

Article

## Copper-Catalyzed Cross-Dehydrogenative N<sub>2</sub>-Coupling of NH-1,2,3-Triazoles with N,N -Dialkylamides: N-Amidoalkylation of NH-1,2,3-Triazoles

Xiacong Deng, Xue Lei, Gang Nie, Lihui Jia, Yuanxiang Li, and Yunfeng Chen

*J. Org. Chem.*, **Just Accepted Manuscript** • Publication Date (Web): 30 May 2017

Downloaded from <http://pubs.acs.org> on May 31, 2017

### Just Accepted

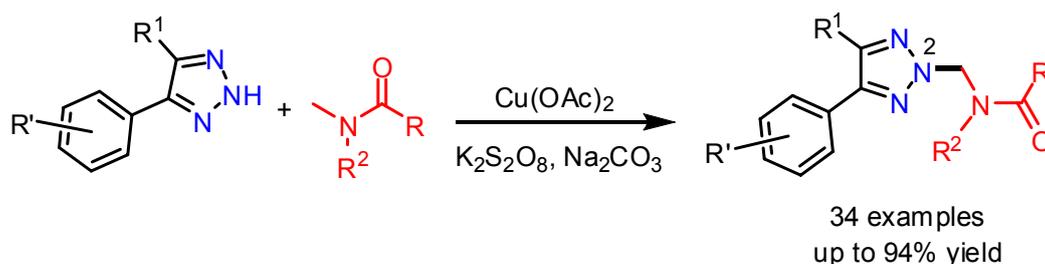
“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

# Copper-Catalyzed Cross-Dehydrogenative $N^2$ -Coupling of $NH$ -1,2,3-Triazoles with $N,N$ -Dialkylamides: $N$ -Amidoalkylation of $NH$ -1,2,3-Triazoles

Xiaocong Deng,<sup>†</sup> Xue Lei,<sup>†</sup> Gang Nie,<sup>†</sup> Lihui Jia,<sup>†</sup> Yuanxiang Li,<sup>\*,‡</sup> and Yunfeng Chen<sup>\*,†</sup>

<sup>†</sup>School of Chemistry and Environmental Engineering, Wuhan Institute of Technology, Wuhan 430073, P. R. China.

<sup>‡</sup>College of Chemistry and Materials Engineering, Huaihua University, Huaihua 418008, P. R. China.



**ABSTRACT:** An efficient copper-catalyzed C-N bond formation by N-H/C-H cross-dehydrogenative coupling (CDC) between  $NH$ -1,2,3-triazoles and  $N,N$ -dialkylamides has been developed. The method provided  $N$ -amidoalkylated 1,2,3-triazoles with moderate to high yields, and the reactions showed high  $N^2$ -selectivities when 4,5-disubstituted  $NH$ -1,2,3-triazoles served as substrates.

## INTRODUCTION

Cross-dehydrogenative coupling (CDC) reactions have emerged as powerful tools for organic synthesis because of their high atom- and step-economical characteristics,<sup>1</sup> which have also been widely applied in the functionalization of heterocyclic compounds in recent years.<sup>2</sup>

1  
2  
3  
4  
5  
6  
7 *N*-alkylamines and alkylamides, are good reaction partners for the CDC reactions. For  
8 example, DMF (*N,N*-dimethylformamide), which is widely recognized as a common solvent,  
9 can also be employed as a cheap, readily available reaction partner in CDC reactions.<sup>3</sup>  
10 Moreover, the CDC reaction of DMF is a straightforward method to introduce an amide  
11 group into a molecule. Indeed, some heterocyclic molecules with amide group have been  
12 found to show a range of interesting biological activities.<sup>4</sup>  
13  
14  
15  
16  
17  
18  
19  
20  
21

22 1,2,3-triazoles have become important building blocks in organic chemistry in the past few  
23 years. They have received considerable attention and shown many applications in medicinal  
24 chemistry,<sup>5</sup> material chemistry,<sup>6</sup> synthetic organic chemistry<sup>7</sup> and other fields<sup>8</sup>. Most  
25 1,2,3-triazoles are obtained by reactions of organic azides with other partners, such as click  
26 reactions of organic azides with alkynes<sup>9</sup>, multicomponent reactions of organic azides with  
27 activated carbonyl compounds<sup>10</sup>, and condensation reactions of organic azides with activated  
28 alkenes<sup>11</sup>. More recently, new azide-free strategies have also been developed.<sup>12</sup> These  
29 methods usually produce the *N*<sup>1</sup> or *N*<sup>1'</sup>-substituted 1,2,3-triazoles, but are not suitable for  
30 producing *N*<sup>2</sup>-substituted 1,2,3-triazole. *N*<sup>2</sup>-substituted 1,2,3-triazoles were obtained mostly  
31 via post-*N*-functionalization of *NH*-1,2,3-triazoles with different reaction partners, including  
32 alkyl halides, alcohols, alkynes and alkenes.<sup>13</sup> Still, advances for the synthesis of diverse  
33 *N*<sup>2</sup>-substituted 1,2,3-triazole were desirable.  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52

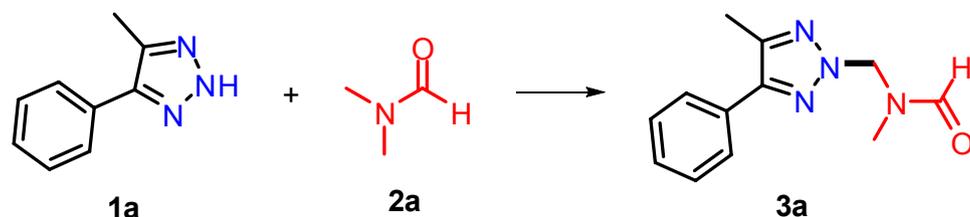
53 In continuation of our research for synthesis of new 1,2,3-triazole related heterocyclic  
54 compounds,<sup>14</sup> herein, we report a copper-catalyzed N-H/C-H cross-dehydrogenative  
55 coupling of *NH*-1,2,3-triazoles with different amides. By tuning the substituents, the  
56  
57  
58  
59  
60

reactions gave  $N^2$ -coupling products with high yield.

## RESULTS AND DISCUSSION

4-Methyl-5-phenyl-2H-1,2,3-triazole (**1a**) and DMF (*N,N*-dimethylformamide) were chosen as model substrates for the initial condition screening, the results are shown in Table 1. We proposed that methyl and phenyl groups could block the  $N^1$  and  $N^1'$  coupling sites, which might lead to high  $N^2$ -coupling selectivity.

**Table 1.** Optimization of Reaction Conditions<sup>a</sup>



entry	catalyst	oxidant	additive	conversion <sup>b</sup>	yield (%) <sup>c</sup>
1	CuCl	air	none	0	0
2	CuCl <sub>2</sub>	air	none	0	0
3	CuCl	TBHP	none	60	54
4	CuBr	TBHP	none	69	61
5	CuCl <sub>2</sub>	TBHP	none	65	59
6	CuBr <sub>2</sub>	TBHP	none	75	68
7	CuO	TBHP	none	64	59
8	Cu(NO <sub>3</sub> ) <sub>2</sub>	TBHP	none	48	43
9	CuSO <sub>4</sub>	TBHP	none	55	51
10	Cu(OAc) <sub>2</sub>	TBHP	none	85	76
11	FeCl <sub>2</sub>	TBHP	none	30	23
12	FeCl <sub>3</sub>	TBHP	none	35	32
13	Cu(OAc) <sub>2</sub>	TBHP	Na <sub>2</sub> CO <sub>3</sub>	100	88
14	Cu(OAc) <sub>2</sub>	TBHP	K <sub>2</sub> CO <sub>3</sub>	93	82
15	Cu(OAc) <sub>2</sub>	TBHP	NaOAc	78	69
16	Cu(OAc) <sub>2</sub>	TBHP	DBU	85	75
17	Cu(OAc) <sub>2</sub>	DTBP	Na <sub>2</sub> CO <sub>3</sub>	<5	trace
18	Cu(OAc) <sub>2</sub>	H <sub>2</sub> O <sub>2</sub>	Na <sub>2</sub> CO <sub>3</sub>	<5	trace

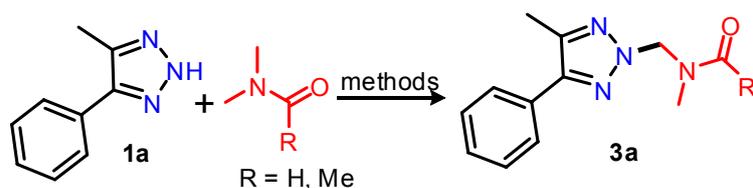
19	Cu(OAc) <sub>2</sub>	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	Na <sub>2</sub> CO <sub>3</sub>	100	92
20	Cu(OAc) <sub>2</sub>	BPO	Na <sub>2</sub> CO <sub>3</sub>	<5	trace
21	Cu(OAc) <sub>2</sub>	DDQ	Na <sub>2</sub> CO <sub>3</sub>	<5	trace
22 <sup>d</sup>	Cu(OAc) <sub>2</sub>	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	Na <sub>2</sub> CO <sub>3</sub>	60	53
23	-	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	Na <sub>2</sub> CO <sub>3</sub>	100	90

<sup>a</sup>Reaction conditions: **1a** (0.3 mmol, 47.7 mg), **2a** (3mL as solvent), and oxidant (0.6 mmol, 2.0 equivalent, 162.2 mg), Na<sub>2</sub>CO<sub>3</sub> (0.33 mmol, 1.1 equivalent, 35.0 mg), catalyst (0.06 mmol, 0.2 equivalent, 12.0 mg), 110 °C, 8 h; <sup>b</sup>Based on **1a**; <sup>c</sup>Isolated yields; <sup>d</sup>80 °C.

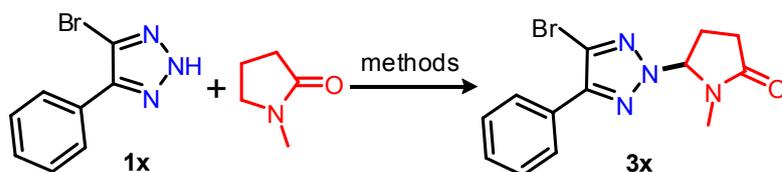
First, the reaction was performed in the presence of **1a** (0.3 mmol), CuCl (0.06 mmol, 0.2 equiv) in 3 mL DMF as solvent with air as the oxidant at 110 °C. After 5 hours there was no reaction at all. When CuCl was changed to CuCl<sub>2</sub>, it gave the same result (Table 1, entries 1 and 2). However, when TBHP (*tert*-butyl hydroperoxide), a common oxidant, was used instead of air, after 5 hours we found that the desired **3a** was formed in 54% yield (entry 3). Inspired by this, other common copper salts were examined, which showed that Cu(OAc)<sub>2</sub> gave the best result (entries 4-10). Furthermore, other inexpensive iron salts were tested, including FeCl<sub>2</sub>, FeCl<sub>3</sub> (entries 11 and 12), however, their catalytic efficiency was inferior to Cu(OAc)<sub>2</sub>. Therefore, Cu(OAc)<sub>2</sub> was chosen as the catalyst for all further reactions. Interestingly, it was found that the addition of a base could promote the reaction, and Na<sub>2</sub>CO<sub>3</sub> gave a better result (entries 13-16). In addition, other different oxidants were also investigated, including DTBP (di-*tert*-butyl peroxide), H<sub>2</sub>O<sub>2</sub>, BPO (benzoyl peroxide) and DDQ (2,3-dichloro-5,6-dicyano-1,4-benzoquinone) (entries 17-21), they failed to promote the reaction; however, when K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> was chosen as the oxidant, the yield of **3a** rose to 92%, which also revealed high *N*<sup>2</sup> selectivity.<sup>15</sup> In addition, a relatively lower conversion and yield

was detected with a reaction temperature of 80 °C (entry 22). If the reaction run without copper catalyst (entry 23),<sup>16</sup> a satisfactory result was also obtained, the isolated yield of **3a** was up to 90%. In order to establish the best reaction condition, comparisons for different methods have also been made (Scheme 1). When DMF was served as reactant, the yield of the TBHP/TBAI protocol was only 36%, much lower than Cu or Cu-free protocols. When DMA or NMP was served as substrate for the coupling reaction under copper-free reaction condition, the yields were inferior to Cu-catalyzed reactions, while TBHP/TBAI was used instead of Cu or Cu-free reaction conditions, the yields of **3a** and **3x** were up to 76% and 88%, which were similar as the Cu-catalyzed protocol. However, the colors of the Cu-free reaction system or TBHP/TBAI protocol were darker than Cu-catalyzed reaction system generally. We suspected that the polymerization of the imine cation intermediates was easier under copper-free conditions. Thus, Cu(OAc)<sub>2</sub>/K<sub>2</sub>S<sub>2</sub>O<sub>8</sub>/Na<sub>2</sub>CO<sub>3</sub> at 110 °C was chosen as the optimal reaction condition.

