

5-DIACETOXYMETHYL-*CYCLO*SAL-D4TMP—A PROTOTYPE OF ENZYMATICALLY ACTIVATED *CYCLO*SAL-PRONUCLEOTIDES

N. Gisch Department of Chemistry, Organic Chemistry, Faculty of Science, University of Hamburg, Martin-Luther-King-Platz 6, 20146 Hamburg, Germany

J. Balzarini Rega-Institute for Medical Research, K.U. Leuven, Minderbroedersstraat 10, 3000 Leuven, Belgium

C. Meier Department of Chemistry, Organic Chemistry, Faculty of Science, University of Hamburg, Martin-Luther-King-Platz 6, 20146 Hamburg, Germany

□ A new class of "lock-in"-modified cycloSal-pronucleotides has been synthesized. On the example of 5-diacetoxymethyl-cycloSal-d4T-monophosphate (5-di-AM-cycloSal-d4TMP), the concept of enzymatically activated cycloSal-pronucleotides is elucidated. Synthesis, hydrolysis studies, and antiviral activities against HIV are presented.

Keywords Pronucleotides; cycloSal; anti-HIV; enzymatic activation; antiviral activity

INTRODUCTION

One established concept for the delivery of therapeutically active nucleoside monophosphates (NMPs) like 2',3'-dideoxy-2',3'-didehydrothymidine monophosphate (d4TMP, 1) into cells is the *cyclo*Sal-concept.^[1] This prodrug concept has been successfully applied to a multitude of nucleoside analogs.^[2] Due to the chemically triggered delivery mechanism^[3] and the lipophilic character of *cyclo*Sal-triester a concentration equilibrium formed through the cell membrane is supposed. This is unfavorable for antiviral efficiency because a high intracellular concentration of the pronucleotide is necessary. Therefore "lock-in"-modified *cyclo*Sal-pronucleotides have been developed. These pronucleotides bear a (carboxy)esterase-cleavable ester site attached to the aromatic ring in order to trap the *cyclo*Sal-triester inside cells by cleavage of the ester group to release a highly polar derivate. To avoid a considerable reduction of the chemical stability of the *cyclo*Saltriester, these groups have been separated from the aromatic ring by an

Address correspondence to N. Gisch, University of Hamburg, Martin-Luther-King-Platz 6, Hamburg 20146, Germany. E-mail: gisch@chemie.uni-hamburg.de



FIGURE 1 Concept of enzymatically activated *cyclo*Sal-Pronucleotides shown on the example of 5diacetoxymethyl-*cyclo*Sal-d4TMP 2.

alkyl spacer. It has been shown that an effective intracellular trapping should be possible, if highly polar *cyclo*Sal-d4TMP acids are released from *cyclo*Sal-d4TMP acid ester. From 5-propionyl-*cyclo*Sal-d4TMP acid, d4TMP was released by the known chemically induced pathway. However, these "lock-in"-compounds led to a delayed drug delivery due to high chemical stability.^[4,5] Here, we present a prototype of the new concept of enzymatically activated *cyclo*Sal-pronucleotides, 5-diacetoxymethyl-*cyclo*Sal-d4TMP **2**. In this concept, after passive transport of **2** into cells (step c, Figure 1), the lipophilic acylal substituent having a weak electron-withdrawing effect attached to the aromatic ring is converted into a highly polar aldehyde group with a strong electron-withdrawing effect by intracellular cleavage (step d, Figure 1).

The formed acceptor group leads to a strong decrease in hydrolysis stability and the rapid formation of a charged intermediate **4** (phosphodiester),^[3] so that compound **3** should not be effluxed (step e, Figure 1) and the phosphodiester intermediate **4** should not be effluxed at all due to the resulting polarity. From the phosphodiester d4TMP **1** is released subsequently (step f, Figure 1). This concept is based on the higher intracellular concentration of esterases compared to the extracellular medium^[6] (Figure 1, step b should not take place) and on a considerable difference



SCHEME 1 Synthesis of 5-diacetoxymethyl-*cyclo*Sal-d4TMP **2**. Method **A**: THF, LiAlH₄, room temp. to reflux, 3 h, 91%; method **B**: acetone, 2,2-dimethoxypropane, *p*TsOH, Na₂SO₄, 40°C, 3 d, 95%; method **C**: THF, *n*-BuLi, DMF, -78° C, 3 h, 94%; method **D**: CH₃CN/H₂O, cat. HCl, 81%; method **E**: i) THF, PCl₃, pyridine, -20° C to room temp., 4.5 h; ii) CH₃CN, DIPEA, d4T, -20° C to room temp., 3 h; iii) CH₃CN, tBuOOH, -20° C to room temp., 1 h, 31%; method **F**: CH₃CN, acetic anhydride, ZrCl₄; room temp., 45 min, 44%.

between the extracellular hydrolysis stability of the 5-diacetoxymethylcycloSal-d4TMP **2** and the intracellular hydrolysis stability of the 5-formyl-cycloSal-d4TMP **3** after enzymatical cleavage (Figure 1, $t_{1/2}$ step a $\gg t_{1/2}$ step d).

