

# Phosphine-Catalyzed Cascade Annulation of MBH Carbonates and Diazenes: Synthesis of Hexahydrocyclopenta[c]pyrazole Derivatives

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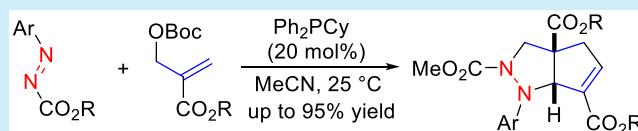
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**ABSTRACT:** A phosphine-catalyzed cascade annulation of Morita–Baylis–Hillman (MBH) carbonates and diazenes was achieved, giving tetrahydropyrazole-fused heterocycles bearing two five-membered rings in moderate to excellent yields. The reaction underwent an unprecedented reaction mode of MBH carbonates, in which two molecules of MBH carbonates were fully merged into the ring system.

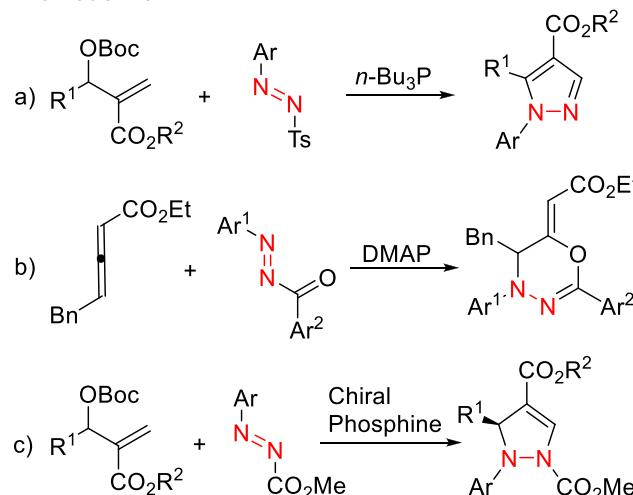


**P**yrazole, dihydropyrazole, and tetrahydropyrazole are among the most widespread motifs in agrochemicals and drugs.<sup>1</sup> The related heterocycles are privileged structures in agrochemical and medicinal chemistry.<sup>2</sup> Therefore, their synthesis is highly desirable and has attracted much attention.

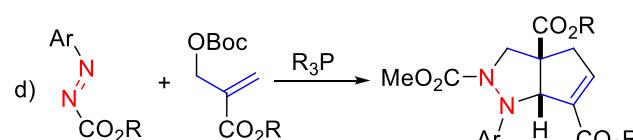
In the past two decades, phosphine-catalyzed annulation reactions have been developed as a powerful tool for synthesis of carbo- and heterocycles.<sup>3</sup> Among them, the annulation reactions with Morita–Baylis–Hillman (MBH) carbonates as phosphine acceptor have been extensively investigated.<sup>3h,j,4–11</sup> MBH carbonates often worked as a three-membered synthon to achieve [3 + 2],<sup>5</sup> [3 + 3],<sup>6</sup> [3 + 4],<sup>7</sup> and [3 + 6]<sup>8</sup> annulation reactions and sometimes functioned as a C1 synthon to furnish [1 + 4] annulation reactions.<sup>9</sup> In these reactions, electron-deficient carbon–carbon,<sup>5a–g</sup> carbon–nitrogen,<sup>5h,i</sup> carbon–oxygen,<sup>5j</sup> and nitrogen–nitrogen<sup>11</sup> double bonds are widely used as electrophiles to accomplish numerous annulation reactions. Obviously, the annulation reactions of nitrogen–nitrogen double bond with MBH carbonates are a concise synthetic tool for pyrazole, dihydropyrazole, and tetrahydropyrazole compounds (Scheme 1).<sup>11</sup> At an early stage, in the presence of an equivalent of phosphine, azodicarboxylates were used as phosphine acceptor for synthesis of dinitrogen-fused compounds.<sup>12</sup> In 2015, Meng and Wang developed a Bu<sub>3</sub>P-catalyzed desulfonylative [3 + 2] cycloaddition of MBH carbonates with arylazosulfones for synthesis of pyrazole derivatives in good to excellent yields (Scheme 1a).<sup>11a</sup> With the use of 4-dimethylaminopyridine (DMAP) as the catalyst, Meng and Wang also achieved a similar [3 + 2] cycloaddition of isatin-derived MBH carbonates with diazenes for the construction of 3-spiropyrazole-2-oxindoles in good to excellent yields.<sup>11c</sup> The same research group also reported a DMAP-catalyzed [2 + 4] cycloaddition of allenotes and N-acyldiazenes to generate 1,3,4-oxadiazine derivatives in moderate to good yields (Scheme 1b).<sup>11b</sup> In 2019, we accomplished a phosphine-catalyzed asymmetric [3 + 2] cycloaddition of diazenes with MBH carbonates to give chiral