**Scheme 1.** Comparisons of the Reaction Conditions



method 1: 92% (R = H); 80% (R = Me);  
method 2: 90% (R = H); 60% (R = Me);  
method 3: 36% (R = H); 76% (R = Me).



method 1: 90%; method 2: 64%; method 3: 88%.

Reaction condition: **1a** and **1x** (200 mg, 1 equivalent), solvent (10 mL); isolated yields.

method 1: Cu(OAc)<sub>2</sub> (0.2 equivalent), K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (2 equivalent), Na<sub>2</sub>CO<sub>3</sub> (1.1 equivalent),

110°C; method 2: K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (2 equivalent), Na<sub>2</sub>CO<sub>3</sub> (1.1 equivalent), 110°C; method 3: TBHP

(3 equivalent), TBAI (0.1 equivalent), 110°C.

Then we set out to study the scope and limitation of this conversion with different

*NH*-1,2,3-triazoles and amides (Scheme 2). In general, the reactions all performed well, the

corresponding *N*<sup>2</sup>-substituted products were less polar than other isomers. Firstly, 4-Me (or

Br)-5-Ar-disubstituted *NH*-1,2,3-triazoles were checked due to the potentially high *N*<sup>2</sup>

regioselectivities. In fact, the Me group and Br atom could indeed block the *N*<sup>1</sup> reaction sites.

In all cases, the *N*<sup>2</sup>-substituted product was formed predominantly. The structures of **3q** was

identified by X-ray single crystal analysis.<sup>17</sup>

Then, other 4,5-disubstituted *NH*-1,2,3-triazoles, such as 4,5-diaryl 1,2,3-triazoles (**3aa-3ac**)

and 4-benzoyl-5-phenyl-1,2,3-triazole(**3y**, **3z**), were suitable for this reaction and showed

high *N*<sup>2</sup> regioselectivities. At the same time, both electron-donating functional groups, such as

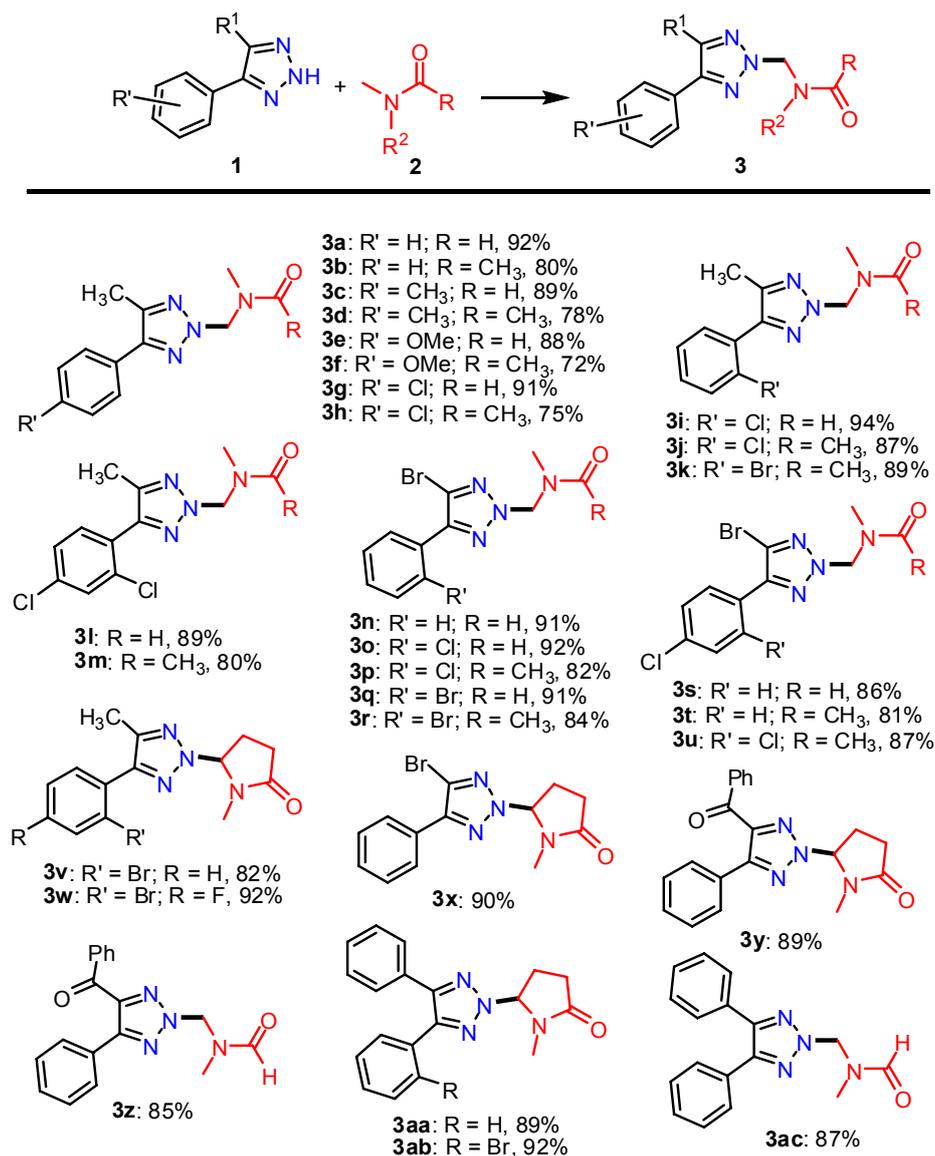
as methoxy or methyl substituents, and electron-withdrawing groups, such as

halogenated moieties in the aryl ring, were compatible with this transformation. The

yield of *N*<sup>2</sup>-coupling product was up to 94%. However, when the *IH*-1,2,3-triazole and

4-ethyl-5-methyl-1,2,3-triazole were tested, the reactions gave complex mixtures and the products were hard to be identified.

**Scheme 2.** Substrate Scope for the Synthesis of *N*<sup>2</sup>-Amidoalkylated 1,2,3-Triazoles<sup>a</sup>



<sup>a</sup>Reaction conditions: **1** (0.3 mmol), **2** (3 mL as solvent), Cu(OAc)<sub>2</sub> (0.06 mmol), K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (0.6 mmol) and Na<sub>2</sub>CO<sub>3</sub> (0.33 mmol), reaction temperature: 110 °C.

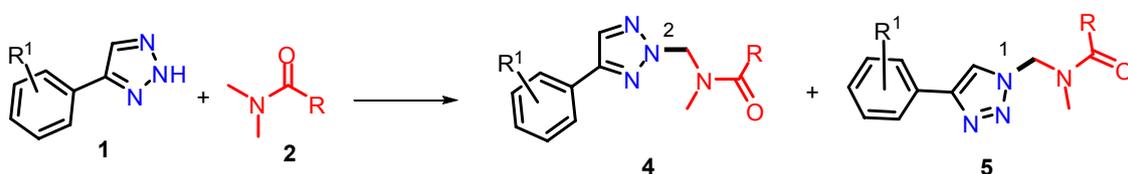
At the same time, commercially available amide solvents, such as DMF, DMA, were

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

suitable reaction partners to couple with *NH*-1,2,3-triazoles, although the reactions with DMA need more time to give the full conversion. However some side reactions or decomposition of the products could be happened over extended reaction time, which resulted in the loss of isolated yields. Furthermore, *N*-methyl-2-pyrrolidinone (NMP) was also a suitable substrate, which gave the *N*<sup>2</sup>-coupling products with high yield. It should be pointed out that the reactions could be finished in 30 mins under standard reaction conditions. However, *N,N*-dimethylaniline piperidine-1-carbaldehyde *N,N*-diethylformamide and *N,N*-dimethylbenzamide cannot react in this reaction system.

To further investigate the reaction substrate scope, we tested 4-aryl substituted *NH*-1,2,3-triazoles (Table 2). The reactions also performed well. The corresponding *N*<sup>2</sup>-coupling product was the major product generally. However, the regioselectivities were not very high, other *N*<sup>1</sup> products were also obtained.<sup>18</sup> The lower regioselectivities were similar to the results for the alkylation, arylation and alkenylation of 4-ary-1,2,3-triazoles.<sup>13</sup>

**Table 2.** 4-Aryl Substituted *NH*-1,2,3-Triazoles as Substrate<sup>a</sup>



entry	R <sup>1</sup>	R	4 yields <sup>b</sup> (%)	5 yields <sup>b</sup> (%)
1	H-	H-	64 ( <b>4a</b> )	18 ( <b>5a</b> )

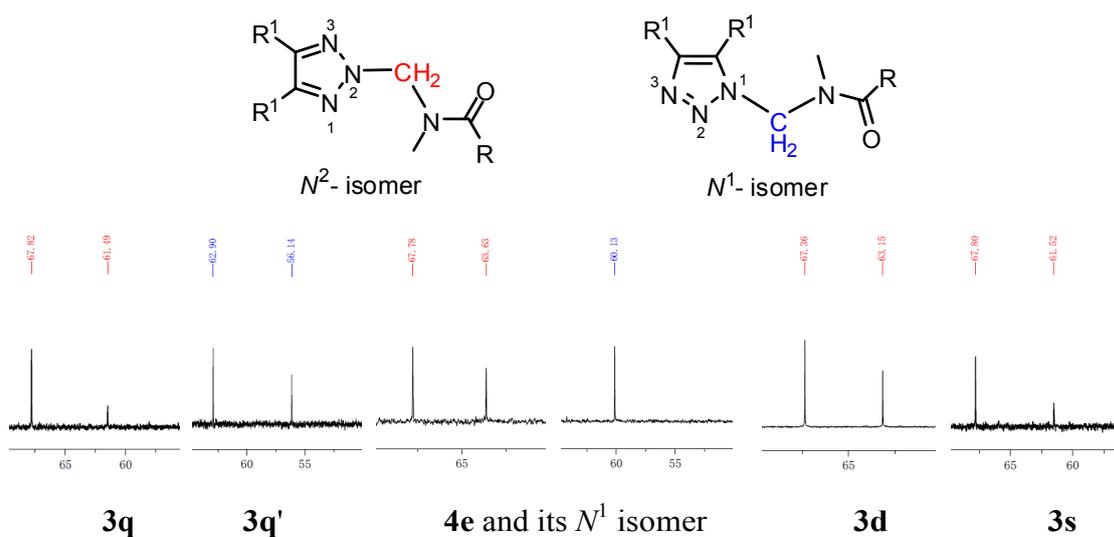
2	H-	CH <sub>3</sub> -	50 ( <b>4b</b> )	25 ( <b>5b</b> )
3	4-CH <sub>3</sub> -	CH <sub>3</sub> -	59 ( <b>4c</b> )	21 ( <b>5c</b> )
4	4-OMe-	CH <sub>3</sub> -	47 ( <b>4d</b> )	27 ( <b>5d</b> )
5	2-Br-	CH <sub>3</sub> -	57 ( <b>4e</b> )	20 ( <b>5e</b> )

<sup>a</sup>Reaction conditions: **1** (0.3 mmol), **2** (3mL as solvent), Cu(OAc)<sub>2</sub> (0.06 mmol), K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (0.6 mmol) and Na<sub>2</sub>CO<sub>3</sub> (0.33 mmol), 110 °C. <sup>b</sup>Isolated yields.

