

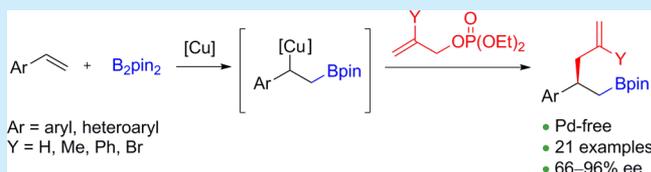
# Copper-Catalyzed Asymmetric Boryllallylation of Vinyl Arenes

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**S** Supporting Information

**ABSTRACT:** A copper-catalyzed, enantioselective method for the boryllallylation of vinyl arenes is reported. The reaction produces enantioenriched and functionalized organoboron compounds by sequentially incorporating boryl and allyl groups onto the C=C bond of vinyl arenes. Copper-catalyzed boryllative coupling of vinyl arenes with allyl phosphates successfully proceeds in a regio- and enantioselective manner in the absence of a palladium cocatalyst.



The installation of different functional groups onto C–C unsaturated bonds in a single process is a powerful method for the preparation of complex organic molecules.<sup>1</sup> In recent years, organocopper intermediates generated by borylcupration of unsaturated substrates with bis(pinacolato)diboron (B<sub>2</sub>pin<sub>2</sub>) have been discovered to be useful organometallic reagents for the formation of various C–C bonds through addition,<sup>2</sup> substitution,<sup>3</sup> and coupling reactions with electrophiles.<sup>4</sup> In these multicomponent reactions, borylated organocopper intermediates were generated in situ from alkenes, allenes, and alkynes in the presence of a copper catalyst.

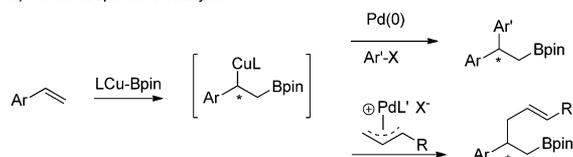
Recently, palladium and copper cooperative catalysis has been applied to coupling reactions of such organocopper intermediates with aryl halides.<sup>5</sup> The dual catalysis was also applied to the enantioselective allylboration of vinyl arenes by Liao and co-workers using *tert*-butyl allyl carbonates as electrophiles (Scheme 1, a).<sup>6</sup> Transition-metal-catalyzed allylation methods are considered valuable transformations due to the versatility of the olefin functional group incorporated

in the allylation.<sup>7</sup> In particular, copper-catalyzed allylic substitution<sup>8</sup> could be a good candidate for a series of tandem reactions of in situ generated organocopper intermediates. While a copper-catalyzed hydroallylation of styrenes has been reported,<sup>9</sup> copper-catalyzed boryllative coupling of simple styrenes with allylic electrophiles has not yet been reported without cooperative palladium catalysis. To date, only copper-catalyzed boryllallylations of allenes, alkynes, and vinyl boronates have been reported.<sup>10</sup> Herein, we report an enantioselective boryllallylation of vinyl arenes catalyzed by a chiral phosphine–copper catalyst that provides access to functionalized alkyboranes.

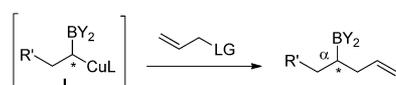
Our group recently reported the copper-catalyzed allylation of enantioenriched  $\alpha$ -boryl organocopper species (I).<sup>11</sup> Given the similar steric factors of  $\alpha$ -boron organocopper species and  $\beta$ -boryl organocopper species (II), we envisioned that catalytic boryllallylation could be achieved by copper catalysis in the absence of palladium. Upon applying our previous reaction conditions<sup>12</sup> developed for  $\alpha$ -boron organocopper species (I) to simple styrene (1a) with allyl phosphate, however, the desired product was obtained in poor NMR yield and ee (30%, 3% ee) along with the protonated side product. Therefore, we screened a series of reaction conditions to optimize the boryllallylation of styrene (Table 1). After extensive investigation starting with a racemic ligand (dppbz, Figure 1), we found that the desired product 3aa could be obtained in 84–87% yield when CuBr or CuCl was used as precatalyst with LiOMe as the base and allyl phosphate (2a) in dimethylacetamide (DMA) solvent. The use of other bases or solvents or other allylic electrophiles resulted in low NMR yields or substantial formation of side products resulting from C–H boryllation (3a') and protoboryllation (3a'').<sup>13</sup> Next, we screened a variety of chiral ligands for their effectiveness (entries 3–10). Among the screened ligands, Josiphos-type ligands and QuinoxP ligand gave promising yields of the desired product 3aa in good ee (entries 7–10). Finally, we

