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### Synthesis of N-acylated 7-amino-2,6,7-trideoxy-D-erythroheptopyranosides from methyl $\alpha$ -D-mannoside

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Abstract—Hexopyranoside methyl  $\alpha$ -D-mannoside (8) was homologated to yield 7-(acylamino)-2,6,7-trideoxy-heptopyranosides 19– 26. A crucial reaction step is the radical cleavage of benzylidene derivative 10 to obtain bromide 11. Since nucleophilic substitution of 11 with KCN provided the bicyclic nitrile 13 instead of nitrile 14, ketone 11 was protected as the dimethyl acetal 15. Nucleophilic substitution of 15 with KCN, subsequent hydrogenation with H<sub>2</sub>/Raney Ni and acylation with various carboxylic acid derivatives yielded 7-(acylamino)heptopyranosides 19-22. © 2005 Elsevier Ltd. All rights reserved.

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### 1. Introduction

The three-dimensional arrangement of pharmacophoric elements is crucial for the interaction of drugs with their targets (enzymes, receptors, DNA, etc.). Monosaccharides represent excellent scaffolds for the introduction of pharmacophoric elements in a defined spatial arrangement, since all positions of a pyranoside ring system are substituted with functional groups. These substituents allow the stereoselective introduction of the desired pharmacophoric groups with defined threedimensional orientation.

We are interested in the development of novel  $\kappa$  receptor agonists and  $\boldsymbol{\sigma}$  receptor ligands. Several of the described high affinity  $\kappa$  receptor agonists and  $\sigma$  receptor ligands comprise (3,4-dichlorophenyl)acetamide substructure as the pharmacophoric element. In Figure 1 the prototypical  $\kappa$  receptor agonist, U-50488 (1a), with (3,4-dichlorophenyl)acetamide substructure is shown.<sup>1</sup> Changing the stereochemistry from the *trans*-(1S,2S)configuration (1a) into the cis-(1R,2S)-configuration

(1b) leads to a highly active  $\sigma$  receptor ligand.<sup>1</sup> (3,4-Dichlorophenyl)acetamide 2a is also a very potent  $\kappa$ receptor agonist with analgesic activity.<sup>2</sup> Formal reduction of amide 2a affords tertiary amine 2b with high affinity toward  $\sigma$  receptors.<sup>3</sup>



Figure 1. Lead compounds with  $\kappa$  and  $\sigma$  affinity.

The focus of this manuscript is the synthesis of a carbohydrate-based scaffold for drug design. In particular, we are interested in pyranosides bearing an aminoethyl moiety in position 5 (compare compound 3 in Figure 2). It should be possible to introduce amino substituents (e.g., pyrrolidin-1-yl) in position 3 and/or 4 (compound 4) to mimic  $\kappa$  and  $\sigma$  ligands 1 and 2. Moreover, it is planned to connect aminoethyl side chain of 3 to the

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Figure 2. Projected pharmacologically active compounds from 7-acylamino-substituted heptopyranoses.

position 1, 3, or 4 to obtain bicyclic systems 5–7. After introduction of amino substituents (e.g., pyrrolidin-1yl) the bicyclic systems 5–7 represent conformationally constrained analogues of the lead compounds 1 and 2. Herein, we describe our studies directed to homologization of the monosaccharide D-mannose to provide 7acylamino-substituted heptopyranose derivatives substituted with functional groups in pyran ring. These substituents will be used for the regio- and stereoselective introduction of further pharmacophoric elements (see Fig. 2).

#### 2. Results and discussion

The synthesis started with methyl  $\alpha$ -D-mannopyranoside (8), which was reacted with 2 equiv of benzaldehyde dimethyl acetal in DMF.<sup>5</sup> Instead of the procedure of Evans, who used a special apparatus to remove DMF and methanol under reduced pressure at 60 °C,<sup>5</sup> the acetalization of 8 was performed in a rotary evaporator at a water bath temperature of 60-70 °C under reduced pressure (about 300 mbar). However, the purification of acetal product 9 turned out to be difficult. In particular, the careful removal of DMF and water was crucial for the further transformation with *n*-butyllithium. Only recrystallization of the crude product from 2-propanol yielded 51% of the pure product 9. In order to improve the yield of 9, various solvents were investigated to replace DMF. Finally, the reaction of 8 with benzaldehyde dimethyl acetal was performed using acetonitrile as solvent at a temperature of 60 °C and at a reduced pressure of 600 mbar in a rotary evaporator to afford the acetal product 9 (mixture of diastereomers) in very high yield

(97%). Employing this product in the next reaction with *n*-butyllithium at -40 °C afforded ketone **10** in 71% yield (Scheme 1).

It was planned to transform hexopyranoside **10** into a heptopyranoside system. For this purpose a leaving group should be introduced in position 6 of hexopyranoside system allowing the nucleophilic substitution with cyanide to incorporate the additional carbon atom. Hence, benzylidene derivative **10** was reacted with *N*-bromosuccinimide (NBS) and azobisisobutyronitrile (AIBN)<sup>7,8</sup> in a free-radical substitution reaction to provide bromobenzoate **11** in 54% yield. The desired methyl  $\alpha$ -D-glycoside **11** was accompanied by small amounts (8%) of the elimination product **12**, which was separated by flash chromatography.

The reaction of bromoketone 11 with KCN in DMSO resulted in a novel product without bromo substituent. Surprisingly, instead of the expected substitution product 14, the bicyclic compound 13 had been formed. The structure of 13 was unambiguously proven by NMR, IR spectroscopy, and MS. The formation of 13 may be explained by nucleophilic attack of NC<sup>-</sup> at carbonyl moiety to yield a cyano-alcoholate, which reacted intramolecularily with bromomethyl group to yield the bicyclic cyanoether 13 instead of the simple substitution product 14 (see Scheme 2).

In order to avoid the cyclization of bromoketone 11, the carbonyl moiety of 11 was transformed into a dimethyl acetal ( $\rightarrow$ 15) with methanol and trimethyl orthoformate. This transformation was catalyzed by *p*toluenesulfonic acid and therefore, led to epimerization of the anomeric center. The  $\alpha$  and  $\beta$  anomers 15 $\alpha$  and 15 $\beta$  were formed in a ratio of 1:1 (total yield 88%) and could be separated by flash chromatography. In a sepa-



Scheme 1. Reagents and conditions: (a) PhCH(OCH<sub>3</sub>)<sub>2</sub>, TosOH, CH<sub>3</sub>CN, 60 °C, 600 mbar, 97%. (b) *n*-BuLi, THF, -40 °C, 71%. (c) NBS, AIBN, BaCO<sub>3</sub>, CCl<sub>4</sub>, reflux, 54%; yield of 12 8%. (d) KCN, DMF, 80 °C, 26%.



Scheme 2.

rate experiment, it was shown that  $\alpha,\beta$ -unsaturated ketone 12, the side product during the radical bromination of the benzylidene compound 10, also reacted with methanol and trimethyl orthoformate to give dimethyl acetals 15 $\alpha$  and 15 $\beta$ . Therefore, a mixture of 11 and 12 could be used for the synthesis of dimethyl acetals 15 $\alpha$ and 15 $\beta$  (Scheme 3).

The nucleophilic substitution of the separated anomeric bromo acetals  $15\alpha$  and  $15\beta$  with KCN proceeded to afford the anomeric cyano acetals  $16\alpha$  and  $16\beta$ , respectively. The cleavage of benzoates  $16\alpha$  and  $16\beta$ was performed with NaOH in methanol to obtain anomeric hydroxynitriles  $17\alpha$  and  $17\beta$ , respectively. However, the <sup>1</sup>H NMR spectra of both hydrolysis products  $17\alpha$  and  $17\beta$  showed significant amounts of the corresponding anomer. Obviously anomerization takes place during saponification of ester moiety. Therefore, the synthesis of substantial amounts of hydroxynitriles  $17\alpha$  and  $17\beta$  was usually performed with anomeric mixtures.

Catalytic hydrogenation (4.1 bar) of nitriles  $17\alpha/17\beta$ using Raney nickel as catalyst<sup>9,10</sup> provided the primary amines  $18\alpha/18\beta$ . Since hydrogenation of nitriles  $17\alpha/$  $17\beta$  with Raney nickel requires basic reaction conditions (4:1 methanol:NaOH (5 M)), the one-pot transformation of cyanobenzoates  $16\alpha/16\beta$  into the primary amines  $18\alpha/18\beta$  was also investigated and yielded 97% of the



Scheme 3. Reagents and conditions: (a) CH<sub>3</sub>OH, CH(OCH<sub>3</sub>)<sub>3</sub>, TosOH, reflux, 88%. (b) KCN, DMSO, 70 °C, 16α 94%, 16β 67%. (c) NaOH, CH<sub>3</sub>OH, 20 °C, 17α 76%, 17β 80%. (d) H<sub>2</sub>, Raney Ni, NaOH, CH<sub>3</sub>OH, 20 °C, 97% (anomeric mixture). (e) R–COX, NEt<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 20 °C (19–21, 23); RCO<sub>2</sub>H, CDI, CH<sub>2</sub>Cl<sub>2</sub>, 20 °C, (22); yields 19: 44%; 20: 60%; 21: -; 22: 52%; 23: 25%.

primary amines  $18\alpha/18\beta$  over two reaction steps. Because of the high degree of purity, the primary amines  $18\alpha/18\beta$  were used for the next reaction steps without further purification.

