The First Regiospecific Synthesis of 8,8-Dimethyl-2*H*,8*H*-pyrano[2,3-*h*]quinolin-2-one and Related Compounds

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The title compound was synthesized in eight steps, starting from 6-acetamido-2,2-dimethyl-2*H*-1-benzopyran-4-one (6). The key steps involved are the regiospecific nitration of 6 and palladium(0)-catalyzed arylation of acrylic acid in aqueous media.

Of the several possible pyranoquinolin-2-one isomers, the angular isomer 1 and the linear isomer 2 are unknown so far. Stimulated by the antithrombotic and antiallergic properties of pyranoquinolin-2-one derivatives, we desired to seek methods for the formation of these two isomers and their analogs. We wish to describe here a highly regiospecific synthesis of the two types of heterocycle 3 and 4, which has led to the preparation of 1 and 2.

Retrosynthetic analysis of 1 and 2 suggested that the chromene derivatives 3 and 4 would be valuable precursors (Scheme 1). Compound 4 had been previously prepared by the nitration of 6-acetamido-2,2-dimethyl-2*H*-1-benzopyran (5). Interestingly, nitration of 5 gave only the 7-nitro compound 4, and the other possible isomer,

the 5-nitro compound 3, was not detectable.² To the best of our knowledge, the 5-nitro compound 3 has not been reported to date. It appeared therefore of interest to develop a regiospecific approach towards 3 in order to synthesize the target molecule 1. In addition, compound 3 can be utilized for the synthesis of a variety of naturally occurring dimethylchromenes.

The present synthetic approach to 3 is based on our observation that the nitration of 6-acetamido-2,2-dimethyl-2H-1-benzopyran-4-one (6), which could be readily prepared in reasonable yield according to the published procedures,³ gave only the 5-nitro compound 7 in 88 % yield (Scheme 2). Surprisingly, the competing orientation effect of acetylamino of compounds 5 and 6 is so different and specific. This fact led to a regiospecific synthesis of 3. Reduction of 7 with sodium borohydride in MeOH followed by dehydration of the chromanol 8 then furnished the key intermediate 3 in 81 % yield for the two steps. The two isomers 3 and 4 were easily distinguished on the basis of aryl protons in the ¹H NMR spectra. The aryl protons of compound 3 displayed two series of double signals (J = 9.0 Hz), whereas the aryl protons of isomer 4 appeared as two singlet signals.

Scheme 1

(a) HNO₃, AcOH; (b) NaBH₄, MeOH; (c) TsOH, Benzene; (d) 5 N HCl, EtOH; (e) i) H₂SO₄ aq., NaNO₂; ii) KI; (f) acrylic acid, Pd(OAc)₂, PPh₃, K₂CO₃, DMF-H₂O; (g) SnCl₂-2H₂O, acetone; (h) 4% HCl

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Deprotection of 3 with 5 N HCl gave the 6-amino-5-nitrochromene (9), which was then converted to the 6-iodo-7-nitrochromene (10) by reaction of the diazonium salt of 9 with KI. Palladium-catalyzed coupling of 10 with acrylic acid in aqueous DMF containing excess K₂CO₃ at 100 °C⁴ afforded nitrocinnamic acid 11 in 80 % yield. While the ferrous sulfate/ammonium hydroxide reduction has been applied frequently in the conversion of o-nitrocinnamic acids to o-aminocinnamic acids, experimental details are complex and the isolated yield of oaminocinnamic acids are very low.5,6 To circumvent the low yield, we employed SnCl₂·2H₂O as a reducing agent so as to obtain an acceptable result. Reduction of 11 with SnCl₂ · 2H₂O in acetone gave aminocinnamic acid 12 in 61 % yield. Cyclization of 12 to 1 was accomplished by heating at reflux in 4% HCl.⁵ In the same way, compound 2 was prepared in 21 % overall yield, starting from 4 (Scheme 3).

Scheme 3

In summary, we have synthesized regiospecifically [2,3-h] and [1,2-b] fused pyranoquinolone heterocycles. This was based on the regiospecific nitration of chromanone 6 and chromene 5, and on the palladium(0)-catalyzed arylation of acrylic acid. The availability of the two novel types of pyranoquinolone heterocycles should assist in the further development of pyranoquinolone derivatives as a biologically important class of compounds.

