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Exploiting Pd^{II} and Ti^{III} Chemistry To Obtain γ-Dioxygenated Terpenoids: Synthesis of Rostratone and Novel Approaches to Aphidicolin and Pyripyropene A

José Justicia, J. Enrique Oltra,* and Juan M. Cuerva*

Department of Organic Chemistry, Faculty of Sciences, University of Granada, E-18071 Granada, Spain

jmcuerva@platon.ugr.es

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In nature there are several terpenoids with a characteristic γ -dioxygenated system on the A ring, and many of them show interesting pharmacological properties. We have developed a novel strategy for the synthesis of these terpenoids involving three stages: (a) the selective epoxidation of commercial polyenes, (b) titanium(III)-catalyzed cyclization of the epoxypolyprenes thus obtained, and (c) Pd-mediated remote functionalization of the equatorial methyl group attached at C-4 on ring A of the cyclic terpenoid thus formed. This strategy has proved to be useful for the synthesis of the natural labdane rostratone (1) and related terpenoids, as well as for advanced synthetic approaches toward the pharmacologically active products aphidicolin (2) and pyripyropene A (3).

Introduction

Some organisms belonging to the kingdoms of plants and fungi are capable of synthesizing small quantities of highly functionalized terpenoids bearing a characteristic γ -dioxygenated system on the A ring (Figure 1), as occurs in the labdane diterpenoid **1** (rostratone) found in the plant *Nolana rostrata*,¹ in the antibiotic aphidicolin (**2**) excreted by *Cephalosporium aphidicola*,² and in the meroterpenoid pyripyropene A (**3**) isolated from the fungus *Aspergillus fumigatus*.³ Many of these terpenoids possess interesting pharmacological properties. Aphidicolin, for example, shows marked activity against Herpes simplex,⁴ and pyripyropene A has proved to be a powerful inhibitor of acyl-CoA:cholesterol acyltransferase (ACAT),⁵ an enzyme related to atherosclerosis. These compounds have consequently attracted the attention of chemists, who have reported some procedures for synthesizing 2^6 and $3.^7$ These syntheses, however, generally require numerous steps, including tedious protection and deprotection protocols, and eventually provide only low overall yields.

 $^{^{*}}$ To whom correspondence should be addressed. Tel: +0034 958 248437. Fax: 0034 958 248090.

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FIGURE 1. γ -Dioxygenated system of the ring A present in several bioactive terpenoids.





 $^a\,$ Key: (a) selective epoxidation; (b) titanocene (III)-catalyzed cyclization; (c) oxidation and Pd-mediated remote functionalization.

A year ago we developed a novel method for the synthesis of complex terpenoids based on the titanocene-(III)-catalyzed⁸ radical cyclization of epoxypolyprenes (such as **5**) prepared from commercial polyenes.⁹

This method, which adheres to the principles of selectivity and atom- and step-economy required in contemporary chemistry,¹⁰ provides 3β -hydroxy terpenoids (such as **6**) with two "unactivated" methyl groups at C-4 (Scheme 1).⁹ With such derivatives in our hands we only needed to oxidize their equatorial methyl group to achieve the γ -dioxygenated system of compounds such as **7** (including **1**-**3**). The advantage of making this transformation by remote functionalization at the end of the synthetic sequence is that protecting groups would not be required during the building of the carbocyclic framework.

Organometallic chemistry affords mild and effective procedures to activate C–H bonds.¹¹ In particular, cyclopalladation reactions have proved to be capable of activating primary C–H bonds for the formation of C–I, C–O, and even C–C bonds.¹² Surprisingly, this kind of chemistry has been largely overlooked in the field of terpenoids.¹³ In a preliminary communication we described how Pd^{II}-mediated C–H activation via palladacycles can be used for the remote functionalization of C-4 methyl groups of terpenoids obtained by titanocene(III)-catalyzed radical cyclization of epoxypolyenes.¹⁴ Here we report on this procedure in full detail and its application to the synthesis of rostratone (1) and related labdanes and advanced approaches toward the pharmacologically active products aphidicolin (2) and pyripyropene A (3).

Results and Discussion

First we explored the efficiency of palladium-mediated C-H activation reactions over a wide range of terpenoid skeletons to facilitate the completion of the sequence depicted in Scheme 1. We began the process by preparing a set of model ketones (commercial **8b** and **9b**-**14b**) with different cyclic skeletons containing five-, six-, and sevenmembered rings (Figure 2) via the oxidation of alcohols **9a**-**14a** with Dess-Martin periodinane. These ketones were subsequently treated with hydroxylamine to form ketoximes **8c**-**14c** at yields ranging from 70% (**9c**) to 92% (**13c**).

Remote functionalization processes were then carried out as depicted in Scheme 2.¹⁵ Reactions between ketoximes **10c**-**14c** and sodium tetrachloropalladate(II)/ NaOAc gave palladacycle dimers **10f**-**14f** (in contrast to **8c** and **9c**, which decomposed after treatment with Na₂-PdCl₄), which were then treated with pyridine and lead tetraacetate to obtain acetoxy oximes **10d**-**14d** in yields ranging from 72% to 100% (Table 1).

To check the possibility of activating C–H bonds of terpenoid (pseudo)axial methyl groups, we made further reactions with sodium tetrachloropalladate(II), this time using acetoxy oximes 10d-14d as substrates. In this way we obtained moderate yields (42-55%) of acetates 15d and 16d (Figure 3) deriving from monocyclic oximes 10d and 11d, although the polycyclic substrates 12d-14d remained unchanged. Presumably the stereochemical rigidity of these compounds prevents a suitable conformation to produce the required palladacycles.

Finally, the hydrolysis of oximes **10d**-**16d** with TiCl₃/ H₂O avoided any undesirable isomerization of the double

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⁽¹⁵⁾ Reactions between ketoximes 10c-14c and sodium tetrachloropalladate(II)/NaOAc can be carried out either in MeOH or AcOH, but giving different yields. The best solvent for each substrate is that indicated in Table 1.





 $\begin{array}{l} \textbf{14a} \ R_1 = OH, H, \ R_2 = H \\ \textbf{14b} \ R_1 = O, \ R_2 = H \\ \textbf{14c} \ R_1 = NOH, \ R_2 = H \\ \textbf{14c} \ R_1 = NOH, \ R_2 = OAc \\ \textbf{14e} \ R_1 = O, \ R_2 = OAc \end{array}$

FIGURE 2. Chemical structure of the alcohols 9a-14a, the model ketones 8b-14b, the corresponding oximes, and the products derived from Pd-mediated remote functionalization.

SCHEME 2. Remote Functionalization of Terpenoids via Dimeric Organopalladium Complexes



 TABLE 1. Yields of Products Obtained by Pd-Mediated

 Remote Functionalization of Oximes 10c-14c, 10d, and

 11d

substrate	${ m solvent^{15}}$	acetoxy oxime (yield)	acetoxy ketone (yield)
10c	AcOH	10d (82%)	10e (82%)
11c	MeOH	11d (88%)	11e (85%)
12c	MeOH	12d (100%)	12e (85%)
13c	MeOH	13d (85%)	13e (85%)
14c	MeOH	14d (72%)	14e (85%)
10d	AcOH	15d (55%)	15e (83%)
11 d	MeOH	$\mathbf{16d}\ (42\%)$	$\mathbf{16e}~(85\%)$

bonds, thus providing good yields of acetoxy ketones 10e-16e (Table 1).

