Lipase-catalyzed Transesterification of Methyl 2-Substituted 3-Hydroxy-4pentenoates and its Synthetic Application to the Taxol Side Chain

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Abstract: *Syn*-and *anti*-methyl 2-substituted 3-hydroxy-4-pentenoates were efficiently resolved in lipase-catalyzed transesterification. This protocol was successfully applied to the synthesis of the taxol side chain.

Key words: *syn-* and *anti-*methyl 2-substituted 3-hydroxy-4-pentenoates, lipase-catalyzed kinetic resolution, 1,2-amino alcohols, the taxol side chain

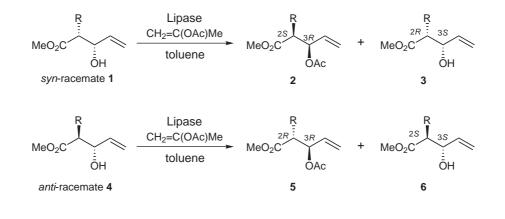
Acyclic 1,2-amino alcohols are not only important constituents of natural products such as sphingosine,¹ statine,² and phenylisoserine³ as the taxol side chain, but also significant chiral auxiliaries for diverse asymmetric syntheses.⁴ Among a number of synthetic methods for acyclic 1,2-amino alcohols reported hitherto, the sequence of Evans' asymmetric aldol reaction/the release of a 2-substituted 3-hydroxycarboxylic acid/Curtius rearrangement⁵ has been well recognized as a useful method owing to the reliability of stereochemistry throughout the process. For these reasons we wanted to devise a more straightforward access to both *syn-* and *anti-*2-substituted 3-hydroxy esters with high optical purities that takes the place of Evans' asymmetric aldol reaction.⁶

To this end, we envisioned to utilize lipase-catalyzed transesterification⁷ of 2-substituted 3-hydroxy esters. A very few related studies were reported so far on lipase-catalyzed transesterifications of 3-hydroxy esters such as *cis*-

1-ethoxycarbonyl-2-hydroxycyclohexane,⁸ *tert*-butyl 3hydroxy-4-pentenoate,⁹ 4-aryloxy-3-hydroxy esters,¹⁰ and *syn*- and *anti*-2-substituted 3-hydroxy esters¹¹ whose substituents at the C-2 position were severely limited to Ph-, MeS-, and PhS- groups. To the best of our knowledge, there has not been reported to date a general and efficient lipase-catalyzed transesterification for acyclic 2substituted 3-hydroxy esters.¹²

In this context, we designed *syn*- and *anti*- methyl 2-substituted 3-hydroxy-4-pentenoates 1^{13} and 4^{13} as enzymatically acceptable substrates of high synthetic flexibility, disclosing that (3*R*)-alcohols¹⁴ for both isomers were acetylated with exceedingly high enantioselectivities in lipase-catalyzed transesterification as shown in Scheme 1.

The reaction with *syn*-racemates **1** was executed as follows. Stirring a solution of **1** (7~38 mmol) and 2-propenyl acetate (3 equiv) in toluene (1.5 mL per mmol of **1**) with Chirazyme[®] (*Candida antarctica*, fraction B, 0.5 g per g of **1**)¹⁵ at room temperature ~70 °C for 40~72 hours resulted in the clean acetylation of one enantiomer to give (2*S*,3*R*)-acetates **2**.¹⁶ The unsolved lipase was filtered off and the filtrate was concentrated and purified by column chromatography (SiO₂). Gratifyingly, both the enantiose-lectivity and the chemical yield of the unreacted (2*R*,3*S*)-alcohols **3**¹⁶ were also found to be excellent. Similarly, the reaction with *anti*-racemates **4** proceeded quite efficiently under the same reaction conditions to afford both (2*R*,3*R*)-





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acetates 5^{16} and the unreacted (2*S*,3*S*)-alcohols 6^{16} in a highly enantioselective manner, the results being summarized in Tables 1 and Table 2.

