

THE SYNTHESIS OF 2-ACETAMIDO-2-DEOXY-6-*O*- $\alpha$ -D-MANNO-PYRANOSYL-D-GLUCOSE\*

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## ABSTRACT

Condensation of tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl bromide with either benzyl 2-acetamido-3-*O*-acetyl- or 3,4-di-*O*-acetyl-2-deoxy- $\alpha$ -D-glucopyranoside (obtained via the 6-*O*-trityl- and 3,4-di-*O*-acetyl-6-*O*-trityl derivatives) gave benzyl 2-acetamido-2-deoxy-6-*O*- $\alpha$ -D-mannopyranosyl- $\alpha$ -D-glucopyranoside penta- and hexa-acetate in 42% and 65% yield, respectively. Removal of the protective *O*-acetyl and *O*-benzyl groups gave the title compound, which was characterized by a hepta-*O*-acetyl derivative. All intermediates were obtained in crystalline form. The title compound is useful as a reference standard for determination of the structure of the carbohydrate core of glycoproteins.

## INTRODUCTION

The present paper describes the synthesis and characterization of 2-acetamido-2-deoxy-6-*O*- $\alpha$ -D-mannopyranosyl-D-glucose (**9**) as part of a program to synthesize fragments of carbohydrate chains of glycoproteins and glycolipids<sup>1-7</sup>.

## DISCUSSION

In order to obtain the disaccharide **9**, two routes were studied. In the first, the known benzyl 2-acetamido-3-*O*-acetyl-2-deoxy- $\alpha$ -D-glucopyranoside<sup>8</sup> (**5**) was condensed with tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl bromide<sup>9</sup> (**1**), to give benzyl 2-acetamido-3-*O*-acetyl-2-deoxy-6-*O*-(tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl)- $\alpha$ -D-glucopyranoside (**7**) in 42% yield; other possible condensation products, such as the  $\beta$  anomer condensed with the hydroxyl group on C-4 of **5**, or the two possible  $\beta$ -D anomers, were not observed.

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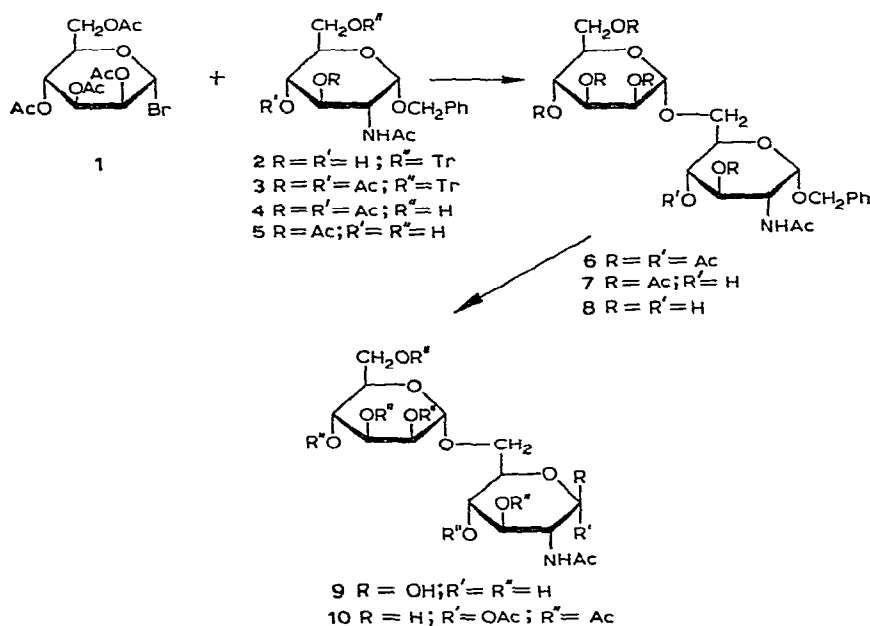
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In a second study, tritylation of benzyl 2-acetamido-2-deoxy- $\alpha$ -D-glucopyranoside<sup>10</sup> gave the 6-*O*-trityl derivative **2**, which was acetylated to give **3**. Compound **3** was detritylated to give the 3,4-di-*O*-acetyl derivative **4**, and condensation of **4** with bromide **1** gave the fully acetylated disaccharide **6** in 65% yield. Saponification of the *O*-acetyl groups of both **6** and **7** gave **8**, which was hydrolyzed to afford the crystalline disaccharide **9**. All intermediates were obtained in crystalline form, and all reactions, except the condensation to form the disaccharides, resulted in excellent yields (over 80%).