RESULTS

The title compound **2** was synthesized starting from the commercial available 5-bromosalicyl aldehyde **5** as outlined in Scheme 1.

Reduction of **5** gave the 4-bromosalicyl alcohol **6** in 91% yield. After protection of **6** as isopropylidene acetal (**7**, 95% yield) the formyl group was introduced via bromo-lithium exchange to yield 4-formylsalicyl alcohol isopropylidene acetal **8** in 94%. By means of acidic deprotection 4-formylsalicyl alcohol **9** could be obtained in 81% yield. Compound **9** was converted to **3** (mixture of two diastereomeres) using chlorophosphite chemistry as described before.^[1] The only modification was an exchange of the solvent. The title compound **2** was synthesized by protection of the formyl group of **3** as an acylal in 44% yield.

The *cyclo*Sal-triesters **2** and **3** were studied for their stability in aqueous 25 mM phosphate buffer (pH = 7.3). As expected, 5-formyl-*cyclo*Sal-d4TMP **3** has a very short half-life of 0.18 h. The half-life of 5-diacetoxymethyl*cyclo*Sal-d4TMP **2** is 6-fold higher ($t_{1/2} = 1.2$ h). In the study of **2** no competing hydrolysis of the acylal ester group was observed. So, the half-life corresponds to the cleavage of the triesters and the exclusive formation of d4TMP **1** (proven by ³¹P-NMR hydrolysis). The cleavage of the acylal group of **2** was shown in hydrolysis studies in T-lymphocyte cell extracts. The halflife was 0.08 h and the corresponding 5-formyl-*cyclo*Sal-d4TMP **3** was formed (HPLC co-elution experiments). *Cyclo*Sal-triester **2** and **3** were tested for their anti-HIV activity in wild type CEM/0 and mutant thymidine-kinasedeficient CEM/TK⁻ cells. As reference d4T (active against HIV-1 and HIV-2 in CEM/0 but weakly active in CEM/TK⁻ cells) was used. Compounds **2** and **3** have the same activity against HIV-1 in wild-type cells and against HIV-2 in CEM/0 cells as d4T (**2**: $0.42\pm0.28 \ \mu$ M and $0.40\pm0.0 \ \mu$ M; **3**: $0.41\pm0.29 \ \mu$ M and $0.15\pm0.08 \ \mu$ M; d4T: $0.48\pm0.45 \ \mu$ M and $0.63\pm0.21 \ \mu$ M).

The activity of triesters **2** and **3** against HIV-2 in CEM/TK⁻ cells is 26-fold, respectively, 33-fold lower as in CEM/0 cells (**2**: 10.5±8.3 μ M; **3**: 5.0±4.6 μ M). However, the activity of d4T decreases even 100-fold (d4T: 47.5±26.3 μ M). So, a partial delivery of the pronucleotide takes place. The loss of antiviral activity in CEM/TK⁻ is comparable with the values for other acceptor substituted *cyclo*Sal-d4TMPs with low hydrolysis half-lives.^[2]

In conclusion, a fast release of NMPs by enzymatic activation out of *cyclo*Sal-prodrugs like compound **2** seems to be possible. Further work will be done to increase the hydrolysis stability in order to achieve full retention of antiviral activity.

REFERENCES

- Meier, C. cycloSal-pronucleotides–Design of chemical trojan horses. Mini Rev. Medicinal Chem. 2002, 2, 219–234.
- Meier, C. cycloSal phosphates as chemical trojan horses for intracellular nucleotide and glycosylmonophosphate delivery–Chemistry meets biology. Eur. J. Organic Chem. 2006, 1081–1102.
- Meier, C. 2004. cycloSal-Pronucleotides–Design of the concept, chemistry, and antiviral activity. In Advances in Antiviral Drug Design, Vol. 4, E. de Clercq, Ed. Elsevier, Amsterdam, pp. 147–213.
- Meier, C.; Ruppel, M.F.H.; Vukadinovic, D.; Balzarini, J. Second generation of *cyclo*Sal-pronucleotides with esterase-cleavable sites: The "lock-in"-concept. *Nucleosides, Nucleotides & Nucleic Acids* 2004, 23, 89–115.
- Meier, C.; Ducho, C.; Jessen, H.; Vukadinovi-Tenter, D.; Balzarini, J. Second-generation cycloSald4TMP pronucleotides bearing esterase-cleavable sites–The "trapping" concept. *Eur. J. Organic Chem.* 2006, 197–206.
- Perigaud, C.; Gosselin, G.; Imbach, J.-L. Minireview: from the pronucleotide concept to the SATE phosphate protecting groups. *Current Topics Med. Chem.* 1997, 2, 15–29.

Copyright of Nucleosides, Nucleotides & Nucleic Acids is the property of Taylor & Francis Ltd and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.