Scheme 1. Lewis Base-Catalyzed Annulation Reactions of Diazenes

**Previous work:**



**This work:**



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dihydropyrazoles in high yields with excellent enantioselectivities (**Scheme 1c**).<sup>11d,13</sup> On the basis of our previous work,<sup>11d,13</sup> we herein disclose an unprecedented phosphine-catalyzed cascade annulation of MBH carbonates and diazenes for synthesis of hexahydrocyclopenta[c]pyrazole derivatives (**Scheme 1d**). Different from previous annulation reactions involving MBH carbonates, two molecules of MBH carbonates were merged into two five-membered rings in the current reaction.

Initially, we began our investigation of cascade annulation between diazene **1a** and MBH carbonate **2a** in dichloromethane at room temperature (**Table 1**). A variety of

**Table 1. Optimization of Reaction Conditions<sup>a</sup>**

| entry           | R <sub>3</sub> P    | solvent                           | T (°C) | yield (%) <sup>b</sup> |
|-----------------|---------------------|-----------------------------------|--------|------------------------|
| 1               | Ph <sub>3</sub> P   | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | trace                  |
| 2               | MePh <sub>2</sub> P | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 48                     |
| 3               | BnPh <sub>2</sub> P | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 42                     |
| 4               | CyPh <sub>2</sub> P | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 64                     |
| 5               | Bn <sub>3</sub> P   | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 62                     |
| 6               | Cy <sub>3</sub> P   | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 58                     |
| 7               | Bu <sub>3</sub> P   | CH <sub>2</sub> Cl <sub>2</sub>   | 25     | 44                     |
| 8               | CyPh <sub>2</sub> P | (CH <sub>2</sub> Cl) <sub>2</sub> | 25     | 67                     |
| 9               | CyPh <sub>2</sub> P | THF                               | 25     | 42                     |
| 10              | CyPh <sub>2</sub> P | toluene                           | 25     | 37                     |
| 11              | CyPh <sub>2</sub> P | MeCN                              | 25     | 82                     |
| 12              | CyPh <sub>2</sub> P | 1,4-dioxane                       | 25     | trace                  |
| 13              | CyPh <sub>2</sub> P | MeOH                              | 25     | trace                  |
| 14 <sup>c</sup> | CyPh <sub>2</sub> P | MeCN                              | 25     | 84                     |
| 15 <sup>d</sup> | CyPh <sub>2</sub> P | MeCN                              | 25     | 82                     |
| 16              | CyPh <sub>2</sub> P | MeCN                              | 40     | 68                     |
| 17 <sup>e</sup> | CyPh <sub>2</sub> P | MeCN                              | 0      | 82                     |

<sup>a</sup>Unless otherwise indicated, reactions of **1a** (0.10 mmol) and **2a** (0.24 mmol) were carried out in the presence of phosphine (0.02 mmol) in 2 mL of the solvent for 3 h. <sup>b</sup>Isolated yield, >20:1 dr. <sup>c</sup>0.3 mmol of **2a** was used in the reaction. <sup>d</sup>0.4 mmol of **2a** was used in the reaction. <sup>e</sup>The reaction time was 13 h.

