

0040-4020(94)E0058-2

# Synthesis of 4-Aryl-2-benzazepine-1,5-diones by Photocyclization of N-(2-Arylethyl)phthalimides#

M. Rita Paleo, Domingo Domínguez\* and Luis Castedo\*

Departamento de Química Orgánica, Facultad de Química, Universidad de Santiago y Sección de Alcaloides del CSIC, 15706 Santiago de Compostela, SPAIN.

Abstract: The photocyclization of N-(2-arylethyl)phthalimides to 4-aryl-2-benzazepine-1,5-diones is described. We found that the presence of electron donating substituents on the aryl ring (as in 10a and b) is necessary for the cyclization process to occur. The procedure also allowed synthesis of 2-benzazepinediones with a carboxylate group at C<sub>3</sub> (11c and d) which were obtained as 1:1 mixture of diastereoisomers. The results of irradiating phathalimides 12, which bear an oxygenated substituent at the benzylic position, depended on the nature of the substituent. Attempts to photocyclize N-(indan-2-yl)phthalimides 16 and the electron-rich phthalimide 23 failed.

### Introduction

Our interest in the total synthesis of ribasine alkaloids  $(1)^1$  led us to investigate <sup>2</sup> the photocyclization of phthalimides 3 as a means of obtaining the 4-aryl-2-benzazepine skeleton (2) that it is latent in these alkaloids (Scheme 1).



#This work is part of the PhD Thesis by M. R.Paleo, University of Santiago, 1992

It is known that simple N-alkylphthalimides (4a) undergo photocyclization in low yield to 2benzazepinediones (5),  $\gamma$ -hydrogen abstraction to a biradical intermediate being followed by cyclization to a highly strained azetidinol which opens in a retrotransannular manner to afford the 2-benzazepinedione system (Scheme 2).<sup>3,4</sup> The corresponding reaction for N-(2-arylethyl)phthalimides (4b) remains largely unexplored. Until recently, the only reference in the literature has been a brief mention of the formation of a complex mixture of minor products upon irradiation of the simplest member of the class, N-(2-phenylethyl)phthalimide (4, R= Ph), <sup>5</sup> though while this manuscript was in preparation we became aware of work by Griesbeck and coworkers <sup>6</sup> on the photochemistry of N-phthaloyl derivatives of aromatic aminoacids which included examples closely related to our compounds 10c and d.



#### **Results and Discussion**

Phthalimides 10a and b were prepared by fusing a mixture of phthalic anhydride (6) and the corresponding commercially available 2-arylethylamine 7a or b at 150°C (Scheme 3). Photolysis of 10a in selected solvents was carried out at room temperature. With methylene chloride as solvent only a 14% yield of 11a was obtained after 6h of irradiation, while irradiation in acetonitrile for 20h did not produce any 2-benzazepinedione. Best results were obtained with acetone, which led to a 62% yield of 11a after 30 min, at which time some starting material was still present. Irradiation for longer periods of time was counterproductive the yield dropping to a 30% after 1.5h as a consequence of photochemical decomposition of the product. Under the same conditions the methylenedioxy-substituted derivative 10b gave, after brief irradiation, a 34% isolated yield of 11b (Scheme 3). These results contrast with the failure to cyclize of the unsubstituted phthalimide 10 ( $R^1=R^2=R^3=H$ )<sup>5</sup> showing the strong influence of the aryl substituents on the photocyclization. Formation of the 2-benzazepinedione is thus favoured by the presence of electron-donating substituents on the aryl ring, probably due to a lowering of its oxidation potential favouring electron transfer from the aryl moiety to the excited phthalimide, which is thought to be the fundamental step in the photochemistry of these systems.<sup>6</sup>

In order to apply this cyclization to the synthesis of ribasine alkaloids (1), we wished to prepare a 2benzazepinedione 11 with a 3-substituent  $\mathbb{R}^3$  suitable for construction of the indane ring (Scheme 3). To this end we studied the photocyclization of phthalimides 10c and d, which have a carboxylate group on the  $\beta$ -position.<sup>7</sup> Both compounds were prepared by fusing a mixture of phthalic anhydride (6) and the corresponding ethyl 2amino-3-aryl propanoate (7c or d), which was prepared by condensing the lithium enolate derived from ethyl benzylidene glycinate 8 with the corresponding benzyl bromide (9a or b), followed by hydrolysis of the imine with 5% HCl.<sup>8</sup>

Phthalimide 10c was irradiated in acetone for 85 min, giving approximate equal amounts of two 2benzazepinediones which were identified, after PTLC purification, as the *cis* ( $J_{H3-H4}=3$  Hz) and *trans* ( $J_{H3-H4}=10.4$  Hz)<sup>6</sup> diastereoisomers of 11c (42% combined yield). This result showed that the carboxylate did not interfere with cyclization.<sup>7,9</sup> A similar result was obtained with the methylenedioxy derivative 10d which led to 2-benzazepinedione 11d (again as a 1:1 mixture of diastereoisomers), although in lower yield.



Scheme	3
--------	---

Next we investigated the effect of including an oxygenated substituent at the benzylic position (where the hydrogen abstraction takes place), since 2-benzazepinediones oxygenated at C<sub>4</sub> were required to form the target ribasine alkaloids. The hydroxylated phthalimide **12a** was prepared from **10a** by benzylic bromination with NBS followed by solvolysis with a mixture of THF and H<sub>2</sub>O (Scheme 4). Irradiation of **12a** in acetone for 50 min gave a complex mixture from which the oxidized phthalimide **14** was isolated in low yield (Scheme 5). This result indicated the need to protect the hydroxyl group, so phthalimides **12b-d** were synthesized.

The acetate 12b was prepared by benzylic oxidation of 10a with DDQ in acetic acid<sup>10</sup> (90% yield). The  $\gamma$ methoxy derivative 12c was obtained similarly to 12a, by treating the intermediate bromide with THF/MeOH (85% yield). Protection of the hydroxyl group of 12a with ClTBDMS and imidazole in DMF gave 12d (93% yield).



Irradiation of 12b gave a mixture of products from which the deoxygenated benzazepinedione 11a was isolated, showing that an acetate group is unstable under the reaction conditions employed. However, irradiation of 12c and d for a short period (15 min) afforded the desired 2-benzazepinediones 13c and d in 27 and 30% yield respectively (Scheme 5).

