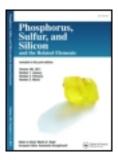
This article was downloaded by: [ECU Libraries] On: 01 June 2014, At: 13:48 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gpss20

# SOME REACTIONS WITH p-ETHOXYPHENYLCYANO-THIOFORAMIDE: SYNTHESIS OF PYRROLE, PYRROLO[2,3c]PYRROLE, IMIDAZO[4,5b]QUINOXALINES AND HYDANTOIN DERIVATIVES

Y. A. Ammar<sup>a</sup>, A. M.SH. El-Sharief<sup>a</sup>, M. M. Ali<sup>a</sup>, Y. A. Mohamed<sup>a</sup> & SH. I. Mohamed<sup>a</sup> <sup>a</sup> Chemistry Department, Faculty of Science, Al-

Azhar University, Nasr City, 11884, Cairo, Egypt Published online: 19 Dec 2006.

To cite this article: Y. A. Ammar , A. M.SH. EI-Sharief , M. M. Ali , Y. A. Mohamed & SH. I. Mohamed (2000) SOME REACTIONS WITH p-ETHOXYPHENYLCYANO-THIOFORAMIDE: SYNTHESIS OF PYRROLE, PYRROLO[2,3-c]PYRROLE, IMIDAZO[4,5b]QUINOXALINES AND HYDANTOIN DERIVATIVES, Phosphorus, Sulfur, and Silicon and the Related Elements, 166:1, 173-185, DOI: <u>10.1080/10426500008076539</u>

To link to this article: http://dx.doi.org/10.1080/10426500008076539

### PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <a href="http://www.tandfonline.com/page/terms-and-conditions">http://www.tandfonline.com/page/terms-and-conditions</a>

Phosphorus, Sulfur and Silicon, 2000, Vol. 166, pp. 173-185 Reprints available directly from the publisher Photocopying permitted by license only © 2000 OPA (Overseas Publishers Association) Amsterdam N.V. Published under license by the Gordon and Breach Science Publishers imprint. Printed in Malaysia

## SOME REACTIONS WITH p-ETHOXYPHENYLCYANO-THIOFORAMIDE: SYNTHESIS OF PYRROLE, PYRROLO[2,3-c]PYRROLE, IMIDAZO[4,5-b]QUINOXALINES AND HYDANTOIN DERIVATIVES

#### Y.A. AMMAR<sup>\*</sup>, A.M.SH. EL-SHARIEF, M.M. ALI, Y.A. MOHAMED and SH.I. MOHAMED

Chemistry Department, Faculty of Science, Al-Azhar University, Nasr City 11884 Cairo, Egypt

(Received April 22, 2000; In final form June 23, 2000)

P-Ethoxyphenylcyanothioformamide (1) was reacted with  $\alpha$ , $\beta$ -unsaturated ketone (2) and N-p-chlorophenylmaleimide (4) to furnish pyrrole and pyrrolo[2,3-c]pyrrol-4,6-diones (3) and (5) respectively. Also, interaction of 1 with anthranilic acid and o-phenylenediamine produced 3-(4'-ethoxyphenyl)-2-thioxoquinazolin-4-one (6) and 2-thioxobenzimidazoles (7). When, 1 was reacted with iso(thio)cyanates caused cyclization to afford 5-imino-4-thioxoimidazolidinones (9). Compound 9 was subjected to some reagents such as hydrazine hydrate, thiosemicarbazide, o-phenylenediamines, hydrogen sulfide and HCl to give 5-hydrazono, 4-thiosemicarbazono, and thiohydantoin derivatives (10–17), respectively.

Keywords: pyrrole; Hydantoin derivatives; hydrogen sulfide

#### INTRODUCTION

The synthesis of cyanothioformamides has attracted considerable attention since these compounds are used as versatile starting materials for the synthesis of a wide variety of fused heterocyclic compounds<sup>1-4</sup>. We have been involved in a program aiming to explore the utility of cyanothioformamides for synthesis of many heterocyclic derivatives<sup>5-8</sup>. As a part of

<sup>\*</sup> Corresponding Author.

this work we report here the results of our investigation on the reactivity of p-ethoxycyanothioformamide  $(I)^{9,10}$ , towards a variety of reagents. The work has resulted in the development of several new approaches for the synthesis of pyrrole, pyrrolo[2,3-c]pyrrole, quinazolinone, imidazolidine and imidazoquinoxaline derivatives. The reactivity of 1 towards some activated double bonds was also discussed.