To identify the structures of *N*-amidoalkylated 1,2,3-triazoles, a gram-scale experiment<sup>19</sup> was done in order to obtain the *N*<sup>1</sup>-coupling regioisomer (**3q'**) of **3q**. The structure was also confirmed by X-ray single crystal analysis.<sup>17</sup> Because the crystal structures of two regioisomers **3q** and **3q'** were obtained, it was beneficial to figure out the methods for the determination of the *N*<sup>2</sup> regioisomers for other reactions.

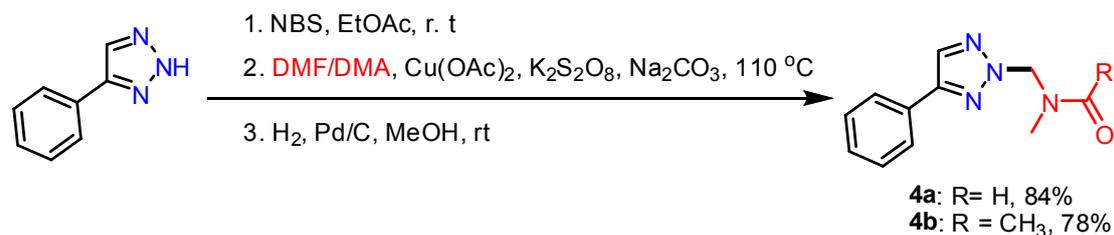
According to the <sup>13</sup>C NMR spectra of **3q** and **3q'**, there were significant differences between the CH<sub>2</sub> groups (which linked to 1,2,3-triazole) of the *N*<sup>1</sup>(**3q'**) and *N*<sup>2</sup>(**3q**) isomers. The chemical shift of CH<sub>2</sub> group for *N*<sup>2</sup> isomer was 67.8 and 61.5 ppm (for two conformers), which located in lower field than *N*<sup>1</sup> isomer's (62.9 and 56.1 ppm, for two conformers). Also, the corresponding <sup>13</sup>C NMR spectra of two regioisomeric products for 4-aryl-substituted 1,2,3-triazoles gave the same information, such as **4e** and its *N*<sup>1</sup> isomer (**5e**). The chemical shift of CH<sub>2</sub> group for **4e** was located in 67.8 and 63.6 ppm. By contrast, the NMR spectra for **5e** showed one conformer, and the <sup>13</sup>C NMR chemical shift for CH<sub>2</sub> group was 60.1 ppm, which located in higher field. For other DMF or DMA coupling products, the corresponding *N*<sup>2</sup> isomers gave higher <sup>13</sup>C NMR chemical shifts for CH<sub>2</sub> group, which located in about 67 and 62 ppm (two conformers), such as **3d** and **3s**, while *N*<sup>1</sup>

isomer gave lower chemical shifts. Moreover, some  $N^1$  isomers only gave one conformer signal (Figure 1). Indeed, the  $N^1$  and  $N^2$  isomers of other triazoles and benzotriazoles showed the similar results.<sup>13, 20</sup> As for the coupling reactions with NMP, it was found that the reactions gave only  $N^2$ -coupling products because of a big steric hindrance.



**Figure 1.** Analysis of  $^{13}\text{C}$  NMR Spectra

In order to selectively synthesize 2,4-disubstituted 1,2,3-triazoles, a stepwise reaction strategy was developed. As shown in Scheme 2, the 4-phenyltriazole was first reacted with NBS (*N*-bromosuccinimide) to introduce a Br atom on the triazole ring<sup>14b,21</sup>, then reacted with DMF or DMA. The corresponding  $N^2$ -amidoalkylated bromotriazoles were further reduced under catalytic hydrogenation condition ( $\text{H}_2$ , Pd/C) to give the 2-amidoalkylated 4-phenyl-1,2,3-triazoles in higher overall yields, which provide a highly selective synthesis of 2-amidoalkylated 4-aryl-1,2,3-triazoles (Scheme 3).

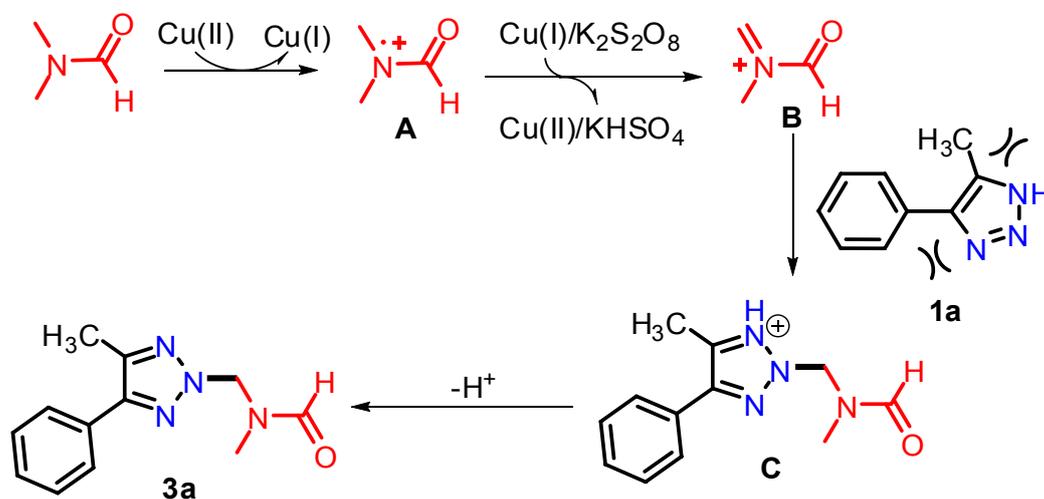


15  
16

**Scheme 3.** An Alternative Synthesis of *N*<sup>2</sup>-Amidoalkylated 4-Aryl-1,2,3-Triazoles

17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33

Based on experimental results and existing literature<sup>2, 22</sup>, a postulated mechanism for this transformation has been put forward (Scheme 3). The initiation step is proposed to be a single electron transfer between Cu(II) and DMF to produce radical cation **A**. Next, the successive hydrogen atom abstraction from the radical cation **A** in the aid of Cu(I)/ K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> forms the iminium ion **B**, which is selective attacked by *N*<sup>2</sup> site of triazole **1a** to form the intermediate **C**. Then lose a proton to form the product **3a**. The addition of a base is to neutralize the formed acid and further improve the reaction conversion (Scheme 4).



52  
53

**Scheme 4.** A Plausible Mechanism for Oxidative *N*<sup>2</sup>-Coupling

54  
55

CONCLUSION

56  
57  
58  
59  
60

In conclusion, we report here a copper-catalyzed oxidative cross-dehydrogenative C-H/N-H

1  
2  
3  
4  
5  
6  
7 coupling reaction between *N,N*-dialkylamides and *NH*-1,2,3-triazoles. This method can give  
8  
9  
10 *N*-amidoalkylated 1,2,3-triazoles with high efficiency. By tuning the substituents on the  
11  
12 *NH*-1,2,3-triazoles, the *N*<sup>2</sup>-amidoalkylated 1,2,3-triazoles can be obtained in high selectivity  
13  
14 and yield.

## 17 EXPERIMENTAL SECTION

20 **General Information.** All reagents were purchased from commercial sources and used  
21  
22 without further purification. <sup>1</sup>H NMR spectra were determined on 400 and 600 MHz  
23  
24 spectrometer as solutions in CDCl<sub>3</sub>. Chemical shifts are expressed in parts per million (δ),  
25  
26 and the signals were reported as s (singlet), d (doublet), t (triplet), m (multiplet), dd (double  
27  
28 doublet), and coupling constants (*J*) were given in Hz. <sup>13</sup>C NMR spectra were recorded at  
29  
30 100 and 150 MHz in CDCl<sub>3</sub> solution. Chemical shifts as internal standard are referenced to  
31  
32 CDCl<sub>3</sub> (δ = 7.26 ppm for <sup>1</sup>H and δ = 77.0 ppm for <sup>13</sup>C NMR) as internal standard.  
33  
34 High-resolution mass spectra (HRMS) were performed with Fourier-transform mass  
35  
36 spectrometer by electrospray ionization and time of flight mass spectrometer by  
37  
38 electrospray ionization. TLC was done on silica gel coated glass slides. All solvents were  
39  
40 dried before use. 4-phenyltriazoles and 4-Methyl-5-phenyl-2*H*-1,2,3-triazoles were  
41  
42 prepared according to literature procedure.<sup>14a</sup>

50 **Typical Experimental Procedure: Synthesis of Br-substituted 1,2,3-triazole.** To a  
51  
52 solution of 4-phenyltriazole (500 mg, 3.45 mmol) in EtOAc (10 mL), was added NBS  
53  
54 (920 mg, 5.18 mmol) at room temperature, the reaction was checked by TLC, After the  
55  
56 completion of the reaction, the mixture was poured into water, and extracted by ethyl  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7 acetate, washed with NaCl (aq), dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, then the solvent was  
8 removed under reduced pressure to obtain a yellow solid, and the crude product was  
9 purified by column chromatography [silica gel, PE-EtOAc (10:1 to 5:10)] to give (**1n**) a  
10 white solid (90%, 694mg).  
11  
12  
13  
14  
15  
16

17 **Typical Experimental Procedure: Synthesis of *N*-amidoalkylated 1,2,3-triazoles.** A 50  
18 mL round-bottom flask equipped with a magnetic stir bar was charged with  
19 4-methyl-5-phenyl-2*H*-1,2,3-triazole (47.7 mg, 0.3 mmol), DMF (3 mL as solvent), and  
20 K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (162.3 mg, 0.6 mmol), Na<sub>2</sub>CO<sub>3</sub> (35 mg, 0.33 mmol), Cu(OAc)<sub>2</sub> (12 mg, 0.05 mmol).  
21 The mixture was then stirred at 110 °C in air for 0.5-6 h (TLC monitoring), then poured into  
22 H<sub>2</sub>O (20 mL), adjusted the pH to neutral using 10% HCl, and extracted with EtOAc (3×20  
23 mL). Next the organic phase was evaporated under vacuum, and the crude product was  
24 purified by column chromatography [silica gel, PE-EtOAc (10:1 to 2:1)] to give (**3a**) a  
25 white solid (89%, 61mg).  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39

40 **The procedure for the synthesis of 4a or 4b:** To a solution of 4-phenyltriazole (200 mg,  
41 1.38 mmol) in EtOAc (4 mL), was added NBS (367.8 mg, 2.07 mmol) at room temperature,  
42 the reaction was checked by TLC. The work-up procedure was the same as the synthesis of  
43 Br-substituted 1,2,3-triazoles, the obtained crude yellow solid can be used for the next step  
44 without further purification. The yellow solid reacted with DMF or DMA (4 mL) to obtain  
45 the *N*<sup>2</sup>-amidoalkylated 4-bromo-triazoles (mentioned above as Typical Experimental  
46 Procedure: Synthesis of *N*-amidoalkylated 1,2,3-triazoles), then the *N*<sup>2</sup>-amidoalkylated  
47 4-bromo-triazoles and 10% Pd/C in methanol (8 mL) was stirred under atmospheric H<sub>2</sub> in a  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

50 mL schlenk bottle for 2 hours. Then triethylamine (3 mmol) was added to the mixture.

After being stirred for 10 min, the mixture was filtered by diatomite. The filtrate was diluted with ethyl acetate (20 mL) and washed with water (20 mL). Then the organic phase was evaporated under vacuum, and the crude product was purified by column chromatography [silica gel, PE-EtOAc (10:1 to 2:1)] to give product **4a** (82%, 333mg) or **4b** (78%, 332mg).

