

Communication

Enantioselective Three-Component Fluoroalkylarylation of Unactivated Olefins Through Nickel-Catalyzed Cross-Electrophile Coupling

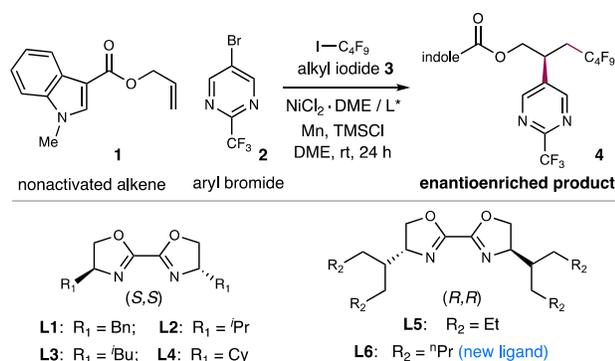
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J. Am. Chem. Soc., **Just Accepted Manuscript** • DOI: 10.1021/jacs.0c03708 • Publication Date (Web): 11 May 2020

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Table 1. Optimization of reaction conditions^a

entry	variations from standard conditions	yield	er
1	L1	21%	20:80
2	L2	55%	12:88
3	L3	24%	10:90
4	L4	75%	9:91
5	L5	87%	94:6
6	L6	81%	94:6
7	-10 °C, instead of rt	76%	95:5
8	Zn or TDAE, instead of Mn	trace	--
9	w/o nickel, L5, or Mn	0%	--
10	w/o TMSCl	66%	93:7

^aReactions were carried out with alkene **1** (0.2 mmol), aryl bromide **2** (0.1 mmol), C₄F₉I (0.2 mmol), NiCl₂·glyme (10 mol%), chiral ligand (12 mol%), TMSCl (0.01 mmol), Mn (0.25 mmol), DME [0.5 M], rt, 24 h. Yields were determined by GC using an internal standard. The er values were determined by HPLC on a chiral stationary phase.

directing group, to facilitate the stereocontrolled radical/nickel combination step, thus fulfilling catalytic enantioselective control in the intermolecular three-component version of unactivated alkenes.

