]⁺, 262 (20) [M + H]⁺. C₁₀H₁₅NO₅S (261.30): C 45.96, H 5.79, N 5.36; found C 45.84, H 5.75, N 5.25.

5-Amino-5-deoxy-1,2-O-isopropylidene-a-D-galactofuranose 5,6-Cyclic Carbamate (36): Carbonylation of 34 (90 mg, 0.41 mmol) by treatment with triphosgene/diisopropylethylamine for 10 min, as described for the preparation of 15, and purification by column chromatography (EtOAc and 40:1 EtOAc/EtOH) yielded 36. Yield: 90 mg (90%). $R_{\rm f} = 0.53$ (20:1 EtOAc/EtOH). $[a]_{\rm D}^{22} = -50.6$ (c = 1.0, MeOH). IR (KBr): \tilde{v}_{max} = 3420, 2935, 1752, 1641, 1379, 1212, 1108, 1061 cm⁻¹. ¹H NMR (300 MHz, CD₃OD): δ = 1.30, 1.49 (2 s, 6 H, CMe₂), 3.81 (dd, $J_{3,4} = 2.8$, $J_{4,5} = 8.5$ Hz, 1 H, H-4), 3.99 (d, 1 H, H-3), 4.13 (td, $J_{5,6b} = 5.2$, $J_{5,6a} = 8.5$ Hz, 1 H, H-5), 4.19 (dd, $J_{6a,6b}$ = 8.5 Hz, 1 H, H-6b), 4.49 (t, 1 H, H-6a), 4.51 (d, $J_{1,2}$ = 3.8 Hz, 1 H, H-2), 5.87 (d, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, CD₃OD): δ = 26.3, 27.4 (CMe₂), 55.4 (C-6), 67.8 (C-6), 76.1 (C-3), 88.6 (C-2), 90.3 (C-4), 106.9 (C-1), 114.1 (CMe₂), 162.0 (CO) ppm. CIMS: m/z (%) = 246 (25) [M + H]⁺. C₁₀H₁₅NO₆ (245.24): C 48.98, H 6.16, N 5.71; found C, 48.92, H 6.08, N 5.68.

(5R,6R,7S,8S,8aR)-5,6,7,8-Tetrahydroxy-3-thioxo-2-oxaindolizidine (8): Compound 35 (102.5 mg, 0.40 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 1.6 mL) and the peracetylated mixture was purified by column chromatography using 1:1 EtOAc/petroleum ether as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 8 as the only reaction product. Yield: 60.9 mg (65%) $R_{\rm f} = 0.31$ (45:5:3 EtOAc/EtOH/H₂O). $[a]_{D}^{22} = -10.5$ (c = 1.0, H₂O). UV (H₂O): $\lambda =$ 244.0 nm (ε = 14.1). ¹H NMR (300 MHz, D₂O): δ = 3.80 (dd, J_{5.6} = 3.7, $J_{6,7}$ = 10.2 Hz, 1 H, H-6), 3.90 (dd, $J_{7,8}$ = 2.7 Hz, 1 H, H-7), 4.01 (br. t, $J_{8,8a} = 2.0$ Hz, 1 H, H-8), 4.44 (ddd, $J_{8a,1b} = 6.5$, $J_{8a,1a} = 9.4$ Hz, 1 H, H-8a), 4.55 (dd, $J_{1a,1b} = 9.4$ Hz, 1 H, H-1b), 4.68 (t, 1 H, H-1a), 5.84 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D_2O): $\delta = 57.7$ (C-8a), 68.2 (C-6), 69.9 (C-8), 70.1 (C-1), 70.2 (C-7), 79.2 (C-5), 187.8 (CS) ppm. CIMS: m/z (%) = 221 (15) [M]⁺, 222 (60) $[M + H]^+$. HRCIMS: m/z = 221.036008; calcd. for C₇H₁₁NO₅S: 221.035794. C₇H₁₁NO₅S (221.24): C 38.00, H 5.01, N 6.33; found C 38.01, H 4.94, N 6.27.