In the next step, the primary amines  $18\alpha/18\beta$  were acylated with various acylating reagents to yield amides and carbamates 19-24 consisting of anomeric mixtures. Interestingly, after acylation of the primary amines  $18\alpha/$ 18 $\beta$  with benzoyl chloride and triethylamine, ketone 23, resulting from additional hydrolysis of acetal moiety, was isolated. The corresponding acetal 21 was only detected by <sup>1</sup>H NMR spectroscopy. With respect to developing novel  $\kappa$  receptor agonists and  $\sigma$  receptor ligands, the introduction of (3,4-dichlorophenyl)acetyl residue was of particular interest. Reaction of the primary amines  $18\alpha/18\beta$  with (3,4-dichlorophenyl)acetyl chloride in a two-phase system (NaOH, methanol, dichloromethane) provided amides  $22\alpha/22\beta$ . However, the yield did not exceed 41% since the acylation was accompanied by substantial hydrolysis of the acetal moiety to afford the corresponding ketones  $24\alpha/24\beta$ . In an alternative synthesis the primary amines  $18\alpha/18\beta$  were acylated with (3,4-dichlorophenyl)acetic acid in the presence of 1,1'-carbonyldiimidazole to obtain amides  $24\alpha/24\beta$  in 50% yield calculated over three steps starting with cyanobenzoates  $16\alpha/16\beta$ . Hydrolysis of dimethyl acetal substructure was not observed with these reaction conditions (Scheme 3).

All attempts to synthesize morphan analogues 27 by intramolecular N/O-acetal formation of amides and carbamate 19–22 $\alpha/\beta$  displaying various nucleophilicity of their nitrogen atoms failed. After heating of (dichlorophenyl)acetamides 22 $\alpha/\beta$  with *p*-toluenesulfonic acid in 1,2-dichloroethane, the  $\alpha/\beta$ -unsaturated ketone 26 was isolated in 53% yield. During this transformation, the intermediate ketones 24 $\alpha/\beta$  were detected by thin-layer chromatography. Performing the same reaction at 0 °C for 4 h led to selective hydrolysis of ketone acetal ( $\rightarrow 24\alpha/\beta$ ), elimination of methanol did not take place. Heating of ketone acetals 22 $\alpha/\beta$  as well as ketones 24 $\alpha/\beta$  with BF<sub>3</sub>·OEt<sub>2</sub> or HCl as catalysts also led to the  $\alpha/\beta$ -unsaturated ketone 26 (Scheme 4).

The same reactivity was observed for benzamides  $21\alpha/\beta$ , which provided ketones  $23\alpha/\beta$  and the  $\alpha/\beta$ -unsaturated ketone 25, depending on the reaction conditions.



Scheme 4. Reagents and conditions: (a) TosOH, 1,2-dichloroethane, reflux, 26 53%. (b) TosOH,  $CH_2Cl_2$ , 0 °C,  $24\alpha/\beta$  80%. (c) TosOH,  $CH_2Cl_2$ , reflux, 25 34%.

### 3. Conclusions

Starting from methyl  $\alpha$ -D-mannopyranoside (8) the anomeric nitriles  $16\alpha/\beta$  were prepared in five steps and 22– 31% overall yields. Reduction of the cyano moiety and acylation of the resulting primary amines  $17\alpha/\beta$  with various carboxylic acid derivatives led to 7-(acylamino)heptopyranosides 19–26. Now, the development of the (acylaminoethyl) substituted pyranosides 19–26 into novel  $\kappa$  and  $\sigma$  receptor ligands requires bridging of the pyran system and introduction of amino moieties.

#### 4. Experimental

### 4.1. Chemistry, general

Unless otherwise noted, moisture-sensitive reactions were conducted under dry nitrogen. Thin-layer chromatography (TLC) was carried out on Silica Gel 60 F<sub>254</sub> plates (E. Merck). Flash chromatography (FC) was performed with Silica Gel 60, 40-63 µm (E. Merck).<sup>11</sup> Parentheses include the following: (1) diameter of the column (cm), (2) eluent, (3) fraction size (mL), and (4)  $R_{\rm f}$ . Melting points (mp) were determined on an SMP2 (Stuart Scientific) melting point apparatus and are uncorrected. Optical rotations were measured on a Perkin-Elmer polarimeter model 241 using a 1.0-dm tube with a solute concentration c (g/100 mL) at a temperature of 20 °C. Elemental analyses (CHN) were determined on a VarioEL (Elementaranalysensysteme GmbH) and a Perkin-Elmer Elemental Analyzer model 240. Mass spectra were recorded on MAT 44S, MAT 312, MAT 8200, and TSQ 7000 instruments (Finnigan)

in either the electron-impact (EI), chemical ionization (CI), electrospray ionization (LC ESI/ESI direct) or atmospheric pressure chemical ionization (APCI/FIA APCI/LC APCI) mode. IR spectra were recorded on a Perkin–Elmer FTIR spectrophotometer 1605 using the Universal ATR Sampling Accessory (ATR FTIR spectra). <sup>1</sup>H NMR spectra (300 MHz) and <sup>13</sup>C NMR spectra (75 MHz) were determined on a Varian Unity 300 FT NMR spectrometer (Varian);  $\delta$  values (in ppm) are given relative to the internal standard tetramethylsilane; the coupling constants *J* (in Hz) are given with 0.5-Hz resolution.

## 4.2. Preparation of methyl 2,3:4,6-di-O-benzylidene- $\alpha$ -D-mannopyranoside (9)<sup>4</sup>

To a solution of methyl  $\alpha$ -D-mannopyranoside (8, 5.0 g, 25.7 mmol) in acetonitrile (30 mL) p-toluenesulfonic acid (0.11 g, 0.6 mmol) and benzaldehyde dimethyl acetal (9.1 mL, 60.4 mmol) were added. The reaction was carried out in a rotary evaporator at 60 °C water bath temperature and a reduced pressure of 600 mbar. After 6 h, the residual mixture was concentrated in vacuo until a colorless solid started to precipitate. For complete precipitation of the product, the mixture was added to NaHCO<sub>3</sub>-containing ice water (3.1 g in 100 mL) under stirring. After filtration of the solvent-water mixture, the colorless solid was dried in vacuo at 60 °C for 24 h. Recrystallization from 2-propanol gave colorless, spicular crystals of 9: Colorless spicules (2-propanol): mp 179 °C [Ref. 4 178 °C]/176–177 °C (after recrystallization); yield 9.3 g (97%)/4.9 g (51%, after recrystallization); IR (neat):  $\tilde{v}$  1082 (vC–O), 736, 696 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  3.42 (s, 3×0.5H, OCH<sub>3</sub>),

3.44 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>), 3.72–3.94 (m, 3H, 4-H, 5-H, 6-H), 4.15 (d, *J* 5.8 Hz, 0.5H, 2-H), 4.30 (d, *J* 6.4 Hz, 0.5H, 2-H), 4.38–4.30 (m, 1H, 6-H), 4.48 (t, *J* 6.7 Hz, 0.5H, 3-H), 4.64 (dd, *J* 7.6/5.5 Hz, 0.5H, 3-H), 5.03 (s, 0.5H, 1-H), 5.09 (s, 0.5H, 1-H), 5.55 (s, 0.5H, PhCHO<sub>2</sub>), 5.65 (s, 0.5H, PhCHO<sub>2</sub>), 5.98 (s, 0.5H, PhCHO<sub>2</sub>), 6.30 (s, 0.5H, PhCHO<sub>2</sub>), 7.35–7.57 (m, 10H, arom. H).

