Synthesis of Aryl and Heterocyclic Acetylenes via Copper Acetylides

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A SIMPLE synthesis of various disubstituted acetylenes, e.g., (I), from copper acetylides and appropriate iodo-heterocyclic derivatives has recently been reported.¹ Unfortunately, monosubstituted acetylenes such as (II) cannot be directly prepared by this method since monocuprous acetylide is unknown. This difficulty can be overcome by the introduction of a substituted cuprous acetylide carrying a readily removable substituent and a synthesis of monosubstituted acetylenes, e.g., (V), based on this approach is now reported.

In a typical example, iodobenzene and the cuprous salt of the tetrahydropyranyl ether of

prop-2-yn-1-ol were heated under reflux in pyridine solution under nitrogen and the product was hydrolysed with 2n-sulphuric acid to give 3-phenylprop-2-yn-1-ol (III). Heterogeneous oxidation of such $\alpha\beta$ -acetylenic alcohols by manganese dioxide in low yield has already been reported;2 with nickel peroxide³ in benzene at room temperature the alcohol (III) gave the aldehyde (IV) in 70% yield. On treatment with aqueous methanolic 2n-sodium hydroxide at 50° deformylation^{4,5} to phenylacetylene (V) occurred (87%). In the same way the alcohols (I),1 (VI; m.p. 44°),† and (VII)6 prepared similarly, were converted into the aldehydes (VIII) (74%),1 (IX) (b.p. 85-86°/0.5 mm., 81%), and (X) (74%), and then into 2-ethynylthiophen (II) (80%),7 1-ethynylnaphthalene (XI) (b.p. $143^{\circ}/25$ mm., 86%), and 5-ethynyl-2,2'bithienyl (XII) (91%).6

The overall sequence (i) could be shortened by direct oxidation of the alcohols with nickel peroxide in aqueous methanolic 2N-sodium hydroxide at 50° to give the corresponding acetylenes in somewhat lower yields (ca. 50%).

(i) RI
$$\xrightarrow{(1) \text{ CuC}: \text{C}\cdot\text{CH}_2\text{OTHP}} \rightarrow \text{RC}: \text{C}\cdot\text{CH}_2\text{OH} \xrightarrow{\text{NiO}_2} \rightarrow \text{RC}: \text{C}\cdot\text{CHO} \xrightarrow{\text{OH}^-} \rightarrow \text{RC}: \text{CHO}$$

(ii) RI
$$\xrightarrow{(1) \text{ CuC}; \text{C}\cdot\text{CH}(\text{OEt})_2}$$
 RC; C·CHO $\xrightarrow{\text{OH}^-}$ RC; CH

(iii) RI
$$\xrightarrow{\text{CuC}: \text{C}\cdot\text{CO}_2\text{Et}}$$
 RC: $\text{C}\cdot\text{CO}_2\text{Et}$ $\xrightarrow{\text{(1)}}$ OH-
$$\text{(2)}$$
 Cu^{II} $\xrightarrow{\text{(2)}}$ RC: CH

C:CR

(VI)
$$R=CH_2OH$$
(IX) $R=CHO$
(XI) $R=CHO$
(XII) $R=H$

(XIII)

Alternative routes (ii) and (iii) are possible. Sequence (ii)¹ gave the required aldehyde directly but there are some practical difficulties in working up these propargylaldehyde derivatives using this method; in our hands sequence (i) is preferred. Similarly, sequence (iii) is feasible but hydrolysis and cupric-catalysed decarboxylation⁸ of the propiolic acids offers no advantage.

This method is an addition to the standard methods of synthesis of monosubstituted aryl and

[†] Compounds (VI), (IX), (XI), and (XIII) gave satisfactory analyses; all other compounds were compared with known standards.

heterocyclic acetylenes. No vigorous dehydrohalogenation is required and some inaccessible acetylenes, e.g., (XIII) [m.p. (of mercury derivative) 201-203°] become readily available.

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