#### **RESULTS AND DISCUSSION**

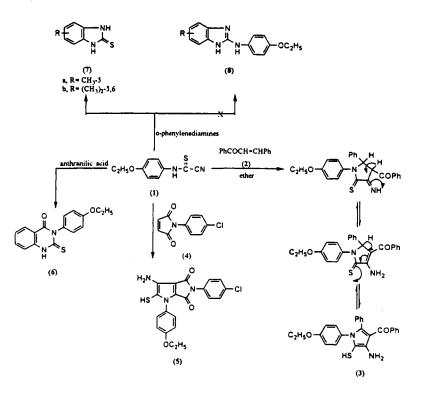
Thus, interaction of 1 with unsaturated ketone (Chalcone) (2) at room temperature in the presence of triethylamine effected cyclization to give 4-benzoyl-1-(4'-ethoxyphenyl)-5-phenyl-3-iminopyrrolidine-2-thione, which has the tautomeric structures of pyrroline and pyrrole (3) (Scheme 1).

Its <sup>1</sup>H NMR spectrum showed signals at  $1.2-1.6(3H,t,CH_3-ethoxy)$ , 4.1-4.3 (2H, q, CH<sub>2</sub>-ethoxy) and its mass spectrum showed a molecular ion peak at m / z 414 (M, 24%), the fragmentation pattern was showed in (scheme 2).

Also, interaction of 1 with N-p-chlorophenylmaleimide (4) as other electrophile in the presence of triethylamine, the product 3-amino-5-(4'-chlorophenyl)pyrrolo[2,3-c]pyrrole-4,6-dione (5) was precipitated after the reaction was stirred at room temperature for 30 min. The suggested product was elucidated by IR spectrum which showed bands at 3319 (broad; NH), 2990(CH-aliphatic) and 1712 (C=O) and its mass spectrum revealed a molecular ion peak at m / z 416 (M; 2.4%) and 281[M-135 (p-H<sub>5</sub>C<sub>2</sub>OC<sub>6</sub>H<sub>4</sub>-N); 6.7%]

Furthermore, condensation of 1 with anthranilic acid furnished a product which gave analytical data fitted with 3-(4<sup>-</sup>-ethoxyphenyl)-2-thioxoquinazolin-4-one (6) via elimination of HCN and H<sub>2</sub>O. Its <sup>1</sup>HNMR spectrum exhibited signals at 1.4–1.7 (3H, q, CH<sub>3</sub>-ethopxy), 4.1–4.3 (2H, q, CH<sub>2</sub>-ethoxy) and mass spectrum showed a molecular ion peak at m / z 298 (M; 100%), 270 [M-28(C=O); 30.2%].

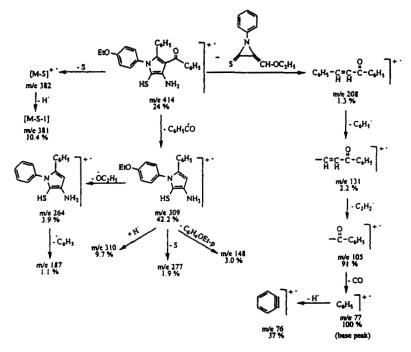
It was reported that cyanothioformamides reacted with o-phenylenediamines to yield 2-substituted anilinobenzimidazole derivatives (through elimination of HCN and  $H_2O$ ).



SCHEME 1

In this paper, condensation of 1 with 4-methyl or 4,5-dimethyl-o-phenylenediamines gave products which were expected to have one of the two structure (7 or 8). The obtained products were found to be sulfur positive and <sup>1</sup>HNMR showed the absence of ethoxy group which favored structure 7 (scheme 3). <sup>1</sup>HNMR spectrum of 7b in DMSO-d<sub>6</sub> showed 2.3(6H,s,2CH<sub>3</sub>), also mass spectrum of 7a exhibited a molecular ion peak at m/z 164(M, 100%) and 132 [M-32(S); 14.3%]. Compounds 7a,b were also confirmed through their synthesis via the reaction of o-phenylenediamines with CS<sub>2</sub>''.

