LUPENE TRITERPENOIDS FROM GLOCHIDION ERIOCARPUM*

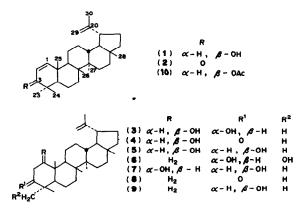
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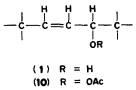
Key Word Index—Glochidion eriocarpum; Euphorbiaceae; lupene triterpenoids; glochidol [lupa-1,20(29)-dien- 3β -ol].

Previous investigations of various Glochidion species [1-7] have shown that this genus is characteristic in yielding lupene derivatives, which include six new compounds: glochidone [lupa-1,20(29)-dien-3-one] (2) [4] glochidiol [lup-20(29)-en-1 β ,3 α -diol] (3) [4], glochidonol [1 β -hydroxylup-20(29)-en-3-one] (4) [1], lup-20(29)-en-1 β ,3 β -diol (5) [2] [4], lup-20(29)-en-3 α ,23-diol (6) [3] and glochilocudiol [lup-20(29)-en-1 α ,3 β -diol] (7) [4] [5]. The present work is a study of the fifth, rather rare local species, *G. eriocarpum* Champ., which is also known as the hairy-fruited abacus plant.

Chromatography of the stems of G. eriocarpum gave lupenone (8), glochidone (2), lupeol (9), sitosterol, and a new compound glochidol (1), glochidonol (4), glochidiol (3) and lup-20(29)-en-1 β ,3 β -diol (5) respectively according to their order of elution from the column. Similar investigation of the leaves led to the isolation of sitosterol and (3).



Glochidol (1), $C_{30}H_{48}O$, gave positive reactions to both tetranitromethane and Liebermann-Burchard. (1) contained an OH group and formed an acetate (10). The PMR spectra of both (1) and (10) showed signals at $\delta 4.59$, 4.68 (1H each, m) and 1.68 (3H, broad s), characteristic of the isopropenyl group of lup-20(29)-ene derivatives. The partial structure



* Part 11 in the series "An Examination of the Euphorbiaceae of Hong Kong." For part 10, see Hui, W. H., Li, M. M. and Ng, K. K. (1975) *Phytochemistry* 14, 816.

was indicated by signals at $\delta 5.27$ (1H, d, J 10, 1.5) 5.83 (1H, d, J 10, 2.5) and 3.86 (1H, t, J 1.5 2.5 Hz) in the spectrum of (1) shifted to $\delta 5.18$, 5.89 and 5.07 respectively in that of (10). This partial structure can exist only in ring A. (1) was thus either lupa-1,20(29)-dien-3\xi-ol or lupa-2,20(29)-dien-1 ξ -ol, and it was proved to be the former by allyllic oxidation with MnO₂ in CHCl₃ to give a conjugated ketone, identical with (2). The configuration of the OH group at C-3 was proved to be β by the formation of (1) from (2) through $LiAlH_4$ reduction, which involved hydride attack from the less hindered α face, leading to the formation of the β -OH group. Further confirmation of this was achieved by catalytic hydrogenation of (10), when a saturated acetate $C_{32}H_{54}O_2$, identical with lupan-3 β -yl acetate prepared from lupenyl acetate, was obtained. The structure (1) for glochidol was thus confirmed.

The NMR proton signals of the six tertiary Me groups in the following lupene derivatives are assigned as in Table 1.

This investigation adds further evidence to our postulation [3] that *Glochidion* species are characteristic in yielding lupene derivatives. The number of new compounds isolated from this species is now seven and all those possessing two oxygen functions have these existing in a 1,3 relationship which might be another significant feature of this genus.

EXPERIMENTAL

IR spectra were recorded in KBr discs; PMR spectra in $CDCl_3$ were determined at 60 MHz using TMS as internal standard, UV spectra in 95% EtOH, and optical rotations in CHCl₃ solns. Petrol was bp 60–80°. Known compounds were identified by TLC, mmp, IR and MS comparisons with authentic samples.

Extraction and isolation of compounds. Milled air-dried stems of G. eriocarpum (14 kg) were extracted $2 \times$ at room temp.

Table 1. NMR Signals of tertiary methyl groups in Lupene derivatives

Compound	C-23	C-24	C-25	C-26	C-27	C-28
(1)	0.98	0.80	0.98	1.07	0.94	0-80
(10)	0.88	0.88	1.00	1.07	0.94	0-79
(2)	1.07	1.07	1.11	1.11	0.96	0.80
(4)	1.05	1.05	0-84	1.05	0-98	0.80
Acetate of (4)	1.07	1.04	0.89	1.07	0.97	0-80
(3)	0.90	0.82	0.90	1.04	0.97	0.78
Diacetate of (3)	0.84	0.90	1.04	1.04	0.98	0.79
(5)	0.96	0.74	0-91	1.06	0.96	0.79
Diacetate of (5)	0.85	0.85	1.03	1.03	0.92	0.79