*4-bromo-5-phenyl-2H-1,2,3-triazole (1n)* White solid (90%, 694mg) m.p. 142–144 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.93 (s, 1H), 7.91 (s, 1H), 7.52–7.49 (m, 1H), 7.47 (d, J = 4.0 Hz, 1H), 7.46–7.42 (m, 1H), 3.52 (s, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 129.2, 128.7, 127.4. HRMS (ESI-FTMS) m/z [M + H]<sup>+</sup> calcd. for C<sub>8</sub>H<sub>6</sub>BrN<sub>3</sub>: 223.9818, found 223.9817.

*4-bromo-5-(2-chlorophenyl)-2H-1,2,3-triazole (1o)* White solid (88%, 633mg) m.p. 81–82 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.54–7.52 (m, 1H), 7.50–7.47 (m, 1H), 7.45–7.41 (m, 1H), 7.40–7.36 (m, 1H), 3.53 (s, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 133.9, 131.8, 130.9, 130.2, 127.1, 126.8, 123.2. HRMS (ESI-FTMS) m/z [M + H]<sup>+</sup> calcd. for C<sub>8</sub>H<sub>6</sub>BrClN<sub>3</sub>: 257.9428, found 257.9427.

*4-bromo-5-(2-bromophenyl)-2H-1,2,3-triazole (1q)* White solid (85%, 575mg) m.p. 92–93 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.70 (d, J = 8.4 Hz, 1H), 7.46–7.31 (m, 3H), 3.53 (d, J = 0.8 Hz, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 133.2, 131.8, 131.0, 127.3, 123.5. HRMS (ESI-FTMS) m/z [M + H]<sup>+</sup> calcd. for C<sub>8</sub>H<sub>6</sub>Br<sub>2</sub>N<sub>3</sub>: 301.89230, found 301.89246.

*4-bromo-5-(4-chlorophenyl)-2H-1,2,3-triazole (1s)* White solid (92%, 662mg) m.p. 192–193 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.89 (d, J = 8.4 Hz, 2H), 7.46 (d, J = 2.0 Hz, 1H), 7.45 (d, J = 2.0 Hz, 1H), 3.50 (s, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 123.0, 128.7.

1  
2  
3  
4  
5  
6  
7 HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  $C_8H_6BrClN_3$ : 257.9428, found 257.9426.

8  
9  
10 *4-bromo-5-(2,4-dichlorophenyl)-2H-1,2,3-triazole (1u)* White solid (90%, 616mg) m.p.  
11  
12 148–152 °C.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.56 (d,  $J = 2.0$  Hz, 1H), 7.42 (d,  $J = 8.0$  Hz,  
13  
14 1H), 7.38–7.36 (m, 1H), 3.51 (d,  $J = 0.8$  Hz, 1H).  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  136.4,  
15  
16 134.8, 132.9, 130.1, 127.3. ESI-MS( $m/z$ ): 290.9  $[M]^+$ . HRMS (ESI-TOF)  $m/z$   $[M + H]^+$   
17  
18 calcd. for  $C_8H_6BrCl_2N_3$ : 291.9038, found 291.9034.

19  
20  
21  
22 *N-methyl-N-((4-methyl-5-phenyl-2H-1,2,3-triazol-2-yl)methyl)formamide (3a)* White solid  
23  
24 (92%, 132 mg) m.p. 60–62 °C.  $^1H$  NMR (600 MHz,  $CDCl_3$ )  $\delta$  8.34 (s, 0.73H), 8.01 (s,  
25  
26 0.25H), 7.56 (t,  $J = 7.8$  Hz, 2H), 7.31 (t,  $J = 7.2$  Hz, 2H), 7.24 (t,  $J = 7.2$  Hz, 1H), 5.71 (s,  
27  
28 0.49H), 5.53 (s, 1.59H), 2.87 (s, 0.7H), 2.80 (s, 2.16H), 2.35 (s, 3H).  $^{13}C$  NMR (150 MHz,  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
 $CDCl_3$ )  $\delta$  162.8, 162.6, 146.0, 145.5, 142.2, 141.8, 130.4, 130.3, 128.4, 128.0, 127.8, 126.9,  
66.4, 60.1, 33.3, 29.1, 11.5. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  $C_{12}H_{15}N_4O$ :  
231.1240; found, 231.1235.

*N-methyl-N-((4-methyl-5-phenyl-2H-1,2,3-triazol-2-yl)methyl)acetamide (3b)* Light yellow  
oil (80%, 123 mg).  $^1H$  NMR (600 MHz,  $CDCl_3$ )  $\delta$  7.67 (t,  $J = 7.8$  Hz, 2H), 7.42 (q,  $J = 7.8$   
Hz, 2H), 7.36–7.33 (m, 1H), 5.87 (s, 0.87H), 5.71 (s, 1.25H), 3.08 (s, 1.23H), 3.03 (s,  
1.82H), 2.46 (s, 3H), 2.43 (s, 1.85H), 2.13 (s, 1.24H).  $^{13}C$  NMR (150 MHz,  $CDCl_3$ )  $\delta$  171.2,  
145.8, 145.5, 142.0, 141.7, 130.8, 130.6, 128.5, 128.1, 127.9, 127.1, 67.3, 63.1, 35.1, 33.0,  
21.7, 11.7. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  $C_{13}H_{17}N_4O$ : 245.1397; found,  
245.1398.

*N-methyl-N-((4-methyl-5-(p-tolyl)-2H-1,2,3-triazol-2-yl)methyl)formamide (3c)* Light

1  
2  
3  
4  
5  
6  
7 yellow solid (89%, 125mg) m.p. 66–68 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.43 (d, *J* = 8.4  
8 Hz, 0.64H), 8.11 (d, *J* = 9.0 Hz, 0.22H), 7.61–7.45 (m, 2H), 7.28–7.15 (m, 2H), 5.80 (d, *J* =  
9  
10 9.0 Hz, 0.48H), 5.61 (d, *J* = 9.0 Hz, 1.53H), 2.99–2.95 (m, 0.7H), 2.92–2.87 (m, 2.12H),  
11 2.45–2.39 (m, 3H), 2.38–2.30 (m, 3H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 162.9, 162.7, 146.3,  
12 145.8, 142.2, 141.8, 138.0, 129.2, 127.5, 126.9, 66.5, 60.2, 33.4, 29.3, 21.1, 11.6. HRMS  
13 (ESI-FTMS) *m/z* [M + H]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>17</sub>N<sub>4</sub>O: 245.1397; found, 245.1400.  
14  
15  
16  
17  
18  
19  
20  
21

22 *N*-methyl-*N*-((4-methyl-5-(*p*-tolyl)-2*H*-1,2,3-triazol-2-yl)methyl)acetamide (**3d**) Colorless  
23 oil (78%, 116 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.61–7.52 (m, 2H), 7.25 (t, *J* = 8.4 Hz,  
24 2H), 5.89 (s, 0.8H), 5.73 (s, 1.21H), 3.10 (s, 1.20H), 3.05 (s, 1.77H), 2.47 (s, 3H), 2.45 (s,  
25 1.76H), 2.39 (d, *J* = 4.8 Hz, 3H), 2.16 (s, 1.14H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 171.4,  
26 171.3, 146.0, 145.7, 141.9, 141.7, 138.1, 137.9, 129.3, 128.1, 127.8, 127.1, 67.4, 63.2, 35.1,  
27 33.2, 29.7, 21.8, 21.3, 11.8. HRMS (ESI-FTMS) *m/z* [M + H]<sup>+</sup> calcd. for C<sub>14</sub>H<sub>19</sub>N<sub>4</sub>O:  
28 259.1553; found, 259.1558.  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39

40 *N*-((4-(4-methoxyphenyl)-5-methyl-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (**3e**)  
41 Light yellow oil (88%, 120 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.45 (s, 0.72H), 8.13 (s,  
42 0.24H), 7.67–7.48 (m, 2H), 7.06–6.85 (m, 2H), 5.81 (s, 0.52H), 5.63 (s, 1.59H), 3.82 (d, *J* =  
43 3.6 Hz, 3H), 2.99 (s, 0.69H), 2.92 (s, 2.10H), 2.44 (d, *J* = 1.2 Hz, 3H). <sup>13</sup>C NMR (150 MHz,  
44 CDCl<sub>3</sub>) δ 163.0, 162.7, 159.5, 146.2, 145.7, 142.0, 141.7, 128.4, 123.0, 114.0, 66.6, 60.3,  
45 55.2, 33.5, 29.4, 11.7. HRMS (ESI-FTMS) *m/z* [M + H]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>17</sub>N<sub>4</sub>O<sub>2</sub>: 261.1346;  
46 found, 261.1340.  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57

58 *N*-((4-(4-methoxyphenyl)-5-methyl-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (**3f**)  
59  
60

1  
2  
3  
4  
5  
6  
7 Light yellow oil (72%, 104 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.60 (t,  $J = 8.4$  Hz, 2H),  
8  
9  
10 6.96 (t,  $J = 7.8$  Hz, 2H), 5.87 (s, 0.79H), 5.71 (s, 1.17H), 3.83 (d,  $J = 3.7$  Hz, 3H), 3.09 (s,  
11  
12 1.11H), 3.04 (s, 1.61H), 2.44 (d,  $J = 3.0$  Hz, 4.79H), 2.15 (s, 1.2H).  $^{13}\text{C}$  NMR (150 MHz,  
13  
14  $\text{CDCl}_3$ )  $\delta$  171.3, 159.6, 159.4, 145.8, 145.5, 141.6, 141.4, 128.5, 123.5, 123.2, 114.1, 67.3,  
15  
16  
17 63.1, 55.3, 35.1, 33.1, 21.8, 11.7. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{14}\text{H}_{19}\text{N}_4\text{O}_2$ :  
18  
19 275.1503; found, 275.1501.  
20  
21

22  
23 *N*-((4-(4-chlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (3g)

24  
25 Light yellow solid (91%, 124mg) m.p. 74–76 °C.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.45 (s,  
26  
27 0.73H), 8.14 (s, 0.23H), 7.60 (t,  $J = 7.8$  Hz, 2H), 7.38 (t,  $J = 8.4$  Hz, 2H), 5.82 (s, 0.5H),  
28  
29 5.65 (s, 1.5H), 3.01 (s, 0.77H), 2.91 (s, 2.23H), 2.44 (s, 3H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$   
30  
31 163.0, 162.8, 145.3, 144.8, 142.4, 142.1, 134.2, 134.0, 129.2, 129.2, 129.0, 128.9, 128.8,  
32  
33 128.3, 66.8, 60.4, 33.6, 29.5, 11.8. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  
34  
35  $\text{C}_{12}\text{H}_{14}\text{ClN}_4\text{O}$ : 265.0851; found, 265.0844.  
36  
37  
38

39  
40  
41 *N*-((4-(4-chlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (3h)

42  
43 Light yellow oil (75%, 108 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.62 (t,  $J = 8.4$  Hz, 2H),  
44  
45 7.43–7.36 (m, 2H), 5.87 (s, 0.93H), 5.73 (s, 1.22H), 3.11 (s, 1.27H), 3.04 (s, 1.7H), 2.46 (s,  
46  
47 3H), 2.44 (s, 1.67H), 2.15 (s, 1.32H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  171.3, 144.9, 144.5,  
48  
49 142.0, 141.8, 134.2, 133.9, 129.4, 129.2, 128.8, 128.4, 67.5, 63.3, 35.3, 33.2, 21.8, 11.8.  
50  
51  
52  
53 HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{13}\text{H}_{16}\text{ClN}_4\text{O}$ : 279.1007; found, 279.1012.  
54  
55

56  
57 *N*-((4-(2-chlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (3i)

58  
59 Light yellow oil (94%, 129 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.42 (s, 0.70H), 8.12 (s,  
60

0.23H), 7.44 (d,  $J = 7.8$  Hz, 1H), 7.35–7.25 (m, 3H), 5.83 (d,  $J = 1.2$  Hz, 0.54H), 5.66 (d,  $J = 1.2$  Hz, 1.59H), 2.97 (d,  $J = 1.2$  Hz, 0.78H), 2.90 (d,  $J = 1.2$  Hz, 2.22H), 2.22 (d,  $J = 1.2$  Hz, 3H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  162.9, 162.7, 145.2, 144.7, 144.3, 143.9, 133.7, 131.6, 130.2, 130.0, 129.8, 129.7, 129.6, 129.3, 126.7, 126.6, 66.7, 60.4, 33.4, 29.4, 10.7. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{12}\text{H}_{14}\text{ClN}_4\text{O}$ : 265.0851; found, 265.0844.