Once we were confident about the possibilities of the Pd-based method, we went on to try and synthesize the target molecule 1 (Scheme 3). As starting material we chose ethylene ketal 17, which was easily prepared from commercially available farnesylacetone.¹⁶ Bromonium-mediated epoxidation of 17, followed by the titanium(III)-catalyzed cyclization of epoxypolyene 18 under anhydrous



FIGURE 3. Chemical structure of products 15 and 16.





conditions, gave exocyclic alkene 19^{9a} (17% yield from farnesylacetone) with high degrees of regio- and stereoselectivity. Further hydrolysis of ketal 19 with CeCl₃/H₂O avoided undesired double-bond isomerization and provided an 85% yield of ketone 20.^{9a} Horner–Emmons olefination of 20 yielded 21a, the methyl ester of the natural labdane 21b found in both the Brazilian Copaiba tree and the Australian plant *Olearia teretifolia*.¹⁷ The chemical synthesis of carboxylic acid 21b has been achieved by Armstrong and Weiler through a 10-step sequence (4% overall yield) that ends with the saponification of ester 21a.¹⁸ Therefore, the titanocene(III)-based preparation of this ester (five steps, 11% overall yield) achieves a substantially improved formal synthesis of 21b.

Dess-Martin oxidation of **21a** led to ketone **22a** (80% yield). Spectroscopic data of this ketone matched those reported for the methyl ester of 3-oxoanticopalic acid (**22b**), found in the needles of *Pinus strobus*.¹⁹ To the best of our knowledge, this is the first total synthesis reported for **22a**, which confirms the structure proposed by Zinkel and Magee for the natural product **22b**.¹⁹

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FIGURE 4. Chemical structure of synthons 26 and 27.

SCHEME 4. Advanced Approach toward Synthon 26



SCHEME 5. Advanced Approach toward Synthon 27



Treatment of ketone **22a** with hydroxylamine led to the oxime **23** (60% yield from **20**) and finally the Pdmediated remote functionalization of **23** and hydrolysis of oxime **24** afforded acetoxy ketone **25** (53% yield from **23**). The natural metabolite (**1**) from *N. rostrata* has been isolated and characterized by Garbarino et al. as its acetate (**25**).¹ Spectroscopic data of synthetic **25** matched those of the acetate reported by Garbarino and coworkers,¹ supporting the structure proposed for the natural product **1**. These authors also describe the selective saponification of the acetate group of **25** to give **1**, and therefore the sequence depicted in Scheme 3 may be regarded as the formal synthesis of **1**. To the best of our knowledge this is the first chemical synthesis of this ketone, which we have called rostratone.

The synthesis of aphidicolin (2), reported by Corey et al.,²⁰ goes through an intermediate (26) closely related to the bicyclic synthon 27 employed by Nagamitsu et al.^{7a} for the synthesis of pyripyropene A (3). This observation prompted us to develop a novel, divergent strategy for the chemical preparation of both intermediates 26 and 27 (Figure 4) starting from isodrimenediol (28), which is readily prepared by titanocene(III)-catalyzed cyclization of 10,11-epoxyfarnesyl acetate.^{9a} From this diol the synthesis split into two branches, one directed toward 26 (Scheme 4) and the other toward 27 (Scheme 5). In the former, the exocyclic double bond of 28 remained after Jones oxidation and treatment with diazomethane, giving

keto ester **29** (60% yield), which was easily transformed into oxime **30**. The Pd-mediated remote functionalization of **30** gave acetate **31**, which was hydrolyzed to acetoxy ketone **32** (50% yield from **29**). Finally, the stereoselective reduction of ketone provided diol **33** (77%). This γ -dihydroxylated sesquiterpenoid contains an exocyclic double bond that might undergo ozonolysis to the corresponding ketone, which after suitable protection should provide **26**. The transformation of **26** into aphidicolin (**2**) was solved by Corey et al. employing the Robinson spiroannulation process.²⁰

In the synthetic branch toward pyripyropene A (Scheme 5), the exocyclic double bond was isomerized to a conjugated position after Dess-Martin oxidation of **28** and subsequent basic treatment, giving α,β -unsaturated aldehyde **34** (65%). Further oxidation of this aldehyde, followed by esterification of the corresponding acid, furnished **35**. The palladium-mediated remote functionalization of **35**, via oxime **36**, furnished acetate **37**, which was hydrolyzed to obtain **38** (32% from **34**). Finally, borohydride reduction of **38** and selective saponification of the acetate group provided **39** (50%). Dibenzylation of **39** would presumably give **27**, thus completing the formal synthesis of pyripyropene A (**3**).

In summary, the results described here prove that Pdmediated C-H activation is a suitable procedure for the remote functionalization of C-4 methyl groups of different terpenoid-like skeletons containing six- and sevenmembered A rings. This procedure might facilitate the chemical preparation of terpenoids with a γ -dioxygenated system on the A ring, and in fact it has proved itself useful for the synthesis of the natural terpenoid rostratone (1) and related labdanes (**21b**, **22a**) and for synthetic approaches to aphidicolin (**2**) and pyripyropene A (**3**). At the moment we are attempting to complete the synthesis of both **2** and **3** using shorter and more efficient synthetic sequences than those reported to date.

Experimental Section

General. For the reactions with titanocene all solvents and additives were thoroughly deoxygenated prior to use. Although all structures have been drawn as one enantiomer the synthesized compounds are racemic. Substances **9a**-**14a**,^{9a,21}**17**,¹⁶**20**,^{9a} and **28**^{9a} were prepared according to known procedures. The following known compounds were isolated as pure samples and showed NMR spectra identical to those of the reported compounds: **8c**,²²**11b**,²³**21**,^{17,18}**22**,¹⁹ and **25**.¹

General Procedure for the Synthesis of Ketones (9b– 14b) by Dess–Martin Oxidation. A solution of alcohol (9a– 14a) (1.0 mmol) and Dess–Martin periodinane (2.0 mmol) in wet CH_2Cl_2 (30 mL) was stirred at room temperature for 2 h. The mixture was diluted with *t*BuOMe, washed with 10% aqueous Na₂S₂O₃ and brine, and dried (Na₂SO₄), and the solvent was removed. The products were isolated by column chromatography of the residue on silica gel (hexane/tBuOMe) and characterized by spectroscopic techniques. The ketones obtained were isolated in the following yields: **9b** (70%), **10b** (88%), **11b** (76%), **12b** (94%), **13b** (72%), **14b** (91%).

Ketone 9b. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 5.01 (br s, 1H), 4.87 (br s, 1H), 4.15–4.05

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(m, 2H), 2.85–2.20 (m, 4H), 1.97 (s, 3H), 1.90–1.55 (m, 1H), 1.19 (s, 3H), 1.05 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 213.7 (C), 170.6 (C), 143.9 (C), 113.8 (CH₂), 63.6 (CH₂), 54.5 (CH), 37.7 (CH₂), 35.6 (C), 31.6 (CH₂), 27.1 (CH₃), 21.1 (CH₃), 20.8 (CH₃).²⁴

Ketone 10b. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 3.05 (dt, J = 14.6, 3.5 Hz, 1H), 2.59 (td, J = 14.5, 5.4 Hz, 1H), 2.49 (td, J = 14.5, 5.4 Hz, 1H), 2.03 (s, 3H), 1.66 (td, J = 14.6, 4.3 Hz, 1H), 1.52 (s, 3H), 1.49–1.44 (m, 1H), 1.10 (s, 3H), 1.08 (s, 3H), 1.05 (d, J = 3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 214.9 (C), 170.1 (C), 82.5 (C), 50.9 (CH), 48.1 (C), 34.3 (CH₂), 34.2 (CH₂), 26.9 (CH₃), 24.6 (CH₃), 24.4 (CH₃), 21.6 (CH₃), 9.1 (CH₃).²⁴

Ketone 12b. Hexane/tBuOMe, 7:3; white solid, mp 150–152 °C; ¹H NMR (400 MHz, CDCl₃) δ 4.95 (br s, 1H), 4.63 (br s, 1H), 4.34 (dd, J = 11.4, 4.1 Hz, 1H), 4.26 (dd, J = 11.4, 8.5 Hz, 1H), 2.66 (td, J = 12.9, 6.4 Hz, 1H), 2.50–2.38 (m, 2H), 2.15–2.05 (m, 3H), 2.04 (s, 3H), 1.88–1.48 (m, 4H), 1.12 (s, 3H), 1.05 (s, 3H), 0.96 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 216.1 (C), 171.3 (C), 145.5 (C), 108.4 (CH₂), 61.4 (CH₂), 55.1 (CH), 53.8 (CH), 47.9 (C), 38.6 (C), 37.5 (CH₂), 37.1 (CH₂), 34.6 (CH₂), 25.9 (CH₃), 24.6 (CH₂), 21.9 (CH₃), 21.1 (CH₃), 14.6 (CH₃); EIMS *m/z* 278 (3, M⁺), 263 (1), 218 (100), 203 (30), 175 (50), 133 (65), 93 (55); HRFABMS calcd for C₁₇H₂₆O₃Na *m/z* 301.1779, found *m/z* 301.1778.

