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Chemoenzymatic synthesis of [3,9-¹³C]-labeled NeuAc and KDN

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Abstract—The chemoenzymatic synthesis of ¹³C-labeled sialic acid (NeuAc) and 3-deoxy-D-*glycero*-D-*galacto*-2-nonulosonic acid (KDN) as useful molecular probes for studying the conformation of sialyl or KDN oligosaccharides attached to proteins was performed by using [6⁻¹³C]-ManNAc, [6⁻¹³C]-Man and [3⁻¹³C]-pyruvic acid sodium salt. In the synthesis of the compounds, 5,6-anhydro intermediates were found to easily provide not only 6⁻¹³C-labeled but also 5-, and 6-modified NeuAc and KDN analogs. Furthermore, it was demonstrated that identical results are obtained by NMR for both [3,9⁻¹³C]-NeuAc and 1:1 mixtures of [3⁻¹³C]-NeuAc. © 2003 Elsevier Science Ltd. All rights reserved.

Sialyl and KDN¹ glycoconjugates are important cell surface components active in a variety of intercellular recognition events. Therefore, a study of the conformation and dynamics of these sialyl and KDN oligosacharides and their analogs should provide useful insights into how these cell surface oligosaccharides interact with the corresponding receptor molecules. However, although the conformational properties^{2,3,4a-c} of sialyloligosaccharide analogs of low molecular weight have been reported by many research groups, the conformation and dynamics of sialyl-oligosaccharides and their analogs attached to glycoproteins have not yet been fully analyzed. To address this problem, ¹³C-labeled NeuAc has been utilized for the conformational analysis of sialyloligosaccharides on artificial membrane surfaces^{5a,b} and for TRNOE experiments.⁶ On the other hand, Kajihara et al. developed a concise [3-¹³C]-labeling method for NeuAc analogs, and reported their preliminary results² for the ¹³C-labeled sialyloligosaccharide, 9-deoxy-9-fluoro-[3-13C]-NeuAc- α -(2-6)-[U-¹³C]-Gal- β -(1-4)-GalNAc- β -sequence of an intact glycoprotein using a CMP-[3-13C]-NeuAc analog and UDP-[U-¹³C]-Glc, and the corresponding NMR analysis. In this work, they were not able to observe the H-7 proton of 9-deoxy-9-fluoro-[3-13C]-NeuAc because the HMQC-TOCSY development stopped at H-6. In 2000, they reported novel NMR techniques,³ HSQC-

TOCSY–NOESY–TOCSY, for observation of all protons of NeuAc, H-3 to H-9, even with only a single ¹³C-atom, at the 3-position of NeuAc, bound to a glycoprotein. However, since combined multi-pulse techniques generally suffer from low sensitivity, we have synthesized [3,9-¹³C]-labeled NeuAc for convenient observation of all protons of NeuAc from H-3 to H-9 by the HMQC–HOHAHA technique. Further, ¹³C-labeled KDN was also synthesized by the same strategy.

In this paper, we describe a novel [3,9-¹³C]-labeling method for NeuAc and KDN via 5,6-anhydro-Man-NAc and -Mannose. Another goal of the research described herein is to demonstrate the synthetic merits of mixed samples of [3-¹³C]- and [9-¹³C]-NeuAc as opposed to [3,9-¹³C]-NeuAc. This is advantageous because the mixed samples result in higher yields due to the use of excess amounts of the non-labeled substrate. Equal measurements for di-labeled NeuAc and mixed mono-labeled NeuAc by NMR are also demonstrated.

Synthesis of [3,9-¹³C]-sialic acid (NeuAc)

The synthesis of $[3,9^{-13}C]$ -NeuAc was achieved as follows. 2-Acetamido-2-deoxy-D-mannose (D-ManNAc) was treated with acetone dimethyl acetal and a catalytic amount of *p*-toluenesulfonic acid in dry acetone at rt. to give the corresponding syrupy di-isopropylydene derivative 1 in 71% yield (Scheme 1). The partial hydrolysis of 1 in 70% aq. AcOH at rt. gave the

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Scheme 1.

corresponding 2,3-*N*,*O*-isopropylidene derivative (4: mp 101–102°C) in 96% yield. Compound **4** was treated with NaIO₄ in MeOH at 0°C to give the carbonyl compound (6: mp 74.0–76.0°C) in 98% yield. Wittig reaction of **6** with PPh₃¹³CH₃I and *n*-BuLi in abs. THF gave the corresponding ¹³C-labeled olefin compound (8: mp 66.5–67.0°C) in 72% yield. ¹³C-Labeled phosphonium salt PPh₃¹³CH₃I was conveniently synthesized by using ¹³CH₃I and triphenylphosphine in toluene in almost quantitative yield. Peracid oxidation of **8** with *m*-CPBA in 1,2-dichloroethane at 70°C gave a mixture of the corresponding 5,6-anhydro derivatives (*manno*-