*N*-((4-(2-chlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (3j)

Light yellow oil (87%, 125 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.50–7.44 (m, 1H), 7.38–7.29(m, 3H), 5.91 (s, 0.87H), 5.76 (s, 1.21H), 3.09 (s, 1.2H), 3.05 (s, 1.68H), 2.43 (s, 1.73H), 2.26 (s, 3H), 2.16 (s, 1.26H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  171.4, 144.8, 144.5, 144.0, 143.7, 133.8, 131.8, 131.7, 130.1, 130.1, 130.0, 129.9, 129.8, 129.6, 126.7, 67.5, 63.3, 35.1, 33.2, 21.8, 10.8. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{13}\text{H}_{16}\text{ClN}_4\text{O}$ : 279.1007; found, 279.1004.

*N*-((4-(2-bromophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (3k)

White solid (89%, 121 mg) m.p. 122–124 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.70–7.66 (m, 1H), 7.40–7.29 (m, 3H), 5.93 (s, 1H), 5.78 (s, 1.18H), 3.10 (s, 1.27H), 3.07 (s, 1.78H), 2.45 (s, 1.8H), 2.27 (s, 3H), 2.18 (s, 1.2H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  171.4, 146.34, 143.8, 133.1, 132.2, 131.5, 130.3, 127.3, 123.9, 67.8, 63.3, 35.0, 33.2, 21.8, 10.8. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{13}\text{H}_{16}\text{BrN}_4\text{O}$ : 323.0502, found 323.0504.

*N*-((4-(2,4-dichlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (3l)

Light yellow oil (89%, 116 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.39 (s, 0.65H), 8.09 (s, 0.21H), 7.45–7.41(m, 1H), 7.27–7.20 (m, 2H), 5.79 (d,  $J = 1.2$  Hz, 0.54H), 5.63 (d,  $J = 1.2$

Hz, 1.54H), 2.95 (d,  $J = 1.2$  Hz, 0.77H), 2.87 (d,  $J = 1.2$  Hz, 2.18H), 2.18 (d,  $J = 1.2$  Hz, 3H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  162.9, 162.7, 144.2, 143.9, 143.7, 135.4, 135.3, 134.4, 132.4, 129.6, 128.2, 128.0, 127.0, 66.8, 60.5, 33.5, 29.4, 10.7. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{H}$ ] $^+$  calcd. for  $\text{C}_{12}\text{H}_{13}\text{Cl}_2\text{N}_4\text{O}$ : 299.0461; found, 299.0465.

*N-((4-(2,4-dichlorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)methyl)-N-methylacetamide (3m)*

Light yellow oil (80%, 110 mg).  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.51 (d,  $J = 9.0$  Hz, 1H), 7.35–7.27 (m, 2H), 5.90 (s, 0.82H), 5.76 (s, 1.05H), 3.10 (s, 1.32H), 3.05 (s, 1.73H), 2.43 (s, 1.68H), 2.25 (s, 3H), 2.16 (s, 1.3H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  171.4, 143.9, 143.6, 135.5, 135.3, 134.6, 132.6, 132.5, 129.8, 128.6, 128.3, 127.1, 67.6, 63.4, 35.2, 33.2, 21.8, 10.8. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{H}$ ] $^+$  calcd. for  $\text{C}_{13}\text{H}_{15}\text{Cl}_2\text{N}_4\text{O}$ : 313.0617; found, 313.0624.

*N-((4-bromo-5-phenyl-2H-1,2,3-triazol-2-yl)methyl)-N-methylformamide (3n)* Light yellow

solid (91%, 118 mg) m.p. 68–70 °C.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.46 (s, 0.73H), 8.16 (s, 0.27H), 7.90 (d,  $J = 7.2$  Hz, 2H), 7.48–7.40 (m, 3H), 5.87 (s, 0.61H), 5.70 (s, 1.66H), 3.05 (s, 0.86H), 2.95 (s, 2.25H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  162.9, 146.8, 146.3, 129.2, 129.0, 128.7, 128.6, 128.5, 128.3, 127.3, 121.4, 121.0, 67.6, 61.3, 33.7, 29.6. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{H}$ ] $^+$  calcd. for  $\text{C}_{11}\text{H}_{12}\text{BrN}_4\text{O}$ : 295.0189; found, 295.0186.

*N-((4-bromo-5-(2-chlorophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylformamide (3o)*

White semisolid (92%, 117 mg).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.47 (s, 0.73H), 8.19 (s, 0.29H), 7.55–7.49 (m, 1H), 7.45–7.34 (m, 3H), 5.92 (s, 0.62H), 5.76 (s, 1.48H), 3.08 (s, 0.83H), 2.99 (s, 2.2H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  162.9, 146.5, 134.0, 131.8, 130.9,

1  
2  
3  
4  
5  
6  
7 130.1, 127.4, 126.7, 124.4, 124.0, 67.8, 61.5, 33.9, 29.7. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$   
8  
9  
10 calcd. for  $C_{11}H_{11}BrClN_4O$ : 328.9799; found, 328.9800.

11  
12  
13 *N-((4-bromo-5-(2-chlorophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylacetamide* (3p)

14  
15 Colorless oil (82%, 109 mg).  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.54–7.49 (m, 1H), 7.46–7.32  
16  
17 (m, 3H), 5.96 (s, 1H), 5.83 (s, 1.07H), 3.16 (s, 1.52H), 3.08 (s, 1.62H), 2.44 (s, 1.62H), 2.19  
18  
19 (s, 1.49H).  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  171.4, 171.2, 146.0, 145.7, 134.0, 131.9, 131.7,  
20  
21 130.8, 130.7, 130.1, 130.0, 127.9, 127.5, 126.7, 126.6, 123.9, 123.5, 68.5, 64.4, 35.4, 33.3,  
22  
23 21.7, 21.6. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  $C_{12}H_{13}BrClN_4O$ : 342.9956, found  
24  
25 342.9962.

26  
27  
28  
29  
30 *N-((4-bromo-5-(2-bromophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylformamide* (3q)

31  
32 White solid (91%, 112 mg) m.p. 78–80 °C.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.48 (s, 0.75H),  
33  
34 8.19 (s, 0.28H), 7.71 (d,  $J = 7.2$  Hz, 1H), 7.46–7.30 (m, 3H), 5.93 (s, 0.6H), 5.76 (s, 1.54H),  
35  
36 3.08 (s, 0.88H), 2.99 (s, 2.27H).  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  162.9, 147.8, 133.2, 131.9,  
37  
38 131.1, 129.4, 127.3, 124.3, 123.7, 67.8, 29.6. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  
39  
40  $C_{11}H_{11}Br_2N_4O$ : 372.9294, found 372.9290.

41  
42  
43  
44  
45 *N-((4-bromo-5-(2-bromophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylacetamide* (3r)

46  
47 Colorless oil (84%, 108 mg).  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.75–7.65 (m, 1H), 7.43–7.38  
48  
49 (m, 2H), 7.37–7.30 (m, 1H), 5.97 (s, 1H), 5.84 (s, 1.11H), 3.15 (s, 1.51H), 3.08 (s, 1.65H),  
50  
51 2.44 (s, 1.66H), 2.19 (s, 1.52H).  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  177.3, 171.4, 147.3, 146.9,  
52  
53 133.2, 131.9, 130.9, 129.89, 129.6, 127.2, 123.9, 123.6, 123.4, 77.3, 68.5, 64.4, 35.4, 33.3,  
54  
55 29.5, 21.7. HRMS (ESI-FTMS)  $m/z$   $[M + H]^+$  calcd. for  $C_{12}H_{13}Br_2N_4O$ : 386.9451, found  
56  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8 386.9449.

9  
10 *N-((4-bromo-5-(4-chlorophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylformamide* (**3s**)

11  
12 White solid (86%, 110 mg) m.p. 65–66 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.48 (s, 0.71H),  
13 8.18 (s, 0.28H), 7.94–7.81 (m, 2H), 7.49–7.37 (m, 2H), 5.88 (s, 0.59H), 5.72 (s, 1.46H),  
14 3.08 (s, 0.88H), 2.97 (s, 2.16H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 162.9, 145.9, 145.4, 135.3,  
15 128.9, 128.6, 126.9, 121.5, 121.0, 67.8, 61.5, 33.8, 29.7. HRMS (ESI-FTMS) m/z [M + H]<sup>+</sup>  
16  
17  
18  
19  
20  
21  
22 calcd. for C<sub>11</sub>H<sub>11</sub>BrClN<sub>4</sub>O: 328.9799; found, 328.9797.

23  
24  
25 *N-((4-bromo-5-(4-chlorophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylacetamide* (**3t**)

26  
27 White solid (81%, 108 mg) m.p. 89–90 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.98–7.76 (m,  
28 2H), 7.53–7.33 (m, 2H), 5.92 (s, 1H), 5.80 (s, 1H), 3.16 (s, 1.47H), 3.06 (s, 1.46H), 2.44 (s,  
29 1.5H), 2.18 (s, 1.53H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 171.4, 171.2, 145.5, 145.0, 135.3,  
30 135.0, 128.7, 127.3, 127.0, 121.0, 120.6, 68.5, 64.5, 35.6, 33.3, 21.7. HRMS (ESI-FTMS)  
31  
32  
33  
34  
35  
36  
37 m/z [M + H]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>13</sub>BrClN<sub>4</sub>O: 342.9956, found 342.9959.

38  
39  
40 *N-((4-bromo-5-(2,4-dichlorophenyl)-2H-1,2,3-triazol-2-yl)methyl)-N-methylacetamide* (**3u**)

41  
42 Colorless oil (87%, 112 mg). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.54 (dd, *J* = 7.2, 1.6 Hz, 1H),  
43 7.40–7.32 (m, 2H), 5.95 (s, 1H), 5.83 (s, 1H), 3.16 (s, 1.5H), 3.07 (s, 1.5H), 2.43 (s, 1.5H),  
44 2.19 (s, 1.5H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 171.5, 171.2, 145.0, 144.7, 136.3, 136.1,  
45 134.9, 132.6, 130.0, 127.1, 126.5, 126.1, 123.9, 123.5, 68.6, 64.6, 35.5, 33.3, 21.7. HRMS  
46  
47  
48  
49  
50  
51  
52  
53 (ESI-FTMS) m/z [M + H]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>12</sub>BrCl<sub>2</sub>N<sub>4</sub>O: 376.9566, found 376.9561.

54  
55  
56 *5-(4-(2-bromophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one* (**3v**) Light  
57  
58  
59  
60 yellow oil (82%, 115 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.68 (d, *J* = 7.8 Hz, 1H), 7.38 (q,

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

$J = 7.2$  Hz, 1H), 7.36–7.32 (m, 1H), 7.32–7.27 (m, 1H), 5.98 (m, 1H), 2.97–2.90 (m, 1H), 2.75 (s, 3H), 2.63–2.56 (m, 2H), 2.48 (m, 1H), 2.26 (s, 3H).  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  175.1, 146.1, 143.6, 133.0, 131.7, 130.3, 127.2, 123.7, 76.8, 28.9, 27.4, 24.9, 10.7. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{H}]^+$  calcd. for  $\text{C}_{14}\text{H}_{16}\text{BrN}_4\text{O}$ : 335.0502; found, 335.0501.