### 4.3. Preparation of methyl 4,6-*O*-benzylidene-2-deoxy- $\alpha$ -D-erythro-hex-3-ulopyranoside (10)<sup>6</sup>

The bisacetal 9 (5.1 g, 19.3 mmol) was dissolved in THF (100 mL). At -40 °C n-butyllithium (2.4 M in n-hexane, 20 mL, 48.0 mmol), was added and the solution was stirred at -40 °C for 3.5 h. The color of the solution changed from yellow to red. The mixture was added to NH<sub>4</sub>Cl-containing ice water (12 g in 100 mL). Then, THF was evaporated at 30 °C water bath temperature without separation of the layers. The aq solution was cooled to 0 °C, and the yellow crystalline product was filtered off. Recrystallization from EtOH gave 10 as colorless crystals. Colorless, crystalline solid, mp 171-172 °C [Ref. 6 170-171 °C]; yield after recrystallization 2.6 g (71%); IR (neat):  $\tilde{v}$  1743 (vC=O), 1129 (vC-O), 908, 736, 650 cm<sup>-1</sup> (γCH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.66 (d, 1H, J 14.7 Hz, 2-H<sub>eq</sub>), 2.82 (dd, J 14.7/4.9 Hz 1H, 2-H<sub>ax</sub>), 3.37 (s, 3H, OCH<sub>3</sub>), 3.90 (t, J 10.1 Hz, 1H, 6-H<sub>ax</sub>), 4.14 (td, J 9.8/4.9 Hz, 1H, 5-H), 4.29 (d, J 9.8 Hz, 1H, 4-H), 4.37 (dd, J 10.1/4.9 Hz, 1H, 6-H<sub>ea</sub>), 5.13 (d, J 4.6 Hz, 1H, 1-H), 5.57 (s, 1H, PhCHO<sub>2</sub>), 7.33-7.35 (m, 3H, arom. H), 7.48-7.51 (m, 2H, arom. H).

### 4.4. Methyl 4-*O*-benzoyl-6-bromo-2,6-didedoxy-α-D*erythro*-hex-3-ulopyranoside (11) and 1,5-anhydro-4-*O*-benzoyl-6-bromo-2,6-dideoxy-D-*erythro*hex-1-en-3-ulose (12)

Ketone **10** (8.1 g, 30.4 mmol) was dissolved in CCl<sub>4</sub> (360 mL) and *N*-bromosuccinimide (8.5 g, 47.8 mmol), barium carbonate (13.8 g, 69.9 mmol) and AIBN (0.44 g, 2.7 mmol) were added. At first the mixture was carefully heated until the reaction started, then it was heated to reflux for 3.5 h. After cooling down the mixture was washed with satd aq NaHSO<sub>3</sub> (150 mL) and satd aq NaCl (150 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo. The residue was purified by FC (8 cm, 80:20 petroleum ether–EtAOc, fractions 50 mL).

**4.4.1. Data for 11.**  $R_{\rm f}$  0.31; colorless solid, mp 92 °C, yield 5.3 g (54%); IR (neat):  $\tilde{\nu}$  1728 (vC=O), 1599, 1492, 1451 (vC=C, arom.), 1247 (vO=C-O), 1109 (vC-O), 733, 706 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.71 (dd, *J* 14.0/0.9 Hz, 1H, 2-H<sub>eq</sub>), 2.93 (ddd, *J* 14.0/4.6/0.9 Hz, 1H, 2-H<sub>ax</sub>), 3.44 (s, 3H,

OCH<sub>3</sub>), 3.61 (dd, J 11.3/6.1 Hz, 1H, CH<sub>2</sub>Br), 3.71 (dd, J 11.3/2.4 Hz, 1H, CH<sub>2</sub>Br), 4.36 (dddd, J 9.8/6.1/2.4/ 0.6 Hz, 1H, 5-H), 5.21 (d, J 4.3 Hz, 1H, 1-H<sub>eq</sub>), 5.47 (dd, J 10.1/0.9 Hz, 1H, 4-H), 7.35 (td, J 7.3/1.3 Hz, 2H, arom. H, *m*-pos.), 7.59 (tt, *J* 7.3/1.3 Hz, 1H, arom. H, p-pos.), 8.04–8.08 (m, 2H, arom. H, o-pos.); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 32.5 (1C, C-6), 46.0 (1C, C-2), 55.3 (1C, OCH<sub>3</sub>), 70.4 (1C, C-4), 75.8 (1C, C-5), 99.6 (1C, C-1), 128.5 (2C, arom. C), 128.8 (1C, arom. C), 130.0 (2C, arom. C), 133.6 (1C, arom. C), 164.8 (1C, PhCOO), 196.9 (1C, CHCOCH<sub>2</sub>); EIMS: *m*/*z* [%] 105 [PhCO<sup>+</sup>, 100], 77 [Ph<sup>+</sup>, 24]; CIMS (NH<sub>3</sub>): *m*/*z* [%] 360/362  $[M + NH_4^+, 79/78], 343/345 [MH^+, 32/32], 328/330$ [MH<sup>+</sup>-CH<sub>3</sub>, 100/95], 311/313 [MH<sup>+</sup>- OCH<sub>3</sub>, 30/28], 105 [PhCO<sup>+</sup>, 25]; Anal. Calcd for  $C_{14}H_{15}BrO_5$  (343.2): C, 48.99; H, 4.41. Found: C, 48.97; H, 4.37.

**4.4.2.** Data for 12.  $R_f 0.17$ ; colorless oil; yield 0.70 g (8%); IR (met):  $\tilde{v}$  1730 (vO=C-O), 1690 (vC=O), 1655 (vC=C), 1597, 1491, 1451 (vC=C, arom.), 1245 (vO=C-O), 1109, 1040 (vC-O), 702, 690 cm<sup>-1</sup> ( $\gamma CH_{oop}$ , arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  3.69 (dd, J 11.9/4.9 Hz, 1H, CH<sub>2</sub>Br), 3.79 (dd, J 11.9/2.7 Hz, 1H, CH<sub>2</sub>Br), 4.79 (ddd, J 12.8/5.2/2.7 Hz, 1H, 2-H), 5.56 (d, J 5.8 Hz, 1H, 5-H), 5.81 (d, J 12.8 Hz, 1H, 3-H), 7.44 (d, J 6.1 Hz, 1H, 6-H), 7.45–7.51 (m, 2H, arom. H, *m*-pos.), 7.62 (tt, J 7.3/1.5 Hz, 1H, arom. H, *p*-pos.), 8.07–8.10 (m, 2H, arom. H, *o*-pos.); CIMS (isobutane): m/z [%] 311/313 [MH<sup>+</sup>, 100/95], 189/191 [M–PhCOO, 4/4], 105 [PhCO<sup>+</sup>, 7]; Anal. Calcd for C<sub>13</sub>H<sub>11</sub>BrO<sub>4</sub> (311.1): C, 50.19; H, 4.56. Found: C, 50.08; H, 4.20.

### 4.5. [(1*R*,3*S*,5*S*,8*R*)-5-Cyano-3-methoxy-2,6-dioxabicyclo[3.2.1]octane-8-yl] benzoate (13)

Bromide 11 (120 mg, 0.35 mmol) was dissolved in DMF (5 mL), and KCN (70 mg, 1.05 mmol) was added. After stirring at 80 °C for 1.5 h, H<sub>2</sub>O (5 mL) was added, and the mixture was extracted with petroleum ether (10 mL). Then, the aq layer was extracted with 95:5 petroleum ether–EtOAc  $(3 \times 10 \text{ mL})$ . The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo. The residue was purified by FC (1 cm, 8:2 petroleum ether-EtOAc, fractions 2 mL, R<sub>f</sub> 0.22). Colorless solid: mp 80 °C; yield 30 mg (26%);  $[\alpha]_{589}$  -62.5 (*c* 0.048, MeOH); IR (neat):  $\tilde{v}$  2255 (vC $\equiv$ N), 1734 (vO=C-O), 1269 (O=C-O), 1100 (vC-O), 909, 734 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 2.39 (dd, J 13.7/4.9 Hz, 1H, 4-H<sub>eq</sub>), 2.45 (dd, J 13.7/7.6 Hz, 1H, 4-H<sub>ax</sub>), 3.50 (s, 3H, OCH<sub>3</sub>), 4.27 (dd, J 10.7/2.7 Hz, 1H, 7-H), 4.32 (d, J 10.7 Hz 1H, 7-H), 4.81 (t, J 2.7 Hz, 1H, 1-H), 4.99 (dd, J 7.3/5.2 Hz, 1H, 3-H), 5.30 (d, J 3.0 Hz, 1H, 8-H), 7.50 (t, J 7.3 Hz, 2H, arom. H, m-pos.), 7.63 (tt, J 7.3/1.7 Hz, 1H, arom. H, p-pos.), 8.08-8.12 (m, 2H, arom. H, o-pos.); EIMS: m/z [%] 289 [M<sup>+</sup>, 1], 184 [M–PhCO, 3], 105 [PhCO<sup>+</sup>, 100], 77 [Ph<sup>+</sup>, 15]; CIMS

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(NH<sub>3</sub>): m/z [%] 307 [M+NH<sub>4</sub><sup>+</sup>, 100], 105 [PhCO<sup>+</sup>, 17]; HREIMS: calcd for C<sub>15</sub>H<sub>15</sub>NO<sub>5</sub>: 289.0950. Found: 289.0950.

