# An Alternative Approach to the Synthesis of Parvistone C

Check for updates

Ramakrishnam Raju Addada,<sup>a,b</sup> Venkata Reddy Regalla,<sup>a,b</sup> Sivarami Reddy Gangireddy Venkata,<sup>b</sup> Venkata Naresh Vema,<sup>b</sup> and Venkateswara Rao Anna<sup>a</sup>\*

<sup>a</sup>Department of Chemistry, Koneru Lakshmaiah Education Foundation, Green fields, Vaddeswaram, Guntur, Andhra Pradesh 522502, India <sup>b</sup>Medicinal Chemistry Division, GVK Biosciences Private Limited, 28A, Nacharam, Hyderabad, Telangana 500076, India

"Medicinal Chemistry Division, GVK Biosciences Private Limited, 28A, Nacharam, Hyderabad, Telangana 500076, India \*E-mail: avrchemistry@gmail.com

Received July 27, 2018

DOI 10.1002/jhet.3455

Published online 00 Month 2019 in Wiley Online Library (wileyonlinelibrary.com).



A stereoselective total synthesis of parvistone C, an oxygenated natural styryllactone, has been accomplished in excellent overall yield by employing asymmetric aldol reaction, asymmetric epoxidation and regioselective epoxide ring opening as the key steps. Our synthetic strategy is very simple, concise and no use of protecting groups.

J. Heterocyclic Chem., 00, 00 (2019).

## **INTRODUCTION**

Chiral lactone rings are widely distributed in many biologically active natural products [1]. Particularly,  $\alpha$ , $\beta$ -unsaturated  $\delta$ -lactone gained the great attention of researchers due to a variety of pharmaceutical properties including antitumor, antifungal, antibacterial, and immunosuppressive properties [2]. A series of new 6*S*-styryllactones parvistone A-E were isolated from a methanolic extract of *Polyalthia parviflora* leaves by Liou *et al.* in 2014 [3]. Their structures were elucidated on the basis of NMR spectroscopy and single-crystal X-ray analysis. These compounds exhibit potent anti-inflammatory and cytotoxic properties. Parvistone C (**1b**) is structurally related to parvistone B (**1a**) and differs in the stereochemical configurations (Figure 1).

Recently, asymmetric total syntheses of parvistone A-E natural products have been reported in the literature [4]. Very recently, Radha Krishna et al. [4e] disclosed the first total synthesis of parvistone C (1b) starting from commercially available 1,3-propane diol in 13 steps with 13% overall yield. They employed a Sharpless asymmetric dihydroxylation, stereoselective aryl Grignard reactions, Still-Gennari olefination, and intramolecular cyclization as the key steps (Scheme 1). Herein, we report a five-step total synthesis of parvistone C (1b) commercially available, inexpensive from transcinnamaldehyde using a protecting group-free strategy.

#### **RESULTS AND DISCUSSION**

The retrosynthetic pathway of parvistone C (**1b**) is presented in Scheme 2. The key step is a *Mukaiyama* asymmetric aldol reaction to install lactone stereogenic center. The remaining two chiral centers could then be introduced by asymmetric epoxidation reaction followed by regioselective epoxide ring opening with methanol.

As shown in Scheme 3, our synthesis started with *Mukaiyama* asymmetric aldol reaction. Thus, transcinnamaldehyde **3** was allowed to react with Chan's diene **4** using catalytic amounts of Ti  $(O'Pr)_4/(S)$ -BINOL complex (10 mol%). This gave  $\delta$ -hydroxy- $\beta$ -ketoester **5** as a >95% enantiomeric excess (Chiral HPLC analysis)



Figure 1. Structures of parvistone B (1a) and C (1b).

Scheme 1. Previous synthetic approach to parvistone C (1b).