The  $\alpha$ -D configuration of the (1 $\rightarrow$ 6)-linkage is clearly indicated by the comparison of the optical rotation of **6**, **7**, and **9** with that of various  $\alpha$ -D-(1 $\rightarrow$ 3) or -(1 $\rightarrow$ 4) analogs<sup>6,7</sup> or with the sum of the optical rotations of the constituents, as described earlier<sup>2-7</sup> (see Table I).

The great difference between the optical rotation of the fully acetylated disaccharide **6** and that of the analog **7** (having a free hydroxyl group at C-4 of the 2-acetamido-2-deoxy-D-glucose moiety) is of interest for an evaluation of the contribution of each asymmetric center to the total optical activity (see Lemieux and Martin<sup>11</sup>, Brewster<sup>12</sup>, and Whiffen<sup>13</sup>). Very little change in the optical rotation was observed when the disaccharide **9** was dissolved in aqueous methanol. Comparison of its molecular rotation with the sum of those of the constituents (see Table I) suggests a  $\beta$ -D configuration for the hexosamine moiety of **9**, but an  $\alpha$ -D configuration of this moiety in the crystalline peracetate **10**.



It is of interest that no formation of a  $\beta$ -D-linked disaccharide was observed in the synthesis of the (1 $\rightarrow$ 3)- (ref. **6**), (1 $\rightarrow$ 4)- (ref. **7**), and (1 $\rightarrow$ 6)-linked 2-acetamido-2-

deoxy-*O*- $\alpha$ -D-mannopyranosyl-D-glucose, despite the use of mercuric cyanide as the catalyst and acid acceptor known to favor the formation of both the  $\alpha$ -D- and  $\beta$ -D-linked disaccharides (see Staněk *et al.*<sup>14</sup> and Evans *et al.*<sup>15</sup>).

TABLE I

MOLECULAR ROTATIONS OF SELECTED DISACCHARIDES, COMPARED TO THE SUM OF THOSE OF THE CONSTITUENTS

Compound	$[M]_D$ (degrees $\times 10^{-2}$ )
Methyl tetra- <i>O</i> -acetyl- $\alpha$ -D-mannopyranoside <sup>a</sup> (11) + benzyl 2-acetamido-3,4,6-tri- <i>O</i> -acetyl-2-deoxy- $\alpha$ -D-glucopyranoside <sup>a</sup> (12)	+622
Methyl tetra- <i>O</i> -acetyl- $\beta$ -D-mannopyranoside <sup>a</sup> (13) + compound 12	+277
Compound 6 <sup>a</sup>	+1108
Compound 7 <sup>a</sup>	+615
Benzyl 2-acetamido-3,4-di- <i>O</i> -acetyl-2-deoxy-3- <i>O</i> -(tetra- <i>O</i> -acetyl- $\alpha$ -D-mannopyranosyl)- $\alpha$ -D-glucopyranoside <sup>a</sup> (ref. 6)	+574
Methyl $\alpha$ -D-mannopyranoside <sup>b</sup> (14) + 2-acetamido-2-deoxy- $\alpha$ -D-glucose <sup>b</sup>	+295
Compound 14 + 2-acetamido-2-deoxy- $\beta$ -D-glucose <sup>b</sup>	+105
Compound 9 <sup>c</sup> at equilibrium	+134
2-Acetamido-2-deoxy-3- <i>O</i> - $\alpha$ -D-mannopyranosyl-D-glucose <sup>c</sup> at equilibrium (ref. 6)	+222
2-Acetamido-2-deoxy-4- <i>O</i> - $\alpha$ -D-mannopyranosyl-D-glucose <sup>c</sup> at equilibrium (ref. 7)	+253
Compound 11 + 2-acetamido-1,3,4,6-tetra- <i>O</i> -acetyl-2-deoxy- $\alpha$ -D-glucose <sup>a</sup>	+529
Compound 11 + 2-acetamido-1,3,4,6-tetra- <i>O</i> -acetyl-2-deoxy- $\beta$ -D-glucose <sup>a</sup>	+176
Compound 10 <sup>a</sup>	+461

<sup>a</sup>Optical rotation determined in chloroform; <sup>b</sup>in water; <sup>c</sup>in 1:1 water-methanol.

