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Synthesis and antimicrobial activity of new (E)-4-[piperidino (4'-methylpiperidino-, morpholino-) N-alkoxy]stilbenes

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Abstract

The synthesis of twenty-one new (E)-4-[piperidino-(4'-methylpiperidino-, morpholino-)N-alkoxy] stilbenes is reported. The compounds were tested for antimicrobial activities against Gram-negative, Gram-positive bacteria, and fungi. In particular, compounds **3b**, **3c**, **3f**, **3g**, **3h**, **3k**, **3l** showed good antibacterial activity against *Staphylococcus aureus* and **3h**, **3k**, **3m**, **3n** also against *Bacillus subtilis*, as well as **3h**, **3n** also against *Streptococcus faecalis*.

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Keywords: (E)-stilbenes; Piperidines; Morpholines; Antimicrobial activity

1. Introduction

(E)-stilbenes, hydroxylated at two to five positions, and their oligomers, are natural compounds, found in various families of plants. Hydroxylated stilbenes have been widely investigated because of their biological role in plant defense against pathogens and for their pharmacological properties [1-10]. Previously published results have shown that grapewines synthesize antimicrobial compounds in response to fungal infection. These compounds, which are referred to as phytoalexins, belong to the family of stilbenes [11–14]. Taking into regard the biological properties of (E)-stilbenols, the synthesis and antimicrobial activity of (E)-acetoxystilbenes [15] and (E)-azastilbenols and their derivatives have been studied previously in our laboratory [16-18]. In view of the fact that the substitution of acetyl group with hydroxy substituent of (E)-stilbenols, as well as that the N-substitution of benzyl and haloalkyl group with annular nitrogen atom of (E)-azastilbenols influences the antimicrobial activity, it may be of interest to direct further synthetic work towards new, yet undescribed (E)-4-[piperidino-(4'-methylpiperidino-, morpholino-) N-alkoxy] stilbenes 3a-3n, 5a-5g. It ought to be pointed out

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that on the other hand N-substituted derivatives of piperidine and morpholine showed a rather significant antimicrobial activity [19–24].

The purpose of this investigation was to elucidate the influence of the presence of the piperidine, 4-methylpiperidine, and morpholine ring, and the effect of the length of the alkyloxy bridge between the (E)-stilbene skeletone and piperidine (4'methylpiperidine, morpholine) ring on the antimicrobial activity of **3a-3n** and **5a-5g** in order to acquire information on the structural characteristic enhancing this activity. This paper presents the synthesis and characteristics of twenty-one new (E)-4-(piperidino-N-alkoxy)stilbenes **3a-3g**, (E)-4-(4'-methylpiperidino-N-alkoxy) stilbenes **3h-3n**, (E)-4-(morpholino-N-alkoxy) stilbenes **5a-5g** and the in vitro estimation of their antimicrobial activity against such microorganismus as Gram-positive and Gram-negative bacteria, yeast, dermatophytes and moulds.

2. Chemistry

The synthetic approach to obtaining (E)-4-(piperidino-N-alkoxy-)stilbenes **3a-3g**, (E)-4-(4`-methylpiperidino-N-alkoxy-) stilbenes **3h-3n**, and (E)-4-(morpholino-N-alkoxy-) stilbenes **5a-5g** followed the reactions shown in Scheme 1. We accomplished the synthesis of these compounds by the reaction of the respective (E)-4-(bromoalkoxy) stilbenes **1a-1g** with piperidine

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Scheme 1. Scheme of synthesis and structure of compounds 1a-1g, 2a-2b, 3a-3n, 4 and 5a-5g.

