Note

An improved synthesis of D-perosamine and some derivatives

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The family of 4-amino-4,6-dideoxy-D-hexoses, first discovered in antibiotics and in bacterial cell-wall¹⁻³, now includes at least seven configurationally distinct representatives⁴⁻⁶. Interest in the biological significance of these unstable monosaccharides and their naturally occurring derivatives has prompted much work on their chemistry⁷⁻⁹. One such sugar, 4-amino-4,6-dideoxy-D-mannose (Dperosamine, 1), is also found *N*-formylated in α -(1->2)-glycosidic linkage as the



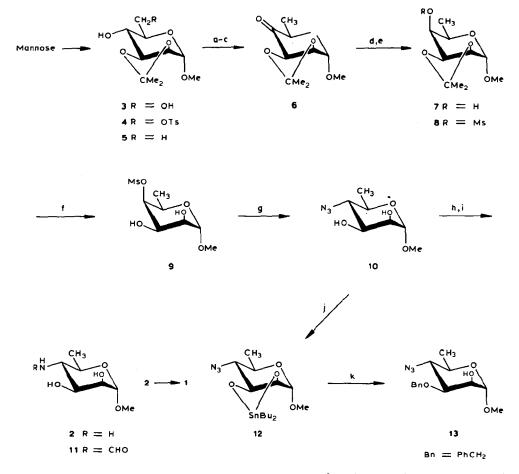
principal structural component of O-antigenic polysaccharides in several Gramnegative organisms. Foremost among these are the pathogens *Vibrio cholera*¹⁰, *Yersinia enterocolitica*¹¹, and *Brucella abortus*¹², which respectively cause cholera and acute gastroenteritis in man, and brucellosis in cattle. These findings not only establish a molecular basis for extensive cross-serological reactions documented in earlier studies, but also suggest an approach to the design of artificial vaccines and specific diagnostic reagents to combat these afflictions.

As part of our program to develop a new, totally synthetic, brucellosis vaccine, we now report an improved route to methyl α -D-perosinamide (2), culminating in the first total synthesis of D-perosamine hydrochloride (1 · HCl). Our approach permits the attachment of various N-acyl substituents, and allows regiochemical differentiation for D-perosamine, of the hydroxyl groups required for the assembly of derived oligosaccharides.

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DISCUSSION

Previous routes to compound 2 used readily available monosaccharides as starting materials, but introduced the nitrogen substituent onto C-4 in unacceptably low yield. In the synthesis of L-2 from L-rhamnose by Brimacombe *et al.*⁸, methyl 6-deoxy-2,3-O-isopropylidene-4-O-(methylsulfonyl)- α -L-talopyranoside (the enantiomer of 8) underwent replacement, by reaction with NaN₃, in only 14% yield. The synthesis by Stevens *et al.*^{6b} of the natural enantiomer 2 from D-mannose relied on NaN₃ opening of methyl 3,4-anhydro-2-O-benzoyl-6-deoxy- α -D-talopyranoside (not shown), which afforded the desired 4-azido-4,6-dideoxy- α -D-manno-



Scheme 1. Synthesis of D-perosamine (1) and some derivatives. [key: (reagents, time, temperature, and yield) (a) TsCl, Et₃N, 24 h, 25°, 96%; (b) LiAlH₄, ether, 6 h, 30°, 100%; (c) oxalyl chloride, dimethyl sulfoxide, Et₃N, 1 h, -55 to 23°, 95%; (d) NaBH₄, 3.5:1 ethanol-water, 15 min, 25°, 98%; (e) MsCl, Et₃N, CH₂Cl₂, 2 h, 25°, 70%; (f) methanol-HCl, 5 h, 25° 100%; (g) NaN₃, dimethyl sulfoxide, 6 h, 100°, 93%; (h) H₂, Pd/C, methanol, 1.5 h, 25° 85%; (i) 2:1 ethyl formate-ethanol, 24 h, 85°, 91%; (j) Bu₂SnO, methanol 1 h, reflux; and (k) neat benzyl bromide, 10 h, 100°, 95%.]

pyranoside (10) in 20-43% yield^{6b}. Our own synthetic sequence, depicted in Scheme 1, also utilizes D-mannose as the starting material, and it shares some features of both of the earlier routes. However, an improved process for amination of C-4, achieved in >90% yield, now makes possible the preparation of D-2 in multigram quantities.