### 4.6. Methyl 4-O-benzoyl-6-bromo-2,6-dideoxy- $\alpha$ -Derythro-hex-3-ulopyranoside dimethyl ketal (15 $\alpha$ ) and methyl 4-O-benzoyl-6-bromo-2,6-dideoxy- $\beta$ -D-erythrohex-3-ulopyranoside dimethyl ketal (15 $\beta$ )

*p*-Toluenesulfonic acid (0.13 g, 0.68 mmol) and trimethyl orthoformate (17 mL, 155 mmol) were added to a solution of ketone **11** (4.1 g, 11.9 mmol) in MeOH (40 mL), and the mixture was heated to reflux for 24 h. After completion of the reaction, a saturated solution of NaHCO<sub>3</sub> (60 mL) was added, and the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 100 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (5.5 cm, 8:2 petroleum ether–EtOAc, fractions 20 mL).

**4.6.1. Data for 15\alpha.**  $R_{\rm f}$  0.35; colorless oil; yield 1.50 g (32%);  $[\alpha]_{589}$  +55.0 (c 0.166, MeOH); IR (neat):  $\tilde{v}$  1721 (v*O*=*C*-O), 1263 (vO=*C*-O), 1110, 1048 (vC-O), 708 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.08 (ddd, J 14.3/3.9/0.9 Hz, 1H, 2-H<sub>eq</sub>), 2.23 (dd, J 14.3/ 4.5 Hz, 1H, 2-H<sub>ax</sub>), 3.25 (s, 3H, OCH<sub>3</sub>), 3.32 (s, 3H, OCH<sub>3</sub>), 3.49 (s, 3H, OCH<sub>3</sub>), 3.53–3.57 (m, 2H, CH<sub>2</sub>Br), 4.33 (td, J 6.7/5.2 Hz, 1H, 5-H), 4.80 (t, J 4.3 Hz, 1H, 1-Hea), 5.23 (dd, J 6.7/0.9 Hz, 1H, 4-H), 7.46 (td, J 7.3/ 1.5 Hz, 2H, arom. H, m-pos.), 7.59 (tt, J 7.3/1.5 Hz, 1H, arom. H, p-pos.), 8.06-8.10 (m, 2H, arom. H, opos.); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  31.5 (1C, C-6), 35.6 (1C, C-2), 48.9 (1C, OCH<sub>3</sub>), 49.5 (1C, OCH<sub>3</sub>), 56.0 (1C, OCH<sub>3</sub>), 73.3 (1C, C-4), 74.1 (1C, C-5), 97.2 (1C, C-1), 100.1 (1C, C-3), 128.5 (2C, arom. C, m-pos.), 128.6 (1C, quart., arom. C), 133.5 (2C, arom. C, o-pos.), 133.5 (1C, arom. C, p-pos.), 165.4 (1C, PhCOO); EIMS: *m*/*z* [%] 357/359 [M–OCH<sub>3</sub>, 6/6]; CIMS (NH<sub>3</sub>): *m*/*z* [%]  $406/408 \ [M+NH_4^+, 4.5/4.5], 357/359 \ [M-OCH_3, 100/$ 97]; Anal. Calcd for C<sub>16</sub>H<sub>21</sub>BrO<sub>6</sub> (389.2): C, 49.37; H, 5.44. Found: C, 49.53; H, 5.43.

**4.6.2.** Data for 15β.  $R_f 0.41$ ; colorless solid; mp 73 °C, yield 0.61 g (13%)];  $[\alpha]_{589} -22.0$  (*c* 0.216, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.87 (dd, *J* 13.4/7.3 Hz, 1H, 2-H<sub>ax</sub>), 2.31 (dd, *J* 13.4/2.7 Hz, 1H, 2-H<sub>eq</sub>), 3.24 (s, 3H, OCH<sub>3</sub>), 3.38 (s, 3H, OCH<sub>3</sub>), 3.52 (dd, *J* 11.0/7.3 Hz, 1H, CH<sub>2</sub>Br), 3.54 (s, 3H, OCH<sub>3</sub>), 3.58 (dd, *J* 11.0/ 4.0 Hz, 1H, CH<sub>2</sub>Br), 4.07 (td, *J* 7.8/3.9 Hz, 1H, 5-H), 4.74 (dd, *J* 7.6/2.7 Hz, 1H, 1-H<sub>ax</sub>), 5.27 (d, *J* 7.9 Hz, 1H, 4-H), 7.47 (t, *J* 7.6 Hz, 2H, arom. H, *m*-pos.), 7.59 (tt, *J* 7.5/1.5 Hz, 1H, arom. H, *p*-pos.), 8.04–8.08 (m, 2H, arom. H, *o*-pos.); <sup>13</sup>C NMR (CDCl<sub>3</sub>): 32.4 (1C, C-6), 37.5 (1C, C-2), 49.1 (1C, OCH<sub>3</sub>), 49.7 (1C, OCH<sub>3</sub>), 56.7 (1C, OCH<sub>3</sub>), 71.9 (1C, C-4), 72.1 (1C, C-5), 97.5 (1C, C-3), 98.4 (1C, C-1), 128.6 (2C, arom. C, *m*-pos.); 129.4 (1C, quart., arom. C), 130.0 (2C, arom. C, *o*-pos.), 133.5 (1C, arom. C, *p*-pos.), 165.3 (1C, Ph*C*OO); Anal. Calcd for  $C_{16}H_{21}BrO_6$  (389.2): C, 49.37; H, 5.44. Found: C, 49.11; H, 5.23.

In addition to the pure samples  $15\alpha$  and  $15\beta$ , a mixture of  $15\alpha/15\beta$ , was isolated as a colorless: oil, yield 2.0 g (43%); total yield 88%.

### 4.7. Methyl 4-O-benzoyl-6-cyano-2,6-dideoxy-α-Derythro-hex-3-ulopyranoside dimethyl ketal (16α)

A solution of  $15\alpha$  (1.5 g, 3.8 mmol) and KCN (1.5 g, 23.7 mmol) in DMSO (20 mL) was stirred at 70 °C for 2.5 h. After cooling to room temperature, water (25 mL) was added, and the aq layer was extracted with 95:5 petroleum ether–EtOAc  $(3 \times 50 \text{ mL})$ . The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (4 cm, 8:2 EtOAcpetroleum ether, fractions 10 mL, R<sub>f</sub> 0.39). Colorless oil: yield 1.2 g (94%); [α]<sub>589</sub> +59.0 (c 0.473, MeOH); IR (neat):  $\tilde{v}$  1720 (vO=C-O), 1266 (vO=C-O), 1105, 1049 (vC–O), 711 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 2.05 (dd, J 14.6/3.9 Hz, 1H, 2-H<sub>ax</sub>), 2.27 (dd, J 14.6/ 3.9 Hz, 1H, 2-H<sub>eq</sub>), 2.62 (dd, 1H, J 16.8/4.3 Hz, CH<sub>2</sub>CN), 2.76 (dd, 1H, J 16.8/9.3 Hz, CH<sub>2</sub>CN), 3.24 (s, 3H, OCH<sub>3</sub>), 3.34 (s, 3H, OCH<sub>3</sub>), 3.48 (s, 3H, OCH<sub>3</sub>), 4.38 (ddd, J 8.9/7.3/4.3 Hz, 1H, 5-H), 4.80 (t, J 3.9 Hz, 1H, 1-H<sub>eq</sub>), 5.11 (d, J 7.3 Hz, 1H, 4-H), 7.46 (t, J 7.6 Hz, 2H, arom. H, m-pos.), 7.60 (tt, J 7.5/ 1.5 Hz, 1H, arom. H, p-pos.), 8.06-8.09 (m, 2H, arom. H, *o*-pos.); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 20.7 (1C, C-6), 35.6 (1C, C-2), 49.0 (1C, OCH<sub>3</sub>), 49.6 (1C, OCH<sub>3</sub>), 55.9 (1C, OCH<sub>3</sub>), 67.4 (1C, C-5 o.C-4), 73.1 (1C, C-4 o.C-5), 97.4 (1C, C-1 o.C-3), 97.9 (1C, C-3 o.C-1), 116.8 (1C, CN), 128.5 (2C, arom. C), 129.0 (1C, arom. C), 130.0 (2C, arom. C), 133.7 (1C, arom. C), 165.6 (1C, PhCOO); EIMS: m/z [%] 304 [M-OCH<sub>3</sub>, 10], 272 [M-HOCH<sub>3</sub>-OCH<sub>3</sub>, 8], 105 [PhCO<sup>+</sup>, 100], 77 [Ph<sup>+</sup>, 24]; CIMS (NH<sub>3</sub>): m/z [%] 353 [M + NH<sub>4</sub><sup>+</sup>, 100], 304  $[M-OCH_3, 99]$ ; Anal. Calcd for  $C_{17}H_{21}NO_6$  (335.4): C, 60.89; H, 6.31; N, 4.18. Found: C, 60.84; H, 6.46; N, 4.15.

