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The Origin of High Stereoselectivity in Di-tert-butylsilylene-directed α -Galactosylation

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Abstract

The origin of the high $\alpha(1,2-cis)$ -stereoselectivity in the reaction of galactosyl and galactosaminyl donors with a di-*tert*-butylsilylene (DTBS) group with several nucleophiles has been elucidated by means of experimental and computational approaches. DTBS overcomes any other cyclic protecting groups examined to date and the $\beta(1,2-trans)$ -directing effect due to the neighboring participation by CO groups at C2. Requirements for the $\alpha(1,2-cis)$ -stereoselectivity are as follows: (1) generation of an oxocarbenium ion; (2) a *galacto*-type glycosyl donor with a cyclic protecting group bridging O4 and O6 to form a six-membered ring; (3) through-space electron donation from O4 and O6 into the empty *p*-orbital of the anomeric carbon to stabilize the oxocarbenium intermediate; (4) steric hindrance due to bulky alkyl substituents on the cyclic protecting group to prevent nucleophilic attack from the β -face; and (5) a 4,6- θ -silylene structure. Furthermore, it was found that the strong stereo-directing effect of the DTBS group was unique and specific among the various cyclic protecting groups examined.

Introduction

The pursuit of stereoselectivity in chemical glycosylation reactions is a key to success in the synthesis of glycoconjugates. To date, many factors in glycosylation have been identified that potentially affect the stereochemistry of glycoside product. Among these factors, the presence of an acyl protecting group at the C2 position of the glycosyl donor is one of the most useful factors for forming 1,2-trans-glycosides. Meanwhile, the synthesis of 1,2-cis-glycosides has been plagued by lower stereoselectivity due to a lack of reliable factors for controlling the glycosylation reaction. Hence, synthetic chemists have focused on developing methods for

1,2-cis-stereoselective glycosylation.³ In particular, the formation of the $\alpha(1,2-cis)$ -glycosidic linkage between galactosamine and serine/threonine has garnered much attention because it is the core linkage of natural O-glycans in glycoproteins. However, the construction of the α-N-acetylgalactosamine (α-GalNAc) linkage is usually difficult because the presence of the 2-acetamide group leads to the formation of the $\beta(1,2-trans)$ -glycoside through neighboring group participation or the undesirable 1,2-oxazoline derivative through intramolecular cyclization. Thus, 2-azido-2-deoxy-galactosyl donors have been used in glycosylation reactions to maximize the anomeric effect, resulting in the formation of α-glycosides.⁵ Although 2-azidogalactosaminyl (GalN) donors are useful for synthesizing glycans containing α -GalNAc, the glycosylation products are often α/β mixtures with moderate stereoselectivity. In 2003, we developed a di-tert-butylsilylene (DTBS)-directed α-stereoselective galactosylation⁶ and found that both GalN and galactosyl donors bearing the DTBS group as a protecting group bridging the C4 and C6 positions gave unusual high α-stereoselectivity (Figure 1). Notably, this method was compatible with a wide variety of leaving groups, activator systems, reaction temperatures, solvent systems, and protecting groups (even in the presence of a neighboring acyl group at C2). High α -selectivity was almost completely independent of the nucleophile (the sugar acceptor), which could be an alcohol, either linear or sterically hindered, a phenol, Ser, Thr and partially protected monosaccharides.⁷ Furthermore, DTBS-directed α-galactosylation was used to efficiently synthesize various α-GalNAc/Gal-containing glycans, such as α-GalNAc-Ser/Thr, α-galactosyl ceramides,⁸ isoglobotrihexosylceramide,⁸ the α-tetrasaccharyl-Ser segment of glycophorin A, p-nitrophenyl O-glycan cores, 4-methylumbelliferyl T-antigen, human ABO histo-blood group antigens,11 and some gangliosides.12 Also, glycosyl donors that are

conformationally restricted by the DTBS group can be effective for a wide variety of stereoselective glycosylations, such as β -arabinofuranosylation, 13 β -galactofuranosylation, 14 α -sialylation, 15 and β -mannosylation. 16 These reports demonstrate that the DTBS group can serve as a factor for controlling glycosylation stereoselectivity. Several groups have reported theoretical and experimental studies regarding the ring restriction of glycosylation. 17,18 responsible for the reactivity of the donor and the stereochemical outcome of glycosylation. 17,18

Figure 1. Outline of the DTBS-directed α -galactosylation

In previous papers, we proposed a reaction mechanism to explain the high α -stereoselectivity of the DTBS-directed reaction, which is summarized as follows (Figure 2). The previous X-ray crystallographic analysis of a DTBS-protected donor showed a six-membered ring composed of a 4,6- α -DTBS ketal moiety and a C4-C5-C6 bond in a nearly half-chair conformation, which resulted in the *tert*-butyl group being positioned close to the anomeric carbon, even in the ground state. This implies that the steric effect of the DTBS group during the course of the glycosylation leads to the α -stereoselectivity. Furthermore, through-space electron donation may contribute to the high α -stereoselectivity. Because of through-space electrostatic stabilization of the oxocarbenium ion by an axially oriented electronegative substituent at C4, the group on the C2 oxygen or nitrogen cannot interact effectively with the oxocarbenium ion in

the intermediate. This lack of interaction, together with the steric effect of the DTBS group, results in the prominent α -stereoselectivity. However, this mechanism for the DTBS-directed α -galactosylation is still only a hypothesis and needs experimental verification. Therefore, to gain insight into the origin of the α -selectivity, we examined how the steric and electronic effects of the substituents on both O4 and O6 of the glycosyl donor influenced stereoselectivity. Here, we report a systematic study using experimental and computational approaches in combination to ascertain the origin of stereoselectivity in DTBS-directed α -galactosylation.