The reactivity of p-ethoxyphenylcyanothioformamide 1 was extended to cover its behavior towards iso(thio)cyanate for obtaining other type of heterocyclic compounds. Thus, interaction of (1) with aryl iso(thio)cyanates in the presence of triethylamine caused cyclization to give 5-amino-4-thi-

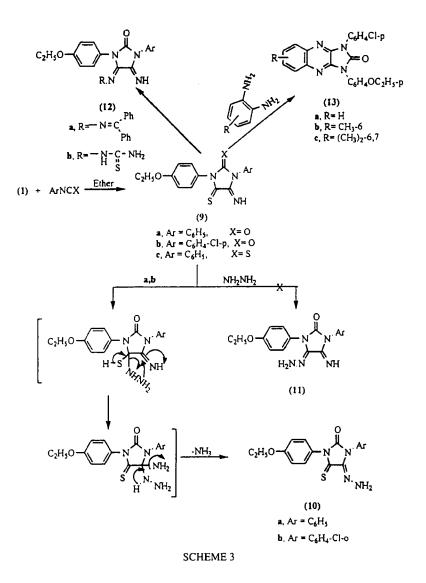


SCHEME 2 Fragmentation pattern of compound (3)

oxoimidazolidin-2-one or 2-thione derivatives (**9a-c**). <sup>1</sup>HNMR spectrum of **9b** in DMSO-d<sub>6</sub> revealed signals at 1.3-1.7(3H,t,CH<sub>3</sub>-ethoxy), 4.2-4.5(2H,q,CH<sub>2</sub>-ethoxy) and 8.8(1H,br,NH).

The imidazolidine derivative (9) contain adjacent imino and thione functions in the 5-and 4- positions which appear promising for further chemical transformations. Thus equimolecular amounts of 9b and hydrazine hydrate gave the monohydrazono derivative for which structure 10 or 11 seemed possible. The positive element test for sulfur and spectral data favored the 5-hydrazono derivative 11. The formation of 11 may be rationalized as in scheme 4. Mass spectrum of 11b showed a molecular ion peak at 375 (M, 100%), 377(M+2,37% due to the chlorine isotope),  $357[M-16; (NH_2), 25.4\%], 347[M-28(CO), 43.7\%]$  and  $196[M-179 {H_5C_2OC_6H_4NCS}, 20\%]$ .

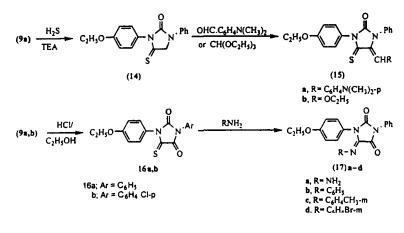
On the other hand, condensation of 9 with benzalhydrazone or thiosemicarbazide furnished the corresponding 4-azine or thiosemicarbazono



derivatives (12a,b). Mass spectrum of 12a showed a molecular ion peak at m/z 492 (M, 100%).

Furthermore, interaction of 9 with o-phenylenediamines, the obtained product gave analytical and spectral data compatible with imiazo[4,5-b]quinoxalines (13a,b) via elimination of  $H_2S$  and  $NH_3$ . <sup>1</sup>HNMR spectrum of **13b** in CDCl<sub>3</sub> showed signals at 1.2–1.5 (3H, t, CH<sub>3</sub>-ethoxy), 2.6 (6H, s, 2CH<sub>3</sub>), 4,1–4.3 (2H, q,.CH<sub>2</sub>-ethoxy).

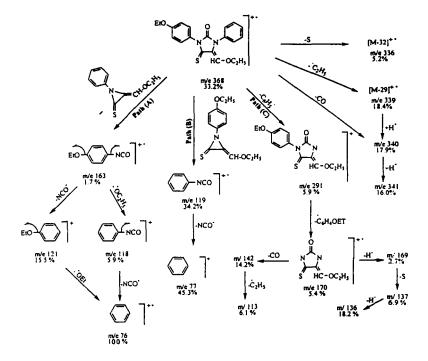
When the iminothione 9a was subjected to a stream of hydrogen sulfide in presence of triethylamine produced 1-(p-ethoxyphenyl)-3-phenylhydantoin (14). Its <sup>1</sup>HNMR spectrum in DMSO-d<sub>6</sub> showed signals at 1.3–1.6 (3H, t, CH<sub>3</sub>-ethoxy), 4.2–4.5 (2H, q, CH<sub>2</sub>-ethoxy) and 5.2 (2H, s, CH<sub>2</sub>-methylene).