Table 1 Chemical and physical data of compounds **3a-3n** and **5a-5g**

2a (or 4'-methylpiperidine 2b, and morpholine 4) in DMF in the presence of triethylamine at room temperature. By manipulating the reaction time, it was found that after 10-25 hours 3a-3n and 5a-5g were the predominant products. (E)-4-(bromoalkoxy-)stilbenes 1a-1g were prepared according to a previously described procedure [25] by reacting equimolar amounts of the respective (E)-stilbenol-4, [(E)-4'-nitrostilbenol-4] and a proper dibromoalkane (1,2-dibromoethane; 1,3-dibromopropane; 1,4-dibromobutane; 1,5-dibromopentane) in 0.3 M NaOH solution of DMSO at room temperature. The structures of all new compounds obtained were determined by examining their UV/VIS, IR, ¹H NMR and ¹³C NMR spectra as well as on the basis of elemental analyses (Tables 1-5). Assignments of the ¹H NMR and ¹³C NMR resonances of these compounds were deduced on the basis of the signal multiplicities, and by the concerted application of the two-dimensional NMR technique HETCOR. The HETCOR results allow unequivocal assignment of the ¹³C NMR spectra proposed on the basis of chemical shift theory, additivity rules and by comparing the measured and calculated chemical shifts. (E) configuration in the stilbene part of the molecules of 3a-3n and 5a-5g was determined on the basis of their UV/VIS and IR spectra. It has been pointed out that in the UV/VIS spectra of **3a-3h** λ_{max} are in the range 321-381 nm, as well as in the spectra of 5a-5g, λ_{max} in the range 321-379 nm (Table 1). According to literature [26–28] (E)-stilbenes exhibit the values of λ_{max} in the range 290-360 nm, and for (Z)-stilbenes the values of λ_{max} are in the range 260-280 nm. The infrared spectra of 3a-3h and 5a-5g show a strong band in the range 964-970 cm⁻¹ which, according to literature [29,30], can be attributed to the out-ofplane deformation vibration of the C-H bond of the (E)-ethy-

Compd.	Yield	M.p.	TLC	IR (KBr)	UV/VIS	Anal
	[%]	[°Ĉ]	[R _f]	$[\mathrm{cm}^{-1}]$		
				CH=CH	λ_{max} (lg ϵ)	
3a	78	80-2	0.57 ^a	967	321 (4.47)	(C ₂₁ H ₂₅ NO) C, H, N
3b	78	108-9	$0.32^{\rm a}$	967	321 (4.46)	(C ₂₂ H ₂₇ NO) C, H, N
3c	76	87-90	0.24 ^a	969	321 (4.47)	(C ₂₃ H ₂₉ NO) C, H, N
3d	72	82-3	0.21 ^a	966	322 (4.44)	(C ₂₄ H ₃₁ NO) C, H, N
3e	42	50-2	0.30 ^a	966	377 (4.40)	(C ₂₂ H ₂₆ N ₂ O ₃) C, H, N
3f	59	75-8	0.24 ^a	967	375 (4.41)	(C ₂₃ H ₂₈ N ₂ O ₃) C, H, N
3g	56	83-6	0.22^{a}	963	378 (4.35)	(C ₂₄ H ₃₀ N ₂ O ₃) C, H, N
3h	25	73-4	0.71 ^b	966	321 (4.46)	(C ₂₂ H ₂₇ NO) C, H, N
3i	13	73-5	0.67^{b}	967	321 (4.39)	(C ₂₃ H ₂₉ NO) C, H, N
3ј	21	75-7	0.68 ^b	967	321 (4.37)	(C ₂₄ H ₃₁ NO) C, H, N
3k	27	82-3	0.69^{b}	966	321 (4.43)	(C ₂₅ H ₃₃ NO) C, H, N
31	65	86-8	0.51 ^b	967	378 (4.51)	(C ₂₃ H ₂₈ N ₂ O ₃) C, H, N
3m	40	78-80	0.50 ^b	968	379 (4.20)	(C ₂₄ H ₃₀ N ₂ O ₃) C, H, N
3n	42	79-80	0.43 ^b	968	381 (4.41)	(C ₂₅ H ₃₂ N ₂ O ₃) C, H, N
5a	58	108-110	0.74^{a}	968	321 (4.45)	(C ₂₀ H ₂₃ NO ₂) C, H, N
5b	56	100-1	$0.74^{\rm a}$	968	322 (4.46)	(C ₂₁ H ₂₅ NO ₂) C, H, N
5c	75	59-61	0.75^{a}	964	322 (4.47)	(C ₂₂ H ₂₇ NO ₂) C, H, N
5d	49	74-8	0.76^{a}	966	321 (4.42)	(C ₂₃ H ₂₉ NO ₂) C, H, N
5e	55	45-8	0.77^{a}	971	378 (4.42)	(C ₂₁ H ₂₄ N ₂ O ₄) C, H, N
5f	38	85-7	0.77^{a}	967	379 (4.41)	(C ₂₂ H ₂₆ N ₂ O ₄) C, H, N
5g	56	69-71	0.76 ^a	970	379 (4.42)	(C ₂₃ H ₂₈ N ₂ O ₄) C, H, N

