

Reaction of Methyl-4-methylene-2,3-O-isopropylidene- β -D-ribofuranoside with *N*-Bromosuccinimide in Aqueous Tetrahydrofuran

N. A. Ivanova, Z. R. Valiullina, O. V. Shitikova, and M. S. Miftakhov

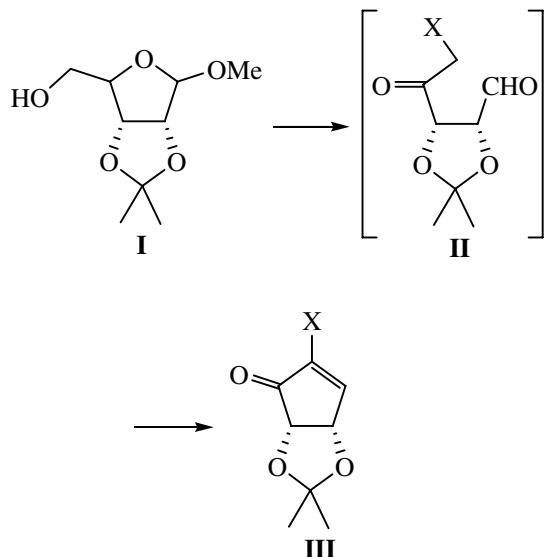
Institute of Organic Chemistry, Ufa Scientific Center, Russian Academy of Sciences, Ufa, 450054 Russia
e-mail: bioreg@anrb.ru

Received May 27, 2006

Abstract—Methyl-4-methylene-2,3-O-isopropylidene- β -D-ribofuranoside prepared from D-ribose reacted in a system NBS–THF–H₂O to give a mixture of stereoisomeric products of regioselective bromohydroxylation of a double bond. The reaction involved a hydrolysis of the glycoside bond, but the acetonide protective group was retained. The mechanism of the selective hydrolysis originating from the ring-chain tautomerism of bromohydrins obtained was proved by the ¹H NMR spectra of the stereoisomeric methyl-5-deoxy-5-bromo-4-hydroxy-2,3-O-isopropylidene- β -D-ribofuranosides. By crotonic cyclization of the formed masked 1,4-dicarbonyl compounds at heating in benzene in the presence of neutral Al₂O₃ a new chiral cyclopentenone block, 2-bromo-4,5-isopropylideneoxycyclopent-2-en-1-one, was obtained in a low yield.

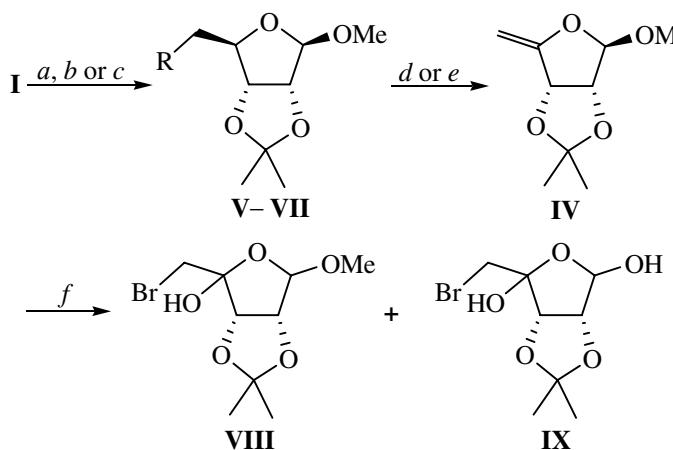
DOI: 10.1134/S1070428007050168

Chiral cyclopentenones prepared from sugars are used in the synthesis of cyclopentanoids (prostanoids, carbocyclic nucleosides etc) [1–5]. In extension of studies in this field we planned to prepare from an available D-ribose derivative I [6] chiral cyclopentenones III (X is an electron-acceptor group). As seen from the scheme, the key reaction stage is based on Knoevenagel intramolecular cyclization of compounds II containing an activated methylene group into bicyclic cyclopentenones III (iodine derivative III is mentioned in [7]).



