An Aryl Stevens Migration

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Summary A Stevens rearrangement in which the migrating group is α -naphthyl has been observed in high yield.

The recent communication of a Stevens rearrangement with aryl migration prompts us to report our findings in this area. Aryl migration occurs in high yield when the migrating aryl

grouping is α -naphthyl, *i.e.*, treatment of α -naphthyltrimethylammonium iodide (I) with two equivalents of sodium amide in liquid ammonia under reflux, yields quantitatively a mixture of 76% α -naphthylmethyldimethylamine (II), 23% naphthalene, and 1% or less of the dealkylation product, NN-dimethyl- α -naphthylamine (III).

The fact that naphthalene is observed in conjunction with rearrangement product is consistent with the following ion-pair mechanism:2

However, a radical-pair mechanism³ would also account for these products.

When β -naphthyltrimethylammonium iodide was subjected to the rearrangement conditions, the major product was β -naphthylamine, apparently resulting from NH₃ addition to naphthyne. No rearranged product could be

detected by n.m.r. as shown by the total absence of signals in the vicinity of $\delta 2.1$. A trace of the dealkylation product NN-dimethyl- β -naphthylamine could be detected by n.m.r. The α -hydrogen of naphthalene is more acidic than the β-hydrogen, possibly accounting in part for naphthyne formation being dominant with the β -quaternary salt.

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¹ H. Heaney and T. J. Ward, Chem. Comm., 1969, 810. ² S. W. Kantor and C. R. Hauser, J. Amer. Chem. Soc., 1951, 73, 4122.

W. Kalltof and C. R. Hadsel, J. Amer. Chem. 301, 133, 1422.
U. Schöllkopf, U. Ludwig, G. Ostermann, and M. Patsch, Tetrahedron Letters, 1969, 39, 3415.
E. N. Yurigina, P. P. Alikhanov, E. A. Izrailevich, P. N. Manochkina, and A. I. Shatenstein, Zhur. fiz. Khim., 1960, 34, 587;
A. I. Shatenstein, Adv. Phys. Org. Chem., 1963, 1, 182.