Figure 2. Plausible mechanism for the α -selectivity in DTBS-directed α -galactosylation

Results and Discussion

In this study, we focused on the relationship between stereoselectivity and the magnitude of steric and electronic effects of substituents on O4/O6 of the glycosyl donor. For this purpose, we designed various GalN and glucosaminyl (GlcN) donors in which the only structural variation is the protecting groups on O4 and O6 (Figure 3). Selected protecting groups were a diisopropylsilylene group (less bulky than the DTBS group; 2), various acetal groups that differed in bulkiness and electronic properties (3–9), cyclic groups that differed in ring size (10, 12), and acyclic groups (11, 13–15). Donors 1–16 were synthesized as outlined in Scheme 1. Starting from the known compound phenyl 2-deoxy-2-phthalimide-1-thio-β-D-galactopyranoside (17),²⁰ 4,6-*O*-benzylidene-protected donor 6 was prepared according to a typical acetalization procedure followed by benzylation at C3.

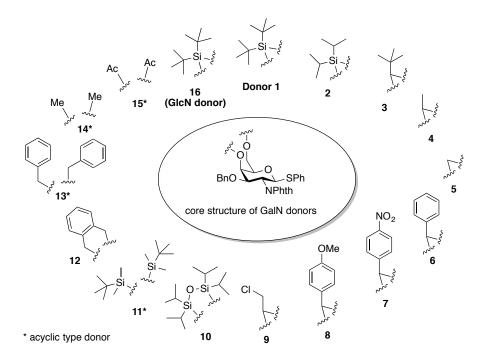
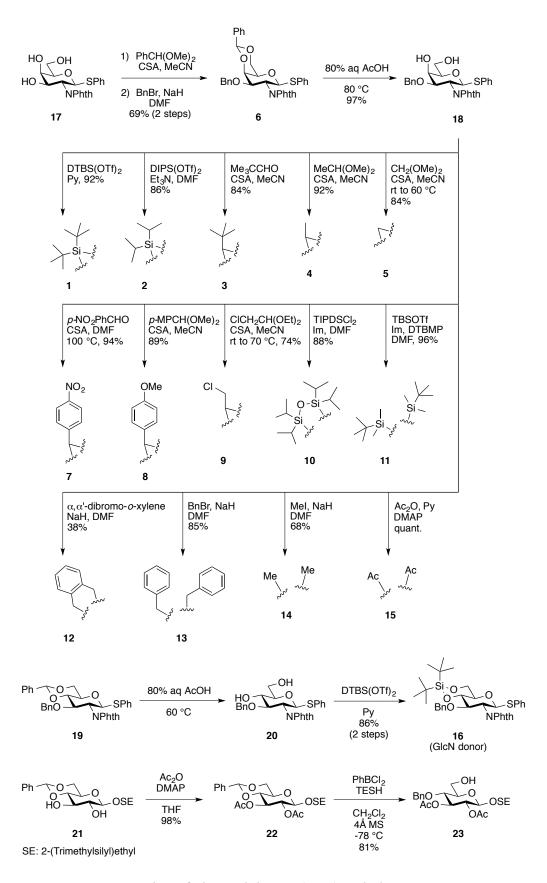


Figure 3. Structures of glycosyl donors used in this study

Acidic hydrolysis then gave 4,6-diol product 18, which was the common intermediate for synthesizing the other GalN donors. Compound 18 was transformed into GalN donors 1-15 using appropriate reaction conditions for each. For silvlene-bridged donors, 18 was reacted with dialkylsilyl ditriflate in the presence of base, giving 1 (92%) and 2 (86%). Treatment of 18 with pivalaldehyde, dimethylacetal, or dimethoxymethane in the presence of camphorsulfonic acid (CSA) afforded alkylidene donors 3 (84%), 4 (92%), and 5 (84%), respectively. Substituted benzylidene donors were derived from 18 by treatment with 4-nitrobenzaldehyde or anisaldehyde dimethyl acetal under acidic conditions, affording 7 (94%) and 8 (89%), respectively. Similarly, the reaction of 18 with 2-chloroacetaldehyde diethyl acetal in the presence of camphorsulfonic acid afforded 9 in 74% yield. 1',1',3',3'-Tetraisopropyldisiloxane (TIPDS)-protected donor was prepared by treating 18 with TIPDSCl₂ and imidazole in DMF, giving 10 in 88% yield. The reaction of 18 with tert-butyldimethylsilyl (TBS) triflate, imidazole along with 2,6-di-tert-butyl-4-methylpyridine as an acid scavenger gave acyclic 4,6-di-O-TBS-protected donor 11 in 96% yield. 4,6-O-Xylylene-protected donor 12 was prepared by the Williamson ether synthesis. Similarly, acyclic donors 13 and 14 were obtained in yield of 85% and 68%, respectively. The reaction of 18 with acetic anhydride in pyridine afforded 4,6-O-acetyl-protected donor 15 in quantitative yield. Meanwhile, glucosamine donor 16 bearing the DTBS group was prepared from the known glucosamine derivative 19²¹ in two steps. Hydrolysis of the benzylidene acetal under acidic conditions gave 20, which was reacted with DTBS(OTf)₂ in pyridine to afford 16 in good yield. The glucosyl acceptor (23) was synthesized as follows and used in all the glycosylation experiments. The hydroxyl groups of the known 2-(trimethylsilyl)ethyl 4,6-O-benzylidene-β-D-glucopyranoside 21²² were acetylated to give **22**, and subsequent reductive opening of the benzylidene acetal furnished 6-OH glucosyl acceptor **23** in 79% yield over the two steps.



Scheme 1. Preparation of glycosyl donors (1-16) and glucose acceptor 23

Next, GalN/GlcN donors 1–16 and glucosyl acceptor 23 were used in glycosidation experiments to examine the relationship between substituents and stereoselectivity in relation to the cyclic protecting groups at C4 and C6 of the glycosyl donor. All glycosylation reactions were carried out using the NIS-TfOH promoter system²³ in CH_2Cl_2 at 0 °C. Moreover, almost all the glycosidations proceeded rapidly, requiring a reaction time less than 30 min. This high reactivity can be explained by Bols' work^{18d} which found that the dihedral angle based on the exocyclic C5–O6 bond can strongly influence the reactivity of carbohydrates and gg conformer is the most reactive among tg, gg, and gt conformers. Specifically, the C5–O6 bond in the 4,6-G-ring-fused galactosyl donors must receive the imposition of the relatively reactive gg conformer. Figure 4 shows the results of glycosidation experiments. The α/β ratios were calculated from each of the isomers isolated after column chromatography.