Ketone 13b. (Hexane/tBuOMe, 4:1); vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 5.35 (t, J = 6.3 Hz, 1H), 2.68 (m, 1H), 2.27 (dt, J = 15.3, 4 Hz, 1H), 2.15–1.77 (m, 3H), 1.73 (s, 3H), 1.69–1.10 (m, 6H), 1.07 (s, 3H), 0.99 (s, 3H), 0.96 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 216.7 (C), 141.4 (C), 121.9 (CH), 59.5 (CH), 48.7 (C), 44.7 (CH₂), 40.9 (CH₂), 35.6 (C), 35.2 (CH₂), 34.3 (CH₂), 25.5 (CH₃), 25.4 (CH₃), 22.7 (CH₂), 21.7 (CH₃), 18.8 (CH₃); EIMS m/z 220 (25, M⁺), 205 (20), 152 (100), 137 (90), 97 (95), 67 (80).²⁴

Ketone 14b. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 4.86 (br s, 1H), 4.53 (br s, 1H), 4.33 (dd, J = 11.1, 3.6 Hz, 1H), 4.18 (dd, J = 11.1, 9 Hz, 1H), 2.56–2.39 (m, 4H), 2.01 (s, 3H), 2.00–1.09 (m, 11H), 1.08 (s, 3H), 1.02 (s, 3H), 0.90 (s, 3H), 0.78 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 217.6 (C), 171.4 (C), 146.2 (C), 107.6 (CH₂), 61.4 (CH₂), 58.9 (CH), 54.9 (CH), 54.9 (CH), 47.3 (C), 39.9 (CH₂), 39.4 (CH₂), 39.1 (C), 37.5 (CH₂), 37.3 (C), 34.1 (CH₂), 26.8 (CH₃), 23.3 (CH₂), 21.2 (CH₃), 21.0 (CH₃), 20.0 (CH₂), 16.2 (CH₃), 15.8 (CH₃); EIMS m/z 346 (5, M⁺), 331 (3), 303 (3), 286 (90), 218 (45), 205 (100), 163 (45), 121 (55), 93 (60); HREIMS [M⁺ – AcOH] calcd for C₂₀H₃₀O m/z 286.2296, found m/z 286.2304.

General Procedure for Oxime Synthesis: A mixture of ketone (8b–14b, 1 mmol), NH₂OH·HCl (2 mmol), and NaOAc (1 mmol) in MeOH (20 mL) was stirred at room temperature until the starting ketone was consumed. The solvent was then removed, and the residue was submitted to flash chromatography (hexane/tBuOMe), giving the corresponding oximes in the following yields: 8c (75%), 9c (70%), 10c (89%), 11c (80%), 12c (88%), 13c (92%), 14c (91%).

Oxime 9c. Hexane.*t*BuOMe, 4:1; white solid, mp 142–144 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.92 (br s, 1H), 4.77 (br s, 1H), 4.19 (dd, J = 11.4, 5.1 Hz, 1H), 4.03 (dd, J = 16.2, 6.3 Hz, 1H), 3.25 (dt, J = 14.4, 4.5 Hz, 2H), 2.50–2.00 (m, 3H), 1.97 (s, 3H), 1.15 (s, 3H), 1.14 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.0 (C), 163.2 (C), 144.5 (C), 113.1 (CH₂), 62.9 (CH₂), 54.4 (CH), 40.5 (C), 29.9 (CH₂), 28.3 (CH₃), 22.9 (CH₃), 20.9 (CH₃), 20.8 (CH₂); HRFABMS calcd for C₁₂H₁₉O₃NNa *m/z* 248.1263, found *m/z* 248.1261.

Oxime 10c. Hexane/tBuOMe, 4:1; white solid, mp 114–116 °C; ¹H NMR (300 MHz, CDCl₃) δ 9.50 (br s, 1H), 3.21 (dt, 1H), 2.93 (dt, 1H), 2.01 (s, 3H), 1.49 (s, 3H), 1.48–1.13 (m, 6H), 1.12 (s, 3H), 1.05 (d, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 170.2 (C), 165.9 (C), 83.3 (C), 51.0 (CH), 40.9 (C), 33.8 (CH₂), 25.6 (CH₃), 25.1 (CH₃), 22.5 (CH₃), 22.4 (CH₃), 17.0

(CH₂), 9.1 (CH₃); EIMS m/z 227 (1, M⁺), 212 (1), 167 (100), 153 (75), 111 (40); HRFABMS calcd for $C_{12}H_{21}O_3NNa$ m/z 250.1419, found m/z 250.1424.

Oxime 11c. Hexane/*t*BuOMe, 4:1; white solid, mp 109–111 °C; ¹H NMR (300 MHz, CDCl₃) δ 5.35 (t, J = 6.6 Hz, 1H), 2.73 (t, J = 6.6 Hz, 2H), 2.29 (t, J = 6.6 Hz, 2H), 2.10 (d, J = 6.9 Hz, 2H), 1.62 (s, 3H), 1.13 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 167.5 (C), 137.6 (C), 120.7 (CH), 43.3 (C), 38.6 (CH₂), 31.9 (CH₂), 27.1 (CH₃), 25.5 (CH₃), 21.4 (CH₂); EIMS *m/z* 167 (1, M⁺), 152 (40), 126 (25), 108 (100), 76 (55); HREIMS calcd for C₁₀H₁₇ON *m/z* 167.1310, found *m/z* 167.1314.

Oxime 12c. Hexane/tBuOMe, 4:1; white solid, mp 125–127 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.87 (br s, 1H), 4.54 (br s, 1H), 4.27 (dd, J = 11.4, 4.5 Hz, 1H), 4.20 (dd, J = 11.4, 7.8 Hz, 1H), 3.25 (dt, J = 18.3, 3.3 Hz, 2H), 2.44–2.38 (m, 1H), 1.99 (s, 3H), 1.92–1.16 (m, 7H), 1.14 (s, 3H), 1.03 (s, 3H), 0.87 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.0 (C), 166.2 (C), 146.0 (C), 108.0 (CH₂), 61.5 (CH₂), 55.2 (CH), 54.1 (CH), 40.8 (C), 38.9 (C), 37.2 (CH₂, two carbons), 26.9 (CH₃), 24.1 (CH₂), 23.3 (CH₃), 21.1 (CH₃), 17.6 (CH₂), 14.8 (CH₃); EIMS *m/z* 293 (8, M⁺), 276 (25), 234 (97), 216 (100), 166 (55), 124 (65), 91 (80), 79 (75); HRFABMS calcd for C₁₇H₂₇O₃NNa *m/z* 316.1888, found *m/z* 316.1889.

