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PRAECOXIN B, C, D AND E, NOVEL ELLAGITANNINS FROM STACHYURUS PRAECOX

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Praecoxin C (6), D (7) and E (8), ellagitannins having depside linkage, and praecoxin B (9), an isomer of tellimagrandin I (1), were isolated from Stachyurus praecox.

KEYWORDS — Stachyurus praecox; Stachyuraceae; ellagitannin; valoneoyl group; depside linkage; C-glucosidic tannin; praecoxin B; praecoxin C; praecoxin D; praecoxin E

The leaves of Stachyurus praecox Sieb. et Zucc. (Japanese name: kibushi, Stachyuraceae) are rich tannin sources, and also have been used as diuretics in Japan. Recently, we found tannins and related compounds, namely casuarictin, pedunculagin, tellimagrandin I (1), strictinin, casuarinin (2), stachyurin, casuariin, $2,3-0-[(S)-hexahydroxydiphenoy1]-\underline{D}-glucose$ (3), rugosin C (4) and praecoxin A (5) in the leaf of S. praecox. $\frac{1-3}{2}$ We now report the isolation and the structure of three ellagitannins of novel type from the leaf, named praecoxin C (6), D (7) and E (8), and an additional new tannin named praecoxin B (9), and also the isolation of $1,2,6-tri-0-galloy1-\beta-\underline{D}-glucose$ and rugosin F (10), $\frac{4}{2}$ a dimeric ellagitannin.

These tannins have been isolated from the ethyl acetate extract of the crude extract of the leaf collected in August, by centrifugal countercurrent chromatography $^{5)}$ combined with the chromatography on Sephadex LH-20 and cellulose column.

Praecoxin B (9), $C_{34}H_{26}O_{22}\cdot 3H_{2}O$, $[\alpha]_{D}$ +49° (c=0.5, MeOH), UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ϵ) 210 (4.85) and 297 (4.34), was obtained as a light tan amorphous powder. The 1 H-NMR spectrum (200 MHz, in acetone- 1 G) shows that 9 forms an anomer mixture (α : β =1:1), and possesses two galloyl groups [δ 7.17 (s, 2H), 7.14 (s, 1H) and 7.11 (s, 1H)], a hexahydroxydiphenoyl (HHDP) group [δ 6.62 (s, 1H) and 6.39 (s, 1H)], and a glucose core [δ 5.66 (t, J=9.5 Hz, α -anomer, H-3), 5.52 (d, J=3.5 Hz, α , H-1), 5.51 (t, J=7 Hz, β , H-3), 5.40 (d, J=9 Hz, β , H-1), 5.21 (dd, J=7, 9 Hz, β , H-2), 5.10 (dd, J=3.5, 9.5 Hz, α , H-2), 4.90 (t, J=9.5 Hz, α , H-4), 4.90 (dd, J=7, 9 Hz, β , H-4), 4.61 - 4.44 (m, H-5 and H-6), 4.31 (d, J=12 Hz, β , H-6') and 4.29 (d, J=12 Hz, α , H-6')]. The HHDP group should be on 0-2 and 0-3, since partial hydrolysis of 9 with tannase yielded 3. The two galloyl groups then should be on 0-4 and 0-6 of the glucose moiety. Therefore, praecoxin B is 4,6-di-O-galloyl-2,3-O-[(S)-HHDP]-D-glucose (9), D0 an isomer of 1. Praecoxin C (6), D0, D1 HDP]-D1 glucose (9), D2 an isomer of 1.

209 (4.93) and 302 (4.47), was obtained as a light tan amorphous powder. Its 1 H-NMR spectrum (200 MHz, in acetone- 1 d) showed the presence of a galloyl group (δ 7.19, s, 2H), and a HHDP and a valoneoyl group [δ 7.21 (s, 1H), 6.96 (s, 1H), 6.60 (s, 1H), 6.46 (s, 1H) and 6.37 (s, 1H)]. Chemical shifts and coupling constants of glucose protons [δ 6.21 (d, J=9 Hz, H-1), 5.46 (dd, J=9, 10 Hz, H-3), 5.26 (t, J=10 Hz, H-4), 5.25 (dd, J=7, 13 Hz, H-6), 5.21 (t, J=9 Hz, H-2), 4.51 (dd, J=7, 10 Hz, H-5) and 3.97 (d, J=13 Hz, H-6')] which are similar to those of 4, 3) indicate the C1 conformation of β -D-glucopyranose.

The treatment of $\mathbf{6}$ with water (37°C, three days) afforded $\mathbf{4}$, quantitatively, and that with diazomethane in methanol gave heptadeca-0-methylrugosin C methyl ester, to show close correlation of $\mathbf{4}$ and $\mathbf{6}$.

