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Oxidant Controlled Regio- and Stereodivergent Azidohydroxylation of Alkenes via I₂ Catalysis

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Abstract: A novel, I_2 catalyzed regio- and stereodivergent vicinal azidohydroxylation of alkenes leading to 1,2-azidoalcohols in high yields (up to 92%) and excellent dr (up to 98%) has been developed. This unprecedented transformation employs NaN₃ and DMF as N- and O- nucleophiles respectively. The role of DMF as O- source in the reaction has been unequivocally proven by ¹⁸O labelling studies.

Vicinal difunctionalization of alkenes is an attractive strategy for the assembly of hetero-functionalized organic compounds. In this context, dioxygenation,1 dinitrogenation2 and oxynitrogenation³ across the alkenes are well studied. However, the direct vicinal azidohydroxylation across unsymmetrical alkenes in requisite regio- and diastereoselective fashion is rare.^{3d,3e} The resulting structural units comprising vicinal azidohydroxy functional groups are recurrent in drugs, natural products and synthetic materials⁴ and find tremendous utility in the synthesis of biologically active amino sugars,⁵ nucleosides,⁶ lactams,⁷ triazoles,8 oxazolines9 and in the chemistry of peptidomimetics¹⁰ and pseudopeptides¹¹. Conventionally, 1,2azidoalcohols are prepared from ring opening of epoxides¹² or displacement of halohydrin¹³ with various azide sources under acidic/basic conditions as well as chemoselective reduction of α -azido ketones.¹⁴ Nevertheless, these reports suffer from disadvantages such as use of prefunctionalized starting materials leading to reduced atom economy, harsh reaction conditions accompanied with low functional group tolerance and often poor regio- and diastereoselectivity. Thus, a mild synthetic procedure that employs readily available starting materials and overcomes the above difficulties is desirable for regio- and stereodivergent synthesis of 1,2-azidoalcohols.

I₂ catalysis has been increasingly explored for C-C, C-O and C-N bond formation as environmentally benign and inexpensive oxidation reagent in place of rare or toxic heavy metal oxidants.¹⁵ However, olefin functionalization using

catalytic electrophilic iodination with stoichiometric cooxidants is rare. We reasoned that, 1,2-additions of two nucleophiles across -C=C- bond could be possible using electrophilic iodine source owing to its ability to form electrophilic iodonium ion. In this context, Komatsu and coworkers have first reported I2-catalyzed aziridination of alkenes using chloramine-T as N-nucleophile.¹⁶ Thereafter, Yoshimura et al. reported the I₂-mediated cyclopropanation of alkenes using malononitrile as C-nucleophile.¹⁷ More recently, we have demonstrated the dihydroxylation of alkenes using benzoic acid and DMSO as O-nucleophiles.¹⁸ In heterodifunctionalization (e.g. azidohydroxylation), switching to one of the possible isomers with required regio- and diastereoselectivity is challenging due to little difference in the relative reactivity of O- and N-nucleophiles under different conditions.



Scheme 1: Azidohydroxylation of alkenes

In 2013, Studer et al. have reported the direct azidooxygenation of alkenes giving azidohydroxy derivative **4** using excess amount of N₃-I(III) reagent and TEMPONa as *N*- and *O*-nucleophiles respectively, thus requiring additional steps to achieve 1,2-azidoalcohols.¹⁹ *N*,*N*-Dimethylformamide is a popular formylating agent;²⁰ however, its role in nucleophilic substitution reactions is quite rare²¹ and opens up a new area of research yet to be established. For the first time, we report, a controlled synthesis of either regioisomers of 1,2-azidoalcohols with high diastereomeric ratios using DMF as *O*-nucleophile and NaN₃ as *N*-nucleophile (**Scheme 1**).