### 4.8. Methyl 4-*O*-benzoyl-6-cyano-2,6-dideoxy-β-Derythro-hex-3-ulopyranoside dimethyl ketal (16β)

A solution of 15 $\beta$  (810 mg, 2.1 mmol) and KCN (843 mg, 12.9 mmol) in DMSO (25 mL) was stirred at 70 °C for 2.5 h. After cooling to room temperature, water (25 mL) was added, and the aq layer was extracted with 95:5 petroleum ether–EtOAc (3 × 50 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (3 cm, 8:2 EtOAc–petroleum ether, fractions 10 mL,  $R_f$  0.43). Colorless solid: mp 112 °C; yield 465 mg (67%); [ $\alpha$ ]<sub>589</sub> –6.0 (*c* 0.360, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.85 (dd,

J 13.7/7.6 Hz, 1H, 2-H<sub>ax</sub>), 2.33 (dd, J 13.7/2.7 Hz, 1H, 2-H<sub>eq</sub>), 2.63 (dd, 1H, J 16.8/4.3 Hz, CH<sub>2</sub>CN), 2.79 (dd, J 16.8/8.9 Hz, 1H, CH<sub>2</sub>CN), 3.24 (s, 3H, OCH<sub>3</sub>), 3.39 (s, 3H, OCH<sub>3</sub>), 3.55 (s, 3H, OCH<sub>3</sub>), 4.13 (ddd, J 8.9/7.9/4.3 Hz, 1H, 5-H), 4.76 (dd, J 7.6/2.7 Hz, 1H, 1-H<sub>ax</sub>), 5.15 (d, J 7.9 Hz, 1H, 4-H), 7.48 (t, J 7.5 Hz, 2H, arom. H, *m*-pos.), 7.61 (tt, *J* 7.5/1.5 Hz, 1H, arom. H, p-pos.), 8.04–8.07 (m, 2H, arom. H, o-pos.); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 21.6 (1C, C-6), 37.4 (1C, C-2), 49.1 (1C, OCH<sub>3</sub>), 49.7 (1C, OCH<sub>3</sub>), 56.8 (1C, OCH<sub>3</sub>), 70.0 (1C, C-5 o.C-4), 73.9 (1C, C-4 o.C-5), 97.2 (1C, C-1 o.C-3), 100.3 (1C, C-3 o.C-1), 117.0 (1C, CN), 128.6 (2C, arom. C), 129.0 (1C, arom. C), 129.9 (2C, arom. C), 133.7 (1C, arom. C), 165.4 (1C, PhCOO); Anal. Calcd for C<sub>17</sub>H<sub>21</sub>NO<sub>6</sub> (335.4): C, 60.89; H, 6.31; N, 4.18. Found: C, 60.55; H, 6.29; N, 4.05.

### 4.9. Methyl 6-cyano-2,6-dideoxy- $\alpha$ -D-*erythro*-hex-3-ulopyranoside dimethyl ketal (17 $\alpha$ )

A mixture of 16α (0.76 g, 2.3 mmol), MeOH (40 mL), and 3 N NaOH (10 mL) was stirred at room temperature for 9 h. After addition of H<sub>2</sub>O (10 mL), it was extracted with  $CH_2Cl_2$  (4 × 50 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (3 cm, 7:3 petroleum ether-EtOAc, fractions 10 mL, Rf 0.31). Colorless crystals: mp 83 °C; yield 0.4 g (76%); [a]<sub>589</sub> +66.2 (c 0.303, MeOH); IR (neat): v 2945 (vC-H), 2279 (vCN), 1127, 1046 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.74 (dd, J 15.0/4.5 Hz, 1H, 2-Hax), 2.31 (dd, J 15.2/1.2 Hz, 1H, 2-H<sub>eq</sub>), 2.60 (dd, J 16.9/7.8 Hz, 1H, CH<sub>2</sub>CN), 2.85 (dd, J 16.5/3.5 Hz, 1H, CH<sub>2</sub>CN), 3.45 (t, J 9.9 Hz, 1H, 4-H), 3.33 (s, 3H, OCH<sub>3</sub>), 3.36 (s, 3H, OCH<sub>3</sub>), 3.37 (s, 3H, OCH<sub>3</sub>), 3.82 (ddd, J 9.5/7.7/3.1 Hz, 1H, 5-H), 4.73 (dd, J 4.5/1.2 Hz, 1H, 1-H<sub>eq</sub>), a signal for the OH-proton was not observed;  ${}^{13}C$  NMR (CDCl<sub>3</sub>):  $\delta$ 20.8 (1C, C-6), 34.8 (1C, C-2), 49.1 (1C, OCH<sub>3</sub>), 51.0 (1C, OCH<sub>3</sub>), 55.4 (1C, OCH<sub>3</sub>), 67.5 (1C, C-5), 74.3 (1C, C-4), 96.5 (1C, C-1 o.C-3), 97.8 (1C, C-3 o.C-1), 117.5 (1C, CN); EIMS: m/z [%] 231 [M<sup>+</sup>, 1], 200 M-OCH<sub>3</sub>. 7], 168 [M-HOCH<sub>3</sub>-OCH<sub>3</sub>, 25], 88  $[-C(OCH_3)_2 - CH_2^+, 100];$  CIMS  $(NH_3): m/z$  [%] 249  $[M + NH_4^+, 74]$ ; CIMS (isobutane): m/z [%] 288  $[M + C_4H_9^+, 13]$ , 232  $[MH^+, 28]$ , 168  $[M-HOCH_3-$ OCH<sub>3</sub>, 47]; Anal. Calcd for C<sub>10</sub>H<sub>17</sub>NO<sub>5</sub> (231.3): C, 51.94; H, 7.41; N, 6.06. Found: C, 52.13; H, 7.54; N, 5.82.

### 4.10. Methyl 6-cyano-2,6-dideoxy-β-D-*erythro*-hex-3ulopyranoside dimethyl ketal (17β)

A mixture of  $16\beta$  (316 mg, 0.9 mmol), MeOH (15 mL), and 3 N NaOH (6 mL) was stirred at room temperature for 9 h. After addition of H<sub>2</sub>O (5 mL), the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (25 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (3 cm, 7:3 petroleum ether-EtOAc, fractions 10 mL,  $R_f$  0.32). Colorless solid: mp 82 °C; yield 174 mg (80%);  $[\alpha]_{589}$  -35.0 (c 0.195, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.59 (dd, J 14.0/9.5 Hz, 1H, 2-H<sub>ax</sub>), 2.29 (dd, J 14.0/2.1 Hz, 1H, 2-H<sub>ea</sub>), 2.64 (dd, J 16.8/7.9 Hz, 1H, CH<sub>2</sub>CN), 2.47 (s, broad, 1H, OH), 2.89 (dd, J 16.8/3.4 Hz, 1H, CH<sub>2</sub>CN), 3.41–3.47 (m, 1H, 4-H), 3.34 (s, 3H, OCH<sub>3</sub>), 3.36 (s, 3H, OCH<sub>3</sub>), 3.50 (s, 3H, OCH<sub>3</sub>), 3.58 (ddd, J 9.3/7.8/3.6 Hz, 1H, 5-H), 4.47 (dd, J 9.5/2.1 Hz, 1H, 1- $H_{ax}$ ); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  21.2 (1C, C-6), 37.0 (1C, C-2), 48.9 (1C, OCH<sub>3</sub>), 50.8 (1C, OCH<sub>3</sub>), 56.5 (1C, OCH<sub>3</sub>), 71.7 (1C, C-5), 74.4 (1C, C-4), 97.3 (1C, C-1 o.C-3), 99.7 (1C, C-3 o.C-1), 117.3 (1C, CN); Anal. Calcd for C<sub>10</sub>H<sub>17</sub>NO<sub>5</sub> (231.3): C, 51.94; H, 7.41; N, 6.06. Found: C, 51.64; H, 7.36; N, 5.95.