Oxime 13c. Hexane/*t*BuOMe, 4:1; white solid, mp 144–146 °C; ¹H NMR (300 MHz, CDCl₃) δ 9.53 (br s, 1H), 5.35 (t, J = 6.6 Hz, 1H), 3.20 (dt, J = 14.7, 3.9 Hz, 2H), 2.16–1.83 (m, 3H), 1.73 (s, 3H), 1.69–1.25 (m, 6H), 1.16 (s, 3H), 0.99 (s, 3H), 0.95 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 167.2 (C), 141.2 (C), 122.3 (CH), 59.9 (CH), 45.3 (CH₂), 41.7 (C), 40.9 (CH₂), 35.8 (C), 34.6 (CH₂), 26.4 (CH₃), 25.5 (CH₃), 23.2 (CH₃), 22.1 (CH₂), 19.0 (CH₃), 17.9 (CH₂); EIMS m/z 235 (40, M⁺), 218 (75), 192 (20), 152 (60), 134 (70), 99 (85), 67 (100); HRFABMS calcd for C₁₅H₂₅ONNa m/z 258.1833, found m/z 258.1831.

Oxime 14c. Hexane/tBuOMe, 4:1; white solid, mp 166–168 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.84 (br s, 1H), 4.51 (br s, 1H), 4.31 (dd, J = 11.1, 3.3 Hz, 1H), 4.17 (dd, J = 11.1, 9.3 Hz, 1H), 2.98 (dt, J = 15.3, 4.6 Hz, 2H), 2.41–2.21 (m, 4H), 2.00 (s, 3H), 1.90–1.17 (m, 9H), 1.14 (s, 3H), 1.04 (s, 3H), 0.90 (s, 3H), 0.76 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.5 (C), 166.9 (C), 146.4 (C), 107.4 (CH₂), 61.5 (CH₂), 59.2 (CH), 55.4 (CH), 55.0 (CH), 40.3 (C), 40.2 (CH₂), 39.4 (CH₂), 39.2 (C), 38.5 (CH₂), 37.6 (C), 37.5 (CH₂), 27.5 (CH₃), 23.1 (CH₂), 23.0 (CH₃), 21.2 (CH₃), 19.4 (CH₂), 16.1 (CH₃), 15.9 (CH₃); EIMS *m/z* 361 (1, M⁺), 331 (7), 286 (90), 271 (15), 243 (10), 205 (100), 187 (17), 163 (35), 121 (55), 93 (65); HREIMS calcd for C₂₂H₃₅O₃N *m/z* 361.2617, found *m/z* 361.2620.

General Procedure for Pd^{II}-Mediated Remote Functionalization. A mixture of oxime (10c-14c, 10d and 11d; 1 mmol), NaOAc (1.2 mmol), and Na₂PdCl₄ (1.2 mmol) in AcOH or MeOH (2-5 mL) (see Table 1) was stirred at room temperature for 48 h. The solvent was removed, and the residue diluted with CH2Cl2, filtered through a Celite pad, and concentrated. The residue obtained and Py (3.2 mmol) in THF (10 mL) were stirred at room temperature for 15 min. The reaction was then cooled to -78 °C, AcOH (66 mmol) and Pb-(OAc)₄ (1.1 mmol) were added, and the resulting mixture allowed to warm to room temperature and stirred for 24 h. Subsequently, tBuOMe was added, and the mixture was washed with saturated NaHCO3 and dried (anhydroud Na2-SO₄), and the solvent was removed. The residue was submitted to flash chromatography (hexane/tBuOMe) giving the corresponding acetoxy oximes in the following yields: 10d (82%), 11d (88%), 12d (100%), 13d (85%), 14d (72%), 15d (55%) and 16d (42%). Oximes 14d-16d were isolated contaminated with minor amounts of their hydrolysis products (14e-16e), so we decided to describe these products directly below.

Acetoxy Oxime 10d. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 4.27 (d, J = 11.4 Hz, 1H), 4.00 (d, J = 11.4 Hz, 1H), 3.17 (dt, J = 15, 3.7 Hz, 1H), 2.89 (dt, J = 14.5, 4 Hz, 1H), 2.02 (s, 3H), 2.01 (s, 3H), 1.98–1.75 (m, 3H), 1.53 (s, 3H), 1.09 (s, 3H), 1.01 (d, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 169.2 (C), 168.2 (C), 159.6 (C), 81.4

⁽²⁴⁾ We could not achieve good quality HRMS for ketones $\mathbf{9b},\,\mathbf{10b},\,$ and $\mathbf{13b}.$

(C), 64.0 (CH₂), 42.2 (CH), 41.7 (C), 31.0 (CH₂), 23.1 (CH₃), 20.9 (CH₃), 19.1 (CH₃), 17.0 (CH₃), 15.3 (CH₂), 6.6 (CH₃); EIMS m/z 285 (1, M⁺) 225 (35), 152 (100), 135 (55), 108 (30); HRFABMS calcd for C₁₄H₂₃O₅NNa m/z 308.1473, found m/z 308.1480.

Acetoxy Oxime 11d. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 5.31 (t, J = 6.3 Hz, 1H), 4.11 (d, J = 11 Hz, 1H), 3.96 (d, J = 11 Hz, 1H), 2.87–2.62 (m, 3H), 2.38 (dd, J = 15, 6 Hz, 1H), 2.29 (br t, 2H), 2.04 (s, 3H), 1.62 (s, 3H), 1.15 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.1 (C), 164.5 (C), 138.2 (C), 119.3 (CH), 69.8 (CH₂), 46.7 (C), 33.1 (CH₂), 322 (CH₂), 25.5 (CH₃), 22.3 (CH₃), 21.8 (CH₂), 20.1 (CH₃); EIMS m/z 225 (1, M⁺), 208 (1), 165 (50), 148 (100), 94 (35); HRFABMS calcd for C₁₂H₁₉O₃NNa m/z 248.1262, found m/z248.1266.

Acetoxy Oxime 12d. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 4.90 (br s, 1H), 4.58 (br s, 1H), 4.35–4.08 (m, 2H), 4.19 (d, J = 11.1 Hz, 1H), 4.03 (d, J = 11.1Hz, 1H), 3.09 (dt, J = 16.3, 3.1 Hz, 1H), 2.44–2.35 (m, 1H), 2.25–2.05 (m, 1H), 2.04 (s, 3H), 2.01 (s, 3H), 2.00–1.10 (m, 7H), 1.02 (s, 3H), 0.85 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 172.1 (C), 171.0 (C), 162.0 (C), 145.5 (C), 108.2 (CH₂), 67.4 (CH₂), 61.4 (CH₂), 53.8 (CH), 47.9 (CH), 43.2 (C), 38.5 (C), 37.1 (CH₂), 35.5 (CH₂), 24.3 (CH₂), 21.1 (CH₃), 21.0 (CH₃), 19.7 (CH₃), 18.1 (CH₂), 14.7 (CH₃); EIMS m/z 351 (1, M⁺), 336 (1), 317 (5), 276 (100), 264 (25), 216 (65), 201 (63), 173 (75), 133 (95), 91 (75); HRFABMS calcd for C₁₉H₂₉O₅NNa m/z 374.1943, found m/z 374.1941.

Acetoxy Oxime 13d. Hexane/MeOtBu, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 8.21 (br s, 1H), 5.35 (t, J = 6.4Hz, 1H), 4.17 (d, J = 11.1 Hz, 1H), 4.10 (d, J = 11.1 Hz, 1H), 3.06 (dt, J = 16, 3 Hz, 2H), 2.26–2.14 (m, 2H), 2.05 (s, 3H), 1.72 (s, 3H), 1.70–1.10 (m, 7H), 0.97 (s, 3H), 0.93 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.2 (C), 163.2 (C), 141.4 (C), 122.1 (CH), 66.7 (CH₂), 51.9 (CH), 45.0 (CH₂), 44.2 (C), 39.4 (CH₂), 35.3 (C), 34.1 (CH₂), 25.4 (CH₃), 22.1 (CH₂), 21.1 (CH₃), 19.5 (CH₃), 19.0 (CH₃), 18.4 (CH₂); EIMS *m/z* 293 (10, M⁺), 276 (5), 234 (15), 220 (65), 178 (18), 152 (98), 134 (60), 93 (55), 67 (70); HRFABMS calcd for C₁₇H₂₇O₃NNa *m/z* 316.1888, found *m/z* 316.1882.