phosphine catalysts were explored (entries 1–7). The Ph<sub>3</sub>P with a low nucleophilicity did not efficiently catalyze the reaction, leading to only a trace of the desired product (entry 1). Other phosphines having at least one alkyl group displayed certain catalytic activity, leading to the desired product in moderate to high yields (entries 2–7). Among these phosphines, CyPh<sub>2</sub>P demonstrated relatively good catalytic ability, resulting in the product in 64% yield (entry 4). With the use of CyPh<sub>2</sub>P as the catalyst, a concise solvent screening was performed (entries 8–13). The chlorinated solvent 1,2-dichloroethane was compatible with the reaction and provided a 67% yield close to that obtained in the solvent CH<sub>2</sub>Cl<sub>2</sub> (entry 8 vs entry 4). Toluene and THF gave inferior results (entries 9 and 10). Polar solvent acetonitrile was efficient, offering a high 82% yield (entry 11). Another two polar solvents, 1,4-dioxane and methanol, were incompatible, affording only a trace of products (entries 12 and 13). Increasing the ratio of MBH carbonate to diazene resulted in a slightly higher 84% yield (entry 14), but further increasing the ratio of MBH carbonate

**2a** to diazene **1a** did not help enhance the yield (entry 15). The effect of temperature was also explored. Increasing reaction temperature to 40 °C reduced the yield (entry 16), while decreasing reaction temperature to 0 °C had no negative effect on the yield but required longer reaction time (entry 17). On the basis of the above screening, the optimal reaction conditions were determined as follows: 0.1 mmol of diazenes and 0.3 mmol of MBH carbonates with 20 mol % CyPh<sub>2</sub>P in 2 mL of MeCN at 25 °C.

Under the optimized reaction conditions, we examined reactions between various diazenes **1** with MBH carbonate **2a** (**Table 2**, entries 1–17). Diazenes with both electron-donating

**Table 2. Scope of Diazenes<sup>a</sup>**

| entry | R/Ar in <b>1</b>   | t (h) | <b>3</b>   | yield (%) <sup>b</sup> |
|-------|--|-------|------------|------------------------|
| 1     | Me/2-MeC <sub>6</sub> H <sub>4</sub> ( <b>1b</b> )                 | 2     | <b>3ba</b> | 45                     |
| 2     | Me/3-MeC <sub>6</sub> H <sub>4</sub> ( <b>1c</b> )                 | 2     | <b>3ca</b> | 84                     |
| 3     | Me/4-MeC <sub>6</sub> H <sub>4</sub> ( <b>1d</b> )                 | 5     | <b>3da</b> | 61                     |
| 4     | Me/2,4-Me <sub>2</sub> C <sub>6</sub> H <sub>3</sub> ( <b>1e</b> ) | 2     | <b>3ea</b> | 65                     |
| 5     | Me/3-MeOC <sub>6</sub> H <sub>4</sub> ( <b>1f</b> )                | 2     | <b>3fa</b> | 82                     |
| 6     | Me/4-MeOC <sub>6</sub> H <sub>4</sub> ( <b>1g</b> )                | 3     | <b>3ga</b> | 59                     |
| 7     | Me/4-EtC <sub>6</sub> H <sub>4</sub> ( <b>1h</b> )                 | 2     | <b>3ha</b> | 89                     |
| 8     | Me/4-i-PrC <sub>6</sub> H <sub>4</sub> ( <b>1i</b> )               | 5     | <b>3ia</b> | 88                     |
| 9     | Me/3-FC <sub>6</sub> H <sub>4</sub> ( <b>1j</b> )                  | 3     | <b>3ja</b> | 48                     |
| 10    | Me/4-FC <sub>6</sub> H <sub>4</sub> ( <b>1k</b> )                  | 1     | <b>3ka</b> | 63                     |
| 11    | Me/3,4-F <sub>2</sub> C <sub>6</sub> H <sub>3</sub> ( <b>1l</b> )  | 2     | <b>3la</b> | 61                     |
| 12    | Me/4-ClC <sub>6</sub> H <sub>4</sub> ( <b>1m</b> )                 | 2     | <b>3ma</b> | 55                     |
| 13    | Me/2-BrC <sub>6</sub> H <sub>4</sub> ( <b>1n</b> )                 | 2     | <b>3na</b> | 46                     |
| 14    | Me/4-BrC <sub>6</sub> H <sub>4</sub> ( <b>1o</b> )                 | 3     | <b>3oa</b> | 70                     |
| 15    | Me/2-naphthyl ( <b>1p</b> )  | 3     | <b>3pa</b> | 77                     |
| 16    | Me/6-Br-2-naphthyl ( <b>1q</b> )                                   | 6     | <b>3qa</b> | 63                     |
| 17    | t-Bu/C <sub>6</sub> H <sub>5</sub> ( <b>1r</b> )                   | 2     | <b>3ra</b> | 68                     |