(5*R*,6*R*,7*S*,8*S*,8a*R*)-5,6,7,8-Tetrahydroxy-3-oxo-2-oxaindolizidine (9): Compound 36 (98.9 mg, 0.40 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 2.6 mL) and the peracetylated mixture was purified by column chromatography using 2:1 EtOAc/petroleum ether as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 9 as the only reaction product. Yield: 73 mg (89%). $R_{\rm f} = 0.56$ (6:3:1 MeCN/H₂O/NH₄OH). $[a]_{\rm D}^{22} = +40.0$ (c = 1.0, H₂O). ¹H NMR (300 MHz, D₂O): $\delta = 3.75$ (dd, $J_{5,6} = 4.1$ Hz, 1 H, H-6), 3.84 (dd, $J_{7,8} = 2.7$, $J_{6,7} = 10.3$ Hz, 1 H, H-7), 3.96 (dd, $J_{8,8a} = 1.8$ Hz, 1 H, H-8), 4.22 (ddd, $J_{8a,1b} = 6.3$, $J_{8a,1a} = 9.0$ Hz, 1 H, H-5), 4.34 (dd, $J_{1a,1b} = 9.0$ Hz, 1 H, H-1b), 4.47 (t, 1 H, H-1a) ppm. ¹³C NMR (125.7 MHz, D₂O): $\delta = 54.9$ (C-8a), 66.7 (C-1), 69.8 (C-6), 71.1 (C-8), 71.5 (C-7), 76.9 (C-5), 160.5 (CS) ppm. CIMS: m/z (%) = 206 (5) [M + H]⁺. C₇H₁₁NO₆ (205.27): C 40.98, H 5.40, N 6.83; found C 40.90, H 5.47, N 6.78.

(5*R*,6*R*,7*S*,8*S*,8*aR*)-5,6,7,8-Tetraacetoxy-3-thioxo-2-oxaindolizidine (37): Conventional acetylation of **8** (46 mg, 0.21 mmol) gave **37**. Yield: 67 mg (82%). $R_{\rm f}$ = 0.54 (2:1 EtOAc/petroleum ether). [*a*]₂₇²⁷ = +74.5 (*c* = 1.0, CH₂Cl₂). UV (CH₂Cl₂): λ = 249.6 nm (ε = 17.3). IR (NaCl): $\tilde{v}_{\rm max}$ = 2999, 2920, 1752, 1752, 1553, 1371, 1212, 1077 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.02–2.20 (4 s, 12 H, 4 MeCO), 4.21 (t, $J_{1a,1b}$ = $J_{8a,1b}$ = 8.5 Hz, 1 H, H-1b), 4.50 (td, $J_{8,8a}$ = 1.8, $J_{8a,1a}$ = 8.5 Hz, 1 H, H-8a), 4.61 (t, 1 H, H-1a), 5.31 (dd, $J_{7,8}$ = 2.6, $J_{6,7}$ = 10.7 Hz, 1 H, H-7), 5.39 (dd, $J_{5,6}$ = 3.8 Hz, 1 H, H-6), 5.53 (dd, 1 H, H-8), 7.23 (d, 1 H, H-5) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.2–20.5 (*Me*CO), 55.1 (C-8a), 65.2 (C-6), 66.7 (C-8), 67.3 (C-1), 67.7 (C-7), 75.4 (C-5), 168.2–170.0 (CO), 186.6 (CS) ppm. FABMS: *m*/*z* (%) = 412 (100) [M + Na]⁺. C₁₅H₁₉NO₉S (389.39): C 46.27, H 4.92, N 3.60; found C 46.55, H 4.67, N 3.59.