## EXPERIMENTAL

*General.* — Melting points were determined with a Mettler FP-2 apparatus and correspond to "corrected melting points". Optical rotations were determined, in semimicrotubes, with a Perkin-Elmer Model 141 polarimeter. The chloroform used was analytical-reagent grade and contained about 0.75% of ethanol. I.r. spectra were recorded, for potassium bromide discs, with a Perkin-Elmer Model 237 spectrophotometer. N.m.r. spectra were recorded with a Varian A-60 n.m.r. spectrometer, for solutions in chloroform-*d* with tetramethylsilane as the internal standard. G.l.c. of the per-*O*-(trimethylsilyl) derivatives was performed with a Perkin-Elmer Model 900 gas chromatograph by use of a column of Chromosorb GHP coated with 3% OV-11 (Supleco Inc., Bellefonte, Pa. 16823, U. S. A.), programmed for a rise of 5°/min from 200 to 232°;  $t'_R$  is given relative to that of hexakis-*O*-(trimethylsilyl)-*myo*-inositol as unity. Column chromatography was performed on Silica Gel Merck (70–325 mesh; E. Merck, Darmstadt, Germany), used without pretreatment. The ratio of weight of substance to weight of adsorbent was 1:80 to 1:120. The volume of the fraction eluted was 3–4 ml per gram of the substance to be chromatographed. The ratio of diameter to length of the columns was 1:25. T.l.c. was performed on precoated

Silica Gel G plates (layer thickness 0.25 mm; E. Merck, Darmstadt, Germany); all compounds showed only one spot. Evaporations were conducted *in vacuo* with the bath temperature below 40°. Solutions in less than 5 ml of volatile solvents were evaporated under a stream of nitrogen. Microanalyses were performed by Dr. W. Manser, Zürich, Switzerland.

*Benzyl 2-acetamido-2-deoxy-6-O-trityl- $\alpha$ -D-glucopyranoside (2).* — A solution of benzyl 2-acetamido-2-deoxy- $\alpha$ -D-glucopyranoside<sup>10</sup> (1.33 g) in dry pyridine (15 ml) was treated with chlorotriphenylmethane (1.33 g) for 48 h at room temperature, and then poured into ice-water. The syrup that separated was dissolved in chloroform, and the solution was washed with water, dried (sodium sulfate), and evaporated; repeated addition and distillation of toluene gave a residue that showed two spots on t.l.c. in 4:1 benzene-methanol, one spot corresponding to triphenylmethanol. Six recrystallizations from ether-hexane or ether-pentane gave 2.1 g (84%) of 2 as fine needles, m.p. 101–102°,  $[\alpha]_D^{20} +39^\circ$  (*c* 1.3, chloroform); i.r. data:  $\nu_{\max}^{\text{KBr}}$  1655 (CONH) and 3350  $\text{cm}^{-1}$  (broad; OH); n.m.r. data (chloroform-*d*):  $\tau$  2.66 (20 H, 4 Ph), 3.66 (deuteratable doublet, *J* 9.0 Hz, NH), 6.20 and 7.13 (deuteratable, two OH), and 8.13 (N-Ac); t.l.c. in 4:1 benzene-methanol:  $R_F$  0.52.

*Anal.* Calc. for  $\text{C}_{34}\text{H}_{35}\text{NO}_6$ : C, 73.76; H, 6.37; N, 2.53; Found: C, 73.78; H, 6.41; N, 2.60.