6-Tosylate 4 was prepared from methyl 2,3-O-isopropylidene- α -D-mannopyranoside¹³ 3, and converted into the known^{6a} crystalline mesylate 8 by modification of the Stevens synthesis^{6a}. Thus, 4 was reduced with LiAlH₄ to the known^{4c} deoxy sugar 5 in high overall yield. Oxidation of 5 by the method of Swern *et al.*¹⁴ with oxalyl chloride in Me₂SO furnished the 6-deoxyhexosid-4-ulose 6 in 95% yield. Reduction of 6 with sodium borohydride gave 7 as reported^{6a,8} (98%), which was then converted into 8 (70%). In fact, 8 was virtually immune to nucleophilic attack by azide under a variety of conditions tested; however by the simple expedient of first removing the adjacent isopropylidene protecting group, the resultant mesylate 9 underwent normal replacement, with NaN₃ in Me₂SO at 100°, to afford azide 10 in 72% yield.

Compound 10 proved to be an extremely versatile intermediate for the synthesis of D-perosamine and its derivatives. Hydrogenation gave methyl α -Dperosaminide (2), which was readily formylated to give 11, which is the repeating structural unit¹² in the *B. abortus* O-polysaccharide. Although earlier workers^{6b} were unsuccessful in hydrolyzing 2 to its parent sugar, we found that treatment of this methyl glycoside with anhydrous HF followed by dilute aqueous HCl readily produced 1 · HCl via the intermediate glycosyl fluoride⁹ (not shown).

To construct short α -(1 \rightarrow 2)-linked oligomers of *N*-formylperosamine¹⁵, the 3-hydroxyl group in **10** was selectively protected by using the elegant method of David *et al.*¹⁶. Benzylation of the stannylene derivative **12** furnished **13** in 95% overall yield from **10**.

EXPERIMENTAL

General methods. — Melting points were determined with a Thomas–Hoover Unimelt apparatus and are uncorrected. Thin-layer chromatography was conducted on Merck precoated plates of silica gel 60F-254. Microanalyses were performed by Galbraith Laboratories, Inc., Knoxville, TN. Infrared spectra were recorded with a Perkin–Elmer 681 spectrophotometer and are calibrated to polystyrene. ¹H-N.m.r. spectra were recorded at 300 MHz with a Bruker WM-300 spectrometer. Samples dissolved in CDCl₃ used CHCl₃ as the internal standard, and samples in D₂O or Me₂SO-d₆ used the HOD peak as internal reference. Chemical-ionization mass spectrometry (c.i.m.s.) using a computerized AEI-MS 902 instrument was performed with isobutane as the reagent gas.

Methyl 2,3-O-isopropylidene-6-O-p-tolylsulfonyl- α -D-mannopyranoside (4). — Diol 3 (17 g, 73 mmol)¹³ was dissolved in CH₂Cl₂ (150 mL) under argon at 25° in a flame-dried, round-bottomed flask fitted with a magnetic stirring bar. Triethylamine (13.1 mL, 94 mmol) was added, followed by solid TsCl (18 g, 94 mmol) in one portion. The resulting solution was stirred for 24 h at 25°. CH₂Cl₂ (250 mL) was added, and the solution was successively washed with saturated NaHCO₃ (5 × 100 mL) and H₂O (3 × 100 mL), dried (MgSO₄), and evaporated to a syrup. Chromatography on silica gel (hexane, and then 1:1 hexane–ethyl acetate) afforded tosylate 4 as a viscous oil (27 g, 96%); $[\alpha]_D$ +12.3° (c 2.75, CH₂Cl₂): R_F 0.36 in 1:1 hexane–EtOAc; ν_{max}^{finn} 3450, 3060, 2940, 1740, 1600, 1450, 1360, 1245, 1220, 1190, 1175, 1090, and 960 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 7.79, 7.33 (4 H, tosyl), 4.82 (s, 1 H, H-1), 4.32–4.24 (m, 2 H, H-6,6'), 4.11–4.06 (m, 2 H, H-2,3), 3.72 (ddd, 1 H, J 3.7, 3.9, and 9.5 Hz, H-5), 3.60 (dd, 1 H, J 4.9, 9.5 Hz, H-4), 3.33 (s, 3 H, OCH₃), 2.43 (s, 3 H, tosyl-CH₃), and 1.46, 1.31 (2 s, each 3 H, CH₃); c.i.m.s. *m/z* 389 (M + 1, 2%) and 357 (M + 1 – H₂O, 100%).