SCHEME 4

The reactivity of the active methylene in compound 14 was proved through its reaction with p-N,N-dimethylaminobenzaldehyde and triethylorthoformate to give the benzylidene and ethoxymethylene derivatives (15a,b). <sup>1</sup>HNMR spectrum of 15a in DMSO-d<sub>6</sub> exhibited signals at 1.3– 1.6 (3H,t,CH<sub>3</sub>-ethoxy), 3.2 [6H,s, N(CH<sub>3</sub>)<sub>2</sub>], 4.1–4.3 (2H, q, CH<sub>2</sub>-ethoxy) and 5.4 (1H, s, CH=). Also, mass spectrum of 15b showed a molecular ion peak at m/z 368 (M,33.2%), the fragmentation pattern was showed in scheme 5.

Hydrolysis of 5-imino-4-thioxoimidazolidin-2-ones 9a,b by using ethanolic hydrochloric acid produced the corresponding 4-thioxoimidazolidin-2,5-diones (16a,b). Condensation of 16a with hydrazine hydrate or aryl amines took place through the thioxo group via elimination of  $H_2S$ and the corresponding 4-hydrazono (17a) and 4-substitutedimino derivatives (17b-d)were obtained.



SCHEME 5 Fragmentation pattern of compound (15b)

#### Experimental

All mps are uncorrected. IR spectra were measured as KBr pellets on a Shimadzu IR 200 spectrophotometer. <sup>1</sup>H NMR spectra were recorded in deutrated chloroform at 200 MHz on a Varian Gemini NMR spectrometer using tetramethylsilane as an internal reference. Mass spectra were performed on a Shimadzu GCMS-QP 1000 EX mass spectrometer at 70 eV. Elemental analyses were carried out at the Microanalytical Center of Cairo University.

#### 4-Benzoyl-1-(4-ethoxyphenyl)-5-phenyl-3-iminopyrrolidine-2-thione (3)

To a solution of (1, 0.01 mol) in ether (20 ml), the chalcone (2, 0.01 mol) and triethylamine (0.5 ml) were added. The reaction mixture was stirred at

room temperature for 3 hr and the solid that obtained after filtration was crystallized to furnish (3; Table I). Its IR spectrum exhibited 3240, 3307 (NH<sub>2</sub>), 2980 (CH-aliphatic), 1640 (C=O) which is lower than the expected value due to the intramolecular hydrogen bond. <sup>1</sup>HNMR spectrum in DMSO-d<sub>6</sub> showed signals 1.2–1.6 (3H, t, CH<sub>3</sub>-ethoxy), 4.1–4.3(2H,q,CH<sub>2</sub>-ethoxy) and 6.8–8.0(16H,m,Ar-H+NH<sub>2</sub>).