General Procedure for Oxime Hydrolysis. A mixture of NH₄OAc (28 mmol), 20% HCl (0.5 mL), H₂O (15 mL), and TiCl₃ (5 mmol) was added to a solution of acetoxy oxime (10d–16d, 1 mmol) in THF (20 mL), and the resulting mixture was stirred at room temperature until the starting oxime was consumed. The reaction was then diluted with *t*BuOMe, washed with saturated NaHCO₃, and dried over anhydroud Na₂SO₄, and the solvent was removed. The residue was submitted to flash chromatography (hexane/*t*BuOMe, 4:1) giving the corresponding acetoxy ketones in the following yields: 10e (82%), 11e (85%), 12e (85%), 13e (85%), 14e (85%), 15e (83%), 16e (85%).

Acetoxy Ketone 10e. Hexane/tBuOMe, 4:1; white solid; mp 155–159 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.27 (d, J = 11.1 Hz, 1H), 3.87 (d, J = 11.1 Hz, 1H), 3.02 (ddd, J = 14.4, 5.5, 3.4 Hz, 1H), 2.49 (td, J = 15.0, 5.6 Hz, 1H), 2.27 (dt, J = 15.8, 4.0 Hz, 1H), 2.03 (s, 3H), 1.99 (s, 3H), 1.89 (q, J = 7.0 Hz, 1H), 1.74 (td, J = 14.3, 4.6 Hz, 1H), 1.58 (s, 3H), 1.07 (s, 3H), 1.04 (d, J = 7.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 211.4 (C), 170.6 (C), 170.0 (C), 82.7 (C), 65.9 (CH₂), 50.9 (C), 43.7 (CH), 34.5 (CH₂), 33.2 (CH₂), 24.6 (CH₃), 22.3 (CH₃), 20.1 (CH₃), 17.5 (CH₃), 8.9 (CH₃); EIMS *m/z* 270 (1, M⁺), 255 (1), 210 (5), 150 (15), 99 (100); HRFABMS calcd for C₁₄H₂₂O₅Na *m/z* 293.1364, found *m/z* 293.1359.

Acetoxy Ketone 11e. Hexane/tBuOMe, 4:1; colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.43 (br t, 1H), 4.11 (d, J = 10.8 Hz, 1H), 4.05 (d, J = 10.8 Hz, 1H), 2.80–2.75 (m, 1H), 2.75– 2.65 (m, 1H), 2.53 (dd, J = 15.8, 5.7 Hz, 1H), 2.30–2.25 (m, 2H), 2.10 (dd, J = 15.9, 7.3 Hz, 1H), 2.02 (s, 3H), 1.67 (s, 3H), 1.09 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 214.5 (C), 170.8 (C), 137.4 (C), 120.6 (CH), 69.5 (CH₂), 53.5 (C), 38.5 (CH₂), 32.2 (CH₂), 31.8 (CH₂), 25.3 (CH₃), 20.9 (CH₃), 20.6 (CH₃); EIMS m/z 210 (10, M⁺), 150 (50), 122 (52), 93 (100), 79 (35).³⁵

Acetoxy Ketone 12e. Hexane/tBuOMe, 4:1; vitreous solid; ¹H NMR (300 MHz, CDCl₃) δ 4.95 (br s, 1H), 4.64 (br s, 1H), 4.37 (dd, J = 11.5, 4.0 Hz, 1H), 4.24 (dd, J = 11.5, 8.5 Hz, 1H), 4.16 (d, J = 10.9 Hz, 1H), 3.95 (d, J = 10.9 Hz, 1H), 2.64– 2.43 (m, 3H), 2.03 (s, 6H), 1.85–1.15 (m, 7H), 0.98 (s, 3H), 0.95 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 212.7 (C), 171.2 (C), 170.7 (C), 145.1 (C), 108.8 (CH₂), 66.9 (CH₂), 61.3 (CH₂), 53.6 (CH), 50.6 (C), 47.6 (CH), 38.3 (C), 37.0 (CH₂), 35.8 (CH₂), 35.1 (CH₂), 24.4 (CH₂), 21.1 (CH₃), 20.9 (CH₃), 18.0 (CH₃), 15.0 (CH₃); EIMS *m*/z 336 (3, M⁺), 276 (100), 234 (20), 216 (65), 201 (60), 161 (63), 133 (75), 91 (73); HRFABMS calcd for C₁₉H₂₈O₅Na *m*/z 359.1834, found *m*/z 359.1834.

Acetoxy Ketone 13e. Hexane/tBuOMe, 4:1; white solid; mp 123–125 °C; ¹H NMR (300 MHz, CDCl₃) δ 5.38 (br t, J = 6.8 Hz, 1H), 4.16 (d, J = 10.9 Hz, 1H), 4.03 (d, J = 10.9 Hz, 1H), 2.65 (ddd, J = 16.6, 7.1, 4.0 Hz, 1H), 2.36 (ddd, J = 16.6, 3.0, 2.5 Hz, 1H), 2.02 (s, 3H), 2.10–1.25 (m, 9H), 1.74 (s, 3H), 1.02 (s, 3H), 0.96 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 213.6 (C), 170.8 (C), 141.5 (C), 121.8 (CH), 66.4 (CH₂), 51.6 (CH), 44.7 (CH₂), 39.6 (CH₂), 35.6 (CH₂), 35.1 (C), 34.0 (CH₂), 25.4 (CH₃), 22.4 (CH₂), 21.0 (CH₃), 18.8 (CH₃), 17.8 (CH₃), (one quaternary-carbon signal was not observed); EIMS *m/z* 278 (40, M⁺), 263 (1), 218 (20), 205 (25), 150 (55), 132 (70), 93 (100), 67 (97); HRFABMS calcd for C₁₇H₂₆O₃Na *m/z* 301.1779, found *m/z* 301.1777.

Acetoxy Ketone 14e. Hexane/tBuOMe, 4:1; colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 4.85 (br s, 1H), 4.52 (br s, 1H), 4.31 (dd, J = 11.1, 3.3 Hz, 1H), 4.17 (dd, J = 11.1, 9.3 Hz, 1H), 4.08–4.02 (m, 2H), 2.50–2.28 (m, 2H), 2.03 (s, 3H), 2.02 (s, 3H), 1.90–1.10 (m, 13H), 0.99 (s, 3H), 0.93 (s, 3H), 0.80 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 213.1 (C), 171.5 (C), 170.8 (C), 146.0 (C), 107.7 (CH₂), 67.7 (CH₂), 61.4 (CH₂), 58.5 (CH), 54.8 (CH), 50.1 (C), 47.9 (CH), 39.6 (CH₂), 39.1 (C), 37.9 (CH₂), 37.4 (C), 36.9 (CH₂), 35.1 (CH₂), 23.2 (CH₂), 21.2 (CH₃), 21.0 (CH₃), 19.9 (CH₂), 17.3 (CH₃), 15.9 (CH₃), 15.8 (CH₃); EIMS m/z 404 (8, M⁺), 344 (30), 302 (15), 284 (27), 217 (40), 203 (100), 161 (55), 133 (85), 93 (97); HREIMS calcd for C₂₄H₃₆O₅ m/z 404.2563, found m/z 404.2568.

