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Metabolism-guided discovery of a potent and orally bioavailable urea-based calcimimetic for the treatment of secondary hyperparathyroidism



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ABSTRACT

A series of urea based calcimimetics was optimized for potency and oral bioavailability. Crucial to this process was overcoming the poor pharmacokinetic properties of lead thiazole **1**. Metabolism-guided modifications, characterized by the use of metabolite identification (ID) and measurement of time dependent inhibition (TDI) of CYP3A4, were essential to finding a compound suitable for oral dosing. Calcimimetic **18** exhibited excellent in vivo potency in a 5/6 nephrectomized rat model and cross-species pharmacokinetics.

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Secondary hyperparathyroidism (secondary HPT) is a condition characterized by the chronic elevation of parathyroid hormone (PTH) that often develops in patients with compromised kidney function.¹ The regulation of PTH levels is governed by the calcium-sensing receptor (CaSR), a class 3 G protein-coupled receptor (GPCR) primarily expressed on the parathyroid gland.² Increases in serum Ca²⁺ concentration raise the level of activation of the CaSR and inhibit PTH secretion, whereas decreases in serum Ca²⁺ concentration reduce CaSR activation and enhance PTH secretion.³ Left untreated, secondary HPT can lead to limb deformities as well as bone and joint pain. The discovery of positive allosteric modulators of the CaSR, type II calcimimetics, represents a novel therapy for the treatment of secondary HPT.⁴ Such agents increase the sensitivity of the CaSR to serum Ca²⁺ and thereby decrease secretion of PTH.

In a groundbreaking publication, Nemeth and co-workers described the in vitro and in vivo pharmacology of the first type

II calcimimetic with the disclosure of R-568 and related analogues.⁵ Subsequent work to identify new calcimimetic agents has focused almost entirely on derivatives that retain a basic α -methylbenzyl amine moiety.⁶ A novel series of benzothiazole urea based calcimimetics arising from R-568 and fendiline was recently disclosed.⁷ Further development of this series led to **1**.⁸ In vivo studies with **1**, using a 5/6 nephrectomized rat model of kidney failure wherein

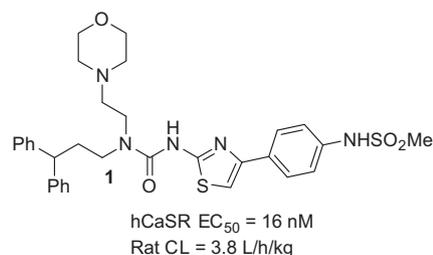


Figure 1. Thiazole **1**, a urea-based calcimimetic.

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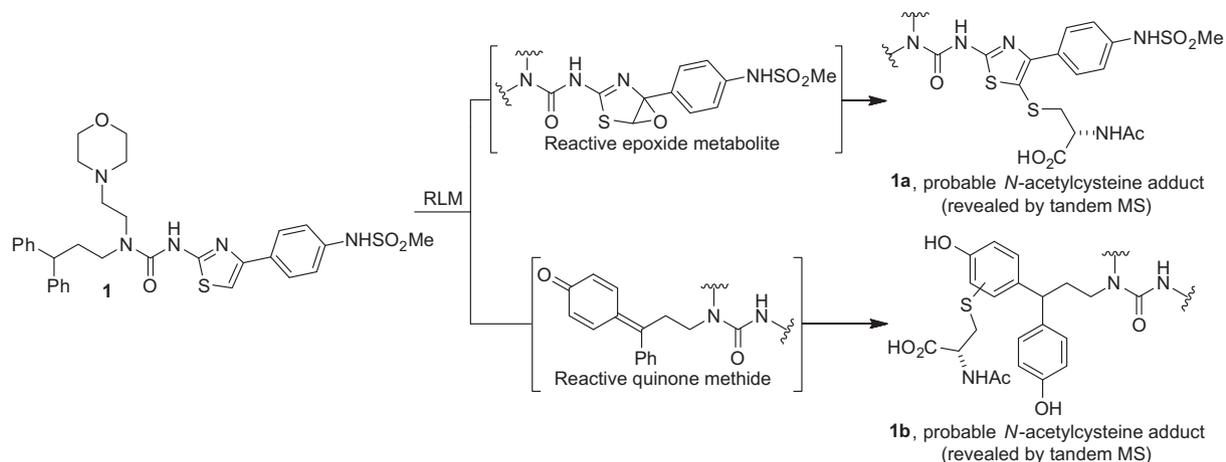
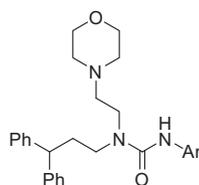


Figure 2. Metabolites of **1** formed from incubation with rat liver microsomes (RLM) and elucidated by tandem mass spectrometry.