Anal. Calc. for C₁₇H₂₄O₈S: C, 52.58; H, 6.19; S, 8.25. Found: C, 52.45; H, 6.31; S, 8.25.

Methyl 6-deoxy-2,3-O-isopropylidene- α -D-mannopyranoside (5). — A solution of tosylate 4 (27 g, 51 mmol) in Et₂O (500 mL) under argon in a flame-dried, three-necked flask fitted with a reflux condenser, thermometer, and magnetic stirring bar was cooled to 0° and LiAlH₄ (2.9 g, 77 mmol) was slowly added, with stirring. After 15 min, the ice bath was removed and the mixture allowed to warm to room temperature, but kept below 30°. After 6 h, the mixture was recooled in ice, and the reaction guenched with saturated aqueous oxalic acid (10 mL). The mixture was warmed to room temperature, stirred for 15 min, and then filtered. The solids were washed with fresh ether, and the filtrate and washings were combined, dried (MgSO₄), and evaporated. Chromatography on SiO₂ gel with 3:2hexane-EtOAc afforded 5 (11 g, 100%) as a colorless syrup; $[\alpha]_{D}$ +16.4° (c 1.4, MeOH), lit.⁴ $[\alpha]_D^{27}$ +15.2°; $R_F 0.42$ (1:1 hexane-EtOAc); ν_{max}^{film} 3460, 2940, 2910, 1455, 1385, 1340, 1240, 1220, 1170, 1140, and 1090 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 4.82 (bs, 1 H, H-1), 4.10 (d, 1 H, J 5.9 Hz, H-2), 4.04 (dd, 1 H, J 5.9, 6.9 Hz, H-3), 3.62 (dq, 1 H, J 6.3, 9 Hz, H-5), 3.37 (ddd, 1 H, J 4.5, 6.9, 9 Hz, H-4), 3.36 (s, 3 H, OCH₃), 1.51, 1.33 (2 s, each 3 H, CH₃), and 1.28 (d, 3 H, J 6.3 Hz, CH₃); c.i.m.s. m/z 219 (M + 1, 27%) and 187 (M + 1 - CH₂OH, 100%).

Methyl 6-deoxy-2,3-O-isopropylidene- α -D-lyxo-hexopyranosid-4-ulose¹⁴ (6). — A solution of oxalyl chloride (5 mL, 55 mmol) in CH₂Cl₂ (125 mL) in a flamedried vessel containing a magnetic stirring bar under argon was cooled to -55° . To it was added Me₂SO (8.5 mL, 110 mmol) in CH₂Cl₂ (25 mL) via a tube, and the solution stirred for 5 min at -55° . Alcohol 5 (11 g, 50.5 mmol) in CH₂Cl₂ (50 mL) was introduced via a tube, and the resultant, cloudy solution maintained for 0.5 h at -55° . Triethylamine (35 mL, 250 mmol) was added, and the mixture was warm allowed to room temperature during \sim 1 h. After quenching with H₂O (250 mL), the layers were separated, and the aqueous phase was extracted with CH₂Cl₂ (3 × 150 mL). The extracts were combined, successively washed with saturated brine (3 × 100 mL) and saturated NaHCO₃ (3 × 150 mL), dried (MgSO₄), and evaporated *in vacuo*, to afford 6 (10.3 g, 95%); [a]_D +100° (c 1.0, EtOH), lit.⁹ [a]_D -105° for the L enantiomer; $R_{\rm F}$ 0.7 in 1:1 hexane–EtOAc; $\nu_{\rm max}^{\rm film}$ 2940, 1745, 1450, 1385, 1375, 1225, 1160, 1135, 1090, and 1050 cm⁻¹; ¹H-n.m.r. (CDCL₃): δ 4.82 (s, 2 H, H-1), 4.42 (d, 1 H, J 6.7 Hz, H-2), 4.39 (d, 1 H, J 6.7 Hz, H-3), 4.22 (q, 1 H, J 6.8 Hz, H-5), 3.44 (s, 3 H, OCH₃), 1.46, 1.38 (d, 3 H, J 6.8 Hz, CH₃), and 1.34 (br. s, 6 H, 2 CH₃); c.i.m.s. *m/z* 219 (M + 1, 47%), and 127 (M + 1 - CH₃OH and Me₂CO, 100%).