Here we report on an approach tested for a hypothetical 1,4-ketoaldehyde II (X = Br). Taking into account that activated 1,4- α -hydroxy(halo)carbonyl compounds similar to substances II are prone to hydration, oligomerization, furanization, (and epimerization!) it seemed feasible to generate them prior to cyclization from enol ether IV available from methoxy derivative I [8]. For preparation of compound IV from alcohol I the latter was converted by known methods into bromide V and iodide VI [9], and also into tosylate VII. However at heating with DBU (80°C, benzene) bromide V did not change, whereas under the same conditions iodide VI was converted to enol ether IV. Regretfully, the *R*_f values of iodide VI and enol ether IV were too close hampering the reaction progress monitoring by TLC and chromatographic purification of the product from residual iodide VI. We avoided these difficulties by reacting tosylate VII with *t*-BuOK to obtain enol ether IV in 65–70% yield.

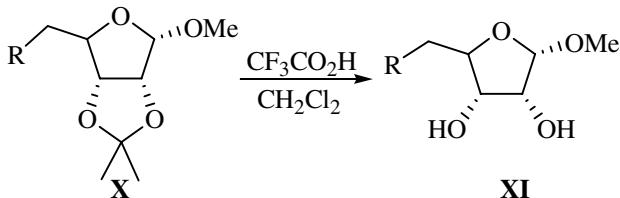
We regarded enol ethers as highly reactive latent equivalents of oxo functions convenient for regiodirected introduction of an electrophilic moiety X⁺ into the α -position with respect to a carbonyl, and thus we studied a reaction of enol ether IV with *N*-bromosuccinimide. The bromohydroxylation in the presence of NBS in a mixture THF–H₂O (3:1) proceeded rapidly (10–15 min)



$R = \text{Br}$ (**V**), I (**VI**), OTs (**VII**). Reagents and conditions: (a) 4.0 equiv CBr_4 , 1.1 equiv PPh_3 , MeCN , 20°C , 30 min (95%); (b) 2.0 equiv I_2 , 1.25 equiv PPh_3 , 1.5 equiv Im , PhMe , MeCN , 70°C , 20 min (76%); (c) 1.5 equiv TsCl , Py , 20°C , 20 h (80%); (d) 1.1 equiv DBU , 90°C , 30 min (68%); (e) 1.2 equiv $t\text{-BuOK}$, THT , $0 \rightarrow 20^\circ\text{C}$, 1.5 h (67%); (f) 1.1 equiv NBS , $\text{THT}-\text{H}_2\text{O}$, 3:1, 20°C , 10 min (98%).

and yielded quantitatively isomeric mixtures of bromohydrins **VIII** and **IX** in a ratio 2:1. A striking feature of this reaction was the formation of deblocked acetal **IX** in considerable amounts within this short time. Methoxybromohydrin **VIII** was converted completely into the corresponding oxybromohydrin **IX** within 12 h at the use of 10 equiv of NBS.

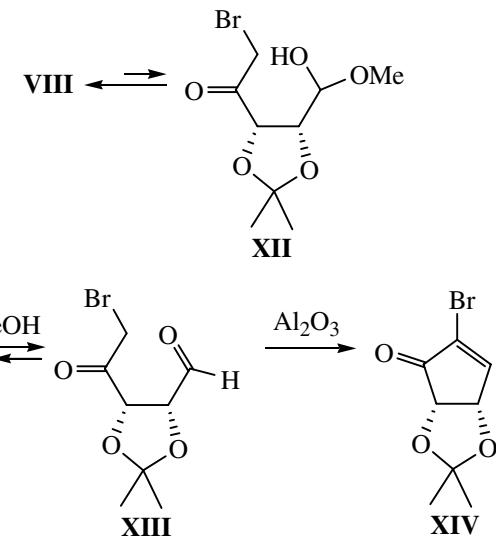
The selective hydrolysis of the glycoside bond with the retention of the isopropylidene protection is obviously of a synthetic interest. The synthetic blocks of sugars containing in the structure both acetonide and methylacetal combination of protective groups, like in compounds **I**, **IV–VII**, are known to be most common among the hydrocarbon synthons. Usually under typical stringent conditions of water-acid hydrolysis an exhaustive hydrolysis occurs of both protective groups. We found in the literature only examples of selective hydrolysis of acetonide groups with retention of the other protection. In [10] α -D-ribofuranoside (**X**) was successfully converted into diol **XI** by a selective hydrolysis of the acetonide group in a mixture $\text{CF}_3\text{CO}_2\text{H}-\text{CH}_2\text{Cl}_2$ without affecting the glycoside. In the β -anomer the acetonide group proved to be stable. The selective hydro-



