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## Synthesis and Potent Anti-HIV Activity of L-3'-Fluoro-2',3'-Unsaturated Cytidine

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## **ABSTRACT**

L-2',3'-Didehydro-2',3'-dideoxy-3'-fluorocytidine (L-3'-Fd4C), a novel potent anti-HIV agent (EC<sub>50</sub> 0.03 μM in PBM cells), has been synthesized from L-xylose in 14 steps.

Nucleoside analogues have been the cornerstone of antiviral therapy over the past thirty years. In the effort to discover effective antiviral agents against AIDS and viral hepatitis, a large number of nucleoside analogues have been synthesized and evaluated. Although structure-activity relationship studies have not led to a pharmacophore model for the antiviral activities of nucleosides, some structural features have been particularly successful. For example, all six of the nucleoside reverse transcriptase inhibitors approved by the FDA for the treatment of AIDS1 can be considered as 2',3'-dideoxynucleosides. In addition, among ring substituents, electronwithdrawing groups such as azido<sup>2</sup> and fluorine<sup>3</sup> have often

produced potent antiviral agents. Another structural feature that is often beneficial for antiviral activity is a 2',3'unsaturated bond. We have extensively explored these substitutions in nucleoside analogues, particularly with the synthesis and biological evaluation of D- and L-2',3'didehydro-2',3'-dideoxy-2'-fluoro nucleosides (Figure 1).3d-f



fluoro nucleosides.

Figure 1. Structures of D- and L-2',3'-didehydro-2',3'-dideoxy-2'-

Among them, the cytosine and 5-fluorocytosine derivatives displayed potent anti-HIV and anti-HBV activity without significant cytotoxicity. For this reason, we decided to explore the chemistry and biology of 2',3'-didehydro-2',3'dideoxy-3'-fluoronucleosides. In this series, the D-cytidine and D-thymidine analogues were previously synthesized and shown to have low to moderate anti-HIV-1 activity without cytotoxicity. 4 The D-adenine derivative also showed moderate anti-HIV activity with some toxicity.4a In view of the fact

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**Scheme 1.** Synthesis of L-2',3'-Didehydro-2',3'-dideoxy-3'-fluorocytidine

<sup>a</sup> Reagents and conditions: (a) ref 5; (b) NaH, THF, from 0 °C to room temperature, 1 h, then BnBr, TBAI, from 0 °C to room temperature, overnight; (c) 1:2 (4.0 M HCl/dioxane)/MeOH, room temp, 1.5 h; (d) (i) PhOC(S)Cl, DMAP, Tol, 90 °C, 3 h, (ii) Bu₃SnH, AIBN, reflux, 1h; (e) H₂ (55 psi), 10% Pd/C, EtOH, room temp, 72 h; (f) CrO₃, Ac₂O, Py, CH₂Cl₂, room temp, 15 min; (g) DAST, CH₂Cl₂, reflux, 36 h; (h) concentrated H₂SO₄, Ac₂O, AcOH, 0 °C, 5 min; (i) persilylated N⁴-benzoylcytosine, TMSOTf, MeCN, from 0 °C to room temperature, 72 h; (j) saturated NH₃/MeOH, room temp, 4 h; (k) MeONa, DMF, room temp, overnight.

that among the 2'-fluoro derivatives L-isomers have potent antiviral activity with no toxicity or less toxicity than their D-counterparts, it was of interest to synthesize L-2',3'-didehydro-2',3'-dideoxy-3'-fluorocytidine (L-3'-Fd4C) 1 (Scheme 1). Our synthetic method can also provide an entry to the L-2',3'-dideoxy-3',3'-difluoro nucleoside 11.

