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SYNTHESIS OF 2- OR 4,6-DINITRO-sec-PENTYL PHENOLS

G. G. S. DUTTON, M. E. D. HILLMAN, AND J. G. MOFFATT

In previous papers we have reported the synthesis and characterization of dinitron-alkyl and iso-alkyl phenols (1-3), compounds which have insecticidal and selective herbicidal properties. In view of the known superiority of 2,4-dinitro-6-sec-butyl phenol compared to the n-isomer (4,5) it was of interest to prepare the sec-pentyl derivatives for physiological tests which will be reported elsewhere.

The isomeric sec-pentyl phenols were prepared by reacting a carbonyl compound with a

TABLE I Properties of alcohols and derived alkenes

	Alcohols $(C_{12}H_{18}O_2)$		Alkenes $(C_{12}H_{16}O)$		Nitrosyl chloride	
	Boiling point °C (mm)	n^{25}	Boiling point °C (mm)	n^{25}	(9, 10) $(C_{12}H_{16}NO_{2}Cl)$, melting point, °C	
2-(2-Methoxyphenyl)-pentanol-2 2-(4-Methoxyphenyl)-pentanol-2	85 (0.02) 100 (0.05)	1.5129 1.5308	67 (0.4) 70 (0.1)	1.5229 1.5370	93 84	
3-(2-Methoxyphenyl)-pentanol-3 3-(4-Methoxyphenyl)-pentanol-3 (12, 13)	76 (0.01) 87 (0.01)	$1.5160 \\ 1.5278$	70 (0.03) 58 (0.03)	$1.5220 \\ 1.5310$	97 79	
2-(2-Methoxyphenyl)-3-methyl- butanol-2 (14)	68 (0.02)	1.5163	55 (0.1)	1.5204	-	
2-(4-Methoxyphenyl)-3-methyl- butanol-2	107 (0.02)	1.5223	65 (0.1)	1.5310	_	

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TABLE II
Properties of alkylanisoles and their derivatives

	Alkylanisole	Sulphon ::: do (11)	
	Boiling point °C (mm)	n^{25}	- Sulphonamide (11) C ₁₂ H ₁₉ NO ₃ S, melting point, °C
2-(2-Methoxyphenyl)-pentane 2-(4-Methoxyphenyl)-pentane	62 (0.3) 68 (0.3)	$1.4980 \\ 1.4970$	60 63
3-(2-Methoxyphenyl)-pentane 3-(4-Methoxyphenyl)-pentane (13)	$50 \ (0.01) \ 56 \ (0.02)$	$1.5010 \\ 1.5030$	107
2-(2-Methoxyphenyl)-2-methylbutane 2-(4-Methoxyphenyl)-2-methylbutane	58 (0.3) 71 (0.5)	$\substack{1.5070\\1.5011}$	111 104

TABLE III
Properties of alkylphenols and their derivatives

	Alkylphenol (C ₁₁ H ₁₆ O)		3,5-Dinitro- benzoate $(C_{18}H_{18}N_2O_6)$,	Dinitro-alkylphenol (C ₁₁ H ₁₄ N ₂ O ₅)	
	Boiling point °C (mm)	n^{25}	melting point,	Boiling point °C (mm)	n^{25}
2-(2-Hydroxyphenyl)-pentane 2-(4-Hydroxyphenyl)-pentane (15)	65 (0.1) 84 (0.5)	$1.5122 \\ 1.5132$	85 81	145 (0.05) 155 (0.1)	$1.5686 \\ 1.5624$
3-(2-Hydroxyphenyl)-pentane 3-(4-Hydroxyphenyl)-pentane	$61 \ (0.1) \\ 83 \ (0.3)$	m.p. 64° m.p. 72°	87 92	138 (0.01) 149 (0.03)	$\frac{1.5668}{1.5664}$
2-(2-Hydroxyphenyl)-3-methylbutane 2-(4-Hydroxyphenyl)-3-methylbutane		1.5157 m.p. 73°	$\begin{array}{c} 72 \\ 109 \end{array}$	$140 \ (0.08) \\ 155 \ (0.1)$	$1.5713 \\ 1.5669$