To determine the absolute configuration of alcohols **6**, **6a** (>99% ee, 4.76 mmol) was transformed into alcohol **9**, an aggregation pheromone of the smaller European elm bark beetle (*Scolytus multistriatus*),¹⁸ as depicted in Scheme 2. Hydrogenation of the terminal olefin (HN=NH/MeOH,¹⁹ 96%) provided saturated alcohol **7**. Protection of the hydroxyl group (CH₂=CHOEt, 93%) followed by reduction of the methyl ester (LiAlH₄, 98%) gave rise to alcohol **8**. Tosylation of **8** (*p*-TsCl, 91%), reduction (LiAlH₄, 92%) and deprotection (3 M HCl, 82%) afforded **9** {[α]_D²⁴ –22.7 (*c* 1.00, hexane), Lit.²⁰ [α]_D¹⁹ –21.4 (c 1.02, hexane)} in 60% overall yield.

To demonstrate the synthetic utility of our method, we executed the synthesis of the taxol side chain. *Anti*-aldol isomer 10^{21} (1.26 g, 5.2 mmol) was subjected to the transesterification [Chirazyme[®] (0.6 g), 2-propenyl acetate (1.72 mL, 15.6 mmol), toluene (7.8 mL)], providing acetate **11** (44%, 98.5%ee) and unreacted alcohol **12** (49%, >99%ee) (Scheme 3).

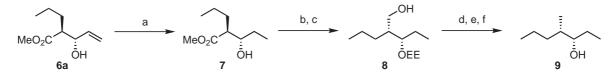
Compound 12 thus obtained was transformed into cyclic carbamate 13 in 80% yield via Curtius rearrangement of the free acid released by palladium-catalyzed hydrogenolysis²² of the allyl ester. Protection of the nitrogen with (Boc)₂O and oxidative cleavage (RuO₂/NaIO₄)²³ of the double bond furnished carboxylic acid 14 in 88% yield. Ring opening (2 M NaOH/MeOH, r.t., 2 h), deprotection of the Boc group, and ensuing benzoylation provided *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine (15) in 71%

Table 1	Lipase-catalyzed Kineti	c Resolution of <i>syn</i> -Racemate 1

Entry	1: R-	Temp. (°C)/Time (h)		Yield (%ee)		
				Acetate 2	Alcohol 3	E value ¹⁷
1	<i>n</i> -C ₃ H ₇ -	1 a	70/72	40 (99): 2a	47 (>99): 3a	>1000
2	(CH ₃) ₂ CHCH ₂ -	1b	70/72	41 (96): 2b	45 (86): 3b	136
3	trans-EtCH=CH-	1c	70/40	48 (99): 2c	49 (99): 3c	>1000
4	(CH ₃) ₂ C=CHCH ₂ -	1d	70/72	49 (>99): 2d	48 (>99): 3d	>1000
5	CH ₂ =CH(CH ₂) ₇ -	1e	70/72	43 (99): 2e	51 (98): 3e	922
6	PhSCH ₂ CH ₂ -	1f	50/40	48 (>99): 2f	50 (>99): 3f	>1000
7	CH ₃ -	1g	r.t./72	48 (89): 2g	42 (>99): 3g	90

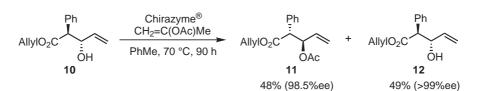
 Table 2
 Lipase-catalyzed Kinetic Resolution of anti-Racemate 4

Entry	4 : R-	Temp. (°C)/Time (h)		Yield (%ee)		
				Acetate 5	Alcohol 6	E value ¹⁷
1	<i>n</i> -C ₃ H ₇ -	4a	70/72	45 (94): 5a	46 (>99): 6a	170
2	(CH ₃) ₂ CHCH ₂ -	4b	70/72	43 (96): 5b	45 (>99): 6b	259
3	trans-EtCH=CH-	4c	70/40	48 (96): 5c	46 (97): 6c	207
4	(CH ₃) ₂ C=CHCH ₂ -	4d	70/72	48 (93): 5d	46 (>96): 6d	145
5	CH ₂ =CH(CH ₂) ₇ -	4e	70/72	49 (>99): 5e	48 (>99): 6e	>1000
6	PhSCH ₂ CH ₂ -	4f	50/40	50 (98): 5f	50 (>99): 6f	525
7	CH ₃ -	4g	r.t./72	46 (96): 5g	44 (>99): 6g	259