^a CH₃OH: CHCl₃ (3:1).

^b CH₃OH: CHCl₃ (1:5).

Table 2

¹H NMR data of compounds **3a-3n** and **5a-5g**



Compd	t_1	t_2	t ₃	t ₄	k ₁	k ₂	k ₃	k ₄	k_5	q_1	q_2	d	Ar, m
													x-
3a	4.13	2.79	2.52	_	-	_	_	1.61	1.45	_	_	_	6.87–7.50
3h	4.12	2.79	2.96	_	_	_	_	_	_	1.64	1.29	0.92	6.87–7.49
5a	4.13	2.81	2.59	3.74	-	_	_	_	_	_	_	_	6.87–7.50
3b	4.00	2.10	2.47	_	2.00	_	_	1.60	1.45	_	_	_	6.86–7.49
3e	4.03	2.45	2.53	_	2.04	_	_	1.60	1.44	_	_	_	6.94-8.20
3i	4.00	2.53	2.96	_	2.00	-	-	-	-	1.66	1.29	0.93	6.86–7.48
31	4.00	2.47	2.93	_	1.97	_	_	_	_	1.61	1.25	0.95	6.90-8.20
5b	4.00	2.47	2.53	3.72	1.98	-	-	-	-	-	-	-	6.86–7.49
5e	4.06	2.49	2.53	3.76	1.99	-	-	-	-	-	-	-	6.90-8.20
3c	3.99	2.39	2.42	_	2.00	1.88	-	1.62	1.44	-	-	-	6.86–7.50
3f	4.04	2.39	2.42	_	2.00	1.87	-	1.61	1.46	-	-	-	6.83-8.21
3j	3.98	2.42	2.96	_	1.97	1.77	_	_	_	1.65	1.32	0.93	6.85–7.49
3m	4.00	2.40	2.91	_	1.94	1.78	-	-	-	1.66	1.25	0.92	6.89-8.20
5c	4.00	2.42	2.49	3.76	1.96	1.83	-	-	-	-	-	-	6.86–7.50
5f	4.02	2.48	2.50	3.75	1.99	1.82	-	-	-	-	-	-	6.84-8.29
3d	3.97	2.38	2.29	_	1.90	1.77	1.46	1.57	1.43	-	-	-	6.47–7.50
3g	4.03	2.45	2.90	_	1.87	1.79	1.45	1.56	1.43	-	-	-	6.68-8.50
3k	3.96	2.34	2.90	_	1.90	1.80	1.46	1.28	-	1.59	1.28	0.91	6.85-7.72
3n	3.98	2.32	2.91	_	1.88	1.84	1.46	1.22	_	1.59	1.30	0.92	6.88-8.21
5d	3.95	2.43	2.35	3.71	1.80	1.80	1.52	_	_	_	_	_	6.80–7.49
5g	3.99	2.47	2.35	3.73	1.84	1.84	1.54	_	_	_	_	_	6.89–8.20