## Scheme 1. Catalytic Allylation Methods

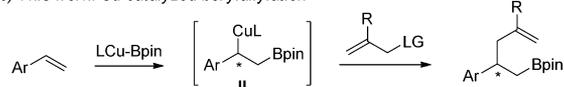
a) Pd/Cu cooperative catalysis



b) Allylation of B- $\alpha$ -chiral organocopper species

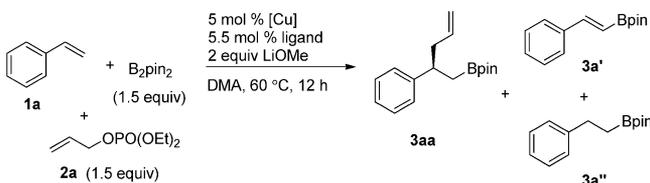


c) This work: Cu-catalyzed boryllallylation



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Table 1. Optimization



entry	[Cu]	ligand	conv <sup>a</sup> (%)	NMR yield <sup>a</sup> (%)		ee <sup>b</sup> of 3aa (%)
				3aa	3a' + 3a''	
1	CuBr	dppbz	94	87 (80) <sup>c</sup>	<2	
2	CuCl	dppbz	92	84	<2	
3	CuBr	( <i>R</i> )-Segphos	90	50	3	75
4	CuBr	Duanphos	100	98 (95) <sup>c</sup>	<2	77
5	CuBr	( <i>S,S</i> )-BPE	43	43	38	56
6	CuBr	( <i>R,R</i> )-Walphos	70	48	<2	18 ( <i>S</i> )
7	CuBr	( <i>R,S</i> )- <i>t</i> -Bu-Josiphos	100	55	19	89
8 <sup>d</sup>	CuBr	( <i>R,S</i> )-Josiphos	92	73	3	87
9 <sup>d</sup>	CuBr	( <i>R,R</i> )-QuinoxP	97	77 (73) <sup>c</sup>	2	88
10 <sup>d</sup>	CuCl	( <i>R,R</i> )-QuinoxP	97	78 (74) <sup>c</sup>	2	90

<sup>a</sup>Determined by <sup>1</sup>H NMR analysis using DMF as an internal standard.

<sup>b</sup>Determined by HPLC analysis on a chiral stationary phase. <sup>c</sup>Isolated yield of the corresponding hydroxy product via oxidation using NaBO<sub>3</sub>. <sup>d</sup>Reaction temperature was 40 °C instead of 60 °C.