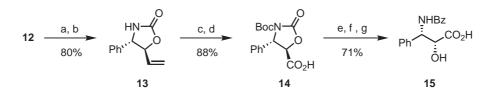


 $\begin{array}{l} \textbf{Scheme 2} \quad (a) \ KO_2 CN = NCO_2 K/HOAc/MeOH, 0 \ ^\circ C \ to \ r.t., 12 \ h, 96\%; (b) \ CH_2 = CHOEt/cat. \ PPTS/CH_2 Cl_2, r.t., 6 \ h, 93\%; (c) \ LiAlH_4/THF, 0 \ ^\circ C, 0.5 \ h, 98\%; (d) \ p-TsCl/cat. \ DMAP/pyridine, r.t., 20 \ h, 91\% \ (e) \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ 3 \ M \ HCl/EtOH, r.t., 4 \ h, 82\% \ LiAlH_4/THF, 0 \ ^\circ C \ to \ r.t., 12 \ h, 92\%; (f) \ R \ h, 82\%; (f) \ R \ h, 82\%;$

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Scheme 3



Scheme 4 (a) $Pd(OAc)_2/PPh_3$, $HCO_2H/Et_3N/THF$, r.t., 10 h; (b) $DPPA/Et_3N/toluene$, r.t., 40 min, then 80 °C, 1.5 h; (c) $(Boc)_2O/Et_3N/DMAP/THF$, r.t., 2 h; (d) $RuO_2/NaIO_4$, $CCl_4/MeCN/H_2O = 2:2:3$, r.t., 15 h; (e) 2 M NaOH/MeOH, r.t., 1 h; (f) TFA, r.t., 2 h; (g) BzCl/aq NaHCO_3, r.t., 6 h

yield { $[a]_D^{25}$ -35.3 (*c* 1.03, EtOH), mp 173.5–176 °C, Lit.²⁴ $[a]_D^{20}$ -35.5 (*c* 1.07, EtOH), mp 175.5–177 °C} as outlined in Scheme 4.

In summary, we have established a practical method for obtaining *syn*- and *anti*-methyl 2-substituted 3-hydroxy-4-pentenoates with high optical purities by means of the lipase-catalyzed transesterification, which would offer access to other synthetically useful intermediates involving 1,2-amino alcohols less available from natural amino acids.

Acknowledgement

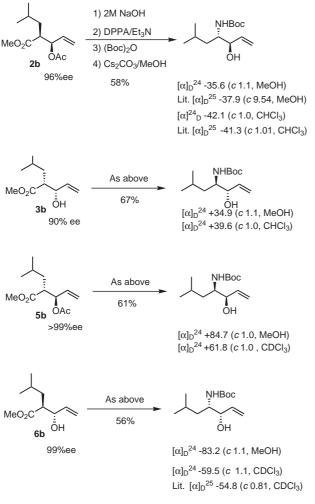
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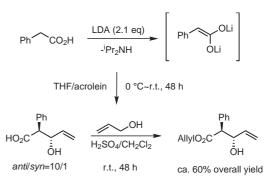
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- (13) For instance, compounds 1b and 4b were prepared as follows. A solution of methyl 4-methylvalerate (7.48 g, 57.5 mmol) in THF (20 mL) was added dropwise to a solution of LDA (60.3 mmol) in hexane (38.7 mL)/THF (50 mL) at -78 °C and the mixture was stirred for 1 h. Then, acrolein (4.61 mL, 68.9 mmol) was added and the mixture was stirred at -78 °C for 2 min. After usual workup, the oil obtained was purified by medium-pressure column chromatography (Yamazen, Ultra PackTM, \emptyset 50 × 300 mm, hexane/ethyl acetate = $6:1 \sim 4:1$ as eluent) to afford the less polar synracemate 1b (5.03 g, 27.0 mmol, 47% yield) and the more polar anti-racemate 4b (4.70 g, 25.2 mmol, 44% yield). Each compound can easily be discriminated by the chemical shifts of methine protons at 2- and 3-positions in ¹H NMR spectra (CDCl₃, 500 MHz) : δ 2.62–2.68 (m, 1 H, CHCOO) and 4.28–4.33 (m, 1 H, CHOH) for **1b**, and δ 2.58–2.63 (m, 1 H, CHCOO) and 4.14–4.20 (m, 1 H, CHOH) for 4b. The chemical shifts of the methine protons at 2- and 3-positions of 1 are generally observed in the down field compared with those of 4. Additionally, syn-racemates 1 are generally less polar than anti-racemates 4 on TLC analysis (hexane/ethyl acetate = $5:1 \sim 3:1$).
- (14) Lipase-catalyzed enantioselective acylation of 1-alkene-3ols has been well examined, see: (a) Ito, T.; Akasaki, E.; Kudo, K.; Shirakami, S. *Chem. Lett.* **2001**, 262. (b) Ito, T.; Kudo, K.; Tanaka, N.; Sakabe, K.; Takagi, Y.; Kihara, H. *Tetrahedron Lett.* **2000**, *41*, 4591.