(5*R*,6*R*,7*S*,8*S*,8*aR*)-5,6,7,8-Tetraacetoxy-3-oxo-2-oxaindolizidine (38): Conventional acetylation of **9** (25 mg, 0.12 mmol) gave 38. Yield: 38 mg (85%). $R_{\rm f} = 0.42$ (2:1 EtOAc/petroleum ether). $[a]_{22}^{22}$ = +59.6 (*c* = 1.0, CH₂Cl₂). IR (NaCl): $\tilde{v}_{\rm max} = 2999$, 2928, 1752, 1537, 1371, 1212, 1077 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta =$ 2.00–2.20 (4 s, 12 H, 4 MeCO), 4.02 (dd, $J_{8,1\rm{b}} = 5.9$, $J_{1a,1\rm{b}} =$ 8.5 Hz, 1 H, H-1b), 4.32 (ddd, $J_{8,8\rm{a}} = 1.7$, $J_{8a,1\rm{a}} = 8.5$ Hz, 1 H, H-8a), 4.41 (t, 1 H, H-1a), 5.29 (dd, $J_{7,8} = 2.3$, $J_{6,7} = 10.9$ Hz, 1 H, H-7), 5.34 (dd, $J_{5,6} = 3.5$ Hz, 1 H, H-6), 5.49 (dd, 1 H, H-8), 6.78 (d, 1 H, H-5) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 20.0-20.4$ (*Me*CO), 51.3 (C-8a), 65.5 (C-6), 67.2 (C-8), 67.8 (C-7), 62.7 (C-1), 72.8 (C-5), 154.1 (CO carbamate), 168.4, 169.2, 169.9, 170.1 (CO ester) ppm. FABMS: *m/z* (%) = 396 (100) [M + Na]⁺. $C_{15}H_{19}NO_{10}$ (373.32): C 48.26, H 5.13, N 3.75; found C 48.15, H 4.95, N 3.74.

1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-B-L-gulofuranose 5,6-Cyclic Thiocarbamate (40): Thiocarbonylation of 1-O-acetyl-5amino-5-deoxy-2,3-O-isopropylidene-β-L-gulofuranose 39 (201 mg, 0.87 mmol) by treatment with CS_2/DCC for 8 h, as above described for the preparation of 14, followed by purification by column chromatography (1:2 \rightarrow 2:1 EtOAc/petroleum ether) gave 40. Yield: 185 mg (70%). $R_{\rm f} = 0.33$ (1:1 EtOAc/petroleum ether). $[a]_{\rm D}^{22} = +73$ $(c = 1.0, CH_2Cl_2)$. UV (CH_2Cl_2) : $\lambda = 249 \text{ nm}$ ($\varepsilon = 22.4$). IR (KBr): $\tilde{v}_{max} = 3330, 2988, 1742, 1505, 1466, 1379, 1265, 1094 \text{ cm}^{-1}$. ¹H NMR (500 MHz, CDCl₃): δ = 1.26, 1.45 (2 s, 6 H, CMe₂), 2.04 (s, 3 H, MeCO), 4.13 (dd, $J_{3,4}$ = 4.0, $J_{4,5}$ = 7.3 Hz, 1 H, H-4), 4.32 $(ddd, J_{5.6b} = 6.5, J_{5.6a} = 9.5 \text{ Hz}, 1 \text{ H}, \text{H-5}), 4.45 (dd, J_{6a.6b} = 9.5 \text{ Hz},$ 1 H, H-6b), 4.71 (d, $J_{2,3}$ = 5.9 Hz, 1 H, H-2), 4.76 (t, 1 H, H-6a), 4.82 (dd, 1 H, H-3), 6.13 (s, 1 H, H-1), 7.51 (br. s, 1 H, NH) ppm. ¹³C NMR (125.7 MHz, CDCl₃): δ = 20.9 (*Me*CO), 24.3, 25.7 (CMe2), 56.2 (C-5), 71.8 (C-6), 78.8 (C-3), 82.4 (C-4), 84.9 (C-2), 100.0 (C-1), 113.8 (CMe₂), 169.2 (CO), 189.6 (CS) ppm. FABMS: m/z (%) = 326 (100) [M + Na]⁺. C₁₂H₁₇O₆NS (303.34): C 47.51, H 5.65, N 4.62; found C 47.60, H 5.53, N 4.67.