10: mp 94.5–95.5°C and *gulo*-11: mp 103.5–104.0°C) in 68 and 29% yields, respectively. The required compound 10 was hydrolyzed with 0.5 M aq. KOH in THF (1:1 v/v) at 75°C to give the labeled 5,6-di-ol derivative 14 in 89% yield. The stereochemistry of 10 was confirmed at this stage by comparing the NMR data of 4 and 14. It is also possible to convert compound 11 into 10 (80% yield) via the 6-*O*-benzoyl-5-*O*-methanesulfonyl intermediate in order to save the labeled compound. Thus, compound 11 was hydrolyzed under reflux conditions (0.5 M aq. KOH/THF) to give a 5,6-di-ol derivative in 85% yield. The di-ol derivative was selectively acylated (91% yield) and sulfonated (quantitative yield) to give 6-*O*-benzoyl-5-*O*-methylsulfonyl derivative in good yield. It was then treated with Bu_4NOH/THF at rt to give 10 in 92% yield. 6-¹³C-labeled D-ManNAc (16) was obtained in 85% yield by the hydrolysis of 14 with Dowex 50W (H⁺) in H₂O. Finally, [3,9-¹³C]-NeuAc (18) was synthesized by the enzymatic reaction⁷ of 16 and 2.5 equiv. [3-¹³C]-pyruvic acid sodium salt in a solution of phosphate buffer (pH 7.5) containing BSA, NaN₃, and NeuAc aldolase for 16 h at 37°C in 70% yield.

Synthesis of [3-¹³C]-sialic acid (NeuAc)

3-¹³C-labeled NeuAc compounds were synthesized by condensation of ¹³C-labeled pyruvic acid with an excess of ManNAc.^{5a} Therefore, **20** was synthesized in a similar manner to that described above using [3-¹³C]-pyruvic acid sodium salt and an excess amount (5.0 equiv.) of non-labeled ManNAc. The yield of this reaction was 82%.

Synthesis of [9-¹³C]-sialic acid (NeuAc)

 $[9^{-13}C]$ -NeuAc (21) was synthesized in a similar manner to that described above using $[9^{-13}C]$ -ManNAc and an excess amount (8.5 equiv.) of pyruvic acid sodium salt. The yield of this reaction was 76%.

Synthesis of [3,9-¹³C]-3-deoxy-D-glycero-D-galacto-2nonulosonic acid (KDN)

[3.9-¹³C]-KDN was synthesized as follows. D-Mannose was treated with acetone dimethyl acetal and a catalytic amount of *p*-toluenesulfonic acid in dry acetone at rt to give the corresponding di-isopropylidene derivative 2 in 82% yield. Then 2 was treated with sodium hydride and benzyl bromide to give the corresponding benzyl α -glycoside (3: mp 56–57°C) and β anomer of 3 in 69 and 27% yields, respectively. In a similar manner as mentioned in the synthesis of [3,9-¹³C]-NeuAc, compounds 5–19 were synthesized as fol-The partial hydrolysis of **3** gave the lows. corresponding 2,3-O-isopropylidene derivative (5: mp 62-63°C) in quantitative yield. Compound 5 was converted into the syrupy carbonyl compound 7 in 96% yield. Wittig reaction of 7 gave the corresponding 6-¹³C-labeled compound 9 in 69% yield. Peracid oxidation of 9 gave a mixture of the corresponding 5,6-anhydro derivatives (manno-12: syrup and gulo-13: mp 85-86°C) in 67 and 29% yields, respectively. The required compound 12 was hydrolyzed under alkaline conditions to give the labeled 5,6-di-ol derivative 15 in 93% yield. The stereochemistry of 12 was confirmed at this stage by comparing the NMR data of 5 and 15. It is also possible to convert compound 13 into 12 in 80% yield (four steps). 6-13C-labeled Dmannose (17) was obtained in 93% yield by the hydrolysis of **15**. Finally, $[3,9^{-13}C]$ -KDN (**19**) was synthesized by enzymatic reaction⁷ of **17** under similar conditions as mentioned in the synthesis of the labeled NeuAc using the same NeuAc aldolase, in 70% yield.

Synthesis of [3-¹³C]-KDN

 $[3-{}^{13}C]$ -KDN (22) was synthesized in a similar way as mentioned above by the use of $[3-{}^{13}C]$ -pyruvic acid sodium salt and an excess amount (5.0 equiv.) of non-labeled ManNAc. The yield of this reaction was 87%.

