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Palladium-Catalyzed C-H Bond Functionalization Reactions Using Phosphate/Sulfonate Hypervalent Iodine Reagents

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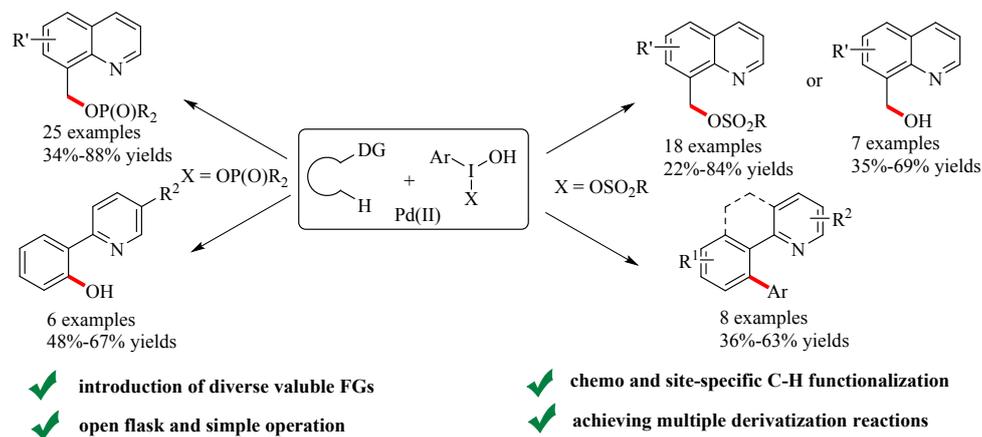
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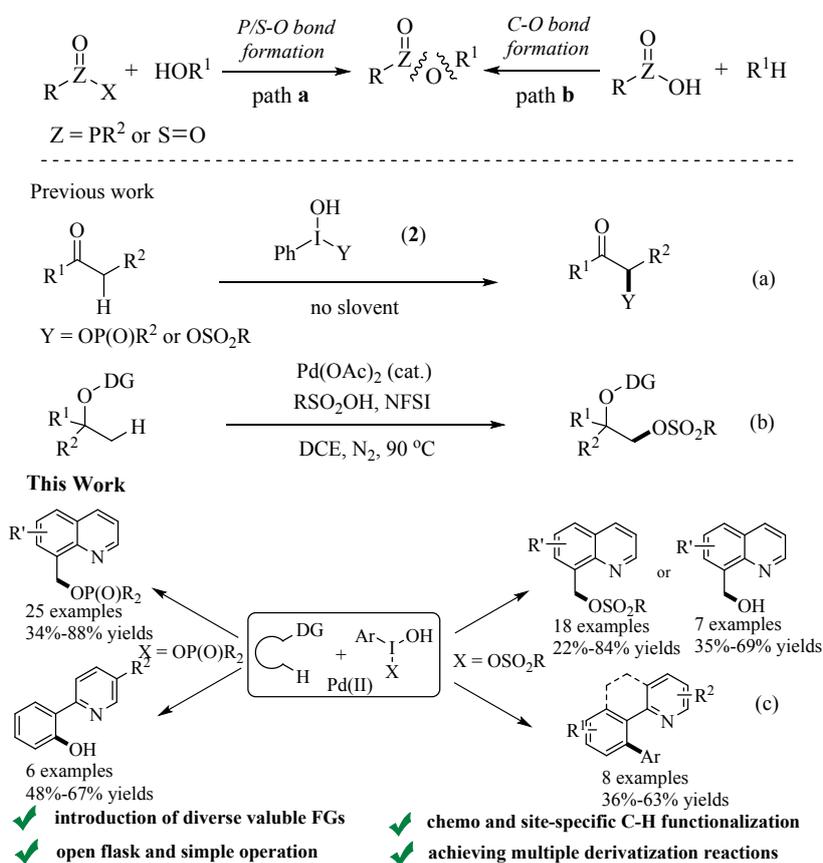
ABSTRACT: A new and operationally simple approach for palladium-catalyzed C-H functionalization reactions utilizing an organophosphorus/sulfonate hypervalent iodine reagent as both an oxidant and the source of a functional group has been developed. Through this method, the oxidative phosphorylation, sulfonation and hydroxylation unactivated benzyl C(sp³)-H bonds, along with the hydroxylation and arylation of aryl C(sp²)-H bonds, are successfully realized under mild conditions and with excellent site-selectivity. The versatile C-OSO₂R bond provides a platform for a wide array of subsequent diversification reactions.

INTRODUCTION

Phosphates and sulfonates, not only widely exist in a variety of important biologically active molecules (e.g., nucleic acids, proteins, carbohydrates, lipids, coenzymes, and steroids),¹ but also have extensive applications in the fields of pharmaceutical chemistry² and agrochemicals³ as well as analytical chemistry.⁴ In addition, these key structures are also useful in organic synthesis,⁵ for example, sulfonates are often utilized as the synthons for S_N2 reactions. For

these reasons, the development of simple and effective synthetic methods for their preparation is of great importance. Historically, the primary means of introducing these important functional groups involve P/S-O bond formation, which is typically accomplished by the nucleophilic substitution of an appropriate phosphoryl/sulfonyl-GLG (GLG: good leaving group) moiety with a suitable alcohol or phenol (Scheme 1, path a).^{6,7} However, this strategy suffers from some drawbacks, including the use of tedious multistep procedures, harsh reaction conditions and hazardous phosphoryl halides. A more convenient method to access these useful structures involving the direct transformation of an unactivated C-H bond into a C-OP(O)R₂/OSO₂R bond is highly desirable (Scheme 1, path b).⁸ While metal-free α -phosphorylation and α -sulfonation reactions of carbonyl compounds mediated by the corresponding phosphate and sulfonate hypervalent iodine reagents were successively reported by Togo's group and Wirth's group in 2002 and 2005 (Scheme 1a),⁹ the substrate scopes were limited to the ketones. Subsequently in 2015, Dong and co-workers described an elegant approach for the synthesis of β -sulfonyloxyated alcohols through a palladium-catalyzed regioselective C(sp³)-H bond functionalization of masked alcohols (Scheme 1b).¹⁰ However, the reaction was performed under harsh reaction conditions. Hence, the realization of a broadly applicable C-H bond phosphorylation and sulfonation in a mild and atom-economic manner is still urgently needed.

Scheme 1 Synthetic strategies for accessing sulfonates/phosphates.



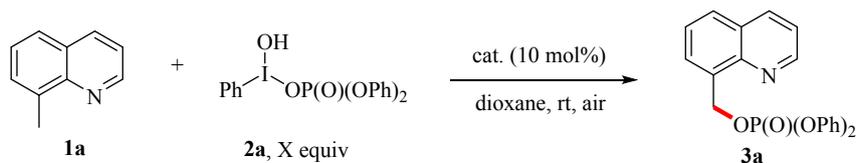
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4 On the other hand, in the past few decades, significant advances have been made in catalytic
5 C-H oxygenations, which have successfully introduced a series of valuable oxygen groups such
6 as OH,¹¹ OCOR,¹² and OR.¹³ In contrast to these strongly nucleophilic groups, however, the
7 installation of weakly nucleophilic and good leaving groups, such as OP(O)R₂ and OSO₂R, to a
8 specific position still faces a number of challenges. First, the complexity of such a catalytic
9 system and the exceedingly low nucleophilicity of OP(O)R₂ and OSO₂R anions, make the
10 proposed phosphorylation and sulfonation likely to suffer from competition from other
11 potential nucleophiles. Second, due to the high reactivity of the target phosphates and
12 sulfonates, product stability during the reaction must be considered and will necessitate
13 relatively mild and simple catalytic systems.
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16 Herein, we reveal a new and efficient approach for the direct, palladium-catalyzed, benzyl C(sp³)-H
17 phosphorylation/sulfonation of 8-methylquinolines,¹⁴ in which an organophosphorus/sulfonate
18 hypervalent iodine reagent is used as both an oxidant and the source of a functional group. In addition,
19 pyridyl-directed aryl C(sp²)-H bond hydroxylation and arylation are also conveniently achieved via the
20 current method (Scheme 1c).
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23 RESULTS AND DISCUSSION

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25 The reaction was initially investigated by using **1a** as the model substrate, Pd(OAc)₂ as the catalyst,
26 PhI(OH)P(O)(OPh)₂ (**2a**) as the oxidant and dioxane as the solvent at room temperature in air. To our
27 delight, desired product **3a** was obtained in 58% yield (Table 1, entry 1). Among the catalysts that were
28 tested, Pd(OAc)₂ displayed the best efficiency (entries 2-4). The presence of additives had an important
29 effect on the reaction. While the addition of 4Å MS could significantly increase the conversion of the
30 reaction (entry 5), other additives such as an acid (e.g. PivOH) or a base (e.g. K₂CO₃) did not improve
31 the yield of the reaction (entries 6-7). Moreover, increasing the amount of the oxidant increased the
32 yields of **3a** (entries 8-10). When the amount of the oxidant reached 2.4 equiv, **3a** was obtained in 87%
33 isolated yield. However, continuing to increase the amount of oxidant did not further increase the
34 yield of **3a** (entry 11). In addition, lowering the catalyst loading to 5 mol% reduced the yield of **3a** to
35 74% (entry 12). To verify the role of the molecular sieves, 5 equiv of H₂O was added, and the yield of
36 **3a** significantly decreased (entry 13). Meanwhile, **3a** was obtained in 85% yield when the reaction was
37 performed under Ar (entry 14). Consequently, the optimum reaction conditions were determined to be
38 **1a** (0.2 mmol) in the presence of Pd(OAc)₂ (10 mol%), **2a** (2.4 equiv), 4Å MS (175 mg) in dioxane (1
39 mL) at room temperature open to air.
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58 **Table 1. Optimization of the Reaction Conditions** ^a



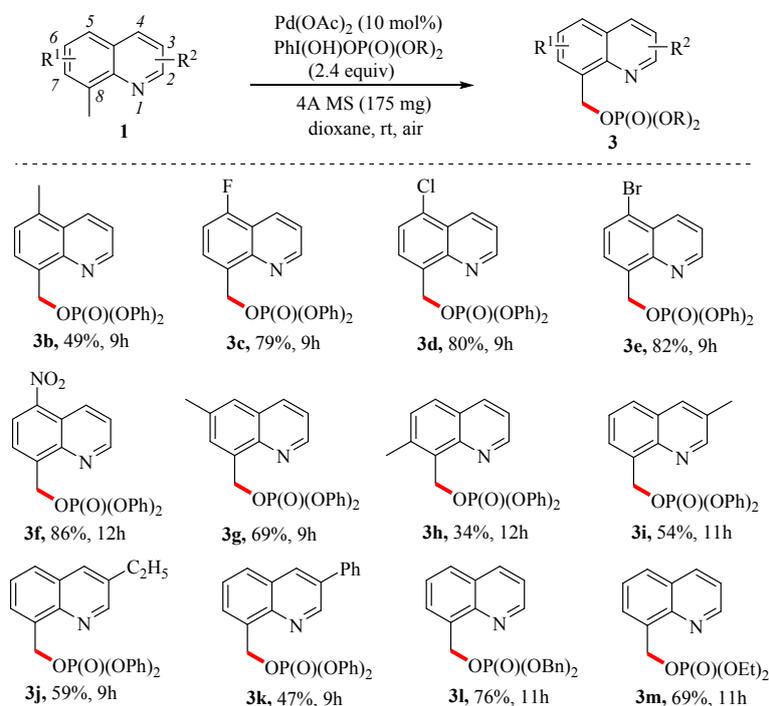
Entry	Catalyst	X	Additive	Yield [%] ^b
1	Pd(OAc) ₂	1.5	-	3a , 58%
2	Pd(OCOCF ₃) ₂	1.5	-	3a , 37%
3	PdCl ₂	1.5	-	3a , 32%
4	Pd(PPh ₃) ₄	1.5	-	3a , <5%
5 ^c	Pd(OAc) ₂	1.5	4Å MS	3a , 68%
6 ^d	Pd(OAc) ₂	1.5	PivOH	3a , 53%
7 ^e	Pd(OAc) ₂	1.5	K ₂ CO ₃	3a , 59%
8 ^c	Pd(OAc) ₂	1.8	4Å MS	3a , 76%
9 ^c	Pd(OAc) ₂	2.1	4Å MS	3a , 81%
10 ^c	Pd(OAc) ₂	2.4	4Å MS	3a , 87%
11 ^c	Pd(OAc) ₂	2.5	4Å MS	3a , 84%
12 ^c	Pd(OAc) ₂	2.4	4Å MS	3a , 74%
13 ^{c, g}	Pd(OAc) ₂	2.4	4Å MS, H ₂ O	3a , 78%
14 ^h	Pd(OAc) ₂	2.4	-	3a , 85%

^a Reaction conditions: **1a** (0.2 mmol), catalyst (10 mol%), solvent (1 mL). ^b Isolated yields. ^c 4Å MS (175 mg) was added. ^d 1 equiv. of acid was added. ^e 1 equiv. of base was added. ^f 5 mol% of catalyst was used. ^g 5 equiv of H₂O was added. ^h in Ar.

With the optimized conditions in hand, the reactions between substrates **1a-1m** and phosphate hypervalent iodine species were first tested. As shown in **Table 2**, the substrates bearing electron-withdrawing substituents such as F, Cl, Br, and NO₂ at the 5-position were tolerated and gave corresponding products **3c-3f** in good yields. However, when the substrates possessed electron-donating groups, the yields were sharply decreased (**3b**, **3g-3k**). These results indicated that the electronic properties of the substituents are crucial to the reaction. When the 7-position of the substrate was blocked by a substituent (e.g., methyl), the yield was obviously reduced, possibly due to steric hinderance (**3h**). Moreover, when the oxidant was replaced by

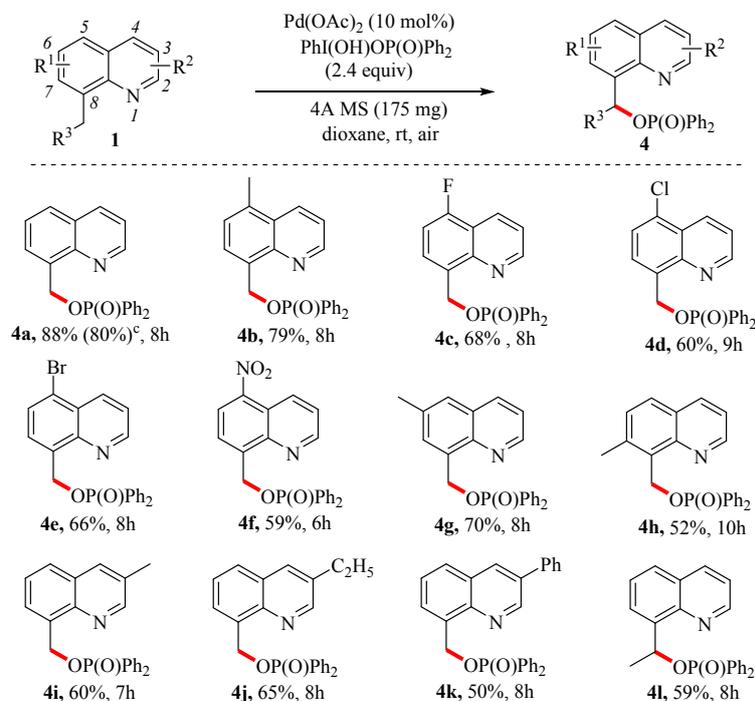
dibenzyl or diethyl phosphate hypervalent iodine, corresponding products **3l** and **3m** were isolated in 76% and 69% yields, respectively.

Table 2 Phosphate esterification of 8-methylquinolines^{a, b}



^a Reaction conditions: **1** (0.2 mmol), $\text{Pd}(\text{OAc})_2$ (0.02 mmol), oxidant (0.48 mmol), 4Å MS (175 mg), dioxane (1 mL) at rt for 9-12 h in air. ^b Isolated yields.

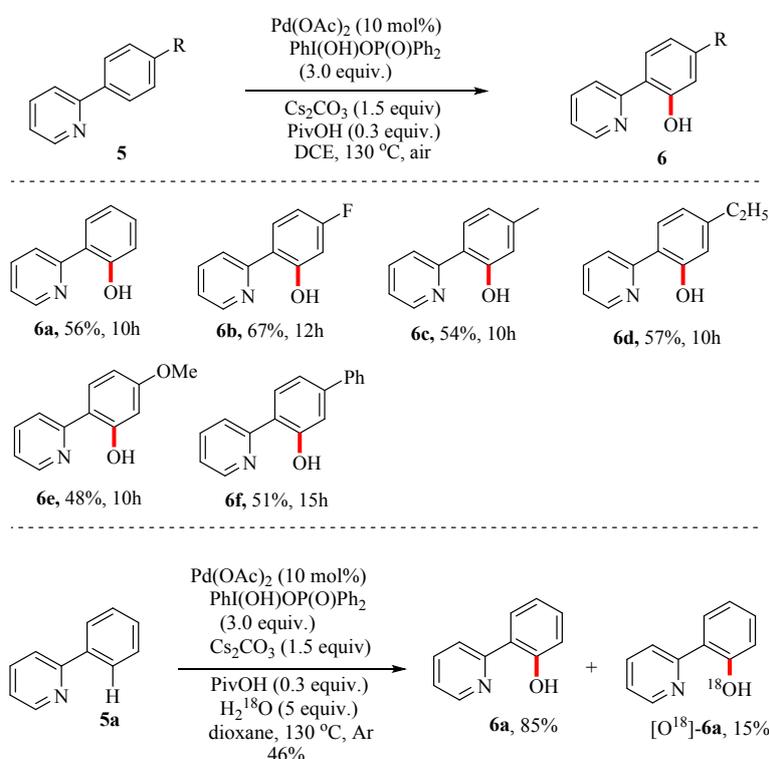
Table 3 Phosphinic esterification of 8-methylquinolines^{a, b}



^a Reaction conditions: **1** (0.2 mmol), $\text{Pd}(\text{OAc})_2$ (0.02 mmol), oxidant (0.48 mmol), 4Å MS (175 mg), dioxane (1 mL) at rt for 6-10 h in air. ^b Isolated yields. ^c 2.0 mmol scale.