Donor 1 bearing the DTBS group was glycosidated with acceptor 23 and the corresponding disaccharide 24 was obtained in 94% isolated yield with prominent α -selectivity. This result was consistent with our previous works. ⁶⁻¹² Notably, the α -stereo-directing effect of the DTBS group overrides the effect of neighboring group participation, which has been reliably used for $\beta(1,2-trans)$ -glycoside formation. We have attributed high α -stereoselectivity to steric effect arising from bulky *tert*-butyl substituents of the DTBS group. Thus, silylene-type donor 2 was examined, in which both *tert*-butyl substituents of 1 were replaced with less bulky isopropyl groups. The reaction gave glycosidated product 25 in 99% yield with 79:21 (α/β) stereoselectivity, which was lower than the α -stereoselectivity obtained with the *tert*-butyl groups. Thus, the bulk of the substituents in the silylene structure played a role in the observed stereoselectivity. Next, to assess the steric effect of substituents on the cyclic protecting group,

cyclic acetal-type donors **3–5** were examined, the steric bulk of which decreased in the order of **3** (steric parameter E_s : $E_{sCMe^3} = -1.54)^{24}$ to **4** ($E_{sMe} = 0.00$) and **5** ($E_{sH} = 1.24$). (S)-tert-Butylmethylidene donor **3** afforded the corresponding disaccharide **26** in 93% yield

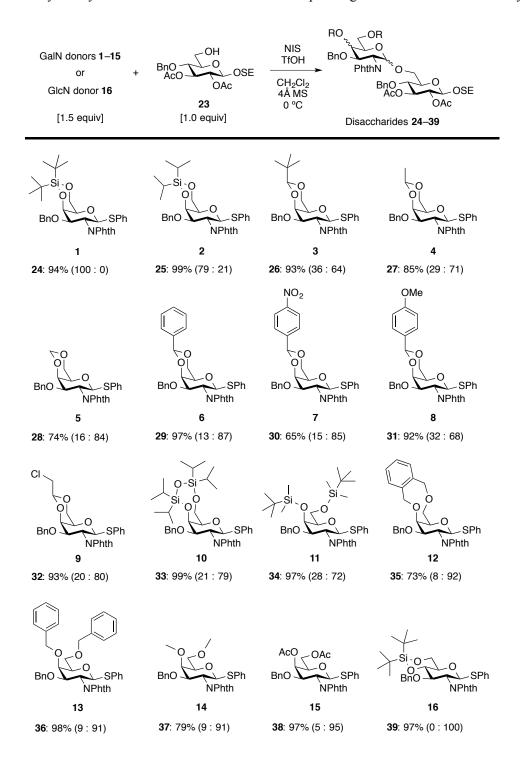


Figure 4. Glycosidation of GalN/GlcN donors 1–16 with glucosyl acceptor 23. Percentages are isolated yields of the corresponding disaccharides obtained after column chromatography. Values in parentheses are α/β stereoselectivities calculated based on the yield of isolated isomers.

along with 36:64 (α/β) stereoselectivity. Similarly, (S)-ethylidene donor 4 and methylidene donor 5 were coupled with 23, giving 27 in 85% yield ($\alpha/\beta = 29.71$) and 28 in 74% yield ($\alpha/\beta = 29.71$) and 28 in 74% yield ($\alpha/\beta = 29.71$) 16:84), respectively. These results strongly suggested that formation of the α-product could decrease as the steric bulk was decreased; in other words, the stereoselectivity was most likely to relate to the bulkiness of substituents at C4 and C6. Importantly, comparison of 1 with 3 showed the significance of the silvlene structure for obtaining high α -selectivity, because both the compounds contained a tert-butyl substituent as a source of steric hindrance, but it was connected to silicon in 1 and carbon in 3. This suggests that the spatial position of the tert-butyl group in the oxocarbenium intermediate and in the ground state might give rise to the stereoselectivity. Based on X-ray crystallographic data of the DTBS donor, 6 the endo-tert-butyl group in 1 is closer to the anomeric carbon compared to an *endo*-substituent of general cyclic acetal protecting groups. When 1 is activated to form the corresponding oxocarbenium intermediate, the tert-butyl group should move much closer to the anomeric carbon and prevent β -face nucleophilic attack by the glycosyl acceptor, leading to the observed α -selectivity. In contrast, the tert-butyl group in 3 is expected to be positioned far from the anomeric carbon in the oxocarbenium intermediate and would not sufficiently block the β -face. Next, we investigated the effect on stereoselectivity of the electronic environment at O4 and O6 of the glycosyl donor, because we expected through-space electron donation to contribute to the stereoselectivity. Glycosidation of benzylidene donor 6 with 23 proceeded smoothly and gave the corresponding disaccharide 29 in excellent yield with 13:87 (α/β) stereoselectivity. Aryl-substituted benzylidene-type donors, namely, 7 with an electron-withdrawing group (p-NO₂) and 8 with an electron-donating group (p-OMe), were then examined for their stereoselectivity. Donor 8 had clearly higher α-selectivity compared with the parent benzylidene donor 6, whereas 7 showed no difference in stereoselectivity compared with 6. These results could be a consequence of the different through-space stabilization of positive charge at the anomeric center affecting the position of substituents on the cyclic protecting group upon formation of the oxocarbenium intermediate stage. To further test this hypothesis, we designed donor 9 with electronegative chlorine atom on the ethylidene moiety; then, we compared the results with those for 4 (85% yield, $\alpha/\beta = 29.71$). Decreased α -selectivity was observed for 9 (93% yield, $\alpha/\beta = 20.80$), even though its chlorine atom ($E_{\text{sCH}_2\text{Cl}} = 0.24$) is bulkier than the hydrogen atom ($E_{\rm sCH_3}$ 0.00) on 4. Next, to examine the stereoselectivity arising from ring size formed by cyclic protecting groups, donors 10 and 12 with larger ring sizes were investigated. First, donor 10 with a TIPDS group, which is bulkier than the DTBS group, was subjected to glycosidation with 23, producing disaccharide 33 in 99% yield with 21:79 (α/β) stereoselectivity. This experiment eliminated the possibility that the α -selectivity was due to steric effects alone. Second, the glycosidation of xylylene-protected donor 12, which contains a nine-membered ring, gave the β -linked disaccharide predominantly ($\alpha/\beta = 8.92$). We then compared acyclic-type donors 11 and 13 with 10 and 12, respectively. 4,6-Di-O-TBS-protected donor 11 gave stereoselectivity similar to that of 10, and 4,6-di-O-Bn-protected donor 13 gave stereoselectivity nearly identical to that of 12. In additional experiments, 4,6-di-O-Me-protected donor 14 with less steric bulk and donor 15 with acyl protecting groups on both O4 and O6 were examined. As expected based on reported findings in carbohydrate chemistry, both the donors showed a strong preference for the formation of the β-product. Finally, 4,6-O-DTBS-protected GlcN donor 16 was coupled with 23 under reaction conditions identical to those used with the GalN donors.

