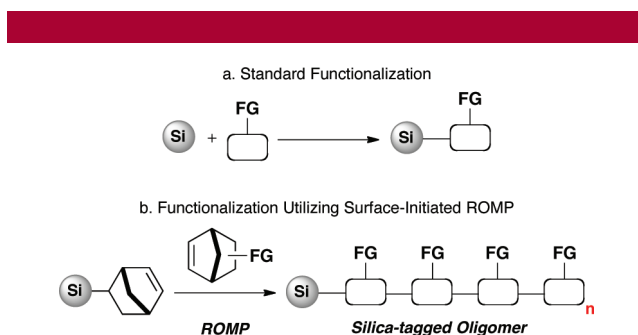




co-workers.<sup>8</sup> Such hybrid nanomaterials combine the physical properties of the inorganic shell (particle size, pore, and shape) with the tunable properties of the grafted organic polymer.<sup>9</sup> Key reports have demonstrated the potential scope of surface-initiated ROM polymerization for the grafting of organic polymers from inorganic nanoparticles,<sup>10</sup> carbon nanotubes,<sup>11</sup> metal surfaces,<sup>12</sup> and resins.<sup>13</sup>

Recently, ROMP-derived oligomeric and polymeric reagents/scavengers have surfaced for application in facilitated synthetic protocols. These reagents possess several inherent characteristics that address classical limitations associated with traditional immobilized reagents, that is, load, tunable properties, heterogeneous reaction kinetics, and so on.<sup>5,14–16</sup> Inspired by these developments, it was envisioned that, by combining the inherent high-load, tunable properties of soluble ROMP-derived oligomers with an insoluble silica support, a new variety of hybrid immobilized functionalized reagents could be achieved. Unlike current methods of functionalizing the surface of silica particles, the application of surface-initiated ROM polymerization addresses the limitation of load by allowing the grafting of multiple monomer units into an oligomeric chain from each Nb-tagged site on the silica particle (Figure 1).

We initially focused on the synthesis of a Si-ROMP-derived bis-acid chloride scavenger (Si-OBAC) by starting with the corresponding Nb-tagged BAC-functionalized monomer utilized for the generation of OBAC soluble oligomers.<sup>17</sup> Previously, simple ROM polymerization of the corresponding Nb-tagged bis-acid chloride monomer **3** using metathesis catalyst [(IMesH<sub>2</sub>)(PCy<sub>3</sub>)(Cl)<sub>2</sub>Ru=CHPh; cat-B],<sup>18,19</sup> gener-

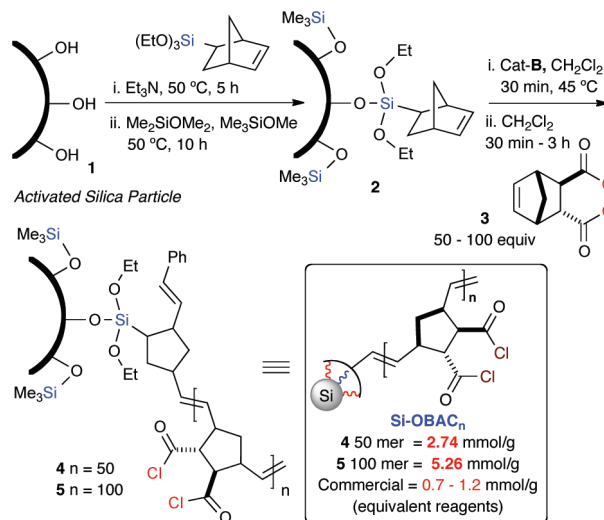


**Figure 1.** (a) Standard functionalization of silica reagents. (b) Functionalization of Nb-tagged silica particles utilizing surface-initiated ROM polymerization.

ated the desired high-load soluble scavenger possessing a theoretical load of 9.1 mmol/g. Therefore, it was envisioned that preactivation of Nb-tagged silica **2** with Grubbs catalyst would generate a catalyst-armed surface (CAS) capable of efficient polymerization off the silica surface with Nb-tagged monomer **3** while maintaining the high-load nature of the reagent.