lysis of acetonide in compound **X** was ascribed to an anchimeric assistance to the hydrolysis of the oxygen atom of the *cis*-oriented glycoside methoxy group. Other examples of chemo- and regioselective deblocking of the acetonide protective group were described for polyhydroxy compounds at the use of heterogeneous catalyst $\text{NaHSO}_3\cdot\text{SiO}_2$ [11], of BiCl_3 [12], $\text{La}(\text{NO}_3)_3\cdot 6\text{H}_2\text{O}$ [13], etc. [12].

The selective hydrolysis of the C^1 -glycoside bond we observed was likely to originate from the ring-chain tautomerism of bromohydrin **VIII**. As seen from the scheme, acyclic form **XII** is easily converted into aldehyde **XIII** whose hydrate gives bromohydrin **IX**.

This suggestion is supported by the data of ^1H NMR spectroscopy that has revealed the presence of four stereoisomeric methoxybromohydrins **VIII** which were isolated in pairs by column chromatography on SiO_2 . The comprehensive analysis of the spectra of methoxybromohydrins **VIII** will be published elsewhere. We failed to isolate individual oxybromohydrins **IX**.



The intramolecular cyclization of compounds **VIII** and **IX** by aldol-crotonic route at heating in benzene in the presence of neutral Al_2O_3 [14] (under optimum conditions for 1,4-dioxo compounds like diol **XI**) led to the formation in low yields (10–20%) of target enone **XIV**. We plan to carry on the search for cyclization conditions of isomeric tetrahydrofurans **IX** by catalysis with protonic and Lewis acids.

EXPERIMENTAL

IR spectra were recorded on spectrophotometers UR-20 and Specord M-80 from films or mulls in mineral oil.

NMR spectra were registered on a spectrometer Bruker AM-300 at operating frequencies 300.13 (¹H) and 75.47 MHz (¹³C) from solutions in CDCl₃ using solvent signals as internal reference (δ_{H} 7.27, δ_{C} 77.00 ppm). The reaction progress was monitored by TLC (Silufol, petroleum ether–ethyl acetate, CH₂Cl₂–MeOH), spots were visualized by 10% solution of anise aldehyde in ethanol with sulfuric acid added. The optical rotation was measured on a polarimeter Perkin Elmer Polarimetre 241-M.

Methyl-2,3-O-isopropylidene-β-D-ribofuranoside (I). To a mixture of 12.40 g of CuSO₄, 5.85 g (38.90 mmol) of D-ribose, 110 ml of anhydrous acetone, and 32 ml of anhydrous MeOH was added dropwise 0.2 ml of concn. H₂SO₄. The reaction mixture was stirred at 40°C for 48 h (TLC monitoring); CuSO₄ was filtered off, the precipitate was washed with a mixture acetone–MeOH, 1:1 v/v. The filtrate was neutralized with a saturated NaHCO₃ solution and evaporated. The residue was extracted with ethyl acetate, the extract was washed with H₂O, saturated solution of NaCl, dried with Na₂SO₄, and the solvent was evaporated in a vacuum. On distilling the residue we obtained 7.95 g (75%) of compound I, bp 110°C (2 mm Hg), R_f 0.44 (CH₂Cl₂–MeOH, 9:1), $[\alpha]_D^{20}$ −75° (c 1, CHCl₃). IR spectrum, ν , cm^{−1}: 1020, 1056, 3485. ¹H NMR spectrum, δ , ppm (J , Hz): 1.32 s (3H, Me), 1.50 s (3H, Me), 3.25 d.d (1H, OH, $^3J_{\text{OH},5B}$ 2.8, $^3J_{\text{OH},5A}$ 10.0), 3.36 s (3H, OMe), 3.62 t.d (1H, H^{5A}, $^3J_{5A,4}$ 3.5, $^2J_{5A,5B}$ = $^3J_{5A,\text{OH}}$ = 10.0), 3.70 d.d.d (1H, H^{5B}, $^3J_{5B,\text{OH}}$ 2.8, $^3J_{5B,4}$ 2.8, $^2J_{5B,5A}$ 10.0), 4.45 d.d (1H, H⁴, $^3J_{4,5B}$ 2.8, $^3J_{4,5A}$ 3.5), 4.60 d (1H, H², $^3J_{2,3}$ 6.0), 4.83 d (1H, H³, $^3J_{3,2}$ 6.0), 4.9 C (1H, H¹). ¹³C NMR spectrum, δ , ppm: 24.60 (Me), 26.25 (Me), 55.35 (OMe), 63.64 (C⁵), 81.40 (C²), 85.65 (C³), 88.16 (C⁴), 109.79 (C¹), 112.00 (Cⁱ-Pr). Found, %: C 52.79; H 7.88. C₉H₁₆O₅. Calculated, %: C 52.93; H 7.90.