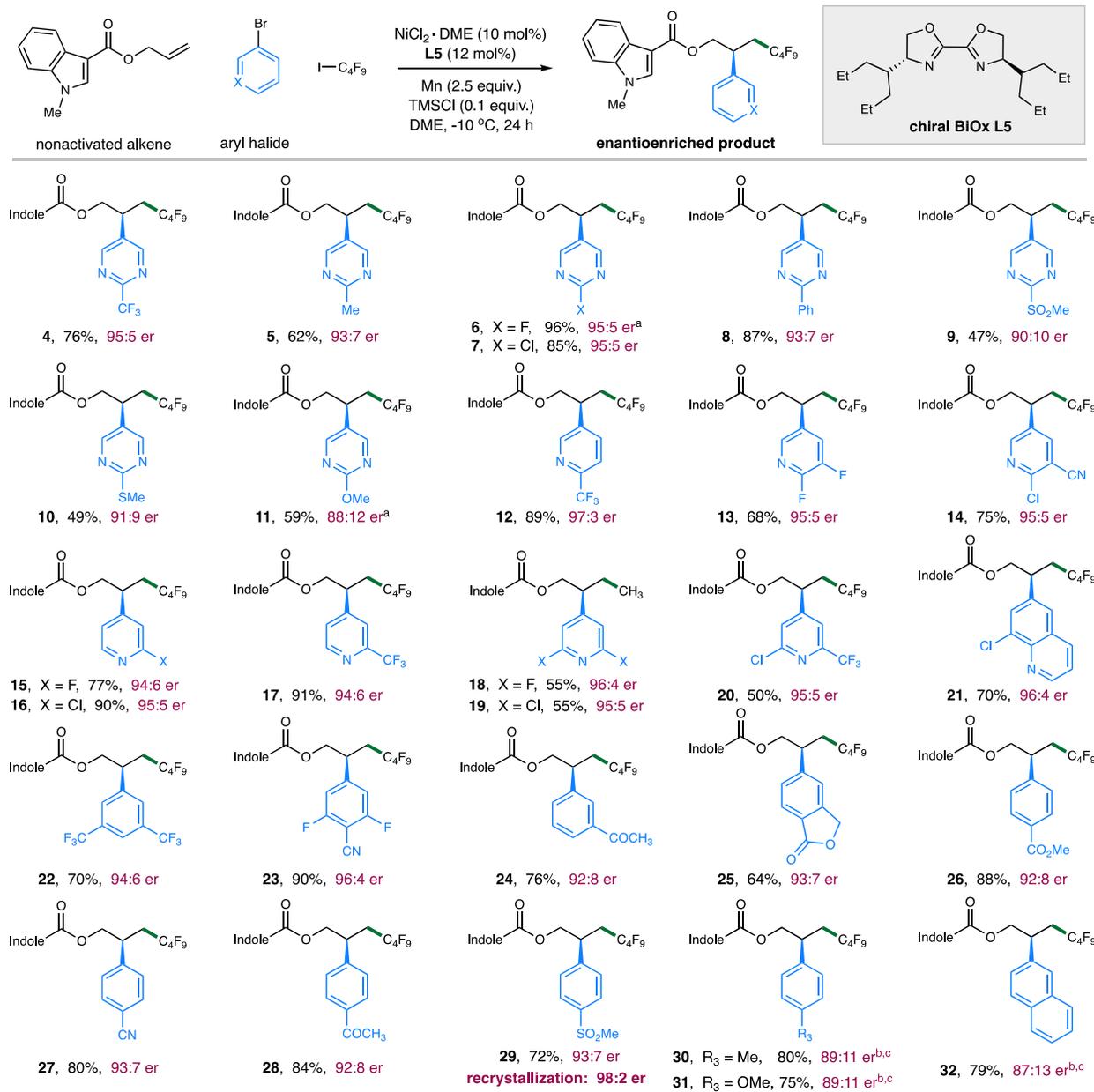
With this idea in mind, we began our investigations by evaluating unactivated alkenes incorporated with different types of directing groups. Pleasingly, we found that allylic ester, the carbonyl group of which could interact with nickel species via a six-membered-chelated ring, was a promising non-conjugated alkene substrate for this enantioselective, three-component dicarbofunctionalization reaction (See Table S1). Employing the catalyst combination of NiCl₂·DME and chiral bioxazoline (Bn-BiOX, L1), reaction of allyl ester **1** with 5-bromo-2-(trifluoromethyl)pyrimidine **2** and C₄F₉I in the presence of stoichiometric Mn and catalytic TMSCl afforded the desired 1,2-fluoroalkylarylation product **4** in 21% yield with an encouraging 20:80 (*S*:*R*) enantiomeric ratio (er) (Table 1, entry 1). Several pendant directing groups such as carbonates, carbamates and benzoates, afforded decreased ers with varied yields (Table S1). Screening of chiral ligands indicated that only BiOx ligands^{4d, 6c, 6h} were effective, while other commonly employed chiral ligands all

failed to promote this three-component reaction under similar conditions (Table S2 and S7). Therefore, we further screened a series of chiral BiOx ligands, and found that BiOx ligands containing longer alkyl chains offered higher yield and enantioselectivity, and (*R,R*)-4-heptyl-BiOX (L5), originally developed for Ni-catalyzed asymmetric reductive cross-couplings,^{6c, 6h} delivered product **4** with 87% yield and 94:6 er at room temperature (entries 2–5). Encouraged by this result,^{6h} we designed and synthesized a new (*R,R*)-5-nonyl-BiOX ligand (L6) bearing alkyl chains with nine carbon atoms, which to our disappointment provided the same er with L5 (entry 6). Finally, conducting the reaction at lower temperature (-10 °C) slightly improved enantioselectivity, furnishing **4** with 95:5 er in 76% yield (entry 7). A trace amount of product was observed when employing Zn or TDAE (tetrakis(dimethylamino)ethylene)¹² as a stoichiometric reductant (entry 8). Control experiments further confirmed that all nickel catalyst, ligand, and Mn dust were required for this asymmetric system (entry 9). Employment of TMSCl, which could activate the surface of Mn, as additive was important to ensure reproducibility for this transformation (entry 10) (See Table S1–S11 for more optimizations).