When an aqueous solution of a mixture of 4 and 6 was extracted with ethyl acetate at pH 6.5, and then at pH 2.5, most of 6 was extracted at pH 6.5, while most of 4 was found in the extract obtained at pH 2.5. This difference in the behavior between $\mathbf{4}$ and $\mathbf{6}$ is similar to that between the tannins having a free carboxyl group such as mallotusinic acid⁷⁾ and chebulinic acid,⁸⁾ and the tannins These results show that the relationship of which lack a free carboxyl group. 4 to 6 is that the free carboxyl group of the valoneoyl group in 4 forms an ester The position of the depside linkage between the carboxyl group of the valoneoyl group and the hydroxyl group at C-10h of the HHDP group was indicated by the comparison of the $^{13}\text{C-NMR}$ spectra of **4** and **6** [δ 164.8 (C-19v in **6**, shifted 4.0 ppm higher from the corresponding carbon in 4), 152.6 (C-10h, >5.9 ppm lower), 133.7 (C-llh, >3.4 ppm higher), 122.9 (C-l4v, 4.4 ppm lower) and 108.1 The occurrence of an aromatic proton in the low field (C-15v, 1.3 ppm lower)]. (δ 6.96) assignable to H-9h in the H-NMR spectrum of **6** indicate that the hydroxyl group at C-10h, and not the group at C-12h, is linked to the valoneoyl group.

These facts and the molecular model show that the orientation of the valoneoyl group at 0-4-0-6 of glucose as in the structure $\bf 6$ allows the carboxyl group of the valoneoyl group to approach spatially to the HHDP group at 0-2-0-3 of glucose, in a way analogous to the phenazine moiety of the phenazine derivative (11) of isoterchebin.

The structure of praecoxin C, including the absolute configurations of the HHDP group and the valoneoyl group, is therefore assigned as 6. This is the first example of ellagitannin forming a depside by the valoneoyl group.

Praecoxin D (7), $C_{41}H_{26}O_{26} \cdot 5H_2O$, [α] $_D$ +81° (c=0.5, MeOH), UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ϵ) 206 (4.85), 235 (sh.) (4.77) and 293 (4.46), was obtained as a light tan amorphous powder. Its 1 H-NMR spectrum (200 MHz, in acetone-d₆) shows that 7 forms an anomer mixture (α : β =4:3), and possesses five aromatic protons [δ 7.19 (s, 3/7H), 7.17 (s, 4/7H), 6.97 (s, 1H), 6.63 (s, 3/7H), 6.62 (s, 4/7H), 6.61 (s, 4/7H), 6.58 (s, 3/7H) and 6.35 (s, 1H)]. Among them, an aromatic proton in the low field (δ 6.97) indicates the presence of a depside linkage in 7, which is analogous to that in 6.

The partial hydrolysis of 6 with tannase afforded 7, and prolonged treatment gave 3. The treatment of 7 with water yielded 5, in a way analogous to the hydrolysis of 6 to 4. Therefore, the structure 7 is assigned to praecoxin D.

These tannins having a depside linkage are not artefacts produced upon the treatment of the extract, since they were detected by HPLC carried out immediately after homogenizing the fresh plant. Concentration of the solutions and desiccation of 4 and 5 at various temperature did not produce 6 and 7.

Praecoxin E (8), $C_{48}H_{30}O_{30}\cdot 4H_{2}O$, $[\alpha]_{D}$ +17° (c=0.8, MeOH), UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ϵ) 205 (4.88) and 276 (4.34), was obtained as a light tan amorphous powder. The glucose protons show a pattern analogous to that of \mathcal{C} -glucosidic casuarinin (2) in the ^{1}H -NMR spectrum (200 MHz, in acetone- $d_{6}-D_{2}O$). The spectrum also shows the protons assignable to a galloyl group (δ 7.11, s, 2H) and to a valoneoyl group [δ 7.09, 6.87 and 6.54 (s, lH each)], and a proton in the low field (δ 7.00, s, lH), which is attributable to a \mathcal{C} -glucosidic HHDP group forming a depside linkage. The \mathcal{C} -glucosidic tannins can be regarded biogenetically as the products of a C-C bond formation between C-1 of glucose and a carbon in the aromatic ring of the tannins which have a glucopyranose ring. 1,2) Co-occurrence of several tannins of these two types in \mathcal{S} . praecox and other species of plant is also known. 2) Combined with the spectral data shown above, the structure of praecoxin E is accordingly assumed to be 8.

Two additional tannins were identified as 1,2,6-tri-0-galloyl- β -p-glucose and rugosin F (10)⁴⁾ which is one of the dimeric ellagitannins recently isolated from Rosa rugosa Thunb.

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REFERENCES

- T. Okuda, T. Yoshida, M. Ashida and K. Yazaki, Chem. Pharm. Bull., <u>30</u>, 766 (1982).
- 2) T. Okuda, T. Yoshida, T. Hatano, K. Yazaki and M. Ashida, Phytochemistry, in press.
- 3) T. Okuda, T. Hatano, K. Yazaki and N. Ogawa, Chem. Pharm. Bull., accepted.
- 4) T. Okuda, T. Hatano and N. Ogawa, Chem. Pharm. Bull., accepted.
- 5) W. Murayama, T. Kobayashi, Y. Kosuge, H. Yano, Y. Nunogaki and K. Nunogaki, J. Chromatogr., 239, 643 (1982).
- 6) This compound has been recently isolated from *Quercus stenophylla* independently: Y. Kashiwada, H. Nishimura, G. Nonaka and I. Nishioka, Abstract Paper of the 29th Annual Meeting of the Japanese Society of Pharmacognosy, p. 47 (Sapporo, 1982).
- 7) T. Okuda and K. Seno, Nippon Kagaku Kaishi, 1981, 671.
- 8) T. Yoshida, T. Okuda, T. Koga and N. Toh, Chem. Pharm. Bull., 30, 2655 (1982) and lit. cited therein.
- 9) T. Okuda, T. Hatano and T. Yasui, Heterocycles, 16, 1321 (1981).

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