The corresponding disaccharide 39 was obtained exclusively as the β -product with perfect stereoselectivity. From these experiments using donors 10-16, we can conclude that just the presence of a silicon atom and a cyclic structure was not sufficient for α -selectivity. The requirements for marked α -selectivity were a silylene structure forming a six-membered ring and distinct steric hindrance due to bulky substituents.

Encouraged by the results of these experimental studies, we next shifted our attention to a

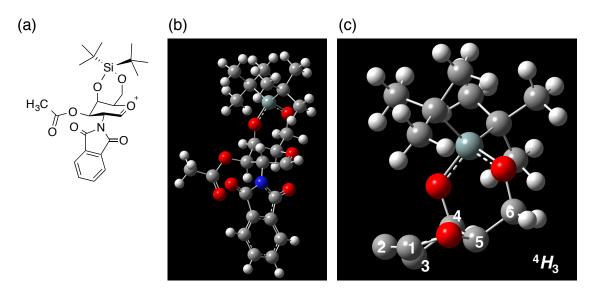


Figure 5. Optimized structure of the oxocarbenium intermediate liberated from a DTBS-protected galactosamine donor obtained by DFT calculations (M062X/6-31G*). (a) Chemical structure produced in ChemDraw. (b) Ball-and-stick representation of the whole structure (c) Ball-and-stick representation of the sugar ring in a partial structure.

computational study to examine the geometry of the postulated oxocarbenium intermediates.

Density functional theory (DFT) calculations were performed to optimize the structures. Figure 5 shows the optimized structure of the oxocarbenium intermediate formed from phenyl 3-*O*-acetyl-2-deoxy-4,6-*O*-di-*tert*-butylsilylene-2-phthalimide-1-thio-β-D-galactoside.

The DFT-optimized geometry revealed that the sugar ring with the DTBS group at the C4 and C6

positions adopts the 4H_3 conformation in the oxocarbenium intermediate. We also calculated the Mulliken and natural charges on atoms in selected glycosyl donors, considering that the electronegativity of oxygen atoms at C4 and C6 might affect the through-space electron donation. As shown in Figure 6, oxygen atoms next to a silicon atom (1, 2, 10, and 11) have a higher electron density compared with oxygen atoms next to a carbon atom (3 and 6). This result suggests the stronger through-space electron donation from oxygen atoms next to a silicon atom means that substituents on O4 and/or O6 are closer to the anomeric center, leading to increased formation of the α -product.

Taking together the experimental results and the computational results, we next propose an

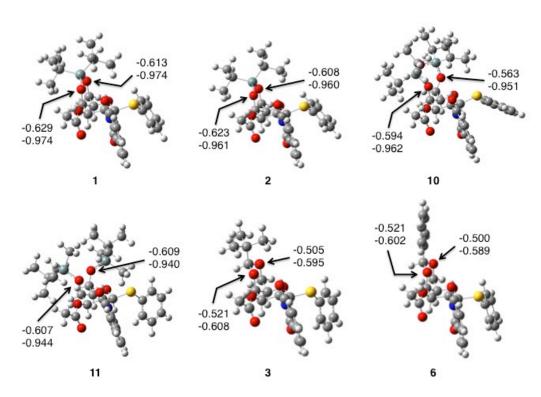


Figure 6. Calculated Mulliken charges (top numbers) and natural charges (bottom numbers) on O4 and O6 of glycosyl donors **1**, **2**, **3**, **6**, **10**, and **11**.

improved reaction mechanism for the DTBS-directed α-galactosylation (Figure 7). First, upon

activation of the phenylthioglycoside under the NIS-TfOH promoter system, the buildup of positive charge on the sulfur atom requires electrons to be stabilized. When the sugar ring adopts the 4C_1 conformation, there is inefficient overlap to provide electron density from O5 to the C1 antibonding orbital. In contrast, when the ring adopts the 1,4B conformation, the anti-periplanar electron pair on O5 can participate as in the E1 elimination reaction. Considering steric effects alone, there is not enough flexibility to provide access to the ^{1,4}B conformation from the 4C_1 conformation due to steric interactions between the pseudoaxial sulfonium ion and the bulky DTBS group. However, a conformational flip from the 4C_1 to the ${}^{1,4}B$ could be explained by stereoelectronic effects. Subsequent departure of the benzenesulfenyl iodide (ultimately giving PhSSPh and I₂) generates the oxocarbenium intermediate. Importantly, although typical pyranosyl oxocarbenium intermediates can usually as adopt a variety of conformations, such as ${}^{4}E$, E_{3} , ${}^{4}H_{3}$, ${}^{3}H_{4}$, ${}^{1}S_{5}$, and $B_{2.5}$, 25 here the conformational freedom of the sugar ring is limited to the half-chair 4H_3 conformer because other conformers are unattainable because of the fused ring system formed by the DTBS group. Once the oxocarbenium intermediate is generated, the C2 phthalate group will interact with the electron-poor anomeric center through neighboring group participation, thus producing a dioxolenium ion. A preliminary computational study suggested that no reaction pathway could exist without neighboring group participation due to the phthalate group. Usually, 1,2-trans glycosides are produced by nucleophilic attack by an alcohol acceptor from β-face onto this sort of dioxolenium ion. In the DTBS-directed galactosylation, however, the dioxolenium ion would be cleaved by through-space electron donation because both O4 and O6 are in an axial orientation. In other words, the coplanar alignment of the O4-C4 and O6-C6 bonds with the empty p-orbital

