

## Organocatalysis

## Disulfonimide-Catalyzed Asymmetric Reduction of N-Alkyl Imines

Vijay N. Wakchaure, Philip S. J. Kaib, Markus Leutzsch, and Benjamin List\*

**Abstract:** A chiral disulfonimide (DSI)-catalyzed asymmetric reduction of N-alkyl imines with Hantzsch esters as a hydrogen source in the presence of Boc<sub>2</sub>O has been developed. The reaction delivers Boc-protected N-alkyl amines with excellent yields and enantioselectivity. The method tolerates a large variety of alkyl amines, thus illustrating potential for a general reductive cross-coupling of ketones with diverse amines, and it was applied in the synthesis of the pharmaceuticals (S)-Rivastigmine, NPS R-568 Hydrochloride, and (R)-Fendiline.

**L** nantiomerically pure amines, in particular  $\alpha$ -chiral *N*-alkyl amines, represent a privileged pharmacophore that can be found in a vast number of pharmaceutical and agrochemical substances, including (*S*)-Rivastigmine (Alzheimer's and Parkinson's diseases), NPS *R*-568 (hyperparathyroidism), and (*R*)-Fendiline (angina pectoris; Figure 1).<sup>[1]</sup>



**Figure 1.**  $\alpha$ -Chiral *N*-alkyl amine pharmaceuticals.

Catalytic asymmetric imine reductions and reductive aminations of carbonyl compounds represent efficient approaches for the construction of optically pure amines.<sup>[2]</sup> Recently, we developed a Brønsted acid catalyzed asymmetric imine reduction and reductive amination of ketones using Hantzsch esters as hydrogen source,<sup>[3]</sup> and independent variations and applications have appeared.<sup>[4–8]</sup> Despite these advances, however, such reductions have been limited to *N*aryl imines. We now report a highly enantioselective chiral disulfonimide (DSI)-catalyzed Hantzsch ester mediated reduction of *N*-alkyl imines.

- [\*] Dr. V. N. Wakchaure, P. S. J. Kaib, M. Leutzsch, Prof. Dr. B. List Max-Planck-Institut für Kohlenforschung
- Kaiser -Wilhelm-Platz 1, 45470 Mülheim an der Ruhr (Germany) E-mail: list@kofo.mpg.de
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N-Alkyl imines are highly attractive substrates for asymmetric reductions since they would directly furnish the Nalkyl amine pharmacophore. Chiral phosphoric acids typically fail in the corresponding Hantzsch ester mediated imine reductions and alternative approaches have only rarely been investigated and remain a major challenge.<sup>[3b,6,9,10]</sup> At the onset of our work, we hypothesized that owing to the high basicity of the desired N-alkyl amine products, a stronger Brønsted acid catalyst may be required for Hantzsch ester reductions to achieve high turnover and enantioselectivity. Indeed, we found that both phosphoric acids (2a-c) and Ntriflyl phosphoramide 2d promoted the reduction of Nmethyl imine 1a with Hantzsch ester 3b to give amine 4 with only disappointing conversion and enantioselectivity (Table 1, entries 1-4). Recently, we introduced chiral disulfonimides (DSIs) as effective precatalysts in silicon-based

*Table 1:* Reaction development.<sup>[a]</sup>



Entry	Cat.	Hantzsch ester	additive	<i>t</i> [h]	Conv.[%] <sup>[b]</sup>	e.r. <sup>[c]</sup>
1	2a	3 b	_	72	18	51.5:48.5
2	2 b	3 b	-	72	6	63:37
3	2 c	3 b	-	72	25	51.5:48.5
4	2 d	3 b	-	72	50	51:49
5	2e	3 b	-	48	84	77:23
6	2 f	3 b	-	72	86	70:30
7	2 g	3 b	-	72	80	76:24
8	2h	3 b	-	48	83	85:15
9	2i	3 b	-	48	83	88:12
10	2i	3 a	-	48	40	89.5:10.5
11	2i	3 c	-	48	81	88.5:11.5
12	2i	3 d	-	24	99	88:12
13 <sup>[d]</sup>	2 i	3 d	-	48	88	93:7
14 <sup>[d,e]</sup>	2i	3 d	-	48	60	95.5:4.5
15 <sup>[d-f]</sup>	2i	3 d	Boc <sub>2</sub> O	48	99	95.5:4.5
16 <sup>[d-f]</sup>	2 b	3 d	$Boc_2O$	48	80	62:38