With optimal reaction conditions in hand, we began to explore the generality of this asymmetric, three-component 1,2-fluoroalkylarylation of unactivated alkenes with respect to aryl and heteroaryl halides. As shown in Scheme 1, a variety of 5-bromopyrimidines bearing electron-withdrawing and electron-donating groups all underwent the desired cross-electrophile coupling smoothly, delivering the corresponding chiral β-fluoroalkyl arylalkanes with good to excellent yields and high enantioselectivity (Products **4–11**, 47%–96% yields, up to 95:5 er); while pyrimidines with electron-donating groups displayed lower efficiency (Products **10** and **11**). Moreover, electron-deficient pyridines and quinolines were also viable coupling partners, furnishing 1,2-fluoroalkylarylation products in good yields and high ers (Products **12–21**, 50%–91% yields, up to 97:3 er). It should be noted that ortho-substituents in pyrimidines and pyridines were found to be crucial to achieving high enantioselective control in this transformation, and simple 2-bromopyridine was ineffective under the standard conditions. We surmised that the presence of 2-substituent would prevent the undesired coordination between nickel and nitrogen atom of heterocycles. Additionally, this catalytic protocol demonstrated excellent chemoselectivity, as exclusive reactivity toward bromides over chlorides was observed in the cases of multiple halogenated heteroarenes (Products **7**, **14**, **16**, **19–21**). Moreover, electron-deficient aryl bromides also worked well to give the three-component coupling products in good yields and high ers (Products **26–29**, 72%–88% yields, up to 98:2 er). Notably, simple recrystallization from hot ethanol furnished product **29** in 96% ee. The mild reaction conditions were well compatible with a wide range of important functionalities, including CF₃, chloride, thioether, sulfone, cyano, ketone, and ester, offering useful handles for further synthetic manipulations (Products **4–29**). Nevertheless, electron-neutral and -rich aryl iodides showed lower reactivity in this catalytic protocol, furnishing the final

products with moderate yields and ers (Products **30–32**, 75%–80% yields, up to 89:11 er).

1-chloro-2-iodotetrafluoroethane proceeded smoothly to afford product **39**, via selective cleavage of C–I bond, in 66%