*5-(4-(2-bromo-4-fluorophenyl)-5-methyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one*  
(**3w**) Colorless oil (92%, 254 mg).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.43–7.38 (m, 1H), 7.32–7.25 (m, 1H), 7.12–7.05 (m, 1H), 5.94 (dd,  $J = 7.2, 2.4$  Hz, 1H), 2.97–2.83 (m, 1H), 2.71 (s, 3H), 2.62–2.52 (m, 2H), 2.51–2.41 (m, 1H), 2.22 (s, 3H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  175.1, 163.7, 161.2, 145.3, 143.7, 132.7, 132.6, 127.93, 127.90, 124.2, 124.1, 120.5, 120.2, 114.7, 114.5, 78.5, 28.9, 27.5, 24.9, 10.7. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{Na}]^+$  calcd. for  $\text{C}_{14}\text{H}_{14}\text{BrFN}_4\text{NaO}$ : 375.0227, found 375.0223.

*5-(4-bromo-5-phenyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one* (**3x**) White solid (90%, 258 mg) m.p. 116–118 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.92 (dd,  $J = 8.4, 1.6$  Hz, 2H), 7.51–7.38 (m, 3H), 5.99 (dd,  $J = 7.6, 1.6$  Hz, 1H), 2.99–2.85 (m, 1H), 2.78 (s, 3H), 2.68–2.56 (m, 1H), 2.55–2.44 (m, 2H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  175.1, 146.3, 129.2, 128.63, 128.57, 127.3, 120.9, 79.6, 28.7, 27.7, 25.1. HRMS (ESI-FTMS)  $m/z$   $[\text{M} + \text{Na}]^+$  calcd. for  $\text{C}_{13}\text{H}_{13}\text{BrN}_4\text{NaO}$ : 343.0165, found 343.0158.

*5-(4-benzoyl-5-phenyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one* (**3y**) White solid (89%, 247 mg) m.p. 112–114 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.10–7.97 (m, 2H), 7.80 (dd,  $J = 6.4, 3.2$  Hz, 2H), 7.61 (t,  $J = 7.6$  Hz, 1H), 7.48 (t,  $J = 7.6$  Hz, 2H), 7.42–7.37 (m, 3H), 6.10 (dd,  $J = 7.6, 1.6$  Hz, 1H), 3.00–2.87 (m, 1H), 2.82 (s, 3H), 2.73–2.62 (m, 1H),

1  
2  
3  
4  
5  
6  
7 2.62–2.46 (m, 2H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  187.6, 175.1, 150.2, 142.5, 136.9, 133.5,  
8  
9  
10 130.3, 129.3, 129.1, 128.6, 128.4, 128.3, 79.6, 28.6, 27.8, 25.3. HRMS (ESI-FTMS)  $m/z$  [ $\text{M}$   
11  
12 +  $\text{Na}$ ] $^+$  calcd. for  $\text{C}_{20}\text{H}_{18}\text{N}_4\text{NaO}_2$ : 369.1322, found 369.1328.

13  
14  
15 *N*-((4-benzoyl-5-phenyl-2H-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (**3z**) White solid  
16  
17 (85%, 218 mg) m.p. 116–118 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.51 (s, 1H), 8.18 (s, 1H),  
18  
19 7.58–7.52 (m, 4H), 7.39–7.33 (m, 6H), 5.94 (s, 1H), 5.75 (s, 2H), 3.07 (s, 1H), 3.01 (s, 2H).  
20  
21  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  163.0, 162.7, 145.8, 145.3, 130.4, 130.2, 128.6, 128.5,  
22  
23 128.42, 128.40, 128.2, 128.1, 67.0, 60.7, 33.6, 29.6. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{Na}$ ] $^+$   
24  
25 calcd. for  $\text{C}_{18}\text{H}_{16}\text{N}_4\text{NaO}_2$ : 343.1166, found 343.1169.

26  
27  
28  
29  
30 *5*-(4,5-diphenyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one (**3aa**) Light yellow solid  
31  
32 (89%, 256 mg) m.p. 92–94 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.57–7.52 (m, 4H), 7.40–7.35  
33  
34 (m, 6H), 6.06 (dd,  $J = 6.8, 2.8$  Hz, 1H), 3.04–2.93 (m, 1H), 2.83 (s, 3H), 2.68–2.58 (m, 2H),  
35  
36 2.56–2.46 (m, 1H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  175.3, 145.4, 130.5, 128.59, 128.55,  
37  
38 128.2, 78.9, 28.9, 27.7, 25.2. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{Na}$ ] $^+$  calcd. for  $\text{C}_{19}\text{H}_{18}\text{N}_4\text{NaO}$ :  
39  
40  
41 341.1373, found 341.1379.

42  
43  
44  
45 *5*-(4-(2-bromophenyl)-5-phenyl-2H-1,2,3-triazol-2-yl)-1-methylpyrrolidin-2-one (**3ab**)  
46  
47 White solid (92%, 243 mg) m.p. 128–130 °C.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.68 (dd,  $J =$   
48  
49 8.0, 0.8 Hz, 1H), 7.50–7.41 (m, 2H), 7.43–7.36 (m, 2H), 7.34–7.24 (m, 4H), 6.07 (t,  $J = 4.8$   
50  
51 Hz, 1H), 3.05–2.92 (m, 1H), 2.81 (s, 3H), 2.72–2.60 (m, 2H), 2.56–2.46 (m, 1H).  $^{13}\text{C}$  NMR  
52  
53 (100 MHz,  $\text{CDCl}_3$ )  $\delta$  175.2, 146.0, 144.5, 133.2, 132.4, 131.8, 130.6, 130.1, 128.54, 128.47,  
54  
55 127.5, 126.8, 124.2, 79.0, 28.9, 27.6, 25.0. HRMS (ESI-FTMS)  $m/z$  [ $\text{M} + \text{Na}$ ] $^+$  calcd. for  
56  
57  
58  
59  
60

C<sub>19</sub>H<sub>17</sub>BrN<sub>4</sub>NaO: 419.0478, found 419.0483.

*N*-((4,5-diphenyl-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (**3ac**) White solid (87%, 229 mg) m.p. 105–107 °C. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.49 (s, 1H), 8.17 (s, 1H), 8.03 (dd, *J* = 16.5, 7.5 Hz, 2H), 7.79–7.76 (m, 2H), 7.59 (d, *J* = 7.5 Hz, 1H), 7.50–7.43 (m, 2H), 7.43–7.34 (m, 3H), 5.96 (s, 1H), 5.79 (s, 1H), 3.07 (s, 1H), 2.99 (s, 2H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 187.6, 187.5, 162.9, 162.8, 150.4, 150.1, 143.0, 142.6, 136.8, 136.7, 133.6, 133.5, 130.3, 130.2, 129.3, 129.2, 129.1, 128.9, 128.6, 128.5, 128.4, 128.32, 128.27, 67.6, 61.3, 33.8, 29.6. HRMS (ESI-FTMS) *m/z* [M + Na]<sup>+</sup> calcd. for C<sub>17</sub>H<sub>16</sub>N<sub>4</sub>NaO: 315.1206, found 315.1208.

*N*-((4-bromo-5-(2-bromophenyl)-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylformamide (**3q'**) White solid (5%, 61 mg) m.p. 196–198 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.84 (s, 0.38H), 7.80 (dd, *J* = 7.8, 1.2 Hz, 0.57H), 7.74 (dd, *J* = 7.8, 1.2 Hz, 0.37H), 7.55–7.42 (m, 2.54H), 7.29–7.26 (m, 1.16H), 5.77 (d, *J* = 13.8 Hz, 0.4H), 5.69 (d, *J* = 4.8 Hz, 0.42H), 5.67 (d, *J* = 5.4 Hz, 0.56H), 5.62 (d, *J* = 14.4 Hz, 0.59H), 2.98 (s, 1.20H), 2.79 (s, 1.77H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 161.8, 161.5, 135.6, 135.1, 133.7, 133.2, 132.8, 132.3, 132.0, 131.7, 128.4, 127.9, 126.0, 125.9, 124.4, 124.2, 122.7, 122.4, 62.9, 56.1, 33.4, 29.1. HRMS (ESI-FTMS) *m/z* [M + H]<sup>+</sup> calcd. for C<sub>11</sub>H<sub>11</sub>Br<sub>2</sub>N<sub>4</sub>O: 372.9294, found 372.9288.

*N*-methyl-*N*-((4-phenyl-2*H*-1,2,3-triazol-2-yl)methyl)formamide (**4a**) Light yellow solid (64%, 89 mg) m.p. 58–60 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.50 (s, 0.74H), 8.16 (s, 0.26H), 7.90 (d, *J* = 3.6 Hz, 1H), 7.79 (d, *J* = 1.2 Hz, 1H), 7.43 (q, *J* = 7.2 Hz, 2H), 7.38 (d, *J* = 7.2 Hz, 1H), 5.92 (s, 0.57H), 5.75 (s, 1.4H), 3.02 (s, 0.76H), 2.94 (s, 2.25H). <sup>13</sup>C NMR (150

MHz, CDCl<sub>3</sub>) δ 162.9, 162.6, 159.8, 148.5, 131.5, 131.2, 127.0, 126.9, 122.0, 114.0, 66.7, 60.3, 54.9, 33.2, 29.1. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>11</sub>H<sub>12</sub>N<sub>4</sub>NaO: 239.0903; found, 239.0900.

*N*-methyl-*N*-((4-phenyl-2*H*-1,2,3-triazol-2-yl)methyl)acetamide (**4b**) Colorless oil (50%, 76 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.89 (s, 1H), 7.79 (t, *J* = 7.8 Hz, 2H), 7.43 (d, *J* = 8.4 Hz, 2H), 7.37 (s, 1H), 5.97 (s, 0.91H), 5.82 (s, 1.01H), 3.11 (s, 1.42H), 3.05 (s, 1.4H), 2.47 (s, 1.48H), 2.16 (s, 1.46H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 171.2, 159.9, 159.8, 148.4, 148.0, 131.2, 127.2, 122.5, 122.3, 114.1, 67.5, 63.4, 55.1, 35.1, 33.0, 21.5. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>14</sub>N<sub>4</sub>NaO: 253.1060; found, 253.1056.

*N*-methyl-*N*-((4-(*p*-tolyl)-2*H*-1,2,3-triazol-2-yl)methyl)acetamide (**4c**) Light yellow solid (59%, 86 mg) m.p. 84–86 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.85 (s, 1H), 7.74–7.58 (m, 2H), 7.30–7.15 (m, 2H), 5.96 (s, 0.89H), 5.82 (s, 0.98H), 3.11 (s, 1.26H), 3.05 (s, 1.31H), 2.47 (s, 1.4H), 2.39 (d, *J* = 3.2 Hz, 3H), 2.17 (s, 1.32H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 163.0, 135.4, 133.7, 130.9, 130.1, 129.9, 127.6, 121.9, 67.1, 60.8, 33.6, 29.6. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>NaO: 267.1216; found, 267.1212.

*N*-((4-(4-methoxyphenyl)-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (**4d**) Light yellow solid (47%, 67 mg) m.p. 81–83 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.81 (s, 1H), 7.71 (t, *J* = 7.8 Hz, 2H), 6.95 (t, *J* = 7.8 Hz, 2H), 5.94 (s, 0.85H), 5.79 (s, 1.13H), 3.82 (d, *J* = 4.2 Hz, 3H), 3.10 (s, 1.29H), 3.04 (s, 1.6H), 2.45 (s, 1.59H), 2.15 (s, 1.26H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 162.9, 162.6, 148.8, 138.5, 131.9, 129.3, 129.2, 126.6, 125.6, 66.8, 60.4, 33.3, 29.2, 21.0. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>NaO<sub>2</sub>: 283.1166;

found, 283.1161.

*N*-((4-(2-bromophenyl)-2*H*-1,2,3-triazol-2-yl)methyl)-*N*-methylacetamide (**4e**) Light yellow solid (57%, 78 mg) m.p. 92–94 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.15 (s, 1H), 7.78–7.64 (m, 2H), 7.40–7.36 (m, 1H), 7.28–7.20 (m, 7.4 Hz, 1H), 5.99 (s, 0.9H), 5.86 (s, 1H), 3.14 (s, 1.38H), 3.07 (s, 1.41H), 2.47 (s, 1.38H), 2.17 (s, 1.37H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 171.3, 148.7, 148.3, 131.9, 128.8, 128.6, 126.0, 67.8, 63.6, 35.2, 33.2, 21.7. HRMS (ESI-FTMS) m/z [M + H]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>14</sub>BrN<sub>4</sub>O: 309.0346; found, 309.0343.