[a] Reactions were run on a 0.1 mmol scale. [b] Determined by GC–MS.[c] Determined by GC or HPLC with a chiral stationary phase.[d] Reactions were run at 10°C. [e] Mesitylene was used as a solvent.

[f] Obtained product **5** a.



Lewis acid catalysis<sup>[11]</sup> and also as strong Brønsted acid catalysts.<sup>[12]</sup> Indeed, our chiral DSIs (2e-i) catalyzed the transformation more rapidly than the alternative phosphoric acid based catalysts and also afforded amine 4 with higher enantioselectivity (entries 5-9). After careful investigation, we identified reaction conditions in which the reduction of imine 1a is performed using 5 mol% of catalyst (R)-DSI-2i, Hantzsch ester 3d (1.4 equiv), and 5 Å molecular sieves in mesitylene at 10°C for 2 days. Amine 4 was obtained with a high enantiomeric ratio (e.r.) of 95.5:4.5, however the reaction was still rather sluggish and the product was isolated in only 41% yield [Eq. (1)]. We speculated that the poor yield of isolated product was still due to catalyst deactivation through salt formation with the highly basic amine 4. It has previously been reported that related product inhibition pathways can be eliminated by in situ protection.<sup>[10c,d,13]</sup> Encouraged by these reports, we investigated the effect of running the reaction in the presence of di-tert-butyl dicarbonate (Boc<sub>2</sub>O). Remarkably, we found that full conversion into the desired N-Boc-protected product 5a in 97% yield with an identical e.r. of 95.5:4.5 was observed under these conditions [Eq. (2)].



Other reagents, such as acetic anhydride and dibenzyl dicarbonate (Cbz<sub>2</sub>O), could also be used and afforded the corresponding N-Ac product in 90% yield with an e.r. of 94.5:5.5 and the N-Cbz product in 92% vield with an e.r. of 95:5, respectively (see the Supporting Information). Encouraged by this result, the scope of the enantioselective reduction of N-methyl imines was next investigated (Table 2). A variety of N-methyl imines were efficiently reduced in the presence of (R)-DSI-2i (5.0 mol%) to afford the corresponding Bocprotected N-methyl amines with high yields and enantioselectivity. Electron-donating para substituents are well tolerated (entries 1-4). For example, tert-butyl-substituted imine 1c provided amine 5c in 91% yield with an e.r. of 98.5:1.5 (entry 3). A *p*-methoxy group in imine 1d decreased the reactivity and required 5 days to achieve high yield and 96:4 e.r. (entry 4). Electron-withdrawing para substituents such as fluorine, chlorine, bromine, or trifluoromethyl groups have no significant influence on enantioselectivity or reactivity (entries 5-8). However, the *p*-cyano group in imine 1i decreased the reactivity and also required a longer reaction time to achieve full conversion with an e.r. of 95:5 (entry 9). Substitutions at the meta position are equally well tolerated and both electron-donating and electron-withdrawing groups **Table 2:** Substrate scope of the asymmetric reduction of *N*-methyl imines.<sup>[a]</sup>