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Table 1: I2-catalyzed regiodivergent azidohydroxylation of styrene	э:
optimization studies ^[a]	

no.	Ph 1a halogen (10 mol %)	halogen source oxidant (2 equiv) NaN ₃ (2 equiv) base (1 equiv) solvent, 8h, 25 °C	Ph Ph 2a base	r_{N_3} + r_{Ph} r_{N_3} r_{Ph} r_{N_3} r_{N	yield of 2a ^[b]
1	I ₂	TBHP	Et ₃ N	DMF	53
2	I_2	TBHP	Et ₃ N	THF+DMF	10
3	I_2	TBHP	Et ₃ N	CH ₃ CN+DMF	48
4	I_2	TBHP	Et ₃ N	CH ₂ Cl ₂ +DMF	18
5	I_2	TBHP	Et ₃ N	DMSO+ DMF	90,38 ^[c]
6	I_2	TBHP	K ₂ CO ₃	DMSO+ DMF	18
7	I_2	TBHP	K'OBu	DMSO+ DMF	44
8	I_2	TBHP	NaH	DMSO+ DMF	32
9	I_2	TBHP	DBU	DMSO+ DMF	65
10	"Bu ₄ NI	TBHP	Et ₃ N	DMSO+ DMF	5
11	NaI	TBHP	Et ₃ N	DMSO+ DMF	11
12	KI	TBHP	Et ₃ N	DMSO+ DMF	14
13 ^[d]	I_2	50%	Et ₃ N	DMSO+ DMF	82,78 ^[e]
		aq.11202			

^[a]Reaction conditions: styrene (1 mmol), NaN₃ (2 mmol), halogen source (10 mol %), base (1 mmol), oxidant (2 mmol); 8 mL of solvent (1:1), 25 °C, 8 h. ^[b]Isolated yields after column chromatographic purification. ^[c] 5 mol % of I₂ was used. ^[d] **3a** was formed as major product. ^[e] 30% aq. H₂O₂ was used.

Initially, when styrene (1a) (1 mmol) was treated with a mixture of I₂ (10 mol %) and TBHP (5-6 M solution in decane contains <4 % water, 2 mmol) followed by addition of Et₃N (1 mmol) and sodium azide (2 mmol) at 0 °C in DMF, to our delight we obtained 2-azido-1-phenylethan-1-ol (2a) as exclusive regioisomer albeit in moderate yield 53% (Table 1). Encouraged by this result, it was of interest to optimize the reaction conditions in order to obtain improved yield without affecting its regioselectivity. Thus, other solvents like THF, CH₃CN and CH₂Cl₂ were screened and found to be unsuitable for this reaction. Hence, we found DMF was crucial in obtaining desired azidoalcohol 2a. Surprisingly, out of solvent combinations screened (entries 2-5), DMSO+DMF (v/v = 1/1) mixture resulted in excellent yield (90%) of the desired product 2a. Decrease in I_2 catalyst loading to 5 mol % however had a deleterious effect on yield (38%) (entry 5). Further modification, either in iodine source or base did not show any significant improvement in the product yield (entries 6-12). On the contrary, when 50% aq. H₂O₂ was used as oxidant instead of TBHP, a complete reversal in product regioselectivity was observed affording 2-azido-2-phenylethan-1-ol (3a) in 82% yield (entry 13). Even, 30% aq. H₂O₂ can be employed that afforded 3a in good yield (78%). Furthermore, with 70% aq. TBHP, it gave a 3:1 mixture of 2a and 3a in 78% combined yield. Other oxidants such as cumene hydroperoxide and oxone afforded 2a in 26 and 14% yields respectively. Thus, the preliminary studies established an oxidant-directed switch to either regioisomers 2a or 3a in excellent yields directly from styrene in a single step.