generated on C1 inevitably causes electrostatic attraction between the partially negatively charged oxygen atom of the C4 alkoxy groups and the positively charged carbon atom of the oxocarbenium ion. The Woerpel, 26 Bols, 27 and Crich 28 groups have independently reported the importance of the electrostatic effect of both the axially oriented C4 alkoxy and the exocyclic C5 alkoxymethyl groups in stabilizing the oxocarbenium ion. Breaking the five-membered ring in the dioxolenium ion would re-form the oxocarbenium ion in the ⁴H₃ conformer, which can then undergo nucleophilic attack by the alcohol acceptor, predominantly from the exo side (α -face) because *endo* attack on the β -face is blocked by the substantial steric hindrance of the tert-butyl group. Endo or exo nucleophilic attack can determine the stereochemical outcome in the glycosylation. First, endo attack and exo attack generate the distinct exploded transition states A and B, respectively. Next, transition states A and B can access a twist-boat-like conformer (C) and a chair-like conformer (D), respectively, to maximize orbital overlap between the approaching alcohol acceptor and the developing lone pair on oxygen, resulting in glycosidated products where the resulting sp³-hydridized orbitals are anti-coplanar.²⁹ The chair-like conformer (D) is more stable than C, which is in a kinetically disfavored twist-boat-like conformation that must undergo a ring-flip to reach the more stable equatorial β-product. Moreover, C clearly suffers from unfavorable steric clash between the tert-butyl substituent and the incoming alcohol acceptor. Hence, the nucleophilic attack predominantly occurs via the pathway leading to the α -product. Importantly, the steric hindrance of the DTBS group can hamper the β-face approach of even the smallest alcohol, methanol, giving rise to predominantly the α-linked methylglycoside. This study has demonstrated that the preferential formation of the ⁴H₃ oxocarbenium intermediate due to the conformational constraints imposed

by the DTBS group and subsequent access to the kinetically stable chair-like transition state provide high stereoselectivity. Indeed, the powerful steric effect of the DTBS group is the most important factor in the observed α -selectivity. Whitfield and co-workers have reported that, to control the stereochemistry of glycosylations, glycosyl donors that can form face-discriminated cations and access only one ring conformation should be used. Furthermore, they proposed that bridged donors are one possible approach. Our results support their account of stereoselective glycosylation.

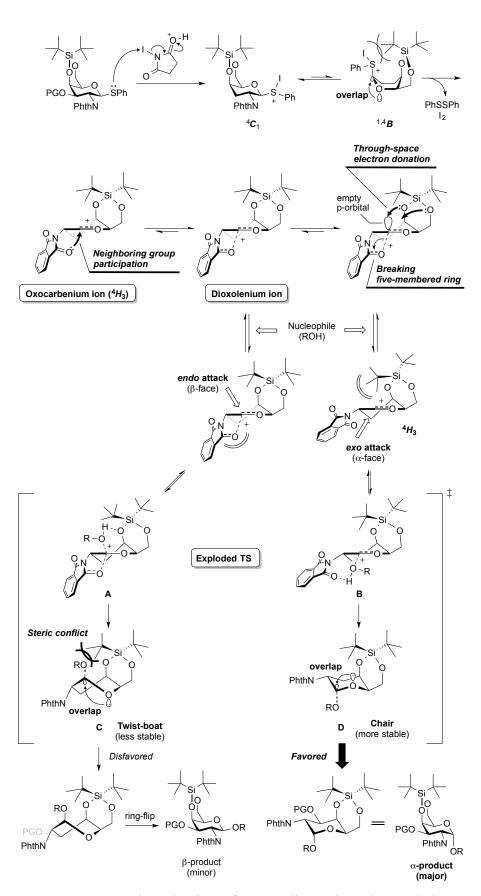


Figure 7. Proposed mechanism of DTBS-directed α -galactosylation.

Figure 8. Proposed mechanism of β -selective glycosidation using a glycosyl bromide donor with activation by insoluble silver silicate.

We have previously made two intriguing findings regarding DTBS-directed α -galactosylation. One is the exceptional case where β -selectivity (6:77, α/β ratio) is obtained by using glycosyl bromide donor with activation by insoluble silver silicate. This remarkable result can be rationalized by an S_N2 -like reaction mechanism. As illustrated in Figure 8, the reaction proceeds on insoluble silver silicate and the nucleophilic attack of the alcohol acceptor mainly occurs along an equatorial trajectory via an S_N2 -like pathway where the oxocarbenium ion is formed but remains linked to silicate anionic surface, in such a way that the nucleophile (HO-R) can only attack by/through the β -face, predominantly forming the β -glycoside. This result suggests that the generation of the "distinct" oxocarbenium ion is required for the stereo-directing effect of the DTBS group.

Our second intriguing finding is the formation of 2-OH α -galactoside as a byproduct when 2-O-benzoyl-protected galactosyl donors were used. As expected, the DTBS-protected

galactosyl donors gave predominantly α-glycosides in >70% yield. However, non-negligible amounts (~15%) of the corresponding 2-OH α-galactosides were also observed as by-products (Figure 9). This phenomenon could be explained by a route involving the formation of an orthoester derivative. Usually, the benzoyl group is selected instead of the acetyl group to prevent the formation of the orthoester and it works well in most cases. However, coexistence of the benzoyl and DTBS group may make access of the alcohol acceptor to anomeric center at the oxocarbenium ion stage limited due to the steric clash, allowing for the nucleophilic attack at relatively less hindered carbonyl carbon on the benzoyl group at C2 to form the orthoester derivative. In addition, nucleophilic attack by the alcohol acceptor at the carbonyl carbon C2, giving the orthoester, could occur from the exo side because of the higher energy barrier to the endo attack.³⁰ Some of the orthoester formed could then be cleaved to the 2-OH oxocarbenium ion to release the benzoylated acceptor and form the 2-OH oxocarbenium ion, which is then glycosidated with the acceptor to furnish the 2-OH \alpha-glycoside. We have not detected the benzoylated acceptor in the reaction mixture so far but the formation of orthoester derivatives has been actually observed. When 2-O-benzyl-protected DTBS-protected galactosyl donors were used, the corresponding α -glycosides were obtained in excellent yields without the formation of the 2-OH byproduct.³²

Figure 9. Proposed mechanism for the formation of 2-OH derivatives in DTBS-directed α -glycosidation using 2-O-benzoyl-protected galactosyl donors.

