Tetrahedron Letters 53 (2012) 2508-2510

Contents lists available at SciVerse ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

$Ga(ClO_4)_3$ -catalyzed synthesis of guinoxalines by cycloaddition of α -hydroxyketones and o-phenylenediamines

Fan Pan^a, Tang-Ming Chen^a, Jia-Jia Cao^a, Jian-Ping Zou^{a,*}, Wei Zhang^{b,*}

^a Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry and Chemical Engineering, Suzhou University, 199 Renai Street, Suzhou, Jiangsu 215123, China ^b Department of Chemistry, University of Massachusetts Boston, 100 Morrissey Boulevard, Boston, MA 02125, USA

ARTICLE INFO

Article history: Received 6 February 2012 Accepted 27 February 2012 Available online 11 March 2012

Keywords: Ouinoxalines α-Hydroxyketones o-Phenylenediamines Cycloaddition Gallium perchlorate

ABSTRACT

A new approach for the synthesis of 2-substituted quinoxalines by $Ga(ClO_4)_3$ -catalyzed cycloaddition of α -hydroxyketones and o-phenylenediamines is introduced. The reaction is catalyzed by 10 mol % of Ga(ClO₄)₃ in EtOH at room temperature. It is performed under simple and mild conditions to afford the product in good yield.

© 2012 Elsevier Ltd. All rights reserved.



As part of our continuous effort on the development of gallium(III) salts-catalyzed reactions, we have recently reported Ga(OTf)₃-promoted synthesis of 1.5-benzodiazepines by the cycloaddition of 1,2-aryldiamine or 2-aminothiophene with ketones or chalkones.¹⁴ We also reported the synthesis of 2,3-substituted quinoxalines by the cycloaddition of o-phenylenediamines and diaryl-1,2-diketones.¹⁵ Introduced in this Letter are Ga(ClO₄)₃-catalyzed reactions of 1,2-aryldiamines with α -hydroxyketones for the synthesis of 2-substituted quinoxalines (Scheme 1).

o-Phenylenediamine 1a and 2-hydroxyacetophenone 2a were selected as the model compounds for the screening of reaction sol-



Scheme 1. Ga(ClO₄)₃-promoted synthesis of quinoxalines.

Table 1 The reaction of **1a** with **2a** under different conditions^a

Entry	Catalyst (10 mol %)	Solvent	Time (h)	Yield ^b (%)
1	$Ga(ClO_4)_3$	EtOH	4	95
2	$Ga(ClO_4)_3$	MeOH	4	96
3	$Ga(ClO_4)_3$	CH ₃ CN	4	62
4	Ga(ClO ₄) ₃	CH_2Cl_2	4	35
5	Ga(ClO ₄) ₃	DMSO	4	71
6	$Ga(ClO_4)_3$	H ₂ O	4	25
7	Ga(OTf) ₃	EtOH	4	55
8	Ga(NO ₃) ₃	EtOH	4	Trace
9	La(OTf) ₃	EtOH	4	72
10	Yb(OTf) ₃	EtOH	4	75
11	None	EtOH	4	Trace

^a All the reactions were performed using **1a** (0.5 mmol) and **2a** (0.5 mmol) at 25 °C. ^b Isolated yield.

vent, catalyst, and related reaction conditions (Table 1). When the reaction was carried out with 10 mol % of $Ga(ClO_4)_3$ in EtOH or MeOH, 3a was obtained in 95% and 96%, respectively, (Table 1, entries 1 and 2). Reactions in other solvents such as MeCN. CH₂Cl₂.





^{*} Corresponding authors. Tel./fax: +86 512 65880336 (J.-P.Z.); tel.: +1 617 287 6147; fax +1 617 287 6030 (W.Z.).

E-mail addresses: jpzou@suda.edu.cn (J.-P. Zou), wei2.zhang@umb.ed (W. Zhang).