Methyl 6-deoxy-2,3-O-isopropylidene- α -D-talopyranoside (7). — To a solution of 6 (12.3 g, 57 mmol) in 7:2 EtOH-H₂O (180 mL) at 0° was added NaBH₄ (1.5 g, 39.6 mmol). After stirring for 15 min at 0° and 15 min at 25°, the reaction was quenched with acetone (10 mL), and the mixture evaporated under diminished pressure at 50° to a syrup which was dissolved in CH₂Cl₂ (500 mL), and the solution washed successively with saturated NaHCO₃ (3 × 150 mL) and H₂O (3 × 150 mL), and dried (MgSO₄). Chromatography on silica gel in 7:3 hexane–EtOAc afforded 7 as a syrup (12.1 g, 98%); $[\alpha]_D$ +48.9° (c 1.3, MeOH); lit.⁸ $[\alpha]_D$ -49° for the L enantiomer; R_F 0.44 in 1:1 hexane–EtOAc; $\nu_{\text{finx}}^{\text{finx}}$ 3520, 2940, 1450, 1380, 1370, 1255, 1215, 1150, 1080, and 1050 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 4.91 (s, 1 H, H-1), 4.18 (dd, 1 H, J 5.1, 6.2 Hz, H-3), 4.01 (d, 1 H, J 6.2 Hz, H-2), 3.81 (q, 1 H, J 6.6 Hz, H-5), 3.54 (dd, 1 H, J 5.5, 6.4 Hz, H-4), 3.38 (s, 3 H, OCH₃), 1.57, 1.36 (2 s, each 3 H, 2 CH₃), and 1.31 (d, 3 H, J 6.6 Hz, CH₃); c.i.m.s. *m*/z 219 (M + 1, 5%) and 187 (M + 1 - CH₃OH, 100%).

Methyl 6-deoxy-2,3-O-isopropylidene-4-O-(methylsulfonyl)- α -D-talopyranoside (8). — To a solution of 7 (12 g, 55 mmol) in CH_2Cl_2 (165 mL) at 0° were added triethylamine (15.3 mL, 110 mmol) and methanesulfonyl chloride (8.5 mL, 110 mmol). After stirring for 10 min at 0° and for 2 h at 25°, the reaction was quenched with saturated NaHCO₃ (10 mL), and the mixture stirred for 0.5 h. Dichloromethane (300 mL) was added, and the aqueous phase was separated. The organic phase was successively washed with saturated NaHCO₃ (3×100 mL), saturated CuSO₄ solution (4 \times 100 mL), and H₂O (3 \times 100 mL), dried (MgSO₄), and evaporated, to afford 14.2 g of impure mesylate which crystallized from 1:4 ethanolhexane, to give 11.4 g (70%) of pure **8**; m.p. 115–117°, lit.^{6a} m.p. 116–117.5°, $[\alpha]_{D}$ +19.0° (c 0.8, MeOH), lit.^{6a} [α]_D -20.2°; $R_{\rm F}$ 0.37 in 1:1 EtOAc-hexane; $\nu_{\rm max}^{\rm film}$ 2950, 1465, 1350, 1250, 1215, 1180, 1160, 1145, 1090, and 1040 cm⁻¹; ¹H-n.m.r. (CDCl₃): 4.86 (br. s, 1 H, J 1.1 Hz, H-1), 4.62 (dd, 1 H, J 1.1, 6.5 Hz, H-2), 4.36 (dd, 1 H, J 5.6, 6.5 Hz, H-3), 4.04 (dd, 1 H, J 2.1, 5.6 Hz, H-4), 3.95 (dq, 1 H, J 2.1, 6.7 Hz, H-5), 3.38 (s, 3 H, OCH₃), 3.07 (s, 3 H, mesyl-CH₃), 1.58, 1.34 (2 s, each 3 H, CH_3), and 1.35 (d, 3 H, J 6.7 Hz, CH_3); c.i.m.s. m/z 297 (M + 1, 63%) and 265 $(M + 1 - CH_3OH, 100\%).$