Methyl-5-deoxy-5-bromo-2,3-O-isopropylidene-β-D-ribofuranoside (V). To a mixture of 0.50 g (2.45 mmol) of compound I and 0.97 g (3.70 mmol) of Ph₃P in anhydrous acetonitrile was added at room temperature 1.23 g (9.70 mmol) of CBr₄, and the mixture was stirred for 30 min (TLC monitoring). The precipitate was filtered off, the solution was evaporated, and the residue was subjected to chromatography on SiO₂ (CH₂Cl₂). Yield 0.62 g (95%), colorless oily substance, R_f 0.30 (petroleum ether–ethyl acetate, 9:1), $[\alpha]_D^{20}$ −74.2° (c 1, CHCl₃). IR spectrum, ν , cm^{−1}: 1040, 1070, 1080, 1105. ¹H NMR spectrum, δ , ppm (J , Hz): 1.33 s (3H, Me), 1.49 s (3H, Me), 3.32 t (1H, H^{5A}, $^2J_{5A,5B}$ 10.0), 3.35 s

(3H, OMe), 3.44 d.d (1H, H^{5B}, $^3J_{5B,4}$ 5.9, $^2J_{5B,5A}$ 10.0), 4.49 d.d (1H, H⁴, $^3J_{4,5B}$ 5.9, $^3J_{4,5A}$ 10.0), 4.62 d (1H, H², $^3J_{2,3}$ 6.0), 4.77 d (1H, H³, $^3J_{3,2}$ 6.0), 5.0 c (1H, H¹). ¹³C NMR spectrum, δ , ppm: 24.90 (Me), 26.43 (Me), 32.48 (C⁵), 55.13 (OMe), 82.59 (C²), 85.13 (C³), 86.63 (C⁴), 109.53 (C¹), 112.71 (Cⁱ-Pr). Found, %: C 40.39; H 5.54; Br 29.79. C₉H₁₅BrO₄. Calculated, %: C 40.47; H 5.66; Br 29.91.