Encouraged by these results, we attempted direct photosynthesis of benzazepinediones 17 (which already incorporate the indane ring of the ribasine alkaloids) from N-(indan-2-yl)phthalimides 16, which can be viewed



as N-(2-arylethyl)phthalimides that are substituted at positions  $\beta$  and  $\gamma$  (Scheme 6). The phthalimides 16 were obtained by heating a mixture of phthalic anhydride and the corresponding 2-aminoindane (15a-d) in DMF.



2-Aminoindane (15a) was obtained by catalytic hydrogenation of 2-indanone oxime (19)  $^{11}$  (Scheme 7). 2-Amino-5,6-(methylenedioxy)indane (15b) was prepared from 5,6-(methylenedioxy)-1-indanone (20b)  $^{12}$  by reaction with isoamyl nitrite to give 21b  $^{13}$  (90% yield) followed by NaBH4 reduction to 22 (98%) and hydrogenolysis (90%).  $^{14}$  Catalytic hydrogenation of 21c in 3/1 EtOH/H<sub>2</sub>O proceeded stereoselectively to give racemic *trans*-15c  $^{15}$  (88%), while under the same conditions 21b afforded the corresponding 2-amino-1indanone hydrochloride.  $^{16}$  By changing the solvent to glacial AcOH the reduction of 21b gives a diastereomeric mixture of 2-amino-1-indanol 15d in 81% yield.  $^{17}$ 

When irradiated in acetone solution, the unsubstituted phthalimide 16a behaved like the acyclic analogue 10 ( $R^1 = R^2 = R^3 = H$ ). Compound 16b, which bears electron-donating substituents that favoured the photoprocess in the acyclic phthalimides 10a-d, again gave a complex mixture, from which we were unable to isolate the desired indanobenzazepinedione 17b. When a hydroxyl group was introduced in the  $\gamma$  position (16c and 16d) the main product was the corresponding indanone, formed by oxidation of the alcohol. Protection as silyl ether 16e led to almost instantaneous fragmentation into a complex mixture of products upon irradiation. The non-cyclization of the *N*-(indan-2-yl)phthalimides 16 was apparently a consequence of the higly strained nature of the cyclic azetidinol intermediate, because of which the biradical generated upon hydrogen abstraction underwent fragmentation rather than cyclization. These findings led us to relinquish direct incorporation of the ribasine indane ring and go back to acyclic 2-arylethylamine precursors with a  $\beta$  substituent that would allow construction of the indane ring after photocyclization (as in 10c and d).



Finally, we investigated the possibility of photocyclization of precursors which already bore the methylendioxy group in the phthalimide portion. To this end we prepared the phthalimide 23 (Scheme 8) and studied its photochemical behaviour. To our surprise this compound did not photocyclize, apparently due to inhibition of the  $\gamma$ -abstraction process by the electron-donating substituents on the phthalimide moiety. For this reason and given the complexity of introducing the 1,3-dioxolane ring at a later stage, we abandoned any further attempt to generate ribasine alkaloids 1 by this photochemical approach.



## Conclusion

We have developed a one step synthesis of 4-aryl-2-benzazepine-1,5-diones by photocyclization of N-(2arylethyl)phthalimides with electron-donating substituents on the aryl ring. The procedure allows the preparation of 4-aryl-2-benzazepinediones with further substituents at positions 3 or 4 of the 2-benzazepine ring. Unfortunately, when the method was applied to N-(indan-2-yl)phthalimides none of the desired indane-2benzazepinediones were isolated. Another limitation is that the photocyclization cannot be applied to compounds bearing electron-donating substituents on the phthalimide moiety.

### **EXPERIMENTAL SECTION**

General Methods. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> at 250 and 62.83 MHz respectively, using TMS as internal reference. Mass spectra were recorded at an ionization voltage of 70 eV. Melting points are uncorrected. All air-sensitive reactions were carried out under dried, deoxygenated Ar using flame-dried glassware and magnetic stirring; reagents were added by syringe through septa. All solvents for air-or moisture-sensitive reactions were dried by standard procedures.<sup>18</sup> The concentration of commercial solutions of *n*-BuLi in hexane (Aldrich) was determined immediately prior to use by titration with diphenylacetic acid.<sup>19</sup>

Preparative irradiation was conducted in a Pyrex immersion well at room temperature using a 450 W watercooled medium-pressure mercury lamp (Hanovia), with magnetic stirring, under Ar atmosphere

Treatment of ethyl glycinate hydrochloride with benzaldehyde in methylene chloride in the presence of triethylamine and anhydrous MgSO4 afforded the Schiff base methyl ester derivative 8.8a

### Preparation of ethyl 2-amino-3-arylpropanoates 7c and 7d

In a flame-dried 25-mL round-bottomed flask equipped with a stirring bar, septum cap and Ar inlet, an LDA solution was prepared by treating diisopropylamine (0.4 mL, 2.9 mmol) in anhydrous THF (10 mL) with 1.6 M *n*-BuLi in hexane (1.8 mL, 2.9 mmol) at -78°C. The cooling bath was removed and the mixture was stirred for 15 min before cooling again to -78°C and addition of the Schiff base 8 (0.5 g, 2.7 mmol) in THF. The yellow solution obtained was stirred for 15 min, a solution of the bromide 9a or 9b (2.7 mmol) in anhydrous THF (5 mL) was added and the resulting mixture was stirred at -78°C for 1 h and then slowly brought to rt. An ice-cold saturated aqueous ammonium chloride/ether mixture (10 mL each) was added, the organic layer was removed and the aqueous layer was further extracted with ether (3x5 mL). The organic extract was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated to leave an oil which was partially hydrolysed to the  $\alpha$ -amino ester by treatment with 5% HCl (5 mL) for 2 h at rt. Ether was added and the aqueous layer was separated, cooled in an ice-water bath and treated with solid K<sub>2</sub>CO<sub>3</sub> until basic and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x10 mL). The combined organic extracts were dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporation of the solvent gave an oil which was used in the next step without further purification.

## General procedure for preparation of N-arylethylphthalimides 10.

In a 25 mL flask a mixture of phthalic anhydride (6) (3 g, 20.27 mmol) and the corresponding 2arylethylamine (7a-d) (20.27 mmol) was fused at 150°C for 90 min and then allowed to cool to rt. The product was purified by recrystallization.