Solvents were purified by standard methods and dried if necessary. Reagents used were of commercial quality. Petroleum ether used had bp $60-90\,^{\circ}$ C. Melting points were performed in open capillaries and are uncorrected. NMR spectra were recorded on a 300 MHz spectrometer with TMS as internal standard. IR spectra were recorded on a Perkin-Elmer 983 spectrometer. Mass spectra were obtained using EI ionization at 70 eV.

6-Acetamido-2,2-dimethyl-2H-1-benzopyran-4-one (6):

Chromanone 6 was prepared according to the literature method;³ mp 162–164 °C.

IR (KBr): $\nu = 3300$, 2960, 1690, 1660, 1600, 1490, 720 cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.40$ (6 H, s, 2 CH₃), 2.10 (3 H, s, CH₃CO), 2.74 (2 H, s, H-3), 6.90 (1 H, d, J = 9.0 Hz, H-8), 7.76 (1 H, dd, J = 1.7, 9.0 Hz, H-7), 8.05 (1 H, d, J = 1.7 Hz, H-5), 9.30 (1 H, br, NH)

MS: m/z (%) = 233 (M⁺, 18), 135 (100).

6-Acetamido-2,2-dimethyl-5-nitro-2H-1-benzopyran-4-one (7):

To a stirred solution of 6 (11.4 g, 0.049 mol) in glacial HOAc (200 mL) at 0 °C was added dropwise a solution of fuming HNO₃ (5.88 mL, 0.061 mol) in glacial HOAc. After stirring for an additional 45 min at 0 °C, the solution was poured onto ice, and the precipitate was collected (12.0 g, 88 %). Recrystallization of a small portion from EtOH gave 7 as pale yellow needles; mp 199–201 °C. IR (KBr): v = 3220, 2990, 1695, 1660, 1550 cm⁻¹.

¹H NMR (CDCl₃): $\delta = 1.49$ (6 H, s, 2 CH₃), 2.16 (3 H, s, COCH₃),

2.78 (2 H, s, H-3), 7.09 (1 H, d, J = 9.2 Hz, H-7), 7.59 (1 H, br, NH), 8.07 (1 H, d, J = 9.2 Hz, H-8).

Anal. Calcd for $C_{13}H_{14}N_2O_5$: C, 56.11, H, 5.07, N, 10.07; Found: C, 56.01, H, 5.11, N, 9.99.

6-Acetamido-3,4-dihydro-2,2-dimethyl-5-nitro-2*H*-1-benzopyran-4-ol (8):

NaBH₄ (1.6 g, 0.042 mol) was added to a stirred suspension of 7 (10.7 g, 0.038 mol) in MeOH (100 mL) at 0 °C and the mixture was stirred at this temperature for 30 min. After stirring for an additional 24 h at r.t., 2 M HCl was added and the product was extracted into EtOAc. The extract was dried (MgSO₄) and evaporated under reduced pressure and the residue was chromatographed on silica gel eluting with EtOAc/petroleum ether (1:1) to give 8 (9.6 g, 90 %) as a yellow solid; mp 175–177°C.

IR (KBr): $\nu = 3301$, 2954, 1674, 1582, 1533, 1482, 1373 cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.35$ (3 H, s, CH₃), 1.48 (3 H, s, CH₃), 2.05 (2 H, m, H-3), 2.17 (3 H, s, COCH₃), 2.60 (1 H, br, OH), 5.14 (1 H, m, H-4), 6.99 (1 H, d, J = 8.8 Hz, H-7), 7.83 (1 H, d, J = 8.8 Hz, H-8), 8.06 (1 H, br, NH).

MS: m/z (%) = 280 (M⁺, 46), 238 (51), 205 (48), 164 (62). Anal. Calcd for $C_{13}H_{16}N_2O_5$: C, 55.71, H, 5.75, N, 9.99; Found: C, 55.71, H, 5.84, N, 10.08.

6-Acetamido-2,2-dimethyl-5-nitro-2H-1-benzopyran (3):

A mixture of 8 (11.1 g, 0.040 mol) and TsOH (0.75 g, 0.004 mol) in toluene (120 mL) was refluxed under N_2 for 10 h until the dehydration was complete. The mixture was then cooled, filtered, and evaporated. The residue was chromatographed on silica gel eluting with EtOAc/petroleum ether (1:4) to give 3 (8.61 g, 83%) as a yellow solid; mp 146–147°C.