Methyl-5-deoxy-5-iodo-2,3-O-isopropylidene-β-D-ribofuranoside (VI). To a stirred mixture of 1.00 g (4.89 mmol) of compound I, 1.60 g (6.11 mmol) of Ph₃P, and 0.49 g (7.33 mmol) of imidazole dissolved in a mixture of 15 ml of toluene and 2.5 ml of acetonitrile at 70°C was added by portions 1.56 g (9.75 mmol) of fine crystalline iodine. The reaction mixture was stirred for 20 min, diluted with ethyl acetate, washed with a saturated solution of Na₂S₂O₃ and with H₂O, and dried with Na₂SO₄. On evaporating the solvent in a vacuum the residue was subjected to chromatography on a column packed with SiO₂ (petroleum ether). Yield 1.18 g (76%), colorless oily substance, R_f 0.22 (petroleum ether–ethyl acetate, 9:1), $[\alpha]_D^{20}$ −79.8° (c 1, CHCl₃). IR spectrum, ν , cm^{−1}: 1020, 1065, 1080, 1095. ¹H NMR spectrum, δ , ppm (J , Hz): 1.32 s (3H, Me), 1.47 s (3H, Me), 3.15 t (1H, H^{5A}, $^2J_{5A,5B}$ 10.0), 3.27 d.d (1H, H^{5B}, $^3J_{5B,4}$ 6.0, $^2J_{5B,5A}$ 10.0), 3.36 s (3H, OMe), 4.42 d.d (1H, H⁴, $^3J_{4,5B}$ 6.0, $^3J_{4,5A}$ 10.0), 4.62 d (1H, H², $^3J_{2,3}$ 6.0), 4.74 d (1H, H³, $^3J_{3,2}$ 6.0), 5.05 s (1H, H¹). ¹³C NMR spectrum, δ , ppm: 6.64 (C⁵), 24.82 (Me), 26.22 (Me), 54.98 (OMe), 82.77 (C²), 85.10 (C³), 87.15 (C⁴), 109.40 (C¹), 112.30 (Cⁱ-Pr). Found, %: C 34.58; H 4.95; I 40.23. C₉H₁₅IO₄. Calculated, %: C 34.41; H 4.81; I 40.40.

Methyl-2,3-O-isopropylidene-5-O-tosyl-β-D-ribofuranoside (VII). To a stirred solution of 2.0 g (9.79 mmol) of alcohol I in 15 ml of pyridine was added at 0°C by portions 2.8 g (14.69 mmol) of TsCl. The reaction mixture was stirred at room temperature for 20 h (TLC monitoring), then it was poured into cold water, the reaction product was extracted into chloroform, the extract was dried with Na₂SO₄, and evaporated. The residue was purified by column chromatography on SiO₂ (CH₂Cl₂) or by recrystallization from petroleum ether–ethyl acetate, 1:1, to obtain 2.8 g (80%) of tosylate VII as colorless crystals, mp 80–81°C, R_f 0.18 (petroleum ether–ethyl acetate, 8:2), $[\alpha]_D^{20}$ −48.7° (c 1, CHCl₃). IR spectrum, ν , cm^{−1}: 814, 838, 1090, 1180, 1354, 1594. ¹H NMR spectrum, δ , ppm (J , Hz): 1.21 s (3H, Me), 1.43 s (3H, Me), 2.44 s (3H, Me_{arom}), 3.22 s (3H, OMe), 3.97 d.d (1H, H^{5A}, $^3J_{5A,4}$ 7.2, $^2J_{5A,5B}$ 10.2), 4.03 d.d (1H,

H^{5B} , $^3J_{5B,4}$ 7.2, $^2J_{5B,5A}$ 10.2), 4.3 t (1H, H^4 , $^3J_{4,5}$ 7.2), 4.52 d (1H, H^3 , $^3J_{3,2}$ 5.9), 4.59 d (1H, H^2 , $^3J_{2,3}$ 5.9), 4.92 s (1H, H^1), 7.35 d (2H, H^O , $^3J_{o,m}$ 8.3), 7.79 d (2H, H^m , $^3J_{m,o}$ 8.3). ^{13}C NMR spectrum, δ , ppm: 21.67 (Me_{arom}), 24.85 (Me), 26.32 (Me), 55.03 (OMe), 69.24 (C^5), 81.35 (C^3), 83.57 (C^4), 84.87 (C^2), 109.45 (C^I), 112.69 (C^{i-Pr}), 127.99 (C^O), 129.96 (C^m), 132.70 (C^O), 145.12 (C^O). Found, %: C 53.55; H 6.12; S 8.83. $C_{16}H_{22}O_7S$. Calculated, %: C 53.62; H 6.19; S 8.95.