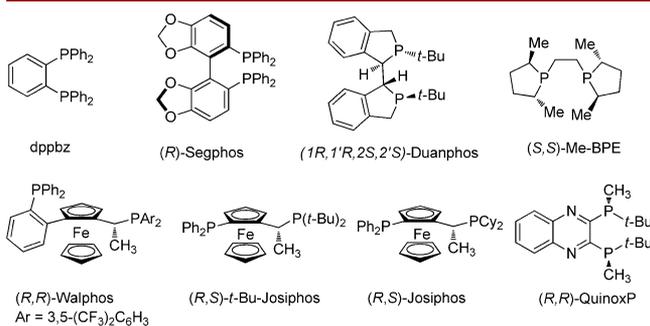
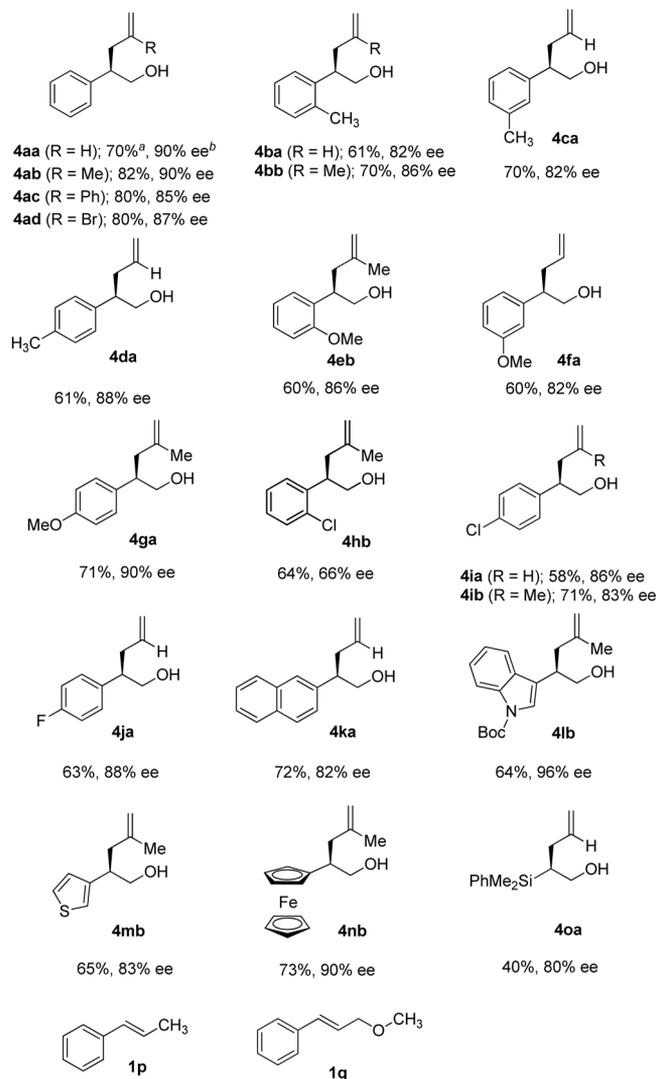
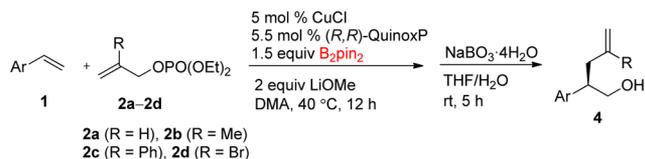


Figure 1. Ligand structure.

chose CuCl/QuinoxP in DMA (entry 10) as an optimized set of reaction conditions.

Having identified optimized conditions, we next investigated the substrate scope of the boryllallylation (Scheme 2). Reactions of styrene (**1a**) with different allylic electrophiles (**2a–d**) proceeded to give good yields of boryllallylated products with good enantioselectivities. Substituted allylic electrophiles (**2b–d**) provided improved yields of the products compared to allyl phosphate **2a**, presumably due to less competing side reaction with the active Cu–Bpin catalyst. In particular, the use of Br-substituted allylic phosphate (**2d**) led to the desired compound containing a vinyl bromide, which could be further utilized in a variety of other organic transformations.<sup>14</sup> *o*-, *m*-, and *p*-methyl and -methoxy-substituted styrenes were all reactive (**1b–g**), but slightly decreased ee values were observed in the *ortho*- and *meta*-substituted cases. Substrates (**1e–j**) with an electron-donating- or electron-withdrawing substituent were suitable substrates in the boryllallylation. However, the vinyl arene (**1h**) with an *o*-Cl exhibited relatively low enantioselectivity compared to other *ortho*-substituted vinyl arenes. The reaction

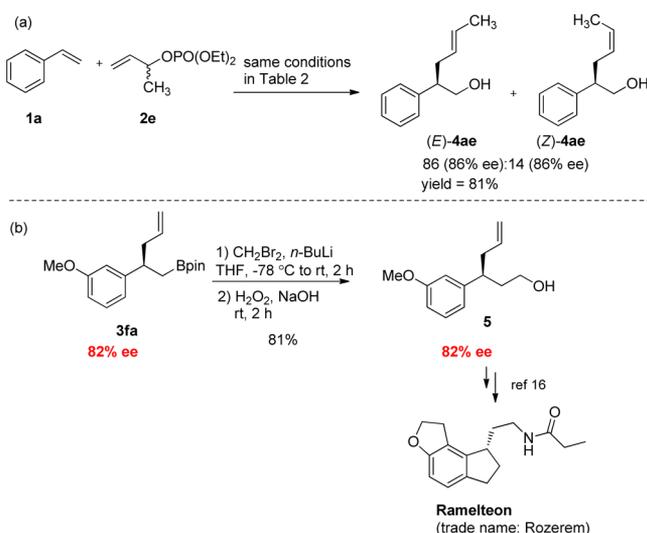
Scheme 2. Substrate Scope in Asymmetric Boryllallylation<sup>a</sup>