¹H NMR (CDCl₃): δ = 1.48 (6 H, s, 2 CH₃), 2.18 (3 H, s, COCH₃), 5.85 (1 H, d, J = 10.2 Hz, H-3), 6.42 (1 H, d, J = 10.2 Hz, H-4), 6.99 (1 H, d, J = 9.0 Hz, H-7), 7.87 (1 H, d, J = 9.0 Hz, H-8), 8.30 (1 H, br, NH).

MS: m/z (%) = 262 (M⁺, 59).

Calcd for $C_{13}H_{14}N_2O_4$: C, 59.54; H, 5.38; N; 10.86. Found: C, 59.43; H, 5.27; N; 10.60.

6-Amino-2,2-dimethyl-5-nitro-2H-1-benzopyran (9):

A solution of 3 (16.2 g, 0.062 mol) in EtOH (180 mL) was refluxed with 5 N HCl for 6 h. The red solution was cooled and poured into $\rm H_2O$, and the red crystals that precipitated were filtered; yield: 13.6 g (95%). Recrystallization of a small portion from EtOH gave 9 as red needles; mp 90–92°C.

IR (KBr): v = 3485, 3368, 2965, 1617, 1505, 1332 cm⁻¹.

¹H NMR (CDCl₃): δ = 1.41 (6 H, s, 2 CH₃), 5.00 (2 H, br, NH₂), 5.74 (1 H, d, J = 9.9 Hz, H-3), 6.62 (1 H, d, J = 8.7 Hz, H-7), 6.71 (1 H, d, J = 9.9 Hz, H-4), 6.89 (1 H, d, J = 8.7 Hz, H-8).

MS: m/z (%) = 220 (M⁺, 36), 205 (100).

Anal. Calcd for $C_{11}H_{12}N_2O_3$; C, 59.99; H; 5.49; N; 12.72. Found: C, 59.88; H; 5.41; N, 12.70.

2,2-Dimethyl-6-iodo-5-nitro-2*H*-1-benzopyran (10):

The nitroamine 9 (5.6 g, 0.025 mol) was dissolved in a mixture of $\rm H_2SO_4$ (45 mL) and $\rm H_2O$ (110 mL) with warming, then cooled to 0°C, and treated dropwise with stirring with a solution of NaNO₂ (1.93 g, 0.028 mol) in $\rm H_2O$ (10 mL). The mixture was stirred for an additional 1 h at 0°C, and then addition of a solution of KI (5.8 g, 0.035 mol) in $\rm H_2O$ (20 mL) followed. The mixture was stirred overnight at r.t. Extraction with $\rm Et_2O$ gave a red gum, which was chromatographed on silica gel eluting with petroleum ether to give 10 (6.7 g, 81 %) as a yellow oil.

IR (KBr): v = 2980, 1637, 1593, 1536, 1357, 1293, 1280 cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.45$ (6 H, s, 2 CH₃), 5.78 (1 H, d,

¹H NMR (CDCl₃): $\delta = 1.45$ (6 H, s, 2 CH₃), 5.78 (1 H, d, J = 10.1 Hz, H-3), 6.19 (1 H, d, J = 10.1 Hz, H-4), 6.68 (1 H, d, J = 8.7 Hz, H-7), 7.56 (1 H, d, J = 8.7 Hz, H-8).

MS: m/z (%) = 331 (M⁺, 26), 316 (100), 143 (49).

3-[2,2-Dimethyl-5-nitro-2*H*-1-benzopyran-6-yl]acrylic Acid (11):

A mixture of 10 (1.02 g, 3.08 mmol), Pd(OAc)₂ (20 mg, 0.089 mmol), PPh₃ (46 mg, 0.18 mmol), K_2CO_3 (2.12 g, 15.4 mmol), acrylic acid (0.4 g, 5.6 mmol), DMF (10 mL) and H_2O (2 mL) was heated under N_2 at 100 °C for 12 h, and then cooled, diluted with H_2O (20 mL), and extracted with Et₂O. Acidification of the aqueous layer gave a yellow solid, which was chromatographed on silica gel eluting with petroleum ether/EtOAc (3:1) to give 11 (0.68 g, 80 %) as a yellow solid; mp 210–212 °C.