TABLE IV
Amine salts of dinitrophenols

	Piperidine $(C_{16}H_{25}N_3O_5)$, m.p., °C	Morpholine $(C_{15}H_{23}N_3O_6)$, m.p., °C	Cyclohexylamine (C ₁₇ H ₂₇ N ₃ O ₅) m.p., °C	
2-(2-Hydroxy-3,5-dinitrophenyl)-pentane	140	146	189	
2-(4-Hydroxy-3,5-dinitrophenyl)-pentane	187	160	169	
3-(2-Hydroxy-3,5-dinitrophenyl)-pentane 3-(4-Hydroxy-3,5-dinitrophenyl)-pentane	$\begin{array}{c} 173 \\ 213 \end{array}$	157 185	$\begin{array}{c} 192 \\ 217 \end{array}$	
2-(2-Hydroxy-3,5-dinitrophenyl)-3-methylbutane 2-(4-Hydroxy-3,5-dinitrophenyl)-3-methylbutane	145	153	213 (decomp.)	
	199	170	217 (decomp.)	

Grignard reagent (e.g. 4-methoxypropiophenone and ethyl magnesium bromide), dehydrating the carbinol (6), and hydrogenating the alkene. The 4-alkyl anisoles were demethylated using pyridine hydrobromide and acetic acid but this method failed with the 2-isomers which were accordingly reacted with hydriodic acid and phenol. Nitration of the phenols was done in acetic acid and each dinitrophenol was characterized by the preparation of crystalline salts with morphine, piperidine, and cyclohexylamine.

The dinitrophenols, their derivatives, and intermediates are listed in the tables. All the compounds reported are new except where noted and satisfactory analyses were obtained for all the compounds. These analyses together with experimental details of the preparations are available elsewhere (7, 8).

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SYNTHESIS OF 2- OR 4,6-DINITRO-sec-HEXYL PHENOLS

G. G. S. DUTTON, M. E. D. HILLMAN, AND J. G. MOFFATT

The synthesis of dinitro-sec-pentyl phenols was reported in a recent note (1), and some corresponding sec-hexyl homologues are now described and characterized as their crystalline amine salts (2). These compounds are of interest because of the known herbicidal activity of 2,4-dinitro-6-sec-butyl phenol (3) and the physiological tests on these compounds will be reported elsewhere.

TABLE I Properties of alcohols and alkenes

	Alcohols (C ₁₃ H ₂₀ O ₂)		Alkenes $(C_{13}H_{18}O)$		Nitrosyl	
•	b.p.° C (0.01 mm)	n^{25}	b.p.° C (mm)	n^{25}	chloride (C ₁₃ H ₁₈ NO ₂ Cl), m.p.° C	
2-(2-Methoxyphenyl)-hexanol-2 2-(4-Methoxyphenyl)-hexanol-2	79 68	$1.5095 \\ 1.5285$	64 (0.04) 70 (0.01)	1.5199 1.5329	103 86	
3-(2-Methoxyphenyl)-hexanol-3 (6) 3-(4-Methoxyphenyl)-hexanol-3 (6)	77 84	$egin{array}{c} 1.5163 \ 1.5204 \end{array}$	$80 \ (0.05) \ 85 \ (0.4)$	$egin{array}{c} 1.5204 \ 1.5260 \end{array}$	73 67	
$\hbox{$2$-(2-Methoxyphenyl)-3-methylpentanol-3}$	73	Partial lehydration	57 (0.04)	1.5170		
2-(4-Methoxyphenyl)-3-methylpentanol-3	sponta dehydi	neous	63 (0.04)	1.5260		
2-(2-Methoxyphenyl)-4-methylpentanol-2 2-(2-Methoxyphenyl)-4-methylpentanol-2	$\begin{array}{c} 68 \\ 63 \end{array}$	$1.5260 \\ 1.5128$	59 (0.01) 68 (0.04)	$1.5134 \\ 1.5290$	104	
$\begin{array}{lll} 3\mbox{-}(2\mbox{-Methoxyphenyl})\mbox{-}2\mbox{-methylpentanol-3} \\ 3\mbox{-}(4\mbox{-Methoxyphenyl})\mbox{-}2\mbox{-methylpentanol-3} \end{array}$	89 8) 82	$1.5175 \\ 1.5196$	70 (0.05) 84 (0.1)	$1.5228 \\ 1.5240$	· — .	

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