Mixture of *N*<sup>l</sup> [*N*-methyl-*N*-((4-phenyl-1*H*-1,2,3-triazol-1-yl)methyl)formamide] and *N*<sup>l'</sup> [*N*-methyl-*N*-((5-phenyl-1*H*-1,2,3-triazol-1-yl)methyl)formamide] product (**5a**) (18%, 25 mg), ratio of *N*<sup>l</sup>:*N*<sup>l'</sup> = 1:3 (based on <sup>1</sup>H NMR).

*N*-methyl-*N*-((4-phenyl-1*H*-1,2,3-triazol-1-yl)methyl)acetamide (**5b**) Light yellow oil (25%, 38 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.94 (s, 1H), 7.71 (d, *J* = 7.2 Hz, 2H), 7.29 (t, *J* = 7.8 Hz, 2H), 7.21 (t, *J* = 7.8 Hz, 1H), 5.72 (s, 2H), 3.03 (s, 3H), 2.00 (s, 3H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 171.8, 147.9, 130.1, 128.6, 128.0, 125.8, 125.5, 125.4, 120.4, 59.9, 35.6, 21.4. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>14</sub>N<sub>4</sub>NaO: 253.1060; found, 253.1058.

*N*-methyl-*N*-((4-(*p*-tolyl)-1*H*-1,2,3-triazol-1-yl)methyl)acetamide (**5c**) Light yellow oil (21%, 31 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.00 (s, 1H), 7.71 (d, *J* = 7.2 Hz, 2H), 7.22 (d, *J* = 7.2 Hz, 2H), 5.84 (s, 2H), 3.16 (s, 3H), 2.36 (s, 3H), 2.13 (s, 3H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 172.0, 148.3, 138.0, 129.5, 129.4, 127.4, 125.9, 125.8, 125.5, 120.1, 60.0, 35.7, 21.5, 21.2. HRMS (ESI-FTMS) m/z [M + Na]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>NaO: 267.1216; found,

1  
2  
3  
4  
5  
6  
7  
8 267.1215.

9  
10 *N*-((4-(4-methoxyphenyl)-1*H*-1,2,3-triazol-1-yl)methyl)-*N*-methylacetamide (**5d**) Light  
11 yellow oil (27%, 38 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.95 (s, 1H), 7.75 (d, *J* = 8.4 Hz,  
12 2H), 6.94 (d, *J* = 8.4 Hz, 2H), 5.83 (s, 2H), 3.83 (s, 3H), 3.16 (s, 3H), 2.13 (s, 3H). <sup>13</sup>C  
13 NMR (150 MHz, CDCl<sub>3</sub>) δ 172.0, 159.6, 148.0, 127.2, 126.9, 122.9, 119.6, 114.2, 114.1,  
14 60.0, 55.2, 35.7, 21.5. HRMS (ESI-FTMS) *m/z* [M + Na]<sup>+</sup> calcd. for C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>NaO<sub>2</sub>:  
15 283.1166; found, 283.1164.  
16  
17  
18  
19

20  
21  
22  
23  
24  
25 *N*-((4-(2-bromophenyl)-1*H*-1,2,3-triazol-1-yl)methyl)-*N*-methylacetamide (**5e**) Light yellow  
26 oil (20%, 27 mg). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.46 (s, 1H), 8.10–8.04 (m, 1H), 7.64 (d, *J*  
27 = 7.8 Hz, 1H), 7.39 (d, *J* = 7.2 Hz, 1H), 7.19 (d, *J* = 0.6 Hz, 1H), 5.89 (s, 2H), 3.20 (s, 3H),  
28 2.14 (s, 3H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 171.9, 145.8, 133.5, 131.1, 130.5, 129.4, 127.5,  
29 123.9, 121.3, 60.1, 35.8, 21.6. HRMS (ESI-FTMS) *m/z* [M + H]<sup>+</sup> calcd. for C<sub>12</sub>H<sub>14</sub>BrN<sub>4</sub>O:  
30 309.0346; found, 309.0343.  
31  
32  
33  
34  
35  
36  
37  
38  
39

## 40 ASSOCIATED CONTENT

### 41 Supporting Information

42  
43  
44  
45 The Supporting Information is available free of charge on the ACS Publications website at  
46 DOI: 10.1021/x0xx00000x. Scanned copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra of the synthesized  
47 compounds (PDF) Crystallographic data for compound **3q** and **3q'** (CIF).  
48  
49  
50

## 51 AUTHOR INFORMATION

### 52 Corresponding Author

53  
54  
55  
56  
57  
58  
59  
60  
\*<sup>†</sup>E-mail: yfchen@wit.edu.cn

\*†E-mail: hhxyliyuanjun@163.com

ORCID

Yunfeng Chen: 0000-0002-6220-5015

Yuanxiang Li: 0000-0001-8917-9294

## ACKNOWLEDGMENTS

We are grateful to the National Natural Science Foundation of China (21002076, and 21441007), Graduate Education Innovation Fund of Wuhan Institute of Technology (CX2016162, CX2016167). Li, Y. thanks the Project of Natural Science Foundation of Hunan Province (2016JJ4072), and the Scientific Research Fund of Hunan Provincial Education Department (16A165).

## REFERENCES

- (a) Li, C. J. *Acc. Chem. Res.*, **2009**, *42*, 335; (b) Scheuermann, C. J. *Chem. Asian J.* **2010**, *5*, 436; (c) Girard, S. A.; Knauber, T.; Li, C. J. *Angew. Chem. Int. Ed.* **2014**, *53*, 74.
- (a) Lao, Z. Q.; Zhong, W. H.; Lou, Q. H.; Li, Z. J.; Meng, X. B. *Org. Biomol. Chem.* **2012**, *10*, 7869; (b) Xia, Q.; Chen, W. J. *J. Org. Chem.* **2012**, *77*, 9366; (c) Nobuta, T.; Tada, N.; Fujiya, A.; Kariya, A.; Miura, T.; Itoh, A. *Org. Lett.* **2013**, *15*, 574; (d) Dhineshkumar, J.; Lamani, M.; Alagiri, K.; Prabhu, K. R. *Org. Lett.* **2013**, *15*, 1092. (e) Aruri, H.; Singh, U.; Sharma, S.; Gudup, S.; Bhogal, M.; Kumar, S.; Singh, D.; Gupta, V. K.; Kant, R.; Vishwakarma, R. A.; Singh, P. P. *J. Org. Chem.* **2015**, *80*, 1929; (f) Habtamu, A.; Siddaiah, V.; Venkateswara, R. B. *RSC Adv.* **2016**, *6*, 82289;

- 1  
2  
3  
4  
5  
6  
7 (g) Singh, M. K.; Akula, H. K.; Satishkumar, S.; Stahl, L.; Lakshman, M. K. *ACS*  
8 *catalysis*. **2016**, *6*, 1921. (h) Suresh, R.; Ganesh, M.; Sourav, K. S.; Bhisma, K. P.  
9  
10 *Org. Lett.* **2015**, *17*, 5586; (i) Cai, H.; Guo, S.; Zhu, Z.; Wang, Y.; Yang, M.; Huang,  
11  
12 L.; Gong, J. *Synlett*. **2016**, *27*, 2705.  
13  
14  
15  
16  
17 3 (a) Ding, S. T.; Jiao, N. *Angew. Chem. Int. Ed.* **2012**, *51*, 1; (b) Jiang, H.; Lin, A.;  
18  
19 Zhu, C.; Cheng, Y. *Chem. Commun.* **2013**, *49*, 819; (c) Ali, W.; Rout, S. K.; Guin, S.;  
20  
21 Modi, A.; Banerjee, A.; Patel, B. K. *Adv. Synth. Catal.* **2015**, *357*, 515.  
22  
23  
24  
25 4 (a) Boonen, J.; Bronselaer, A.; Nielandt, J.; Veryser, L.; Tré, G. D.; Spiegeleer, B. D.  
26  
27 *J. Ethnopharmacol.* **2012**, *142*, 563; (b) Roughley, S. D.; Jordan, A. M. *J. Med.*  
28  
29 *Chem.* **2011**, *54*, 3451.  
30  
31  
32  
33 5 (a) Bakunov, S. A.; Bakunova, S. M.; Wenzler, T.; Ghebru, M.; Werbovets, K. A.;  
34  
35 Brun, R.; Tidwell, R. R. *J. Med. Chem.* **2010**, *53*, 254; (b) Agalave, S. G.; Maujan, S.  
36  
37 R.; Pore, V. S. *Chem. Asian. J.* **2011**, *6*, 2696; (c) Hsieh, H. Y.; Lee, W. C.; Senadi, G.  
38  
39 C.; Hu, W. P.; Liang, J. J.; Tsai, T. R.; Chou, Y. W.; Kuo, K. K.; Chen, C. Y.; Wang, J.  
40  
41 *J. J. Med. Chem.* **2013**, *56*, 5422; (d) Mohammed, I.; Kummetha, I. R.; Singh, G.;  
42  
43 Sharova, N.; Lichinchi, G.; Dang, J.; Stevenson, M.; Rana, T. M. *J. Med. Chem.* **2016**,  
44  
45 *59*, 7677; (e) Massarotti, A.; Aprile, S.; Mercalli, V.; Del Grosso, E.; Grosa, G.;  
46  
47 Sorba, G.; Tron, G. C. *Chem. Med. Chem.* **2014**, *9*, 2497; (f) Bai, H.; Zhu, P.; Wu, W.;  
48  
49 Li, J.; Ma, Z.; Zhang, W.; Cheng, Y.; Du, L.; Li, M. *Med. Chem. Commun.* **2015**, *6*,  
50  
51 418.  
52  
53  
54  
55  
56  
57  
58  
59  
60

- 1  
2  
3  
4  
5  
6  
7  
8 (a) Juricek, M.; Kouwer, P. H.; Rowan, A. E. *Chem. Commun.* **2011**, 47, 8740; (b)  
9  
10 Yan, F.; Lartey, M.; Jariwala, K.; Bowser, S.; Damodaran, K.; Albenze, E.; Luebke,  
11  
12 D. R.; Nulwala, H. B.; Smit, B.; Haranczyk, M. *J. Phys. Chem. B.* **2014**, 118, 13609;  
13  
14 (c) Kantheti, S.; Narayan, R.; Raju, K. V. S. N. *RSC Adv.* **2015**, 5, 3687; (d) Liang, P.;  
15  
16 Li, Z.; Mi, Y.; Yang, Z.; Wang, D.; Cao, H.; He, W.; Yang, H. *J. Electron. Mater.*  
17  
18 **2015**, 44, 2883; (e) Sood, R.; Donnadio, A.; Giancola, S.; Kreis, A.; Jones, D. J.;  
19  
20 Cavaliere, S. *ACS Appl. Mat. Interfaces.* **2016**, 8, 16897; (f) Reed, D. A.; Xiao, D. J.;  
21  
22 Gonzalez, M. I.; Darago, L. E.; Herm, Z. R.; Grandjean, F.; Long, J. R. *J. Am. Chem.*  
23  
24 *Soc.* **2016**, 138, 5594.
- 25  
26  
27  
28  
29  
30 7 Selected examples: (a) Chattopadhyay, B.; Gevorgyan, V. *Angew. Chem. Int. Ed.*  
31  
32 **2012**, 51, 862; (b) Johnson, T. C.; Totty, W. G.; Wills, M. *Org. Lett.* **2012**, 14, 5230;  
33  
34 (c) Selander, N.; Worrell, B. T.; Chuprakov, S.; Velaparthi, S.; Fokin, V. V. *J. Am.*  
35  
36 *Chem. Soc.* **2012**, 134, 14670; (d) Liu, S.; Sawicki, J.; Driver, T. G. *Org. Lett.* **2012**,  
37  
38 14, 3744; (e) Spangler, J. E.; Davies, H. M. L. *J. Am. Chem. Soc.* **2013**, 135, 6802; (f)  
39  
40 Xing, Y.; Sheng, G.; Wang, J.; Lu, P.; Wang, Y. *Org. Lett.* **2014**, 16, 1244; (g) Su, Y.;  
41  
42 Petersen, J. L.; Gregg, T. L.; Shi, X. *Org. Lett.* **2015**, 17, 1208.
- 43  
44  
45  
46  
47  
48 8 (a) Moses, J. E.; Moorhouse, A. D. *Chem. Soc. Rev.* **2007**, 36, 1249; (b) Heal, W. P.;  
49  
50 Wickramasinghe, S. R.; Leatherbarrow, R. J.; Tate, E. W. *Org. Biomol. Chem.* **2008**,  
51  
52 6, 2308; (c) Schneider, G. *Nat. Rev. Drug. Discovery.* **2010**, 9, 273.
- 53  
54  
55  
56 9 Selected examples: (a) Törnøe, C. W.; Christensen, C.; Meldal, M. *J. Org. Chem.*  
57  
58 **2002**, 67, 3057; (b) Bock, V. D.; Hiemstra, H.; van Maarseveen, J. H. *Euro. J. Org.*  
59  
60

