Rhodium(III)-Catalyzed Cyclative Capture Approach to Diverse 1-Aminoindoline Derivatives at Room Temperature**

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In memory of Ekkehard Winterfeldt

Abstract: A Rh^{III} -catalyzed C–H activation/cyclative capture approach, involving a nucleophilic addition of $C(sp^3)$ –Rh species to polarized double bonds is reported. This constitutes the first intermolecular catalytic method to directly access 1-aminoindolines with a broad substituent scope under mild conditions.

he 1-aminoindoline unit is prevalent in numerous commercial drugs and biologically active compounds (Figure 1),^[1] and has also been widely utilized as a chemical feedstock to access a large number of important skeletal motifs.^[2] 1-Aminoindolines are typically formed by the nitrosylation/reduction of free indolines or by intramolecular amination.^[3] These methods suffer from disadvantages such as multistep synthe-



Figure 1. Some biologically active 1-aminoindolines and -indoles.

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ses, moderate to poor yields, harsh reaction conditions, and limited substrate scope. Moreover, indoline as the parent heterocycle must be prepared in advance.^[4] To date, no intermolecular catalytic method is available to directly access the 1-aminoindoline core.^[5]

Rh^{III}-catalyzed directed C-H functionalization has emerged as a versatile synthetic approach to access complex molecules.^[6] Recently, several groups have demonstrated that C(sp²)-Rh intermediates formed by insertion of alkynes into the C-Rh bond could undergo Grignard-type addition to ketone-, imine-, amide-, and ester-based directing groups (Figure 2A).^[7] However, the nucleophilic attack of C(sp³)-Rh species, generated upon alkene insertion, to polarized double bonds has never been reported, presumably due to problematic preferential β -hydride elimination (Figure 2B).^[8] We wondered whether nucleophilic additions of C(sp³)-Rh species to polarized unsaturated directing groups might be possible if β -hydride elimination is inhibited. Herein, we introduce diazenecarboxylate as a directing group to trigger a new cyclative capture approach where alkenes undergo insertion followed by intramolecular addition of the C(sp³)-Rh species to the N=N bond to afford diverse 1-aminoindolines without any external oxidants (Figure 2C). This reaction proceeds at room temperature with a wide range of functionalized substrates, and, moreover, has been extended to the synthesis of pharmaceutically important 1-aminoindoles.

Besides disfavoring β -hydride elimination of the C(sp³)– Rh species, two other potential difficulties must be overcome:

- Even though 1,2-diaryl- and 1-alkyl-2-aryl-substituted diazenes have been used as directing groups, diazenecarboxylate has never been used as a directing group;^[9]
- Although the addition of organometallic reagents to N=N bonds has been reported,^[10] it has not been observed in Rh^{III}-catalyzed C-H activation. Furthermore, the issue of regioselectivity in additions to N=N bonds must also be addressed.

We initially focused on the reaction of *tert*-butyl 2phenyldiazenecarboxylate **1a** and *n*-butyl acrylate (**2a**) with $[(Cp*RhCl_2)_2]$ (2.5 mol%) in 1,2-dichloroethane (DCE, 1 mL). After extensive screening, we found that the tandem C-H bond activation/N=N bond addition occurred smoothly in the presence of $[{RhCp*Cl_2}_2]$ (2.5 mol%; Cp* = C₅Me₅) and AgOAc (10 mol%) in a mixture of HOAc and DCE (1:3) at room temperature for 24 h to afford 1-aminoindoline **3a** in 88% yield. To confirm the structure, we further converted the oily **3a** to the corresponding 1-aminoindole which formed single crystals suitable for X-ray analysis (see the Supporting

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A) Rh^{III}-catalyzed directed C-H activation/addition cascade with alkynes



Challenges: • Diazenecarboxylate as directing group

Competition between β-hydride elimination and N=N bond addition

• Reactivity and regioselectivity in the addition of C(sp³)-Rh species to the N=N bond



Information).^[11] Subsequently, the scope of the aryl-substituted diazenecarboxylate substrates was investigated. A series of 1-aminoindolines bearing substituents at the 5-, 6-, and/or 7-positions were synthesized in good yields (Scheme 1; **3a–I**). A variety of important functional groups such as halogens (F, Cl, and Br), ester, and methoxy groups were well tolerated (Scheme 1; **3e–j**). In addition, we proved that the reaction can be conducted on a gram scale without a significant decrease in yield (ca. 0.9 g of **3a**, 86% yield). It is also worth noting that a remarkable yield was also obtained with a lower catalyst loading (85% yield, 1 mol% [{RhCp*Cl₂}₂]).