Table 1
Heterocyclic replacements of thiazole



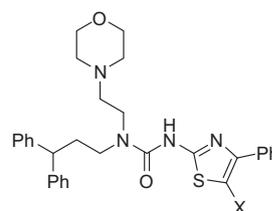
Compound	Ar	hCaSR EC ₅₀ (nM) ^a
2		300
3		1030
4		286
5		175
6		23

^a Values are an average of at least two determinations.

one kidney and 2/3 of the second kidney have been surgically removed, showed promise demonstrating a modest reduction in serum PTH levels. Despite these positive features, **1** exhibited substantial liabilities including poor pharmacokinetics (PK). This account describes our systematic efforts to address the metabolism of **1** (Fig. 1).

To understand the high clearance of **1**, a series of metabolite ID studies was performed. Incubation of **1** in the presence of rat liver microsomes (RLM) indicated sites of oxidation at the thiazole, morpholine, and *gem*-diphenyl moieties. Further studies to elaborate the specific nature of the oxidations were performed by incubating **1** in RLM containing *N*-acetylcysteine (Fig. 2). These studies suggested the intermediacy of a reactive epoxide in the formation of metabolite **1a** and implicated oxidation of the phenyl ring to a quinone methide in the formation of metabolite **1b**. The intermediacy of specific oxidations in the formation of metabolites **1a** and **1b** suggested a potential strategy to improve their stability.

Table 2
Impact of electron withdrawing groups on TDI potential

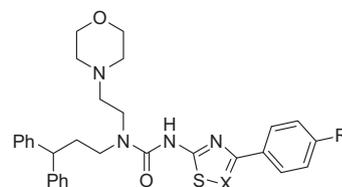


Compound	X	%3A4 Activity Remaining ^a	hCaSR EC ₅₀ (nM) ^b
7	H	22	36
8	F	28	53
9	SO ₂ Me	59	17
10	CN	76	157
11	Cl	91	18

^a CYP3A4 experiments were performed as previously described Ref. 11

^b Values are an average of at least two determinations.

Table 3
Chlorothiazole and thiadiazole analogues of **1**



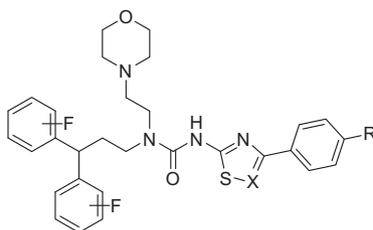
Compound	X	R	hCaSR EC ₅₀ (nM) ^a	PK Data ^b CL (L/h/kg)/%F
12	CCl	NHSO ₂ Me	3	1.4/37
13	CCl	SO ₂ NH ₂	2	0.42/16
14	N	NHSO ₂ Me	6	4.0/23
15	N	SO ₂ NH ₂	3	3.8/23

^a Values are an average of at least two determinations.

^b Pharmacokinetic studies were conducted in male Sprague–Dawley rats and were administered at 0.5 mg/kg IV and 2 mg/kg PO.

Numerous reports indicate thiazoles as elements of oxidative susceptibility.⁹ Our strategy to prevent epoxidation was to replace the thiazole nucleus with other heterocycles lacking the potential for epoxidation (Table 1). Despite considerable effort, only 1,2,4-thiadiazole **6** (hCaSR EC₅₀ = 23 nM)¹⁰ was identified as a competent bioisostere of **1**.

