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During investigation of the properties of 1,2-hydroxyaminooximes (Ia, b) and their derivatives [1] we found that 1,2-(acyloxyamino)oximes (IIa-c), obtained by acylation of 1,2-hydroxyaminooximes (Ia, b), are converted into 4,5-dihydro-1,2,4-oxadiazoles (IIIa) (mp 78-79°C) and (IIIb) (mp 135-137°C) with yields of $\sim 50-70\%$ when treated with alkali in dioxane. The nitrogen atoms, which are in vicinal positions in compounds (IIa-c), become geminal in 4,5-dihydro-1,2,4-oxadiazoles (IIIa, b). From (IIc) the 1,2-hydroxyaminooxime (Ib) was obtained with a yield of $\sim 15\%$ in addition to compound (IIIb). Treatment of 4,5-dihydro-1,2,4-oxadiazoles (IIIa, b) with an aqueous solution of hydrochloric acid leads to the known acetamidooxime (IVa) and benzamidooxime (IVb) [2], which form the same 4,5-dihydro-1,2,4-oxadiazoles (IIIa, b) during acid-catalyzed condensation with acetone or its diethyl acetal [2]:

$$\begin{array}{c} R^{1}-C=NOH \\ CH_{3}-C-NHOH \\ CH_{3}-C-NHOH \\ CH_{3}-C-NHOCR^{2} \\ CH_{3} \\ (Ia, b) \\ (IIa-c) \\ R^{1}-C=NOH \\ R^{1}-C=NOH$$

 $R^1 = CH_3$ (a); $R^1 = C_6H_5$ (b, c); $R^2 = CH_3$ (IIa, b); $R^2 = CH_2CI$ (IIc).

The structure of 3,5,5-trimethyl-4,5-dihydro-1,2,4-oxadiazole (IIIa) was established by x-ray crystallographic analysis. The elemental analysis and IR, UV, and PMR spectra of compounds (II, III) are consistent with the given structure.

LITERATURE CITED

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