## 4.11. Methyl 7-amino-2,6,7-trideoxy- $\alpha$ - and $\beta$ -D-*erythro*-hept-3-ulo-1,5-pyranoside dimethyl ketal (18 $\alpha$ and 18 $\beta$ )

A solution of **16** (anomeric mixture, 1.0 g, 3.1 mmol) in MeOH (60 mL) and 5 N NaOH (15 mL) was stirred at room temperature for 2 h. After completion of the reaction, Raney nickel was added to the resulting hydrolysis product (17, anomeric mixture), and the mixture was shaken under an H<sub>2</sub> atmosphere (4.1 bar) at room temperature for 24 h. Raney Ni was removed by filtration through Celite<sup>®</sup> AFA. The solution was concentrated to a volume of about 20 mL. After addition of water (10 mL), the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>  $(4 \times 50 \text{ mL})$ . The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo to yield a yellow oil (yield 0.70 g, 97%), which was pure enough for further reactions. In order to characterize the primary amine 18, a sample (103 mg) of the residue was purified by FC [2 cm, 8:2 ethanol-acetone, 2% N-ethyl-N,N-dimethylamine, fractions 5 mL,  $R_f$  0.11, yield 17 mg (<10%)]; IR (neat):  $\tilde{v}$  2945 (vC–H), 1128, 1053 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 1.47 (dd, J 13.7/9.8 Hz, 0.45H, 2- $H_{ax}$ , β-is.), 1.69 (dd, J 14.9/4.6 Hz, 0.55H, 2- $H_{ax}$ , αis.), 1.70–1.86 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>, α+β-is.), 1.92– 2.03 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>, α+β-is.), 2.21 (dd, J 12.8/ 2.1 Hz, 0.45H, 2-H<sub>eq</sub>,  $\beta$ -is.), 2.23 (dd, J 14.0/2.0 Hz, 0.55H, 2-Heg, α-is.), 2.94-3.07 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>,  $\alpha+\beta$ -is.), 2.80–2.91 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>,  $\alpha+\beta$ -is.), 3.29-3.53 (m, 0.45H, 5-H,  $\beta$ -is./1H, 4-H,  $\alpha+\beta$ -is./2H, NH<sub>2</sub>,  $\alpha + \beta$ -is.), 3.28 (s, 3 × 0.55H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.30 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.31 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>, β-is.), 3.34 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>, α-is.), 3.37 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>,  $\beta$ -is.), 3.42 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>,  $\beta$ is.) 3.67 (td, J 8.6/4.0 Hz, 0.55H, 5-H, α-is.), 4.39 (dd, J 9.8/1.8 Hz, 0.45H, 1-H<sub>ax</sub>,  $\beta$ -is.), 4.63 (dd, J 4.4/ 2.1 Hz, 0.55H, 1-H<sub>eq</sub>,  $\alpha$ -is.), a signal for the OH-proton was not found; the ratio of  $\alpha$ - and  $\beta$ -anomers was 55:45; EIMS: *m*/*z* [%] 204 [M–OCH<sub>3</sub>, 7], 173 [M–2×OCH<sub>3</sub>,

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3], 88 [C(OCH<sub>3</sub>)<sub>2</sub>-CH<sub>2</sub><sup>+</sup>-, 100]; CIMS (NH<sub>3</sub>): m/z [%] 236 [MH<sup>+</sup>, 17], 204 [M-OCH<sub>3</sub>, 100]; HREIMS: calcd for C<sub>9</sub>H<sub>18</sub>NO<sub>5</sub>: 220.1185. Found: 220.1185.

## 4.12. Methyl 7-(acetylamino)-2,6,7-trideoxy- $\alpha$ - and $\beta$ -D-*erythro*-hept-3-ulo-1,5-pyranoside dimethyl ketal (19 $\alpha$ and 19 $\beta$ )

Under N<sub>2</sub> atmosphere, Et<sub>3</sub>N (0.4 mL, 2.9 mmol) and  $Ac_2O$  (0.2 mL, 2.1 mmol) were added to a solution of crude primary amine 18 (anomeric mixture, 101 mg, 0.37 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL). After stirring the mixture at room temperature for 2.5 h, the mixture was concentrated in vacuo, and the residue was purified by FC (2 cm, 9:1 EtOAc–MeOH, fractions 5 mL,  $R_{\rm f}$ 0.31). Yellow oil: yield 47 mg (44% referring to nitrile **16**); IR (neat):  $\tilde{v}$  3299 (vN–H), 2943 (vC–H), 1650 (vO=C-NH, amide I), 1553 (δN-H, amide II), 1129, 1047 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.52 (dd, J 14.0/9.8 Hz, 0.45H, 2-H<sub>ax</sub>,  $\beta$ -is.), 1.61–1.76 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 1.73 (dd, J 14.9/4.3 Hz, 0.55H, 2-H<sub>ax</sub>,  $\alpha$ -is.), 1.96 (s, 3H, NHCOCH<sub>3</sub>,  $\alpha$ + $\beta$ -is.), 2.00-2.19 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha$ + $\beta$ -is.), 2.25 (dd, J 14.9/ 1.8 Hz, 0.55H, 2-H<sub>eq</sub>,  $\alpha$ -is.), 2.30 (dd, J 14.0/1.8 Hz, 0.45H, 2-H<sub>eq</sub>, β-is.), 2.42 (d, J 7.6 Hz, 0.45H, OH, βis.), 2.53 (d, J 9.5 Hz, 0.55H, OH, α-is.), 3.24–3.59 (m, 1H, 4-H,  $\alpha+\beta$ -is./2H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha+\beta$ -is.), 3.32 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.33 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>,  $\alpha$ is.), 3.34 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>,  $\beta$ -is.), 3.36 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.38 (s, 3×0.45H, OCH<sub>3</sub>,  $\beta$ -is.), 3.49 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>, β-is.), 3.64 (td, J 9.2/2.7 Hz, 1H, 5-H,  $\alpha + \beta$ -is.), 4.69 (dd, J 4.5/2.0 Hz, 0.55H, 1-H<sub>eq</sub>,  $\alpha$ is.), 4.40 (dd, J 9.8/2.1 Hz, 0.45H, 1-H<sub>ax</sub>,  $\beta$ -is.), 6.00 (s, broad, 1H, NH,  $\alpha+\beta$ -is.); the ratio of  $\alpha$ - and  $\beta$ -anomers was 55:45; EIMS: m/z [%] 246 [M-OCH<sub>3</sub>, 2], 215  $[M-2 \times OCH_3, 2];$  ESIMS: m/z [%] 300  $[M+Na^+, 100],$ 214 [M-HOCH<sub>3</sub>-OCH<sub>3</sub>, 31]; CIMS (NH<sub>3</sub>): *m*/*z* [%] 295  $[M + NH_4^+, 3]$ , 278  $[MH^+, 13]$ , 246  $[M-OCH_3,$ 100], 214 [M-HOCH<sub>3</sub>-OCH<sub>3</sub>, 83]. HREIMS: calcd for C<sub>11</sub>H<sub>20</sub>NO<sub>5</sub>: 246.1342. Found: 246.1338.

# 4.13. Methyl 7-(benzyloxycarbonylamino)-2,6,7-trideoxy- $\alpha$ - and $\beta$ -D-*erythro*-hept-3-ulo-1,5-pyranoside dimethyl ketal (20 $\alpha$ and 20 $\beta$ )

Under N<sub>2</sub> atmosphere, Et<sub>3</sub>N (0.4 mL, 2.9 mmol) and benzyl chloroformate (0.4 mL, 2.8 mmol) were added to a solution of crude primary amine **18** (anomeric mixture, 180 mg, 0.77 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL). After stirring the reaction mixture at room temperature for 5 h, it was concentrated in vacuo. The residue was purified by FC (2 cm, 6:4 petroleum ether–EtOAc, fractions 5 mL,  $R_{\rm f}$  0.22). Colorless oil: yield 177 mg (60%, referring to nitrile **16**); IR (neat):  $\tilde{\nu}$  3347 (vN–H), 1702 (v*O*=*C*– NH, amide I), 1525 ( $\delta$ N–H, amide II), 1250, 1128, 1051 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  50 (dd, *J*  14.0/9.8 Hz, 0.5H, 2-H<sub>ax</sub>,  $\beta$ -is.), 1.61–1.72 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 1.69 (dd, J 14.9/4.6 Hz, 0.5H, 2-Hax, α-is.), 2.04-2.18 (m, 1H, m, 1H, CH2CH2NH,  $\alpha + \beta$ -is.), 2.23 (dd, J 14.0/2.1 Hz, 0.5H, 2-H<sub>eq</sub>,  $\beta$ -is.), 2.27 (dd, J 14.0/1.8 Hz, 0.5H, 2-H<sub>eq</sub>,  $\alpha$ -is.), 2.28 (d, J 8.5 Hz, 0.5H, OH, β-is.), 2.44 (d, J 9.8 Hz, 0.5H, OH,  $\alpha$ -is.), 3.25–3.48 (m, 1H, 4-H,  $\alpha$ + $\beta$ -is./2H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha + \beta$ -is.), 3.28 (s, 3 × 0.5H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.29 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.31 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ β-is.), 3.33 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>, α/β-is.), 3.35 (s,  $6 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.59 (td, J 9.2/2.4 Hz, 1H, 5-H,  $\alpha + \beta$ -is.), 4.35 (dd, J 9.8/2.1 Hz, 0.5H, 1-H<sub>ax</sub>,  $\beta$ -is.), 4.65 (dd, J 4.3/1.8 Hz, 0.5H, 1-H<sub>eq</sub>, α-is.), 5.07 (s, 2H, OCH<sub>2</sub>Ph,  $\alpha+\beta$ -is.), 5.13 (s, broad, 1H, NH,  $\alpha+\beta$ -is.), 7.26–7.38 (m, 5H, arom. H,  $\alpha+\beta$ -is.); the ratio of  $\alpha$ and  $\beta$ -anomers was 1:1; EIMS: m/z [%] 91 [CH<sub>2</sub>Ph<sup>+</sup>, 100]. ESIMS: *m*/*z* [%] 392 [M+Na<sup>+</sup>, 100]. CIMS  $(NH_3): m/z \ [\%] \ 287 \ [M + NH_4^+, 3], \ 338 \ [M - OCH_3, 16],$ 306 [M<sup>+</sup>-HOCH<sub>3</sub>-OCH<sub>3</sub>, 100]. HREIMS: calcd for C<sub>17</sub>H<sub>24</sub>NO<sub>6</sub>: 338.1604. Found: 338.1611.