## N-[2-(3,4-Dimethoxyphenyl)ethyl]phthalimide (10a):

5.43 g (86%); m. p. : 168-169°C (EtOH). IR (CHCl3): 1720, 1755 cm-1. UV (MeOH)  $\lambda_{max}$ : 220, 280 nm. <sup>1</sup>H NMR:  $\delta$  2.95 (m, 2H), 3.82 (s, 3H), 3.84 (s, 3H), 3.91 (m, 2H), 6.75 (s, 1H), 6.79 (s, 2H), 7.71 (m, 2H), 7.83 (m, 2H). <sup>13</sup>C NMR:  $\delta$  33.97, 39.24, 55.77, 55.82, 111.47, 112.13, 120.91, 123.17, 130.57, 132.13, 133.88, 147.86, 149.02, 168.20. MS m/z (%): 311 (M<sup>+</sup>, 14), 164 (84), 151 (100), 107 (33), 77 (62). Anal. Calc. for C18H17NO4: C, 69.44; H, 5.50; N, 4.50. Found: C, 69.03; H, 5.71; N, 4.52.

## N-[2-(3,4-Methylenedioxyphenyl)ethyl]phthalimide (10b)

5.09 g (90%); m.p.: 136-137°C (EtOH). IR (CHCl3): 1710, 1770 cm-1.UV (MeOH)  $\lambda_{max}$ : 218, 288 nm. <sup>1</sup>H NMR:  $\delta$  2.90 (t, 2H, J= 7.6 Hz), 3.87 (t, 2H, J= 7.6 Hz), 5.92 (s, 2H), 6.70 (m, 3H), 7.71 (m, 2H), 7.83 (m, 2H). <sup>13</sup>C NMR:  $\delta$  34.26, 39.41, 100.86, 108.31, 109.27, 121.81, 123.24, 131.81, 132.16, 133.91, 146.35, 147.81, 168.20. MS m/z (%): 295 (M<sup>+</sup>, 7), 160 (23), 148 (64), 135 (48), 105 (22), 77 (100). Anal. Calc. for C17H13NO4: C, 69.15; H, 4.44; N, 4.74. Found: C, 68.81; H, 4.24; N, 4.65.

## N-[1-(Ethoxycarbonyl)-2-(3,4-dimethoxyphenyl)ethyl]phthalimide (10c).

3.97 g (63%); foam. IR (CHCl3): 1710, 1740, 1775 cm<sup>-1</sup>. UV (CH3OH)  $\lambda_{max}$ : 220, 282 nm. <sup>1</sup>H NMR:  $\delta$  1.25 (t, 3H, J= 7.1 Hz), 3.50 (m, 2H), 3.68 (s, 3H), 3.75 (s, 3H), 4.23 (q, 2H, J= 7.1 Hz), 5.12 (dd, 1H, J= 6.2, 10.5 Hz), 6.65 (m, 3H), 7.67 (m, 2H), 7.74 (m, 2H). <sup>13</sup>C NMR:  $\delta$  14.00, 34.11, 53.38, 55.65, 55.73, 61.87, 111.46, 112.08, 121.08, 123.38, 129.33, 131.74, 134.06, 147.94, 148.94, 167.50, 168.86. Anal. Calc. for C21H21NO6: C, 65.79; H, 5.52; N, 3.65. Found: C, 65.41; H, 5.29; N, 3.79.

## N-[1-(Ethoxycarbonyl)-2-(3,4-methylenedioxyphenyl)ethyl]phthalimide (10d).

3.31 g (52%); m.p.: 74-75°C (MeOH). IR (CHCl3): 1720, 1740, 1780 cm-1. UV (CH3OH)  $\lambda_{max}$ : 218, 288 nm. <sup>1</sup>H NMR:  $\delta$  1.25 (t, 3H, J= 7.1 Hz), 3.47 (m, 2H), 4.25 (q, 2H, J= 7.1 Hz), 5.06 (dd, 1H, J= 6.0, 10.4 Hz), 5.85 (d, 1H, J= 1.3 Hz), 5.86 (d, 1H, J= 1.3 Hz), 6.59 (s, 2H), 6.67 (s, 1H), 7.70 (m, 2H), 7.79 (m, 2H). <sup>13</sup>C NMR:  $\delta$  13.99, 34.30, 53.60, 61.92, 100.82, 108.25, 109.19, 121.96, 123.47, 130.56, 131.70, 134.07, 146.43, 147.73, 167.54, 168.77. MS m/z (%): 367 (M<sup>+</sup>, 6), 294 (4), 220 (96), 175 (54), 160 (22), 135 (100), 104 (73), 77 (66). Anal. Calc. for C<sub>20</sub>H<sub>17</sub>NO<sub>6</sub>: C, 65.39; H, 4.66; N, 3.81. Found: C, 65.38; H, 4.88; N, 3.99.

## N-[2-(Hydroxy)-2-(3,4-dimethoxyphenyl)ethyl] phthalimide (12a).

A suspension of **10a** (2.0 g, 6.43 mmol) and NBS (1.14 g, 6.43 mmol) in anhydrous CCl4 (60 mL) was heated at reflux and irradiated with a 60 W lamp until the NBS has been transformed into colourless succinimide suspended in the reaction mixture. The suspension was cooled, filtered and the residue was washed with CCl4. The combined filtrate and washings were evaporated to dryness and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and washed with brine. The organic layer was separated, dried over anh. Na<sub>2</sub>SO<sub>4</sub> and evaporated *in vacuo* to give a residue which was dissolved in THF/H<sub>2</sub>O (12mL/8mL) and stirred overnight at rt. CH<sub>2</sub>Cl<sub>2</sub> was added and the organic layer separated, dried and evaporated to dryness. The product was purified by column chromatography on silica gel, with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (1%) as eluant, giving 1.5 g (71% yield) of a white solid; m.p.: 173-174°C (MeOH). IR (CHCl<sub>3</sub>): 1710, 1770, 3460 cm-1. UV (MeOH)  $\lambda_{max}$ : 220, 280 nm. <sup>1</sup>H NMR:  $\delta$  3.87 (s, 3H), 3.88 (s, 3H), 3.93 (dd, 1H, J= 3.8, 14.2 Hz), 4.03 (dd, 1H, J= 8.4, 14.2 Hz), 5.04 (dd, 1H, J= 3.8, 8.4 Hz), 6.85 (d, 1H, J= 8.7 Hz), 6.98 (m, 2H), 7.73 (m, 2H), 7.85 (m, 2H). <sup>13</sup>C NMR:  $\delta$  45.62, 55.90, 72.22, 109.18, 111.32, 118.31, 123.44, 131.97, 133.80, 134.12, 148.95; 149.28, 168.76. MS m/z (%): 327 (M<sup>+</sup>, 12), 180 (24), 167 (100), 160 (62), 139 (92), 124 (63), 104 (56), 77 (78). Anal. Calc. for C1<sub>8</sub>H<sub>17</sub>NO<sub>5</sub>: C, 66.05; H, 5.23; N, 4.28. Found: C, 66.16; H, 5.26; N, 4.15.