Phosphinic hypervalent-iodines were subsequently evaluated (Table 3). Regardless of if the substrates possessed electron-donating groups such as a methyl substituent, or electron-withdrawing substituents, such as F, Cl, Br, or NO₂, the reaction proceeded smoothly and afforded target products **4a-4k** in moderate to good yields. However, the phosphate product was not obtained when a strong electron-donating group such as OMe was present on the substrate. To our delight, when the primary C-H bond was replaced with a secondary C-H bond at the 8-position of the substrate, the reaction still occurred and gave product **4l** in 59% yield. However, when converting to more sterically hindered tertiary C-H bonds, the reaction did not proceed. To prove the practicality of this approach, a 2.0 mmol scale reaction was carried out, and product **4a** was isolated in a yield of 80%, which was comparable to that of the small-scale reaction.

Table 4 Hydroxylation of 2-phenylpyridines ^{a, b}

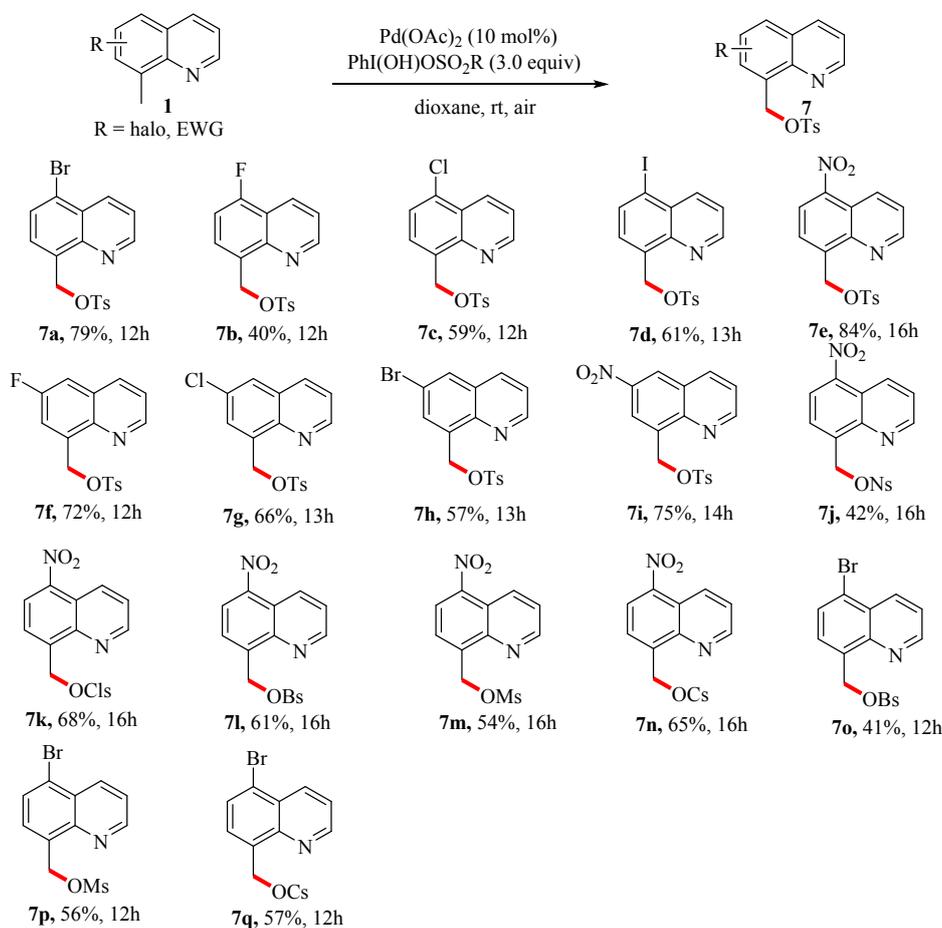


^a Reaction conditions: **5** (0.2 mmol), Pd(OAc)₂ (0.02 mmol), oxidant (0.6 mmol), Cs₂CO₃ (0.3 mmol), PivOH (0.06 mmol), DCE (1 mL) at 130 °C for 10-15 h in air. ^b Isolated yields.

When 2-phenylpyridines **5** were utilized as substrates, an OH group other than the OP(O)Ph₂ group was unexpectedly introduced (Table 4).¹⁵ Substrates bearing the electron-donating groups such as Me, Et, OMe, and Ph substituents or the electron-withdrawing groups such as F substituent at *para*-position of the benzene smoothly underwent the reaction and produced target products **6a-6f** in moderate to good yields. Moreover, when 5 equiv. of H₂¹⁸O was added to the reaction, products **6a** and [O¹⁸]-**6a** were obtained in an approximate ratio of 5.7 (see SI),

indicating that the oxygen atom in the hydroxy product likely originated from the phosphinic hypervalent iodine rather than adventitious H₂O.

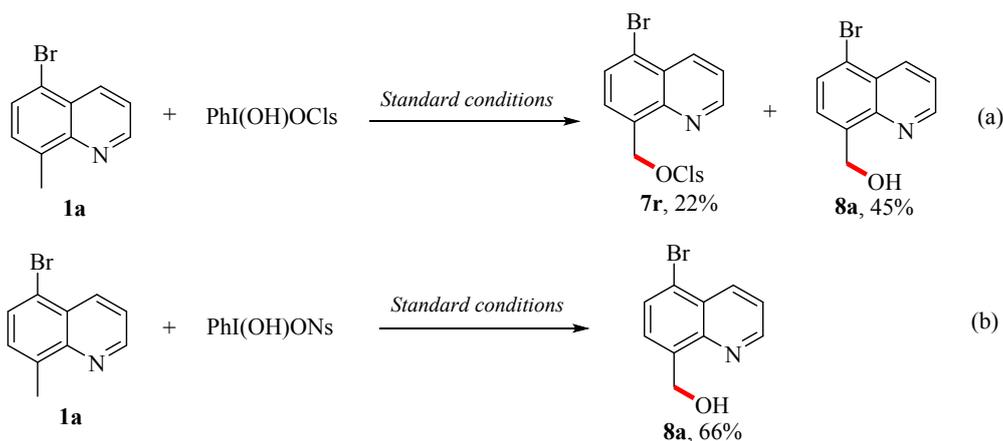
Table 5 Sulfonation of 8-methylquinolines^{a, b}



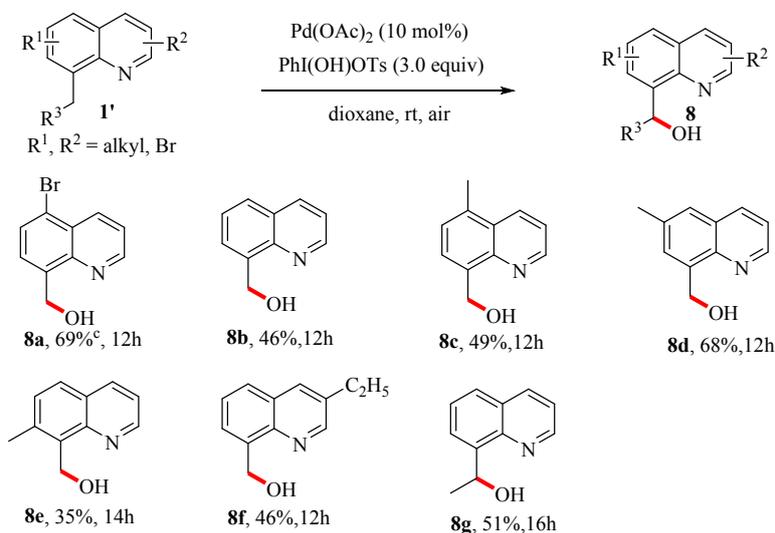
^a Reaction conditions: **1** (0.2 mmol), Pd(OAc)₂ (0.02 mmol), oxidant (0.60 mmol), dioxane (1 mL) for 12-16 h. ^b Isolated yields. ^c Ns, 4-NO₂-C₆H₄SO₂; Cls, 4-Cl-C₆H₄SO₂; Bs, C₆H₅SO₂; Ms, MeSO₂; Cs, (1R)-10-camphorylsulfonyl.

When sulfonate hypervalent iodine reagents were used as the oxidant, the reactions proceeded smoothly, and the results are presented in **Table 5**. The OTs products were produced in moderate to excellent yields from reactions with substrates bearing electron-withdrawing substituents under similar conditions (**7a-7i**). Preliminary screenings indicated that more strongly electron-withdrawing substituents (e.g., NO₂ versus F, Cl, Br, and I) favored the reaction. Other sulfonate hypervalent iodine species were examined, and various sulfonates (e.g., ONs, OClS, OBs, OMs, and OCs) were expediently introduced in moderate to good yields when using substrates bearing a NO₂ group at the 5-position (**7j-7n**). However, when the 5-position of the substrate was taken up by a Br group, some changes had occurred. For alkyl or phenyl sulfonates, the expected products were obtained (**7o-7q**); however, when the *para*-position of the sulfonates were substituted with electron-withdrawing groups such Cl or NO₂, the OH product was obtained as the major or only product (Scheme 2a and 2b).

Scheme 2 Competitive reactions.



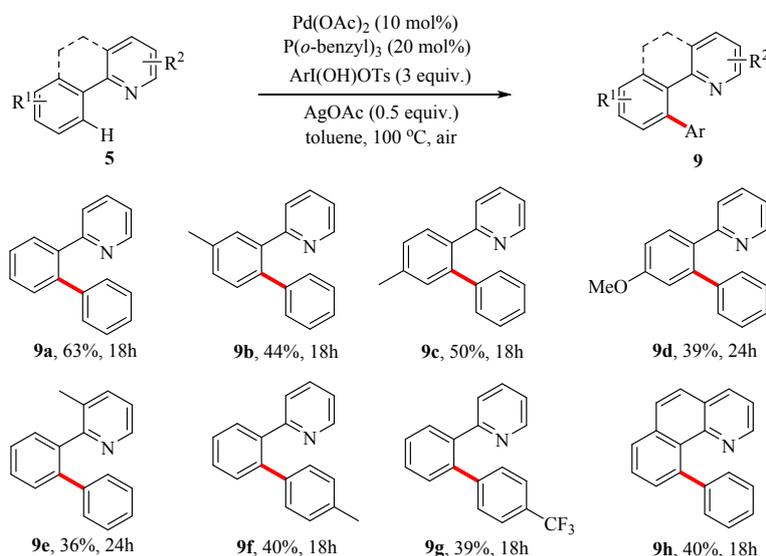
When the substrates bearing electron-donating groups were subjected to the reaction, surprisingly, OH other than OTs products were obtained (**8a-8g**). Interestingly, for substrate **1a**, OTs product **7a** was isolated in 79% yield following purification with a more polar eluent (PE: EA = 2:1), while hydroxy product **6a** was obtained in 69% yield when using a less polar eluent (PE: EA = 20:1). Thus, the reaction not only offered a new approach for hydroxylation, but also revealed that the hydroxy source was OTs other than the OH of HTIB. To our delight, when the primary C-H bond at 8-position of the substrate was replaced by secondary C-H bond, the reaction still proceeded smoothly and gave product **6g** in 51% yield. However, when moving to more sterically hindered tertiary C-H bonds, the reaction did not occur.

Table 6 Hydroxylation of 8-methylquinolines ^{a, b}

^a Reaction conditions: **1** (0.2 mmol), Pd(OAc)₂ (0.02 mmol), oxidant (0.60 mmol), dioxane (1 mL) for 12-16 h. ^b Isolated yields. ^c Less polar developing solvent (PE: EA = 20:1) was used during the column chromatography.

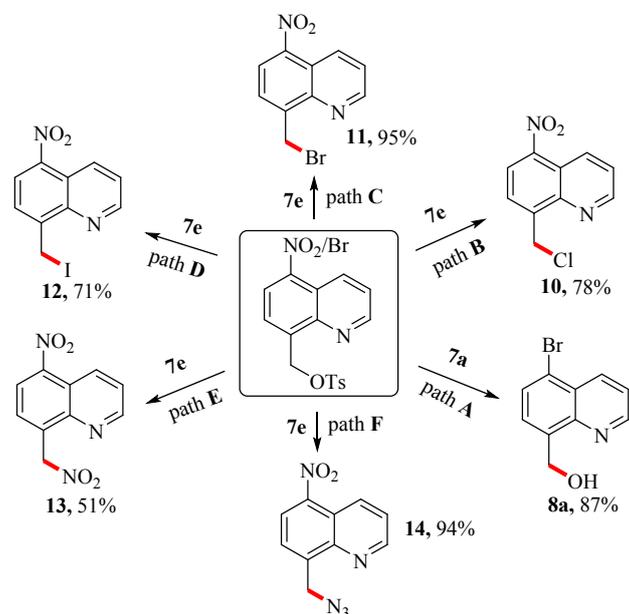
The reactions between 2-phenylpyridine analogues and sulfonate hypervalent iodine species are also discussed (Table 7). In this process, an aryl group other than the OTs of the sulfonate hypervalent iodine was unexpectedly introduced, thus affording an alternative method for arylation.¹⁶

Table 7 Arylation of 2-phenylpyridines^{a, b}



^a Reaction conditions: **6** (0.2 mmol), $\text{Pd}(\text{OAc})_2$ (0.02 mmol), ligand (0.04 mmol), oxidant (0.60 mmol), AgOAc (0.1 mmol), toluene (1 mL) for 18–24 h. ^b Isolated yields.

Scheme 3 Synthetic applications.^a



^a Conditions: (a) **5a** (0.2 mmol), SiO_2 (0.6 mmol), PE : EA = 1 mL : 0.05 mL, rt, 5 h; (b) **5e** (0.2 mmol), *t*-BuCl (0.24 mmol), $[\text{pmIm}]\text{Br}$ (0.4 mmol), 60 °C, 4 h; (c) **5e** (0.2 mmol), *t*-BuBr (0.24 mmol), $[\text{pmIm}]\text{Br}$ (0.4 mmol), sonication, 0.5 h; (d) **5e** (0.2 mmol), *t*-BuI (0.24 mmol), $[\text{pmIm}]\text{Br}$ (0.4 mmol), sonication, 0.5 h; (e) **5e** (0.2 mmol), AgNO_3 (0.4 mmol), *t*-AmOH, 120 °C, 6 h; (f) **5e** (0.2 mmol), NaN_3 (0.3 mmol), THF/ H_2O , reflux, 3 h. ^b Isolated yields.

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3 The synthetic utility of the OTs products was exemplified by further transformations of
4 products **7a/7e** (Scheme 3). Substrate **7a** was hydrolyzed to product **8a** in 87% yield under
5 reaction conditions similar to the column chromatography conditions. Given that OTs is a good
6 leaving group, several substitution reactions were conducted.¹⁷ *tert*-Butyl chloride, bromide and
7 iodide in combinations with the ionic liquid [pmIm]Br had been used to convert the OTs
8 derivative into the corresponding chloride, bromide and iodide products under sonication
9 conditions (or heating) in good yields. In addition, the NO₂ and N₃ products were also produced
10 by reacting the OTs derivatives with AgNO₃ and NaN₃, respectively.
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14 CONCLUSIONS

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17 In summary, by utilizing organophosphorus/sulfonate hypervalent iodine reagents as both an
18 oxidant and the source of a functional group, we have realized a convenient and efficient
19 approach for the direct, palladium-catalyzed C(sp³)-H phosphorylation/sulfonation of
20 8-methylquinolines. Various weakly nucleophilic and good leaving groups, such as phosphates,
21 phosphonates and sulfonates, were successfully introduced under mild conditions, providing a
22 new strategy for the preparation of such species. However, when using the current method for
23 2-phenylpyridine analogues, the hydroxyl or phenyl products are unexpectedly produced.
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30 EXPERIMENTAL SECTION

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32 **General Information.** Reactions were monitored by using thin-layer chromatography (TLC) on
33 commercial silica gel plates (GF254). Visualization of the developed plates was performed under UV
34 lights (254 nm). Flash column chromatography was performed on silica gel (200-300 mesh). ¹H, ¹³C
35 and ³¹P NMR spectra were recorded on a Bruker AV300, 400 or 500 spectrometer. Chemical shifts (δ)
36 were reported in ppm referenced to the CDCl₃ residual peak (δ 7.26) or the DMSO-d₆ residual peak (δ
37 2.50) for ¹H NMR. Chemical shifts of ¹³C NMR were reported relative to CDCl₃ (δ 77.0) or D₆-DMSO
38 (δ 39.5). Chemical shifts of ³¹P NMR were reported relative to 85% H₃PO₄ (δ = 0). The following
39 abbreviations were used to describe peak splitting patterns when appropriate: br s = broad singlet, s =
40 singlet, d = doublet, t = triplet, q = quartet, m = multiplet. Coupling constant, *J*, was reported in Hertz
41 unit (Hz). Melting points (mp) were taken on a MEL-TEMP® apparatus and were uncorrected. High
42 resolution mass spectra (HRMS) were obtained on an ESI-LC-MS/MS spectrometer.
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50 **General procedure: Preparation of the substrates**

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52 All substrates **1a-1l** used in this work are known compounds. 8-Methylquinolines **1a**, **1l** and
53 2-phenylpyridines **5** are commercially available. The substrates **1b-1k** were prepared as published
54 before, and the ¹H NMR spectral data matched those of previous reported.¹⁸
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57 **General procedure for the synthesis of the substates **1b-1h**^{18a-18c}**

58 Glycerin (5.6 g, 60 mmol) was added dropwise over a period of 0.5h to a solution of substituted
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3 aniline (50 mmol) and NaI (0.1 g, 0.6 mmol) in 80% aqueous H₂SO₄ (27 g, 220 mmol) and stirred in a
4 140 °C oil-bath. The mixture was then heated at 140-145 °C for 3.5h while distilling the water formed
5 during this period. Upon cooling to room temperature, the dark solution was carefully poured into ice
6 (50 g) and then neutralized with 25% NaOH (55 g, 0.34 mol) to basic pH 8~11. The mixture was
7 extracted with ethyl acetate (3×50 mL). The combined organic solution was washed with water (2×50
8 mL), brine (50 mL), dried with anhydrous sodium sulfate, and concentrated under reduced pressure.
9 The crude product was purified by column chromatography on silica gel (petroleum ether/EtOAc = 8:1)
10 to afford the corresponding substituted 8-methyl-quinoline **1b-1h**.