Table 4
Stabilization of the diphenyl motif by fluorination



Compound	F	X	R	hCaSR EC ₅₀ (nM) ^a	PK Data ^b CL (L/h/kg)/%F
16	2-F	CCl	SO ₂ NH ₂	4	2.3/20
17	3-F	CCl	SO ₂ NH ₂	19	—
18	4-F	CCl	SO ₂ NH ₂	10	0.32/36
19	4-F	CCl	NHSO ₂ Me	9	0.81/1.0
20	4-F	N	SO ₂ NH ₂	9	1.6/39
21	4-F	N	NHSO ₂ Me	20	1.3/48

^a Values are an average of at least two determinations.

^b Pharmacokinetic studies were conducted in male Sprague–Dawley rats and were administered at 0.5 mg/kg IV and 2 mg/kg PO.

Table 5
PK summary for compound **18**

Species	CL (L/h/kg)	V _{ss} (L/kg)	MRT IV (h)	t _{1/2} (h)	%F
Rat	0.32	2.0	6.3	7.0	36
Nonhuman Primate	0.024	0.21	8.9	8.0	15
Beagle Dog	0.098	1.2	13	11	47

An alternate strategy to reduce the oxidative susceptibility of the thiazole in **1** was to modulate reactivity by substitution at the 5-position of the thiazole. We measured time-dependent inactivation (TDI) of CYP3A4¹¹ as an estimate of reactive metabolite

formation. Using **7**⁸ as a baseline, substitution with –F, –SO₂Me, –CN, and –Cl was explored (Table 2 and **8–11**). Reduced TDI potential was observed upon substitution with chloro analog **11** proving most effective. Fluorine did not significantly decrease bioactivation of the thiazole ring (**7** vs **8**).

The chloro substituted thiazole analogue **12**⁸ demonstrated a nearly 3-fold improvement in clearance relative to **1** (Table 3). Interestingly, the reverse sulfonamide analogue **13**⁸ resulted in a further reduction in clearance. Unfortunately, both thiazole analogues **14** and **15** were rapidly cleared in vivo. Further improvements to PK were realized by modification of the *gem*-diphenyl group.

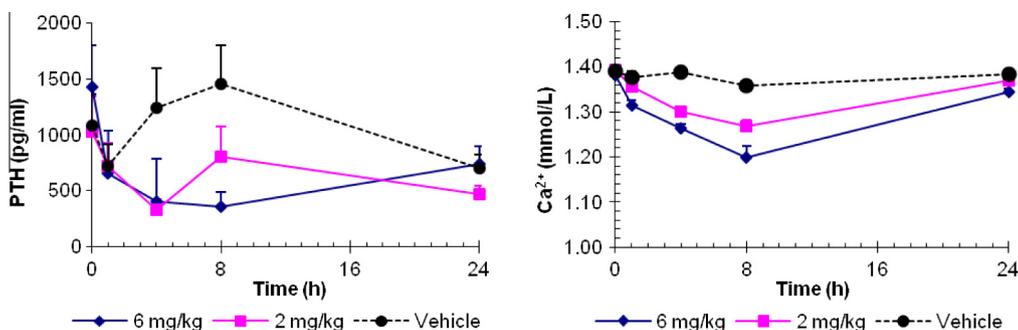
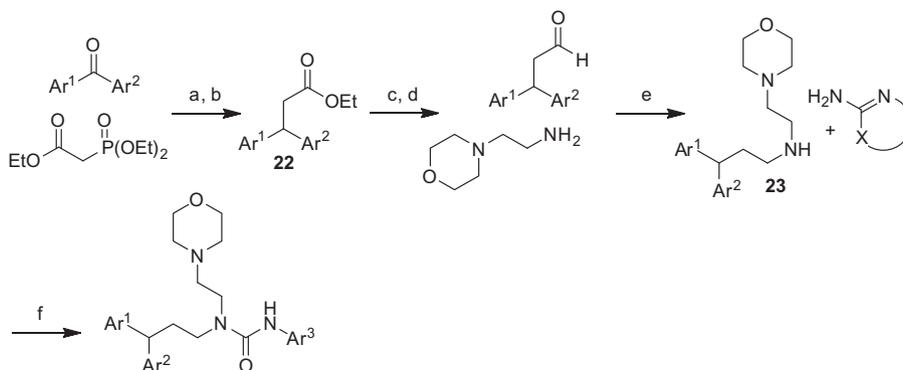


Figure 3. Sustained reduction of PTH and ionized calcium after oral administration of **18** to 5/6 nephrectomized rats.