In order to determine its scope, a variety of olefins were evaluated under the optimized reaction conditions and the results are summarized in **Table 2**. When styrenes, with $-CH_3$ (**1b**), -OH (**1c**), -Br (**1d**), $-NO_2$ (**1e**) groups on aromatic ring

Table	2:	I ₂ -catalyzed	regio-	and	stereodivergent
azidohy	droxy	lation of alkenes	: substrat	te scope	

он	I ₂ (10 mol %) I ₂ (10 mo anhyd.TBHP R ₁ 50% aq.H	1%) H2O2 Bts I
$R_1 \rightarrow R_3$	(2 equiv) Et N (1 equiv) R2 R3 (2 equiv) Et N (1 equiv)	
22 N3	NaN ₃ (2 equiv) NaN ₃ (2 e DMSO:DMF (1:1) 1a-n DMSO:D	equiv) OH MF(1:1) 3a-n
Products ^[a]	substrates	Products ^[b,c]
(2a–n)	(1a-n)	(3a -n)
OH N3		N ₃ OH
R	R	R
2a (90%)	1a R = H	3a (82%) (12:1)
2b (88%)	$1b R = CH_3$	3b (89%) (9:1)
2c (76%)	Ic $R = OH$	 N3
		ССССИН
Br	Br	Br
2d 82%	1d	3d (86%) (10:1)
N ₃		ССССОН
ų.	NO	Y
NO2 20779/	10	\dot{NO}_2 3 $(769/) (10.1)$
26 / / 70	Ie	Je (70%) (10.1)
N ₃		С КОН
2f (78%)	1f	3f (83%) (9:1)
N₃		
с₅н₁́ Ү он	C o ™ii °	G ₅ H₁₁Ť ↓ N ₃
2g (79%)	1g	3g (83%) (9:1)
V ^{OH} N ₃		N ₃
BnO	BnÓ	BnO
2h (84%)	1h	3h (74%) (8:1)
HQ_N₃	>	N3 OH
/OBn	∕ ⊆OBn	OBn
2i (74%)	1i	3i (78%) (5:1)
OH N3	\bigcirc	, N3
\bigcup	\bigtriangledown	\bigcup
2j (87%) dr = 95:	5 1j	3j (92%) dr = 95:5 ^[d]
		С С С С С С С С С С С С С С С С С С С
2k(87%) dr = 983	2 1 k	$3\mathbf{k}$ (82%) dr = 92:8
	R	N ₅
R1 N3 R	R	R1 OH
2l (88%) dr = 93:	7 11 $R = H, R_1 = H$	31 (86%) dr = 97:3
2m (80%) dr = 92 2n (82%) dr = 96%	$1m R = OH R_1 = H$	3m (78%) dr = 94:6 3n (80%) dr = 09:2
4000000 = 90	\rightarrow $\mathbf{n} = \mathbf{n} = \mathbf{n} = \mathbf{n}$	31100/0101 - 90/2

^[a]Reaction condition A: styrene (1 mmol), I_2 (10 mol %), Et_3N (1 mmol), 5-6 M TBHP in decane (2 mmol), NaN_3 (2 mmol), DMSO+DMF (4 mL each), 0 °C to 25 °C, 8 h; ^[b] condition B: styrene (1 mmol), I_2 (10 mol %), Et_3N (1 mmol), 50% aq. H_2O_2 (2 mmol), NaN_3 (2 mmol) DMSO+DMF (4 mL each) 0 °C to 25 °C, 8 h; ^[c] ratio of **3**:**2**; ^[d] diastereomeric ratios were determined using ¹H NMR and HPLC analysis.

were treated with I_2 (10 mol %), TBHP (2 equiv), NaN₃ (2 equiv) and Et₃N (1 equiv) in DMF:DMSO (1:1) at 25 °C, the corresponding 1,2-azido alcohols **2a–e** were obtained in 76-