N	( <i>R</i> )-DSI- <b>2i</b> (5 mol%)	, <b>3d</b> (1.4 eq	uiv), MS 5 Å	Boc N
R <sup>1</sup>	R <sup>2</sup> Boc <sub>2</sub> O (1.2 equiv) I 10	, mesitylene °C, 2 d	е (0.036 м)	R <sup>1</sup> R <sup>2</sup> 5
Entry	Products		Yield [%]	e.r. <sup>[b]</sup>
1	Boc N Ph	5 a	97	95.5:4.5
2	4-MeC <sub>6</sub> H <sub>4</sub>	5 b	88	95:5
3	4- <i>t</i> BuC <sub>6</sub> H <sub>4</sub>	5 c	91	98.5:1.5
4 <sup>[c]</sup>	4-MeOC <sub>6</sub> H <sub>4</sub>	5 d	93	96:4
5	4-FC <sub>6</sub> H <sub>4</sub>	5 e	94	95:5
6	4-CIC <sub>6</sub> H <sub>4</sub> Boc	5 f	96	95.5:4.5
7	4-BrC <sub>6</sub> H <sub>4</sub>	5 g	95	96:4
8	4-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> Boc	5 h	96	97:3
9 <sup>[c]</sup>	4-CNC <sub>6</sub> H <sub>4</sub>	5 i	95	95:5
10	3-MeC <sub>6</sub> H <sub>4</sub>	5 j	90	95:5
11 <sup>[d,e]</sup>	3-MeOC <sub>6</sub> H <sub>4</sub>	5 k	92	94:6
12	3-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	51	94	95:5
13 <sup>[c]</sup>	2-MeC <sub>6</sub> H <sub>4</sub> Boc	5 m	67	67.5:33.5
14 <sup>[c]</sup>	3,4-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	5 n	95	97:3
15	Boc	50	92	95.5:4.5
16	Boc	5 p	89	90.5:9.5
1 7 <sup>[d]</sup>	Ph	5 q	74	52.5: 47.5
18 <sup>[c]</sup>	cHex Boc	5 r	63	53.5:46.5
19	Ph	5 s	90	71:29

[a] Reactions were run on a 0.25 mmol scale. [b] Determined by HPLC with a chiral stationary phase. The absolute configuration was determined by preparing an authentic sample of 5a and by comparing various known optical rotation values (see the Supporting Information). [c] Reaction time 5 d. [d] Reaction time 3 d. [e] The S-enantiomer of disulfonimide 2i was used.



provided the corresponding amines in high yields and enantioselectivity (entries 10–12). *ortho* Substitution proved detrimental to both reactivity and selectivity, while *meta, para* disubstitution led to good results with an e.r. of 97:3 (entries 13 and 14). Interestingly, *N*-methyl imine **10**, which is derived from  $\alpha$ -tetralone, was reduced in high yield with an e.r. of 95.5:4.5 (entry 15). *N*-Methyl imine **1p**, which is derived from 1-indanone, gave less satisfactory results (entry 16). An aliphatic imine can also be employed but gave poor enantioselectivity (entries 17 and 18). Propiophenone-derived imine **1s** was reduced in 90% yield with an e.r. of 71:29 (entry 19).

An additional strength of our method is revealed upon exploring longer and substituted *N*-alkyl groups. Gratifyingly, the reduction of *N*-ethyl imine **6a** proceeded in 83 % yield and with an e.r. of 97:3 (Table 3, entry 2). Remarkably, even

Table 3: Scope of the asymmetric reduction of N-alkyl imines.[a]

	∧ Alkyl ∥ ( <i>R</i> )-DSI-2i (5 mol%), 3d	(1.4 equiv), N	IS 5 Å	N Alkyl
L 1a	Boc <sub>2</sub> O (1.2 equiv), me or <b>6</b>	esitylene (0.03 2 d	6 м)	5a or 7
Entry	Products		Yield [%]	e.r./d.r. <sup>[b]</sup>
1	Boc N Ph	5 a	97	95.5:4.5
2	Ph	7 a	83	97:3
3	Ph	7 b	90	97.5:2.5
4	Ph	7c	82	97.5:2.5
5	Boc N Ph	7 d	77	95.5:4.5
6	Boc N Ph	7e	94	97:3
<b>7</b> <sup>[c]</sup>	Boc N 3,4-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	7 f	83	95.5:4.5
8	Ph	7 g	94	97.5:2.5
<b>9</b> <sup>[c]</sup>		7 h	75	93:7
10 <sup>[c,d]</sup>	Boc N CO2Me	ent- <b>7 h</b>	79	97.5:2.5