## N-[2-(Acetoxy)-2-(3,4-dimethoxyphenyl)ethyl] phthalimide (12b).

A solution of 1.0 mmol of DDQ (230 mg) in 5 mL of anhydrous acetic acid was slowly added at 25°C to a solution of **10a** (300 mg, 0.97 mmol) in 10 mL of anhydrous acetic acid and heated at 70°C overnight. 20 mL of CH<sub>2</sub>Cl<sub>2</sub> were added and the organic layer was washed with water (3x10 mL), 10% aqueous NaOH (10 mL) and water (10 mL).The extract was dried (Na<sub>2</sub>SO<sub>4</sub>) and cocentrated to give an oil which solidified on cooling. Recrystallization from a CH<sub>2</sub>Cl<sub>2</sub>/Hexane solvent pair afforded 307 mg (90%), m.p.: 139-141°C. IR (CHCl<sub>3</sub>): 1720, 1730, 1770 cm<sup>-1</sup>. UV (MeOH)  $\lambda_{max}$ : 220, 280 nm. <sup>1</sup>H NMR:  $\delta$  2.02 (s, 3H), 3.87 (s, 3H), 3.89 (s, 3H), 3.93 (m, 1H), 4.15 (dd, 1H, J= 9.0, 14.1 Hz), 6.09 (dd, 1H, J= 4.0, 9.0 Hz), 6.85 (d, 1H, J= 8.2 Hz), 7.01 (m, 2H), 7.73 (m, 2H), 7.84 (m, 2H). <sup>13</sup>C NMR:  $\delta$  20.77, 42.55, 55.74, 55.82, 72.88, 109.96, 111.22, 119.31, 123.23, 129.57, 131.81, 133.95, 149.11, 149.30, 167.80, 170.05. MS m/z (%) 369 (M<sup>+</sup>, 4), 209 (13), 167 (100), 160 (48), 139 (35), 77 (40). Anal. Calc. for C<sub>2</sub>0H<sub>1</sub>9NO<sub>6</sub>: C, 65.03; H, 5.18; N, 3.79. Found: C, 65.10; H, 5.38; N, 3.73.

## N-[2-(Methoxy)-2-(3,4-dimethoxyphenyl)ethyl] phthalimide (12c).

The same procedure as for 12a, changing the treatment of the intermediate bromide with THF/H<sub>2</sub>O for THF/MeOH afforded 12c in 85% yield; m.p.: 122-123°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1715, 1775 cm-1. UV (MeOH)  $\lambda_{max}$ : 220, 280 nm. <sup>1</sup>H NMR:  $\delta$  3.11 (s, 3H), 3.68 (dd, 1H, J= 4.8, 13.8 Hz), 3.81 (s, 3H), 3.82 (s, 3H), 3.96 (dd, 1H, J= 8.9, 13.8 Hz), 4.48 (dd, 1H, J= 4.8, 8.9 Hz), 6.78 (d, 1H, J= 8.6 Hz), 6.86 (m, 2H), 7.65 (m, 2H), 7.77 (m, 2H). <sup>13</sup>C NMR:  $\delta$  43.87, 55.84, 55.93, 56.78, 80.40, 109.56, 111.08, 119.77, 123.30, 131.19, 132.12, 133.93, 149.11, 149.39, 168.26. MS m/z (%) 341 (M<sup>+</sup>, 3), 181 (100), 166 (73), 160 (41), 151 (54), 77 (71). Anal. Calc. for C19H19NO5: C, 66.85; H, 5.61; N, 4.10. Found: C, 66.76; H, 5.79; N, 4.11.

## N-[2-(t-Butyldimethylsilyloxy)-2-(3,4-dimethoxyphenyl)ethyl] phthalimide (12d).

In a flame-dried 25-mL round-bottomed flask equipped with a stirring bar, septum cap and Ar inlet, a solution of **12a** (0.5 g, 1.5 mmol) in anhydrous DMF (10 mL), CITBDMS (277 mg, 1.8 mmol) and imidazole (260 mg, 3.8 mmol) were stirred under Ar for 24 h at 25°C, then poured over an ice/water mixture and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was separated, dried (Na<sub>2</sub>SO<sub>4</sub>) and the solvent evaporated. Crystallization from a CH<sub>2</sub>Cl<sub>2</sub>/Hexane solvent pair afforded 628 mg (93%), m.p.: 138-140°C. IR (CHCl<sub>3</sub>): 1710, 1770 cm-1. UV (MeOH)  $\lambda_{max}$ : 218, 226, 280 nm. <sup>1</sup>H NMR:  $\delta$  -0.22 (s, 3H), -0.20 (s, 3H), 0.75 (s, 9H), 3.68 (dd, 1H, J= 4.2, 13.6 Hz), 3.88 (s, 3H), 3.89 (s, 3H), 3.94 (dd, 1H, J= 9.2, 13.6 Hz), 5.03 (dd, 1H, J= 4.2, 9.2 Hz), 6.82 (d, 1H, J= 8.2 Hz), 6.96 (m, 2H), 7.73 (m, 2H), 7.86 (m, 2H). <sup>13</sup>C NMR:  $\delta$  -5.43, -4.99, 17.77, 25.48, 46.24, 55.86, 71.82, 109.40, 110.99, 118.64, 123.19, 132.23, 133.94, 134.69, 148.72, 149.12, 168.29. MS m/z (%) 441 (M<sup>+</sup>, 0.3), 384 (9), 281 (100), 204 (20), 160 (14), 77 (20), 73 (67). Anal. Calc. for C<sub>24H31</sub>NO<sub>5</sub>Si: C, 65.28; H, 7.07; N, 3.17. Found: C, 65.01; H, 6.84; N, 2.93.

## General procedure for irradiation of phthalimides 10 and 12.

A solution of the phthalimide (0.2 g) in acetone (200 mL) was irradiated, under Ar atmosphere, by using a Hanovia 450 W medium-pressure Hg lamp in a pyrex inmersion well mantained at room temperature. The solution was degassed by bubbling Ar through it for a period of 10 min prior to irradiation. Solvent was removed *in vacuo* and the residue chromatographied by preparative TLC on silica gel with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (3%) as eluant.