11 **General procedure for the synthesis of the substates 1i-1j^{18d-18e}**

12 Unsaturated aldehyde (5.6 g, 60 mmol) was added dropwise over a period of 0.5h to a solution of
13 substituted aniline (50 mmol) and NaI (0.1 g, 0.6 mmol) in 80% aqueous H₂SO₄ (27 g, 220 mmol) and
14 stirred in a 110 °C oil-bath. The mixture was then heated at 110 °C for 2h while distilling the water
15 formed during this period. Upon cooling to room temperature, the dark solution was carefully poured
16 into ice (50 g) and then neutralized with 25% NaOH (55 g, 0.34 mol) to basic pH 8~11. The mixture
17 was extracted with ethyl acetate (3×50 mL). The combined organic solution was washed with water
18 (2×50 mL), brine (50 mL), dried with anhydrous sodium sulfate, and concentrated under reduced
19 pressure. The crude product was purified by column chromatography on silica gel (petroleum
20 ether/EtOAc = 8:1) to give the corresponding substituted 8-methyl-quinoline **1i-1j**.

21 **General procedure for the synthesis of the substate 1k^{18f}**

22 Aniline (5.0 mmol) and styrene oxide (10.0 mmol) were added to a mixture of MeSO₃H (1.5 mL) and
23 Al₂O₃ (0.5 g). The mixture was stirred at room temperature in solvent-free conditions for 0.5 h. After
24 completion of the reaction, the mixture was diluted with ethyl acetate, and filtered. The filtrate was
25 washed with a solution of NaHCO₃ (5%; 3 x 30 mL) and then 150 mL deionized water. The solution
26 was dried over magnesium sulfate; the solvent was evaporated to give the crude product, which was
27 purified by silica gel column chromatography employing petroleum ether/ethyl acetate (8:1) as eluent.

28 **General procedure: Palladium-Catalyzed phosphorylation/sulfonation reactions 29 of 8-methylquinolines**

30 *[Caution] Since the hypervalent iodine reagents are generally explosive, so please operate them in a
31 fume hood.*

32 **Procedure A:** A oven-dried 15 mL Schlenk tube was charged with a mixture of **1** (0.2 mmol),
33 Pd(OAc)₂ (10 mol%), oxidant (0.48 mmol), 4Å MS (175 mg) in dioxane (1.0 mL) and was stirred at
34 room temperature for 6-12 h. After the reaction was complete (as determined by TLC analysis), the
35 reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was
36 extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated
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brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 4:1) to afford the targeted products **3a-3m** or **4a-4l**.

Diphenyl (quinolin-8-ylmethyl) phosphate (3a)

Yield: 87% (68.1 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.92 (d, *J* = 4.2 Hz, 1H), 8.17 (d, *J* = 8.4 Hz, 1H), 7.82 (t, *J* = 8.1 Hz, 2H), 7.53 (t, *J* = 7.7 Hz, 1H), 7.47-7.42 (m, 1H), 7.35-7.24 (m, 8H), 7.18 (t, *J* = 7.1 Hz, 2H), 6.09 (d, *J* = 8.1 Hz, 2H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 150.6 (d, *J* = 7.1 Hz), 149.8, 145.5, 136.2, 133.5 (d, *J* = 7.4 Hz), 129.7, 128.2 (d, *J* = 16.1 Hz), 128.0, 126.3, 125.3, 121.4, 120.2 (d, *J* = 4.9 Hz), 67.3 (d, *J* = 5.6 Hz); ³¹P NMR (121 MHz, CDCl₃): δ -11.69; IR (KBr): ν 3070, 3015, 2952, 1594, 1487, 1290, 1215, 1187, 1163, 876, 823, 789, 764; HRMS (ESI-TOF) *m/z*: [M+Na]⁺ Calcd for C₂₂H₁₈NO₄PNa 414.0866, Found 414.0870.

(5-methylquinolin-8-yl)methyl diphenyl phosphate (3b)

Yield: 49% (39.7 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.92 (d, *J* = 4.2 Hz, 1H), 8.34 (d, *J* = 8.4 Hz, 1H), 7.12 (d, *J* = 7.2 Hz, 1H), 7.49-7.45 (m, 1H), 7.37-7.14 (m, 11H), 6.04 (d, *J* = 8.1 Hz, 2H), 2.69 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 150.6 (d, *J* = 7.1 Hz), 149.3, 145.8, 135.3, 132.6, 131.5 (d, *J* = 7.1 Hz), 129.7, 128.1, 127.4, 126.6, 125.2, 121.0, 120.2 (d, *J* = 4.9 Hz), 67.5 (d, *J* = 5.6 Hz), 18.7; ³¹P NMR (121 MHz, CDCl₃): δ -10.58; IR (KBr): ν 3065, 2922, 2853, 1594, 1491, 1412, 1240, 1203, 1160, 848, 771, 719; HRMS (ESI-TOF) *m/z*: [M+Na]⁺ Calcd for C₂₃H₂₀NO₄PNa 428.1022, Found 428.1025.

(5-fluoroquinolin-8-yl)methyl diphenyl phosphate (3c)

Yield: 79% (64.7 mg), Colorless oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.95 (d, *J* = 3.3 Hz, 1H), 8.42 (d, *J* = 8.4 Hz, 1H), 7.75 (t, *J* = 6.6 Hz, 1H), 7.51-7.47 (m, 1H), 7.33-7.15 (m, 11H), 5.99 (d, *J* = 8.4 Hz, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 157.9 (d, *J*_{CF} = 254.9 Hz), 150.6 (t, *J* = 11.0 Hz), 146.2 (d, *J* = 3.1 Hz), 129.7, 129.4 (d, *J* = 4.5 Hz), 128.4 (d, *J* = 9.1 Hz), 125.2, 121.4 (d, *J* = 2.8 Hz), 120.1 (d, *J* = 4.9 Hz), 118.8 (d, *J* = 16.5 Hz), 109.7 (d, *J* = 19.4 Hz), 66.8 (d, *J* = 5.6 Hz); ³¹P NMR (121 MHz, CDCl₃): δ -11.68; IR (KBr): ν 3050, 2962, 2930, 1599, 1489, 1409, 1287, 1217, 1185, 1160, 898, 868, 776, 756; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₂H₁₈FNO₄P 410.0952, Found 410.0954.

(5-chloroquinolin-8-yl)methyl diphenyl phosphate (3d)

Yield: 80% (68.1 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.95 (d, *J* = 3.3 Hz, 1H), 8.59 (d, *J* = 8.7 Hz, 1H), 7.73 (d, *J* = 7.5 Hz, 1H), 7.61-7.54 (m, 2H), 7.34-7.16 (m, 10H), 6.02 (d, *J* = 8.4 Hz, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 150.5 (d, *J* = 6.9 Hz), 150.3, 146.0, 133.0, 131.6, 129.7, 127.8, 126.3, 126.0, 125.3, 122.1, 120.1 (d, *J* = 4.9 Hz), 66.9 (d, *J* = 5.5 Hz); ³¹P NMR (121 MHz, CDCl₃): δ -11.72; IR (KBr): ν 3047, 2932, 2848, 1587, 1489,

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4 1444, 1282, 1220, 1187, 1162, 871, 848, 774, 754; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for
5 $C_{22}H_{18}ClNO_4P$ 426.0656, Found 426.0658.

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7 **(5-bromoquinolin-8-yl)methyl diphenyl phosphate (3e)**

8 Yield: 82% (77.1 mg), Pale yellow oil, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300
9 MHz, $CDCl_3$): δ 8.72 (d, J = 4.2 Hz, 1H), 8.47 (d, J = 8.6 Hz, 1H), 7.75 (d, J = 7.8 Hz, 1H), 7.63 (d, J
10 = 7.8 Hz, 1H), 7.50-7.46 (m, 1H), 7.33-7.23 (m, 8H), 7.16 (t, J = 7.1 Hz, 2H), 6.00 (d, J = 8.1 Hz, 2H);
11 $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 150.5 (t, J = 7.1 Hz), 145.9, 135.5, 133.7 (d, J = 7.3 Hz), 130.0,
12 129.8 (d, J = 0.5 Hz), 128.2, 127.2, 125.4 (d, J = 1.1 Hz), 122.5, 122.2, 120.1 (d, J = 5.0 Hz), 67.0 (d, J
13 = 5.4 Hz); ^{31}P NMR (121 MHz, $CDCl_3$): δ -11.71; IR (KBr): ν 3097, 3057, 3042, 2942, 1589, 1482,
14 1452, 1295, 1215, 1187, 1145, 871, 828, 776, 761; HRMS (ESI-TOF) m/z : $[M+Na]^+$ Calcd for
15 $C_{22}H_{17}BrNO_4PNa$ 491.9961, Found 491.9955.

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21 **(5-nitroquinolin-8-yl)methyl diphenyl phosphate (3f)**

22 Yield: 86% (75.1 mg), Yellow oil, R_f = 0.35 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300
23 MHz, $CDCl_3$): δ 9.01 (t, J = 7.7 Hz, 2H), 8.34 (d, J = 7.8 Hz, 1H), 7.87 (d, J = 8.1 Hz, 1H), 7.71-7.67
24 (m, 1H), 7.38-7.20 (m, 10H), 6.12 (d, J = 7.8 Hz, 2H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 150.7, 150.4
25 (d, J = 7.1 Hz), 145.0 (d, J = 14.0 Hz), 141.6 (d, J = 7.7 Hz), 132.2, 129.9, 125.5, 124.8, 124.4, 124.2,
26 120.7, 120.1 (d, J = 4.9 Hz), 66.9 (d, J = 5.0 Hz); ^{31}P NMR (121 MHz, $CDCl_3$): δ -11.78; IR (KBr): ν
27 3072, 3057, 3040, 2965, 1591, 1524, 1479, 1452, 1404, 1292, 1212, 1180, 1155, 871, 851, 834, 799,
28 774, 721; HRMS (ESI-TOF) m/z : $[M+Na]^+$ Calcd for $C_{22}H_{17}N_2O_6PNa$ 459.0716, Found 459.0710.

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34 **(6-methylquinolin-8-yl)methyl diphenyl phosphate (3g)**

35 Yield: 69% (55.9 mg), Yellow oil, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300
36 MHz, DMSO): δ 8.87 (d, J = 4.1 Hz, 1H), 8.29 (d, J = 8.4 Hz, 1H), 7.75 (s, 1H), 7.57-7.53 (m, 2H),
37 7.44-7.39 (m, 4H), 7.27-7.22 (m, 6H), 5.92 (d, J = 8.1 Hz, 2H), 2.44 (s, 3H); $^{13}C\{^1H\}$ NMR (75 MHz,
38 DMSO): δ 152.1 (d, J = 6.9 Hz), 147.8, 141.3, 139.2, 138.2, 137.3, 130.6, 130.0, 128.6, 126.0, 124.4,
39 122.0, 120.4 (d, J = 5.0 Hz), 59.9, 21.8; ^{31}P NMR (121 MHz, $CDCl_3$): δ -12.34; IR (KBr): ν 3072, 2918,
40 2853, 1589, 1489, 1439, 1290, 1160, 1122, 868, 776, 711; HRMS (ESI-TOF) m/z : $[M+Na]^+$ Calcd for
41 $C_{23}H_{20}NO_4PNa$ 428.1027, Found 428.1034.

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47 **(7-methylquinolin-8-yl)methyl diphenyl phosphate (3h)**

48 Yield: 34% (27.6 mg), Yellow oil, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300
49 MHz, $CDCl_3$): δ 8.93 (d, J = 4.2 Hz, 1H), 8.12 (d, J = 8.3 Hz, 1H), 7.76 (d, J = 8.4 Hz, 1H), 7.42-7.36
50 (m, 2H), 7.31-7.12 (m, 10H), 6.20 (d, J = 6.9 Hz, 2H), 2.61 (s, 3H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ
51 150.7 (d, J = 7.1 Hz), 150.1, 146.8, 140.8, 135.9, 130.1, 129.8, 129.6, 128.8, 126.5, 125.0, 120.5, 120.2
52 (d, J = 5.0 Hz), 63.1 (d, J = 5.7 Hz), 19.7; ^{31}P NMR (121 MHz, $CDCl_3$): δ -11.88; IR (KBr): ν 3050,
53 2920, 2850, 1594, 1482, 1437, 1285, 1220, 1190, 1163, 831, 776, 749; HRMS (ESI-TOF) m/z :
54 $[M+Na]^+$ Calcd for $C_{23}H_{20}NO_4PNa$ 428.1022, Found 428.1023.
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(3-methylquinolin-8-yl)methyl diphenyl phosphate (3i)

Yield: 54% (43.8 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.75 (d, $J = 1.5$ Hz, 1H), 7.89 (s, 1H), 7.76-7.70 (m, 2H), 7.48 (t, $J = 7.7$ Hz, 1H), 7.34-7.24 (m, 8H), 7.17 (t, $J = 7.1$ Hz, 2H), 6.07 (d, $J = 8.1$ Hz, 2H), 2.51 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 151.8, 150.6 (d, $J = 7.1$ Hz), 143.8, 134.8, 133.2 (d, $J = 7.3$ Hz), 130.9, 129.7, 127.8 (d, $J = 10.3$ Hz), 127.1, 126.3, 125.3 (d, $J = 1.0$ Hz), 120.2 (d, $J = 5.0$ Hz), 67.4 (d, $J = 5.6$ Hz), 18.7; ^{31}P NMR (121 MHz, CDCl_3): δ -11.74; IR (KBr): ν 3065, 2922, 2850, 1586, 1544, 1489, 1439, 1290, 1217, 1187, 1165, 881, 768, 711; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{23}\text{H}_{20}\text{NO}_4\text{PNa}$ 428.1022, Found 428.1018.

(3-ethylquinolin-8-yl)methyl diphenyl phosphate (3j)

Yield: 59% (49.5 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.79 (d, $J = 2.4$ Hz, 1H), 7.92 (d, $J = 2.1$ Hz, 1H), 7.77-7.74 (m, 2H), 7.52-7.47 (m, 1H), 7.34-7.24 (m, 8H), 7.20-7.14 (m, 2H), 6.08 (d, $J = 8.1$ Hz, 2H), 2.89-2.81 (m, 2H), 1.36 (t, $J = 7.5$ Hz, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 151.3, 150.6 (d, $J = 7.1$ Hz), 144.0, 137.0, 133.6, 133.2 (d, $J = 7.3$ Hz), 129.7, 127.9 (d, $J = 6.1$ Hz), 127.2, 126.3, 125.2 (d, $J = 1.1$ Hz), 120.2 (d, $J = 5.0$ Hz), 67.4 (d, $J = 5.6$ Hz), 26.2, 15.2; ^{31}P NMR (121 MHz, CDCl_3): δ -11.72; IR (KBr): ν 3072, 2967, 2925, 2870, 1589, 1489, 1457, 896, 876, 828, 776; HRMS (ESI-TOF) m/z : $[\text{M}+\text{K}]^+$ Calcd for $\text{C}_{24}\text{H}_{22}\text{NO}_4\text{PK}$ 458.0918, Found 458.0926.