*Benzyl 2-acetamido-3,4-di-O-acetyl-2-deoxy-6-O-trityl- $\alpha$ -D-glucopyranoside (3).* — A solution of 2 (1.0 g) in dry pyridine (10 ml) was cooled to 0° and treated with acetic anhydride (15 ml). After 48 h at room temperature, the solution was poured into ice-water, and the mixture was extracted with chloroform. The extract was dried (sodium sulfate) and evaporated, and toluene was repeatedly added to and distilled from the residue. Crystallization from chloroform-hexane gave 1.0 g (91%) of fine needles, m.p. 80–81°,  $[\alpha]_D^{20} +87^\circ$  (*c* 1.2 chloroform); i.r. data:  $\nu_{\max}^{\text{KBr}}$  1655 (CONH) and 1745  $\text{cm}^{-1}$  (OAc); n.m.r. data (chloroform-*d*):  $\tau$  2.63 (20 H, 4 Ph), 4.18 (deuteratable doublet, *J* 9.5 Hz, NH), 8.02, 8.10 (two OAc), and 8.28 (N-Ac); t.l.c. in 19:1 chloroform-ethanol:  $R_F$  0.56.

*Anal.* Calc. for  $\text{C}_{38}\text{H}_{39}\text{NO}_8$ : C, 71.57; H, 6.16; N, 2.20. Found: C, 71.48; H, 6.15; N, 2.25.

*Benzyl 2-acetamido-3,4-di-O-acetyl-2-deoxy- $\alpha$ -D-glucopyranoside (4).* — A solution of 3 (1.0 g) in glacial acetic acid (10 ml) was cooled to 10° and treated with a 32% solution of hydrogen bromide in glacial acetic acid (1 ml). The mixture was shaken for 75 sec, and then filtered rapidly on a sintered-glass funnel, into ice-water. The filtrate was extracted with chloroform, and the extract was successively washed with saturated sodium hydrogen carbonate solution and water, dried (sodium sulfate), and evaporated, to give a residue which was crystallized from benzene-hexane, affording 0.4 g (82%) of 3 as long needles, m.p. 166–167°,  $[\alpha]_D^{20} +128^\circ$  (*c* 1.2, chloroform); i.r. data:  $\nu_{\max}^{\text{KBr}}$  1655 (CONH), 1750 (OAc), and 3350  $\text{cm}^{-1}$  (OH); n.m.r. data (chloroform-*d*):  $\tau$  2.63 (5 H, Ph), 4.20 (doublet, *J* 9.5 Hz, NH), 7.58 (deuteratable, OH), 7.95 (6 H, two OAc), and 8.10 (N-Ac); t.l.c. in 4:1 benzene-methanol:  $R_F$  0.45.

*Anal.* Calc. for  $C_{19}H_{25}NO_8$ : C, 57.71; H, 6.37; N, 3.54. Found: C, 57.86; H, 6.29; N, 3.52.

*Benzyl 2-acetamido-3-O-acetyl-2-deoxy- $\alpha$ -D-glucopyranoside (5).* — This compound was prepared as previously described<sup>8</sup>. When the product of the debenzylidenation was crystallized from acetone-ether-pentane, hygroscopic microcrystals were obtained, m.p. 55–56°,  $[\alpha]_D^{20} + 106^\circ$  (*c* 1.4, chloroform); i.r. data:  $\nu_{\max}^{KBr}$  1660 (CONH), 1750 (OAc), and  $3400\text{ cm}^{-1}$  (broad; OH); n.m.r. data (chloroform-*d*):  $\tau$  2.70 (5 H, Ph), 3.59 (doublet, *J* 9.0 Hz, NH), 8.0 (OAc), and 8.20 (NAc); t.l.c. in 7:3 benzene-methanol:  $R_F$  0.55.

*Anal.* Calc. for  $C_{17}H_{32}NO_7$ : C, 57.78; H, 6.56; N, 3.96; O, 31.69. Found: C, 57.87; H, 6.55; N, 3.88; O, 31.70.