## 2,3,4,5-Tetrahydro-4-(3,4-dimethoxyphenyl)-1H-2-benzazepine-1,5-dione (11a).

Irradiation period: 30 min. Yield: 62%, m.p.: 166-167°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1670, 1690 cm-1. UV (MeOH)  $\lambda_{max}$ : 208, 280 nm. <sup>1</sup>H NMR:  $\delta$  3.68 (m, 1H), 3.82 (m, 1H), 3.83 (s, 3H), 3.84 (s, 3H), 4.12 (dd, 1H, J= 4.1, 10.2 Hz), 6.79 (m, 3H), 7.56 (m, 1H), 7.6-7.8 (m, 2H), 7.94 (m, 2H); (CDCl<sub>3</sub>-D<sub>2</sub>O):  $\delta$  3.66 (dd, 1H, J= 4.1, 14.9 Hz), 3.82 (dd, 1H, J= 14.9, 10.2 Hz), 3.83 (s, 3H), 3.84 (s, 3H), 4.14 (dd, 1H, J= 4.1, 10.2 Hz), 6.79 (m, 3H), 7.56 (m, 1H), 7.6-7.8 (m, 2H), 7.94 (m, 1H). <sup>13</sup>C NMR  $\delta$  44.25, 55.94, 61.32, 111.78, 120.08, 128.53, 129.79, 130.01, 131.52, 132.33, 132.60, 137.30, 148.95, 149.51, 170.86, 204.32. Anal. Calc. for C1<sub>8</sub>H<sub>1</sub>7NO4: C, 69.44; H, 5.50; N, 4.50. Found: C, 69.42; H, 5.86; N, 4.60.

## 2,3,4,5-Tetrahydro-4-(3,4-Methylenedioxyphenyl)-1H-2-benzazepine-1,5-dione (11b).

Irradiation period: 33 min. Yield: 34%, m.p. 153-154°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1660, 1690, 3020 cm-1. UV (MeOH)  $\lambda_{max}$ : 208, 238, 286 nm. <sup>1</sup>H NMR:  $\delta$  3.64 (m, 1H), 3.78 (m, 1H), 4.07 (dd, 1H, J=

4.1, 10.0 Hz), 5.93 (s, 2H), 6.61 (m, 2H), 6.75 (d, 1H, J= 8.4 Hz), 7.56 (d, 1H, J= 7.6 Hz), 7.67 (m, 2H), 7.95 (d, 1H, J= 7.6 Hz); (CDCl<sub>3</sub>-D<sub>2</sub>O):  $\delta$ 3.62 (dd, 1H, J= 4.1, 14.9 Hz), 3.78 (dd, 1H, J= 10.0, 14.9 Hz), 4.07 (dd, 1H, J= 4.1, 10.0 Hz), 5.93 (s, 2H), 6.61 (m, 2H), 6.75 (d, 1H, J= 8.4 Hz), 7.56 (d, 1H, J= 7.6 Hz), 7.67 (m, 2H), 7.95 (d, 1H, J= 7.6 Hz). <sup>13</sup>C NMR:  $\delta$  44.13, 61.37, 101.24, 108.41, 108.81, 121.47, 128.61, 129.95, 130.97, 131.45, 132.38, 132.64, 137.15, 147.36, 148.21, 170.90, 204.23. Anal. Calc. for C<sub>17H13</sub>NO4: C, 69.15; H, 4.44; N, 4.74. Found: C, 68.81; H, 4.37; N, 4.59.

## 2,3,4,5-Tetrahydro-3-ethoxycarbonyl-4-(3,4-dimethoxyphenyl)-1*H*-2-benzazepine-1,5-dione (11c).

Irradiation time: 85 min. Yield: 42% (1/1 mixture of cis/trans isomers).

*cis*-11c: less polar product, m.p.: 68-70°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1670, 1740 cm-1. UV (MeOH)  $\lambda_{max}$ : 214, 236, 282 nm. <sup>1</sup>H NMR:  $\delta$  1.25 (t, 3H, J= 7.3 Hz), 3.81 (s, 3H), 3.86 (s, 3H), 4.16 (m, 2H), 4.40 (d, 1H, J= 3.0 Hz), 5.05 (dd, 1H, J= 5.6, 3.0 Hz), 6.6-6.9 (m, 4H), 7.6-7.8 (m, 3H), 7.9-8.1 (m, 1H). <sup>13</sup>C NMR:  $\delta$  14.05, 55.85, 55.94, 56.03, 62.59, 63.00, 111.50, 113.00, 122.52, 126.71, 129.00, 130.24, 132.45, 133.13, 136.55, 149.19, 149.35, 167.83, 201.81. Anal. Calc. for C<sub>21</sub>H<sub>21</sub>NO<sub>6</sub>: C, 65.79; H, 5.52; N, 3.65. Found: C, 65.41; H, 5.82; N, 3.45.

*trans*-11c: more polar product, m.p.: 59-61°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1670, 1740 cm-1. UV (MeOH)  $\lambda_{max}$ : 214, 236, 282 nm. <sup>1</sup>H NMR:  $\delta$  0.89 (t, 3H, J= 7.1 Hz), 3.81 (s, 3H), 3.86 (s, 3H), 3.94 (q, 2H, J= 7.1 Hz), 4.27 (d, 1H, J= 10.4 Hz), 4.68 (dd, 1H, J= 5.9, 10.4 Hz), 6.6-6.8 (m, 4H), 7.4-8.0 (m, 4H). <sup>13</sup>C NMR:  $\delta$  13.58, 55.97, 57.47, 62.19, 65.10, 77.19, 111.75, 120.63, 123.65, 128.63, 129.89, 131.56, 132.56, 134.34, 137.23, 149.49, 168.72, 169.06, 201.80. Anal. Calcd. for C<sub>21H21NO6</sub>: C, 65.79; H, 5.52; N, 3.65. Found: C, 65.75; H, 5.64; N, 3.83.

## 2,3,4,5-Tetrahydro-3-ethoxycarbonyl-4-(3,4-methylenedioxyphenyl)-1*H*-2-benzazepine-1,5-dione (11d).

Irradiation time: 65 min. Yield: 25% (1/1 mixture of cis/trans isomers).