			•	•		
M.P.ª	Yield (%)	Mol. Formula (Mol. wt)	Elemental analyses, % Required/ Found			
Compd. M.P. <sup>a</sup> No. [°C]			С	Н	N	S
155–7 <sup>b</sup>	65	C <sub>25</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub> S	72.44	5.35	6.76	7.73
		(414.52)	72.30	5.20	6.50	7.60
5 135–6 <sup>b</sup>	64	C <sub>20</sub> H <sub>16</sub> N <sub>3</sub> O <sub>3</sub> SCI	58.04	3.89	10.15	7.75
		(413.87)	58.10	3.90	10.10	7.80
6 310–12 <sup>b</sup>	68	C <sub>16</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S	64.41	4.73	9.39	10.75
		(298.35)	64.40	4.70	9.40	10.70
7a 275–7 <sup>b</sup>	70	C <sub>8</sub> H <sub>8</sub> N <sub>2</sub> S <sub>3</sub>	58.51	4.91	17.06	19.52
		(164.22)	58.50	4.90	17.00	19.50
7b 3002 <sup>b</sup>	70	$C_9H_{10}N_2S$	60.65	5.65 <sup>-</sup>	15.71	17.99
		(178.25)	60.60	5.70	15.80	17.00
9a 125-7 <sup>b</sup>	68	C <sub>17</sub> H <sub>15</sub> N <sub>3</sub> O <sub>2</sub> S	62.75	4.65	12.91	9.85
		(325.38)	62.70	4.70	12.90	9.80
9b 167-8 <sup>b</sup>	71	C <sub>17</sub> H <sub>14</sub> ClN <sub>3</sub> O <sub>2</sub> S	65.75	3.92	11.68	8.91
		(359.82)	65.70	3.90	11.60	8.90
9c 135-6 <sup>b</sup>	69	C <sub>17</sub> H <sub>15</sub> N <sub>3</sub> OS <sub>2</sub>	59.80	4.43	12.31	18.78
		(341.45)	59.90	4.40	12.40	18.70
10a 210 <sup>b</sup>	70	C <sub>17</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> S	59.99	4.74	16.46	9.42
		(340.40)	59.90	4.80	16.40	9.50
10b 195 <sup>b</sup>	69	C <sub>17</sub> H <sub>15</sub> ClN <sub>4</sub> O <sub>2</sub> S	54.47	4.03	14.95	8.55
		(374.84)	54.40	4.10	14.90	8.70
12a 218 <sup>b</sup>	65	$C_{30}H_{24}CIN_5O_2$	69.03	4.63	13.42	
		(522.01)	69.00	4.80	13.40	
190 <sup>b</sup>	69	C <sub>18</sub> H <sub>18</sub> CIN <sub>6</sub> O <sub>2</sub> S	51.86	4.11	20.16	
		(416.88)	51.90	4.10	20.30	
220 <sup>b</sup>	64	C <sub>23</sub> H <sub>14</sub> N <sub>4</sub> O <sub>2</sub> Cl	66.27	4.11	13.44	
	[°C] 155–7 <sup>b</sup> 135–6 <sup>b</sup> 310–12 <sup>b</sup> 275–7 <sup>b</sup> 300–2 <sup>b</sup> 125–7 <sup>b</sup> 167–8 <sup>b</sup> 135–6 <sup>b</sup> 210 <sup>b</sup> 195 <sup>b</sup> 218 <sup>b</sup> 190 <sup>b</sup>	$[^{\circ}C]$ $(\%)$ $155-7^{b}$ $65$ $135-6^{b}$ $64$ $310-12^{b}$ $68$ $275-7^{b}$ $70$ $300-2^{b}$ $70$ $300-2^{b}$ $70$ $125-7^{b}$ $68$ $167-8^{b}$ $71$ $135-6^{b}$ $69$ $210^{b}$ $70$ $195^{b}$ $69$ $218^{b}$ $65$ $190^{b}$ $69$		$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

TABLE I Characteristics of the synthesized compounds

Compd. No.	M.P.ª	Yield (%)	Mol. Formula (Mol. wt)	Elemental analyses, % Required/ Found			
	(°C)			С	Н	N	s
			(416.87)	66.30	4.20	13.60	
13b 200 <sup>b</sup>	200 <sup>b</sup>	68	$C_{24}H_{19}N_4O_2CI$	66.90	4.44	13.00	
		(430.89)	66.80	4.60	13.20		
13c 270 <sup>b</sup>	270 <sup>b</sup>	70	$C_{25}H_{21}N_4O_2C1$	67.49	4.76	12.59	
			(444.92)	67.40	4.90	12.80	
14 175 <sup>c</sup>	175 <sup>c</sup>	67	$C_{17}H_{16}N_2O_2S$	65.37	5.16	8.97	10.26
		(312.38)	65.50	5.00	8.80	10.10	
15a 155 <sup>b</sup>	155 <sup>b</sup>	63	$C_{26}H_{25}N_3O_2S$	70.41	5.68	9.47	7.23
			(443.55)	70.60	5.80	9.30	7.40
15b 168 <sup>b</sup>	168 <sup>b</sup>	40	$C_{20}H_{20}N_2O_3S$	65.20	5.47	7.60	8.69
			(368.44)	65.30	5.60	7.80	8.50
16a 140 <sup>b</sup>	140 <sup>b</sup>	70	C <sub>17</sub> H <sub>14</sub> N <sub>2</sub> O <sub>3</sub> S	62.57	4.32	8.58	9.82
			(326.36)	62.70	4.30	8.40	9.90
16b 150–2 <sup>d</sup>	150-2 <sup>d</sup>	68	C <sub>17</sub> H <sub>13</sub> N <sub>2</sub> O <sub>3</sub> SCI	56.59	3.63	7.76	8.88
			(360.80)	56.70	3.80	7.60	8.70
17a 210 <sup>d</sup>	210 <sup>d</sup>	66	C <sub>17</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub>	62.96	4.97	17.28	
			(324.34)	62.70	5.00	17.40	
17b 220 <sup>b</sup>	220 <sup>b</sup>	66	C <sub>23</sub> H <sub>19</sub> N <sub>3</sub> O <sub>3</sub>	71.68	4.97	10.90	
			(385.42)	71.80	4.70	10.70	
17c 2	215 <sup>b</sup>	69	$C_{24}H_{21}N_{3}O_{3}$	72.17	5.29	10.52	
			(399.45)	72.30	5.40	10.60	
17d	240 <sup>b</sup>	71	C <sub>23</sub> H <sub>18</sub> N <sub>3</sub> O <sub>3</sub> Br	59.49	3.91	9.05	
			(464.31)	59.60	3.80	9.20	

a. Solvent cryst.