Recrystallization from ethyl acetate-2-isopropoxypropane, as previously described<sup>8</sup>, gave needles having m.p. 121°,  $[\alpha]_D^{20} + 111^\circ$  (*c* 0.9 pyridine),  $[\alpha]_D^{20} + 106^\circ$  (*c* 1.4, chloroform), identical with the product previously described {lit.<sup>8</sup> m.p. 119–121°;  $[\alpha]_D^{25} + 132^\circ$  (*c* 1, pyridine)\*}.

*Benzyl 2-acetamido-3,4-di-O-acetyl-2-deoxy-6-O-(tetra-O-acetyl- $\alpha$ -D-mannopyranosyl)- $\alpha$ -D-glucopyranoside (6).* — A mixture of dry **4** (1.1 g) and mercuric cyanide (1 g) in dry 1:1 benzene-nitromethane (180 ml) was concentrated to 130 ml under atmospheric pressure and then cooled to room temperature. A solution of tetra-O-acetyl- $\alpha$ -D-mannopyranosyl bromide<sup>9</sup> (**1**, 1.5 g) in dry 1,2-dichloroethane (15 ml) was added, and the mixture was stirred for 3 days at room temperature. Additional amounts of the bromide (0.5 g) and mercuric cyanide (0.5 g) were added, and the mixture was stirred for a further 24 h. The mixture was diluted with 1,2-dichloroethane (100 ml), washed successively with a cold, saturated solution of sodium hydrogen carbonate and water, dried (sodium sulfate), and evaporated to a syrup (3 g) which was chromatographed on silica gel with 19:1 chloroform-ethanol, to give a fraction enriched in compound **6**. Chromatography of this fraction on another column of silica gel, with 1:1 ethyl acetate-ether, gave 1.3 g (65%) of **6**, which crystallized from ether-hexane as microcrystals having m.p. 76–78°,  $[\alpha]_D^{20} + 143^\circ$  (*c* 1.2, chloroform); i.r. data:  $\nu_{\max}^{KBr}$  1675 (CONH) and  $1750\text{ cm}^{-1}$  (OAc); n.m.r. data (chloroform-*d*):  $\tau$  2.60 (5 H, Ph), 4.20 (doublet, *J* 9.5 Hz, NH), and 7.67 (21 H, 6 OAc+NAc); t.l.c. in 1:1 ethyl acetate-ether:  $R_F$  0.41.

*Anal.* Calc. for  $C_{33}H_{43}NO_{17}$ : C, 54.61; H, 5.97; N, 1.92; O, 37.41. Found: C, 54.46; H, 5.94; N, 2.11; O, 37.27.

*Benzyl 2-acetamido-3-O-acetyl-2-deoxy-6-O-(tetra-O-acetyl- $\alpha$ -D-mannopyranosyl)- $\alpha$ -D-glucopyranoside (7).* — Compounds **5** (537 mg) and **1** (1.2 g) were condensed in the way just described for compound **6**. The residue obtained after processing of the reaction mixture was chromatographed on a column of silica gel with 1:1 ethyl acetate-ether. A fraction was obtained that crystallized from acetone-pentane to give 520 mg (42%) of **7** as microcrystals, m.p. 82–83°,  $[\alpha]_D^{20} + 90^\circ$  (*c* 0.8,

\*A re-examination of the sample previously described<sup>8</sup> showed  $[\alpha]_D^{20} + 113^\circ$  (*c* 0.9, pyridine).

chloroform); i.r. data:  $\nu_{\text{max}}^{\text{KBr}}$  1660 (CONH), 1745 (OAc), and  $3350\text{ cm}^{-1}$  (OH); t.l.c. in 1:1 ethyl acetate–ether:  $R_F$  0.28.

*Anal.* Calc. for  $\text{C}_{31}\text{H}_{41}\text{NO}_{16}$ : C, 54.48; H, 6.05; N, 2.05; O, 37.44. Found: C, 54.37; H, 6.03; N, 2.15; O, 37.50.