Diphenyl ((3-phenylquinolin-8-yl)methyl) phosphate (3k)

Yield: 47% (43.9 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 9.15 (d, $J = 1.2$ Hz, 1H), 8.28 (d, $J = 1.2$ Hz, 1H), 7.83 (d, $J = 10.2$ Hz, 2H), 7.70 (t, $J = 3.6$ Hz, 2H), 7.55-7.52 (m, 3H), 7.46-7.43 (m, 1H), 7.31-7.23 (m, 8H), 7.15 (t, $J = 3.6$ Hz, 2H), 6.09 (d, $J = 4.2$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 150.6 (d, $J = 7.1$ Hz), 149.3, 144.6, 137.7, 134.1, 133.5 (d, $J = 7.1$ Hz), 133.3, 129.7, 129.3, 128.5, 128.3, 128.1, 127.8, 127.4, 126.8, 125.3, 120.2 (d, $J = 4.9$ Hz), 67.3 (d, $J = 5.6$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ -11.71; IR (KBr): ν 3067, 2922, 2850, 1584, 1484, 1444, 1285, 1215, 1187, 1162, 898, 871, 826, 771; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{28}\text{H}_{22}\text{NO}_4\text{PNa}$ 490.1179, Found 490.1187.

Dibenzyl (quinolin-8-ylmethyl) phosphate (3l)

Yield: 76% (63.7 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.90 (t, $J = 2.4$ Hz, 1H), 8.16 (d, $J = 8.4$ Hz, 1H), 7.79 (d, $J = 7.8$ Hz, 2H), 7.52 (t, $J = 7.8$ Hz, 1H), 7.45-7.33 (m, 11H), 5.85 (d, $J = 7.5$ Hz, 2H), 5.11 (d, $J = 7.8$ Hz, 4H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 149.8, 145.6, 136.2, 135.9 (d, $J = 7.1$ Hz), 134.1 (d, $J = 7.6$ Hz), 128.5 (d, $J = 7.6$ Hz), 128.0 (d, $J = 11.0$ Hz), 126.3, 121.3, 69.3 (d, $J = 5.5$ Hz), 66.0 (d, $J = 5.0$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ -0.73; IR (KBr): ν 3032, 2920, 2853, 1596, 1502, 1454, 1275, 1215, 1187, 876, 821, 786, 736; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{24}\text{H}_{22}\text{NO}_4\text{PNa}$ 442.1179, Found 442.1183.

Diethyl (quinolin-8-ylmethyl) phosphate (3m)

Yield: 69% (40.8 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.91 (d, $J = 3.9$ Hz, 1H), 8.16 (d, $J = 8.1$ Hz, 1H), 7.89 (d, $J = 6.9$ Hz, 1H), 7.79 (d, $J = 7.8$ Hz, 1H), 7.56 (t, $J = 7.5$ Hz, 1H), 7.45-7.41 (m, 1H), 5.83 (d, $J = 7.2$ Hz, 2H), 4.21-4.11 (m, 4H), 1.32 (t, $J = 6.9$ Hz, 6H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 149.7, 145.5, 136.2, 134.4 (d, $J = 7.5$ Hz), 127.9, 127.7, 126.3, 121.3, 65.7 (d, $J = 5.0$ Hz), 63.9 (d, $J = 5.7$ Hz), 16.1 (d, $J = 6.8$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ -0.82; IR (KBr): ν 2980, 2930, 2905, 1596, 1559, 1499, 1462, 1265, 1162, 876, 826, 791; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{14}\text{H}_{18}\text{NO}_4\text{PNa}$ 318.0866, Found 318.0874.

Quinolin-8-ylmethyl diphenylphosphinate (4a)

Yield: 88% (63.3 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.84 (d, $J = 2.1$ Hz, 1H), 8.13 (d, $J = 4.2$ Hz, 1H), 7.95-7.88 (m, 5H), 7.75 (d, $J = 3.9$ Hz, 1H), 7.55-7.48 (m, 3H), 7.44-7.38 (m, 5H), 5.83 (d, $J = 3.6$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 149.6, 145.6, 136.1, 134.9 (d, $J = 7.0$ Hz), 132.5, 132.1 (d, $J = 2.8$ Hz), 131.8 (d, $J = 10.1$ Hz), 130.7, 128.5 (d, $J = 13.1$ Hz), 127.8 (d, $J = 10.4$ Hz), 126.3, 121.2, 63.3 (d, $J = 5.0$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ 32.42; IR (KBr): ν 3049, 2930, 2863, 1589, 1576, 1496, 1437, 1222, 1160, 1125, 863, 833, 786, 749, 729; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{22}\text{H}_{18}\text{NO}_2\text{PNa}$ 382.0967, Found 382.0970.

(5-methylquinolin-8-yl)methyl diphenylphosphinate (4b)

Yield: 79% (59.0 mg), Pink oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.85 (d, $J = 3.9$ Hz, 1H), 8.28 (d, $J = 8.7$ Hz, 1H), 7.95-7.89 (m, 4H), 7.82 (d, $J = 7.5$ Hz, 1H), 7.53-7.35 (m, 8H), 5.82 (d, $J = 7.5$ Hz, 2H), 2.65 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 149.1, 146.0, 134.7, 132.8 (d, $J = 7.1$ Hz), 132.5 (d, $J = 4.4$ Hz), 132.1 (d, $J = 2.8$ Hz), 131.8 (d, $J = 10.1$ Hz), 130.8, 128.5 (d, $J = 13.1$ Hz), 127.8, 127.3, 126.7, 120.8, 63.5 (d, $J = 5.1$ Hz), 18.6; ^{31}P NMR (121 MHz, CDCl_3): δ 32.16; IR (KBr): ν 3069, 3015, 2927, 1596, 1501, 1434, 1227, 1175, 1128, 1113, 851, 826, 774, 746, 724; HRMS (ESI-TOF) m/z : $[\text{M}+\text{K}]^+$ Calcd for $\text{C}_{23}\text{H}_{20}\text{NO}_2\text{PK}$ 412.0863, Found 412.0865.

(5-fluoroquinolin-8-yl)methyl diphenylphosphinate (4c)

Yield: 68% (51.3 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.91 (d, $J = 3.0$ Hz, 1H), 8.41 (d, $J = 8.1$ Hz, 1H), 7.92-7.86 (m, 5H), 7.52-7.45 (m, 7H), 7.21 (t, $J = 8.7$ Hz, 1H), 5.78 (d, $J = 7.5$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 157.6 (d, $J_{\text{CF}} = 254.3$ Hz), 150.4, 146.2, 132.1 (d, $J = 2.8$ Hz), 131.8 (d, $J = 10.3$ Hz), 131.0, 129.4 (d, $J = 4.6$ Hz), 128.5 (d, $J = 13.1$ Hz), 128.2 (d, $J = 9.0$ Hz), 121.3 (d, $J = 2.8$ Hz), 118.7 (d, $J = 16.4$ Hz), 109.8 (d, $J = 19.3$ Hz), 62.9 (d, $J = 5.0$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ 32.56; IR (KBr): ν 3017, 2918, 2848, 1594, 1474, 1437, 1407, 1222, 1130, 863, 798, 751, 729; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{22}\text{H}_{18}\text{FNO}_2\text{P}$ 378.1054, Found 378.1057.

(5-chloroquinolin-8-yl)methyl diphenylphosphinate (4d)

Yield: 60% (47.3 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.90 (d, $J = 3.3$ Hz, 1H), 8.57 (d, $J = 8.4$ Hz, 1H), 7.94-7.87 (m, 5H), 7.63 (d, $J = 7.5$ Hz, 1H), 7.53-7.46 (m, 7H), 5.81 (d, $J = 7.5$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 150.1, 146.1, 134.3 (d, $J = 11.6$ Hz), 133.0, 132.2 (d, $J = 4.9$ Hz), 131.8 (d, $J = 17.0$ Hz), 131.2, 130.5, 128.6 (d, $J = 21.9$ Hz), 127.7, 126.4, 125.9, 122.0, 63.0 (d, $J = 8.1$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ 32.69; IR (KBr): ν 3052, 2920, 2848, 1576, 1494, 1437, 1220, 1122, 1113, 853, 816, 749, 729; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{22}\text{H}_{18}\text{ClNO}_2\text{P}$ 394.0758, Found 394.0760.

(5-bromoquinolin-8-yl)methyl diphenylphosphinate (4e)

Yield: 66% (57.9 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.84 (d, $J = 4.1$ Hz, 1H), 8.48 (t, $J = 7.5$ Hz, 1H), 7.93-7.87 (m, 4H), 7.87 (s, 2H), 7.54-7.41 (m, 7H), 5.79 (d, $J = 7.5$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 150.1, 146.1, 135.6, 135.0 (d, $J = 7.0$ Hz), 132.3 (d, $J = 2.9$ Hz), 131.8 (d, $J = 10.2$ Hz), 130.4, 130.1, 128.6 (d, $J = 13.1$ Hz), 128.1, 127.2, 122.3, 121.7, 63.0 (d, $J = 5.0$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ 32.81; IR (KBr): ν 3052, 3012, 2940, 2865, 1589, 1571, 1491, 1437, 1220, 1125, 1110, 853, 771, 736, 724; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{22}\text{H}_{17}\text{BrNO}_2\text{PNa}$ 460.0072, Found 460.0075.

(5-nitroquinolin-8-yl)methyl diphenylphosphinate (4f)

Yield: 59% (47.7 mg), Yellow oil, $R_f = 0.30$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.98 (d, $J = 8.9$ Hz, 1H), 8.90 (d, $J = 4.1$ Hz, 1H), 8.39 (d, $J = 7.8$ Hz, 1H), 8.07 (d, $J = 7.8$ Hz, 1H), 7.96-7.89 (m, 4H), 7.64-7.60 (m, 1H), 7.56-7.45 (m, 6H), 5.88 (d, $J = 7.5$ Hz, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 150.5, 144.9 (d, $J = 10.7$ Hz), 143.0 (d, $J = 6.9$ Hz), 132.5 (d, $J = 2.8$ Hz), 132.2, 131.8 (t, $J = 6.9$ Hz), 130.1, 128.7 (d, $J = 13.1$ Hz), 124.9, 124.6, 124.0, 120.6, 63.2 (d, $J = 4.7$ Hz); ^{31}P NMR (121 MHz, CDCl_3): δ 33.38; IR (KBr): ν 3104, 3052, 2945, 1589, 1506, 1439, 1404, 1230, 1192, 1130, 861, 833, 803, 721; HRMS (ESI-TOF) m/z : $[\text{M}+\text{K}]^+$ Calcd for $\text{C}_{22}\text{H}_{17}\text{N}_2\text{O}_4\text{PK}$ 443.0558, Found 443.0564.

(6-methylquinolin-8-yl)methyl diphenylphosphinate (4g)

Yield: 70% (52.3 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.83 (d, $J = 4.2$ Hz, 1H), 8.05 (d, $J = 8.3$ Hz, 1H), 7.61 (s, 1H), 7.53 (s, 1H), 7.40-7.25 (m, 9H), 7.20-7.15 (m, 2H), 6.06 (d, $J = 8.1$ Hz, 2H), 2.47 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 150.6 (d, $J = 7.1$ Hz), 148.9, 144.1, 136.1, 135.5, 133.0 (d, $J = 7.2$ Hz), 130.2, 129.7, 128.1, 127.0, 125.3 (d, $J = 1.1$ Hz), 121.4, 120.2 (d, $J = 5.0$ Hz), 67.3 (d, $J = 5.6$ Hz), 21.7; ^{31}P NMR (121 MHz, CDCl_3): δ 32.35; IR (KBr): ν 3059, 2930, 2855, 1589, 1496, 1434, 1263, 1227, 1130, 1115, 861, 828, 768, 726, 701; HRMS (ESI-TOF) m/z : $[\text{M}+\text{K}]^+$ Calcd for $\text{C}_{23}\text{H}_{20}\text{NO}_2\text{PK}$ 412.0863, Found 412.0871.

(7-methylquinolin-8-yl)methyl diphenylphosphinate (4h)

Yield: 52% (38.8 mg), Yellow oil, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300

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4 MHz, CDCl₃): δ 8.92 (d, *J* = 4.1 Hz, 1H), 8.09 (d, *J* = 8.1 Hz, 1H), 7.74 (d, *J* = 8.4 Hz, 1H), 7.40-7.34
5 (m, 2H), 7.31-7.22 (m, 8H), 7.15 (t, *J* = 6.9 Hz, 2H), 6.21 (d, *J* = 6.9 Hz, 2H), 2.60 (s, 3H); ¹³C{¹H}
6 NMR (75 MHz, CDCl₃): δ 150.7 (d, *J* = 7.1 Hz), 150.1, 149.9, 140.7, 135.9, 130.1 (d, *J* = 7.7 Hz),
7 129.7 (d, *J* = 9.8 Hz), 128.9, 126.5, 125.1, 120.6, 120.2 (d, *J* = 4.9 Hz), 63.1 (d, *J* = 5.9 Hz), 19.7; ³¹P
8 NMR (121 MHz, CDCl₃): δ 31.79; IR (KBr): ν 3055, 2922, 2850, 1501, 1464, 1437, 1222, 1130, 1110,
9 876, 833, 793, 729; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₃H₂₁NO₂P 374.1304, Found
10 374.1309.

11 12 13 14 **Methyl 3-benzoylimidazo[1,2-a]pyridine-7-carboxylate (4i)**

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16 Yield: 60% (44.8 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300
17 MHz, CDCl₃): δ 8.68 (s, 1H), 7.95-7.85 (m, 6H), 7.66 (d, *J* = 8.1 Hz, 1H), 7.51-7.42 (m, 7H), 5.83 (d, *J*
18 = 7.2 Hz, 2H), 2.45 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 151.5, 143.8, 134.8, 134.5 (d, *J* = 7.1
19 Hz), 132.4, 132.2 (d, *J* = 2.8 Hz), 131.8 (d, *J* = 10.2 Hz), 130.6 (d, *J* = 6.2 Hz), 128.5 (d, *J* = 13.1 Hz),
20 127.8, 127.2, 126.8, 126.3, 63.4 (d, *J* = 5.1 Hz), 18.7; ³¹P NMR (121 MHz, CDCl₃): δ 32.46; IR (KBr):
21 ν 3049, 2930, 2862, 1589, 1499, 1482, 1437, 1227, 1128, 1110, 896, 838, 821, 769, 729; HRMS
22 (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₃H₂₁NO₂P 374.1304, Found 374.1313.

23 24 25 26 **(3-ethylquinolin-8-yl)methyl diphenylphosphinate (4j)**

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28 Yield: 65% (50.4 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300
29 MHz, CDCl₃): δ 8.73 (d, *J* = 1.8 Hz, 1H), 7.95-7.87 (m, 6H), 7.70 (d, *J* = 8.1 Hz, 1H), 7.53-7.40 (m,
30 7H), 5.85 (d, *J* = 6.9 Hz, 2H), 2.84-2.77 (m, 2H), 1.32 (t, *J* = 7.5 Hz, 3H); ¹³C{¹H} NMR (75 MHz,
31 CDCl₃): δ 151.0, 144.1, 136.8, 134.5 (d, *J* = 7.1 Hz), 133.5, 132.4, 132.2 (d, *J* = 2.8 Hz), 131.8 (d, *J* =
32 10.1 Hz), 130.6, 128.5 (d, *J* = 13.1 Hz), 127.9, 127.4, 126.9, 126.3, 63.4 (d, *J* = 5.1 Hz), 26.2, 15.2; ³¹P
33 NMR (121 MHz, CDCl₃): δ 32.27; IR (KBr): ν 3050, 2957, 2895, 1589, 1486, 1444, 1215, 1130, 1113,
34 868, 729, 701; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₂₄H₂₃NO₂P 388.1461, Found 388.1469.

35 36 37 38 **(3-phenylquinolin-8-yl)methyl diphenylphosphinate (4k)**

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40 Yield: 50% (43.6 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300
41 MHz, CDCl₃): δ 9.14 (d, *J* = 1.8 Hz, 1H), 8.30 (d, *J* = 2.1 Hz, 1H), 7.98-7.92 (m, 5H), 7.85 (d, *J* = 8.1
42 Hz, 1H), 7.72 (d, *J* = 7.5 Hz, 2H), 7.63-7.44 (m, 10H), 5.90 (d, *J* = 7.2 Hz, 2H); ¹³C{¹H} NMR (75
43 MHz, CDCl₃): δ 149.1, 144.7, 137.8, 134.8 (d, *J* = 7.1 Hz), 133.9, 133.3, 132.5, 132.2 (d, *J* = 2.8 Hz),
44 131.8 (d, *J* = 10.1 Hz), 130.7, 129.2, 128.6 (d, *J* = 13.1 Hz), 128.1 (d, *J* = 11.9 Hz), 127.8 (d, *J* = 10.2
45 Hz), 127.4, 126.8, 63.3 (d, *J* = 5.0 Hz); ³¹P NMR (121 MHz, CDCl₃): δ 32.44; IR (KBr): ν 3054, 2922,
46 2865, 1584, 1484, 1437, 1275, 1222, 1128, 1110, 866, 769, 729; HRMS (ESI-TOF) *m/z*: [M+H]⁺
47 Calcd for C₂₈H₂₃NO₂P 436.1461, Found 436.1470.