Conclusions

By experimental and computational studies, we have elucidated the origin of the α -selectivity in DTBS-directed α-galactosylation. Glycosidation experiments using various GalN donors, which differed in terms of only the protecting group at the C4 and C6 positions, demonstrated that the presence of the DTBS group was essential for strong α-selectivity. The following main requirements were found for α -selectivity. (1) Generation of the oxocarbenium ion:³³ A well-defined S_N1-like mechanism that produces the oxocarbenium ion results in the electron-poor anomeric center necessary for modulating the stereoselectivity. (2) Galacto-type donors with cyclic protecting group bridging O4 and O6 to form six-membered ring: These donors preferentially adopt the 4H_3 conformation upon formation of the oxocarbenium ion, allowing for nucleophilic attack by the glycosyl acceptor preferentially from exo side (α -face) because of both stereoelectronic and steric effects. (3) Through-space electron donation: The axially oriented C4 alkoxy and the exocyclic C5 alkoxy groups exert a powerful electrostatic effect on the empty p-orbital that is formed after liberation of the leaving group, thus stabilizing the oxocarbenium ion. Moreover, the effect is strengthened by the C4-O4 and C6-O6 bonds being coplanar with the empty p-orbital. This effect is maximized when the donor has a cyclic protecting group enabling the formation of a six-membered ring. (4) Steric hindrance due to bulky alkyl substituents appended to cyclic protecting group: Steric bulk hinders nucleophilic attack on the β-face and also affects the stability in the transition state. (5) 4,6-O-Silylene structure: X-ray crystallographic analysis of the DTBS donor showed that one of the bulky alkyl groups in the silylene structure is positioned very close to the anomeric carbon, even in the ground state. The substituent can be positioned much closer to the anomeric carbon in the oxocarbenium intermediate, resulting in severe steric clash during the glycosylation (this is supported by the DFT calculations).

The DTBS group fulfills all these requirements well. These characteristics of the DTBS group may explain the high stereoselectivity of DTBS-directed α -galactosylation. Furthermore, of all the cyclic protecting groups examined, only the DTBS group provided strong α -selectivity. Thus, we suggest that the unique and specific effect of the DTBS group in α -selective galactosylation be called the "DTBS effect". Encouraged by the discovery of the DTBS effect, we are working on the development of stereoselective glycosylations using other ring-restricted glycosyl donors, the results of which will be reported in due course.

Experimental Section

General procedure for chemical synthesis: All reactions were carried out under a positive pressure of argon unless otherwise noted. All chemicals were purchased from commercial suppliers and used as received without further purification. Molecular sieves were dried at 300 °C for 2 h in a muffle furnace prior to use. Solvents for reaction media were dried over molecular sieves and used without purification. Thin-layer chromatography (TLC) was performed on TLC plates (silica gel 60F₂₅₄ on glass plate). Compounds were detected on TLC plates by either exposure to UV light (2536 Å) or development in 10% H₂SO₄ in ethanol followed by heating. Silica gel (80 mesh and 300 mesh) was used for flash column chromatography. The quantity of silica gel was usually 100 to 200-fold the weight of the crude sample to be purified. Solvent systems for chromatography are specified in volume ratios.

Evaporation and concentration were carried out *in vacuo*. 1 H-NMR and 13 C-NMR spectra were recorded on a 400/500/600 MHz spectrometer. Chemical shifts in 1 H-NMR spectra are expressed in ppm (δ) relative to the Me₄Si signal adjusted to δ 0.00 ppm. Data are presented as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, dd = double doublet, td = triple doublet, m = multiplet and/or multiple resonances), coupling constant in hertz (Hz), integration, and position of the corresponding proton. 1 H- 1 H COSY method was used to confirm the NMR peak assignments. High-resolution mass spectra (ESI-TOF MS) were obtained. Optical rotations were measured with a high-sensitivity polarimeter.

Computational details: All DFT calculations were carried out in the GAUSSIAN09 software package.³⁴ We chose the M062X exchange-correlation functional³⁵ and the 6-31G* electronic basis set for calculations. Initial structures for the calculations were built with the aid of GaussView 5 visualization software.³⁶ To investigate the desired reactions, we constructed a six-membered ring as chair-like form in the initial structure. The geometries of complexes were fully optimized at the M062X/6-31G* level. We also performed normal mode analysis to characterize the optimized structures and confirmed that all optimized structures have no imaginary frequencies.

Typical procedure for glycosidation experiments: Molecular sieves, 4Å (100 mg per 1 mL of solvent) were added to a mixture of the glycosyl donor (1.5 equiv with respect to the acceptor) and acceptor 23 in CH₂Cl₂ at room temperature. After stirring for 30 min, the mixture was cooled to 0 °C and *N*-iodosuccinimide (2.0 equiv with respect to the donor) and TfOH (0.2 equiv with respect to the donor) were added. The mixture was stirred for 30 min to 1 h while monitoring the reaction by TLC. Once all the acceptor was consumed, the reaction was quenched by the addition of saturated aq. NaHCO₃. The reaction mixture was diluted with CHCl₃ and filtered through Celite. The filtrate was then washed sequentially with saturated aq. NaHCO₃, saturated aq. Na₂S₂O₃, and brine. The organic layer was collected, dried over Na₂SO₄, and concentrated. The resulting residue was purified by silica gel column chromatography to give the glycosylated products.