IR (KBr): v = 2500-3200, 1693, 1628, 1604, 1565, 1531, 1478, 1365, 1295, 1275, 1203, 1120, 811 cm⁻¹.

¹H NMR (CDCl₃): δ = 1.48 (6 H, s, 2 CH₃), 5.83 (1 H, d, J = 10.1 Hz, H-3′), 6.25 (1 H, d, J = 10.1 Hz, H-4′), 6.35 (1 H, d, J = 15.7 Hz, H-3), 6.93 (1 H, d, J = 8.6 Hz, H-7′), 7.47 (1 H, d, J = 8.6 Hz, H-8′), 7.58 (1 H, d, J = 15.7 Hz, H-2), 12.59 (1 H, br, CO₂H).

MS: m/z (%) = 275 (M⁺, 23), 260 (92), 214 (100), 169 (32). Anal. Calcd for $C_{14}H_{13}NO_5$: C, 61.09; H, 4.76; N, 5.09. Found: C, 61.13; H, 4.79; N, 4.96.

3-[5-Amino-2,2-dimethyl-2*H*-1-benzopyran-6-yl]acrylic Acid (12):

A solution of 11 (5.0 g, 0.018 mol) and $SnCl_2 \cdot 2H_2O$ (16.7 g, 0.072 mol) in acetone (150 mL) was refluxed for 6 h. After evaporation of solvent, the residue was basified with 28 % aq ammonia (50 mL), and filtered through a layer of Celite. The precipitate was washed with H_2O (100 mL) and the combined filtrate was acidified to pH 5 with concentrated HCl to give a light brown solid. The crude product was purified by reprecipitation from aqueous ammonia to give 12 (2.72 g, 61 %) as a golden yellow crystalline solid; mp 178–179 °C.

IR (KBr): $v = 3467, 3393, 2972, 1667, 1608, 1581, 1478, 1202 \text{ cm}^{-1}.$ ¹H NMR (DMSO- d_6): $\delta = 1.35$ (6 H, s, 2 CH₃), 5.58 (1 H, d, J = 9.6 Hz, H-3′), 6.05 (1 H, d, J = 9.6 Hz, H-4′), 6.10 (1 H, d, J = 15.5 Hz, H-3), 6.80 (1 H, d, J = 8.8 Hz, H-8′), 7.20 (1 H, d, J = 8.8 Hz, H-7′), 7.77 (1 H, d, J = 15.5 Hz, H-2).

MS: m/z (%) = 245 (M⁺, 46), 230 (100), 212 (35), 186 (44).

Anal. Calcd for $C_{14}H_{15}NO_3 \cdot 0.25H_2O$: C, 67.32; H, 6.25; N, 5.61. Found: C, 67.37; H, 6.03; N, 5.36.

8,8-Dimethyl-2H,8H-pyrano[2,3-h]quinolin-2-one (1):

A mixture of 12 (0.7 g, 2.86 mmol) and 4% aq HCl (30 mL) was refluxed for 1.5 h. Cooling and filtration gave a crude product, which was purified by chromatography on silica gel eluting with EtOAc/petroleum ether (1:2) to give 1 (0.47 g, 72%) as a white solid; mp 244–245°C.

IR (KBr): $v = 3160, 3050, 1643, 1604, 1555, 1258, 1198, 1123 \text{ cm}^{-1}.$ ¹H NMR (DMSO- d_6): $\delta = 1.40$ (6 H, s, 2 CH₃), 5.78 (1 H, d, J = 10.1 Hz, H-9), 6.32 (1 H, d, J = 9.5 Hz, H-4), 6.65 (1 H, d, J = 8.5 Hz, H-6), 7.21 (1 H, d, J = 10.1 Hz, H-10), 7.45 (1 H, d, J = 8.5 Hz, H-5), 7.81 (1 H, d, J = 9.5 Hz, H-3), 11.29 (1 H, br, NH). MS: m/z (%) = 227 (M⁺, 24), 212 (100), 184 (6).

Anal. Calcd for $C_{14}H_{13}NO_2$: C, 73.99; H, 5.77; N, 6.16. Found: C, 73.80; H, 5.54; N, 6.12.