*cis*-11d: less polar product, m.p.: 74-75°C (CH<sub>2</sub>Cl<sub>2</sub>/Hexane). IR (CHCl<sub>3</sub>): 1670, 1740 cm-1. UV (MeOH)  $\lambda_{max}$ : 212, 236, 288 nm. <sup>1</sup>H NMR:  $\delta$  1.26 (t, 3H, J= 7.1 Hz), 4.17 (m, 2H), 4.37 (d, 1H, J= 3.1 Hz), 5.05 (dd, 1H, J= 5.5, 3.1 Hz), 5.96 (s, 2H), 6.61 (m, 2H), 6.78 (m, 2H), 7.6-7.8 (m, 3H), 7.99 (d, 1H, J= 6.3 Hz). <sup>13</sup>C NMR:  $\delta$  14.04, 56.00, 62.65, 63.08, 101.31, 108.57, 110.07, 123.62, 127.72, 128.95, 130.24, 132.02, 132.47, 133.07, 136.63, 147.88, 148.07, 167.80, 201.68. Anal. Calc. for C<sub>20</sub>H<sub>17</sub>NO<sub>6</sub>: C, 65.39; H, 4.66; N, 3.81. Found: C, 65.18; H, 4.91; N, 3.92.

*trans*-11d: more polar product, m.p.: 70-71°C (CH<sub>2</sub>Cl<sub>2</sub>/Hexane). IR (CHCl<sub>3</sub>): 1670, 1740 cm-1. UV (MeOH)  $\lambda_{max}$ : 212, 236, 288 nm. <sup>1</sup>H NMR:  $\delta$  0.93 (t, 3H, J= 7.1 Hz), 3.96 (m, 2H), 4.22 (d, 1H, J= 10.5 Hz), 4.64 (dd, 1H, J= 5.9, 10.5 Hz), 5.95 (s, 2H), 6.58 (m, 2H), 6.75 (m, 2H), 7.49 (m, 1H), 7.69 (m, 2H), 7.95 (m, 1H). <sup>13</sup>C NMR:  $\delta$  13.56, 57.48, 62.21, 65.13, 101.32, 108.41, 108.79, 122.04, 128.73, 129.71, 129.86, 131.48, 132.65, 147.70, 148.22, 168.64, 168.91, 201.93. Anal. Calc. for C<sub>20</sub>H<sub>17</sub>NO<sub>6</sub>: C, 65.39; H, 4.66; N, 3.81. Found: C, 65.26; H, 4.85; N, 3.89.

### N-(3,4-Dimethoxyacetophenyl)phthalimide (14)

Irradiation period: 50 min. Yield: 30%, m.p.: 194-195°C (CH<sub>2</sub>Cl<sub>2</sub>/Hexane). IR (CHCl<sub>3</sub>): 1690, 1720, 1775 cm<sup>-1</sup>. UV (MeOH)  $\lambda_{max}$ : 220, 278, 302 nm. <sup>1</sup>H NMR:  $\delta$  3.92 (s, 3H), 3.97 (s, 3H), 5.10 (s, 2H), 6.94 (d, 1H, J= 8.2 Hz), 7.52 (s, 1H), 7.65 (d, 1H, J= 8.2 Hz), 7.76 (m, 2H), 7.88 (m, 2H). <sup>13</sup>C NMR:  $\delta$  43.84,

56.03, 56.13, 110.38, 110.48, 122.78, 123.57, 127.81, 132.41, 134.13, 149.48, 154.23, 168.05, 189.63. Anal. Calc. for C18H15NO5: C, 66.46; H, 4.65; N, 4.31. Found: C, 66.80; H, 4.77; N, 4.51.

## 2,3,4,5-Tetrahydro-4-methoxy-4-(3,4-dimethoxyphenyl)-1*H*-2-benzazepine-1,5-dione (13c).

Irradiation period: 15 min. Yield: 27%; m.p.: 148-150°C (CH<sub>2</sub>Cl<sub>2</sub>/Hexane). IR (CHCl<sub>3</sub>): 1660, 1670 cm-1. UV (MeOH)  $\lambda_{max}$ : 216, 232, 278 nm. <sup>1</sup>H NMR:  $\delta$  3.33 (s, 3H), 3.70 (dd, 1H, J= 14.9, 6.5 Hz), 3.83 (dd, 1H, J= 14.9, 6.5 Hz), 3.88 (s, 6H), 6.39 (broad, 1H, NH), 6.80 (m, 2H), 6.90 (d, 1H, J= 1.9 Hz), 7.53 (m, 1H), 7.68 (m, 2H), 7.90 (m, 1H). <sup>13</sup>C NMR:  $\delta$  49.28, 53.15, 55.93, 56.08, 87.55, 110.57, 111.15, 120.65, 128.62, 129.62, 130.03, 132.18, 132.32, 132.56, 137.47, 149.51, 149.64, 170.27, 201.58. Anal. Calc. for C19H19NO5: C, 66.85; H, 5.61; N, 4.10. Found: C, 66.97; H, 5.27; N, 3.77.

## 2,3,4,5-Tetrahydro-4-*t*-butyldimethylsilyloxy-4-(3,4-dimethoxyphenyl)-1*H*-2benzazepine-1,5-dione (13d).

Irradiation period: 15 min. Yield: 30%; m.p.:  $153^{\circ}$ C (Ether/Hexane). IR (CHCl3): 1660, 1670 cm<sup>-1</sup>. UV (MeOH)  $\lambda_{max}$ : 220, 280 nm. <sup>1</sup>H NMR:  $\delta$  0.03 (s, 3H), 0.21 (s, 3H), 1.00 (s, 9H), 3.46 (dd, 1H, J= 6.9, 15.0 Hz), 3.84 (s, 6H), 3.91 (dd, 1H, J= 5.9, 15.0 Hz), 6.57 (broad, 1H, NH), 6.64-6.77 (m, 2H), 7.03 (m, 1H), 7.5-7.98 (m, 4H). <sup>13</sup>C NMR:  $\delta$  -3.22, -3.16, 18.93, 26.03, 51.89, 55.85, 55.90, 86.71, 109.22, 110.96, 117.93, 128.78, 130.10, 131.26, 132.21, 132.67, 133.15, 137.63, 149.13, 149.23, 170.24, 204.68; Anal. Calc. for C24H31NO5Si: C, 65.28; H, 7.07; N, 3.17. Found: C, 65.42; H, 7.15; N, 3.27.