Diacetoxy Ketone 15e. Hexane/tBuOMe, 4:1; colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.09 (d, J = 11.4 Hz, 1H), 4.67 (d, J = 11.4 Hz, 1H), 4.06 (d, J = 11.4 Hz, 1H), 3.96 (d, J = 11.4 Hz, 1H), 3.14 (dt, J = 14.6, 5.0 Hz, 1H), 2.76 (td, J = 15.0, 5.6, 1H), 2.25 (dt, J = 15.0, 4.0, 1H), 2.12 (s, 3H), 2.10– 1.90 (m, 1H), 2.02 (s, 3H), 1.98 (s, 3H), 1.74 (dt, J = 14.3, 4.6 Hz, 1H), 1.61 (s, 3H), 1.12 (d, J = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 207.4 (C), 170.7 (C), 170.6 (C), 169.9 (C), 82.4 (C), 62.9 (CH₂), 62.7 (CH₂), 54.8 (C), 44.8 (CH), 34.9 (CH₂), 33.8 (CH₂), 24.5 (CH₃), 22.4 (CH₃), 21.0 (CH₃), 20.8 (CH₃), 8.9 (CH₃); EIMS m/z 328 (1, M⁺), 268 (4), 208 (6), 195 (5), 166 (15), 148 (20), 99 (100); HRFABMS calcd for C₁₆H₂₄O₇Na m/z351.1419, found m/z 351.1427. The acidic hydrolysis of the oxime precursor produced minor amounts of a C-3 epimer.

Diacetoxy Ketone 16e. Hexane/tBuOMe, 4:1; colorless oil; ¹H NMR (300 MHz, CDCl₃) δ 5.41 (br t, 1H), 4.32 (d, J = 11.3 Hz, 2H), 4.10 (d, J = 11.3 Hz, 2H), 2.77 (t, J = 6.6 Hz, 2H), 2.37–2.30 (m, 4H), 2.03 (s, 6H), 1.69 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 211.4 (C), 170.7 (C), 138.8 (C), 119.3 (CH), 64.8 (CH₂), 56.6 (C), 38.6 (CH₂), 31.6 (CH₂), 27.3 (CH₂), 25.4 (CH₃), 20.9 (CH₃); EIMS *m*/*z* 268 (2, M⁺), 208 (25), 166 (8), 148 (65), 120 (25), 106 (100), 79 (40); HREIMS [M⁺ – AcOH] calcd for C₁₂H₁₆O₃ *m*/*z* 208.1099, found *m*/*z* 208.1104.

Synthesis of 21a. A mixture of NaH (61 mg, 2.6 mmol) and methyl diethylphosphonoacetate in THF (15 mL) was stirred at room temperature for 30 min. A solution of ketone 20 (74 mg, 0.25 mmol) in THF (5 mL) was added, and the mixture was stirred at 50 °C for 48 h. The mixture was then diluted with tBuOMe, washed with water, and dried over anhydrous Na₂SO₄, and the solvent was removed. The residue was submitted to flash chromatography (hexane/tBuOMe, 65: 35) giving methyl esther 21a (65 mg, 75%) as 9:1 mixture of

E/Z stereoisomers: vitreous solid; ¹H NMR (300 MHz, CDCl₃) (major isomer) δ 5.62 (br s, 1H), 4.84 (br s, 1H), 4.49 (br s, 1H), 3.66 (s, 3H), 3.22 (dd, J=11.5, 4.2 Hz, 1H), 2.40–2.20 (m, 3H), 2.13 (s, 3H), 2.00–1.80 (m, 4H), 1.75–1.05 (m, 7H), 0.97 (s, 3H), 0.75 (s, 3H), 0.66 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) (major isomer) δ 160.9 (C), 147.7 (C), 122.7 (C), 115.1 (CH), 106.8 (CH₂), 78.8 (CH), 55.9 (CH), 54.7 (CH₃), 50.8 (CH), 43.3 (C), 39.8 (CH₂), 39.8 (C), 38.2 (CH₂), 37.1 (CH₂), 28.4 (CH₃), 28.0 (CH₂), 24.1 (CH₂), 21.7 (CH₂), 18.9 (CH₃), 15.5 (CH₃), 14.5 (CH₃); EIMS *m*/z 334 (1), 319 (10), 301 (15), 260 (15), 203 (18), 175 (20), 135 (100), 114 (90), 82 (50); HRFABMS calcd for C₂₁H₃₄O₃Na *m*/z 357.2405, found *m*/z 357.2402.

Synthesis of 22a. Oxidation of alcohol **21a** with Dess–Martin periodinane (see general procedure described above) yielded ketone **22a** (80%) as a vitreous solid: hexane/tBuOMe, 4:1; ¹H NMR (300 MHz, CDCl₃) δ 5.63 (br s, 1H), 4.91 (br s, 1H), 4.56 (br s, 1H), 3.67 (s, 3H), 2.70–2.26 (m, 5H), 2.14 (s, 3H), 2.05–1.05 (m, 9H), 1.07 (s, 3H), 1.00 (s, 3H), 0.85 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 216.6 (C), 160.5 (C), 147.0 (C), 122.5 (C), 115.2 (CH), 107.7 (CH₂), 55.2 (CH), 55.2 (CH₃), 50.9 (CH), 47.8 (C), 39.7 (CH₂), 39.6 (CH₂), 39.4 (C), 37.9 (CH₂), 34.8 (CH₂), 26.2 (CH₃), 25.2 (CH₂), 22.1 (CH₂), 21.8 (CH₃), 19.0 (CH₃), 14.1 (CH₃).

Preparation of Oxime 23. Starting from ketone **22a** and following the general procedure described above for oxime synthesis, we obtained **23** (100%) as a vitreous solid: hexane/ MeOtBu, 85:15; ¹H NMR (300 MHz, CDCl₃) δ 5.62 (br s, 1H), 4.90 (br s, 1H), 4.53 (br s, 1H), 3.67 (s, 3H), 3.21 (dt, J = 14.7, 3.6 Hz, 2H), 2.43–2.20 (m, 2H), 2.13 (s, 3H), 2.08–1.17 (m, 10H), 1.14 (s, 3H), 1.02 (s, 3H), 0.79 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.7 (C), 160.8 (C), 147.4 (C), 122.5 (C), 115.2 (CH), 107.3 (CH₂), 35.5 (CH₃), 55.5 (CH₃), 50.9 (CH), 40.8 (C), 39.7 (CH₂), 39.6 (C), 37.9 (CH₂), 37.4 (CH₂), 27.1 (CH₃), 24.6 (CH₂), 23.2 (CH₃), 21.9 (CH₂), 19.0 (CH₃), 17.7 (CH₂), 14.2 (CH₃); EIMS *m/z* 347 (2), 332 (5), 317 (8), 281 (20), 258 (40), 207 (100), 174 (35), 159 (42), 121 (55), 95 (70), 55 (90); HRFABMS calcd for C₂₁H₃₃O₃NNa *m/z* 370.2358, found *m/z* 370.2360.

Synthesis of Acetoxy Oxime 24. Palladium-mediated remote functionalization of oxime 23 (see general procedure) gave acetoxy oxime 24 (76%) as a vitreous solid. NMR spectra indicated that, after flash chromatography, oxime 24 was contaminated by some proportion of its hydrolysis product (25). Therefore, this contaminated oxime was used in the next hydrolysis step without further purification.

Synthesis of Ketone 25. Standard oxime hydrolysis of compound **24** gave acetoxy ketone **25** (70%) as a vitreous solid. Spectroscopic data matched those reported in ref 1.