48 49 50 51 **1-(quinolin-8-yl)ethyl diphenylphosphinate (4l)**

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53 Yield: 59% (44.1 mg), Colorless oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300
54 MHz, CDCl₃): δ 8.75 (d, *J* = 1.8 Hz, 1H), 8.08 (d, *J* = 4.2 Hz, 1H), 8.01 (d, *J* = 3.6 Hz, 1H), 7.91-7.88
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(m, 2H), 7.73-7.70 (m, 3H), 7.56 (t, $J = 3.9$ Hz, 1H), 7.51 (t, $J = 3.6$ Hz, 1H), 7.46-7.43 (m, 2H), 7.36-7.31 (m, 2H), 7.26-7.23 (m, 2H), 6.81-6.76 (m, 1H), 1.79 (d, $J = 3.3$ Hz, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3): δ 149.3, 144.5, 141.1 (d, $J = 4.6$ Hz), 136.0, 133.3, 132.7, 132.0 (d, $J = 2.8$ Hz), 131.8 (d, $J = 2.6$ Hz), 131.7 (d, $J = 2.6$ Hz), 131.5, 130.9, 128.5, 128.3 (d, $J = 6.2$ Hz), 128.1, 128.0, 127.4, 126.4, 125.7, 121.0, 70.9 (d, $J = 5.3$ Hz), 25.2; ^{31}P NMR (121 MHz, CDCl_3): δ 30.57; IR (KBr): ν 3060, 2972, 2855, 1594, 1576, 1499, 1437, 1250, 1160, 1127, 826, 811, 754, 724; HRMS (ESI-TOF) m/z : $[\text{M}+\text{Na}]^+$ Calcd for $\text{C}_{23}\text{H}_{20}\text{NO}_2\text{PNa}$ 396.1124, Found 396.1121.

General Procedure for the Synthesis of **4a** in a 2.0 mmol Scale.

A oven-dried 100 mL Schlenk tube was charged with a mixture of **1a** (2.0 mmol), $\text{Pd}(\text{OAc})_2$ (10 mol%), oxidant (4.8 mmol), 4Å MS (1750 mg) in dioxane (10.0 mL) and was stirred at room temperature for 8 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 100 mL of water was added. Then, the mixture was extracted with ethyl acetate (30 mL \times 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na_2SO_4 . The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 4:1) to afford the targeted product **4a** in 80% yield (573 mg).

Procedure B: A 15 mL Schlenk tube was charged with a mixture of **5** (0.2 mmol), $\text{Pd}(\text{OAc})_2$ (10 mol%), oxidant (0.60 mmol), C_2CO_3 (0.3 mmol), PivOH (0.06 mmol) in DCE (1.0 mL) and was stirred in a 130 °C oil-bath for 10-15 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL \times 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na_2SO_4 . The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 2:1) to afford the targeted products **6a-6f**.

2-(pyridin-2-yl)phenol (**6a**)¹⁵

Yield: 56% (19.2 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ^1H NMR (300 MHz, CDCl_3): δ 14.40 (br s, 1H), 8.53 (d, $J = 4.2$ Hz, 1H), 7.94 (d, $J = 8.4$ Hz, 1H), 7.88-7.81 (m, 2H), 7.36-7.25 (m, 2H), 7.06 (d, $J = 8.4$ Hz, 1H), 6.93 (d, $J = 6.9$ Hz, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 160.0, 157.9, 145.9, 137.7, 131.5, 126.1, 121.5, 119.1, 118.8, 118.7, 118.6.

5-fluoro-2-(pyridin-2-yl)phenol (**6b**)¹⁵

Yield: 67% (25.4 mg), Mp 102-103 °C, Yellow solid, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ^1H NMR (300 MHz, CDCl_3): δ 14.84 (br s, 1H), 8.50 (d, $J = 4.8$ Hz, 1H), 7.86-7.75 (m, 3H), 7.26 (t, $J = 3.3$ Hz, 1H), 6.74 (d, $J = 10.5$ Hz, 1H), 6.68-6.61 (m, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 165.7, 163.7, 162.1, 162.0, 157.3, 145.7, 137.9, 127.6, 127.5, 121.4, 118.8, 115.4, 106.4, 106.2, 105.3, 105.1.

5-methyl-2-(pyridin-2-yl)phenol (6c)¹⁵

Yield: 54% (20.0 mg), Colorless oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ¹H NMR (300 MHz, CDCl₃): δ 14.36 (br s, 1H), 8.50 (d, $J = 3.9$ Hz, 1H), 7.91-7.80 (m, 2H), 7.70 (d, $J = 8.1$ Hz, 1H), 7.22 (t, $J = 6.0$ Hz, 1H), 6.87 (s, 1H), 6.75 (d, $J = 7.8$ Hz, 1H), 2.37 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 159.9, 158.0, 145.8, 142.1, 137.6, 125.9, 121.0, 119.9, 118.9, 118.7, 116.3, 21.4.

5-ethyl-2-(pyridin-2-yl)phenol (6d)¹⁵

Yield: 57% (22.7 mg), Yellow oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ¹H NMR (300 MHz, CDCl₃): δ 14.35 (br s, 1H), 8.51 (d, $J = 3.3$ Hz, 1H), 7.91-7.80 (m, 2H), 7.73 (d, $J = 8.1$ Hz, 1H), 7.23 (t, $J = 5.1$ Hz, 1H), 6.90 (s, 1H), 6.78 (d, $J = 7.8$ Hz, 1H), 2.70-2.63 (m, 2H), 1.28 (t, $J = 7.5$ Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 160.0, 158.0, 148.4, 145.8, 137.6, 126.0, 121.0, 118.7, 117.6, 116.5, 28.7, 15.1.

5-methoxy-2-(pyridin-2-yl)phenol (6e)¹⁵

Yield: 48% (19.3 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ¹H NMR (300 MHz, CDCl₃): δ 14.76 (br s, 1H), 8.46 (d, $J = 3.6$ Hz, 1H), 7.80 (t, $J = 3.3$ Hz, 2H), 7.72 (d, $J = 9.0$ Hz, 1H), 7.20-7.16 (m, 1H), 6.57-6.50 (m, 2H), 3.85 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 162.4, 161.9, 157.9, 145.6, 137.6, 127.1, 120.5, 118.2, 112.1, 106.6, 102.2, 55.3.

4-(pyridin-2-yl)-[1,1'-biphenyl]-3-ol (6f)¹⁵

Yield: 51% (25.2 mg), Mp 144-145 °C, White solid, $R_f = 0.40$ (petroleum ether/ethyl acetate = 2:1); ¹H NMR (300 MHz, CDCl₃): δ 14.51 (br s, 1H), 8.55 (d, $J = 3.9$ Hz, 1H), 7.97 (d, $J = 8.4$ Hz, 1H), 7.89 (d, $J = 8.4$ Hz, 2H), 7.69 (d, $J = 7.5$ Hz, 2H), 7.48 (t, $J = 6.9$ Hz, 2H), 7.40 (d, $J = 6.6$ Hz, 1H), 7.30 (d, $J = 9.9$ Hz, 2H), 7.20 (d, $J = 8.1$ Hz, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 160.3, 157.7, 145.9, 144.2, 140.3, 137.8, 128.8, 127.7, 127.0, 126.5, 121.4, 119.0, 117.8, 117.7, 116.9.

Procedure C: A oven-dried 15 mL Schlenk tube was charged with a mixture of **1** (0.2 mmol), Pd(OAc)₂ (10 mol%), oxidant (0.60 mmol) in dioxane (1.0 mL) and was stirred at room temperature for 12-16 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 4:1) to afford the targeted products **7a-7r**.

(5-bromoquinolin-8-yl)methyl 4-methylbenzenesulfonate (7a)

Yield: 79% (62.0 mg), Mp 114-116 °C, White solid, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.85 (d, $J = 4.2$ Hz, 1H), 8.52 (d, $J = 8.4$ Hz, 1H), 7.83 (t, $J = 8.4$ Hz, 3H), 7.67 (d, $J = 7.8$ Hz, 1H), 7.55-7.50 (m, 1H), 7.31 (d, $J = 8.1$ Hz, 2H), 5.78 (s, 2H), 2.44 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 150.4, 146.1, 144.7, 135.6, 133.2, 132.2, 130.0, 129.7, 128.9, 128.1, 127.3, 122.6, 122.5, 68.2, 21.6; IR (KBr): ν 3087, 2920, 2853, 1586, 1491, 1459, 1439, 1295, 1200,

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1120, 871, 813, 778, 759; HRMS (ESI-TOF) m/z : $[M+Na]^+$ Calcd for $C_{17}H_{14}BrNO_3SNa$ 413.9770, Found 413.9766.

(5-fluoroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7b)

Yield: 40% (26.5 mg), Colorless oil, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.88 (s, 1H), 8.41 (d, $J = 8.1$ Hz, 1H), 7.85 (d, $J = 7.8$ Hz, 2H), 7.75 (t, $J = 6.9$ Hz, 1H), 7.50-7.46 (m, 1H), 7.30 (t, $J = 4.8$ Hz, 2H), 7.20 (t, $J = 8.7$ Hz, 1H), 5.76 (s, 2H), 2.44 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$): δ 158.1 (d, $J_{CF} = 255.8$ Hz), 150.7, 146.2, 144.6, 133.5, 129.6, 129.5, 129.2, 129.1, 128.1, 121.5, 118.9, 109.8, 109.7, 68.1, 21.6; IR (KBr): ν 2922, 2850, 1556, 1491, 1477, 1409, 1205, 1180, 1122, 871, 843, 816, 786; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{17}H_{15}FNO_3S$ 332.0751, Found 332.0745.

(5-chloroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7c)

Yield: 59% (41.0 mg), Pale yellow oil, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.87 (t, $J = 2.7$ Hz, 1H), 8.55 (t, $J = 7.2$ Hz, 1H), 7.84 (d, $J = 8.1$ Hz, 2H), 7.73 (d, $J = 7.8$ Hz, 1H), 7.60 (d, $J = 7.8$ Hz, 1H), 7.55-7.50 (m, 1H), 7.31 (d, $J = 8.1$ Hz, 2H), 5.78 (s, 2H), 2.44 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$): δ 150.3, 146.0, 144.7, 133.3, 133.0, 132.0, 131.4, 129.7, 128.5, 128.1, 126.2, 126.0, 122.2, 68.2, 21.6; IR (KBr): ν 3092, 2920, 2853, 1591, 1571, 1496, 1449, 1290, 1172, 1120, 858, 826, 778, 731; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{17}H_{15}ClNO_3S$ 348.0456, Found 348.0452.

(5-iodoquinolin-8-yl)methyl 4-methylbenzenesulfonate (7d)

Yield: 61% (53.6 mg), Pale yellow oil, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.84 (d, $J = 2.7$ Hz, 1H), 8.39 (d, $J = 8.7$ Hz, 1H), 8.11 (d, $J = 7.8$ Hz, 1H), 7.83 (d, $J = 8.1$ Hz, 2H), 7.55-7.49 (m, 2H), 7.29 (d, $J = 7.5$ Hz, 2H), 5.78 (s, 2H), 2.43 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$): δ 150.3, 144.7, 141.2, 137.6, 133.3, 133.0, 129.9, 129.7, 128.9, 128.0, 126.1, 123.0, 99.1, 68.0, 21.6; IR (KBr): ν 3090, 2918, 2850, 1594, 1559, 1494, 1447, 1290, 1165, 1120, 861, 811, 778, 719; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{17}H_{15}INO_3S$ 439.9812, Found 439.9808.

(5-nitroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7e)

Yield: 84% (60.2 mg), Mp 94-96 °C, Yellow solid, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 9.00-8.94 (m, 2H), 8.36 (d, $J = 8.1$ Hz, 1H), 7.94-7.88 (m, 3H), 7.69-7.64 (m, 1H), 7.36 (d, $J = 8.1$ Hz, 2H), 5.86 (s, 2H), 2.46 (s, 3H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 150.7, 145.2, 144.9, 140.1, 132.8, 132.2, 129.9, 128.1, 125.4, 124.3, 124.2, 120.7, 67.9, 21.7; IR (KBr): ν 3104, 2922, 2853, 1594, 1521, 1504, 1442, 1245, 1192, 1170, 1157, 858, 811, 778, 731; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{17}H_{15}N_2O_5S$ 359.0696, Found 359.0699.

(6-fluoroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7f)

Yield: 72% (47.7 mg), Mp 103-104 °C, White solid, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.79 (t, $J = 2.1$ Hz, 1H), 8.09 (d, $J = 8.4$ Hz, 1H), 7.89 (d, $J = 8.1$ Hz,

2H), 7.60 (d, $J = 8.1$ Hz, 1H), 7.45-7.33 (m, 4H), 5.81 (s, 2H), 2.45 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 160.0 (d, $J_{\text{CF}} = 247.3$ Hz), 148.9, 144.9, 142.5, 135.5, 135.4, 133.1, 129.8, 128.8, 128.7, 128.1, 122.2, 118.4, 118.2, 110.9, 110.8, 67.7, 21.6; IR (KBr): ν 3065, 2925, 2853, 1594, 1561, 1499, 1447, 1290, 1172, 1127, 878, 813, 783, 761; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{17}\text{H}_{15}\text{FNO}_3\text{S}$ 332.0751, Found 332.0750.

(6-chloroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7g)

Yield: 66% (45.9 mg), Mp 123-124 °C, Pale yellow solid, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.83 (d, $J = 2.4$ Hz, 1H), 8.06 (d, $J = 8.1$ Hz, 1H), 7.88 (d, $J = 8.1$ Hz, 2H), 7.74 (d, $J = 12.9$ Hz, 2H), 7.47-7.42 (m, 1H), 7.34 (d, $J = 7.8$ Hz, 2H), 5.79 (s, 2H), 2.45 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 149.9, 144.8, 143.8, 135.2, 134.3, 133.2, 132.2, 129.8, 128.9, 128.6, 128.1, 126.7, 122.3, 67.7, 21.6; IR (KBr): ν 3060, 2925, 2853, 1591, 1559, 1491, 1442, 1240, 1212, 1120, 873, 816, 783, 751; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{17}\text{H}_{14}\text{ClNO}_3\text{S}$ 348.0456, Found 348.0459.

(6-bromoquinolin-8-yl)methyl 4-methylbenzenesulfonate (7h)

Yield: 57% (44.7 mg), Mp 137-138 °C, White solid, $R_f = 0.50$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 8.84 (t, $J = 2.4$ Hz, 1H), 8.05 (d, $J = 8.4$ Hz, 1H), 7.94 (s, 1H), 7.85 (t, $J = 10.2$ Hz, 3H), 7.46-7.42 (m, 1H), 7.33 (d, $J = 8.1$ Hz, 2H), 5.79 (s, 2H), 2.45 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 150.0, 144.8, 144.1, 135.1, 134.3, 133.2, 131.4, 130.2, 129.8, 129.1, 128.1, 122.3, 120.2, 67.6, 21.6; IR (KBr): ν 3065, 2918, 2848, 1596, 1584, 1494, 1449, 1292, 1232, 1182, 1135, 861, 811, 783, 749; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{17}\text{H}_{15}\text{BrNO}_3\text{S}$ 391.9951, Found 391.9947.

(6-nitroquinolin-8-yl)methyl 4-methylbenzenesulfonate (7i)

Yield: 75% (53.8 mg), Mp 140-141 °C, Yellow solid, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 9.04 (d, $J = 2.4$ Hz, 1H), 8.76 (d, $J = 2.1$ Hz, 1H), 8.50 (s, 1H), 8.37 (d, $J = 8.1$ Hz, 1H), 7.91 (d, $J = 8.4$ Hz, 2H), 7.64-7.59 (m, 1H), 7.36 (d, $J = 8.1$ Hz, 2H), 5.86 (s, 2H), 2.46 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 153.1, 147.3, 145.2, 145.1, 138.0, 135.2, 133.0, 129.9, 128.1, 126.9, 124.6, 123.2, 121.3, 67.3, 21.6; IR (KBr): ν 3097, 2922, 2848, 1591, 1534, 1491, 1444, 1240, 1210, 1180, 1137, 881, 846, 788, 741; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{17}\text{H}_{15}\text{N}_2\text{O}_5\text{S}$ 359.0696, Found 359.0702.

(5-nitroquinolin-8-yl)methyl 4-nitrobenzenesulfonate (7j)

Yield: 42% (32.7 mg), Mp 130-132 °C, Yellow solid, $R_f = 0.40$ (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300 MHz, CDCl_3): δ 9.00 (d, $J = 8.7$ Hz, 1H), 8.93 (d, $J = 2.4$ Hz, 1H), 8.41 (d, $J = 9.0$ Hz, 3H), 8.22 (d, $J = 8.7$ Hz, 2H), 7.96 (d, $J = 8.1$ Hz, 1H), 7.71-7.67 (m, 1H), 5.96 (s, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 150.9, 145.8, 145.1, 141.6, 138.8, 132.4, 129.5, 126.1, 124.5, 124.4, 124.1, 120.8, 68.9; IR (KBr): ν 3104, 2920, 2850, 1546, 1519, 1496, 1437, 1240, 1177, 1152, 1108, 871, 856, 811, 739; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{12}\text{N}_3\text{O}_7\text{S}$ 390.0390, Found 390.0394.