HO OH Association and the second seco

Scheme 2. Retrosynthetic analysis of parvistone C (1b).



in 75% yield [5]. The aldol adduct **5** was then subjected to a carbonyl reduction with NaBH<sub>4</sub> to give the diol as the product. After, we have converted the diol to lactone **2** 

*via* hydrolysis of ester followed by intramolecular lactonization in 62% yield over two steps [6]. The elimination of hydroxyl group in  $\beta$ -hydroxy- $\delta$ -lactone **2** was performed smoothly to furnish the desired  $\alpha$ , $\beta$ -unsaturated  $\delta$ -lactone (**6**) in 76% yield [7]. Next, the compound **6** was subjected to Han's reaction conditions [8] using (+)-(*R*,*R*)-salen-Mn (III) as the catalyst and Oxone as the oxidant at 0°C to furnish the desired goniothalamin oxide **7** with excellent regioselectivity and diastereoselectivity (dr, 98:2) in 81% yield. Finally, the



 Table 1

 Comparison of <sup>1</sup>H-NMR and <sup>13</sup>C-NMR data of natural parvistone C and synthetic 1b.



	<sup>1</sup> H-NMR			<sup>13</sup> C-NMR			
Н	Natural	Synthetic	$(\Delta\delta)_{N,S}$	С	Natural	Synthetic	$(\Delta\delta)_{N,S}$
				2	164.0	164.0	0.0
3	6.05 (dd, 9.5, 2.4)	6.04 (ddd, 9.6, 2.7, 0.8)	-0.01	3	120.8	120.8	0.0
4	6.97 (dt, 9.5, 6.4, 2.4)	6.97 (ddd, 9.6, 6.1, 2.1)	0.00	4	146.0	145.9	-0.1
5	2.26 (ddd, 18.2, 6.4, 4.0)	2.25 (dddd, 18.1, 6.1, 3.7, 0.7)	-0.01	5	26.2	26.1	-0.1
	2.84 (ddt, 18.2, 13.2, 2.4)	2.83 (ddt, 18.1, 12.7, 2.1)	-0.01				
6	4.95 (ddd, 12.6, 4.0, 1.2)	4.95 (ddd, 12.7, 3.8, 1.5)	0.00	6	76.1	76.1	0.0
7	3.61 (dt, 8.4, 1.2)	3.61 (dd, 9.0, 1.5)	0.00	7	74.8	74.8	0.0
8	4.38 (d, 8.4)	4.38 (d, 8.8)	0.00	8	81.9	81.9	0.0
Ph	7.41–7.36 (m, 5H)	7.44–7.31 (m, 5H)		9	138.5	138.5	0.0
OMe	3.22 (s)	3.22 (s)	0.00	10	127.8	127.8	0.0
				11	128.7	128.6	-0.1
				12	128.5	128.5	0.0
				13	128.7	128.6	-0.1
				14	127.8	127.8	0.0
				OMe	56.7	56.7	0.0

stereoselective and regioselective epoxide ring-opening of goniothalamin oxide 7 with methanol in the presence of Lewis acid Eu (OTf)<sub>3</sub> provided the target natural product parvistone C (**1b**) in 95% yield with 96:4 dr. The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR data of natural parvistone C were well agreed with our synthetic compound **1b** (Table 1).

### **CONCLUSIONS**

The protecting group-free stereoselective total synthesis of lactone natural product parvistone C (1b) was achieved in simple five steps without using any protecting group from readily available starting material transcinnamaldehvde with 27% overall vield. Kev transformations of our synthetic scheme include an asymmetric aldol reaction and asymmetric epoxidation and regioselective epoxide ring opening.

#### **EXPERIMENTAL**

General. Solvents were dried over standard drying agents and freshly distilled prior to use. All moisturesensitive reactions were carried out under nitrogen. Organic solutions were dried over anhydrous MgSO<sub>4</sub> and concentrated below 40°C *in vacuo*. All column chromatographic separations were performed on silica gel of 60–120 mesh. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra were recorded in CDCl<sub>3</sub> solvent at 300, 400, 500 MHz, and 75, 100, 125 MHz, respectively, using tetramethylsilane as internal standard. Optical rotations were measured with a digital polarimeter at 25°C. Mass spectra were recorded on a mass spectrometer by electrospray ionization (ESI-MS). High-resolution mass spectra (ESI-HRMS) were recorded on an ESI-QTOF mass spectrometer.