The starting material of our synthetic approach (Scheme 1) was L-xylose, which was converted to the protected L-ribose analogue 2 in four steps in 73% overall yield by a well-known procedure in our laboratory. 5 Benzylation of 2 was easily accomplished by treatment with sodium hydride, followed by benzyl bromide and catalytic tetrabutylammonium iodide. Methanolysis of the resulting benzyl ether gave the intermediate 3 as the sole isomer. <sup>1</sup>H NMR showed the signal related to the H-1 as a singlet, which indicates the  $\beta$ -stereochemistry. 6 Comparison of the proton spectrum with that of the known enantiomer<sup>7</sup> confirmed the assignment. Conversion of 3 to the phenoxythiocarbonyl derivative followed by the radical deoxygenation of the latter gave protected L-2-deoxyribose 4. Compound 4 was rather unreactive toward catalytic hydrogenation, and its palladiumcatalyzed debenzylation required treatment with hydrogen at 55 psi for 3 days. Although the yield was modest (60%), most of the unreacted starting material could be recovered and recycled. Oxidation of the debenzylated product 5 by chromic anhydride/pyridine/acetic anhydride gave ketone 6 in 88% yield. Treatment with (diethylamino)sulfur trifluoride (DAST) afforded difluorinated intermediate **7**<sup>8</sup> in 66% yield. Compound 7 was converted to the acetate 8 by the modification of a known literature method.9 Condensation of the acetate 8 with persilvlated  $N^4$ -benzoylcytosine was effected under Vorbrüggen conditions using trimethylsilyl trifluoromethanesulfonate (TMSOTf) as a catalyst. The epimeric products 9 and 10 were chromatographically separable, and the  $\beta$ -isomer was more abundant. In fact, <sup>1</sup>H NMR of the crude reaction mixture showed an epimeric ratio of 5:4. Deprotection of each isomer was accomplished by ammonolysis to give difluorinated nucleosides 11<sup>10</sup> and 12. Elimination by treatment with sodium methoxide in DMF

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<sup>(8)</sup> Yellow oil:  $[\alpha]^{23}_{\rm D}$  40.78° (c 4.28, CHCl<sub>3</sub>);  $^{1}{\rm H}$  NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.08 (m, 2H), 7.57 (m, 1H), 7.45 (m, 2H), 5.15 (dt, 1H, H<sub>1</sub>, J = 5.7, 1.7 Hz), 4.63–4.41 (m, 3H, H<sub>4</sub>, H<sub>5</sub>), 3.40 (s, 3H), 2.67 (tdd, 1H, H<sub>2 $\alpha$ </sub>, J = 16.6, 15.0, 5.7 Hz), 2.49 (tdd, 1H, H<sub>2 $\beta$ </sub>, J = 15.0, 9.0, 2.0 Hz);  $^{13}{\rm C}$  NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  166.05, 133.14, 129.71, 128.38, 128.32 (t,  $J_{\rm C-F}$  = 252.8 Hz), 103.70 (dd,  $J_{\rm C-F}$  = 7.1, 4.2 Hz), 79.46 (dd,  $J_{\rm C-F}$  = 32.1, 24.9 Hz), 62.94 (dd,  $J_{\rm C-F}$  = 7.5, 4.5 Hz), 55.38, 42.26 (t,  $J_{\rm C-F}$  = 24.3 Hz); HRMS (FAB) m/z found 273.0950, calcd for  $C_{13}H_{15}F_2O_4$  273.0938 (MH<sup>+</sup>). Anal. Calcd for  $C_{13}H_{14}F_2O_4$ : C, 57.35; H, 5.18. Found: C, 57.64; H, 5.30. (9) Mikhailopulo, I. A.; Poopeiko, N. E.; Pricota, T. I.; Sivets, G. G.;

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<sup>(10)</sup> White solid: mp 194–196 °C (dec);  $[\alpha]^{24}_D$  –51.89° (c 1.15, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  276.5 ( $\epsilon$  18 160) (pH 2), 268.0 ( $\epsilon$  13 280) (pH 7), 268.5 ( $\epsilon$  13 580) (pH 11); <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz)  $\delta$  7.97 (d, 1H, H<sub>6</sub>, J = 7.3 Hz), 6.27 (t, 1H, H<sub>1</sub>, J = 6.8 Hz), 5.93 (d, 1H, H<sub>5</sub>, J = 7.3 Hz), 4.17 (m, 1H, H<sub>4</sub>), 3.83 (m, 2H, H<sub>5</sub>), 2.90 (m, 1H, H<sub>2</sub>), 2.51 (m, 1H, H<sub>2</sub>); <sup>13</sup>C NMR (DMSO- $d_6$ , 100 MHz)  $\delta$  168.19, 158.46, 142.62, 128.71 (dd,  $J_{C-F}$  = 255.1, 247.4 Hz), 97.04, 84.73 (dd,  $J_{C-F}$  = 6.7, 4.9 Hz), 83.74

afforded the target unsaturated nucleosides 1<sup>11</sup> and 13 in 53 and 51% yields, respectively.