Methyl 6-deoxy-4-O-(methylsulfonyl)- α -D-talopyranoside (9). — To a solution of 8 (13.5 g, 45.6 mmol) in anhydrous MeOH (500 mL) was added saturated, methanolic HCl (10 mL), and the mixture was stirred for 5 h at 25°, and evaporated under diminished pressure at 20°, to afford crystalline 9 (11.7 g, 100%). A portion was recrystallized from 1:1 ether-pentane; m.p. 102–104°, $[\alpha]_D$ –95.7° (c 1.0, MeOH); R_F 0.18 in 2:1 EtOAc-hexane; ν_{max}^{KBr} 3500, 2940, 1340, 1170, 1130, 1100,

1065, and 1020 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 4.77 (d, 1 H, J 1.8 Hz, H-1), 4.76 (dd, 1 H, J 1, 3.7 Hz, H-4), 4.00 (dq, 1 H, J 1, 6.6 Hz, H-5), 3.94 (ddd, 1 H, J 3.7, 3.7, 8.5 Hz, H-3), 3.71 (ddd, 1 H, J 1.8, 3.7, 11.1 Hz, H-2), 3.37 (s, 3 H, OCH₃), 3.16 (s, 3 H, mesyl-CH₃), and 1.33 (d, 3 H, J 6.6 Hz, CH₃); c.i.m.s. *m/z* 257 (M + 1, 23%), 239 (M + 1 - H₂O, 58%), and 225 (M + 1 - H₂O and CH₃OH, 100%).

Anal. Calc. for C₈H₁₆O₇S: C, 37.50; H, 6.25; S, 12.50. Found: C, 37.69; H, 6.44; S, 12.50.

Methyl 4-azido-4,6-dideoxy- α -D-mannopyranoside (10). — Mesylate 9 (15 g, 59 mmol) was stirred with NaN₃ (20 g, 295 mmol) in Me₂SO (150 mL) for 6 h at 100°. Most of the solvent was removed by distillation at 90°/67 Pa. The residue was suspended in CH₂Cl₂ (500 mL), and the suspension filtered through a pad of Celite. The filtrate was evaporated in a rotary evaporator to a final volume of 15 mL. Chromatography on a column of silica gel eluted with 3:2 hexane–EtOAc afforded crystalline 10 (8.6 g, 72%). A portion was recrystallized from hexane, to give needles; m.p. 83–84°, lit.^{6b} m.p. 81.5–82.5°, [α]_D +131.8° (*c* 1.0, MeOH), lit.^{6b} [α]_D +126.9°; R_F 0.45 in 2:1 EtOAc–hexane; ν_{max}^{KBr} 3860–3240, 2940, 2120 (N₃), 1465, 1370, 1290, 1265, 1200, 1140, 1100, 1070, and 1040 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 4.66 (d, 1 H, J 1.4 Hz, H-1), 3.89 (ddd, 1 H, J 1.4, 3.3, 4.4 Hz, H-2), 3.81 (ddd, 1 H, J 3.3, 6.6, 9.8 Hz, H-3), 3.55 (dq, 1 H, J 6.3, 10.5 Hz, H-5), 3.33 (s, 3 H, OCH₃), 3.26 (dd, 1 H, J 9.8, 10.5 Hz, H-4), and 1.33 (d, 3 H, J 6.3 Hz, CH₃); c.i.m.s. *m/z* 204 (M + 1, 5%) and 172 (M + 1 – CH₃OH, 100%).