### 4.14. Methyl 7-[2-(3,4-dichlorophenyl)acetylamino]-2,6,7trideoxy- $\alpha$ - and $\beta$ -D-*erythro*-hept-3-ulo-1,5-pyranoside dimethyl ketal (22 $\alpha$ and 22 $\beta$ )

Under N<sub>2</sub> atmosphere, a solution of (3,4-dichlorophenyl)acetic acid (540 mg, 2.6 mmol) and 1,1'-carbonyldiimidazole (420 mg, 2.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was stirred at room temperature for 1 h. Then, the unpurified primary amine 18 (anomeric mixture, 498 mg, 2.1 mmol) dissolved in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added dropwise to the mixture under ice cooling. The mixture was stirred at room temperature for 6 h. After completion of the transformation the solvent was removed in vacuo, and the residue was purified by FC (3 cm, 95:5 EtOAcacetone, fractions 10 mL, R<sub>f</sub> 0.31). Pale-yellow oil: yield 468 mg (52%, referring to nitrile 16); IR (neat):  $\tilde{v}$  3301 (vN-H), 2942 (vC-H), 1647 (vO=C-NH, amide I), 1552 (δN-H, amide II), 1129, 1048 cm<sup>-1</sup> (νC-O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.49 (dd, J 14.0/9.8 Hz, 0.5H, 2-H<sub>ax</sub>,  $\beta$ -is.), 1.59–1.75 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha$ + $\beta$ -is.), 1.64 (dd, J 14.9/4.3 Hz, 0.5H, 2-H<sub>ax</sub>,  $\alpha$ -is.), 2.00–2.14 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 2.24 (dd, J 14.9/1.5 Hz, 0.5H, 2-H<sub>eq</sub>,  $\alpha$ -is.), 2.28 (dd, J 14.0/1.8 Hz, 0.5H, 2-H<sub>eq</sub>, β-is.), 2.36 (d, J 8.6 Hz, 0.5H, OH, β-is.), 2.46 (d, J 10.1 Hz, 0.5H, OH, α-is.), 3.21–3.62 (m, 1H, 4-H,  $\alpha+\beta$ -is./1H, 5-H,  $\alpha+\beta$ -is./2H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha+\beta$ -is.), 3.17 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.30 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.33 (s, 3 × 0.5H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.34 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.36 (s,  $3 \times 0.5$ H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.42 (s, 3×0.5H, OCH<sub>3</sub>,  $\alpha/\beta$ -is.), 3.47 (s,  $2 \times 0.5$ H, COC $H_2$ Ph,  $\alpha/\beta$ -is.), 3.50 (s,  $2 \times 0.5$ H, COCH<sub>2</sub>Ph, α/β-is.), 4.33 (dd, J 9.5/2.0 Hz, 0.5H, 1- $H_{ax}$ , β-is.), 4.45 (dd, J 4.3/1.5 Hz, 0.5H, 1- $H_{ea}$ , α-is.), 6.00 (s, broad, 0.5H, NH,  $\alpha/\beta$ -is.), 6.13 (s, broad, 0.5H, NH,  $\alpha/\beta$ -is.), 7.12 (dd, J 8.2/2.1 Hz, 0.5H, arom.

H, 6'H, α/β-is.), 7.13 (dd, J 8.2/2.1 Hz, 0.5H, arom. H, 6'-H, α/β-is.), 7.37 (d, J 2.4 Hz, 0.5H, arom. H, 2'-H, α/ β-is.), 7.38 (d, J 2.4 Hz, 0.5H, arom. H, 2'-H, α/β-is.), 7.39 (d, J 8.2 Hz, 0.5H, arom. H, 5'-H, α/β-is.), 7.42 (d, J 8.2 Hz, 0.5H, arom. H, 5'-H, α/β-is.); the ratio of α- and β-anomers was 1:1; EIMS: m/z [%] 358/360/362 [M<sup>+</sup>-HOCH<sub>3</sub>-OCH<sub>3</sub>, 7.7/5.3/0.9]; CIMS (NH<sub>3</sub>): m/z[%] 438/440/442 [M+NH<sub>3</sub>, 4.1/2.7/0.5], 422/424/426 [MH<sup>+</sup>, 17/11/2], 358/360/362 [M-HOCH<sub>3</sub>-OCH<sub>3</sub>, 100/65/12]; Anal. Calcd for C<sub>18</sub>H<sub>25</sub>Cl<sub>2</sub>NO<sub>6</sub> (422.3): C, 51.19; H, 5.97; N, 3.32. Found: C, 50.98; H, 6.11; N, 3.47.

### 4.15. Methyl 7-(benzoylamino)-2,6,7-trideoxy- $\alpha$ - and $\beta$ -Derythro-hept-3-ulo-1,5-pyranoside (23 $\alpha$ and 23 $\beta$ )

Under  $N_2$  atmosphere, benzoyl chloride (0.33 mL, 2.8 mmol) and Et<sub>3</sub>N (0.54 mL, 3.9 mmol) were added to a solution of the unpurified primary amine 18 (anomeric mixture, 468 mg, 2.1 mmol) in  $CH_2Cl_2$  (25 mL). The mixture was stirred at room temperature for 22 h. The mixture was concentrated in vacuo, and the residue was purified by FC (2 cm, EtOAc, fractions 5 mL,  $R_{\rm f}$ 0.39). Pale-yellow solid: mp 154 °C; yield 180 mg (25%, referring to nitrile 16); IR (neat):  $\tilde{v}$  3340 (vN-H), 2943 (vC-H), 1712 (vC=O), (1638 (vO=C-NH), amide I), 1536 ( $\delta$ N–H, amide II), 1051 (vC–O), 711, 693 cm<sup>-1</sup> ( $\gamma CH_{oop}$ , arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.99–2.13 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 2.22–2.35 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 2.71 (dd, J 14.0/1.2 Hz, 0.6H, 2-H<sub>eq</sub>, α-is.), 2.70 (ddd, J 14.3/9.2/1.5 Hz, 0.4H, 2-H<sub>ax</sub>,  $\beta$ -is.), 2.83 (ddd, J 14.0/4.6/1.2 Hz, 0.6H, 2-H<sub>ax</sub>,  $\alpha$ -is.), 2.87 (dd, J 14.3/2.7 Hz, 0.4H, 2-H<sub>eq</sub>, β-is.), 3.23 (s,  $3 \times 0.6$ H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.52 (s,  $3 \times 0.4$ H, OCH<sub>3</sub>,  $\beta$ -is.), 3.36–3.91 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha$ + $\beta$ -is./1H, OH,  $\alpha$ + $\beta$ is./1H, 5-H, α+β-is.), 3.95 (dd, J 10.1/1.5 Hz, 0.4H, 4-H,  $\beta$ -is.), 3.97 (dd, J 9.8/1.2 Hz, 0.6H, 4-H,  $\alpha$ -is.), 4.58 (dd, J 9.2/2.4 Hz, 0.4H, 1-H<sub>ax</sub>, β-is.), 5.14 (d, J 4.0 Hz, 0.6H, 1-H<sub>eq</sub>,  $\alpha$ -is.), 6.76 (s, broad, 1H, NH,  $\alpha$ + $\beta$ -is.), 7.39–7.53 (m, 3H, arom. H,  $\alpha+\beta$ -is.), 7.75–7.79 (m, 2H, arom. H,  $\alpha + \beta$ -is.); the ratio of  $\alpha$ - and  $\beta$ -anomers was 3:2; EIMS: m/z [%] 243 [M-HOCH<sub>3</sub>-H<sub>2</sub>O, 9], 105 [PhCO<sup>+</sup>, 100]; CIMS (isobutane): m/z [%] 294  $[MH^+, 100]$ , 262  $[M-OCH_3, 85]$ ; Anal. Calcd for C<sub>15</sub>H<sub>19</sub>NO<sub>5</sub> (293.3): C, 61.42; H, 6.52; N, 4.78. Found: C, 61.71; H, 6.54; N, 4.69.