(5-nitroquinolin-8-yl)methyl 4-chlorobenzenesulfonate (7k)

1
2
3 Yield: 68% (51.5 mg), Mp 98-100 °C, Yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1);
4
5 ^1H NMR (300 MHz, CDCl_3): δ 9.01 (d, J = 8.7 Hz, 1H), 8.94 (d, J = 3.6 Hz, 1H), 8.38 (d, J = 7.8 Hz,
6
7 1H), 7.95 (d, J = 8.4 Hz, 3H), 7.70-7.66 (m, 1H), 7.55 (d, J = 8.4 Hz, 2H), 5.89 (s, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR
8
9 (125 MHz, CDCl_3): δ 150.8, 145.5, 145.0, 140.8, 139.5, 134.3, 132.3, 125.7, 124.3, 124.2, 120.8, 68.3;
10
11 IR (KBr): ν 3095, 2925, 2853, 1589, 1531, 1496, 1439, 1280, 1185, 1160, 1115, 861, 831, 801, 756;
12
13 HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{12}\text{ClN}_2\text{O}_5\text{S}$ 379.0150, Found 379.0150.

14 **(5-nitroquinolin-8-yl)methyl benzenesulfonate (7l)**

15 Yield: 61% (42.0 mg), Yellow oil, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1); ^1H NMR (300
16
17 MHz, CDCl_3): δ 9.01 (d, J = 8.7 Hz, 1H), 8.94 (t, J = 2.4 Hz, 1H), 8.37 (d, J = 8.1 Hz, 1H), 8.02 (d, J =
18
19 7.5 Hz, 2H), 7.94 (d, J = 7.8 Hz, 1H), 7.68 (t, J = 6.0 Hz, 2H), 7.58 (t, J = 7.8 Hz, 2H), 5.90 (s, 2H);
20
21 $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 150.7, 145.3, 145.0, 139.9, 135.9, 134.0, 132.3, 129.3, 128.1,
22
23 125.5, 124.3, 124.2, 120.7, 68.1; IR (KBr): ν 3107, 2920, 2848, 1589, 1519, 1501, 1447, 1245, 1185,
24
25 1157, 848, 801, 759, 734; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{16}\text{H}_{13}\text{N}_2\text{O}_5\text{S}$ 345.0540, Found
26
27 345.0540.

28 **(5-nitroquinolin-8-yl)methyl methanesulfonate (7m)**

29 Yield: 54% (30.5 mg), Mp 130-132 °C, Yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1);
30
31 ^1H NMR (300 MHz, CDCl_3): δ 9.05 (t, J = 8.7 Hz, 2H), 8.42 (d, J = 7.8 Hz, 1H), 8.00 (d, J = 7.5 Hz,
32
33 1H), 7.75-7.70 (m, 1H), 6.05 (s, 2H), 3.20 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 151.0, 145.6,
34
35 145.3, 139.7, 132.4, 126.1, 124.3, 120.9, 67.6, 37.8; IR (KBr): ν 3012, 2920, 2848, 1591, 1516, 1464,
36
37 1409, 1260, 1160, 866, 803, 776, 724; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{11}\text{H}_{11}\text{N}_2\text{O}_5\text{S}$
38
39 283.0383, Found 283.0383.

40 **(5-nitroquinolin-8-yl)methyl (7,7-dimethyl-2-oxobicyclo[2.2.1]heptan-1-yl)methanesulfonate (7n)**

41 Yield: 65% (54.4 mg), Mp 116-118 °C, Yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 4:1);
42
43 ^1H NMR (300 MHz, CDCl_3): δ 9.05 (t, J = 5.4 Hz, 2H), 8.43 (d, J = 8.1 Hz, 1H), 8.00 (d, J = 8.1 Hz,
44
45 1H), 7.74-7.70 (m, 1H), 6.13 (s, 2H), 3.85 (d, J = 15.0 Hz, 1H), 3.19 (d, J = 15.0 Hz, 1H), 2.61-2.52 (m,
46
47 1H), 2.43 (d, J = 18.6 Hz, 1H), 2.17-2.07 (m, 2H), 1.99 (d, J = 18.6 Hz, 1H), 1.80-1.71 (m, 1H),
48
49 1.53-1.44 (m, 1H), 1.16 (s, 3H), 0.93 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3): δ 214.5, 150.9, 145.4,
50
51 145.2, 140.3, 132.3, 125.9, 124.4, 124.3, 120.8, 68.2, 58.1, 48.1, 47.4, 42.8, 42.5, 26.9, 25.0, 19.8, 19.7;
52
53 IR (KBr): ν 3102, 2947, 2918, 2848, 1587, 1519, 1452, 1409, 1245, 1167, 1152, 1130, 836, 813, 793,
54
55 741; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ Calcd for $\text{C}_{20}\text{H}_{23}\text{N}_2\text{O}_6\text{S}$ 419.1271, Found 419.1284.

56 **(5-bromoquinolin-8-yl)methyl benzenesulfonate (7o)**

57 Yield: 41% (31.0 mg), Mp 98-99 °C, White solid, R_f = 0.50 (petroleum ether / ethyl acetate = 4:1); ^1H
58
59 NMR (300 MHz, CDCl_3): δ 8.84 (d, J = 2.7 Hz, 1H), 8.52 (d, J = 8.4 Hz, 1H), 7.97 (d, J = 7.5 Hz, 2H),
60
61 7.82 (d, J = 7.8 Hz, 1H), 7.68-7.61 (m, 2H), 7.55-7.50 (m, 3H), 5.81 (s, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz,
62
63 CDCl_3): δ 150.4, 146.1, 136.3, 135.7, 133.7, 132.0, 129.9, 129.1, 129.0, 128.0, 127.4, 122.8, 122.6,
64
65 68.4; IR (KBr): ν 3060, 2922, 2848, 1586, 1571, 1491, 1444, 1230, 1192, 1175, 853, 826, 759, 736;

HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{16}H_{13}BrNO_3S$ 377.9794, Found 377.9791.

(5-bromoquinolin-8-yl)methyl methanesulfonate (7p)

Yield: 56% (35.4 mg), Mp 94-96 °C, White solid, R_f = 0.50 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.97 (d, J = 2.1 Hz, 1H), 8.60 (d, J = 8.4 Hz, 1H), 7.89 (d, J = 7.8 Hz, 1H), 7.75 (d, J = 7.5 Hz, 1H), 7.61-7.57 (m, 1H), 5.94 (s, 2H), 3.15 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$): δ 150.7, 146.4, 135.9, 131.9, 130.1, 129.9, 127.6, 123.3, 122.7, 68.1, 37.8; IR (KBr): ν 3007, 2930, 2850, 1569, 1491, 1444, 1422, 1262, 1197, 1165, 866, 821, 776, 721; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{11}H_{11}BrNO_3S$ 315.9638, Found 315.9642.

(5-bromoquinolin-8-yl)methyl (7,7-dimethyl-2-oxobicyclo[2.2.1]heptan-1-yl)methanesulfonate (7q)

Yield: 57% (51.6 mg), Mp 108-110 °C, White solid, R_f = 0.50 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.97 (s, 1H), 8.58 (d, J = 8.4 Hz, 1H), 7.88 (d, J = 7.5 Hz, 1H), 7.75 (d, J = 6.9 Hz, 1H), 7.58 (d, J = 4.2 Hz, 1H), 6.00 (t, J = 12.3 Hz, 2H), 3.85 (d, J = 15.0 Hz, 1H), 3.16 (d, J = 15.0 Hz, 1H), 2.57-2.38 (m, 2H), 2.13-1.93 (m, 3H), 1.80-1.71 (m, 1H), 1.46 (d, J = 8.7 Hz, 1H), 1.13 (s, 3H), 0.90 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$): δ 214.5, 150.7, 146.4, 135.8, 132.3, 130.1, 129.7, 127.6, 123.1, 122.7, 68.3, 58.1, 48.0, 47.3, 42.8, 42.5, 26.9, 25.0, 19.9, 19.7; IR (KBr): ν 3080, 2952, 2915, 1571, 1494, 1452, 1407, 1227, 1170, 1147, 1113, 846, 788, 754; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{20}H_{23}BrNO_4S$ 452.0526, Found 452.0521.

(5-bromoquinolin-8-yl)methyl 4-chlorobenzenesulfonate (7r)

Yield: 22% (18.2 mg), Colorless oil, R_f = 0.50 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.84 (d, J = 3.3 Hz, 1H), 8.54 (d, J = 8.4 Hz, 1H), 7.89-7.82 (m, 3H), 7.67 (d, J = 7.5 Hz, 1H), 7.56-7.52 (m, 1H), 7.48 (d, J = 8.1 Hz, 2H), 5.81 (s, 2H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 150.5, 140.3, 135.8, 134.8, 131.6, 129.9, 129.4, 129.3, 127.4, 123.1, 122.6, 68.7; IR (KBr): ν 3090, 2922, 2848, 1586, 1559, 1494, 1474, 1297, 1227, 1170, 1120, 828, 778, 754, 706; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{16}H_{12}BrClNO_3S$ 411.9404, Found 411.9406.

Procedure D: A oven-dried 15 mL Schlenk tube was charged with a mixture of **1** (0.2 mmol), $Pd(OAc)_2$ (10 mol%), oxidant (0.60 mmol) in dioxane (1.0 mL) and was stirred at room temperature for 12-16 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL \times 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na_2SO_4 . The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 4:1) to afford the targeted products **8a-8g**.

(5-bromoquinolin-8-yl)methanol (8a)

Yield: 69% (32.9 mg), Mp 113-115 °C, Pale yellow solid, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.91 (d, J = 4.2 Hz, 1H), 8.60 (d, J = 8.4 Hz, 1H), 7.80 (d, J = 7.8 Hz, 1H), 7.59-7.55 (m, 1H), 7.49 (d, J = 7.8 Hz, 1H), 5.17 (s, 2H), 4.76 (br s, 1H); $^{13}C\{^1H\}$ NMR (75

MHz, CDCl₃): δ 149.7, 147.7, 138.3, 136.3, 130.1, 128.1, 127.8, 122.3, 121.3, 64.2; IR (KBr): ν 3361, 3055, 2920, 2850, 1559, 1494, 1467, 1429, 1287, 1247, 1147, 1113, 876, 831, 778, 729; HRMS (ESI-TOF) m/z : [M+H]⁺ Calcd for C₁₀H₉BrNO 237.9862, Found 237.9872.

Quinolin-8-ylmethanol (8b)

Yield: 46% (14.6 mg), Mp 68-70 °C, Yellow solid, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, DMSO): δ 8.89 (d, J = 4.2 Hz, 1H), 8.36 (d, J = 8.1 Hz, 1H), 7.84 (t, J = 7.5 Hz, 2H), 7.63-7.53 (m, 2H), 5.17 (s, 2H); ¹³C{¹H} NMR (75 MHz, DMSO): δ 149.9, 145.3, 140.3, 136.9, 127.9, 126.9, 126.8, 126.6, 121.8, 60.0; IR (KBr): ν 3299, 2920, 2868, 1594, 1576, 1499, 1447, 1295, 1232, 1757, 1127, 866, 821, 783, 759; HRMS (ESI-TOF) m/z : [M+H]⁺ Calcd for C₁₀H₁₀NO 160.0757, Found 160.0754.

(5-methylquinolin-8-yl)methanol (8c)

Yield: 49% (17.0 mg), Pale yellow oil, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.88 (d, J = 4.2 Hz, 1H), 8.37 (d, J = 8.4 Hz, 1H), 7.50-7.46 (m, 2H), 7.32 (d, J = 7.2 Hz, 1H), 5.17 (s, 2H), 2.68 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 148.5, 147.4, 136.1, 134.2, 133.3, 127.9, 127.5, 126.7, 120.7, 64.8, 18.5; IR (KBr): ν 3366, 3207, 2963, 2918, 2850, 1599, 1501, 1457, 1409, 1292, 1235, 1210, 1152, 838, 811, 776, 709; HRMS (ESI-TOF) m/z : [M+H]⁺ Calcd for C₁₁H₁₂NO 174.0913, Found 174.0915.

(6-methylquinolin-8-yl)methanol (8d)

Yield: 68% (23.6 mg), Pale yellow oil, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.81 (d, J = 4.2 Hz, 1H), 8.12 (d, J = 8.4 Hz, 1H), 7.53 (s, 1H), 7.44-7.40 (m, 2H), 5.19 (s, 2H), 2.53 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 148.1, 145.8, 137.7, 136.3, 136.1, 130.1, 128.7, 126.0, 121.2, 64.8, 21.5; IR (KBr): ν 3361, 3259, 2920, 2848, 1584, 1494, 1467, 1439, 1247, 1205, 1120, 881, 856, 776, 701; HRMS (ESI-TOF) m/z : [M+H]⁺ Calcd for C₁₁H₁₂NO 174.0913, Found 174.0912.

(7-methylquinolin-8-yl)methanol (8e)

Yield: 35% (12.1 mg), Yellow oil, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.82 (d, J = 3.9 Hz, 1H), 8.13 (d, J = 8.4 Hz, 1H), 7.64 (d, J = 8.4 Hz, 1H), 7.39-7.35 (m, 2H), 5.28 (s, 2H), 2.56 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 148.6, 147.6, 136.8, 136.5, 134.8, 130.2, 127.0, 126.6, 120.2, 60.5, 19.6; IR (KBr): ν 3426, 2957, 2918, 2850, 1596, 1556, 1504, 1457, 1270, 1180, 1122, 896, 873, 836, 798; HRMS (ESI-TOF) m/z : [M+H]⁺ Calcd for C₁₁H₁₂NO 174.0913, Found 174.0911.

(3-ethylquinolin-8-yl)methanol (8f)

Yield: 46% (17.2 mg), Yellow oil, R_f = 0.30 (petroleum ether / ethyl acetate = 4:1); ¹H NMR (300 MHz, CDCl₃): δ 8.76 (d, J = 2.4 Hz, 1H), 7.97 (d, J = 1.8 Hz, 1H), 7.72 (d, J = 8.4 Hz, 1H), 7.54-7.44 (m, 2H), 5.21 (s, 2H), 2.90-2.83 (m, 2H), 1.37 (t, J = 7.5 Hz, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ

150.4, 145.6, 137.8, 136.8, 134.2, 128.5, 127.0, 126.8, 126.4, 64.8, 26.3, 15.2; IR (KBr): ν 3414, 2965, 2922, 2850, 1584, 1494, 1457, 1419, 1247, 1177, 1155, 893, 871, 768, 716; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{12}H_{14}NO$ 188.1070, Found 188.1066.

1-(quinolin-8-yl)ethan-1-ol (8g)

Yield: 51% (17.7 mg), Yellow oil, $R_f = 0.30$ (petroleum ether / ethyl acetate = 4:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.87 (d, $J = 4.2$ Hz, 1H), 8.21 (d, $J = 8.4$ Hz, 1H), 7.75 (d, $J = 7.8$ Hz, 1H), 7.60 (d, $J = 6.3$ Hz, 1H), 7.51 (t, $J = 7.8$ Hz, 1H), 7.47-7.42 (m, 1H), 6.20 (br s, 1H), 5.52-5.45 (m, 1H), 1.76 (d, $J = 6.6$ Hz, 3H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 148.4, 146.5, 141.8, 137.1, 128.7, 127.1, 126.4, 120.9, 70.2, 24.1; IR (KBr): ν 3416, 2967, 2922, 2848, 1596, 1576, 1496, 1447, 1277, 1237, 1162, 1113, 876, 826, 791, 761; HRMS (ESI-TOF) m/z : $[M+H]^+$ Calcd for $C_{11}H_{12}NO$ 174.0913, Found 174.0916.

Procedure E: A 15 mL Schlenk tube was charged with a mixture of **5** (0.2 mmol), $Pd(OAc)_2$ (10 mol%), ligand (0.04 mmol), sulphonate hypervalent-iodine (0.6 mmol), $AgOAc$ (0.1 mmol) in toluene (1.0 mL) and was stirred in a 100 °C oil-bath for 12-18 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL \times 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na_2SO_4 . The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 8:1) to afford the targeted product **9a-9h**.

2-([1,1'-biphenyl]-2-yl)pyridine (9a)¹⁶

Yield: 63% (29.1 mg), Mp 73-75 °C, Yellow solid, $R_f = 0.40$ (petroleum ether/ethyl acetate = 8:1); 1H NMR (300 MHz, DMSO): δ 8.55 (d, $J = 4.8$ Hz, 1H), 7.61-7.48 (m, 4H), 7.44-7.41 (m, 1H), 7.26-7.20 (m, 4H), 7.09 (t, $J = 4.8$ Hz, 2H), 6.94 (d, $J = 7.8$ Hz, 1H); $^{13}C\{^1H\}$ NMR (75 MHz, DMSO): δ 159.1, 149.7, 141.3, 140.6, 139.7, 136.1, 130.9, 130.8, 129.7, 129.0, 128.6, 128.0, 127.2, 125.3, 122.2; HRMS (ESI): Exact mass calcd for $C_{17}H_{13}N$ $[M+H]^+$, 232.1121; Found: 232.1127.

2-(4-methyl-[1,1'-biphenyl]-2-yl)pyridine (9b)¹⁶

Yield: 44% (21.6 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 8:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.66 (d, $J = 4.5$ Hz, 1H), 7.55 (s, 1H), 7.41-7.34 (m, 2H), 7.31-7.23 (m, 4H), 7.17-7.09 (m, 3H), 6.88 (d, $J = 7.8$ Hz, 1H), 2.47 (s, 3H); $^{13}C\{^1H\}$ NMR (75 MHz, $CDCl_3$): δ 159.4, 149.4, 141.3, 139.2, 137.8, 137.4, 135.2, 131.1, 130.5, 129.7, 129.3, 128.0, 126.5, 125.5, 121.3, 21.1; HRMS (ESI): Exact mass calcd for $C_{18}H_{15}N$ $[M+H]^+$, 246.1277; Found: 246.1280.