Methyl 4-amino-4,6-dideoxy-α-D-mannopyranoside (2). — To a solution of 10 (1.22 g, 6 mmol) in MeOH (100 mL) was added 10% Pd-on-C (200 mg). Hydrogen was bubbled through an E-frit-tipped glass tube into the stirred solution for 1.5 h at 25°. The catalyst was filtered off through Celite, and the filtrate was evaporated *in vacuo*, to afford a solid which was recrystallized from 1:3 ethanolether, to give 2 as prisms (0.9 g, 85%); m.p. 150–152°, lit.^{6b} m.p. 152–153°, $[\alpha]_D$ +83.2° (*c* 1.0, MeOH), lit.^{6b} $[\alpha]_D$ +82.5°; R_F 0.73 in 7:4:1 CHCl₃–MeOH-33% NH₄OH; ν_{max}^{Nujol} 3500, 3340, 2740, 1450, 1400, 1380, 1355, 1240, 1120, 1100, and 1085 cm⁻¹; ¹H-n.m.r. (D₂O): δ 4.54 (br. s, 1 H, J 1.4 Hz, H-1), 3.68 (dd, 1 H, J 1.4, 3.3 Hz, H-2), 3.45 (dq, 1 H, J 6.3, 10 Hz, H-5), 3.41 (dd, 1 H, J 3.3, 9.9 Hz, H-3), 3.22 (s, 3 H, OCH₃), 2.56 (dd, 1 H, J 9.9, 10 Hz, H-4), and 1.11 (d, 3 H, J 6.3 Hz, CH₃); c.i.m.s. *m/z* 178 (M + 1, 88%) and 146 (M + 1 – CH₃OH, 100%).

D-(-)-Perosamine (1). — In a 29.6-mL (1-oz.) polypropylene bottle fitted with a screw cap and a Teflon-coated, magnetic stirring bar was placed 200 mg (1.13 mmol) of compound 2. The apparatus was cooled to -78° , and anhydrous HF (5 mL) was rapidly transferred into it *via* a polypropylene syringe fitted with a teflon needle. The bottle was capped, warmed to room temperature, and the mixture stirred for 1.5 h, after which time, the HF was removed under a stream of argon. The resulting, pale-yellow gum was stored at 13.3 Pa for 12 h, whereupon it solidified. The solid was dissolved in distilled water (100 mL), the solution treated with Norit (50 mg), the suspension filtered through a pad of Celite, and the filtrate lypophilized. The crude α -glycosyl fluoride was ground to a fine powder, transferred to a 100-mL, round-bottomed flask fitted with a stirring bar, dissolved in water (30 mL), and the pH adjusted to 1 with M HCl (~5 mL). The solution was stirred for 24 h at room temperature, diluted to 100 mL with water, and lyophilized, to afford pure D-(-)-perosamine hydrochloride (220 mg, 98%) as an amorphous mixture of anomers; $[\alpha]_D^{23} -23^\circ$ (c 1.3, H₂O), lit.³ $[\alpha]_D^{23} -20^\circ$ (c 1.3, H₂O); β anomer: ¹H-n.m.r. (D₂O): δ 4.74 (d, 1 H, J 1 Hz, H-1), 4.02 (dq, 1 H, J 6.4, 10.4 Hz, H-5), 3.86 (dd, 1 H, J 3.3, 10.4 Hz, H-3), 3.78 (dd, 1 H, J 1.7, 3.3 Hz, H-2), 3.01 (t, 1 H, J 10.4 Hz, H-4), and 1.18 (d, 3 H, J 6.4 Hz, CH₃); α anomer: δ 5.01 (d, 1 H, J 1.7 Hz, H-1), 3.82 (dd, 1 H, J 3.2 Hz, H-2), 3.68 (dd, 1 H, J 3.2, 10.2 Hz, H-3), 3.57 (dq, 1 H, J 6.3, 10.2 Hz, H-5), 2.92 (t, 1 H, J 10.2 Hz, H-4), and 1.21 (d, 3 H, J 6.3 Hz, CH₃); c.i.m.s. m/z 164 (M + 1, 6%), 146 (M + 1 - H₂O, 12%), and 112 (100%).