Table 3 ¹³C NMR data of **3a-3g**

Carbon atom				Compd			
	3a	3b	3c	3d	3e	3f	3g
C-1	137.56	137.58	137.60	137.70	144.29	144.36	144.13
C-2,6	127.09	126.40	127.00	127.47	128.36	128.39	128.28
C-3,5	128.53	128.52	128.53	128.62	124.08	124.11	123.98
C-4	127.58	127.07	127.00	127.15	146.37	146.41	146.13
C-α	126.15	126.13	126.14	126.23	123.93	123.95	123.98
С-α'	127.70	127.57	127.59	127.68	132.93	133.00	132.74
C-1'	130.08	129.91	129.94	129.95	128.81	128.80	128.84
C-2',6'	128.13	128.14	128.20	128.19	126.42	126.46	126.34
C-3',5'	114.76	114.67	114.70	114.89	114.87	114.89	114.68
C-4'	158.47	158.66	158.66	158.87	159.25	159.80	159.35
C-I	66.09	66.62	66.06	67.88	68.80	67.87	67.78
C-II	57.96	56.01	56.00	59.40	55.73	58.90	57.00
C-III	_	26.92	27.20	26.66	26.50	27.30	28.66
C-IV	_	-	23.00	24.18	-	23.28	23.55
C-V	_	-	-	29.17	-	-	28.60
C-VI,X	55.12	54.70	54.58	54.63	54.45	54.49	52.87
C-VII,IX	26.04	26.06	25.91	25.93	25.60	25.76	23.62
C-VIII	24.28	24.52	24.46	24.44	24.20	24.30	23.55

lene bridge of the stilbene skeletone (Table 1). In the ¹H NMR spectra of **3a-3n** and **5a-5g** there are two triplets of O-CH₂ (t₁) and N-CH₂ (t₂) protons of N-alkoxy chain linking (E)-stilbene and piperidine (4-methylpiperidine, morpholine) rings. The values of the chemical shifts of the signals of these protons were established in the range 3.95-4.06 δ and 2.39-2.81 δ , respectively (Scheme 2, Table 2). The ¹H NMR spectra of **3a-3g**

show two quintets (k_4 and k_5) of protons of methylene groups of piperidine ring (Scheme 2, Table 2). The values of the chemical shifts of the signals of these protons fall in the range 1.56-1.62 δ and 1.43-1.46 δ , respectively. The presence of these signals in the ¹H NMR spectra of **3a-3g** indicates the occurrence of the piperidine ring in the molecules of these compounds. The ¹H NMR spectra of **3h-3n** show a dublet of

¹³ C NMR data of 3h-3n	Table 4		
	¹³ C NMR	data	of 3h-3n

Carbon atom	Compd									
	3 h	3i	3ј	3k	31	3m	3n			
C-1	137.58	137.47	137.51	137.61	144.23	144.27	144.28			
C-2,6	127.15	127.02	127.03	127.11	128.36	128.35	128.35			
C-3,5	128.59	128.47	128.49	128.58	124.11	124.08	124.08			
C-4	127.64	127.52	127.55	127.63	146.34	146.27	146.27			
C-α	126.19	126.07	126.09	126.17	129.81	123.84	123.82			
C-α'	127.82	128.64	127.35	128.58	132.82	132.91	132.92			
C-1'	130.09	129.84	129.84	129.89	129.03	130.84	128.65			
C-2',6'	128.14	128.08	128.11	128.20	128.77	128.74	126.40			
C-3',5'	114.89	114.59	114.57	114.60	115.06	114.78	114.77			
C-4'	158.47	158.52	158.57	158.76	159.19	159.68	159.74			
C-I	66.16	66.43	67.71	67.79	68.81	67.71	67.86			
C-II	57.49	55.58	58.93	58.97	55.54	58.54	59.03			
C-III	_	26.85	27.41	26.74	26.74	27.29	26.24			
C-IV	_	_	23.45	24.16	-	23.47	24.15			
C-V	_	_	_	29.14	_	_	29.10			
C-VI,X	54.48	54.09	54.90	54.02	54.07	53.93	54.10			
C-VII,IX	34.19	34.19	34.13	34.22	34.17	34.29	34.32			
C-VIII	30.52	30.79	30.79	30.76	30.78	30.71	30.81			
C-XI	21.86	21.89	21.87	21.87	21.89	21.83	21.89			