2-(5-methyl-[1,1'-biphenyl]-2-yl)pyridine (9c)¹⁶

Yield: 50% (24.5 mg), Pale yellow oil, $R_f = 0.40$ (petroleum ether/ethyl acetate = 8:1); 1H NMR (300 MHz, $CDCl_3$): δ 8.65-8.63 (m, 1H), 7.64 (d, $J = 7.8$ Hz, 1H), 7.41-7.35 (m, 1H), 7.32-7.24 (m, 5H), 7.20-7.16 (m, 2H), 7.12-7.07 (m, 1H), 6.88 (t, $J = 8.4$ Hz, 1H), 2.47 (s, 3H); $^{13}C\{^1H\}$ NMR (75 MHz,

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CDCl₃): δ 159.2, 149.4, 141.5, 140.5, 138.4, 136.7, 135.1, 131.3, 130.5, 129.7, 129.6, 128.4, 128.0, 127.6, 126.6, 125.4, 121.1, 21.2; HRMS (ESI): Exact mass calcd for C₁₈H₁₅N [M+H]⁺, 246.1277; Found: 246.1286.

2-(5-methoxy-[1,1'-biphenyl]-2-yl)pyridine (9d)¹⁶

Yield: 39% (20.4 mg), Pale yellow oil, R_f = 0.40 (petroleum ether/ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 8.63-8.61 (m, 1H), 7.68 (d, *J* = 8.4 Hz, 1H), 7.39-7.33 (m, 1H), 7.28-7.25 (m, 3H), 7.21-7.16 (m, 2H), 7.10-7.02 (m, 2H), 6.98 (d, *J* = 2.7 Hz, 1H), 6.83 (d, *J* = 7.8 Hz, 1H), 3.90 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 159.7, 158.9, 149.3, 142.0, 141.4, 135.1, 132.3, 131.9, 129.6, 129.5, 128.1, 127.6, 126.9, 125.4, 120.9, 115.7, 113.3, 55.4; HRMS (ESI): Exact mass calcd for C₁₈H₁₅NO [M+H]⁺, 262.1236; Found: 262.1238.

2-([1,1'-biphenyl]-2-yl)-3-methylpyridine (9e)¹⁶

Yield: 36% (17.7 mg), Yellow oil, R_f = 0.40 (petroleum ether/ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 8.51 (d, *J* = 4.5 Hz, 1H), 7.50-7.40 (m, 4H), 7.33-7.30 (m, 1H), 7.19-7.10 (m, 6H), 1.77 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 159.5, 146.5, 141.1, 140.7, 139.4, 137.5, 131.7, 129.9, 129.7, 129.3, 128.4, 127.8, 127.4, 126.6, 122.1, 18.8; HRMS (ESI): Exact mass calcd for C₁₈H₁₅N [M+H]⁺, 246.1277; Found: 246.1282.

2-(4'-methyl-[1,1'-biphenyl]-2-yl)pyridine (9f)¹⁶

Yield: 40% (19.6 mg), Pale yellow oil, R_f = 0.40 (petroleum ether/ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 8.66 (d, *J* = 5.1 Hz, 1H), 7.72-7.69 (m, 1H), 7.49-7.38 (m, 4H), 7.15-7.07 (m, 5H), 6.92 (d, *J* = 8.4 Hz, 1H), 2.34 (s, 3H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 149.4, 140.6, 139.4, 138.4, 136.4, 135.2, 130.5, 130.5, 129.6, 129.5, 128.8, 128.5, 128.4, 127.4, 125.4, 121.3, 21.1; HRMS (ESI): Exact mass calcd for C₁₈H₁₅N [M+H]⁺, 246.1277; Found: 246.1286.

2-(4'-(trifluoromethyl)-[1,1'-biphenyl]-2-yl)pyridine (9g)¹⁶

Yield: 39% (23.3 mg), Pale yellow oil, R_f = 0.40 (petroleum ether/ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 8.64-8.62 (m, 1H), 7.73-7.70 (m, 1H), 7.54-7.43 (m, 6H), 7.28 (t, *J* = 5.7 Hz, 2H), 7.18-7.13 (m, 1H), 6.95 (d, *J* = 7.8 Hz, 1H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 158.8, 149.5, 145.1, 139.6, 139.2, 135.6, 130.6, 130.4, 129.9, 128.7, 128.3, 125.2, 125.0, 125.0, 121.7; HRMS (ESI): Exact mass calcd for C₁₈H₁₂F₃N [M+H]⁺, 300.0995; Found: 300.0999.

10-phenylbenzo[h]quinoline (9h)¹⁶

Yield: 40% (20.4 mg), Yellow oil, R_f = 0.40 (petroleum ether/ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 8.46-8.43 (m, 1H), 8.11 (d, *J* = 8.1 Hz, 1H), 7.95 (d, *J* = 8.1 Hz, 1H), 7.88 (d, *J* = 8.7 Hz, 1H), 7.73-7.68 (m, 2H), 7.57 (d, *J* = 7.2 Hz, 1H), 7.43-7.32 (m, 6H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 146.9, 146.8, 146.4, 141.7, 135.2, 135.0, 131.5, 129.0, 128.3, 127.9, 127.4, 127.2, 127.0, 125.9, 125.7, 121.1, 29.7; HRMS (ESI): Exact mass calcd for C₁₉H₁₃N [M+H]⁺, 256.1121; Found: 256.1124.

General procedure: Synthetic applications

Path A: A oven-dried 15 mL Schlenk tube was charged with **7a** (0.2 mmol), SiO₂ (0.6 mmol) in a mixture of PE (1 mL) and EA (0.05 mL), which was stirred at room temperature for 5 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 2:1) to afford the corresponding product **8a**.

Yield: 87% (41.4 mg), Mp 113-115 °C, Pale yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 2:1); ¹H NMR (300 MHz, CDCl₃): δ 8.91 (d, *J* = 4.2 Hz, 1H), 8.60 (d, *J* = 8.4 Hz, 1H), 7.80 (d, *J* = 7.8 Hz, 1H), 7.59-7.55 (m, 1H), 7.49 (d, *J* = 7.8 Hz, 1H), 5.17 (s, 2H), 4.76 (br s, 1H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 149.7, 147.7, 138.3, 136.3, 130.1, 128.1, 127.8, 122.3, 121.3, 64.2; IR (KBr): ν 3361, 3055, 2920, 2850, 1559, 1494, 1467, 1429, 1287, 1247, 1147, 1113, 876, 831, 778, 729; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₉BrNO 237.9862, Found 237.9872.

Path B: A oven-dried 15 mL Schlenk tube was charged with a mixture of **7e** (0.2 mmol), *t*-BuCl (0.24 mmol), [pmIm]Br (0.4 mmol) and was stirred in a 60 °C oil-bath for 4 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 8:1) to afford the targeted product **10**.

Yield: 78% (34.7 mg), Mp 99-100 °C, Pale yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 9.11 (t, *J* = 3.9 Hz, 1H), 9.03 (d, *J* = 8.7 Hz, 1H), 8.35 (d, *J* = 7.8 Hz, 1H), 7.95 (d, *J* = 8.1 Hz, 1H), 7.72-7.68 (m, 1H), 5.25 (s, 2H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 151.1, 145.5, 143.7, 132.3, 128.5, 127.6, 124.3, 124.2, 121.3, 27.9; IR (KBr): ν 3047, 2922, 2853, 1594, 1516, 1496, 1404, 1237, 1212, 1155, 1122, 873, 831, 791, 739; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₈ClN₂O₂ 223.0269, Found 223.0269.

Path C: A oven-dried 15 mL Schlenk tube was charged with a mixture of **7e** (0.2 mmol), *t*-BuBr (0.24 mmol), [pmIm]Br (0.4 mmol) and was sonicated in an ultrasonic bath for 0.5 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 8:1) to afford the corresponding product **11**.

Yield: 95% (50.7 mg), Mp 106-107 °C, Pale yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate =

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8:1); ¹H NMR (300 MHz, CDCl₃): δ 9.12 (d, *J* = 2.1 Hz, 1H), 9.03 (d, *J* = 8.7 Hz, 1H), 8.36 (d, *J* = 7.8 Hz, 1H), 7.96 (d, *J* = 7.8 Hz, 1H), 7.73-7.69 (m, 1H), 5.26 (s, 2H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 151.1, 145.5, 143.7, 132.3, 128.5, 124.3, 124.2, 121.3, 27.9; IR (KBr): ν 2965, 2920, 2853, 1591, 1511, 1496, 1404, 1250, 1217, 1160, 876, 831, 791, 714; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₈BrN₂O₂ 266.9764, Found 266.9756.

Path D: A oven-dried 15 mL Schlenk tube was charged with a mixture of **7e** (0.2 mmol), *t*-BuI (0.24 mmol), [pmIm]Br (0.4 mmol) and was sonicated in an ultrasonic bath for 0.5 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 8:1) to afford the corresponding product **12**.

Yield: 71% (44.6 mg), Mp 126-127 °C, Pale yellow solid, *R_f* = 0.40 (petroleum ether / ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 9.13 (d, *J* = 2.4 Hz, 1H), 9.06 (d, *J* = 8.7 Hz, 1H), 8.31 (d, *J* = 7.8 Hz, 1H), 7.92 (d, *J* = 8.1 Hz, 1H), 7.72-7.68 (m, 1H), 5.21 (s, 2H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 150.6, 145.9, 145.2, 132.2, 127.6, 124.6, 124.4, 121.6, 0.87; IR (KBr): ν 3037, 2922, 2853, 1561, 1511, 1494, 1402, 1235, 1165, 876, 851, 833, 793; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₈IN₂O₂ 314.9625, Found 314.9627.

Path E: A oven-dried 15 mL Schlenk tube was charged with a mixture of **7e** (0.2 mmol), AgNO₃ (0.4 mmol) in *t*-AmOH (1 mL) and was stirred in a 120 °C oil-bath for 6 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate = 8:1) to afford the corresponding product **13**.

Yield: 51% (23.8 mg), Yellow oil, *R_f* = 0.40 (petroleum ether / ethyl acetate = 8:1); ¹H NMR (300 MHz, CDCl₃): δ 9.05 (d, *J* = 9.0 Hz, 2H), 8.40 (d, *J* = 8.1 Hz, 1H), 7.88 (d, *J* = 7.8 Hz, 1H), 7.75-7.71 (m, 1H), 6.28 (s, 2H); ¹³C{¹H} NMR (75 MHz, CDCl₃): δ 151.2, 145.6, 138.7, 132.3, 126.1, 124.4, 124.1, 121.1, 70.2; IR (KBr): ν 3104, 2913, 2850, 1594, 1516, 1499, 1402, 1277, 1232, 1190, 1157, 858, 833, 806, 778, 736; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₈N₃O₄ 234.0509, Found 234.0511.

Path F: To a solution of sodium azide (0.30 mmol) in water (1 mL) in a two necked round bottom flask equipped with a stirring bar and reflux condenser was added a solution of **7e** (0.20 mmol) in THF (0.2 mL) and the mixture was heated to reflux in an oil-bath for 3 h. After the reaction was complete (as determined by TLC analysis), the reaction was cooled to room temperature and 10 mL of water was

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4 added. Then, the mixture was extracted with ethyl acetate (10 mL × 3). The combined organic layers
5 were washed with saturated brine and dried by anhydrous Na₂SO₄. The solvent was evaporated under
6 reduced pressure, and the residue was purified on a silica gel column (petroleum ether/ethyl acetate =
7 8:1) to afford the corresponding product **14**.

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10 Yield: 94% (43.1 mg), Mp 93-94 °C, Yellow solid, R_f = 0.40 (petroleum ether / ethyl acetate = 8:1); ¹H
11 NMR (300 MHz, CDCl₃): δ 9.05 (t, *J* = 5.1 Hz, 2H), 8.40 (d, *J* = 8.1 Hz, 1H), 7.88 (d, *J* = 7.8 Hz, 1H),
12 7.72-7.68 (m, 1H), 5.19 (s, 2H); ¹³C {¹H} NMR (75 MHz, CDCl₃): δ 150.9, 145.8, 145.2, 142.1, 132.3,
13 126.3, 124.4, 124.2, 121.1, 51.1; IR (KBr): ν 3047, 2922, 2853, 1594, 1519, 1501, 1432, 1282, 1195,
14 1162, 1130, 878, 846, 803, 776; HRMS (ESI-TOF) *m/z*: [M+H]⁺ Calcd for C₁₀H₈N₅O₂ 230.0673,
15 Found 230.0676.
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20 ASSOCIATED CONTENT

21 Supporting Information

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23 The Supporting Information is available free of charge on the ACS Publications website at DOI:
24 10.1021/acs.joc.xxxxxx.
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28 ¹H and ¹³C spectra for **3a-3m**, **4a-4l**, **6a-6f**, **7a-7r**, **8a-8g**, **9a-9h**, **10-14** and ³¹P spectra for **3a-3m**, **4a-4l**
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37 Notes

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52 REFERENCES

- 53
54 (1) Corbridge, D. E. C. Phosphorus 2000. *Chemistry, Biochemistry & Technology*; Elsevier:
55 Oxford, **2000**; Chapters 10 and 11.
56
57 (2) (a) Hecker, S. J.; Erion, M. D. Prodrugs of Phosphates and Phosphonates. *J. Med. Chem.* **2008**,
58 *51*, 2328; (b) Pisani, L.; Barletta, M.; Soto-Otero, R.; Nicolotti, O.; Mendez-Alvarez, E.; Catto,
59
60

- M.; Introcaso, A.; Stefanachi, A.; Cellamare, S.; Altomare, C.; Carotti, A. Discovery, Biological Evaluation, and Structure-Activity and -Selectivity Relationships of 6'-Substituted (E)-2-(Benzofuran-3(2H)-ylidene)-N-methylacetamides, a Novel Class of Potent and Selective Monoamine Oxidase Inhibitors. *J. Med. Chem.* **2013**, *56*, 2651; (c) Pradere, U.; Garnier-Amblard, E. C.; Coats, S. J.; Amblard, F.; Schinazi, R. F. Synthesis of Nucleoside Phosphate and Phosphonate Prodrugs. *Chem. Rev.* **2014**, *114*, 9154; (d) Park, J.-H.; Lee, G.-E.; Lee, S.-D.; Hien, T. T.; Kim, S.; Yang, J. W.; Cho, J.-H.; Ko, H.; Lim, S.-C.; Kim, Y.-G.; Kang, K.-W.; Kim, Y.-C. Discovery of Novel 2,5-Dioxoimidazolidine-Based P2X₇ Receptor Antagonists as Constrained Analogues of KN62. *J. Med. Chem.* **2015**, *58*, 2114.
- (3) (a) Morales-Rojas, H.; Moss, R. A. Phosphorolytic Reactivity of *o*-Iodosylcarboxylates and Related Nucleophiles. *Chem. Rev.* **2002**, *102*, 2497; (b) Segall, Y.; Quistad, G. B.; Sparks, S. E.; Casida, J. E. Major Intermediates in Organophosphate Synthesis (PCl₃, POCl₃, PSCl₃, and Their Diethyl Esters) Are Anticholinesterase Agents Directly or on Activation. *Chem. Res. Toxicol.* **2003**, *16*, 350.
- (4) (a) Pauff, S. M.; Miller, S. C. Synthesis of Near-IR Fluorescent Oxazine Dyes with Esterase-Labile Sulfonate Esters. *Org. Lett.* **2011**, *13*, 6196; (b) Gao, X.; Tang, Z.; Lu, M.; Liu, H.; Jiang, Y.; Zhao, Y.; Cai, Z. Suppression of Matrix Ions by *N*-Phosphorylation Labeling Using Matrix-Assisted Laser Desorption-Ionization Time-of-Flight Mass Spectrometry. *Chem. Commun.* **2012**, *48*, 10198; (c) Pauff, S. M.; Miller, S. C. A Trifluoroacetic Acid-labile Sulfonate Protecting Group and Its Use in the Synthesis of a Near-IR Fluorophore. *J. Org. Chem.* **2013**, *78*, 711.
- (5) (a) Avitabile, B. G.; Smith, C. A.; Judd, D. B. Pentafluorophenyl Sulfonate Ester as a Protecting Group for the Preparation of Biaryl- and Heterobiaryl Sulfonate Esters. *Org. Lett.* **2005**, *7*, 843; (b) Adams, C. J.; Bedford, R. B.; Carter, E.; Gower, N. J.; Haddow, M. F.; Harvey, J. N.; Huwe, M.; Cartes, M. Á.; Mansell, S. M.; Mendoza, C.; Murphy, D. M.; Neeve, E. C.; Nunn, J. Iron (I) in Negishi Cross-Coupling Reactions. *J. Am. Chem. Soc.* **2012**, *134*, 10333; (c) Li, F.; Liu, T.-X.; Wang, G.-W. Synthesis of [60] Fullerene-Fused Sultones via Sulfonic Acid Group-Directed C-H Bond Activation. *Org. Lett.* **2012**, *14*, 2176; (d) DeBergh, J. R.; Niljianskul, N.; Buchwald, S. L. Synthesis of Aryl Sulfonamides via Palladium-Catalyzed Chlorosulfonylation of Arylboronic Acids. *J. Am. Chem. Soc.* **2013**, *135*, 10638.
- (6) For the preparation of the phosphates, see: (a) Jones, S.; Selitsianos, D. A Simple and Effective Method for Phosphoryl Transfer Using TiCl₄ Catalysis. *Org. Lett.* **2002**, *4*, 3671; (b) Sakakura, A.; Katsukawa, M.; Ishihara, K. The Oxorhenium (VII)-Catalyzed Direct Condensation of Phosphoric Acid with an Alcohol. *Angew. Chem., Int. Ed.* **2007**, *46*, 1423; (c) Xiong, B.; Feng,