1  
2  
3  
4  
5  
6  
7  
8 *Chem.* **2006**, *2006*, 51; (c) Hein, J. E.; Fokin, V. V. *Chem. Soc. Rev.* **2010**, *39*, 1302;  
9  
10 (d) Gonda, Z.; Novak, Z. *Dalton. Trans.* **2010**, *39*, 726; (e) Haldon, E.; Nicasio, M.  
11  
12 C.; Perez, P. J. *Org. Biomol. Chem.* **2015**, *13*, 9528; (f) Wang, W.; Peng, X.; Wei, F.;  
13  
14 Tung, C.; Xu, Z. *Angew. Chem. Int. Ed.* **2016**, *55*, 649.

15  
16  
17 10 Selected examples: (a) Ramachary, D. B.; Ramakumar, K.; Narayana, V. V. *Chem.*  
18  
19 *Eur. J.* **2008**, *14*, 9143; (b) Belkheira, M.; Abed, D. El; Pons, J. M.; Bressy, C. *Chem.*  
20  
21 *Eur. J.* **2011**, *17*, 12917; (c) Wang L.; Peng S.; Danence, L. J.; Gao, Y.; Wang, J.  
22  
23 *Chem. Eur. J.* **2012**, *18*, 6088; (d) Danence, L. J.; Gao, Y.; Li, M.; Huang, Y.; Wang, J.  
24  
25 *Chem. Eur. J.* **2011**, *17*, 3584; (e) Ramachary, D. B.; Shashank, A. B. *Chem. Eur. J.*  
26  
27 **2013**, *19*, 13175; (f) Ramasastry, S. S. *Angew. Chem. Int. Ed.* **2014**, *53*, 14310; (g) Li,  
28  
29 W.; Du, Z.; Huang, J.; Jia, Q.; Zhang, K.; Wang, J. *Green Chem.* **2014**, *16*, 3003;  
30  
31 (h) Thomas, J.; Jana, S.; Liekens, S.; Dehaen, W. *Chem. Commun.* **2016**, *52*, 9236.

32  
33  
34  
35  
36  
37 11 (a) Amantini, D.; Fringuelli, F.; Piermatti, O.; Pizzo, F.; Zunino, E.; Vaccaro, L. J.  
38  
39 *Org. Chem.* **2005**, *70*, 6526; (b) Janreddy, D.; Kavala, V.; Kuo, C. W.; Chen, W. C.;  
40  
41 Ramesh, C.; Kotipalli, T.; Kuo, T. S.; Chen, M. L.; He, C. H.; Yao, C. F.; *Adv. Synth.*  
42  
43 *Catal.* **2013**, *355*, 2918; (c) Li, W.; Wang, J. *Angew. Chem. Int. Ed.* **2014**, *53*, 14186;  
44  
45  
46 (d) Li, W.; Du, Z.; Zhang, K.; Wang, J. *Green Chem.*, **2015**, *17*, 781; (e) Chen, Y.;  
47  
48 Nie, G.; Zhang, Q.; Ma, S.; Li, H.; Hu, Q. *Org. Lett.* **2015**, *17*, 1118; (f) Rohilla, S.;  
49  
50 Patel, S. S.; Jain, N. *Eur. J. Org. Chem.* **2016**, 847.

51  
52  
53  
54  
55 12 For selected examples, see: (a) Van Berkel, S. S.; Brauch, S.; Gabriel, L.; Henze, M.;  
56  
57 Stark, S.; Vasilev, D.; Wessjohann, L. A.; Abbas, M.; Westermann, B. *Angew. Chem.*  
58  
59  
60

- 1  
2  
3  
4  
5  
6  
7  
8 *Int. Ed.* **2012**, *51*, 5343; (b) Chen, Z.; Yan, Q.; Liu, Z.; Xu, Y.; Zhang, Y. *Angew.*  
9  
10 *Chem. Int. Ed.* **2013**, *52*, 13324; (c) Cheng, G.; Zeng, X.; Shen, J.; Wang, X.; Cui, X.  
11  
12 *Angew. Chem. Int. Ed.* **2013**, *52*, 13265; (d) Li, W.; Jia, Q.; Du, Z.; Wang, J. *Chem.*  
13  
14 *Commun.* **2013**, *49*, 10187; (e) Wang, J.-P.; Cao, S.; Liu, Y. *J. Org. Chem.* **2015**, *80*,  
15  
16 9028; (f) Bai, H.-W.; Cai, Z.-J.; Wang, S.-Y.; Ji, S.-J. *Org. Lett.* **2015**, *17*, 2898; (g)  
17  
18 Thomas, J.; Goyvaerts, V.; Liekens, S.; Dehaen, W. *Chem. Eur. J.* **2016**, *22*, 9966.  
19  
20  
21  
22  
23 13 (a) Chen, Y.; Liu, Y.; Petersen, J. L.; Shi, X. *Chem. Commun.* **2008**, 3254; (b) Liu, Y.;  
24  
25 Yan, W.; Chen, Y.; Petersen, J. L.; Shi, X. *Org. Lett.* **2008**, *10*, 5389; (c) Wang, X. J.;  
26  
27 Zhang, L.; Lee, H.; Haddad, N.; Krishnamurthy, D.; Senanayake, C. H. *Org. Lett.*  
28  
29 **2009**, *11*, 5026; (d) Duan, H.; Yan, W.; Sengupta, S.; Shi, X. *Bioorg. Med. Chem.*  
30  
31 *Lett.* **2009**, *19*, 3899; (e) Wang, X. J.; Zhang L.; Krishnamurthy, D.; Senanayake, C.  
32  
33 H.; Wipf, P. *Org. Lett.* **2010**, *12*, 4632; (f) Ueda, S.; Su, M.; Buchwald, S. L. *Angew.*  
34  
35 *Chem. Int. Ed.* **2011**, *50*, 8944; (g) Zhang, Y.; Li, X.; Li, J.; Chen, J.; Meng, X.; Zhao,  
36  
37 M.; Chen, B. *Org. Lett.* **2012**, *14*, 26; (h) Yan, W.; Wang, Q.; Chen, Y.; Petersen, J. L.;  
38  
39 Shi, X. *Org. Lett.* **2010**, *12*, 3308; (i) Zhu, L. L.; Xu, X. Q.; Shi, J. W.; Chen, B. L.;  
40  
41 Chen, Z. *J. Org. Chem.* **2016**, *81*, 3568.  
42  
43  
44  
45  
46  
47  
48 14 (a) Hu, Q.; Liu, Y.; Deng, X.; Li, Y.; Chen, Y. *Adv. Synth. Catal.* **2016**, *358*, 1689; (b)  
49  
50 Chen, Y.; Zhou, S.; Ma, S.; Liu, W.; Pan, Z.; Shi, X. *Org. Biomol. Chem.* **2013**, *11*,  
51  
52 8171; (c) Fan, M.; Liu, Y.; Hu, Q.; Jia, L.; Chen, Y. *Eur. J. Org. Chem.* **2016**, 5470.  
53  
54  
55  
56 15 From the crude <sup>1</sup>H NMR, *N*<sup>1</sup> and *N*<sup>1</sup>-coupling products are less than 5% in all, which  
57  
58 cannot be separated and identified.  
59  
60

1  
2  
3  
4  
5  
6  
7 16 During the revision of this manuscript, Singh and coworkers reported a metal-free  
8  
9 cross-dehydrogenative coupling reaction between *NH*-azoles and  $\alpha$ -C(sp<sup>3</sup>)-H  
10  
11 containing amides, where *N*-amidoalkylation of *NH*-1,2,3-triazoles has been carried  
12  
13 out under TBAI/TBHP oxidative conditions: Aruri, H.; Singh, U.; Kumar, M.;  
14  
15 Sharma, S.; Aithagani, S. K.; Gupta, V. K.; Mignani, S.; Vishwakarma, R. A.; Singh,  
16  
17 P. P. *J. Org. Chem.* **2017**, *82*, 1000.  
18  
19

20  
21  
22 17 Further information can be found in the CIF files. These crystal structures were  
23  
24 deposited in the Cambridge Crystallographic Data Centre and assigned as CCDC  
25  
26 (No: 1510061 and 1510062).  
27  
28

29  
30 18 The separations of *N*<sup>1</sup> products sometimes include some *N*<sup>1'</sup> impurities (*see*  
31  
32 *supporting information*).  
33  
34

35  
36 19 To a solution of 4-bromo-5-(2-bromophenyl)-*IH*-1,2,3-triazole (1 g, 3.3 mmol) in  
37  
38 DMF (20 mL), was added Cu(OAc)<sub>2</sub> (132 mg, 0.66 mmol), K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (1.78 g, 6.6  
39  
40 mmol) and Na<sub>2</sub>CO<sub>3</sub> (0.38 g, 3.6 mmol), the mixture was stirred at 110 °C for 12 h.  
41  
42 the isomer **3q'** was separated by silica gel chromatography from the mixture  
43  
44 products in 5% yield.  
45  
46  
47

48 20 (a) Katritzky, A. R.; Perumal, S.; Savage, G. P. *J. Chem. Soc. Perkin Trans. 2* **1990**,  
49  
50 921; (b) Kalisiak, J.; Sharpless, K. B.; Fokin, V. V. *Org. Lett.* **2008**, *10*, 3171.  
51  
52

53 21 Chen, Y.; Wu, J.; Ma, S.; Zhou, S.; Meng, X.; Jia, L.; Pan, Z.; *J. Mol. Struct.* **2015**,  
54  
55 1809, 1.  
56  
57  
58  
59  
60

- 1  
2  
3  
4  
5  
6  
7 22 (a) Lou, S. J.; Xu, D. Q.; Shen, D. F.; Wang, Y. F.; Liu, Y. K.; Xu, Z. Y. *Chem.*  
8 *Commun.* **2012**, *48*, 11993; (b) Li, Y.; Xue, D.; Lu, W.; Wang, C.; Liu, Z. T.; Xiao, J.  
9  
10 *Org. Lett.* **2014**, *16*, 66; (c) Zhao, M. N.; Hui, R. R.; Ren, Z. H.; Wang, Y. Y.; Guan,  
11  
12 Z. H. *Org. Lett.* **2014**, *16*, 3082; (d) Modak, A.; Dutta, U.; Kancherla, R.; Maity, S.;  
13  
14  
15 Bhadra, M.; Mobin, S. M.; Maiti, D. *Org. Lett.* **2014**, *16*, 2602; (e) Wu, X.; Zhao, Y.;  
16  
17 Ge, H. *J. Am. Chem. Soc.* **2015**, *137*, 4924; (f) Xiao, J.; Li, Q.; Chen, T.; Han, L. B.  
18  
19 *Tetrahedron Lett.* **2015**, *56*, 5937; (g) Zhang, L.; Lu, P.; Wang, Y. *Org. Biomol. Chem.*  
20  
21 **2015**, *13*, 8322.  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60