Preparation of Keto Ester 29. A sample of Jones' reagent (1 mL) was added to a solution of diol 28 (275 mg, 1.15 mmol) in acetone (35 mL) and stirred at room temperature for 1 h. The solvent was then removed, and the residue was dissolved in tBuOMe, washed with brine, dried over anhydrous Na₂SO₄, and concentrated. The residue was stirred with a saturated solution of CH_2N_2 in Et_2O (5 mL) for 30 min. The solvent was removed, and the residue was submitted to flash chromatograpy (hexane/tBuOMe, 7:3) giving 29 (187 mg, 62%) as a white solid: mp 119–121 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.87 (s, 1H), 4.67 (s, 1H), 3.62 (s, 3H), 2.68 (s, 1H), 2.66 (m, 1H), 2.43 (dt, J = 15, 3.3 Hz, 1H), 2.27 (ddq, J = 15, 3.3, 1.5 Hz, 1H), 2.04 (m, 1H), 1.77 (m, 1H), 1.65-1.44 (m, 4H), 1.22 (s, 3H), 1.05 (s, 3H), 1.02 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 215.5 (C), 171.6 (C), 142.6 (C), 109.6 (CH_2), 61.9 (CH), 54.6 (CH), 51.3 (CH₃), 47.8 (C), 38.7 (C), 37.3 (CH₂), 35.5 (CH₂), 34.6 (CH₂), 25.7 (CH₃), 23.9 (CH₂), 22.0 (CH₃), 13.7 (CH₃); EIMS m/z 264 (80), 232 (40), 189 (30), 147 (70), 123 (85), 91 (100); HRFABMS calcd for $C_{16}H_{24}O_3Na m/z$ 287.1623, found m/z 287.1623.

Preparation of Oxime 30. This oxime was obtained from ketone **29** following the general procedure for oxime synthesis (85% yield): white solid (hexane/tBuOMe, 1:1) mp 179–181 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.86 (s, 1H), 4.66 (s, 1H), 3.62 (s, 3H), 3.26 (dt, J = 15, 3.6 Hz, 1H), 2.75 (s, 1H), 2.43

Synthesis of Acetoxy Oxime 31. The palladium-mediated remote functionalization of oxime **30** (see general procedure) gave acetoxy oxime **31** (90%) as a vitreous solid: hexane/ tBuOMe, 7:3; ¹H NMR (300 MHz, CDCl₃) δ 4.86 (br s, 1H), 4.67 (br s, 1H), 4.12 (d, J = 11.2 Hz, 1H), 4.02 (d, J = 11.2 Hz, 1H), 3.64 (s, 3H), 3.09 (ddd, J = 16.5, 5.1, 2.6 Hz, 2H), 2.82 (s, 1H), 2.44–2.40 (m, 2H), 2.03 (s, 3H), 1.72–1.32 (m, 6H), 1.14 (s, 3H), 1.03 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.6 (C), 171.0 (C), 166.8 (C), 142.5 (C), 109.4 (CH₂), 67.4 (CH₂), 62.1 (CH₃), 51.1 (CH₃), 47.6 (CH), 43.1 (C), 38.6 (C), 35.7 (CH₂), 35.6 (CH₂), 23.6 (CH₂), 21.0 (CH₃), 19.8 (CH₃), 18.1 (CH₂), 13.9 (CH₃); EIMS *m/z* 337 (70), 321 (10), 277 (85), 260 (88), 218 (55), 186 (75), 134 (77), 91 (100), 79 (65); HRFABMS calcd for C₁₈H₂₇O₅NNa *m/z* 360.1786, found *m/z* 360.1787.

Synthesis of Acetoxy Ketone 32. Standard hydrolysis of oxime 31 gave acetoxy ketone 32 (65%) as a vitreous solid: hexane/tBuOMe, 7:3; ¹H NMR (300 MHz, CDCl₃) δ 4.91 (br s, 1H), 4.71 (br s, 1H), 4.11 (d, J = 10.8 Hz, 1H), 3.94 (d, J = 10.8 Hz, 1H), 3.67 (s, 3H), 2.88 (br s, 1H), 2.65–2.53 (m, 1H), 2.48–2.34 (m, 2H), 2.01 (s, 3H), 1.90–1.73 (m, 2H), 1.69–1.40 (m, 4H), 1.24 (s, 3H), 0.99 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 212.4 (C), 171.5 (C), 170.0 (C), 142.2 (C), 109.9 (CH₂), 69.9 (CH₂), 61.9 (CH₃), 51.2 (CH₃), 50.5 (C), 47.1 (CH), 38.3 (C), 35.7 (CH₂), 35.4 (CH₂), 35.0 (CH₂), 23.9 (CH₂), 20.9 (CH₃), 18.1 (CH₃), 13.8 (CH₃); EIMS *mlz* 322 (15, M⁺), 290 (10), 250 (45), 230 (80), 203 (55), 161 (65), 133 (100), 91 (98), 55 (70); HRFABMS calcd for C₁₈H₂₆O₅Na *m/z* 345.1677, found *m/z* 345.1674.

Obtention of y-Dihydroxylated Synthon 33. L-Selectride (0.32 mL, 0.32 mmol) was added to a solution of ketone 32 (9 mg, 0.032 mmol) in THF (3 mL) at -78 °C; the resulting mixture was stirred for 22 h, extracted with tBuOMe, washed with 2 N HCl, and dried over anhydrous Na₂SO₄; and the solvent was removed. The residue was submitted to flash chromatography (hexane/tBuOMe, 1:1) giving diol 33 (7 mg, 77%) as a white solid: mp 203-205 °C; ¹H NMR (300 MHz, $CDCl_3$) δ 4.84 (s, 1H), 4.66 (s, 1H), 3.70 (br t, J = 2.0 Hz, 1H), 3.65 (s, 3H), 3.48 (d, J = 11.4 Hz, 1H), 3.39 (d, J = 11.4 Hz, 1H), 2.96 (br s, 1H), 2.41 (ddd, J = 15.0, 3.5, 2.1 Hz, 1H), 2.22-1.80 (m, 8H), 1.10 (s, 3H), 0.70 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 177.1 (C), 143.6 (C), 108.7 (CH₂), 76.8 (CH), 71.2 (CH₂), 62.8 (CH), 51.1 (CH₃), 41.4 (CH), 40.6 (C), 38.8 (C), 36.0 $(CH_2), 31.3 (CH_2), 26.8 (CH_2), 22.7 (CH_2), 18.0 (CH_3), 14.5$ (CH₃); EIMS m/z 282 (10, M⁺), 251 (35), 235 (100), 191 (45), 173 (40), 147 (42), 91 (55); HRFABMS calcd for C₁₆H₂₆O₄Na m/z 305.1728, found m/z 305.1725.

Synthesis of Keto Aldehyde 34. Dess-Martin periodinane (1.20 g, 3.0 mmol) was added to a solution of diol 28 (237 mg, 0.99 mmol) in CH₂Cl₂ (25 mL), and the mixture was stirred at room temperature for 4 h. Then tBuOMe was added, and the solution was washed with a 1:1 mixture of aqueous 10% Na₂S₂O₃ and saturated NaHCO₃, and brine. The organic layer was dried (anhydrous Na₂SO₄), and the solvent was removed. The residue was dissolved in 0.5 M methanolic K₂CO₃ (20 mL) at 0 °C and stirred for 3 h. The mixture was then diluted with tBuOMe and washed with 2 N HCl and brine. The organic layer was dried (anhyd. Na₂SO₄), and the solvent was removed. Flash chromatography (hexane/tBuOMe, 7:3) of the residue provided keto aldehyde 34 (148 mg, 64%) as a vitreous solid: ¹H NMR (300 MHz, CDCl₃) δ 9.99 (s, 1H), 2.73 (q, J = 6.4 Hz, 1H), 2.43 (dd, J = 8, 7 Hz, 2H), 2.27 (dd, J = 8, 4.2 Hz, 2H), 2.04 (s, 3H), 1.68-1.40 (m, 3H), 1.32-1.20 (m, 1H), 1.12 (s, 3H), 1.03 (s, 3H), 0.99 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 217.3 (C), 191.6 (CH), 156.1 (C), 141.7 (C), 51.2 (CH), 47.0 (C), 38.1 (C), 36.4 (CH₂), 34.5 (CH₂), 34.1 (CH₂), 27.3 (CH₃), 20.8 (CH₃), 19.3 (CH₃), 19.2 (CH₂), 18.9 (CH₃); EIMS m/z 234 (100), 206 (40), 149 (50), 123 (80), 91 (85); HRFABMS calcd for C₁₅H₂₂O₂Na m/z 257.1517, found m/z 257.1521.