Methyl-2,3-O-isopropylidene-4-methylene- β -D-erythofuranoside (IV). *a.* A mixture of 0.20 g (0.64 mmol) of iodide VI and 0.1 g (0.70 mmol) of DBU was stirred at 90°C for 30 min. The reaction mixture was subjected to column chromatography on SiO_2 (petroleum ether–ethyl acetate, 98:2) to obtain 0.08 g (68%) of enol ether IV.

b. To a solution of 0.46 g (1.28 mmol) of tosylate VII in 15 ml of anhydrous THF was added at 0°C while stirring 0.22 g (1.92 mmol) of *t*-BuOK. The reaction mixture was stirred at room temperature for 1.5 h (TLC monitoring), the precipitate was filtered off, and the solution was evaporated in a vacuum. The residue was purified by column chromatography on SiO_2 (petroleum ether–ethyl acetate, 95:5). Yield 0.16 g (67%), R_f 0.22 (petroleum ether–ethyl acetate, 9:1), $[\alpha]_D^{20} +55.2^\circ$ (*c* 1.15, $CHCl_3$). IR spectrum, ν , cm^{-1} : 890, 1060, 1085, 1670, 3085. 1H NMR spectrum, δ , ppm (J , Hz): 1.35 s (3H, Me), 1.48 s (3H, Me), 3.41 s (3H, OMe), 4.38 br.s (1H, H^{5A}), 4.49 d (1H, H^3 , $^3J_{3,2}$ 5.90), 4.59 br.s (1H, H^{5B}), 5.95 d (1H, H^2 , $^3J_{2,3}$ 5.90), 5.10 s (1H, H^1). ^{13}C NMR spectrum, δ , ppm: 25.73 (Me), 26.73 (Me), 55.65 (OMe), 78.69 (C^3), 82.66 (C^2), 88.70 (C^5), 108.35 (C^I), 113.20 (C^{i-Pr}), 161.23 (C^4). Found, %: C 57.98; H 7.39. $C_9H_{14}O_4$. Calculated, %: C 58.05; H 7.58.

Reaction of enol ether IV with *N*-bromo-succinimide. To a solution of 0.2 g (1.07 mmol) of enol ether IV in 7 ml of a mixture $THF-H_2O$, 3:1, was added 0.21 g (1.2 mmol) of NBS, and the stirring continued for 10 min (TLC monitoring). The reaction mixture was evaporated, the residue was extracted with $CHCl_3$, the extract was washed with a saturated NaCl solution, dried with Na_2SO_4 , and evaporated in a vacuum. The residue was purified by column chromatography on SiO_2 (petroleum ether–ethyl acetate, 95:5) to isolate 0.2 g (66%) of a mixture of methoxybromohydrins VIII as two pairs of stereoisomers, and 0.09 g (32%) of compound IX as three stereoisomers in a ratio 4:2:1.8 (measured by integral intensities of CH_3 peaks in the 1H NMR spectrum).

(4S)-Methyl-5-bromo-4-hydroxy-5-deoxy-2,3-O-isopropylidene- β -D-ribofuranoside (4S, β -VIII). R_f 0.3 (petroleum ether–ethyl acetate, 8:2, 2 runs). IR spectrum, ν , cm^{-1} : 682, 2986, 3420. 1H NMR spectrum, δ , ppm (J , Hz): 1.38 s (3H, Me), 1.56 s (3H, Me), 3.38 s (3H, OMe), 3.40 s (1H, OH), 3.64 d (1H, H^{5A} , $^2J_{5A,5B}$ 10.6), 3.69 d (1H, H^{5B} , $^2J_{5B,5A}$ 10.6), 4.63 d (1H, H^3 , $^3J_{3,2}$ 5.9), 4.70 d (1H, H^2 , $^3J_{2,3}$ 5.9), 4.95 s (1H, H^1). ^{13}C NMR spectrum, δ , ppm: 24.64 (Me), 25.96 (Me), 35.73 (C^5), 55.14 (OMe), 84.44 (C^3), 85.20 (C^2) 104.81 (C^I), 109.98 (C^{i-Pr}), 113.72 (C^4). Found, %: C 38.32; H 5.45; Br 28.37. $C_9H_{15}BrO_5$. Calculated, %: C 38.18; H 5.34; Br 28.22.