Scheme 1. (a) NaH, PhMe, 0 °C for 1 h, then 50 °C, 3 d; (b) Pd/C, 1 atm H₂, EtOH, 25 °C, 16 h; (c) LiAlH₄, THF/PhMe, –78 °C to 0 °C; (d) SO₃-pyridine, Et₃N, DMSO, 25 °C; (e) NaBH(OAc)₃, ClCH₂CH₂Cl, 25 °C; (f) Ar³NH₂, DMAP, CDI, 40 °C, 2.5 d.

Metabolite ID studies with *N*-acetylcysteine had indicated the formation of covalent adducts at the phenyl rings suggesting the intermediacy of a quinone methide (Fig. 2, 1b). As with the thiazole, the installation of electron withdrawing groups on the unsubstituted phenyl ring was hypothesized to reduce metabolism. Efforts focused on the incorporation of electron withdrawing elements. As such, a series of fluorinated derivatives were prepared (Table 4 and 16–21). *Ortho*-fluorination resulted in a 2-fold decrease in potency relative to 13. *Meta*- and *para*-fluorination compromised activity by 10- and 5-fold, respectively, relative to 13. Unfortunately, the *ortho*-fluorinated derivative 16 was cleared more rapidly in rats than the parent analogue 13.¹³ In all cases, *para*-fluorination had lower clearance than their unsubstituted congeners, especially in the case of the thiadiazoles 20 and 21.

Additional pharmacokinetic profiling of 18 in higher species confirmed the good cross-species pharmacokinetics (Table 5). These results demonstrate the pharmacokinetic improvements realized by a metabolism-guided approach to analogue design. A screen of off-target activities using a functional GPCR platform at CEREP indicated that 18 showed no confounding off-target activity.¹⁴ Moreover, 18 exhibited robust lowering of PTH and ionized calcium (Ca²⁺), relative to vehicle, in a 5/6 nephrectomized rat model of kidney failure when dosed orally at 6 mg/kg (Fig. 3).¹⁵

The modular route by which analogues were prepared is described in Scheme 1. The *gem*-diphenyl ester 22 formed from the corresponding benzophenone via a Horner–Wadsworth–Emmons reaction. Amine 23 was formed in three steps from 22 by a reduction–oxidation–reductive amination sequence. The urea functionality was formed by coupling ArNH₂ with carbonyldiimidazole (CDI) in the presence of 4-dimethylaminopyridine (DMAP).

In summary, we identified 18, a novel calcimimetic with the potential to lower elevated parathyroid hormone levels associated with secondary hyperparathyroidism. A metabolism guided approach to analogue design, highlighted by the use of metabolite ID experiments and measurement of TDI of CYP3A4, addressed PK issues and enabled the identification of 18. Most consequential was the use of thiazole chlorination in combination with fluorination of the *gem*-diphenyl moiety. Lastly, 1,2,4-thiadiazoles appear to be equipotent replacements of the chlorothiazole although the chlorothiazoles had lower rat IV clearance.

Acknowledgments

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- Compound 18 was tested in a CEREP panel of 34 enzymes, 19 receptors, 5 transporters, and 5 channels at 10 μM. Three assays with POC < 50 (adenosine transporter, guanylyl cyclase, and carbonic anhydrase II) were not expected to interfere with the observed in vivo pharmacology. Solubility data for 18: 0.01 N HCl = 0.017 mg/mL, Simulated Intestinal Fluid (SIF) = 0.040 mg/mL, Phosphate Buffered Saline (PBS) = 0.001 mg/mL.
- Compound 18 was dosed orally to 5/6 nephrectomized male Sprague–Dawley rats as a suspension in 2% HPMC/1% Pluronic F68/5% Captisol in water. A statistically significant (repeated measures ANOVA, *p* < 0.05) decrease in PTH relative to vehicle was measured at 4 h for both doses and at 8 h for the 6 mg/kg dose. The decrease in blood ionized calcium was statistically significant (repeated measures ANOVA, *p* < 0.05) relative to vehicle at 1, 4, and 8 h for the 6 mg/kg dose and 4 and 8 h for the 2 mg/kg dose.