Methyl 4,6-*dideoxy-4-formamido*- α -D-*mannopyranoside* (11). — A solution of compound 2 (200 mg, 1.13 mmol) in EtOH (4 mL) was diluted with ethyl formate (12 mL), refluxed for 24 h at 85°, cooled, and evaporated under diminished pressure. Chromatography on silica gel, using 9:1 CHCl₃-MeOH, followed by recrystallization from 1:8 ethanol-ether, gave 11 (210 mg, 91%); m.p. 173-174°, [α]_D +82.7° (*c* 1.0, MeOH); R_F 0.36 in 9:1 CHCl₃-MeOH; ν_{max}^{KBr} 3420, 3320, 2940, 2760, 1670, 1535, 1450, 1405, 1390, 1320, 1300, and 1130 cm⁻¹; ¹H-n.m.r. (Me₂SO-*d*₆) revealed two rotational isomers; major: δ 8.06 (s, 1 H, CHO), 7.85 (d, 1 H, *J* 11.3 Hz, NH), 4.47 (br. s, 1 H, H-1), 3.74 (q, 1 H, *J* 10 Hz, H-4), 3.60-3.45 (m, 3 H, H-2,3,5), 3.23 (s, 3 H, OCH₃), 1.04 (d, 3 H, *J* 6.2 Hz, CH₃); minor: δ 7.85 (d, 1 H, *J* 11.3 Hz, CHO), 7.48 (dd, 1 H, *J* 9.2, 11.3 Hz, NH), 4.47 (br. s, 1 H, H-1), 3.60-3.45 (m, 3 H, H-2,3,5), 3.30 (s, 3 H, OCH₃), 3.07 (q, 1 H, *J* 10 Hz, H-4), and 1.09 (d, 3 H, *J* 6.2 Hz, CH₃); c.i.m.s. *m/z* 206 (M + 1, 37%), 174 (M + 1 – CH₃OH, 100%), and 156 (M + 1 – CH₃OH and H₂O, 100%).

Anal. Calc. for C₈H₁₅NO₅: C, 46.83; H, 7.32; N, 6.83. Found: C, 46.87; H, 7.55; N, 6.71.

Methyl 4-azido-3-O-benzyl-4,6-dideoxy- α -D-mannopyranoside (13). — To a solution of compound 10 (203 mg, 1 mmol) in anhydrous MeOH (25 mL) was added dibutyltin oxide (249 mg, 1 mmol). The stirred suspension became a clear solution after boiling for 1 h under reflux. The solvent was evaporated under diminished pressure, and the crude stannylene compound 12 was dried under high vacuum for 2 h. The residue was dissolved in benzyl bromide (2 mL), and the clear solution was stirred for 10 h at 100°. Solvent was removed, via Kugelrohr, at 1.07 to 1.33 kPa, and the crude material was chromatographed twice on silica gel using 3:2 hexane-EtOAc, to afford 13 (277 mg, 95%) as a colorless syrup, $[\alpha]_{\rm D}$ +134° (c 1.0, MeOH); ν_{max}^{film} 3460, 2940, 2110 (N₃), 1500, 1455, 1380, 1370, 1285, 1130, 1100, and 1065 cm⁻¹; ¹H-n.m.r. (CDCl₃): δ 7.40-7.26 (m, 5 H, benzyl), 4.67, 4.65 (AB q, 2 H, J 11.4 Hz, PhCH₂), 4.69 (d, 1 H, J 1.6 Hz, H-1), 3.96 (ddd, 1 H, J 1.6, 1.9, 3.2 Hz, H-2), 3.69 (dd, 1 H, J 3.2, 9.6 Hz, H-3), 3.50 (dq, 1 H, J 4.1, 6.1 Hz, H-5), 3.38 (dd, 1 H, J 4.1, 9.6 Hz, H-4), 3.33 (s, 3 H, OCH₃), and 1.31 (d, 3 H, J 6.1 Hz, (CH_3) ; c.i.m.s. m/z 266 (M + 1 – N₂, 27%), 234 (M + 1 – N₂ and CH₃OH, 48%), and 222 (100%).

Anal. Calc. for C₁₄H₁₉N₃O₄: C, 57.34; H, 6.48; N, 14.33. Found: C, 57.38; H, 6.66; N, 14.67.

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