(4R)-Methyl-5-bromo-4-hydroxy-5-deoxy-2,3-O-isopropylidene- α -D-ribofuranoside (4R, α -VIII). 1H NMR spectrum, δ , ppm (J , Hz): 1.32 s (3H, Me), 1.47 s (3H, Me), 3.43 m (1H, H^{5A}), 3.45 s (3H, OMe), 3.65 d (1H, H^{5B} , $^2J_{5B,5A}$ 11.0), 4.42 d (1H, OH, $^4J_{OH,5A}$ 1.6), 4.69 d (1H, H^3 , $^3J_{3,2}$ 5.67), 4.80 d (1H, H^2 , $^3J_{2,3}$ 5.67), 5.03 s (1H, H^1). ^{13}C NMR spectrum, δ , ppm: 24.80 (Me), 26.16 (Me), 35.88 (C^5), 55.59 (OMe), 78.72 (C^3), 84.19 (C^2) 104.54 (C^I), 106.06 (C^{i-Pr}), 113.18 (C^4).

(4S)-Methyl-5-bromo-4-hydroxy-5-deoxy-2,3-O-isopropylidene- α -D-ribofuranoside (4S, α -VIII). 1H NMR spectrum, δ , ppm (J , Hz): 1.35 s (3H, Me), 1.48 s (3H, Me), 2.99 d (1H, OH, $^3J_{OH,I}$ 9.8), 3.40 s (3H, OMe), 3.60 d (1H, H^{5A} , $^2J_{5A,5B}$ 11.3), 3.61 d (1H, H^{5B} , $^2J_{5B,5A}$ 11.3), 4.68 d (1H, H^3 , $^3J_{3,2}$ 5.6), 4.71 d (1H, H^2 , $^3J_{2,3}$ 5.6), 5.35 d (1H, H^1 , $^3J_{1,OH}$ 9.8). ^{13}C NMR spectrum, δ , ppm: 24.89 (Me), 26.25 (Me), 35.96 (C^5), 55.68 (OMe), 84.56 (C^3), 85.31 (C^2) 104.61 (C^I), 113.30 (C^{i-Pr}), 113.83 (C^4).

(4R)-Methyl-5-bromo-4-hydroxy-5-deoxy-2,3-O-isopropylidene- β -D-ribofuranoside (4R, β -VIII).

1H NMR spectrum, δ , ppm (J , Hz): 1.41 s (3H, Me), 1.55 s (3H, Me), 3.31 s (3H, OMe), 3.60 d (1H, H^{5A} , $^2J_{5A,5B}$ 11.3), 3.61 d (1H, H^{5B} , $^2J_{5B,5A}$ 11.3), 4.03 d (1H, OH, $^3J_{OH,I}$ 12.5), 4.58 d (1H, H^3 , $^3J_{3,2}$ 5.9), 4.63 d.d (1H, H^2 , $^3J_{2,1}$ 3.6, $^3J_{2,3}$ 5.9), 5.24 d.d (1H, H^1 , $^3J_{1,2}$ 3.6, $^3J_{1,OH}$ 12.5). ^{13}C NMR spectrum, δ , ppm: 24.72 (Me), 25.05 (Me), 35.76 (C^5), 55.22 (OMe), 78.85 (C^2), 84.31 (C^3), 104.96 (C^I), 106.11 (C^{i-Pr}), 110.09 (C^4).

5-Bromo-4-hydroxy-5-deoxy-2,3-O-isopropylidene-D-ribofuranose (IX). R_f 0.11 (petroleum ether–ethyl acetate, 8:2, 2 runs). IR spectrum, ν , cm^{-1} : 634, 2944, 3412. 1H NMR spectrum, δ , ppm (J , Hz): 1.32 s (3H, Me), 1.38 s (3H, Me), 1.40 s (3H, Me), 1.48 s (3H, Me), 1.55 s (3H, Me), 1.56 s (3H, Me), 3.60–3.75 m (6H, H^5), 4.60–4.85 m (6H, H^2 , H^3), 5.40–5.50 m (3H, H^1).