## General procedure for preparation of 16a-d.

In a 25-mL flame-dried round-bottomed flask a suspension of NaH (oil dispersion 80%, 90 mg, 3.0 mmol, washed with THF 2 x 1 mL) in anhydrous DMF was prepared and cooled to 0°C. A solution of the corresponding hydrochloride 15 (2.7 mmol) in DMF (3 mL) was added, stirred for 15 min and allowed to warm to rt. Phthalic anhydride (400 mg, 2.7 mmol) was added and the resulting suspension was heated at 120°C for between 5 and 12h depending on the aminoindane (TLC analysis). After cooling to rt a 1/1 H2O/CH2Cl2 mixture (20 mL) was added. The organic phase was washed with water (5 x 5 mL), dried over Na2SO4, and evaporated to dryness. The residue was recrystallyzed from MeOH.

## N-(Indan-2-yl)phthalimide (16a).

Yield: 78%; m.p.: 194-195°C (Cl<sub>2</sub>CH<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1715, 1775 cm-1. UV (MeOH)  $\lambda_{max}$ : 220, 242, 290 nm. <sup>1</sup>H NMR:  $\delta$  3.18 (dd, 2H, J= 8.8, 15.3 Hz), 3.62 (dd, 2H, J= 9.3, 15.3 Hz), 5.15 (m, 1H), 7.20 (m, 4H), 7.73 (m, 2H), 7.84 (m, 2H). <sup>13</sup>C RMN:  $\delta$  36.08, 50.08, 123.23, 124.44, 126.74, 132.19, 133.99, 140.96, 168.41. MS m/z 263 (M<sup>+</sup>, 13), 148 (73), 130 (70), 116 (100), 104 (68), 89 (52), 77 (76), 76 (87). Anal. Calcd. for C<sub>17</sub>H<sub>13</sub>NO<sub>2</sub>: C, 77.55; H, 4.98; N, 5.32; Found: C, 77.35; H, 5.16; N, 5.40.

## N-[5,6-(Methylenedioxy)indan-2-yl]phthalimide (16b).

Yield: 92%; m.p.: 195-196°C (EtOH). IR (CHCl<sub>3</sub>): 1710, 1770 cm<sup>-1</sup>. UV (CH<sub>3</sub>OH)  $\lambda_{max}$ : 220, 296 nm; <sup>1</sup>H RMN  $\delta$  3.07 (dd, 2H, J= 9.0, 14.6 Hz), 3.50 (dd, 2H, J= 9.0, 14.6 Hz), 5.15 (q, 1H, J= 9.0 Hz), 5.92 (d, 1H, J= 1.4 Hz), 5.93 (d, 1H, J= 1,4 Hz), 6.69 (s, 2H), 7.72 (m, 2H), 7.83 (m, 2H); <sup>13</sup>C RMN  $\delta$  35.98, 50.34, 100.85, 105.06, 123.23, 132.15, 133.40, 133.99, 146.80, 168.37; m/z 307 (M<sup>+</sup>, 4), 160 (100), 130 (62), 102 (97), 76 (67); Anal. Calcd. for C18H13NO4: C, 70.35; H, 4.26; N, 4.56. Found: C, 70.05; H, 4.31; N, 4.60.

## trans-N-(1-Hydroxyindan-2-yl)phthalimide(16c).

Yield: 70%; m.p.: 217-219°C (MeOH). IR (CHCl3): 1710, 1770 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{max}$ : 210, 220, 292 nm. <sup>1</sup>H NMR:  $\delta$  3.18 (dd, 1H, J=8.6, 15.1 Hz), 3.53 (dd, 1H, J=10.0, 15.1 Hz), 4.76 (m, 1H), 5.85 (d, 1H, J=7.4 Hz), 7.27 (m, 3H), 7.43 (m, 1H), 7.75 (m, 2H), 7.86 (m, 2H). <sup>13</sup>C NMR:  $\delta$  32.71, 60.42, 77.07, 123.39, 123.95, 124.82, 127.39, 128.54, 132.10, 134.15, 138.80, 142.45, 168.70. Anal. Calcd. for C<sub>17</sub>H<sub>13</sub>NO<sub>3</sub>: C, 73.11; H, 4.69; N, 5.01. Found: C; 72.80; H, 4.65; N, 4.82.

## N-[1-Hydroxy-5,6-(methylenedioxy)indan-2-yl]phthalimide (16d).

cis-16d: less polar product. Yield: 27%, m.p.: 203-204°C (EtOH). IR (CHCl<sub>3</sub>): 1710, 1770, 3400-3520 cm-1; UV (MeOH)  $\lambda_{max}$ : 218, 296 nm. <sup>1</sup>H NMR:  $\delta$  3.05 (dd, 1H, J= 8.8, 15.8 Hz), 3.91 (dd, 1H, J= 8.0, 15.8 Hz), 5.09 (m, 2H), 5.97 (d, 1H, J= 1.3 Hz), 5.98 (d, 1H, J= 1.3 Hz), 6.72 (s, 1H), 6.93 (s, 1H), 7.74 (m, 2H), 7.85 (m, 2H). <sup>13</sup>C NMR:  $\delta$  32.47, 53.23, 76.17, 101.28, 104.98, 105.64, 123.46, 132.02, 133.66, 134.20, 135.54, 147.44, 149.13, 169.59.

*trans*-16d: more polar product. Yield: 51%, m.p.: 203-204°C (EtOH). IR (CHCl3): 1710, 1770, 3400-3520 cm-1. UV (MeOH)  $\lambda_{max}$ : 220, 296 nm. <sup>1</sup>H NMR:  $\delta$  3.08 (dd, 1H, J= 8.7, 14.9 Hz), 3.38 (dd, 1H, J= 9.7, 14.9 Hz), 4.73 (m, 1H), 5.69 (d, 1H, J= 7.6 Hz), 5.95 (d, 1H, J= 1.1 Hz), 5.97 (d, 1H, J= 1.1 Hz), 6.67 (s, 1H), 6.87 (s, 1H), 7.73 (m, 2H), 7.84 (m, 2H). <sup>13</sup>C NMR:  $\delta$  32.66, 60.54, 77.16, 101.13, 104.52, 105.17, 123.39, 131.99, 132.07, 134.15, 135.45, 147.43, 148.34, 168.64.