### 4.16. Methyl 7-[2-(3,4-dichlorophenyl)acetylamino]-2,6,7trideoxy- $\alpha$ - and $\beta$ -D-*erythro*-hept-3-ulo-1,5-pyranoside (24 $\alpha$ and 24 $\beta$ )

A solution of acetal **22** (anomeric mixture, 178 mg, 0.42 mmol) and *p*-toluenesulfonic acid monohydrate (53 mg, 0.28 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was stirred at 0 °C for 4 h. After completion of the transformation, the mixture was washed with satd aq NaHCO<sub>3</sub>

 $(2 \times 10 \text{ mL})$ . The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (2 cm, 95:5 EtOAc-acetone, fractions 5 mL,  $R_{\rm f} = 0.23$ ). Colorless solid, mp 147 °C, yield 127 mg (80%); IR (neat): v 3423 (vN-H), 3287 (vO-H), 2968 (vC-H), 1627 (vO=C-NH, amide I), 1566 (δN-H, amide II), 1118 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 1.79–1.96 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 2.05–2.18 (m, 1H,  $CH_2CH_2NH$ ,  $\alpha+\beta$ -is.), 2.65 (dd, J 14.0/ 9.2 Hz, 0.45H, 2-H<sub>ax</sub>, β-is.), 2.66 (d, J 14.0 Hz, 0.55H, 2-Heq, α-is.), 2.74 (dd, J 14.0/4.3 Hz, 0.55H, 2-Hax, αis.), 2.85 (dd, J 14.3/2.7 Hz, 0.45H, 2-H<sub>eq</sub>,  $\beta$ -is.), 3.20 (s, 2H, COC $H_2$ Ph,  $\alpha+\beta$ -is.), 3.26–3.68 (m, 1H, 5-H,  $\alpha+\beta$ -is./1H, OH,  $\alpha+\beta$ -is./2H, CH<sub>2</sub>CH<sub>2</sub>NH,  $\alpha+\beta$ -is.), 3.48 (s,  $3 \times 0.45$ H, OCH<sub>3</sub>,  $\beta$ -is.), 3.51 (s,  $3 \times 0.55$ H, OCH<sub>3</sub>,  $\alpha$ -is.), 3.84 (d, J 9.8 Hz, 1H, 4-H,  $\alpha$ + $\beta$ -is.), 4.49 (dd, J = 9.2/2.4 Hz, 0.45H, 1-H<sub>ax</sub>,  $\beta$ -is.), 4.90 (d, J4.3 Hz, 0.55H, 1-Heq, a-is.), 5.81 (s, broad, 0.45H, NH, β-is.), 5.94 (s, broad, 0.55H, NH, α-is.), 7.11 (dd, J 8.2/2.1 Hz, 0.45H, arom. H, 6',  $\beta$ -is.), 7.12 (dd, J 8.2/2.4 Hz, 0.55H, arom. H, 6'-H, α-is.), 7.37 (d, J 2.1 Hz, 0.55H, arom. H, 2'-H, α-is.), 7.38 (d, J 1.8 Hz, 0.45H, arom. H, 2'-H, β-is.), 7.41 (d, J 8.2 Hz, 0.45H, arom. H, 5'-H, β-is.), 7.42 (d, J 8.2 Hz, 0.55H, arom. H, 5'-H,  $\alpha$ -is.); the ratio of  $\alpha$ - and  $\beta$ -anomers was 55:45; EIMS: m/z [%] 343/345/347 [M-HOCH<sub>3</sub>, 16/ 11/2], 159/161/163 [-CH<sub>2</sub>PhCl<sup>+</sup><sub>2</sub>, 100/66/11]; CIMS (NH<sub>3</sub>): m/z [%] 393/395 [M+NH<sub>4</sub><sup>+</sup>, 10/7], 376/378/ 380 [MH<sup>+</sup>, 100/63/15], 344/346/348 [M-HOCH<sub>3</sub>, 95/60/11]; Anal. Calcd for  $C_{16}H_{19}Cl_2NO_5$  (376.3): C, 51.08; H, 5.09; N, 3.72. Found: C, 51.84; H, 5.58; N, 3.72.

### 4.17. 1,5-Anhydro-7-(benzoylamino)-2,6,7-trideoxy-Derythro-hept-1- en-3-ulose (25)

A solution of ketone 23 (anomeric mixture, 97 mg, 0.29 mmol) and p-toluenesulfonic acid monohydrate (62 mg, 0.32 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was heated to reflux for 30 min. Then, the mixture was washed with satd aq NaHCO<sub>3</sub> (10 mL). The organic layer was dried (MgSO<sub>4</sub>) and concentrated in vacuo, and the residue was purified by FC (2 cm, EtOAc, fractions 2 mL,  $R_{\rm f} = 0.42$ ). Colorless oil: yield 25 mg (34%); IR (neat): ṽ 3346 (vN−H), 2934 (vC−H), 1721 (vC=O), 1622 (vO=C-NH, amide I), 1598 (vC=C), 1537 (δN-H, amide II), 1256, 1113 (vC–O), 698 cm<sup>-1</sup> ( $\gamma$ CH<sub>oop</sub>, arom.); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 2.15–2.27 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH), 2.30–2.40 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH), 3.68– 3.85 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>NH), 4.13 (d, J 13.1 Hz, 1H, 4-H), 4.23 (ddd, J 13.4/7.0/3.8 Hz, 1H, 5-H), 5.49 (d, J 5.8 Hz, 1H, 2-H), 6.62 (s, broad, 1H, NH), 7.52-7.38 (m, 3H, arom. H), 7.48 (d, J 5.5 Hz, 1H, 1-H), 7.75-7.79 (m, 2H, arom. H), a signal for the OHproton was not observed in the spectrum;  $C_{14}H_{15}NO_4$ (261.3). CIMS (isobutane): m/z [%] 262 (MH<sup>+</sup>, 100).

### 4.18. 1,5-Anhydro-7-[2-(3,4-dichlorophenyl)acetylamino]-2,6,7-trideoxy-D-*erythro*-hept-1-en-3-ulose (26)

A solution of acetal 22 (anomeric mixture, 152 mg, 0.36 mmol) and *p*-toluenesulfonic acid monohydrate (60 mg, 0.32 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) was heated to reflux for 45 min. Then, the mixture was washed with satd aq NaHCO<sub>3</sub> (10 mL). The organic layer was dried  $(MgSO_4)$ , concentrated in vacuo, and the residue was purified by FC (2 cm, 95:5 EtOAc-acetone, fractions 2 mL, R<sub>f</sub> 0.23). Pale-yellow solid: mp 142 °C; yield 65 mg (53%); IR (neat):  $\tilde{v}$  3286 (vN–H), 1670 (vO=C– NH, amide I), 1620 (νC=C), 1592 (δN-H, amide II), 1114, 1032 cm<sup>-1</sup> (vC–O); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.03 (dt, J 13.4/6.7 Hz, 1H, CH<sub>2</sub>CH<sub>2</sub>NH), 2.14–2.24 (m, 1H, CH<sub>2</sub>CH<sub>2</sub>NH), 3.46–3.59 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>NH), 3.50 (s, 2H, COCH<sub>2</sub>), 4.00 (d, J 13.1 Hz, 1H, 4-H), 4.09 (ddd, J 13.1/6.7/3.4 Hz, 1H, 5-H), 5.47 (d, J 5.8 Hz, 1H, 2-H), 5.78 (s, broad, 1H, NH), 7.12 (dd, J 8.2/2.1 Hz, 1H, arom. H, 6'-H), 7.29 (d, J 5.8 Hz, 1H, 1-H), 7.38 (d, J 1.8 Hz, 1H, arom. H, 2'-H), 7.42 (d, J 8.2 Hz, 1H, arom. H, 5'-H), a signal for the OH-proton was not observed; EIMS: m/z [%] 343/  $345/347 \text{ [M}^+, 5.8/3.4/0.8], 159/161/163 \text{ [-CH<sub>2</sub>PhCl<sub>2</sub><sup>+</sup>, -CH<sub>2</sub>PhCl<sub>2</sub><sup>+</sup>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCl<sub>2</sub>, -CH<sub>2</sub>PhCL<sub>2</sub>PhCL<sub>2</sub>, -CH<sub>2</sub>PhCL<sub>2$ </sub> 94/60/12; CIMS (isobutane): m/z [%] 344/346/348[MH<sup>+</sup>, 100/65/13], 310/312 [MH<sup>+</sup>-Cl, 67/24], 274  $[M-2 \times Cl, 35]$ ; Anal. Calcd for  $C_{15}H_{15}Cl_2NO_4$ (344.2): C, 52.34; H, 4.39; N, 4.07. Found: C, 52.17; H, 4.68; N, 4.02.

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