Intermediate 7 could have been obtained from 2 through a more straightforward route (Scheme 2) involving oxidation

of 2 to the known ketone 14, difluorination of 14 to the difluororibose derivative 15, followed by methanolysis and deoxygenation. Since 14 can also be obtained from L-xylose in three steps,<sup>5</sup> this route would have saved four steps in our synthetic scheme. Unfortunately, ketone 14 proved to be unreactive in the classic fluorinating conditions. Even when the reaction was run in neat DAST, no products could be detected, and attempts to increase the reactivity by increasing the temperature only caused decomposition of 14. It is known that the steric hindrance as well as electronic effects of the isopropylidene moiety of 14 (Scheme 2) can prevent nucleophilic attack on position 3 of the furanose ring.<sup>12</sup> Besides, literature examples show that high temperatures are required to difluorinate sterically hindered ketones. Furthermore, in the only example of difluorination of an  $\alpha,\alpha'$ trans-disubstituted five-membered cyclic ketone found in the literature, the carbonyl group is relatively unhindered and the yield is still low (25%).<sup>13</sup>

(dd,  $J_{C-F}$  = 29.5, 25.0 Hz), 60.73 (t,  $J_{C-F}$  = 4.8 Hz), 42.14 (t,  $J_{C-F}$  = 23.4 Hz), 12.29; HRMS (FAB) m/z found 248.0850, calcd for  $C_9H_{12}F_2N_3O_3$  248.0847 (MH<sup>+</sup>). Anal. Calcd for  $C_9H_{11}F_2N_3O_3$ ·0.1H<sub>2</sub>O: C, 43.41; H, 4.53; N, 16.88. Found: C, 43.63; H, 4.50; N, 16.48.

(11) White solid: mp 182–183 °C (dec);  $[\alpha]^{22}_{D}$  6.67° (c 0.54, MeOH); UV (MeOH)  $\lambda_{\text{max}}$  276.0 ( $\epsilon$  11 990) (pH 2), 267.5 ( $\epsilon$  8010) (pH 7), 264.5 ( $\epsilon$  8060) (pH 11); <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz)  $\delta$  8.14 (d, 1H, H<sub>6</sub>, J = 7.5 Hz), 7.01 (m, 1H, H<sub>I'</sub>), 5.90 (d, 1H, H<sub>5</sub> J = 7.5 Hz), 5.46 (m, 1H, H<sub>2'</sub>), 4.76 (m, 1H, H<sub>4'</sub>), 3.80 (m, 1H, H<sub>5'</sub>), 7.9 (m, 1H, H<sub>5'</sub>); <sup>13</sup>C NMR (CD<sub>3</sub>-OD, 100 MHz)  $\delta$  168.31, 163.40 (d,  $J_{C-F}$  = 284.8 Hz), 159.05, 144.05, 102.34 (d,  $J_{C-F}$  = 9.7 Hz), 95.68, 88.42 (d,  $J_{C-F}$  = 14.9 Hz), 81.96 (d,  $J_{C-F}$  = 24.7 Hz), 61.82 (d,  $J_{C-F}$  = 2.2 Hz); HRMS (FAB) m/z found 228.0778, calcd for C<sub>9</sub>H<sub>11</sub>FN<sub>3</sub>O<sub>3</sub> 228.0784 (MH<sup>+</sup>). Anal. Calcd for C<sub>9</sub>H<sub>10</sub>FN<sub>3</sub>O<sub>3</sub>: C, 47.58; H, 4.44; N, 18.50. Found: C, 47.47; H, 4.44; N, 18.17.