Phenyl

3-*O*-benzyl-2-deoxy-4,6-*O*-di-*tert*-butylsilylene-2-phthalimide-1-thio-β-D-galactopyranosid **e** (1). To a solution of **18** (200 mg, 0.407 mmol) in pyridine (2.0 mL) was added di-*tert*-butylsilyl bis(trifluoromethanesulfonate) (223 μL, 0.611 mmol) at 0 °C. After stirring for 1 h at rt, the completion of the reaction was confirmed by TLC (1:3 EtOAc–*n*-hexane). The reaction was quenched by the addition of dry methanol at 0 °C. Solvents were removed by co-evaporation with toluene, and then the residue was diluted with CHCl₃, washed with 2M HCl, H₂O, satd aq NaHCO₃, and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:5 EtOAc–*n*-hexane) to give **1** (236 mg, 92%) as a colorless viscous

compound. [α]_D +112.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.87–7.07 (m, 14 H, NPhth, 2 Ph), 5.54 (d, $J_{1,2}$ = 10.5 Hz, 1 H, 1-H), 4.87 (t, $J_{2,3}$ = 10.7 Hz, 1 H, 2-H), 4.64 (d, J_{gem} = 12.3 Hz, 1 H, PhC H_2), 4.60 (d, $J_{3,4}$ = 2.7 Hz, 1 H, 4-H), 4.44 (d, 1 H, PhC H_2), 4.27 (dd, 1 H, 3-H), 4.25 (m, 2 H, 6a-H, 6b-H), 3.50 (s, 1 H, 5-H), 1.20, 1.07 (2 s, 18 H, 2 ^tBu); ¹³C NMR (100 MHz, CDCl₃): δ = 168.4, 167.6, 137.9, 134.1, 133.9, 133.6, 132.0, 131.8, 128.8, 128.3, 127.6, 127.4, 123.6, 123.1, 84.9, 75.6, 75.2, 70.0, 69.4, 67.5, 50.7, 27.7, 27.6, 23.5, 20.8. HRMS (ESI-TOF): m/z calcd for C₃₅H₄₁NO₆SSi: 654.2316 [M+Na]⁺; found: 654.2318.

Phenyl

3-O-benzyl-2-deoxy-4,6-O-diisopropylsilylene-2-phthalimide-1-thio-β-D-galactopyranoside

(2). To a solution of **18** (187 mg, 0.380 mmol) in DMF (3.8 mL) were added diisopropylsilyl bis(trifluoromethanesulfonate) (170 μ L, 0.580 mmol) and triethylamine (370 μ L, 2.66 mmol) at 0 °C. After stirring for 10 min at rt, the consumption of the starting material was confirmed by TLC (2:3 EtOAc–*n*-hexane). The reaction was quenched by the addition of dry methanol at 0 °C. Solvents were removed by co-evaporation with toluene, and then the residue was diluted with CHCl₃, washed with H₂O and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:5 EtOAc–*n*-hexane) to give **2** (198 mg, 86%) as a colorless viscous compound. [α]_D +108.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.90–6.99 (m, 14 H, NPhth, 2 Ph), 5.53 (d, J_{1,2} = 10.5 Hz, 1 H, 1-H), 4.80 (t, J_{2,3} = 10.6 Hz, 1 H, 2-H), 4.63 (d, J_{gem} = 12.4 Hz, 1 H, PhCH₂), 4.51 (d, J_{3,4} = 2.7 Hz, 1 H, 4-H), 4.43 (d, 1 H, PhCH₂), 4.27–4.17 (m, 3 H, 3-H, 6a-H, 6b-H), 3.53 (s, 1 H, 5-H), 1.26–1.03 (m, 14 H, 2 ⁱPr); ¹³C NMR (100 MHz, CDCl₃): δ = 168.4, 167.4, 137.8, 134.0, 133.8, 132.9, 132.4, 131.7, 128.7, 128.2, 127.6, 127.6,

127.5, 123.6, 123.1, 84.4, 75.8, 75.3, 70.4, 69.2, 66.9, 50.7, 29.7, 17.1, 17.0, 16.9, 13.9, 12.7.

HRMS (ESI-TOF): *m/z* calcd for C₃₃H₃₇NO₆SSi: 626.2003 [*M*+Na]⁺; found: 626.2002.

Phenyl

3-O-benzyl-4,6-O-tert-butylmethylidene-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranosi de (3). To a solution of 18 (50 mg, 0.105 mmol) in MeCN (349 μL) were added pivalaldehyde (46 μL, 0.418 mmol) and (±)-camphor-10-sulfonic acid (CSA) (5 mg, 21.0 μmol) at rt. After stirring for 7.5 h at rt as the reaction was monitored by TLC (1:2 EtOAc-n-hexane, 1:15 MeOH-CHCl₃), the reaction was quenched by the addition of triethylamine at 0 °C and the reaction mixture was then evaporated. The resulting residue was purified by silica gel column chromatography (1:5 EtOAc-n-hexane) to give 3 (49 mg, 84%) as a white foamy compound. $[\alpha]_D + 2.3$ (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.85 - 7.07$ (m, 14 H, NPhth, 2 Ph), 5.53 (d, $J_{1,2} = 10.4$ Hz, 1 H, 1-H), 4.70 (t, $J_{2,3} = 10.7$ Hz, 1 H, 2-H), 4.54 (d, $J_{gem} = 12.4$ Hz, 1 H, $PhCH_2$), 4.38 (d, 1 H, $PhCH_2$), 4.35 (dd, $J_{3,4} = 3.5$ Hz, 1 H, 3-H), 4.27 (dd, $J_{gem} = 12.2$ Hz, $J_{5,6a} = 12.2$ Hz, 1.4 Hz, 1 H, 6a-H), 4.14 (s, 1 H, ${}^{t}BuCH <$), 4.05 (d, 1 H, 4-H), 3.81 (dd, $J_{5,6b} = 1.7$ Hz, 1 H, 6b-H), 3.49 (d, 1 H, 5-H), 0.99 (s, 9 H, 'Bu); 13 C NMR (125 MHz, CDCl₃): $\delta = 168.8$, 167.0, 138.0, 134.0, 133.8, 132.8, 131.8, 131.8, 131.6, 128.7, 128.1, 127.7, 127.5, 127.3, 123.6, 123.1, 107.1, 83.0, 74.9, 72.1, 70.6, 69.2, 50.6, 35.2, 29.7, 24.7. HRMS (ESI-TOF): m/z calcd for $C_{32}H_{33}NO_6S$: 582.1921 [M+Na]⁺; found: 582.1922.