[a] Reactions were run on a 0.25 mmol scale. [b] Determined by HPLC with a chiral stationary phase. [c] Reaction time 3 d. [d] The S-enantiomer of disulfonimide **2i** was used.

the *N*-*n*-propyl, *N*-*n*-butyl, *N*-allyl, *N*-2-phenylethyl, *N*-2-(3,4dimethoxy)-phenylethyl, and *N*-3-phenylpropyl imine substrates (**6b–6g**) are well tolerated and reacted with excellent enantioselectivity (entries 3–8). Interestingly, the reduction of imine **6h**, which is derived from methyl (tert-butoxycarbonyl)-L-lysinate, with (*R*)-DSI-**2i** resulted in a d.r. of 93:7, while (*S*)-DSI-**2i** gave a d.r. of 97.5:2.5, which is indicative of a matched combination (entries 9 and 10).



**Scheme 1.** Synthesis of (S)-Rivastigmine, NPS R-568 Hydrochloride, and (R)-Fendiline.

To illustrate the synthetic utility of our method, several marketed pharmaceuticals were prepared (Scheme 1). For example, a short synthesis of (S)-Rivastigmine, which is currently used for treating Alzheimer's disease, was realized.<sup>[14]</sup> Boc-protected *N*-methyl amine **5k** (Table 2, entry 11) was converted into N,N-dimethyl amine 8 in 93 % yield upon LiAlH<sub>4</sub> reduction. This transformation illustrates an additional benefit of the Boc moiety, which not only enables turnover in the asymmetric reduction but also introduces a useful N-methyl precursor. Next, cleavage of the O-methyl group using aqueous HBr provided (S)-3-(1-dimethylaminoethyl)phenol 9 in 95% yield. A single recrystallization from CH<sub>2</sub>Cl<sub>2</sub>/pentane gave product 9 in 76% yield with an e.r. of 99.6:0.4. Finally, carbamovlation of alcohol (S)-9 with commercially available N-ethyl-N-methylcarbamoyl chloride 10 in the presence of sodium hydride furnished (S)-Rivastigmine in 98% yield with an e.r. of 99.6:0.4.

Furthermore, the calcimimetic compound NPS *R*-568 hydrochloride, which has been shown to be effective against both primary and secondary hyperparathyroidism, was synthesized.<sup>[15]</sup> In this case, imine generation from commercially available ketone **11** and amine **12**, asymmetric reduction, and Boc deprotection, were all performed in one pot, thereby providing NPS *R*-568 in 82% yield with an e.r. of 97:3. A single recrystallization of the corresponding HCl salt from CH<sub>2</sub>Cl<sub>2</sub>/MTBE gave NPS *R*-568 hydrochloride in 82% yield with an e.r. of 99.7:0.3.

In a similar manner, an enantioselective synthesis of (R)-Fendiline, an effective antianginal drug for the treatment of coronary heart diseases, was achieved in 78% yield with an e.r. of 97.5:2.5.<sup>[16]</sup>

In summary, we have developed a highly enantioselective Brønsted acid catalyzed reduction of *N*-alkyl imines. The



reaction is catalyzed by a disulfonimide catalyst with a Hantzsch ester as a hydrogen source in the presence of  $Boc_2O$ , and it provides facile access to several enantioenriched Boc-protected *N*-alkyl amines. The in situ protection of the basic *N*-alkyl amine product with the easily removable Boc group not only solves the problem of product inhibition but also introduces a convenient *N*-methyl precursor. Our method was applied to the synthesis of the pharmaceuticals (*S*)-Rivastigmine, NPS *R*-568 hydrochloride, and (*R*)-Fendiline. The high tolerance for various *N*-alkyl amines immediately suggests the potential of our method as a general approach for the reductive cross-coupling of ketones with diverse amines. Our explorations of disulfonimide-catalyzed reactions continue.

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**Keywords:** Brønsted acids · disulfonimide · *N*-alkyl amines · organocatalysis · reduction

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