^{0040-4039/\$ -} see front matter © 2012 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.tetlet.2012.02.113

Table 2	
Ga(ClO ₄) ₃ -catalyzed reactions of diamines and α -hydroxyketone	sa

Entry	1,2-Diamine 1	α -Hydroxyketones 2	Product 3	Yield % (ratio)
1	NH ₂ NH ₂ 1a	O 2a		95
2	1a	O OH 2b		90
3	1a	MeO 2c	N 3c OMe	86
4	1a	Br 2d	N 3d Br	88
5	1a	CI 2e OH		84
6	1a	O O ₂ N 2f		71
7	NH ₂ NH ₂ 1b	2a		87 (1:1)
8	16	2d	3g and 3g'	82 (1:1)
9	NH ₂ NH ₂	2a		88
10	CI NH ₂ NH ₂ 1d	2a	CI N $CI $ N $CI $ N CI N N N CI N	87 ^b (2:1)
11	O NH ₂ NH ₂	2a		75 ^b
12	O ₂ N NH ₂ If	2a		61 ^b
13	1a	O OH 2g		70 ^c

^a Reactions were performed using 1 (0.5 mmol) and 2 (0.5 mmol) with Ga(ClO₄)₃ (10 mol %) at 25 °C.
 ^b Performed at 35 °C.
 ^c Reflux.



Scheme 2. Proposed reaction mechanism.

DMSO, and H₂O gave the product in much lower yields (Table 1, entries 3–6). Reactions with other gallium salts or with lanthanum and ytterbium salts in EtOH were also conducted. It was found that $Ga(OTf)_3$ gave 55% yield of **3a** (Table 1, entry 7), whereas, $La(OTf)_3$ and Yb(OTf)₃ gave **3a** in 72% and 75%, respectively. Thus, using equal molar substrates of **1a** and **2a** with 10 mol % of $Ga(ClO_4)_3$ in EtOH was selected as the optimized condition for the synthesis of 2-substituted quinoxalines **3**.

Using the optimized condition, we conducted the cycloadditions of various o-phenylenediamines 1 and α -hydroxyketones 2 (Table 2). It was found that α -hydroxyketones **2** containing electron-donating and weak electron-withdrawing substitution groups such as Me, MeO, Br, and Cl afforded products in good yields (84-95%) (Table 2, entries 2-5). However, strong electron-withdrawing substitution groups such as NO₂ reduced the yield to 71% yield (Table 2, entry 6). The reactivity and regioselectivity of the substituents on o-phenylenediamines 1 were also investigated. It was found that o-phenylenediamines **1b** and **1d**, each has a group at the 4-position, afforded corresponding guinoxalines as two isomers (Table 2, entries 7, 8, 10). Interestingly, reactions of 1e and 1f yielded the corresponding quinoxalines 3k and 3l as single isomers, respectively (Table 2, entries 11 and 12). The reaction of 2hydroxy-1,2-diphenylethanone 2g with 1a was also proceeded smoothly to give 2,3-diphenylquinoxaline **3m** in 70% yield (Table 2, entry 13).

Two possible routes for $Ga(ClO_4)_3$ -catalyzed reaction of diamines and α -hydroxyketones are proposed in Scheme 2. The first route involving intermediates **4** and **5** is proceeded through the substitution/cyclization/aromatization steps to form quinoxaline **3**. The reaction is also possible to go through an alternative pathway in which diaryl-1,2-diketone **2** is oxidized to 1,2-diketone **4** and then undergoes cycloaddition with 1,2-aryldiamine **1** to afford product **3**. In summary, a new Ga(ClO₄)₃-catalyzed reaction of diamines with α -hydroxyketones to afford 2-substituted quinoxalines is developed. Compared to the literature methods, this reaction has the advantages of simple manipulation, mild reaction conditions, and high product yields.

Acknowledgment

J.-P.Z. thanks the financial support from The National Natural Science Foundation of China (Nos. 20772088 & 21172163).

Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2012.02.113.