## trans-N-[1-t-Butyldimethylsilyloxy-5,6-(methylenedioxy)indan-2-yl]phthalimide (16e).

Obtained from *trans*-16d by a procedure analogous to the preparation of 12d. Yield: 95%, m.p.: 67-69°C (CH<sub>2</sub>Cl<sub>2</sub>-Hexane). IR (CHCl<sub>3</sub>): 1710, 1770 cm-1. UV (MeOH)  $\lambda_{max}$ : 210, 222, 296 nm. <sup>1</sup>H NMR:  $\delta$  -0.19 (s, 3H), 0.08 (s, 3H), 0.85 (s, 9H), 2.99 (dd, 1H, J= 8.8, 14.7 Hz), 3.35 (dd, 1H, J= 9.8, 14.7 Hz), 4.84 (m, 1H), 5.79 (d, 1H, J= 7.5 Hz), 5.93 (d, 1H, J= 0.4 Hz), 5.94 (d, 1H, J= 0.4 Hz), 6.64 (s, 1H), 6.73 (s, 1H), 7.75 (m, 2H), 7.86 (m, 2H). <sup>13</sup>C NMR:  $\delta$  -4.75, -4.57, 17.74, 25.59, 32.39, 59.87, 76.59, 101.01, 104.57, 105.13, 123.34, 131.53, 131.99, 134.15, 136.36, 147.21, 147.91, 168.47. Anal. Calcd.for C<sub>24H27</sub>NO5Si: C, 65.88; H, 6.22; N, 3.20. Found: C, 65.82; H, 6.17; N, 3.16.

## N-[2-(3,4-dimethoxyphenyl)ethyl]-4,5-(methylenedioxy)phthalimide (23).

In a 25 mL flask 4,5-methylenedioxyphthalic anhydride (0.5 g, 2.6 mmol) and 3,4-dimethoxyphenethylamine (7a) (0.47 g, 2.6 mmol) were fused at 160°C for 80 min and then allowed to cool to rt. The product was purified by recrystallization from EtOH (85% yield), m.p.: 183°C. IR (CHCl<sub>3</sub>): 1700, 1760 cm<sup>-1</sup>. <sup>1</sup>H RMN  $\delta$  2.92 (m, 2H), 3.84 (s, 3H), 3.85 (s, 3H), 3.88 (m, 2H), 6.23 (s, 2H), 6.77 (m, 3H), 7.01 (d, 1H, J=7.8 Hz), 7.38 (d, 1H, J=7.8 Hz).

Acknowledgements: The authors would like to express their gratitude to Prof. R. Suau (University of Málaga) for fruitful discussions and the micro-analytical determinations. This work was supported by the Spanish Ministry of Education and Science under DGICYT proyect No. PB90-0764 and through an FPI predoctoral fellowship awarded to M. R. Paleo.

## **REFERENCES AND NOTES**

- (a) Boente, J. M.; Castedo, L.; Cuadros, R.; Saá, J. M.; Suau, R.; Perales, A.; Martínez-Ripoll, M.; Fayos, J. Tetrahedron Lett. 1983, 24, 2029. (b) Boente, J. M.; Campello, M. J.; Castedo, L.; Domínguez, D.; Saá, J. M.; Suau, R.; Vidal, M.C. Tetrahedron Lett. 1983, 24, 4481. (c) Allais, D. P.; Guinaudeau, H. J. Nat. Prod. 1990, 53, 1280. (d) Alonso, R.; Castedo, L.; Domínguez, D. Tetrahedron Lett. 1986, 27, 3539.
- 2. Part of this work has been published in preliminary form: Paleo, M. R.; Domínguez, D.; Castedo, L. *Tetrahedron Lett.* 1993, 34, 2369.
- Kanaoka, Y.; Migita, Y.; Koyama, K.; Sato, Y.; Nakai, H.; Mizoguchi, T. Tetrahedron Lett. 1973, 1193.
- 4. For a review of the photochemistry of phthalimides see: Coyle, J. D. in Synthetic Organic Photochemistry; Horspool, W. M., Ed.; Plenum Press: New York, 1984. Chapter 4.
- 5. Kanaoka, Y.; Migita, Y. Tetrahedron Lett. 1974, 3693.
- 6. We thank Prof. Griesbeck for sending us a preprint of their manuscript before publication. Griesbeck. A. G.; Henz, A.; Hirt, J.; Ptatschek, V.; Engel, T.; Löffler, D.; Schneider, F. W. *Tetrahedron*, 1993, in press.
- After completion of our work Griesbeck et al. reported on the photochemistry of aliphatic N-phthaloyl αamino acid esters: (a) Griesbeck, A. G., Mauder, H. Angew. Chem. Int. Ed. Engl. 1992, 31, 73.
  (b) Griesbeck, A. G.; Mauder, H.; Müller, I. Chem. Ber. 1992, 125, 2467.
- (a) Stork, G.; Leong, A. Y. W.; Touzin, A. M. J. Org. Chem. 1976, 41, 3491. (b) Oguri, T.; Shioiri, T.; Yamada, S. Chem. Pharm. Bull. 1977, 25, 2287.
- 9. Griesbeck has reported a (9/1) trans /cis mixture in the case of N-Phthaloyl DOPA derivatives (ref. 6). In our experiments we could not determine the composition of the crude reaction mixture by <sup>1</sup>H-NMR due to its complexity. After separation of unreacted starting material and byproducts by PTLC, a 1/1 ratio of diastereoisomers was evident from <sup>1</sup>H-NMR. This final composition might be the result of epimerization at the carbon α to the carbonyl during work-up and does not necessarily imply a non-stereoselective reaction.
- 10. Bouquet, M.; Guy, A.; Lemaire, M.; Guette, J.P. Synth. Commun. 1985, 15, 1153.
- 11. Rosen, W.E.; Green, M.J. J. Org. Chem. 1963, 28, 2797.
- 12. Reeve, W.; Myers, H. J. Am. Chem. Soc. 1951, 73, 1371.
- 13. Perkin, W.H. Jr.; Robinson, R. J. Chem. Soc. 1907, 91, 1073.
- 14. Cushman, M.; Dikshit, D.K. J. Org. Chem. 1980, 45, 5064.
- 15. Rimek, H-J.; Yupraphat, T.; Zymalkowski, F. Liebigs Ann. Chem. 1969, 725, 116.
- 16. Coutts, R.T.; Malicky, J.L. Can. J. Chem. 1974, 52, 381.
- 17. Kim, J.C. Bull. Chem. Soc. Japan 1981, 3197.
- 18. Perrin, D.D.; Armarego, W.L.F. Purification of Laboratory Chemicals, 3rd.Ed., Pergamon, Oxford, 1988.
- 19 Kofron, W.G.; Baclawsky, L.M. J. Org. Chem. 1976, 41, 1879.

(Received in UK 2 December 1993; accepted 14 January 1994)