Utilizing a protocol reported by Buchmeiser and co-workers in 2000,<sup>8</sup> activated spherical silica **1** (70 Å, 20 μm particle size) was tagged with 5-(bicycloheptenyl)-triethoxy silane, followed by capping with trimethoxymethylsilane and dimethoxydimethylsilane to yield the desired Nb-tagged silica (Si–Nb) **2** (Scheme 1). With this tagged nanoparticle in

**Scheme 1.** Synthesis of Silica-Grafted Oligomeric Bis-Acid Chloride **4** and **5** (Si-OBAC)



hand, the surface was armed with metathesis catalyst cat-B (0.6–0.8 equiv), followed by addition of the Nb-tagged BAC monomer **3** to rapidly generate the desired hybrid material, Si-OBAC<sub>50</sub> **4**.<sup>20,21</sup> A number of key parameters were critical, including reaction concentration and prearming of the silica surface with 0.8 equiv of cat-B before the addition of the Nb-tagged monomeric species. This was important to ensure

(7) (a) Graf, C.; Vossen, D. I. J.; Imhof, A.; van Blaaderen, A. *Langmuir* **2003**, *19*, 6693–6700. (b) Schneider, G.; Decher, G. *Nano Lett.* **2004**, *4*, 1833–1839.

(8) (a) Buchmeiser, M. R.; Atzl, N.; Bonn, G. K. *J. Am. Chem. Soc.* **1997**, *119*, 9166–9174. (b) Ambrose, D.; Fritz, J. S.; Buchmeiser, M. R.; Altz, N.; Bonn, G. K. *J. Chromatogr. A* **1997**, *786*, 259–268. (c) Buchmeiser, M. R.; Sinner, F.; Mupa, M.; Wurst, K. *Macromolecules* **2000**, *33*, 32–39. (d) Eder, K.; Reichel, E.; Schottenberger, H.; Huber, C. G.; Buchmeiser, M. R. *Macromolecules* **2001**, *34*, 4334–4341.

(9) Kim, N. Y.; Jeon, N. L.; Choi, I. S.; Takami, S.; Harada, Y.; Finnie, K. R.; Girolami, G. S.; Nuzzo, R. G.; Whitesides, G. M.; Laibinis, P. E. *Macromolecules* **2000**, *33*, 2793–2795.

(10) (a) Kong, B.; Lee, J. K.; Choi, I. S. *Langmuir* **2007**, *23*, 6761–6765. (b) Faulkner, C. J.; Fischer, R. E.; Jennings, G. K. *Macromolecules* **2010**, *43*, 1203–1209.

(11) Jeong, W.; Kessler, M. R. *Chem. Mater.* **2008**, *20*, 7060–7068.

(12) Rutenberg, I. M.; Scherman, O. A.; Grubbs, R. H.; Jiang, W.; Garfunkel, E.; Bao, Z. *J. Am. Chem. Soc.* **2004**, *126*, 4062–4063.

(13) Lapinte, V.; Montembault, V.; Houdayer, A.; Fontaine, L. J. *Mol. Catal. A: Chem.* **2007**, *276*, 219–225.

(14) (a) Long, T.; Maity, P. K.; Samarakoon, T. B.; Hanson, P. R. *Org. Lett.* **2010**, *12*, 2904–2907. (b) Rolfe, A.; Probst, D.; Volp, K.; Omar, I.; Flynn, D.; Hanson, P. R. *J. Org. Chem.* **2008**, *73*, 8785–8790. (c) Stoianova, D. S.; Yao, L.; Rolfe, A.; Samarakoon, T.; Hanson, P. R. *Tetrahedron Lett.* **2008**, *49*, 4553–4555. (d) Herpel, R. H.; Vedantham, P.; Flynn, D. L.; Hanson, P. R. *Tetrahedron Lett.* **2006**, *47*, 6429–6432. (e) Harned, A. M.; He Song, H.; Toy, P. H.; Flynn, D. L.; Hanson, P. R. *J. Am. Chem. Soc.* **2005**, *127*, 52–53. (f) Barrett, A. G. M.; Hopkins, B. T.; Love, A. C.; Tedeschi, L. *Org. Lett.* **2004**, *6*, 835–837. (g) Arstad, E.; Barrett, A. G. M.; Tedeschi, L. *Tetrahedron Lett.* **2003**, *44*, 2703–2707.