General procedure for the synthesis of (*R*)-6-((1*R*,2*R*)-1hydroxy-2-methoxy-2-phenylethyl)-5,6-dihydro-2H-pyran-2one (Parvistone C) (1b). To a stirred solution of compound **8** (40 mg, 0.18 mmol) in methanol (1 mL) was added Eu (OTf)<sub>3</sub> (12.2 mg, 0.02 mmol) at room temperature and stirred for 12 h. The methanol was removed under vacuum, and the residue was purified by flash column chromatography (hexane: EtOAc, 6:4) to provide parvistone C (**1b**) (43 mg, 95%) as a white solid.  $[\alpha]_D^{25} = -219$  (*c* 0.5, CHCl<sub>3</sub>), HRESIMS *m*/*z* 271.0936 [M + Na]<sup>+</sup> (Calcd for C<sub>14</sub>H<sub>16</sub>O<sub>4</sub>Na, 271.0940).

Acknowledgments. R. K. R. thanks K L University for constant encouragement during this research program. R. K. R. is also grateful to GVK Biosciences for providing basic research facility.

#### **REFERENCES AND NOTES**

[1] (a) Negishi, E.; Kotora, M. Tetrahedron 1997, 53, 6707; (b) Collins, I. J Chem Soc Perkin Trans 1999, 1, 1377.

[2] (a) Chen, W. Y.; Wu, C. C.; Lan, Y. H.; Chang, F. R.; Teng, C. M.; Wu, Y. C. Eur J Pharmacol 2005, 522, 20; (b) Boucard, V.; Broustal, G.; Campagne, J. M. Eur J Org Chem 2007, 2007, 225.

[3] Jing-Ru, L.; Tung-Ying, W.; Tran, D. T.; Tsong-Long, H.; Chin-Chun, W.; Yuan-Bin, C.; Michael, Y. C.; Yu-Hsuan, L.; Mohamed, E.; Shwu-Li, W.; Ludger, B.; Shyng-Shiou, Y.; Ming-Feng, H.; Shu-Li, C.; Fang-Rong, C.; Yang-Chang, W. J Nat Prod 2014, 77, 2626.

[4] (a) Li, Z.; Tong, R. Synthesis 2016, 48, 1630; (b) Ramesh, P.;
Rao, T. P. J Nat Prod 2016, 79, 2060; (c) Ramesh, P. Synthesis 2016, 48, 4300; (d) Ramesh, P. ChemistrySelect 2016, 1, 3244; (e) Jala, R.;
Nomula, R.; Krishna, P. R. Synth Commun 2017, 47, 1879; (f) Ramesh, P.;
Reddy, Y. N.; Reddy, T. N.; Srinivasu, N. Tetrahedron Asymmetry 2017, 28, 246.

[5] (a) Brownbridge, P.; Chan, T. H.; Brook, M. A.; Kang, G. J. Can J Chem 1983, 61, 688; (b) Rayala, N. K.; Meshram, H. M. Tetrahedron Lett 2011, 52, 1003; (c) Soriente, A.; Rosa, M. D.; Stanzione, M.; Villano, R.; Scettri, A. Tetrahedron Asymmetry 2001, 12, 959; (d) Xu, Q.; Yu, J.; Han, F.; Hu, J.; Chen, W.; Yang, L. Tetrahedron Asymmetry 2010, 21, 156.

[6] (a) Birgitta, H.; Annemarie, K.; Hans, S. Tetrahedron Asymmetry 1993, 4, 153; (b) Hirokazu, U.; Tetsuji, M.; Fumie, S. Tetrahedron Lett 1992, 33, 4183; (c) Bhaskar, R. G.; Minami, T.; Hanomoto, T.; Hiyama, T. J Org Chem 1991, 56, 5752.

[7] Sato, M.; Sunami, S.; Sugita, Y.; Kaneko, C. Chem Pharm Bull 1994, 42, 839.

[8] Lim, J.; Kim, I.-H.; Kim, H. H.; Ahn, K.-S.; Han, H. Tetrahedron Lett 2001, 42, 4001.