Table 5

¹³C NMR data of 5a-5g

Contorn	- 4

Carbon atom				Compd			
	5a	5b	5c	5d	5e	5f	5g
C-1	137.64	137.53	137.52	137.46	144.27	144.13	144.30
C-2,6	127.25	126.49	126.45	126.32	128.37	128.27	128.33
C-3,5	128.66	128.53	128.53	128.48	124.00	124.01	124.10
C-4	127.72	127.11	127.10	127.03	146.40	146.18	146.40
C-α	126.27	126.14	126.13	126.04	124.09	123.85	123.96
C-α'	127.27	127.59	127.59	127.54	132.90	132.77	132.94
C-1'	130.38	130.01	129.96	129.80	128.89	128.69	128.80
C-2',6'	128.16	128.11	128.11	128.06	128.37	126.34	126.44
C-3',5'	114.84	114.64	114.59	114.51	114.87	114.71	114.84
C-4'	158.47	158.57	158.55	158.57	159.70	159.47	159.78
C-I	66.69	67.02	67.66	67.74	68.81	67.66	67.86
C-II	57.42	55.58	58.59	58.94	55.41	58.51	58.88
C-III	-	26.56	27.23	26.31	26.34	27.13	26.18
C-IV	_	_	23.03	24.02	_	22.95	23.91
C-V	_	-	-	29.18	-	-	29.04
C-VI,X	54.10	53.80	53.64	53.75	53.70	53.58	53.72
C-VII,IX	65.94	66.19	66.81	66.92	66.92	66.73	66.89

$$t_1 t_2$$

O-CH₂-CH₂-N $<$

$$\mathbf{t}_1 \quad \mathbf{k}_1 \quad \mathbf{t}_2$$

O-CH2-CH2-CH2-N.



Scheme 2. The numbering of protons of N-alkoxypiperidine, N-alkoxy-4methylpiperidine and N-alkoxymorpholine groups of 3a-3n; 5a-5g.

protons of a methyl group in the range 0.91-0.95 δ . The presence of this signal in the ¹H NMR spectra of **3h-3n** proves the occurrence of the 4-methylpiperidine ring in the molecules of these compounds. The ¹H NMR spectra of **5a-5g** in the range of 3.71-3.75 δ show a triplet of proton signals of O-CH₂

groups of morpholine ring. These signals indicate the occurrence of the morpholine ring in the investigated molecules (Scheme 2, Table 2).

The presence of piperidine (4-methylpiperidine, morpholine) ring in the molecules of 3a-3n and 5a-5g is also indicated in the ¹³C NMR spectra of these compounds (carbons VI, X; VII, IX and VIII - Tables 3-5).

3. Biology

New (E)-4-[piperidino-]N-alkoxystilbenes 3a-3g, (E)-4-[4]methylpiperidino-]N-alkoxystilbenes 3h-3n and (E)-4-[morpholino-]N-alkoxystilbenes 5a-5g were tested in vitro in order to evaluate their antimicrobial effect. The potential antimicrobial activity of compounds 3a-3n and 5a-5g was estimated in vitro by determining the MIC against a wide spectrum of microorganism: Gram-positive cocci (Staphylococcus aureous



Scheme 3. The numbering of carbons of (E)-stilbene, N-alkoxypiperidine, Nalkoxy-4-methylpiperidine and N-alkoxymorpholine groups of 3a-3n; 5a-5g.

209P FDA, Streptococcus faecalis ATCC 8040), aerobic bacilli (Bacillus subtilis ATCC 1633), Gram-negative rods (Escherichia coli PZH 026 B6, Klebsiella pneumoniae 231, Pseudomonas aeruginosa SR1), veasts (Candida albicans PCM 1409 PZH) dermatophytes (Microsporum Gypseum K1), moulds (Aspergillus fumigatus C1) (Scheme 3).