Synthesis of [9-¹³C]-KDN

 $[9^{-13}C]$ -KDN (23) was synthesized in a manner analogous to that described above using $[9^{-13}C]$ -Man-NAc and an excess amount (8.5 equiv.) of pyruvic acid sodium salt. The yield of this reaction was 78%.

NMR data for each ¹³C-labeled compound **18–23** are given in Ref. 8. NMR data demonstrating identical results for the HMQC–HOHAHA spectra of [3,9-¹³C]-NeuAc and a 1:1 mixture of $[3-^{13}C]$ - and $[9-^{13}C]$ -NeuAc are shown in Figures 1, 2, and 3. It is possible to analyze the chemical shifts and *J* values of all protons by this technique.

As described above, a practical synthesis of minimal ¹³C-labeled NeuAc and KDN should facilitate studies on the conformational properties and dynamic behavior of oligosaccharides that contain NeuAc and KDN. This minimal and efficient labeling method may also be extended to other important component monosaccharides, such as mannose, mannosamine, galactose, and fucose which are currently difficult to study due to ${}^{3}J=0$ coupling constants.



Figure 1. HMQC-HOHAHA spectrum of [3,9-¹³C]-NeuAc.

ppm

4.0	3.5 4	3.0	2.5	2.0	
H-6 #44 # H-4 H-5					
					th
•					
H-8 (4 • H-9	H-9' 4 () H-7				+c s ⊧s

Figure 2. HMQC-HOHAHA spectrum of a 1:1 mixture of $[3^{-13}C]$ - and $[9^{-13}C]$ -NeuAc.



Figure 3. HMQC–HOHAHA spectrum of a 1:1 mixture of $[3^{-13}C]$ - and $[9^{-13}C]$ -NeuAc (expansion).