Synthesis of Keto Ester 35. A mixture of NaClO₂ (422 mg, 4.72 mmol) and NaH₂PO₄·H₂O (305 mg, 3.54 mmol) was added to a solution of 34 (138 mg, 0.59 mmol) in a mixture of tBuOH/H₂O/2-methyl-2-butene (4.5:1.6:1 ratio, 20 mL). The solution was stirred at room temperature for 20 h before being extracted with tBuOMe, washed with brine, dried over anhydrous Na₂SO₄, and concentrated. The residue was stirred with a saturated solution of CH_2N_2 in $Et_2O\ (5\ mL)$ for 30 min. The solvent was removed and the residue submitted to flash chromatograpy (hexane/MeOtBu, 3:1) giving **35** (90 mg, 60%) as a vitreous solid: ¹H NMR (300 MHz, CDCl₃) & 3.68 (s, 3H), 2.59-2.47 (m, 2H), 2.41 (t, J = 5.3 Hz, 1H), 2.09 (t, J = 6.8Hz, 4H), 1.77-1.60 (m, 2H), 1.61 (s, 3H), 1.24 (s, 3H), 1.06 (s, 3H), 1.02 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 216.4 (C), 170.3 (C), 136.2 (C), 134.0 (C), 51.2 (CH₃), 50.1 (CH), 47.2 (C), 38.2 (C), 34.9 (CH₂), 34.2 (CH₂), 32.0 (CH₂), 26.5 (CH₃), 22.2 (CH₃), 21.9 (CH₃), 19.5 (CH₃), 18.2 (CH₂); EIMS m/z 264 (45), 217 (55), 182 (70), 147 (75), 91 (100), 79 (70); HRFABMS calcd for C₁₆H₂₄O₃Na m/z 287.1623, found m/z 287.1626.

Preparation of Oxime 36. Starting from ketone **35** and following the general procedure described above, we obtained oxime **36** (85% yield) as a white solid: hexane/tBuOMe, 1:1; mp 151–153 °C; ¹H NMR (300 MHz, CDCl₃) δ 3.69 (s, 3H), 3.10 (ddd, J = 16.0, 5.3, 3.3 Hz, 1H), 2.27–2.05 (m, 4H), 1.60 (s, 3H), 1.55–1.30 (m, 4H), 1.26 (s, 3H), 1.15 (s, 3H), 1.07 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 170.5 (C), 165.9 (C), 137.1 (C), 133.5 (C), 51.2 (CH₃), 50.4 (CH), 40.1 (C), 36.6 (C), 34.6 (CH₂), 32.2 (CH₂), 27.49 (CH₃), 23.27 (CH₃), 20.93 (CH₃), 19.97 (CH₃), 19.06 (CH₂), 17.52 (CH₂); EIMS *m/z* 279 (1), 262 (35), 222 (15), 158 (50), 119 (60), 91 (100); HREIMS calcd for C₁₆H₂₅-NO₃ *m/z* 279.1912, found *m/z* 279.1912.

Synthesis of Acetoxy Oxime 37. The palladium-mediated remote functionalization of oxime **36** (see general procedure) gave acetoxy oxime **37** (83%) as a vitreous solid: hexane/ tBuOMe, 7:3; ¹H NMR (300 MHz, CDCl₃) δ 8.50 (br s, 1H), 4.07 (br s, 2H), 3.72 (s, 3H), 2.90 (dq, J = 17.5, 2.7 Hz, 2H), 2.47–2.35 (m, 1H), 2.12–2.02 (m, 2H), 2.01 (s, 3H), 1.80–1.67 (m, 2H), 1.66 (s, 3H), 1.22 (s, 3H), 1.08 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.1 (C), 170.5 (C), 161.9 (C), 136.6 (C), 133.9 (C), 68.6 (CH₂), 51.2 (CH₃), 43.6 (CH), 42.7 (C), 36.3 (C), 33.3 (CH₂), 32.0 (CH₂), 21.0 (CH₃), 20.8 (CH₃), 19.7 (CH₃), 19.5 (CH₃), 19.1 (CH₂), 18.6 (CH₂); EIMS *m*/z 337 (35), 306 (20), 278 (18), 260 (40), 215 (20), 189 (25), 153 (100), 121 (30), 91 (55); HRFABMS calcd for C₁₈H₂₇O₅NNa *m*/z 360.1786, found *m*/z 360.1785. **Synthesis of Acetoxy Ketone 38.** Standard hydrolysis of oxime **37** gave acetoxy ketone **38** (71%) as a vitreous solid: hexane/tBuOMe, 7:3; ¹H NMR (300 MHz, CDCl₃) δ 4.06–4.02 (m, 2H), 3.74 (s, 3H), 2.61–2.40 (m, 2H), 2.18–2.00 (m, 3H), 1.99 (s, 3H), 1.85–1.75 (m, 4H), 1.65 (s, 3H), 1.28 (s, 3H), 1.01 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 213.5 (C), 170.7 (C), 170.3 (C), 136.1 (C), 134.5 (C), 67.7 (CH₂), 51.3 (CH₃), 50.1 (C), 43.3 (CH), 36.1 (C), 35.2 (CH₂), 33.9 (CH₂), 31.8 (CH₂), 21.0 (CH₃), 20.9 (CH₃), 19.7 (CH₃), 19.3 (CH₂), 17.4 (CH₃); EIMS *m/z* 322 (50, M⁺), 291 (50), 262 (35), 215 (100), 187 (45), 175 (80), 147 (75), 91 (83); HRFABMS calcd for C₁₈H₂₆O₅Na *m/z* 345.1677, found *m/z* 345.1674.

Obtention of γ **-Dihydroxylated Synthon 39.** A sample of NaBH₄ (26 mg, 0.81 mmol) was added to a solution of 38 (26 mg, 0.081 mmol) in DME (5 mL) and stirred at 0 °C for 1 h. The mixture was then diluted with *t*BuOMe, washed with 2 N HCl, and dried (anhydrous Na₂SO₄), and the solvent was removed. The residue was dissolved in 0.5 M methanolic K₂-CO₃ (20 mL) at 0 °C and stirred for 2 h. The mixture was then diluted with *t*BuOMe and washed with 2 N HCl and brine. The organic layer was dried (anhydrous Na_2SO_4), and the solvent was removed. Flash chromatography (hexane/tBuOMe, 1:4) of the residue provided diol 39 (11 mg, 50%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.73 (d, J = 10.3 Hz, 1H), 3.71 (s. 3H), 3.68 (dd, J = 11.4, 4.6 Hz, 1H), 3.42 (d, J = 10.3Hz, 1H), 2.11-2.04 (m, 2H), 1.75-1.10 (m, 7H), 1.59 (s, 3H), 1.23 (s, 3H), 0.92 (s, 3H); $^{13}\mathrm{C}$ NMR (75 MHz, CDCl_3) δ 170.58 (C), 137.9 (C), 132.9 (C), 76.6 (CH), 71.8 (CH₂), 51.1 (CH₃), 44.5 (CH), 41.8 (C), 36.4 (C), 34.6 (CH₂), 31.9 (CH₂), 27.1 (CH₂), 20.9 (CH₃), 20.8 (CH₃), 18.4 (CH₂), 11.1 (CH₃); EIMS m/z 282 $(10, M^+), 251 (35), 235 (100), 191 (45), 173 (40), 147 (42), 91$ (55); HRFABMS calcd for $C_{16}H_{26}O_4Na m/z$ 305.1728, found m/z305.1725.

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Supporting Information Available: ¹H or ¹³C NMR spectra for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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