4. Results

Table 6 lists the antimicrobial susceptibility results against Gram-positive cocci, aerobic bacilli, Gram-negative rods, yeast, dermatophytes and moulds. (E)-4-[piperidino-N-ethoxy-]stilbene **3a**, (E)-4-[piperidino-N-propioxy-]stilbene **3b**, (E)-4-[piperidino-N-butoxy-]stilbene 3c, (E)-4-[piperidino-N-butoxy-] 4'-nitrostilbene **3f**, (E)-4-[piperidino-N-pentoxy-]4'-nitrostilbene 3g, (E)-4-[4`-methylpiperidino-N-ethoxy-]stilbene 3h, (E)-4-[4'-methylpiperidino-N-pentoxy-]stilbene 3k, (E)-4-[4'methylpiperidino-N-propioxy-]4`-nitrostilbene 31, (E)-4-[4`methylpiperidino-N-butoxy-]4'-nitrostilbene 3m and (E)-4-[4'methylpiperidino-N-pentoxy-]4'-nitrostilbene **3n** tested in this study exhibited considerable antimicrobial activity against Gram-positive cocci Staphylococcus aureus. (E)-4-[4'-methylpiperidino-N-ethoxy-]stilbene 3h, (E)-4-[4'-methylpiperidino-Npentoxy-]4'-nitrostilbene 3n exhibited also a high activity against Streptococcus feacalis and aerobic bacilli- Bacillus subtilis. (E)-4-[4'-methylpiperidino-N-pentoxy-]stilbene 3k and (E)-4-[4'-methylpiperidino-N-butoxy-]4'-nitrostilbene 3m showed considerable antibacterial activity agaist Bacilus subtilis. Compounds 3d, 3e, were endowed with a medium activity against cocci Staphylococcus aureous, Streptococcus faecalis as well as Bacillus subtilis. A medium activity against, Streptococcus faecalis and Bacillus subtilis was exhibited also by 3a, 3e, 3f, 3g and 3l. Coumpound 3a was also endowed with a medium activity against Gram-negative rods- Escherichia coli. (E)-4-[morpholino-N-butoxy-]stilbene 5c, (E)-4-[morpholino-N-pentoxy-]stilbene 5d, (E)-4-[morpholino-N-propioxy-]4`-nitrostilbene 5e, (E)-4-[morpholino-N-butoxy-]4'-nitrostilbene 5f and (E)-4-[morpholino-N-pentoxy-]4'-nitrostilbene 5g were en-

Table 6 Antimicrobial activity of 3a-3n and 5a-5g

Compd	Minimal inhibitory concentration (MIC) µg/cm ³ *									
-	SA	SF	BS	EC	KP	PA	CA	MG	AF	
3a	10	100	100	100	_	_	50	_	50	
3b	5	_	_	_	_	_	_	_	100	
3c	5	100	100	_	_	_	_	_	100	
3d	100	100	100	_	_	_	_	_	50	
3e	100	100	100	-	_	_	_	_	_	
3f	7.5	100	100	_	_	_	_	_	_	
3g	10	100	100	_	_	_	_	_	50	
3h	7.5	7.5	5	-	_	_	50	_	50	
3i	_	_	_	_	_	_	50	_	50	
3j	_	_	_	-	_	_	50	_	50	
3k	7.5	100	5	_	_	_	50	_	10	
31	7.5	100	100	-	_	_	_	_	100	
3m	10	100	10	-	_	_	100	_	10	
3n	10	7.5	10	_	_	_	_	_	50	
5a	_	_	_	-	_	_	_	_	_	
5b	_	_	_	_	_	_	_	_	_	
5c	100	_	_	-	_	_	_	_	50	
5d	100	_	_	_	_	_	_	_	50	
5e	100	_	_	-	_	_	_	_	50	
5f	100	_	_	_	_	_	_	_	50	
5g	100	_	_	_	_	_	_	_	50	
А	5.0	5.0	5.0	5.0	50.0	50.0	_	_	_	
В	_	_	_	-	_	_	0.5	10.0	1.0	