b. from ethanol

c. from chloroform

d. from benzene

# 3-Amino-5-(4'-chlorophenyl)-1-(4'-ethoxyphenyl)-2-mercaptopyrrolo [2,3-c]pyrrol-4,6-dione (5)

A mixture of (1, 0.01 mol), N-p-chlorophenylmaleimide (4, 0.01 mol) and triethylamine (0.5 ml) in ether (20 ml) was stirred at room temperature for 3 hr. The obtained product after treatment with pet-ether (40-60) and fil-

#### Y.A. AMMAR et al.

tration was crystallized to give (5; Table I). IR spectrum showed a broad band around  $3319(NH_2)$ , 2930(CH-aliphatic) and 1712 (C=O). Also its mass spectrum showed m/z 416 (2.4%) with a base peak at m/z =86 and other significant peaks 344 (2.0%), 307 (5.6%), 274 (6.7%), 226 (7.7%), 191 (5.0%), 180 (12.3%), 148 (12.1%), 129 (16.5%) and 127 (46.2%).

#### 3-(4'-ethoxyphenyl)-2-thioxoquinazolin-4-one (6)

A solution of (1, 0.01mol) in DMF (15ml) was treated with anthranilic acid (0.01mol) and piperidine (0.5ml) then refluxed for 3hr. The solid that obtained after cooling and filtration was crystallized to afford (6; Table I). <sup>1</sup>HNMR in DMSO-d<sub>6</sub> Its spectrum showed signals at 1.4-1.7(3H,t,CH<sub>3</sub>-etoxy), 4.1-4.3(2H,q,CH<sub>2</sub>-ethoxy) and 7.2-8.3 (9H, m, Ar-H+NH) and mass spectrum exhibited m/z 299 (M + 1,14.1%), 298 (M<sup>+</sup>, 100%), 270 (30.2%), 237 (4.7%), 211 (10.7%), 163 (35.7%), 135 (24.4%), 119 (17.7%), 90 (10.0%) and 77(20.0).

#### 2-Thioxobenzimidazoles (7a,b)

**a** - A mixture of (1, 0.01mol), o-phenylenediamines (0.01mol) and piperidine (0.5ml) in DMF (15ml) was heated under reflux for 3hr. Cooling, filtered and the obtained product was crystallized to give (**7a,b**; Table I). <sup>1</sup>HNMR spectrum of **7a** in DMSO-d<sub>6</sub> showed signals at 2.3(3H,s,CH<sub>3</sub>), 7.2-7.7(5H,m,Ar+NH) and <sup>1</sup>HNMR of **7b** in DMSO-d<sub>6</sub>2.3(6H,s,2CH<sub>3</sub>) and 7.2(2H,s,Ar-H). Mass spectrum of **7a** showed m/z 164 (M<sup>+</sup>, 100%), 132[M-32(S), 14.3%], 106(24.5%) and 77(15.6%).

**b** - Compound **7a**,**b** were prepared according to the reported method<sup>11</sup>.

#### 5-Imino-4-thioxoimidazolidin-2-ones and -2,4-dithione (9a-c)

To a solution of (1, 0.01 mol) in ether (20ml), iso(thio)cyanates (0.01mol) and triethylamine (0.5 ml) were added, the reaction mixture was stirred at room temperature for 2 hr. The obtained products were crystallized from the proper solvent to afford (**9a-c**; Table I). IR of **9a** showed bands at 3240 (NH), 2980 (CH-aliphatic) and 1740 (C=O). While IR spectrum of **9c** showed the absence of CO group. <sup>1</sup>HNMR spectrum of **9a** in DMSO-d<sub>6</sub> 1.3–1.6 (3H, t, CH<sub>3</sub>-ethoxy), 4.2–4.5 (2H, q, CH<sub>2</sub>) and 7.3–8.0 (10H, m,

Ar-H+ NH) and <sup>1</sup>HNMR spectrum of **9b** in DMSO-d<sub>6</sub> 1.3–1.7 (3H, t, CH<sub>3</sub>), 4.2–4.5 (2H, q, CH<sub>2</sub>), 7.8–8.6 (8H, m, Ar-H) and 8.8 (1H, br, NH; cancelled with  $D_2O$ ).