- (15) The quantity of the lipase was not optimized. The lipase collected was washed with ether, dried in air for 5 min and stored below 5 °C. The lipase retains full activity and can be reused at least three times.
- (16) The enantiomeric purity of alcohols **3** and **6** were determined by Chiralcel OD-H (hexane-2-propanol, 230 nm or 254 nm). Acetates 2 and 5 were also analyzed by Chiralcel OD-H (hexane/2-propanol, 230 nm or 254 nm) after having been converted to the corresponding alcohols via methanolysis (K₂CO₃/MeOH, r.t., 1 h). Some of them were transformed into known 1,2-amino alcohols to confirm their absolute configuration. For instance, acetates 2b (96% ee), 3b (90% ee after resubjection to the lipase-catalyzed transesterification), 5b (>99% ee), and 6b (99% ee) were converted to N-protected amino alcohols in 58%, 67%, 61%, and 56% overall yields, respectively via sequential alkaline hydrolysis (2 M NaOH/MeOH, r.t., 2 h), Curtius rearrangement (DPPA/Et₃N/toluene, r.t., 4 h and 80 °C, 1 h), N-protection [(Boc)₂O/Et₃N/cat. DMAP/THF, r.t., 2 h], and ring opening (0.4 equiv. of Cs_2CO_3 /MeOH, r.t., 11 h),²⁵ the results being summarized below (Scheme 5). The optical rotation values were compared with those of reported in the literatures. See: (a) Leanna, M. R.; DeMattei, J. A.; Li, W.; Nichols, P. J.; Rasmussen, M.; Morton, H. E. Org. Lett. 2000, 2, 3627. (b) DeMattei, J. A.; Leanna, M. R.; Li, W.; Nichols, P. J.; Rasmussen, M. W.; Morton, H. E. J. Org. Chem. 2001, 66, 3330.



Scheme 5

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- (21) (a) The attempted alkaline hydrolysis of the corresponding methyl ester of 10 failed entirely due to the intensive retroaldol reaction. Thus we adopted the allyl ester 10 cleanly convertible to the free carboxylic acid by the palladiumcatalyzed hydrogenolysis under a neutral condition. The compound 10 was prepared according to the Mulzer's protocol: Mulzer, J.; Segner, J.; Brüntrup, G. Tetrahedron Lett. 1977, 4651. (b) A solution of phenylacetic acid (13.6 g, 100 mmol) in THF (40 mL) was added dropwise to a stirred solution of LDA (210 mmol) in THF-hexane (140 mL/135 mL) at 0 °C and the reaction mixture was stirred at 0 °C for 45 min and at r.t. for 2 h. The solvent was removed under reduced pressure and the residual viscous material was dried in vacuo at 70 °C for 2 h to give pale yellow solids. The lithium enolate thus obtained was dispersed in THF (100 mL) and acrolein (8.02 mL, 120 mmol) was added dropwise at 0 °C. After being stirred at r.t. for 48 h, the solvent was evaporated and ice-cold 3 N HCl (120 mL) was added to the residue. Extraction with CHCl₃ gave 2-phenyl-3-hydroxy-4pentenoic acid (*anti/syn* = ca. 10:1) as a viscous oil (17.4 g) which without purification was esterified with allyl alcohol (14.9 mL, 220 mmol) in methanol-free CH₂Cl₂ (150 mL) in the presence of concd H_2SO_4 (2 mL) at r.t. for 48 h to give a diastereomeric mixture of allyl esters. The anti-ester 10 was isolated in 60% overall yield (Scheme 6) by mediumpressure column chromatography (SiO₂, toluene–EtOAc = 10:1~5:1).



Scheme 6

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