(15) (a) Barrett, A. G. M.; Cramp, S. M.; Roberts, R. S. *Org. Lett.* **1999**, *1*, 1083–1086. (b) Roberts, R. S. *J. Comb. Chem.* **2005**, *7*, 21–32.

(16) (a) Zhang, M.; Flynn, D. L.; Hanson, P. R. *J. Org. Chem.* **2007**, *72*, 3194–3198. (b) Harned, A. M.; Sherrill, W. M.; Flynn, D. L.; Hanson, P. R. *Tetrahedron* **2005**, *61*, 12093–12099. (c) Zhang, M.; Moore, J. D.; Flynn, D. L.; Hanson, P. R. *Org. Lett.* **2004**, *6*, 2657–2660.

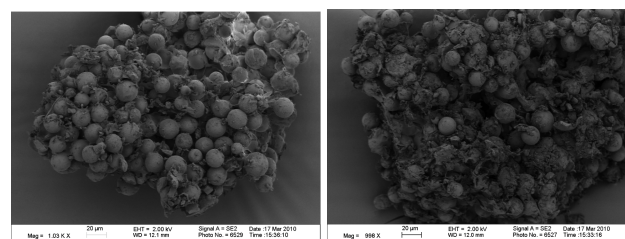
(17) Moore, J. D.; Byrne, R. J.; Vedantham, P.; Flynn, D. L.; Hanson, P. R. *Org. Lett.* **2003**, *5*, 4241–4244.

that all the metathesis catalyst was attached from the Nb-tagged silica surface and potentially preventing polymerization of the Nb-tagged monomer **3** by any free catalyst in solution.<sup>20</sup>

After optimization of the reaction conditions, Si-OBAC<sub>50</sub> **4** possessing a theoretical load of 2.74 mmol/g, was readily accessed as a free-flowing solid on the gram scale.<sup>22,23</sup> Utilizing the inherent tunable characteristics of ROMP-derived oligomers, the theoretical load of this hybrid material was further increased by extending the oligomeric chain length to 100 monomeric units (100-mer), yielding Si-OBAC<sub>100</sub> **5** possessing a theoretical load of 5.26 mmol/g (Scheme 1). This key advancement exploits the inherent properties of Si-ROMP hybrid materials, whereby simply increasing the ratio of monomer **3** to silica **2** (i.e., 50:1 to 100:1) increases the polymer chain length and hence gives rise to increased theoretical loads. Furthermore, Si-ROMP scavengers **4** and **5** are highly favorable in comparison to commercially available Si-immobilized scavengers of nucleophiles with loads ranging from 0.7–1.2 mmol/g.<sup>24</sup> Taken collectively, this represents a significant cost savings and environmental impact to the field.

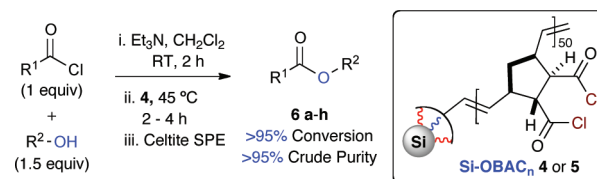
When scanning electron microscopy (SEM) was used, detailed images of the hybrid materials surface morphology were obtained. Notably, the comparison between Si-OBAC<sub>50</sub> **4** and the longer chained Si-OBAC<sub>100</sub> **5** is visually noticed by the SEM images (Figure 2).