SA: Staphylococcus aureus 209P FDA; CA: Candida albicans PCM 1409 PZH; SF: Streptococcus faecalis ATCC 8040; MG: Microsporum gypseum K1; BS: Bacillus subtilis ATCC 1633; AF: Aspergillus fumigatus C1; EC: Escherichia coli PZH 026 B6; A: chloroamphenicole; KP: Klebsiella pneumoniae 231; B: amphoterricin B; PA- Pseudomonas aeruginosa SP1; - Not tested because MIC value is higher than 100 μ g/cm³.

dowed only with medium antimicrobial activity against Grampositive cocci- Staphylococcus aureus and moulds- Aspergillus *fumigatus*. All the compounds tested in this study displayed not significant effect against the Gram-negative rods Klebsiella pneumoniae and Pseudomonas aeruginosa, in general, MIC > 100g/cm^3 .

The newly synthesised compounds (E)-[4'-metylpiperidino-N-pentoxy-]stilbene 3k and (E)-4-[4'-methylpiperidino-N-butoxy-]stilbene 3m showed good antimicrobial properties against moulds- Aspergillus fumigatus. The compounds 3a, 3b, 3c, 3d, 3g, 3h, 3i, 3j, 3l and 3n as well as 5a-5g showed a medium activity against Aspergillus fumigatus. Additionally, a medium activity against yeast- Candida albicans was exhibited by compounds 3a, 3 h, 3i, 3j, 3k and 3m. It is noteworthy that **3a** displayed a wide spectrum of antimicrobial activity against 6 strains (SA, SF, BS, EC, CA, CF - Table 6) as well as 3h, 3k, and 3m against 5 strains (SA, SF, BS, EC, AF-Table 6).

5. Discussion

This work presents a novel class of potent, wide-spectrum antimicrobial compounds. The results are worth noting because in recent years increasing rates of antimicrobial resistance among community and nosocomial pathogenes has severely limited the therapeutic options for treating infections caused by such organisms. From among the compounds tested 3h, 3k and 3m proved to be the most effective.

As regards the structure-activity relationship of the substituted derivatives of (E)-stilbenol-4 and (E)-4'-nitrostilbenol-4, the introduction of 4-alkoxy-N-piperidine group and 4-alkoxy-N-4'-methylpiperidine group as substitutent in the para position of the (E)-stilbene [(E)-4'-nitrostilbene] moiety noticeably enhanced the antibacterial activity displayed by the unsubstituted stilbenols [15]. The introduction of 4-alkoxy-N-morpholine group as substituent in the para position of the (E)-stilbene [(E)-4'-nitrostilbene] moiety did not noticeably enhance the antimicrobial activity displayed by the unsubstituted stilbenols [15]. (E)-4-(piperidino-N-propioxy-]stilbene **3b** and (E)-[piperidino-N-butoxy-]stilbene 3c displayed the antibacterial activity against Staphylococcus aureus comparative to that of the reference drug chloramphenicol (Table 6). (E)-4-[4'-methylpiperidino-N-pentoxy-]stilbene 3k and (E)-4-[4'-methylpiperidino-Nethoxy-]stilbene **3h** displayed the same antibacterial activity against Bacillus subtilis similar to that of the reference drug chloramphenicol (Table 6).

In conclusion, taking into account that **3a**, **3h**, **3k** and **3m** are wide spectrum antimicrobial substances, it can be concluded that they are promising new agents for treatment of microbial infections.