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90% yields with excellent regioselectivity (>95%). In the case of disubstituted α -methyl styrene (1f), the azidohydroxylation product 2f was obtained in high regioselectivity consistent with our initial findings. Also, 1-octene (1g) gave the desired products 2g in high yield (79%) with high regioselectivity (>19:1). Interestingly, sterically congested and abundantly available isoprenol and prenol derivatives 1h and 1i too afforded 2h and 2i respectively in high yields with excellent regioselectivity. Cyclohexene (1j) and indene (1k) also participated in azidohydroxylation reaction affording the respective products 2j and 2k in predicted regioselectivity accompanied with high diastereoselectivity (Table 2) in favour of syn isomer. Interestingly, other open chain internal alkene 11 gave the corresponding 1,2-azido alcohol 21 with excellent diastereomeric ratio which is also a key intermediate for the syntheses of the psychostimulant drug, norpseudoephedrine,²² as well as the antidepressant drug, cathinone.²² Other allylic alcohol 1m and 1n were also tested and the diastereoselectivity obtained in each case remained high. Specifically, substrate with free hydroxy group was also found tolerent under this mild protocol.

On the contrary, when 50% aqueous H₂O₂ was used as cooxidant in the reaction, azidoalcohols with complementary regioselectivity were obtained. Here again, substituted styrenes 1a-f afforded 3a-f in high yields (76-89%) and regioselectivities (88-95%). Also, electronically unbiased aliphatic alkenes such as octene 1g, terminal disubstituted alkene 1h and trisubstituted alkene 1i smoothly underwent this transformation affording the respective azido alcohols 3g, 3h and 3i in high yields (74-84%) albeit in moderate regioselectivity (5:1). Remarkably, when symmetrical and unsymmetrical disubstituted alkenes 1j and 1k were tested for their diastereoselectivity under the present conditions, 1,2azidoalcohols 3j and 3k were obtained with major anti isomer this time. No discrepancy in anti-diastereoselectivity was observed in substrates 11, 1m and 1n thus affording 31, 3m and 3n in high yields (78-86%). However, electron deficient conjugated alkenes failed to undergo catalvtic azidohydroxylation under either condition.

The results of experiments for the mechanistic understanding of the reaction are summarized in Table 3: (i) DMF as oxygen source and role of Et₃N in hydrolvsis: When styrene was treated with I₂ (1 equiv) in DMF, the corresponding iodoformate 5 was isolated in 72% yield. In the presence of Et₃N, iodoalcohol 6 was obtained along with 5 which suggest the role of Et₃N in hydrolysis of formate 5 (entries 1 and 2). (ii) Co-oxidants do not compete with DMF as O-nucleophile: In the presence of cooxidants (entries 3 and 4), iodoformate 5 was still formed. Further, when ¹⁸O labelled DMF was used in the reaction, styrene gave 6% yield of 2a-18O and 78% of 3a-18O under condition B (Scheme 2). This clearly establishes that DMF serves as a source of oxygen and thus proves it to be more potent O-nucleophile as compared to TBHP, H2O2, H2O or DMSO under present reaction condition. (iii) Possibility of formation of hypervalent iodine is ruled out: Further, on adding NaN₃ in the absence of co-oxidants (entry 5), azido alcohol product 2a was indeed obtained in 48% yield. Therefore,



Table 3: Stoichiometric contr	ol experiments	for azidohydro	xylation of
styrene ^[a]			

	Ph 🔨 1a	l ₂ (1 equiv oxidant (2 azide (2 er base (1 er solvent 25 °C, 3 h) equiv) quiv)	OCHO Ph 5 X = I 8 X = N ₃	+ Ph → 6 X = I	X + Ph 7
no.	solve	nt	base	oxidant	azide	products [b]
1	DM	F				5(72%)
2	DM	F	$\mathrm{E}t_3\mathrm{N}$			5 (38%) + 6 42%)
3	DMF + D	MSO	Et ₃ N	anhyd. TBHP		5 (34%) + 6 (22%) + 7 (11%)
4	DMF + D	MSO	Et ₃ N	aq.50% H ₂ O ₂		5 (32%) + 6 (26%) + 7 (18%)
5	DMF + D	MSO	Et ₃ N		NaN ₃	8 (22%) + 6 (18%) + 2a (48%)
6	DMF + D + H_2	MSO C	Et ₃ N		NaN ₃	5 (35%) + 3a (48%)