2.5 mol% [{RhCp*Cl₂}₂] 10 mol% AgOAc COOⁿBu COOⁿBu DCE/HOAc (3:1) RT, 24 h Boc 2a 3 NHBoc COOⁿBu COOⁿBu COOⁿBu NHBoc NHBoc NHBoc 3a, 88% (86%)^[a] 3c, 80% 3b. 91% H₃CO COOⁿBu COOⁿBu COOⁿBu OCH3 NHBoc NHBoc NHBoc 3e, 89% 3d, 93% 3f, 90% Br COOⁿBu COOⁿBu -COOⁿBu NHBoc NHBoc NHBoc 3g, 85% 3h, 82% 3i, 91% MeOOC COOⁿBu COOⁿBu COOⁿBu NHBoc NHBoc NHBoc 3j, 76% 3k. 87% 31.93%

We next examined the scope of the alkenes (Scheme 2). A variety of electrondeficient alkenes could be efficiently converted into the corresponding products (Scheme 2; 4a-f). Unexpectedly, the reaction with acrolein directly afforded the indole-2-carbaldehyde **4g**.^[12] Moreover, various substituted styrenes also reacted with 1a to furnish the products in moderate yields (Scheme 2; 4h-k). In addition to styrenes, alkyl alkenes could also give the desired products (Scheme 2; 41,m). It is important to note that 1,3-dienes are welltolerated in this reaction (Scheme 2; 4n,o). Interestingly, when we switched the substrate to vinyl acetate, we could directly access the aminoindole product 4p in excellent yield; this reaction might involve an elimination of acetic acid after cyclization.[13]

To date, synthetic approaches to 1aminoindoles, which also display very important pharmacological properties,^[14] have been quite limited.^[15] We wondered

if we could utilize the present transformation for the straightforward synthesis of 1-aminoindoles by introducing an additional oxidant. We found that 1-aminoindole 5a could be formed when AgOAc was used as the external oxidant under the standard conditions. Further optimization lead to



Scheme 2. Scope of alkenes. Reaction conditions: **1** (0.2 mmol), **2** (1.5 equiv); see SI. [a] 3.0 mmol scale. [b] Acrolein was used. [c] 0.4 mmol scale. [d] 2.5 mmol scale. [e] Vinyl acetate was used.

Scheme 1. Scope of aryl-substituted diazenecarboxylates. Reaction conditions: **1** (0.2 mmol), **2a** (1.5 equiv); see SI. [a] 3.0 mmol scale.

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2

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82% yield when the reaction temperature was increased to 90°C. Using this procedure, a variety of aryl-substituted diazenecarboxylates and alkenes were tested to access various 1-aminoindoles. As shown in Scheme 3, the desired products were generally obtained in good yields.

Furthermore, we demonstrated the synthetic usefulness of the 1-aminoindolines by conducting five additional transformations with our products (Scheme 4). First, the N-N bond of the 1-aminoindoline was easily cleaved to yield the free indolines **6a,b** using TiCl₃.^[16] We could also submit compound 4h to an initial deprotection, followed by a photocatalytic oxidation with [Ir{dF(CF₃)ppy}₂(dtbbpy)]PF₆ (dF- $(CF_3)ppy = 2-(2,4-difluorophenyl)-5-trifluoromethylpyridine,$ dtbbpy = 4,4'-di-tert-butyl-2,2'-bipyridine), visible light, and air to afford the free indole 6c.^[17] Cinnoline 6d could also be prepared from 2-phenyl-1-aminoindole in good yield.^[18] The tetracyclic fused indole skeleton 6e, which exists in numerous privileged pharmaceuticals and natural alkaloids,^[2g-i] could also be constructed from 1-aminoindoline by a Fischer indole synthesis (Scheme 4; 6e). In addition, we also made the backbone of indolo[1,2-b]indazoles, which have been shown to be potent inhibitors to DNA topoisomerases I and II (Scheme 4; 6 f).^[19]



Scheme 3. Synthesis of 1-aminoindoles. Reaction conditions: 1 (0.2 mmol), **2** (1.5 equiv); see SI. [a] 3.0 mmol scale.



Scheme 4. Applications of 1-aminoindolines/1-aminoindoles. DPE-phos = Bis[(2-diphenyl-phosphino)phenyl] ether, TFA = trifluoroacetic acid.