1-O-Acetyl-5-amino-5-deoxy-2,3-O-isopropylidene-β-L-gulofuranose 5,6-Cyclic Carbamate (41): Carbonylation of **39** (211 mg, 0.81 mmol) by treatment with triphosgene/diisopropyl ethylamine for 45 min, as described for the preparation of **15**, and purification by column chromatography (1:1 EtOAc/petroleum ether → EtOAc) yielded **41**. Yield: 100 mg (43%). $R_{\rm f} = 0.24$ (3:1 EtOAc/petroleum ether). $[a]_{\rm D^2}^{22} = +78$ (c = 1.0, CH₂Cl₂). IR (KBr): $\tilde{v}_{\rm max} = 3405$, 2955, 1753, 1537, 1414, 1238, 1094 cm⁻¹. ¹H NMR (500 MHz, CDCl₃):
$$\begin{split} &\delta=1.25,\,1.42\,(2~{\rm s},\,6~{\rm H},\,{\rm CMe}_2),\,2.01\,({\rm s},\,3~{\rm H},\,{\rm MeCO}),\,4.05\,({\rm dd},\,J_{3,4}\\ &=3.8,\,J_{4,5}=7.1~{\rm Hz},\,1~{\rm H},\,{\rm H-4}),\,4.13\,({\rm m},\,1~{\rm H},\,{\rm H-5}),\,4.15\,({\rm dd},\,J_{5,6b}\\ &=5.7,\,J_{6a,6b}=6.9~{\rm Hz},\,1~{\rm H},\,{\rm H-6b}),\,4.49\,({\rm m},\,1~{\rm H},\,{\rm H-6a}),\,4.68\,({\rm d},\,J_{2,3}=5.9~{\rm Hz},\,1~{\rm H},\,{\rm H-2}),\,4.78\,({\rm dd},\,1~{\rm H},\,{\rm H-3}),\,5.75\,({\rm br},\,{\rm s},\,1~{\rm H},\,{\rm NH}),\\ &6.11\,({\rm s},\,1~{\rm H},\,{\rm H-1})~{\rm ppm}.\,\,^{13}{\rm C}\,\,{\rm NMR}\,\,(125.7~{\rm MHz},\,{\rm CDCl}_3):\,\delta=20.8\,\\(Me{\rm CO}),\,24.3,\,25.7\,({\rm CMe}_2),\,52.2\,({\rm C-5}),\,66.3\,({\rm C-6}),\,78.8\,({\rm C-3}),\,83.3\,\\({\rm C-4}),\,85.0\,({\rm C-2}),\,100.2\,({\rm C-1}),\,113.6\,({\rm CMe}_2),\,158.9\,({\rm CO}\,\,{\rm carbamate}),\,169.1\,({\rm CO}\,\,{\rm ester})\,{\rm ppm}.\,\,{\rm FABMS}:\,m/z\,\,(\%)=283\,\,(30)\,[{\rm M}\,+\,{\rm H}]^+,\,{\rm C}_{12}{\rm H}_{17}{\rm O}_7{\rm N}\,(287.28):\,{\rm C}\,\,50.17,\,{\rm H}\,\,5.96,\,{\rm N}\,\,4.87;\,{\rm found}\,\,{\rm C}\,\,50.07,\,{\rm H}\,\,5.85,\,{\rm N}\,\,4.75.\,\\ \end{split}$$

(5R,6S,7S,8R,8aS)-5,6,7,8-Tetrahydroxy-3-thioxo-2-oxaindolizidine (10): Compound 40 (90 mg, 0.31 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 1.6 mL) and the peracetylated mixture was purified by column chromatography using 100:1 CH₂Cl₂/MeOH as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 10 as the only reaction product. Yield: 34 mg (50%). $R_{\rm f} = 0.45$ (45:5:3 EtOAc/EtOH/H₂O). $[a]_{D}^{22} = +77 (c = 1.0, H_2O)$. UV (CH₂Cl₂): $\lambda =$ 247nm (ε = 10.8). ¹H NMR (500 MHz, D₂O): δ = 3.97 (dd, J_{6.7} = 3.4, $J_{5.6} = 4.4$ Hz, 1 H, H-6), 4.06 (dd, $J_{4.5} = 1.9$, $J_{7.8} = 3.4$ Hz, 1 H, H-8), 4.13 (t, 1 H, H-7), 4.63 (dd, $J_{8a,1b} = 6.7$, $J_{1a,1b} = 9.0$ Hz, 1 H, H-1b), 4.70 (ddd, $J_{8a,1a}$ = 9.0 Hz, 1 H, H-8a), 4.79 (t, 1 H, H-1a), 5.86 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, D₂O): δ = 55.4 (C-8a), 66.6 (C-6), 70.9 (C-8), 71.4 (C-1), 73.7 (C-7), 81.2 (C-5), 189.5 (CS) ppm. FABMS: m/z (%) = 222 (60) [M + H]⁺. C₇H₁₁O₅NS (221.24): C 38.00, H 5.01, N 6.33; found C 37.68, H 4.69, N 6.33.