6. Experimental

6.1. Chemistry

The purity of all described compounds was checked by m. p.'s TLC and elemental analysis. Melting points (uncorrected) were determined on Böetius microscope hot stage. Rf values refer to TLC silica gel F254 TLC plates (Merck) developed with CHCl₃ : MeOH (5:1) and observed under UV light (λ =254 and 366 nm). UV/VIS spectra were recorded with a Specord UV/ VIS spectrophotometer in CHCl₃. IR spectra were recorded with a FTIR Bruker IFS-113 V spectrophotometer in KBr pellets. The ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) spectra were recorded on a Varian Mercury Spectrometer operating at 300.07 MHz (proton) or 75.46 mHz (carbon). The data were obtained from CDCl₃ solutions. The chemical shifts were referenced to tetramethylsilane. Chemical shifts are given in the δ scale (ppm) and coupling constants in Hz. ¹H NMR (300.07) spectra were recorded with spectral width 9 KHz, acquisition time 2.0 s, pulse width 6 µs and double precision acquisition time. ¹³C NMR (75.460 MHz) spectra were recorded with spectral width 18.76 KHz, acquisition time 1.0 s, recycle delay 1.0 s and pulse width 15 μ s. Heteronuclear 2D¹³C NMR – ¹H NMR chemical shift correlation experiments were carried out using HETCOR spectra. The spectra were acquired with 2K data points, 256 increments and spectral width 19.63 KHz for ¹³C and 4.97 KHz for ¹H. Elemental analyses were performed with a Vector Euro EA 3000 Analyzer.

(E)-4-bromoethoxystilbene **1a** [31], (E)-4-bromopropioxystilbene **1b** [31], (E)-bromobutoxystilbene **1c** [25], (E)-4-bromopentoxystilbene **1d** [25], (E)-4'-nitro-4-bromopropioxystilbene **1e** [25], (E)-4'-nitro-4-bromobutoxystilbene **1f** [25], (E)-4'-nitro-4-bromopentoxystilbene **1g** [25], were obtained according to literature. Piperidine, 4-methylpiperidine and morpholine were purchased from Aldrich.

General procedure for the synthesis of compounds 3a-3n and 5a-5g.

To a solution of the corresponding (E)-4-bromoalkoxystilbene **1a-1g** (1 mmol) in DMF (70 ml) triethylamine (8 mmol) and piperidine (4'-methylpiperidine, or morpholine) (1 mmol) were added on constant stirring. After stirring for 10-25 h (i.e. **3a**, **3b**, **5a**, **5b** – 25 h; **3c**, **3f** – 24 h; **3g**, **5e**, **5f**, **5g** – 21 h, **3d**, **3i**, **3e**, **5c**, **5d** – 18 h; **3h**, **3k** – 15 h; **3m**, **3j** – 11 h; **3h**, **3l** – 10 h) at room temperature, distilled water (60 ml) was added to the reaction mixture. The precipitated solids of **3a-3n** and **5a**-**5g** were then filtered off, dried and recrystallized from DMF: H₂O (3:2).

6.2. Microbiology

6.2.2. Determination of minimum inhibitory concentration (MIC)

The microorganisms used were supplied by – National Institute of Hygiene in Warsaw (*Staphylococcus aureus 209P FDA*, *Escherichia coli PZH 026 B6*, *Candida albicans PCM 1409 PZH*), American type Culture Collection (*Streptococcus faecalis ATCC 8040*, *Bacillus subtilis ATCC* 1633), Department of Microbiology, Poznań University of Medical Sciences (*Klebsiella pneumoniae 231*, *Pseudomonas aeruginosa SP1*), Department of Medical Mycology, Poznań University of Medical Sciences (*Aspergillus fumigatus C1*, *Microsporum gypseum K1*).

The compounds were dissolved using DMSO (Serva); to form solutions of the concentration 1000 μ g/cm³. A series of dilutions with concentrations ranging form 10 to 1000 μ g/cm³ were prepared for each compound.

The MIC values of the compounds were determined, with reference to standard microorganisms, by introducing 1 cm³ of the corresponding solutions at various concentrations into a series of tubes (each 12×100 mm), 0.1 cm³ of a standarized 1:1000 diluted suspension of a microorganism was added. The MIC values were determined after 18 hours of incubation at 37°C. Penassay Broth (Difco) was used as the test medium for bacteria; in all assays both bacterial culture sterility and standard bacterial growth were checked. Sabouraud dextrise broth (Difco) was used as a test medium for fungi; MIC values were determined after 3-7 days of incubation at 25°C. In all assay both fungi culture sterility and standard fungi growth were checked. The MIC values determined were compared with those of the standards, chloramphenicol (the reference bactericidal drug) and amphotericine B (the reference antifungal drug). The results are listed in Table 6.

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