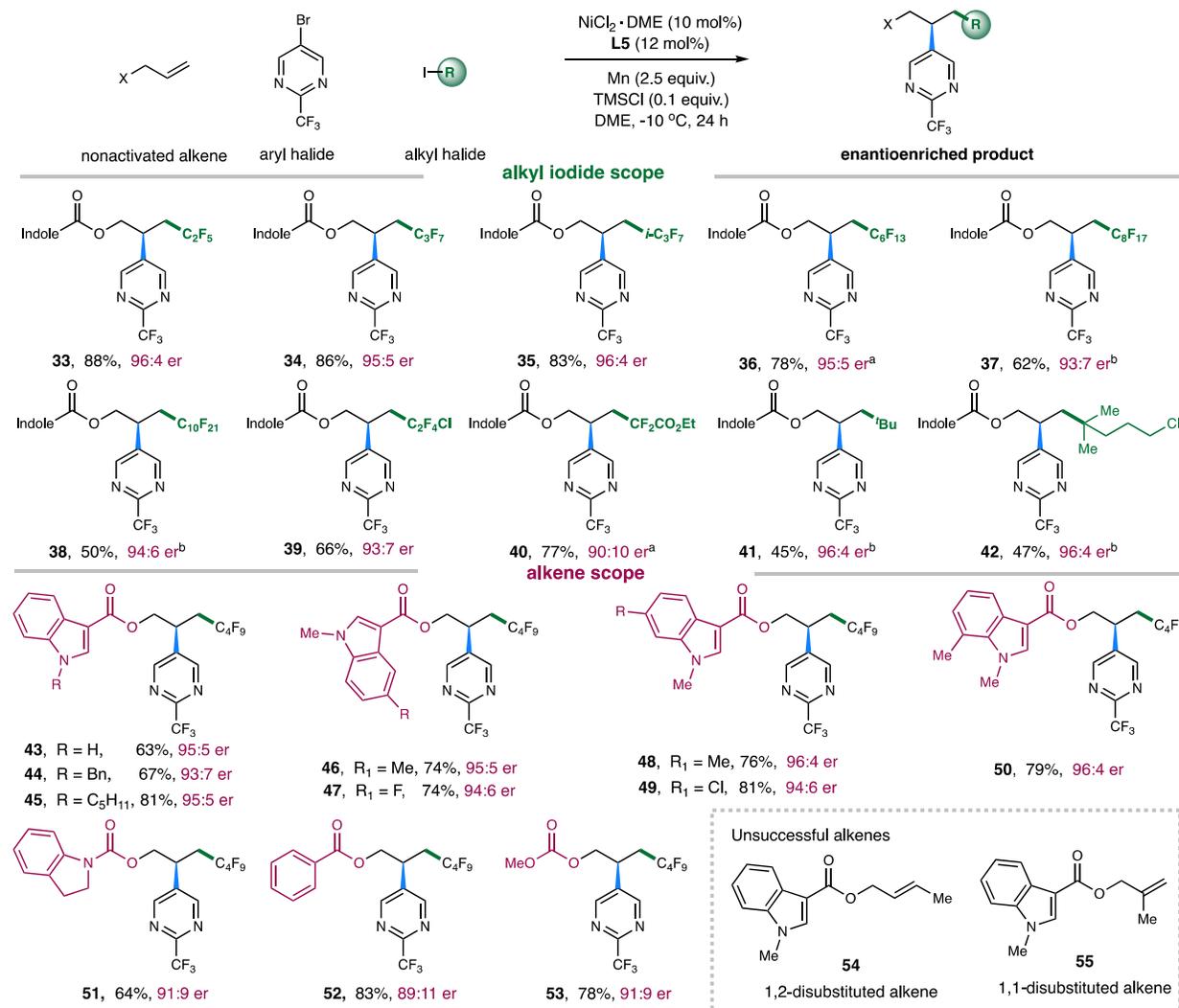


Scheme 1. Scope of aryl halides. Reaction conditions: alkene **1** (0.2 mmol), aryl bromide (0.1 mmol), $\text{C}_4\text{F}_9\text{I}$ (0.2 mmol), $\text{NiCl}_2 \cdot \text{glyme}$ (10 mol%), **L5** (12 mol%), TMSCl (0.01 mmol), Mn (0.25 mmol), DME [0.5 M], -10°C , 24 h. Isolated yields. The er values were determined by HPLC on a chiral stationary phase. ^aConducted at 0°C . ^bConducted at room temperature. ^cWith aryl iodide. Indole = 1-methyl-1*H*-indole.

Next, we examined the scope of the fluoroalkyl precursors. As depicted in Scheme 2, a wide array of perfluoroalkyl iodides (from C_2 to C_{10}) could be employed as competent C-radical precursors, leading to the corresponding enantioenriched fluoroalkyl-containing motifs in moderate to high yields and high ers (Products **33–38**, 50%–88% yields, up to 96:4 er). While reaction efficiency decreased to a small extent as the fluoroalkyl carbon chains grew longer, probably due to the lower solubility of larger perfluoroalkyl iodides in the current system (Products **37** and **38**). The reaction with

yield and 93:7 er. Moreover, ethyl difluoroiodoacetate also underwent this asymmetric three-component protocol smoothly (Product **40**, 77% yield, 90:10 er). Pleasingly, we also found that tertiary alkyl iodides could be employed as effective nonfluorinated C-radical precursors to afford the 1,2-alkylarylated products **41** and **42** with 96:4 ers, albeit in relatively low yields (45% and 37% yield, respectively).¹³

Finally, we also examined the reactivity of unactivated alkenes under the optimal conditions (Scheme 2). Allyl esters derived from different indole-3-carboxylic acids all



Scheme 2. Scope of alkenes and alkyl halides. Reaction conditions: alkene (0.2 mmol), aryl bromide (0.1 mmol), alkyl halide (0.2 mmol), $\text{NiCl}_2 \cdot \text{glyme}$ (10 mol%), **L5** (12 mol%), TMSCl (0.01 mmol), Mn (0.25 mmol), DME [0.5 M], -10°C , 24 h. Isolated yields. The er values were determined by HPLC on a chiral stationary phase. ^aConducted at 0°C . ^bConducted at room temperature. Indole = 1-methyl-1*H*-indole