<sup>a</sup>Unless otherwise indicated, all reactions were carried out with **1** (0.1 mmol), **2a** (0.3 mmol), and CyPh<sub>2</sub>P (0.02 mmol) in 2 mL of MeCN at 25 °C. <sup>b</sup>Isolated yield, >20:1 dr.

and -withdrawing groups on the phenyl ring were well tolerated to produce the corresponding annulation products in moderate to good yields. Various diazenes bearing functional groups such as methoxy (**1f** and **1g**), fluorine (**1j**–**1l**), chlorine (**1m**) and bromine (**1n** and **1o**) behaved properly in this catalytic system. The substrates having an *ortho*-substituted phenyl (**1b** and **1n**) gave relatively low yields of the corresponding products, indicating that steric hindrance has a remarkable influence on the reactivity of the substrates (entries 1 and 13). Diazenes with disubstituted phenyl such as **1e** and **1l** were proven to be suitable substrates for the cascade cycloaddition, giving the hexahydrocyclopenta[c]pyrazole derivatives in good yields (entries 4 and 11). To our delight, 2-naphthyl and substituted naphthyl group (**1p** and **1q**) were also carried out, and the corresponding products (**3pa** and **3qa**) were obtained in high yields (entries 15 and 16). The relative configuration of the annulation products has been confirmed by X-ray crystallography of the product **3aa**.

As shown in **Table 3**, the scope of MBH carbonates was also evaluated (entries 1–7). The reactions of several MBH carbonates with different alkyl groups in the ester moiety

Table 3. Scope of MBH Carbonates<sup>a</sup>

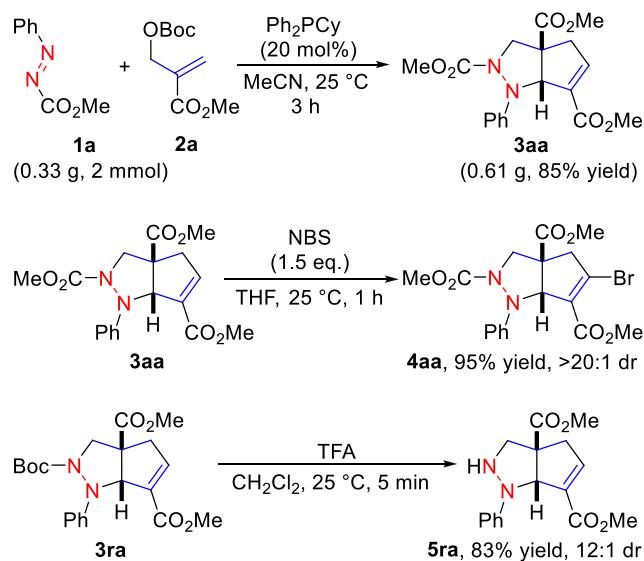
| entry | R         | t (h) | 3   | yield (%) <sup>b</sup> |
|-------|-----------|-------|-----|------------------------|
| 1     | Et (2b)   | 2     | 3ab | 71                     |
| 2     | n-Pr (2c) | 2     | 3ac | 59                     |
| 3     | Cy (2d)   | 2     | 3ad | 95                     |
| 4     | Bn (2e)   | 2     | 3ae | 57                     |
| 5     | i-Pr (2f) | 2     | 3af | 60                     |
| 6     | Ad (2g)   | 2     | 3ag | 53                     |
| 7     | t-Bu (2h) | 24    | 3ah | trace                  |

<sup>a</sup>Unless otherwise indicated, all reactions were carried out with **1a** (0.1 mmol), **2** (0.3 mmol), and CyPh<sub>2</sub>P (0.02 mmol) in 2 mL of MeCN at 25 °C. Ad: 1-adamantyl. <sup>b</sup>Isolated yield, >20:1 dr.