(5R,6S,7S,8R,8aS)-5,6,7,8-Tetrahydroxy-3-oxo-2-oxaindolizidine (11): Compound 41 (90 mg, 0.31 mmol) was deacetalated by treatment with 90% TFA/water (5 mL). The reaction mixture was concentrated and the residue coevaporated several times with water to eliminate traces of acid. The resulting residue was subjected to conventional acetylation with Ac₂O/pyridine (1:1, 2.6 mL) and the peracetylated mixture was purified by column chromatography using 100:1 CH₂Cl₂/MeOH as eluent. The resulting crude peracetylated mixture of furanose and indolizidine isomers, according to NMR, was deacetylated with methanolic NaOMe to give 11 as the only reaction product. Yield: 32 mg (50%). $R_f = 0.28 (45:5:3)$ EtOAc/EtOH/H₂O). $[a]_D^{22} = +52$ (c = 1.0, MeOH). ¹H NMR (300 MHz, D₂O): δ = 3.92 (dd, $J_{6,7}$ = 3.2, $J_{5,6}$ = 4.5 Hz, 1 H, H-6), 4.00 (dd, $J_{8.8a}$ = 1.8, $J_{7.8}$ = 4.2 Hz, 1 H, H-8), 4.09 (t, 1 H, H-7), 4.40 (dd, $J_{8a,1b} = 8.6$, $J_{1a,1b} = 8.6$ Hz, 1 H, H-1b), 4.48 (ddd, $J_{8a,1a} = 7.5$ Hz, 1 H, H-8a), 4.57 (t, 1 H, H-6a), 5.34 (d, 1 H, H-1) ppm. ¹³C NMR (75.5 MHz, D_2O): $\delta = 48.8$ (C-8a), 64.4 (C-1), 64.5 (C-6), 68.2 (C-8), 71.4 (C-7), 75.2 (C-5), 158.6 (CO) ppm. CIMS: m/z (%) = 206 (30) [M + H]⁺. C₇H₁₁O₆N (205.17): C 40.98, H 5.40, N 6.83; found C 41.03, H 5.40, N 6.65.

(5*R*,6*S*,7*S*,8*R*,8a*S*)-5,6,7,8-Tetraacetoxy-3-thioxo-2-oxaindolizidine (42): Conventional acetylation of 10 (35 mg, 0.16 mmol) gave 42. Yield: 54 mg (87%). $R_f = 0.53$ (70:1 CH₂Cl₂/MeOH). $[a]_D^{22} = -50.4$ (c = 1.0, CH₂Cl₂). IR (KBr): $\tilde{v}_{max} = 2988$, 1750, 1371, 1227, 1094 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 2.00-2.17$ (4 s, 12 H, 4 MeCO), 4.23 (m, 1 H, H-1b), 4.64 (m, 2 H, H-8a, H-1a), 5.14 (dd, $J_{8,8a} = 2.0$, $J_{7,8} = 4.0$ Hz, 1 H, H-8), 5.29 (t, $J_{6,7} = 4.0$ Hz, 1 H, H-7), 5.36 (t, $J_{5,6} = 4.0$ Hz, 1 H, H-6), 7.14 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 20.4-20.7$ (*Me*CO), 52.8 (C-8a), 64.4 (C-1), 66.0 (C-6), 66.8 (C-7), 67.6 (C-8), 74.8 (C-5), 168.3– 169.6 (CO ester), 187.2 (CS) ppm. FABMS: *m/z* (%) = 412 (100) [M + Na]⁺. $C_{15}H_{19}O_9NS$ (389.39): C 46.27, H 4.92, N 3.60; found C 46.27, H 4.80, N 3.53.