References and notes

- Selected reviews and book chapters for the synthesis and bioactivities of quinoxalines: (a) Husain, A.; Madhesia, D. J. Pharm. Res. 2011, 4, 924–929; (b) Patidar, A. K.; Jeyakanda, M.; Mobiya, A. K.; Selvam, G. Int. J. PharmTech. Res. 2011, 3, 386–392; (c) Saifina, D. F.; Mamedov, V. A. Russ. Chem. Rev. 2010, 79, 351–370; (d) Lv, M.; Xu, H. Comb. Chem. High Throughput Screen. 2010, 13, 293– 301; (e) Gobec, S.; Urleb, U. In Science of Synthesis; 2004; Vol. 16, p 845–911
- Sarges, R. H.; Howard, R.; Browne, R. G.; Lebel, L. A.; Seymour, P. A. J. Med. Chem. 1990, 33, 2240–2254.
- 3. Seitz, L. E.; Suling, W. J.; Reynolds, R. C. J. Med. Chem. 2002, 45, 5604-5606.
- Dailey, S.; Feast, J. W.; Peace, R. J.; Sage, I. C.; Till, S.; Wood, E. L. J. Mater. Chem. 2001, 11, 2238–2243.
- (a) Bhosale, R. S.; Sarda, S. R.; Ardhapure, S. S.; Jadhav, W. N.; Bhusare, S. R.; Pawar, R. P. *Tetrahedron Lett.* **2005**, *46*, 7183–7186; (b) More, S. V.; Sastry, M. N. V.; Wang, C. C.; Yao, C. F. *Tetrahedron Lett.* **2005**, *46*, 6345–6348; (c) Darabi, H. R.; Mohandessi, S.; Aghapoor, K.; Mohsenzadeh, F. *Catal. Commun.* **2007**, *8*, 389– 392; (d) Heravi, M. M.; Taheri, S.; Bakhtiari, K.; Oskooie, H. A. *Catal. Commun.* **2007**, *8*, 211–214; (e) Huang, T. K.; Wang, R.; Shi, L.; Lu, X. X. *Catal. Commun.* **2008**, 9, 1143–1147; (f) Srinivas, C.; Kumar, C. N. S. S. P.; Rao, V. J.; Palaniappan, S. J. Mol. *Catal. A: Chem.* **2007**, *265*, 227–230.
- (a) Hazarika, P.; Gogoi, P.; Konwar, D. Synth. Commun. 2007, 37, 3447–3454; (b) More, S. V.; Sastry, M. N. V.; Yao, C. F. Green Chem. 2006, 8, 91–95.
- 7. Antoniotti, S.; Donach, E. Tetrahedron Lett. 2002, 43, 3971-3973.
- Das, B.; Venkateswarlu, K.; Suneel, K.; Majhi, A. Tetrahedron Lett. 2007, 48, 5371–5374.
- Madhav, B.; Narayana Murthy, S.; Prakash Reddy, V.; Rama Rao, K.; Nageswar, Y. V. D. Tetrahedron Lett. 2009, 50, 6025–6028.
- 10. Wan, J. P.; Gan, S. F.; Wu, J. M.; Pan, Y. J. Green Chem. 2009, 11, 1633-1637.
- Meshram, H. M.; Santosh Kumar, G.; Ramesh, P.; Chennakesava Reddy, B. Tetrahedron Lett. 2010, 51, 2580–2585.
- (a) Raw, S. A.; Wilfred, C. D.; Taylor, R. J. K. Chem. Commun. 2003, 2286–2287;
 (b) Raw, S. A.; Wilfred, C. D.; Taylor, R. J. K. Org. Biomol. Chem. 2004, 2, 788–796;
 (c) Robinson, R. S.; Taylor, R. J. K. Synlett 2005, 1003–1005;
 (d) Kim, S. Y.; Park, K. H.; Chung, Y. K. Chem. Commun. 2005, 1321–1323;
 (e) Cho, C. S.; Oh, S. G. J. Mol. Catal. A: Chem. 2007, 276, 205–210.
- 13. Venkateswara Rao, K. T.; Sai Prasad, P. S.; Lingaiah, N. J. Mol. Catal. A: Chem. 2009, 312, 65–69.
- Pan, X. Q.; Zou, J. P.; Huang, Z. H.; Zhang, W. Tetrahedron Lett. 2008, 49, 5302– 5308.
- 15. Cai, J. J.; Zou, J. P.; Pan, X. Q.; Zhang, W. Tetrahedron Lett. 2008, 49, 7386–7390.