With Si-OBAC<sub>50</sub> **4** in hand, the hybrid material was evaluated for the scavenging of nucleophilic species. A variety of alcohols were benzoylated to yield the corresponding esters **6a–h**, whereby excess alcohol (0.5 equiv) was scavenged efficiently with Si-OBAC<sub>50</sub> **4** (Scheme 2).<sup>25</sup> When 1.0 equiv of Si-OBAC<sub>50</sub> **4** was used, the desired esters **6a–h** were isolated in high crude purity and conversion after simple filtration thru a Celite SPE, demonstrating the efficient ability



**Figure 2.** SEM images of Si-OBAC<sub>50</sub> **4** 50-mer (left) and Si-OBAC<sub>100</sub> 100-mer **5** (right).

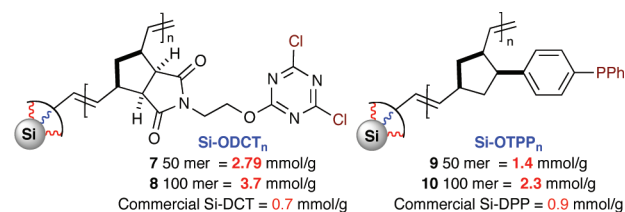
**Scheme 2.** Benzoylation Reactions Utilizing Si-OBAC<sub>50</sub> **4** To Scavenge Excess Alcohol



of **4** to work as a facile scavenger without the need for conventional silica chromatography. Identical results were observed when utilizing the high-load, hybrid Si-OBAC<sub>100</sub> **5**, demonstrating comparable scavenging efficiency to the previously reported oligomeric OBAC derivative.<sup>17</sup>

Building on these results, the project was expanded to the synthesis of additional Si-ROMP hybrid reagents and scavengers. In this regard, investigations focused on the synthesis of the corresponding Si-hybrid materials

Si-ODCT<sub>50</sub> **7** and Si-OTPP<sub>50</sub> **9**, synthesized via the grafting of the corresponding Nb-tagged dichlorotriazine,<sup>14b</sup> and Nb-tagged triphenylphosphine (Figure 3).<sup>14c</sup> When the same



**Figure 3.** Synthesis of silica-grafted oligomeric dichlorotriazine **7** and **8** (Si-ODCT<sub>50</sub>) and triphenylphosphine **9** and **10** (Si-OTPP<sub>50</sub>).

protocol reported for the synthesis of Si-OBAC<sub>50</sub> **4** (Scheme 1) was used, Si-ODCT<sub>50</sub> **7** (theoretical load = 2.79 mmol/g) and Si-OTPP<sub>50</sub> **9** (theoretical load = 1.4 mmol/g) were successfully isolated as free-flowing powders. The theoretical load of both reagents was further increased by extending the oligomeric length to 100 monomer units, yielding the corresponding materials Si-ODCT<sub>100</sub> **8** (3.7 mmol/g)<sup>26</sup> and Si-OTPP<sub>100</sub> **10** (2.3 mmol/g), respectively.<sup>24</sup> Both reagents

(18) Scholl, M.; Ding, Lee, C. W.; Grubbs, R. H. *Org. Lett.* **1999**, *1*, 953–956.

(19) A variety of metathesis catalysts were investigated, including [(PCy<sub>3</sub>)<sub>2</sub>(Cl)<sub>2</sub>Ru=CHPh; cat-A], [(IMesH<sub>2</sub>)(PCy<sub>3</sub>)(Cl)<sub>2</sub>Ru=CHPh; cat-B], and the Hoveyda-Grubbs second-generation catalyst. In all cases studied, cat-B was ideal in both thermal stability and initiation rate.

(20) Si-OBAC<sub>50</sub> denotes the inorganic surface [Si], followed by the ROMP-derived oligomer grafted [OBAC], followed by the theoretical number of Nb monomer units grafted.

(21) We have previously found that there is a good correlation between the mol % of Grubbs catalyst added and the Gaussian distribution of oligomers formed, which is the case with several other oligomers formed [see ref 5d]. We have made these reagents several times with good reproducibility and consistency. MALDITOF and GPC data are normally attained on all oligomers; however, both methods have failed to give good results for previously published reactive oligomers.

(22) Control reaction was run with activated silica, 50 equiv of OBAC monomer in CH<sub>2</sub>Cl<sub>2</sub> (0.8 M), for 30 min. After such time, the resulting crude mixture did not gel as observed for Si-OBAC hybrid materials and was instead isolated via precipitation in Et<sub>2</sub>O.