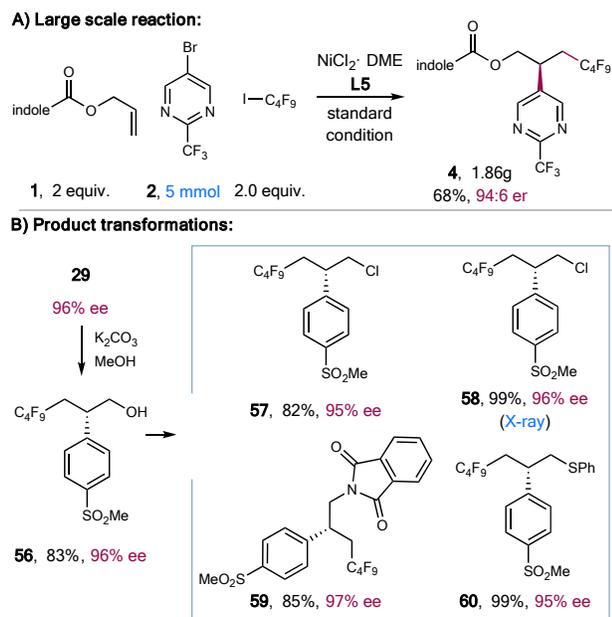
underwent the desired three-component coupling reaction in good yields and high ers (Products **43–50**, 63%–81% yields, up to 96:4 er). Substituents on nitrogen atom or the indole ring had no significant influence on the yield and enantioselectivity of this transformation. Unprotected indole was also well-tolerated under mild conditions (Product **43**). Interestingly, allyl esters incorporating other types of directing groups, exemplified by benzoate, carbonate, and carbamate, were also applicable in this system, albeit with only moderate enantioselectivity (Products **51–53**, 64%–83% yields, up to 91:9 er). Nevertheless, 1,1-disubstituted alkenes and 1,2-disubstituted internal alkenes were generally ineffective (e.g. **54** and **55**).

This asymmetric, three-component difunctionalization protocol could be feasibly scaled up, yield and enantioselectivity of product **4** on a gram scale were comparable to a smaller scale (Scheme 3A). Furthermore, the ester directing group of product **29** could be easily cleavage

via K_2CO_3 in MeOH,¹⁴ affording the enantioenriched γ -fluoroalkyl- β -aryl alcohol **56** (96% ee) without racemization (Scheme 3B). To further demonstrate the synthetic utility of this asymmetric method, several diversifications of alcohol **56** were performed (Scheme 3B). γ -Fluoroalkyl alcohol was easily converted to the corresponding alkyl chloride **57** (95% ee) and bromide **58** (96% ee),¹⁵ versatile intermediates in organic synthesis, in high yields without loss of enantioselectivity. Substitutions of **56** with phenyl disulfide and phthalimide afforded fluoroalkyl thiol **60** (95% ee) and fluoroalkyl amide **59** (97% ee) in 99% and 85% yields,¹⁶ respectively. It should be noted that the absolute configuration of compound **58** was assigned by X-ray diffraction analysis and the configuration of the other products was assigned by analogy (see SI for X-ray diffraction data for **58**).

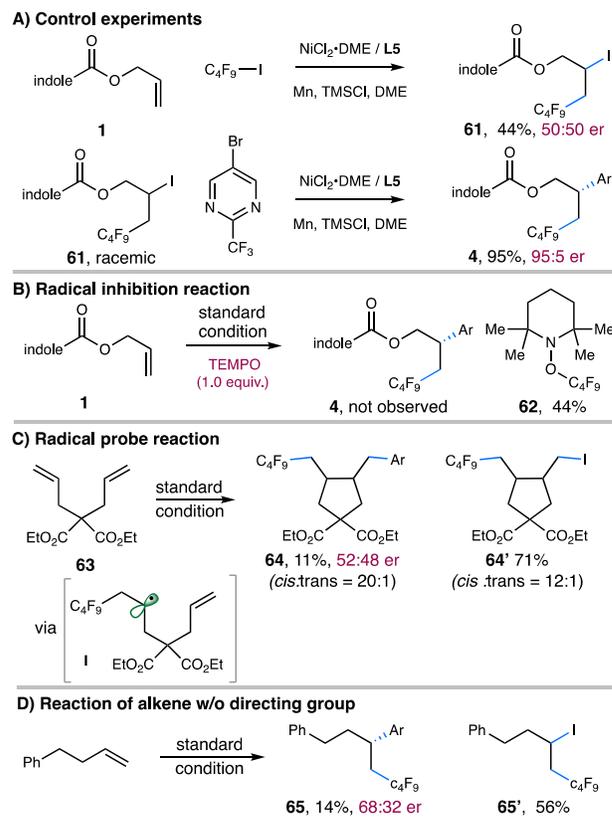
To shed some light on the plausible mechanism of this novel nickel-catalyzed enantioselective three-component