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- 8. ¹H and ¹³C NMR data for ¹³C-labeled compounds 18–23: **18**: ¹H NMR (D₂O, 500 MHz) 3.95 (1H, dddd, $J_{4,3ax} = 11.9$ H, $J_{4,3eq} = 5.0$ Hz, $J_{4,5} = 10.4$ Hz, $J_{4,C} = 1.6$ Hz, H-4), 3.93 (1H, dd, $J_{6,5}$ =10.4 Hz, $J_{6,7}$ =1.0 Hz, H-6), 3.80 (1H, dd, H-5), 3.71 (1H, ddd, $J_{9,8} = 2.8$ Hz, $J_{9,9'} = 11.9$ Hz, $J_{9,C} =$ 144.8 Hz, H-9), 3.62 (1H, dddd, J_{9',8}=6.4 Hz, H-8), 3.50 (1H, ddd, $J_{9',C} = 143.8$ Hz, H-9'), 3.43 (1H, dd, $J_{7,8} = 9.2$ Hz, H-7), 2.19 (1H, ddd, $J_{3eq,3ax} = 12.4$ Hz, $J_{3eq,C} = 132.4$ Hz, H-3eq), 1.92 (3H, s, -NHAc), 1.76 (1H, dddz, J_{3ax,C}= 130.0 Hz, H-3ax); ¹³C NMR (D₂O, 125 MHz) 64.12 (C-9), 40.20 (C-3); 19: ¹H NMR (D₂O, 500 MHz) 3.88 (1H, dddd, $J_{4,3ax} = 11.6$ Hz, $J_{4,3eq} = 4.9$ Hz, $J_{4,5} = 9.5$ Hz, $J_{4,C} = 2.1$ Hz, H-4), 3.84 (1H, dd, $J_{6,5} = 9.5$ Hz, $J_{6,7} = 1.2$ Hz, H-6), 3.78 (1H, ddd, $J_{9,C}$ =143.1 Hz, $J_{9,8}$ =2.8 Hz, $J_{9,9'}$ = 11.6 Hz, H-9), 3.75 (1H, dd, J_{7.8}=8.9 Hz, H-7), 3.67 (1H, ddd, $J_{8,9'} = 6.4$ Hz, H-8), 3.57 (1H, ddd, $J_{9',C} = 142.8$ Hz, H-9'), 3.49 (1H, dd, H-5), 2.10 (1H, ddd, J_{3eq,3ax}=13.0 Hz, $J_{3eq,C} = 132.4$ Hz, H-3eq), 1.66 (1H, ddd, $J_{3ax,C} = 128.8$ Hz, H-3ax); ¹³C NMR (D₂O, 125 MHz) 63.5 (C-9), 39.25 (C-3); 20: ¹H NMR (D₂O, 500 MHz) 3.91 (1H, dddd, $J_{4,3ax} = 11.9$ Hz, $J_{4,3eq} = 4.6$ Hz, $J_{4,5} = 10.4$ Hz, $J_{4,C} = 1.6$ Hz, H-4), 3.89 (1H, dd, $J_{6.5} = 10.4$ Hz, $J_{6.7} = 1.0$ Hz, H-6), 3.80 (1H, dd, H-5), 3.73 (1H, dd, $J_{9.8} = 2.4$ Hz, $J_{9.9'} = 11.9$ Hz, H-9), 3.65 (1H, ddd, J_{8.7}=9.2 Hz, J_{8.9}=6.4 Hz, H-8), 3.51 (1H, dd, H-9'), 3.42 (1H, br dd, H-7), 2.12 (1H, ddd, $J_{3eq,3ax} = 12.8$ Hz, $J_{3eq,C} = 132.8$ Hz, H-3eq), 1.95 (3H, s, -NHAc), 1.73 (1H, ddd, $J_{3ax,C}$ =129.4 Hz, H-3ax); ¹³C NMR (D₂O, 125 MHz) 39.67 (C-3); 21: ¹H NMR (D₂O, 500 MHz) 3.92 (1H, ddd, $J_{4,3ax} = 10.0$ Hz, $J_{4,3eq} = 4.9$ Hz, $J_{4,5} = 10.1$ Hz, H-4), 3.88 (1H, dd, $J_{6,5} = 10.4$ Hz, $J_{6,7} = 1.0$ Hz, H-6), 3.81 (1H, dd, H-5), 3.73 (1H, ddd, J_{9.8}=2.8 Hz, $J_{9,9'} = 11.9$ Hz, $J_{9,C} = 144.0$ Hz, H-9), 3.66 (1H, ddd, H-8), 3.50 (1H, ddd, $J_{9',8} = 6.4$ Hz, $J_{9',C} = 138.9$ Hz, H-9'), 3.41 (1H, br dd, J_{7.8}=9.2 Hz, H-7), 2.11 (1H, dd, J_{3eq.3ax}=13.1 Hz, H-3eq), 1.95 (3H, s, -NHAc), 1.73 (1H, dd, H-3ax); 13 C NMR (D₂O, 125 MHz) 64.11 (C-9); **22**: ¹H NMR (D₂O, 500 MHz) 3.83 (1H, dddd, $J_{4,3ax}$ =12.0 Hz, $J_{4,3eq}$ = 5.2 Hz, $J_{4,5}=9.7$ Hz, $J_{4,C}=2.0$ Hz, H-4), 3.79 (1H, dd, $J_{6.5} = 9.7$ Hz, $J_{6.7} = 1.2$ Hz, H-6), 3.68 (1H, dd, $J_{9.8} = 2.8$ Hz, J_{9.9} = 11.5 Hz, H-9), 3.70 (1H, dd, J_{7,8} = 8.9 Hz, H-7), 3.65 (1H, ddd, J_{8.9'}=6.3 Hz, H-8), 3.50 (1H, dd, H-9'), 3.43 (1H, dd, H-5), 2.05 (1H, ddd, $J_{3eq,3ax} = 13.2$ Hz, $J_{3eq,C} = 132.8$ Hz, H-3eq), 1.61 (1H, ddd, $J_{3ax,C} = 129.4$ Hz, H-3ax); ¹³C NMR (D₂O, 125 MHz) 39.67 (C-3); 23: ¹H NMR (D₂O, 500 MHz) 3.82 (1H, ddd, $J_{4,3ax}$ =12.0 Hz, $J_{4,3eq} = 5.2$ Hz, $J_{4,5} = 9.7$ Hz, H-4), 3.78 (1H, dd, $J_{6,5} = 9.7$ Hz, $J_{6,7} = 1.1$ Hz, H-6), 3.64 (1H, ddd, $J_{9,C} = 146.1$ Hz, $J_{9,8} = 2.5$ Hz, $J_{9,9'} = 12.0$ Hz, H-9), 3.71 (1H, dd, $J_{7,8} = 8.5$ Hz, H-7), 3.62 (1H, ddd, J_{8,9}=6.3 Hz, H-8), 3.50 (1H, ddd, J_{9',C}=140.7 Hz, H-9'), 3.43 (1H, dd, H-5), 2.02 (1H, dd, $J_{3eq,3ax} = 13.2$ Hz, H-3eq), 1.62 (1H, dd, H-3ax); ¹³C NMR (D₂O, 125 MHz) 63.8 (C-9).