with petrol for 7 days. Combined extracts were evaporated to give a residue (30 g), which was chromatographed on Al₂O₃ (600 g). Elution with petrol gave lupenone (30 mg), mp 170–171° $[\alpha]_D$ 56.0°; then glochidone (1 g), mp 169–170°, $[\alpha]_D + 69.0°$. Elution with petrol-C.H. (1:1) yielded first lupeol (10 mg), mp 199–202°, next sitosterol (2 g), mp 139–140°, and finally a solid, which on repeated recrystallisation from CHCl₃–MeOH gave needles of glochidol (1, 70 mg), mp 201–203°, $[\alpha]_D + 93.7°$ (Found: M⁺ 424. C₃₀H₄₈O requires: M⁺ 424) IR v_{max} : 3380 (OH), 3080, 1645, 885 cm⁻¹ (=CH₂); MS: m/e (%) 424 (100), 409 (70), 381 (8), 285 (32), 229 (83), 218 (28), 205 (37), 203 (46), 189 (57). Elution with C₆H₆ afforded glochidonol (25 mg), mp 268–269°, then lup-20(29)-en-1 β ,3 β -diol (30 mg), mp 250–251°, $[\alpha]_D + 28.0°$. The residue (60 g) of the petrol extract of leaves (55 kg) was chromatographed on alumina (1.2 kg). Elution with C₆H₆–CHCl₃ (1:1) glochidiol (10 mg).

Acetylation of (1). (1) (30 mg) was treated with Ac₂O and C₅H₅N at room temp. for 2 days. The product was recrystallized from petrol to give plates of glochidyl acetate (10, 30 mg), mp 207–208°, $[\alpha]_{\rm p}$ + 117.8° (Found: C, 82.6; H, 10.6. C₃₂H₅₀O₂ requires: C, 82.35; H, 10.8%), IR v_{max}: 1745, 1250 ÕAc), 3080, 1645, 880 cm⁻¹ (=CH₂), MS: m/e (%) 466 (27), 451 (13), 423 (11), 407 (100), 406 (95), 391 (29), 285 (35), 247 (12), 229 (86), 218 (17), 203 (57), 189 (60), 187 (37), NMR: δ 2.07 (3H, s, OCOCH₃).

 MnO_2 oxidation of (1) to glochidone (2). (1) (25 mg) in CHCl₃ (25 ml) was shaken with MnO_2 (30 mg) at room temp. for 3 days. The product was recrystallized from petrol to give prisms (18 mg), mp 169–171°, $[\alpha]_D + 68.0^\circ$, M⁺ 422, IR v_{max} : 1670, 1630 (C=C-C=O), 3050, 1650, 880 cm⁻¹ (=CH₂), UV λ_{max} : 229 (ϵ 9,850), identical with glochidone (2).

Reduction of (2) to (1). (2) (50 mg) in dry Et_2O (30 ml) was

refluxed with LiAlH₄ (50 mg) for 3 hr. The product (45 mg), mp 201-203° (from CHCl₃-MeOH), M⁺ 424, IR ν_{max} : 3380 (OH), 3080, 1645, 885 cm⁻¹ (=CH₂), was identical with (1) [acetate, mp 207-208°, identical with (10)].

Hydrogenation of (10) to lupan-3 β -yl acetate. (10) (25 mg) in CHCl₃ (20 ml) was shaken with Adam's catalyst in H₂ for 2 hr. The product (20 mg), mp 247-248°, $[\alpha]_D - 3 \cdot 0^\circ$, M⁺ 470, IR v_{max} : 1740, 1255 cm⁻¹ (OAc), NMR: δ 4.48 (1H, q, $J_{ax/eq}$ 8 Hz, $J_{ax/ax}$ 9 Hz, CHOAc), was identical with lupan-3 β -yl acetate prepared from lupenyl acetate by hydrogenation.

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ISOLATION OF TINGENONE AND PRISTIMERIN FROM MAYTENUS CHUCHUHUASCA

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Key Word Index-Maytenus chuchuhuasca; Celastraceae; tingenone, pristimerin; quinonoid triterpenes.

Plant and source: Maytenus chuchuhuasca Raymond Hamet (syn. Maytenus krukovii A. C. Smith (?) The plant was collected on the upper Rio Napo, Valley of Misahualli, Ecuador, and identified by one of us (A.P.) A voucher sample is deposited under the number T-10 in the Laboratory of Phytochemistry, Escuela Politecnica Nacional, Quito (Ecuador).

The plant is known locally as "chuchuhuaso" or "curicaspi" and is used in alcoholic solution for skin cancer treatment.

Previous work. This species was studied by Raymond-Hamet and Colas [1] who have described the isolation of an alkaloid maytenine, which was later demonstrated to be a dicinnamoyl spermidine [2]. No other substances were investigated.

Present work. In the course of our research on American species of Celastraceae [3] we examined a sample of Maytenus chuchuhuasca. The powdered bark of the trunk (100 g) was extracted at room temperature with a mixture (1/1) of hexane–EtOAc and gave on evaporation a residue (2.5 g), showing two orange pigments by TLC (C_6H_6 –EtOAc 9:1). The residue was chromatographed on silica gel. Elution with the above solvent led to isolation of pristimerin (1) (250 mg) and tingenone (2)(200 mg) identified by TLC IR and NMR in comparison with authentic samples.

Pristimerin and tingenone were previously isolated from various species belonging to the closely related families of Celastraceae and Hippocrateaceae. [4]. It is of interest to find high concentration of the two phenol