<sup>a</sup>Isolated yield. <sup>b</sup>Determined by HPLC analysis on a chiral stationary phase.

of vinyl naphthalene and other heteroaromatic vinyl arenes gave the desired products in good yields and ee. In particular, 3-vinyl indole (**11**) and vinyl ferrocene substrate (**1n**) afforded the corresponding products **41b** and **4nb** in 96% and 90% ee, respectively. The reaction of a vinylsilane gave product **4oa** in 40% yield and 80% ee. Finally, we found that internal alkene substrates (**1p** and **1q**) did not afford the desired coupled products due to competitive reactions of the borylcopper catalyst with allyl phosphate.

This three-component coupling was also effective with secondary allylic phosphate **2e** (Scheme 3, a), affording the corresponding allylation products (*E*)- and (*Z*)-**4ae** in a 6:1 ratio in 86% ee. These results indicate the S<sub>N</sub>2'-substitution occurs without much influence of the stereogenic center of the allylic phosphate. However, 1,2-disubstituted internal allylic

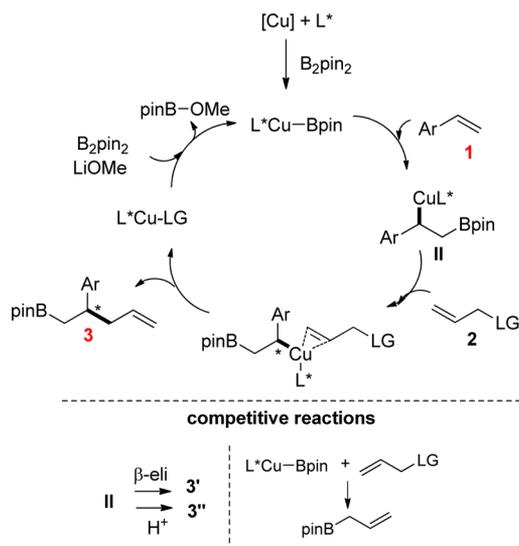
### Scheme 3. Boryllallylation with Allyl Phosphate 2e and Application of Product 3



phosphates with increased steric demand at the terminal olefin carbon were not efficient (see the Supporting Information for details). In a brief application, the boryllallylated product 3fa was homologated<sup>15</sup> without erosion of the ee to a known intermediate in the synthesis of ramelteon<sup>16</sup> (Scheme 3, b).

A proposed catalytic cycle for this reaction is shown in Scheme 4. The active catalyst L<sup>\*</sup>Cu–Bpin reacts with vinyl

### Scheme 4. Proposed Catalytic Cycle and Competitive Reaction Pathways



arene to form the borylcuprated intermediate II, which undergoes S<sub>N</sub>2'-allylic substitution with an allylic electrophile to yield the desired product (3). A catalyst regeneration step completes the catalytic cycle. Other undesired products can form from intermediate II by side reactions and direct boryl substitution of the allylic electrophile with the Cu–Bpin catalyst.

In conclusion, we have described a copper-catalyzed, enantioselective method for the boryllallylation of vinyl arenes in the absence of a palladium cocatalyst. Enantioenriched and functionalized organoboron compounds can be efficiently

obtained by sequentially incorporating boryl and allyl groups onto the C=C bond of vinyl arenes in a regio- and enantioselective manner. Efforts to develop more widely applicable boryllallylation protocols are underway.

### ■ ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.7b03022.

Experimental procedures, characterization of products, and NMR spectra (PDF)

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#### Notes

The authors declare no competing financial interest.

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(12) A combination of 5 mol % of CuCl, 5.5 mol % of (*R,R*)-Walphos, and 2 equiv of LiOt-Bu was used in THF at 40 °C. Compounds **3a** and **3a''** were produced in 30% and 11% NMR yields, respectively.

(13) See the [Supporting Information](#) for details and [Scheme 4](#).

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