fluoroalkylarylation reaction, we have conducted several preliminary mechanistic experiments (Scheme 4). Since the iodoperfluoroalkylation byproduct was detected in our reactions,¹⁷ we performed a control reaction of alkene **1** with C₄F₉I in the absence of aryl bromide, which gave the racemic



Scheme 3. A) Large scale reaction; B) Cleavage of directing group and product transformations. a. CCl₄, PPh₃, CH₂Cl₂; b. CBr₄, PPh₃, CH₂Cl₂; c. PhSPh, PBu₃, THF; d. phthalimide, PPh₃, DIAD, THF.

alkyl iodide **61** in 44% yield (Scheme 4A). Two-component reaction of racemic **61** with aryl bromide **2** in the presence of Ni(II)/(*R,R*)-**L5** and Mn also furnished product **4** with excellent yield and enantioselectivity (Scheme 4A).¹⁸ Moreover, the time courses for this reaction disclosed that the conversion of alkene **1** into alkyl iodide **61** was achieved at the beginning of the reaction, and then the final product **4** gradually formed as the consumption of intermediate **61**, suggesting alkyl iodide was an on-cycle reactive intermediate in this catalytic system (Fig. S1). On the other hand, the desired reaction was completely inhibited by TEMPO, and TEMPO-C₄F₉ adduct **62** was observed via ¹⁹F NMR (Scheme 4B). The radical probe reaction of diene **63** with aryl bromide **2** and C₄F₉I resulted in the cyclized product **64** in 11% yield (*cis/trans* 20:1, 52:48 er)¹⁹ along with 71% yield of the cyclized alkyl iodide **64'** (*cis/trans* 12:1)²⁰ (Scheme 4C). These *cis/trans* ratios were consistent with the involvement of a radical intermediate (Scheme 4C).^{9a, 19, 21} To further diagnose the importance of chelation effect for this remarkable enantioselective control, we submitted 4-phenyl-1-butene, a non-activated alkene without a chelation group, into our standard condition, which only afforded the desired product **65** in 14% yield and 68:32 er, along with 56% yield of alkyl iodide **65'** (Scheme 4D). We anticipated that the weak chelating effect of the pendant indole carbonate group would be beneficial to the enantio-determining step of this



Scheme 4. Preliminary mechanistic studies.

transformation.²² Nevertheless, further studies are required to fully elucidate the possible reaction pathway.^{9f, 23}

In conclusion, we have developed the first enantioselective, three-component 1,2-fluoroalkylarylation of unactivated alkenes with aryl halides and fluoroalkyl iodides via a chelation-assisted Ni-catalyzed multicomponent cross-electrophile coupling. The benign protocol allows for the facile construction of a wide range of functionalized chiral β -fluoroalkyl arylalkanes with high efficiency and excellent enantioselectivity from readily available starting materials. The pendant ester group plays a crucial role in achieving high levels of enantioselectivity and efficiency in this three-component, asymmetric difunctionalization of unactivated alkenes. Moreover, the chelating group could be readily cleaved to give enantioenriched alcohols, further transformations of which generate a series of chiral fluoroalkyl-containing motifs that could be useful in the areas of pharmaceuticals and agrochemicals.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

- Experimental details, analytical data, and NMR spectra for all new compounds; synthetic applications; effect of reaction parameters; mechanistic studies (PDF)
- X-ray crystallographic details for Compound **58** (CIF)

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ACKNOWLEDGEMENT

We thank the National Natural Science Foundation of China (21991123, 21971036, 21702029), the “Thousand Plan” Youth program, and the Fundamental Research Funds for the Central Universities for financial support. We thank Ms. L. Wei and Prof. Y.-B. Zhang for beneficial discussion on X-ray determination of absolute configuration with support from Analytical Instrumentation Center (# SPST-AIC10112914) at ShanghaiTech University.

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