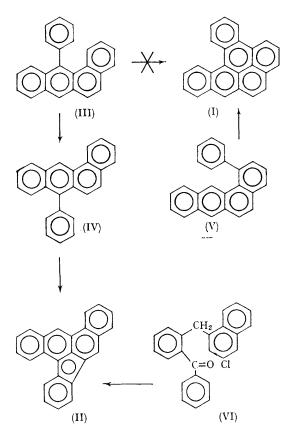
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The Synthesis of Dibenzo [a,l] pyrene

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LAVIT-LAMY and Buu-Hoï¹ have recently reported that what has been considered as "dibenzo[a,l]pyrene" (I)² is in fact dibenzo [a,e] fluoranthrene (II). The difficulty in attempting to prepare (I) from 12-phenylbenz[a]anthracene (III) is that (III) rearranges to 7-phenylbenz[a]anthracene (IV) which then yields (II). It appeared to us that a more suitable precursor to (I) would be 1-phenylbenz[a]anthracene (V) since (a) dehydrogenation would involve the active 12-position of the benz[a]anthracene moiety, and (b) the phenyl ring in (V) is less hindered than it is in (III) and hence should be less prone to rearrange.

We now report the first successful synthesis of dibenzo[a,l]pyrene (I) (m.p. 162—163°, pale yellow plates from benzene-ethanol). When (V) was treated with SnCl₄-AlCl₃ in benzene a 60% yield of (I) resulted. A sample of this interesting new hydrocarbon is being submitted for carcinogenic activity studies.

In support of Lavit-Lamy and Buu-Hoï's work we find that 2-(3-chloro-1-naphthylmethyl)benzophenone³ (VI) on treatment with alumina⁴ at 240° yields 40% of (II). Presumably the first step is an aromatic cyclodehydration4 to yield 6-chloro-7phenylbenz[a]anthracene which undergoes dehydrohalogenation to give (II).

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¹ D. Lavit-Lamy and N. P. Buu-Hoï, Chem. Comm., 1966, 92.

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⁴ F. A. Vingiello and A. Borkovec, J. Amer. Chem. Soc., 1956, 78, 3205.