 $^{[a]}$ Reaction condition: styrene (1 mmol), I₂ (1 mmol), oxidant (2 mmol), NaN₃ (2 mmol), base (2 mmol), solvent, 25 °C, 3 h; $^{[b]}$ Isolated yields after column chromatographic purification

 $\begin{array}{c} Ph & \begin{array}{c} I_2(10 \text{ mol \%}) \\ 50\% \text{ aq. } H_2O_2(2 \text{ equiv}) \\ \hline NaN_3(2 \text{ equiv}) \\ \hline Et_6N(1 \text{ equiv}) \\ 1a \\ DMF.^{18}O:DMSO(1:1) \\ 25\ ^{\circ}C, 8\ h \end{array} & \begin{array}{c} I^{18}OH \\ Ph \\ \hline N_3 \\ Ph \\$

Scheme 2. Isotopic labelling experiment using DMF-¹⁸O regeneration of I₂. (iv) *Account for reversal in regio and stereoselectivity of the product*: However, when the above reaction was performed under aqueous conditions (DMSO: DMF:H₂O = 1:1:0.4), iodoformate **5** and product **3a** were isolated (entry 6). Further reaction of iodo formate **5** with NaN₃ (2 equiv) and Et₃N (1 equiv) in DMF:DMSO:H₂O (1:1:0.4) gave **3a** in 54% yield suggesting the formation of cyclic intermediate **III**, which can account for the reversal in regio- and stereoselectivity of azidohydroxylation under the aqueous conditions.



Scheme 3: Plausible mechanism for $\mathsf{I}_2\text{-}\mathsf{catalyzed}$ azidohydroxylation of alkenes

Based on the above experiments and literature precedence,^{18,21} the above mechanism has been proposed (**Scheme 3**). Initially, alkene reacts with iodine to form iodonium ion which

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undergoes regioselective ring opening with DMF to give the corresponding iodo intermediate I followed by subsequent stereoselective displacement with azide ion to form species II. This on hydrolysis affords *syn* azido alcohols 2. On the other hand, under aq. H_2O_2 condition, the iodo intermediate I is hydrolyzed in situ to form iodoformate 5. The proposed species III formed from 5 by the anchimeric assistance shown by the formate group, reacts with azide anion in a regioselective manner to give *anti* azido alcohol 3 with the liberation of iodide ion, which is then reoxidized with TBHP/H₂O₂ to regenerate I₂ in the catalytic cycle.

The application of this novel method is amply illustrated in the concise total synthesis of (\pm) -chloramphenicol (9),²³ an antibiotic drug and (\pm) -cytoxazone (11),²⁴ a cytokine modulator. 1,2-*syn* azido alcohol **2m**, prepared by the present protocol was subjected to simple reduction-protection sequence



Scheme 4. Total Synthesis of (±)-chloramphenicol and (±)-cytoxazone

followed by regioselective nitration gave (\pm) -chloramphenicol (9) in 71% yield. Also, azidodiol **3n**, prepared from **1n** using condition B, on reduction-Boc protection sequence, gave aminodiol **10** in 76% yield, which under basic condition underwent cyclization to give (\pm) -cytoxazone (**11**) in 90% yield (Scheme 4).

To summarize, we have developed, for the first time, regio- and diastereoselective azidohydroxylation of alkenes to give vicinal azidoalcohols in high yields. The regio- and stereodivergence observed in the process is driven by the oxidants chosen in combination with catalytic I₂. Mechanistic study revealed that DMF is crucial for regiodivergence and acts as an *O*-source in this azidohydroxylation reaction. This methodology has successfully been applied to the concise syntheses of (\pm) -chloramphenicol and (\pm) -cytoxazone. We believe that this mild, environmentally benign and operationally simple method would find tremendous application in streamlining the synthesis of various drugs and synthetically useful intermediates. Chiral version of this azidohydroxylation process is currently underway in our laboratory and will be reported in due course.

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