(**2b–2g**) proceeded smoothly to give the desired products (**3ab–3ag**) in moderate to high yields (entries 1–6). Particularly, when the cyclohexyl-bearing MBH carbonate **2d** was used, 95% yield of the annulation product **3ad** was obtained (entry 3). The *tert*-butyl substituted MBH carbonate **2h** was not reactive, leading to only a trace of product **3ah** (entry 7), probably because of its steric hindrance. It should be noted that those MBH carbonates derived from other aldehydes, such as benzaldehyde, did not work in this reaction.

To our delight, the scale-up reaction still worked well under standard reaction conditions. As indicated in Scheme 2, 2

### Scheme 2. Scale-up Reaction and Further Transformations of the Products

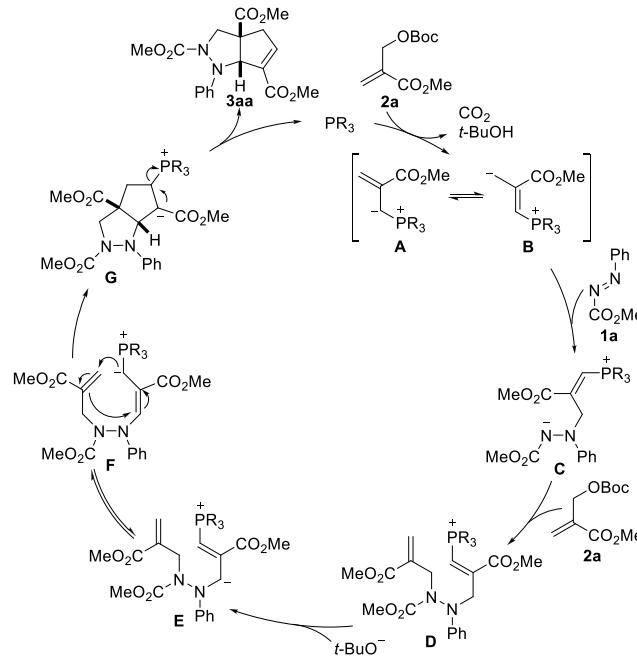


mmol of the diazene **1a** (0.33 g) reacted with MBH carbonate **2a** (6 mmol, 1.30 g) in acetonitrile at room temperature for 3 h, leading to the cycloaddition product **3aa** in 85% yield (0.61 g). No loss of yield was observed in the scale-up reaction, compared with the reaction at 0.1 mmol scale. Further transformations of the product **3** also were investigated (Scheme 2). Treatment of **3aa** with N-bromosuccinimide (NBS) gave the brominated product **4aa** in excellent yield with >20:1 dr. In the presence of trifluoroacetic acid (TFA), the

Boc group in the product **3ra** was easily removed to give the derivative **5ra** in high yield with good dr.

On the basis of phosphine catalysis principles, a plausible mechanism was proposed for this cascade annulation reaction (Scheme 3). Initially, the phosphine catalyst attacks the MBH

### Scheme 3. A Plausible Mechanism



carbonate **2a** to form zwitterionic intermediate **A**. Its tautomeric isomer **B** then attacks the diazene **1a** to give the ylide **C**, which reacts with another molecule of MBH carbonate **2a** to generate the intermediate **D**. Subsequent deprotonation of the intermediate **D** through *tert*-butoxy anion from decarboxylation of the substrate **2a** leads to the ylide **E**, which performs an isomerization to afford the intermediate **F**. Consequent intramolecular nucleophilic addition triggers highly diastereoselective cascade annulation to furnish the intermediate **G** having two five-membered rings. Subsequent formation of the carbon–carbon double bond with simultaneous regeneration of phosphine catalyst produces the final product **3aa**.

In summary, we achieved the cascade annulation of MBH carbonates with diazenes under mild reaction conditions. A variety of tetrahydropyrazole-fused heterocycles were obtained in moderate to excellent yields. The reaction undergoes a very interesting reaction pathway and demonstrates an unprecedented reaction mode of MBH carbonates under phosphine catalysis conditions.

### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c01975>.

Experimental procedure, characterization data, NMR spectra, and X-ray crystallographic data (PDF)

#### Accession Codes

CCDC 2055670 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge

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#H.L. and W.S. contributed equally to this work.

### Notes

The authors declare no competing financial interest.

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