#### 3-(4'-Ethoxyphenyl)-5-hydrazono-1-substitutedphenyl-4thioxoimidazolidin-2-ones (10a,b)

A mixture of (**9a or b**, 0.01mol) and hydrazine hydrate (0.012mol) in ethanol (20 ml) was stirred at room temperature for one hr. The obtained product after filtration was crystallized to give (**10a,b**; Table I). IR spectrum of **10a** showed bands at 3430, 3366 (NH<sub>2</sub>), 2993 (CH-aliphatic) and 1734(C=O). Also mass spectrum of **10b** showed 377 (M+2, 37.0% chlorine isotope), 375 (M<sup>+</sup>, 100%), 359 (M-16; NH<sub>2</sub>; 25.4%), 347(M-28; C=O; 43.7), 317 (12.2%), 284 (4.0%), 248 (4.9%), 196 (7.0%), 179 (20%), 149 (48.7%), 135 (84.9%), 121 (50%) and 107 (52.7%).

#### Interaction of (9b) with diphenylhydrazone or thiosemicarbazide

To a solution of (**9b**, 0.01mol) in ethanol (20ml), diphenylhydrazone or thiosemicarbazide (0.01mol) was added and the reaction mixture was refluxed for 3hr. The obtained product was filtered, washed with ethanol and crystallized to furnish (**12a,b**; Table I). IR spectrum of **12b** showed bands at 3263, 3163 (NH<sub>2</sub>,NH) and 1763 (C=O) and mass spectrum of **12a** exhibited m/z 523 (M+1, 1.1%), 522 (M<sup>+</sup>, 1.6%), 492 (100%), 444 (23.3%), 416 (3.9%), 362 (4.0), 358 (13.3%), 357 (37.8), 330 (4.8%), 292 (50.2%), 281 (4.1%), 264 (2.7%), 232 (5.6%), 182 (12.8%), 180 (58.9%), 153 (12.3%), 135 (12.2%), 125 (10.5%), 104 (13.6%), 77 (78.0%).

#### Imidaz[4,5-b]quinoxalines (13a-c)

A solution of (**9b**, 0.01mol) in ethanol (20 ml) was treated with o-phenylenedi amines (0.01mol) and triethylamine (0.5 ml). The reaction mixture was refluxed for 3hr and the solid that obtained after filtration was crystallized to give (**13a-c**, Table I). IR spectrum of **13a** showed 2997 (CH-aliphatic), 1748 (C=O) and 1604 (C=N), <sup>1</sup>H NMR spectrum of **13c** in DMSO-d<sub>6</sub> exhibited signals at 1.2–1.5 (3H, t, CH<sub>3</sub>-ethoxy), 2.6 (6H, s, 2CH<sub>3</sub>), 4.1–4.3 (2H, q, CH<sub>2</sub>-ethoxy) 7 7.2–8.8 (10H, m, Ar-H) and mass spectrum of **13a** showed m/z 417 (M<sup>+</sup>, 100%), 389 (67.5%), 360 (3.0%), 324 (2.9%), 277 (10.2%), 235 (16.2%), 206 (32.7%), 180 (4.6%), 148 (1.3%). 107 (2.7%) and 90 (10.4%).

#### 1-(4'-Ethoxyphenyl)-3-phenylthiohydantoin (14)

A solution of (**9a**, 0.01mol) in ethanol (20 ml) was treated with triethylamine (1 ml) and subjected to a stream of hydrogen sulfide for half an hour at room temperature. The obtained solid after filtration and washed with ethanol was crystallized to afford (**14**; Table I). <sup>1</sup>HNMR spectrum in DMSO-d<sub>6</sub> showed signals 1.3–1.6 (3H, t, CH<sub>3</sub>-ethoxy), 4.2–4.5 (2H, q, CH<sub>2</sub>-ethoxy), 5.2 (2H, s, CH<sub>2</sub>-methylene) and 7.3–8.1(9H,m,Ar-H).