- X.; Zhu, L.; Chen, T.; Zhou, Y.; Au, C. T.; Yin, S-F. Direct Aerobic Oxidative Esterification and Arylation of P(O)-OH Compounds with Alcohols and Diaryliodonium Triflates. *ACS Catal.* **2015**, *5*, 537; (d) Huang, H.; Denne, J.; Yang, C.-H.; Wang, H.; Kang, J. Y. Direct Aryloxylation/Alkyloxylation of Dialkyl Phosphonates for the Synthesis of Mixed Phosphonates. *Angew. Chem. Int. Ed.* **2018**, *57*, 6624.
- (7) For the preparation of the sulfonates, see: (a) Choudary, B. M.; Chowdari, N. S.; Kantam, M. L. Montmorillonite Clay Catalyzed Tosylation of Alcohols and Selective Monotosylation of Diols with *p*-Toluenesulfonic Acid: An Enviro-Economic Route. *Tetrahedron.* **2000**, *56*, 7291; (b) Das, B.; Reddy, V. S. ZrCl₄ as an Efficient Catalyst for Selective Tosylation of Alcohols with *p*-Toluenesulfonic Acid. *Chem. Lett.* **2004**, *33*, 1428; (c) Velusamy, S.; Kumar, J. S. K.; Punniyamurthy, T. Cobalt(II) Catalyzed Tosylation of Alcohols with *p*-Toluenesulfonic Acid. *Tetrahedron Lett.* **2004**, *45*, 203; (d) Jalalian, N.; Petersen, T. B.; Olofsson, B. Metal-Free Arylation of Oxygen Nucleophiles with Diaryliodonium Salts. *Chem.-Eur. J.* **2012**, *18*, 14140.
- (8) (a) Jalalian, N.; Petersen, T. B.; Olofsson, B. Metal-Free Arylation of Oxygen Nucleophiles with Diaryliodonium Salts. *Chem. -Eur. J.* **2012**, *18*, 14140; (b) Shen, C.; Yang, M.; Xu, J.; Chen, C.; Zheng, K.; Shen, J.; Zhang, P. Iodobenzene-Catalyzed Synthesis of Aryl Sulfonate Esters from Aminoquinolines via Remote Radical C-O Cross-Coupling. *RSC Adv.* **2017**, *7*, 49436; (c) Wang, C.-S.; Dixneuf, P. H.; Soulé, J.-F. Metal-Free C(sp³)-H Bond Sulfonyloxylation of 2-Alkylpyridines and Alkylnitrones. *Org. Biomol. Chem.* **2018**, *16*, 4954.
- (9) (a) Nabana, T.; Togo, H. Reactivities of Novel [Hydroxy (tosyloxy) iodo] arenes and [Hydroxy (phosphoryloxy) iodo] arenes for α -Tosyloxylation and α -Phosphoryloxylation of Ketones. *J. Org. Chem.* **2002**, *67*, 4362; (b) Yusubov, M. S.; Wirth, T. Solvent-Free Reactions with Hypervalent Iodine Reagents. *Org. Lett.* **2005**, *7*, 519; (c) Merrih, E. A.; Carneiro, V. M. T.; Silva Jr, L. F.; Olofsson, B. Facile Synthesis of Koser's Reagent and Derivatives from Iodine or Aryl Iodides. *J. Org. Chem.* **2010**, *75*, 7416.
- (10) (a) Xu, Y.; Yan, G.; Ren, Z.; Dong, G. Diverse sp³C-H Functionalization Through Alcohol β -Sulfonyloxylation. *Nat. Chem.* **2015**, *7*, 829; (b) Yang, Z. W.; Zhang, Q.; Jiang, Y. Y.; Li, L.; Xiao, B.; Fu, Y. Palladium-catalyzed directing group-assisted C8-triflation of naphthalenes. *Chem. Commun.* **2016**, *52*, 6709; (c) Yang, Q.-L.; Li, Y.-Q.; Ma, C.; Fang, P.; Zhang, X.-J.; Mei, T.-S. Palladium-Catalyzed C(sp³)-H Oxygenation via Electrochemical Oxidation. *J. Am. Chem. Soc.* **2017**, *139*, 3293; (d) Zhao, R.; Lu, W. Palladium-Catalyzed β -Mesylation of Simple Amide via Primary sp³C-H Activation. *Org. Lett.* **2017**, *19*, 1768; (e) Chen, X.-Y.; Ozturk, S.; Sorensen, E. J. Pd-Catalyzed *Ortho* C-H Hydroxylation of Benzaldehydes Using a Transient Directing Group. *Org. Lett.* **2017**, *19*, 6280; (e) Nappi, M.; He, C.; Whitehurst, W. G.; Chappell, B. G. N.; Gaunt, M. J. Selective Reductive Elimination at Alkyl Palladium(IV)

- 1
2
3
4 by Dissociative Ligand Ionization: Catalytic C(sp³)-H Amination to Azetidines. *Angew. Chem. Int. Ed.* **2018**, *57*, 3178; (f) Wang, D.; Liu, Z.; Wang, Z.; Ma, X.; Yu, P. Metal- and Base-Free
5
6 Regioselective Thiolation of the Methyl C(sp³)-H Bond in 2-Picoline N-Oxides. *Green Chem.*
7
8 **2019**, *21*, 157.
- (11) (a) Lu, Q.; Zhang, J.; Wei, F.; Qi, Y.; Wang, H.; Liu, Z.; Lei, A. Aerobic Oxysulfonylation of
9
10 Alkenes Leading to Secondary and Tertiary β -Hydroxysulfones. *Angew. Chem., Int. Ed.* **2013**,
11
12 *52*, 7156; (b) Yang, X.; Shan, G.; Rao, Y. Synthesis of 2-Aminophenols and Heterocycles by
13
14 Ru-Catalyzed C-H Mono- and Dihydroxylation. *Org. Lett.* **2013**, *15*, 2334; (c) Liu, W.;
15
16 Ackermann, L. *Ortho*- and *Para*-Selective Ruthenium-Catalyzed C(sp²)-H Oxygenations of
17
18 Phenol Derivatives. *Org. Lett.* **2013**, *15*, 3484; (d) Li, X.; Liu, Y.-H.; Gu, W.-J.; Li, B.; Chen,
19
20 F.-J.; Shi, B.-F. Copper-Mediated Hydroxylation of Arenes and Heteroarenes Directed by a
21
22 Removable Bidentate Auxiliary. *Org. Lett.* **2014**, *16*, 3904.
- (12) (a) Ye, Z.; Wang, W.; Luo, F.; Zhang, S.; Cheng, J. Rhodium-Catalyzed *ortho*-Benzylation
23
24 of sp²C-H Bond. *Org. Lett.* **2009**, *11*, 3974; (b) Rit, R. K.; Yadav, M. R.; Sahoo, A. K.
25
26 Pd(II)-Catalyzed *ortho*-C-H Oxidation of Arylacetic Acid Derivatives: Synthesis of
27
28 Benzofuranones. *Org. Lett.* **2014**, *16*, 968; (c) Wang, Z.; Kuang, C. Palladium-Catalyzed
29
30 Acyloxylation of 2-Substituted 1,2,3-Triazoles via Direct sp²C-H Bond Activation. *Adv. Synth.*
31
32 *Catal.* **2014**, *356*, 1549; (d) Raghuvanshi, K.; Rauch, K.; Ackermann, L. Ruthenium
33
34 (II)-Catalyzed C-H Acyloxylation of Phenols with Removable Auxiliary. *Chem.-Eur. J.* **2015**,
35
36 *21*, 1790.
- (13) (a) Roane, J.; Daugulis, O. Copper-Catalyzed Etherification of Arene C-H Bonds. *Org. Lett.*
37
38 **2013**, *15*, 5842; (b) Yin, Z.; Jiang, X.; Sun, P. Palladium-Catalyzed Direct *ortho* Alkoxylation
39
40 of Aromatic Azo Compounds with Alcohols. *J. Org. Chem.* **2013**, *78*, 10002; (c) Jiang, Q.;
41
42 Wang, J.-Y.; Guo, C. Iodine (III)-Mediated C-H Alkoxylation of Aniline Derivatives with
43
44 Alcohols under Metal-Free Conditions. *J. Org. Chem.* **2014**, *79*, 8768; (d) Zhang, L.-B.; Hao,
45
46 X.-Q. Zhang, S.-K.; Liu, Z.-J.; Zheng, X.-X. Gong, J.-F.; Niu, J.-L.; Song, M.-P.
47
48 Cobalt-Catalyzed C(sp²)-H Alkoxylation of Aromatic and Olefinic Carboxamides. *Angew.*
49
50 *Chem., Int. Ed.* **2015**, *54*, 272.
- (14) For recent reports on the C(sp³)-H functionalization of 8-methylquinolines, see: (a) Dick, A.
51
52 R.; Hull, K. L.; Sanford, M. S. A Highly Selective Catalytic Method for the Oxidative
53
54 Functionalization of C-H Bonds. *J. Am. Chem. Soc.* **2004**, *126*, 2300; (b) Iglesias, Á.;
55
56 Álvarez, R.; de Lera, Á. R.; Muñiz, K. Palladium-Catalyzed Intermolecular C(sp³)-H
57
58 Amidation. *Angew. Chem., Int. Ed.* **2012**, *51*, 2225; (c) McMurtrey, K. B.; Racowski, J. M.;
59
60 Sanford, M. S. Pd-Catalyzed C-H Fluorination with Nucleophilic Fluoride. *Org. Lett.* **2012**,
14, 4094; (d) Wang, D.; Zavalij, P. Y.; Vedernikov, A. N. Aerobic C-H Acetoxylation of

- 8-Methylquinoline in Pd^{II}-Pyridinecarboxylic Acid Systems: Some Structure-Reactivity Relationships. *Organometallics*. **2013**, *32*, 4882; (e) Liu, B.; Zhou, T.; Li, B.; Wang, B. Rhodium(III)-Catalyzed Alkenylation Reactions of 8-Methylquinolines with Alkynes by C(sp³)-H Activation. *Angew. Chem., Int. Ed.* **2014**, *53*, 4191; (f) Kong, L.; Zhou, X.; Xu, Y.; Li, X. Rhodium (III)-Catalyzed Acylation of C (sp³)-H Bonds with Cyclopropanones. *Org. Lett.* **2017**, *19*, 3644; (g) You, C.; Pi, C.; Wu, Y.; Cui, X. Rh(III)-Catalyzed Selective C8-H Acylmethylation of Quinoline N-Oxides. *Adv. Synth. Catal.* **2018**, *360*, 4068; (h) Xie, L.-Y.; Peng, S.; Tan, J.-X.; Sun, R.-X.; Yu, X.; Dai, N.-N.; Tang, Z.-L.; Xu, X.; He, W.-M. Waste-Minimized Protocol for the Synthesis of Sulfonylated N-Heteroaromatics in Water. *ACS Sustainable Chem. Eng.* **2018**, *6*, 16976; (i) Muhammad, M. H.; Chen, X.-L.; Yu, B.; Qu, L.-B.; Zhao, Y.-F. Applications of H-phosphonates for C element bond formation. *Pure Appl. Chem.* **2019**, *91*, 33.
- (15) (a) Chen, X.; Hao, X.-S.; Goodhue, C. E.; Yu, J.-Q. Cu(II)-Catalyzed Functionalizations of Aryl C-H Bonds Using O₂ as an Oxidant. *J. Am. Chem. Soc.* **2006**, *128*, 6790; (b) Yan, Y.; Feng, P.; Zheng, Q.-Z.; Liang, Y.-F.; Lu, J.-F.; Cui, Y.; Jiao, N. PdCl₂ and N-Hydroxyphthalimide Co-Catalyzed C-H Hydroxylation by Dioxygen Activation. *Angew. Chem., Int. Ed.* **2013**, *52*, 5827; (c) Yamaguchi, T. E.; Tada, N.; Itoh, A. Direct *ortho*-Hydroxylation of 2-Phenylpyridines Using Palladium(II) Chloride and Hydrogen Peroxide. *Adv. Synth. Catal.* **2015**, *357*, 2017; (d) Das, P.; Saha, D.; Guin, J. Aerobic Direct C(sp²)-H Hydroxylation of 2-Arylpyridines by Palladium Catalysis Induced with Aldehyde Auto-Oxidation. *ACS Catal.* **2016**, *6*, 6050; (e) Wu, Y.; Zhou, B. Rhodium (III)-Catalyzed Selective C-H Acetoxylation and Hydroxylation Reactions. *Org. Lett.* **2017**, *19*, 3532; (f) Thongpaen, J.; Schmid, T. E.; Toupet, L.; Dorcet, V.; Mauduit, M.; Baslé, O. Directed *ortho* C-H Borylation Catalyzed Using Cp*Rh (iii)-NHC Complexes. *Chem. Commun.* **2018**, *54*, 8202.
- (16) (a) Kalyani, D.; Deprez, N. R.; Desai, L. V.; Sanford, M. S. Oxidative C-H Activation/C-C Bond Forming Reactions: Synthetic Scope and Mechanistic Insights. *J. Am. Chem. Soc.* **2005**, *127*, 7330; (b) Hull, K. L.; Sanford, M. S. Catalytic and Highly Regioselective Cross-Coupling of Aromatic C-H Substrates. *J. Am. Chem. Soc.* **2007**, *129*, 11904; (c) Wang, X. S.; Leow, D.; Yu, J.-Q. Pd(II)-Catalyzed *para*-Selective C-H Arylation of Monosubstituted Arenes. *J. Am. Chem. Soc.* **2011**, *133*, 13864; (d) Luan, Y.-X.; Zhang, T.; Yao, W.-W.; Lu, K.; Kong, L.-Y.; Lin, Y.-T. Ye, M. Amide-Ligand-Controlled Highly *para*-Selective Arylation of Monosubstituted Simple Arenes with Arylboronic Acids. *J. Am. Chem. Soc.* **2017**, *139*, 1786.
- (17) (a) Ranu, B. C.; Jana, R. Direct Halogenation of Alcohols and Their Derivatives with *tert*-Butyl Halides in the Ionic Liquid [pmIm]Br under Sonication Conditions - A Novel,

- 1
2
3
4 Efficient and Green Methodology. *Eur. J. Org. Chem.* **2005**, 755; (b) Swetha, M.; Ramana, P.
5 V.; Shirodkar, S. G. Simple and Efficient Method for the Synthesis of Azides in Water-THF
6 Solvent System. *Org. Prep. Proced. Int.* **2011**, 43, 348; (c) Li, C.; Deng, H.; Li, C.; Jia, X.; Li,
7 J. Palladium-Catalyzed Synthesis of Δ^2 -Isoxazoline from Toluene Derivatives Enabled by the
8 Triple Role of Silver Nitrate. *Org. Lett.* **2015**, 17, 5718.
9
10
11 (18) (a) Zhang, J.; Khaskin, E.; Anderson, N. P.; Zavalij, P. Y.; Vedernikov, A. N. Catalytic Aerobic
12 Oxidation of Substituted 8-Methylquinolines in Pd^{II}-2,6-Pyridinedicarboxylic Acid Systems.
13 *Chem. Commun.* **2008**, 3625; (b) Wang, N.; Li, R.; Li, L.; Xu, S.; Song, H.; Wang, B. Rhodium
14 (III)-Catalyzed Intermolecular Amidation with Azides via C (sp³)-H Functionalization. *J. Org.*
15 *Chem.* **2014**, 79, 5379; (c) Liu, B.; Hu, P.; Zhou, X.; Bai, D.; Chang, J.; Li, X.
16 Cp*Rh(III)-Catalyzed Mild Addition of C(sp³)-H Bonds to α,β -Unsaturated Aldehydes and
17 Ketones. *Org. Lett.* **2017**, 19, 2086; (d) O'Murchu, C. Ozonolysis of Quinolines: A Versatile
18 Synthesis of Polyfunctional Pyridines. *Synthesis.* **1989**, 11, 880; (e) Cai, W.; Wang, S.; Jalani, H.
19 B.; Wu, J.; Lu, H.; Li, G. Oxidative Cascade Reaction of *N*-(Hetero)Aryl-3-alkylideneazetidines
20 and Carboxylic Acids: Access to Fused Pyridines. *Org. Lett.* **2018**, 20, 3833; (f) Sharghi, H.;
21 Aberi, M.; Khataminejad, M.; Shiri, P. Solvent-Free and Room Temperature Synthesis of
22 3-Arylquinolines from Different Anilines and Styrene Oxide in the Presence of Al₂O₃/MeSO₃H.
23 *Beilstein J. Org. Chem.* **2017**, 13, 1977.
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
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