¹³C NMR spectrum, δ, ppm, major isomer: 24.66 (Me), 26.04 (Me), 35.80 (C⁵), 82.93 (C³), 84.39 (C²), 103.68 (C¹), 106.10 (C⁴), 113.18 (Cⁱ-Pr); second isomer: 24.49 (Me), 25.82 (Me), 36.22 (C⁵), 84.71 (C³), 86.16 (C²), 96.00 (C¹), 99.66 (C⁴), 113.70 (Cⁱ-Pr); minor isomer: 24.32 (Me), 25.60 (Me), 36.35 (C⁵), 78.44 (C³), 79.06 (C²), 98.01 (C¹), 103.68 (C⁴), 113.77 (Cⁱ-Pr). Found, %: C 35.58; H 5.04; Br 29.53. C₈H₁₃BrO₅. Calculated, %: C 35.71; H 4.87; Br 29.69.

2-Bromo-4,5-O-isopropylidene-2-cyclopenten-1-one (XIV). To a dispersion of neutral Al₂O₃ in benzene was added under an argon atmosphere 0.68 g of a mixture of bromohydrins **VIII** and **IX** in benzene. The reaction mixture was stirred for 3 h at reflux, then Al₂O₃ was filtered off, and the solution was evaporated. The residue was subjected to a chromatography on a column packed with SiO₂ (petroleum ether–ethyl acetate, 95:5) to isolate 0.08 g (15%) of enone **XIV** as colorless crystals, mp 86.5–88°C, R_f 0.35 (petroleum ether–ethyl acetate, 7:3), [α]_D²⁰+4.4° (c 1.15, CHCl₃). IR spectrum, ν, cm⁻¹: 1582, 1744. ¹H NMR spectrum, δ, ppm (J, Hz): 1.40 s (3H, Me), 1.42 s (3H, Me), 4.58 d (1H, H⁵, 2J_{5,4} 5.4), 5.23 d.d (1H, H⁴, 3J_{4,3} 3.6, 3J_{4,5} 5.4), 7.59 d (1H, H³, 3J_{3,4} 3.6). ¹³C NMR spectrum, δ, ppm: 26.33 (Me), 27.45 (Me), 75.29 (C⁴), 77.52 (C⁵), 115.91 (Cⁱ-Pr), 128.52 (C²), 157.10 (C³), 195.45 (C¹). Found, %: C 41.11; H 3.78; Br 34.19. C₈H₉BrO₃. Calculated, %: C 41.23; H 3.89; Br 34.28.

REFERENCES

1. Ferrier R.J. and Middleton S., *Chem. Rev.* 1993, vol. 93, p. 2779.
2. Chu C.K., Jin Y.H., Baker R.O., and Huggins J., *Bioorg. and Med. Chem. Lett.* 2003, vol. 13, p. 9.
3. Berecibar A., Grandjean C., and Siriwardena A., *Chem. Rev.* 1999, vol. 99, p. 779.
4. Ali S.M., Borchardt K.R., and Borchardt R.T., *Tetrahedron Lett.*, 1990, vol. 31, p. 1509.
5. Elhalem E., Comin M.J., Leitofuter Y., Garcia-Linares G., and Rodrigues J.B., *Tetrahedron: Asymmetry*, 2005, vol. 16, p. 425.
6. Ghosh A.K. and Liu W., *J. Org. Chem.* 1996, vol. 61, p. 6175.
7. Tanaka K., Taniguchi T., and Ogasaware K., *Tetrahedron Lett.* 2001, vol. 24, p. 1049.
8. Hill J.M., Hutchinson E.J., Le Grand D.M., Roberts S.M., Thorpe A.J., and Turner N.J., *J. Chem. Soc., Perkin Trans. 1*, 1994, p. 1483.
9. Lerner L.M., *Carbohydr. Res.* 1977, vol. 53, p. 177.
10. Wakharkar R.D., Sahasrabuddhe M.B., Borate H.B., and Jarjar M.K., *Synthesis*, 2004, vol. 11, p. 1830.
11. Ramn R., Ramesh C., and Das B., *Chem. Lett.* 2003, vol. 32, p. 734.
12. Swamy N.R. and Venkatesvarlu Y., *Tetrahedron Lett.*, 2002, vol. 43, p. 7549.
13. Reddy S.M., Reddy Y.V., and Venkatesvarlu Y., *Tetrahedron Lett.*, 2005, vol. 46, p. 7439.
14. Hudlicky T., Luna H., Barbieri J., and Kwart L.D., *J. Am. Chem. Soc.*, 1988, vol. 110, p. 4735.