*Benzyl 2-acetamido-2-deoxy-6-O- $\alpha$ -D-mannopyranosyl- $\alpha$ -D-glucopyranoside (8).* — From 6. A solution of 6 (200 mg) in methanol (5 ml) was treated with 0.1M sodium methoxide in methanol (0.5 ml) for 24 h at 4°. The solution was deionized with Dowex-50 ( $\text{H}^+$ ) ion-exchange resin (1 ml) and then evaporated. The residue was crystallized from methanol–ethyl acetate to give 119 mg (91%) of 8 as microcrystals containing 0.5 molecule of water per molecule, m.p. 117–118°,  $[\alpha]_{\text{D}}^{20} +167^\circ$  (c 0.7, methanol); i.r. data:  $\nu_{\text{max}}^{\text{KBr}}$  1650 (CONH) and  $3370\text{ cm}^{-1}$  (broad; OH); g.l.c. data: peak at  $t'_R$  31.60; t.l.c. in 1:1 benzene–methanol:  $R_F$  0.33.

*Anal.* Calc. for  $\text{C}_{21}\text{H}_{31}\text{NO}_{11} \cdot 0.5\text{ H}_2\text{O}$ : C, 52.26; H, 6.69; N, 2.90. Found: C, 52.67; H, 6.59; N, 2.84.

From 7. De-O-acetylation of 7 by the procedure just described gave 8, having m.p., mixed m.p. (116–118°), i.r. spectrum, and  $R_F$  value identical with those of the compound obtained from 6.

*2-Acetamido-2-deoxy-6-O- $\alpha$ -D-mannopyranosyl- $\beta$ -D-glucose (9).* — A solution of 8 (292 mg) in a mixture of 95% ethanol (49 ml) and acetic acid (1 ml) was hydrogenolyzed with hydrogen under pressure (3.4 atm.) in the presence of 10% palladium-on-charcoal (300 mg) for 72 h, the catalyst was filtered off, and the solution was evaporated. Crystallization of the product from methanol–acetone gave 222 mg (94%) of 9 as hygroscopic needles containing 0.5 molecule of water per molecule, softening at 136°, and melting at 142–144°;  $[\alpha]_{\text{D}}^{20} +38 \rightarrow +35^\circ$  (equilibrium after 48 h; c 1.2, 50% methanol); i.r. data:  $\nu_{\text{max}}^{\text{KBr}}$  1650 (CONH) and  $3350\text{ cm}^{-1}$  (broad; OH); g.l.c. data: double peak at  $t'_R$  15.4 and 17.1.

*Anal.* Calc. for  $\text{C}_{14}\text{N}_2\text{O}_{11} \cdot 0.5\text{ H}_2\text{O}$ : C, 42.85; H, 6.68; N, 3.57. Found: C, 42.97; H, 6.73; N, 3.43.

*2-Acetamido-1,3,4-tri-O-acetyl-2-deoxy-6-O-(tetra-O-acetyl- $\alpha$ -D-mannopyranosyl)- $\alpha$ -D-glucose (10).* — A solution of compound 9 (100 mg) in pyridine (2 ml) was treated with acetic anhydride (3 ml) for 24 h at room temperature. Evaporation gave a residue that crystallized from benzene–pentane to give 10 as microcrystals (133 mg, 86%), m.p. 75–77°,  $[\alpha]_{\text{D}}^{20} +68^\circ$  (c 0.6, chloroform); i.r. data:  $\nu_{\text{max}}^{\text{KBr}}$  1665 (CONH) and  $1750\text{ cm}^{-1}$  (OAc); t.l.c. in 4:1 benzene–methanol:  $R_F$  0.49.

*Anal.* Calc. for  $\text{C}_{28}\text{H}_{39}\text{NO}_{18}$ : C, 49.63; H, 5.80; N, 2.07; O, 42.50. Found: C, 49.52; H, 5.75; N, 2.16; O, 42.54.

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*Anal.* Calc. for  $\text{C}_{21}\text{H}_{31}\text{NO}_{11} \cdot 0.5\text{ H}_2\text{O}$ : C, 52.26; H, 6.69; N, 2.90. Found: C, 52.67; H, 6.59; N, 2.84.

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*Anal.* Calc. for  $\text{C}_{28}\text{H}_{39}\text{NO}_{18}$ : C, 49.63; H, 5.80; N, 2.07; O, 42.50. Found: C, 49.52; H, 5.75; N, 2.16; O, 42.54.

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