2,2-Dimethyl-6-iodo-7-nitro-2*H*-1-benzopyran (13):

Compound 13^7 was prepared from 4^2 following the procedure described for 10, to give 13 (2 steps, 76%) as a yellow oil.

IR (KBr): $v = 3095, 2979, 2104, 1640, 1564, 1525, 1473, 1355, 1337, 889 \text{ cm}^{-1}$.

¹H NMR (CDCl₃): δ = 1.45 (6 H, s, 2 CH₃), 5.82 (1 H, d, J = 10 Hz, H-3), 6.28 (1 H, d, J = 10 Hz, H-4), 7.32 (1 H, s, H-8), 7.57 (1 H, s, H-5).

MS: m/z (%) = 331 (M⁺, 32), 316 (100), 270 (19), 143 (64).

3-[2,2-Dimethyl-7-nitro-2*H*-1-benzopyran-6-yl]acrylic Acid (14):

Compound 14 was prepared from 13 following the procedure described for 11, to give 14 (87%) as a yellow solid; mp 225-227°C. IR (KBr): v = 2500-3200, 1693, 1629, 1604, 1570, 1516, 1484, 1412, 1336, 1281, 1262, 1216, 1196, 977, 765 cm⁻¹.

¹H NMR (DMSO- d_6): δ = 1.42 (6 H, s, 2 CH₃), 6.03 (1 H, d, J = 9.9 Hz, H-3′), 6.46 (1 H, d, J = 15.7 Hz, H-3), 6.55 (1 H, d, J = 9.9 Hz, H-4′) 7.37 (1 H, s, H-8′), 7.72 (1 H, s, H-5′), 7.76 (1 H, d, J = 15.7 Hz, H-2), 12.60 (1 H, br, CO₂H).

MS: m/z (%) = 275 (M⁺, 12), 260 (100), 229 (42), 214 (47), 185 (36), 169 (47).

Anal. Calcd for $C_{14}H_{13}NO_5$: C, 61.09; H, 4.76; N, 5.09; Found: C, 61.33; H, 4.70; N, 5.06.

3-[7-Amino-2,2-dimethyl-2*H*-1-benzopyran-6-yl|acrylic Acid (15):

Compound 15 was prepared from 14 following the procedure described for 12, to give 15 (58%) as a golden yellow crystalline solid; mp 166-168°C.

IR (KBr): v = 3362, 3047, 2976, 1697, 1635, 1614, 1496, 1369, 1132 cm⁻¹.

¹H NMR (DMSO- d_6): δ = 1.33 (6 H, s, 2 CH₃), 5.49 (1 H, d, J = 9.8 Hz, H-3′), 6.07 (1 H, s, H-8′), 6.12 (1 H, d, J = 15.6 Hz, H-3), 6.28 (1 H, d, J = 9.8 Hz, H-4), 7.18 (1 H, s, H-5′), 7.67 (1 H, d, J = 15.6 Hz, H-2).

MS: m/z (%) = 245 (M⁺, 26), 230 (100), 212 (65).

Anal. Calcd for $C_{14}H_{15}NO_3 \cdot 0.25H_2O$: C, 67.32; H, 6.25; N, 5.61. Found: C, 67.35; H, 6.28; N, 5.53.

8,8-Dimethyl-2*H*,8*H*-pyrano[1,2-*b*]quinolin-2-one (2):

Compound 2 was prepared from 15 following the procedure described for 1, to give 2 (70%) as a white solid; mp 236-238°C.

IR (KBr): $v = 3138, 2980, 1650, 1629, 1567, 1468, 1267, 1154 \text{ cm}^{-1}$. ¹H NMR (DMSO- d_6): $\delta = 1.40$ (6 H, s, H-8), 5.80 (1 H, d, J = 9.9 Hz, H-7), 6.29 (1 H, d, J = 9.6 Hz, H-4), 6.47 (1 H, d, J = 9.9 Hz, H-6), 6.64 (1 H, s, H-10), 7.32 (1 H, s, H-5), 7.75 (1 H, d, J = 9.6 Hz, H-3), 11.58 (1 H, br, NH).

MS: m/z (%) = 227 (M⁺, 19), 212 (100), 184 (5).

Anal. Calcd for $C_{14}H_{13}NO_2$: C, 73.99; H, 5.77; N, 6.16; Found: C, 73.82; H, 5.80; N, 5.98.

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