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Direct reaction of intermediate **7** with  $N^4$ -benzoylcytosine in the standard Vorbrüggen conditions did not give the expected protected nucleosides. Instead, the inseparable epimeric mixture **16**<sup>14</sup> (Scheme 3) was isolated in 60% yield.

**Scheme 3.** Ring-Opening of **7** in Vorbrüggen Conditions

The stereochemistry of condensation products **9** and **10** has been established by 2D NMR NOESY experiments. Thus, the NOESY spectrum of epimer **9** showed clear correlations between protons  $H_{1'}$  and  $H_{4'}$ . Correlations between  $H_{4'}$  and  $H_{2'\alpha}$  and between the latter and  $H_{1'}$  were also observed. In the case of **10**, the  $H_{4'}$  proton strongly correlated with one of the  $H_{2'}$  protons (which could thus be identified as  $\alpha$ ), while the other  $H_{2'}(\beta)$  correlated with  $H_{1'}$ .

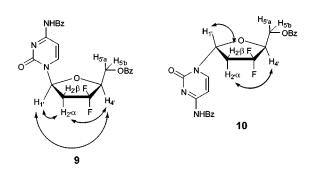


Figure 2. NOE correlations and stereochemistry of 9 and 10.

In summary, L-2',3'-dideoxy-3',3'-difluoro- and L-2',3'-didehydro-2',3'-dideoxy-3'-fluoro nucleosides have been synthesized from L-xylose in 13 and 14 steps, respectively. The target compounds were obtained by condensation of a difluorinated intermediate **8** with a nucleobase in a process of general applicability. The key difluorination step of ketone **6** proceeded with a 66% yield. Our approach seems to be more convenient and versatile than the reported synthesis of the D-thymidine analogue.<sup>15</sup>

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<sup>(14)</sup> Yellow oil:  $^1{\rm H}$  NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  8.75 (bs, 2H), 8.06—7.45 (m, 24H), 6.14—6.10 (m, 2H), 7.57 (m, 2H), 4.56 (dd, 1H, J=11.5, 3.2 Hz), 4.51 (dd, 1H, J=11.7, 4.2 Hz), 4.37 (m, 2H), 4.27 (m, 2H), 3.40 (s, 3H), 3.38 (s, 3H), 2.57 (m, 2H), 2.42 (m, 2H), 0.17 (s, 9H), 0.16 (s, 9H);  $^{13}{\rm C}$  NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  166.13, 166.07, 154.90, 154.81, 143.12, 133.20, 133.16, 129.60, 129.56, 128.97, 128.45, 128.41, 127.60, 121.33 (t,  $J_{\rm C-F}=248.2$  Hz), 121.10 (t,  $J_{\rm C-F}=248.2$  Hz), 97.69, 97.53, 83.91, (dd,  $J_{\rm C-F}=5.1,$  3.1), 83.76 (dd,  $J_{\rm C-F}=7.1,$  3.1), 72.13 (t,  $J_{\rm C-F}=29.3$  Hz), 72.08 (t,  $J_{\rm C-F}=28.0$  Hz), 64.20, 64.16, 57.08, 57.01, 37.92 (t,  $J_{\rm C-F}=23.5$  Hz), 37.81 (t,  $J_{\rm C-F}=23.6$  Hz), 0.11; MS (FAB) m/z 560 (MH $^+$ ).

Preliminary biological evaluation of the synthesized compounds showed that L-3'-Fd4C 1 has potent anti-HIV activity (EC<sub>50</sub> 0.03  $\mu$ M in PBM cells) with little or no significant toxicity (IC<sub>50</sub> = 86.9  $\mu$ M in PBM cells and IC<sub>50</sub> > 100  $\mu$ M in CEM cells). <sup>16</sup> The difluorinated analogue 11 was inactive. These promising results prompt us to synthesize other pyrimidine and purine analogues in order to study the

(16) Whereas AZT showed EC $_{50}$  of 0.004  $\mu$ M, with IC $_{50}$  > 100 in PBM cells and IC $_{50}$  = 14.3 in CEM cells (ref 3d).

full structure—activity relationships, which is in progress in our laboratory.

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