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carried out. First, we submitted 1-methyl-2-phenyldiazene as the substrate to the standard conditions with *n*-butyl acrylate (Scheme 5 a). The β -hydride elimination takes place preferentially to afford the product **7a** in moderate yield.^[20] However, if we change the Boc group to -COOEt, the cyclative capture step occurs smoothly to afford 1-aminoindoline **7b** in 69% yield (Scheme 5 a). These observations imply that the carboxyl substituent in our directing group might be crucial in precluding β -hydride elimination. When the reaction was performed with AcOD to roughly 70% conversion, deuterium incorporation was observed at the C7position of **3a** (Scheme 5b). This deuteration might be induced by reversible C–H activation or acid-catalyzed tautomerism. Also, deuterium incorporation was not observed at the C2-position of the product (Scheme 5b),

A series of mechanistically insightful experiments were



Scheme 5. Mechanistic experiments.

which indicates that migratory insertion of alkene might be irreversible. Furthermore, deuterium incorporation was also observed using substrate **1a** in AcOD/DCE in the absence of Rh catalyst implying that the acidcatalyzed tautomerism might occur during the reaction (see SI). The kinetic isotope effect (KIE) experiment was also carried out to independently assess the rate of reaction for *ortho*-C-H versus *ortho*-C-D activation. The value of $k_{\rm H}/k_{\rm D} = 4.1$ from two parallel reactions as well as the value of KIE = 6.3 from intermolecular competition indicated that the C–H bond-cleavage process is likely involved in the rate-determining step (Scheme 5 c).^[21] Additionally, it was observed that the reactivities of aryl-substituted diazene-carboxylates with an electron-donating substituent (e.g., OCH₃) are obviously higher than those with an electron-withdrawing substituent (e.g., *p*-COOMe), which is consistent with an electrophilic

C-H activation mechanism.^[22]

Furthermore, we prepared the rhodacycle Α (Scheme 5d),^[23] which can be easily transformed to rhodacycle B in the presence of stoichiometric silver acetate (Scheme 5e). Importantly, when 4 mol% rhodacycle A served as the catalyst in the presence of 4 mol% AgOAc in HOAc/DCE (1:3) at RT, the reaction of 1a and 2a works efficiently (Scheme 5 f), suggesting the plausible intermediacy of a cyclometalated complex **B** in the catalytic cycle. Furthermore, we monitored the reaction of rhodacycle B with methyl acrylate in CD₂Cl₂ by ¹H NMR spectroscopy (see SI). No reaction was observed in the absence of HOAc. When HOAc was added to the reaction system, the signal of product 4a was detected immediately. Presumably HOAc promotes the dissociation of the coordinated acetate of rhodacycle B to form the key intermediate cation C to initiate this reaction.^[22]

On the basis of our preliminarily mechanistic experiments and literature precedence,^[6,7] we propose that a directed C–H bond cleavage forms intermediate **B** as the first step after generation of [RhCp*(OAc)₂]. An acetate ligand of rhodacycle **B** dissociates to give the cationic complex **C** with assistance from HOAc;^[22] this is followed by alkene coordination and insertion, thus affording the rhodacycle **E** (Scheme 6). The resulting seven-membered metallacycle presumably rearranges to the more stable six-membered coordinately saturated Rh species **F** by chelation with the Boc substituent,^[24] which efficiently prevents competitive β -hydride elimination.^[25] Complex **F** might undergo a nucleophilic addition to the N=N bond to form intermediate **G**, which is finally protonated by HOAc to yield the desired 1-aminoindoline.



Scheme 6. Proposed reaction mechanism.

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In summary, we have developed the first Rh^{III}-catalyzed synthesis of 1-aminoindolines and 1-aminoindoles from arylsubstituted diazenecarboxylates and alkenes. Mechanistic studies support a pathway that proceeds via reversible C–H activation, alkene insertion, and Grignard-type addition. This intermolecular annulation proceeds under mild conditions, does not require external oxidants, and displays a broad scope with respect to the substituents.

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$$\underbrace{N_{N}}_{N} \underbrace{Rh^{|||} \text{ catalysis}}_{COO''Bu} \underbrace{N_{N}}_{COO''Bu} \\ \beta - \text{hydride elimination intermediate} \\ \underbrace{Michael \ addition}_{\text{isomerization}} \underbrace{V_{N}}_{Ta} \underbrace{N_{N}}_{COO''Bu} \\ \underbrace{N_{N}}_{Ta} \underbrace{N_{N}}_{Ta} \underbrace{N_{N}}_{Ta} \underbrace{N_{N}}_{Ta} \\ \underbrace{N_{N}}_{Ta} \underbrace{N_{$$

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Angew. Chem. Int. Ed. 2014, 53, 1-6

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Rhodium(III)-Catalyzed Cyclative Capture Approach to Diverse 1-Aminoindoline Derivatives at Room Temperature



Buckle up! The nucleophilic addition of C(sp³)–Rh species to polarized double bonds is the key step in a Rh^{III}-catalyzed C–H activation/cyclative capture reaction. This constitutes the first intermo-

lecular catalytic method to directly access the 1-aminoindoline core with a broad substituent scope under mild conditions (Boc = tert-butoxycarbonyl, DG = directing group).

6 www.angewandte.org