#### 1-(4'-Ethoxyphenyl)-4-(4'-N,N-dimethylaminophenylmethylene)-3phenylthiohydantoin (15a)

A mixture of (14, 0.01 mol), p-N,N-dimethylaminobenzaldehyde (0.01 mol) and triethylamine (0.5 ml) in ethanol (20 ml) was refluxed for 3hr. The solid that obtained after cooling and filtration was crystallized to give (15a; Table I). <sup>1</sup>Hnmr spectrum in DMSO-d<sub>6</sub> showed signals at 1.3–1.6 (3H, t, CH<sub>3</sub>-ethoxy), 3.2 [6H, s, N(CH<sub>3</sub>)<sub>2</sub>], 4.1–4.3 (2H, q, CH<sub>2</sub>-ethoxy), 6.1 (1H, s, CH=) and 7.0.8.2 (13H, m, Ar-H).

#### 1-(4'-Ethoxyphenyl)-4-ethoxymethylene-1-phenylthiohydantoin (15b)

Compound (14, 0.01 mol) was added to a mixture of acetic anhydride and triethylorthoformate (1:1; 10 ml) and was refluxed for 4hr. The product that obtained after cooling and filtration was refluxed to afford (15b; Table I). Mass spectrum displayed m/z 368 ( $M^+$ , 33.2%), 352 (17.4%), 326 (100%), 309 (17.4%), 288 (7.1%), 224 (11.4%), 193 (12.8%), 177 (26.5%), 150 (54.2%), 119 (34.2%), 104 (15.0%) and 77 (45.0%).

#### 3-(4'-Ethoxyphenyl)-1-(phenyl or 4'-chlorophenyl)-4 thioxoimidazolidine-2,5-dione (16a,b)

To a solution of (**9a,b**, 0.01mol) in ethanol (20ml), Conc.HCl (5ml) was added with stirring at room temperature for 1/2 hr. The solid that precipi-

tated was filtered, washed with ethanol and crystallized to yield (16a,b; Table I). IR spectrum showed the complete disappearance of NH band already present in the parent compound.

#### 3-(4'-Ethoxyphenyl)-4-hydrazono-1-phenylimidazolidine-2,5dione (17a)

To a solution of (16, 0.01mol) in ethanol (20ml), hydrazine hydrate (0.012mol) was added and the solution was stirred at room temperature for 2hr. The solid that obtained was crystallized to give (17a; Table I). Its IR spectrum showed bands 3230, 3310 (NH<sub>2</sub>), 2970 (CH-aliphatic) and 1750, 1710 (2C=O).

#### 3-(4'-Ethoxyphenyl)-1-phenyl-4-substitutediminoimidazolidine-2,5dione (17b-c)

A mixture of (16, 0.01mol) and the requisite amine (0.01mol) in ethanol (20ml) was refluxed for 3hr. The obtained product after cooling and filtration was crystallized to afford (17b-c; Table I). Mass spectrum of 17b showed m/z  $385(M^+, 5.7\%)$ , 238 (2.8%), 209 (2.6%), 163 (3.6%), 135 (4.2%), 119 (15.2%), 93 (100%) and 77 (14.3%).

#### References

- I. Kattak, R. Ketcham, E. Schaumann and G. Adiwidjaja, J. Org. Chem., 50 (19), 3431 (1985).
- [2] H. Jamin, G. Michaol, J. Heterocycl. Chem., 24 (6), 1781 (1987).
- [3] R. Ketcham and E. Schaumann, Synthesis, 869 (1980).
- [4] R. Ketcham and E. Schaumann, J. Org. Chem., 45, 3748 (1980).
- [5] Y.A. Mohamed, A.M. Sh. El-Sharief, Y.A. Ammar and M.M. Ali, *Die Pharmazie*, 44 (11), 765 (1989).
- [6] Y.A. Mohamed, A.M. Sh. El-Sharief and M.M. Ali, J. Chin. Chem. Soc., 37, 511, (1990).
- [7] A.M. Sh. El-Sharief, R. Ketcham and E. Schaumann, *Phosphorus Sulfur and Silicon*, 46, 83 (1989).
- [8] A.M. Sh. El-Sharief, Y.A. Ammar, Y.A. Mohamed and M.S.A. El-Gaby, *Phosphorus Sulfur and Silicon*, 1-14 (1999).
- [9] W. Dieckmann and H. Kammerer, Chem. Ber., 38, 2977 (1905).
- [10] W. Dieckmann and H. Kammerer, Chem. Ber., 40, 3737 (1907).
- [11] Y. Saegusa, S. Horad and S. Nakamura, J. Heterocyclic Chem., 25, 1337 (1988).