(5*R*,6*S*,7*S*,8*R*,8*aS*)-5,6,7,8-Tetraacetoxy-3-oxo-2-oxaindolizidine (43): Conventional acetylation of 11 (30 mg, 0.14 mmol) gave 43. Yield: 47 mg (90%). $R_f = 0.47$ (20:1 CH₂Cl₂/MeOH). $[a]_{12}^{22} = -48.4$ (c = 1.0, CH₂Cl₂). IR (KBr): $\bar{v}_{max} = 2963$, 1750, 1559, 1374, 1258, 1094 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.99-2.17$ (4 s, 12 H, 4 MeCO), 4.03 (dd, $J_{8a,1b} = 6.0$, $J_{1a,1b} = 8.5$ Hz, 1 H, H-1b), 4.41 (t, $J_{8a,1a} = 8.5$ Hz, 1 H, H-1a), 4.46 (ddd, $J_{8,8a} = 2.0$ Hz, 1 H, H-8a), 5.11 (dd, $J_{7,8} = 4.5$ Hz, 1 H, H-8), 5.25 (dd, $J_{6,7} = 3.5$, $J_{5,6} = 4.5$ Hz, 1 H, H-6), 5.35 (dd, 1 H, H-7), 6.68 (d, 1 H, H-5) ppm. ¹³C NMR (125.7 MHz, CDCl₃): $\delta = 20.4-20.7$ (*Me*CO), 48.8 (C-8a), 63.1 (C-1), 64.7 (C-6), 66.3 (C-7), 67.2 (C-8), 72.2 (C-5), 154.8 (CO carbamate), 168.6–169.7 (CO ester) ppm. FABMS: *m/z* (%) = 396 (100) [M + Na]⁺. C₁₅H₁₉O₁₀N (373.32): C 48.26, H 5.13, N 3.75; found C 48.25, H 5.17, N 3.76.

General Procedure for the Inhibition Assay: Inhibitory potencies were determined by spectrophotometrically measuring the residual hydrolytic activities of the glycosidases against the respective o- (for β -glucosidase from bovine liver) or *p*-nitrophenyl α - or β -D-glycopyranoside, or α, α' -trehalose (for trehalase) in the presence of the corresponding iminoindolizidine derivative. Each assay was performed in phosphate or phosphate/citrate buffer (for α-mannosidase and amyloglucosidase) at the optimal pH for each enzyme. The $K_{\rm m}$ values for the different glycosidases used in the tests and the corresponding working pHs are listed as follows: α-glucosidase (yeast), $K_{\rm m} = 0.35 \,\mathrm{mM} \,\mathrm{(pH \, 6.8)}$; β -glucosidase (almonds), $K_{\rm m} =$ 3.5 mM (pH 5.5 and 7.0); β -glucosidase (bovine liver), $K_{\rm m} = 1.8$ mM (pH 6.8); trehalase (pig kidney), $K_{\rm m} = 5.0 \text{ mM}$ (pH 6.0); α -galactosidase (coffee beans), $K_{\rm m}$ = 2.02 mM (pH 6.8); α -mannosidase (Jack beans), $K_{\rm m} = 2.0 \text{ mM}$ (pH 5.5); α -fucosidase (bovine kidney), $K_{\rm m} =$ 0.2 mM (pH 5.5); amyloglucosidase (Aspergillus niger), $K_{\rm m}$ = 3.0 mM (pH 4.5). The reactions were initiated by addition of the enzyme to a solution of the substrate in the absence or presence of various concentrations of inhibitor. After incubating the mixture for 10-30 min at 37 or 55 °C (for amyloglucosidase) the reaction was quenched by the addition of 1 M Na₂CO₃ or a solution of GLC-Trinder (Sigma, for trehalase). The absorbance of the resulting mixture was determined at 405 or 505 nm (for trehalase). The K_i value and enzyme inhibition mode were determined from the slope of Lineweaver-Burk plots and double reciprocal analysis using a Sigma Plot program (version 4.14, Jandel Scientific).

Acknowledgments

We thank the Spanish Ministerio de Educación y Ciencia for financial support (contract numbers BQU2003–00937 and CTQ2004– 05854/BQU).

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Received: January 28, 2005 Published Online: June 2, 2005