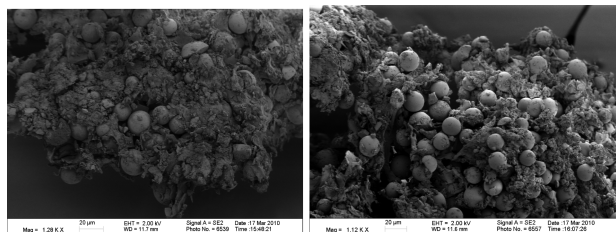
(23) OBAC load stated as mmol/g of acid chloride (ROCl) and further details are reported in ref 17. For load calculation, see Supporting Information S4.

(24) www.biotage.com, www.silicycle.com, and www.fluorous.com (accessed June 2010).

(25) In comparison to their soluble variants (OBAC, ref 17 and ODCT ref 14b) under identical conditions, no difference in reaction stoichiometry and kinetics was observed. Currently there is no commercially available silica immobilized bis-acid chloride variant for direct comparison, however, Si-ROMP reagents reported within this manuscript produced similar conversions/yields under identical conditions to their soluble variants (OBAC and ODCT).



are highly favorable in comparison to commercially available immobilized silica reagents.<sup>24</sup> When SEM was used, detailed images of **7** and **9** were obtained, demonstrating once again the morphology of the hybrid materials obtained via surface-initiated ROMP off the spherical silica particles (Figure 4).



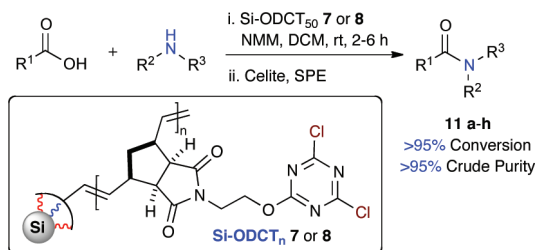
**Figure 4.** SEM images of Si-ODCT<sub>50</sub> 50-mer **7** (left) and (b) Si-OTPP<sub>50</sub> 50-mer **9** (right).

We have previously reported the versatile utilization of oligomeric dichlorotriazine (ODCT) as an efficient coupling reagent, scavenger, and activator in parallel synthesis.<sup>14b</sup> In this regard, Si-ODCT<sub>50</sub> was investigated as a coupling reagent for the synthesis of amides **11a–h** from simple acids and amines under mild conditions (Scheme 3).

The desired amides **11 a–h** were efficiently coupled together utilizing Si-ODCT<sub>50</sub> **7** in high conversion and crude purity without the need for standard chromatography. It is key to note that Si-ODCT<sub>50</sub> **7** demonstrated the same efficiency in comparison to the soluble oligomeric version ODCT for the formation of amides.<sup>26</sup>

In conclusion, the combination of Nb-tagged silica particles and functionalized Nb-tagged monomers efficiently yields high-load, hybrid Si-ROMP reagents. A catalyst-armed

### Scheme 3. Amide Coupling Reactions Utilizing Si-ODCT



surface (CAS)-initiated polymerization was key to functionalization of units off the silica particle surface. When this technology was used, a high load Si-ROMP derived bis-acid chloride (Si-OBAC) scavenger was successfully synthesized and utilized in batch. Additionally, high-load, Si-ROMP-derived dichlorotriazine (Si-ODCT) and triphenylphosphine (Si-OTP) were readily prepared. SEM imaging was utilized to demonstrate the successful grafting of the corresponding oligomer and the inherent morphology of the hybrid materials. Current efforts are focusing on the application and utilization of high load Si-ROMP reagents in parallel synthesis and continuous flow-through protocols for in-line purification and diversification of small molecules.

**Acknowledgment.** This research was made possible by the National Institute of General Medical Sciences [NIH-STTR R41 GM076765, NIH P050-GM069663]. We acknowledge Dr. David S. Moore and Ms. Heather Shinogle for carrying out SEM analysis (KU). We thank Materia Inc. for providing metathesis catalyst.

**Supporting Information Available:** Experimental details and spectral characterization for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL102239H

(26) For comparison to reported uses of silica-immobilized dichlorotriazine, see: Kowalski, J. A.; Leonard, S. F.; Lee, G. E., Jr. *J. Comb. Chem.* **2006**, *8*, 774–779.