Phenyl 3-O-benzyl-2-deoxy-4,6-O-ethylidene-2-phthalimide-1-thio-β-D-galactopyranoside (4). To a solution of **18** (281 mg, 0.572 mmol) in MeCN (1.9 mL) were added dimethylacetal (182 μL, 1.72 mmol) and CSA (13 mg, 57.0 μmol) at rt. After stirring for 2.5 h at rt as the reaction was monitored by TLC (1:30 MeOH–CHCl₃), the reaction was quenched by the

addition of triethylamine at 0 °C and the reaction mixture was then evaporated. The resulting residue was purified by silica gel column chromatography (1:2 \rightarrow 2:3 EtOAc–n-hexane) to give 4 (273 mg, 92%) as a white foamy compound. [α]_D –105.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.71–7.00 (m, 14 H, NPhth, 2 Ph), 5.54 (d, $J_{1,2}$ = 10.4 Hz, 1 H, 1-H), 4.75–4.68 (m, 2 H, 2-H, CH₃CH<), 4.59 (d, J_{gem} = 12.6 Hz, 1 H, PhC H_2), 4.37 (d, 1 H, PhC H_2), 4.32 (dd, $J_{3,4}$ = 3.4 Hz, $J_{2,3}$ = 10.7 Hz, 1 H, 3-H), 4.23 (dd, $J_{5,6a}$ = 1.5 Hz, J_{gem} = 12.2 Hz, 1 H, 6a-H), 4.07 (d, 1 H, 4-H), 3.83 (dd, $J_{5,6b}$ = 1.2 Hz, 1 H, 6b-H), 3.48 (d, 1 H, 5-H), 1.46 (d, $J_{CH3CH<CH3CH<}$ = 5.0 Hz, 3 H, CH_3 CH<); ¹³C NMR (125 MHz, CDCl₃): δ = 168.5, 167.1, 137.6, 137.6, 134.0, 133.8, 132.8, 132.2, 131.8, 131.7, 128.7, 128.2, 128.0, 127.7, 127.7, 123.6, 123.1, 99.1, 83.5, 74.7, 72.1, 71.0, 70.1, 69.0, 50.7, 29.7, 21.1. HRMS (ESI-TOF): m/z calcd for C_{29} H₂₇NO₆S: 540.1451 [M+Na]⁺; found: 540.1453.

Phenyl

3-O-benzyl-2-deoxy-4,6-O-methylidene-2-phthalimide-1-thio-β-D-galactopyranoside (5).

To a solution of **18** (200 mg, 0.407 mmol) in MeCN (814 μ L) were added dimethoxymethane (2.53 mL, 28.5 mmol) and CSA (19 mg, 81.4 μ mol) at rt. After stirring for 14 h at rt as the reaction was monitored by TLC (1:30 MeOH–CHCl₃), the reaction was warmed to 40 °C and the stirring was continued for 10 h at the same temperature. Then the reaction was warmed to 60 °C and the stirring was continued for 10 h at the same temperature. The reaction was quenched by the addition of triethylamine at 0 °C and the reaction mixture was then evaporated. The resulting residue was purified by silica gel column chromatography (1:1 EtOAc–n-hexane) to give **5** (173 mg, 84%) as a white foamy compound. [α]_D +0.2 (c 4.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.88–7.00 (m, 14 H, NPhth, 2 Ph), 5.56 (d, $J_{1,2}$ = 10.5 Hz, 1 H, 1-H), 5.28 (d,

Phenyl 3-O-benzyl-4,6-O-benzylidene-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranoside (6). To a solution of 17 (14.3 g, 35.5 mmol) in MeCN (71.0 mL) were added benzaldehyde dimethyl acetal (BDA) (5.83 mL, 39.1 mmol) and CSA (411 mg, 1.77 mmol) at 0 °C. After stirring for 1.5 h at rt as the reaction was monitored by TLC (1:1 EtOAc-toluene), the reaction was quenched by the addition of triethylamine at 0 °C and the reaction mixture was then evaporated. The resulting residue was purified by silica gel column chromatography $(1:3\rightarrow 1:1$ EtOAc-toluene) to give the 4,6-O-benzylidenated product (15.4 g, 31.6 mmol), which was subsequently dissolved in DMF (158 mL). To the mixture was added BnBr (9.41 mL, 79.1 mmol) at rt. After cooling to 0 °C, NaH (60% in oil; 2.53 g, 63.3 mmol) was added to the mixture at the same temperature. After stirring for 2.5 h at rt as the reaction was monitored by TLC (2:3 EtOAc-n-hexane, 1:10 EtOAc-toluene), the reaction was quenched by the addition of triethylamine and satd aq NH₄Cl at 0 °C. The reaction mixture was co-evaporated with toluene. The residue was diluted with EtOAc and washed with H₂O, satd aq NaHCO₃, and brine. The organic phase was dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (2:7 EtOAc-n-hexane) to give 6 (14.2 g, 69%) over two steps) as a white foamy compound, which was then recrystallized from EtOAc*n*-hexane system. M.p. = 211 to 216 °C. [α]_D +48.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.86–7.01 (m, 19 H, NPhth, 3 Ph), 5.58 (d, $J_{1,2}$ = 10.5 Hz, 1 H, 1-H), 5.51 (s, 1 H, PhC*H*<), 4.77 (t, $J_{2,3}$ = 10.6 Hz, 1 H, 2-H), 4.59 (d, J_{gem} = 12.4 Hz, 1 H, PhC*H*₂), 4.42–4.37 (m, 3 H, 3-H, 6a-H, PhC*H*₂), 4.26 (d, $J_{3,4}$ = 3.2 Hz, 1 H, 4-H), 4.03 (d, J_{gem} = 11.9 Hz, 1 H, 6b-H), 3.60 (s, 1 H, 5-H); ¹³C NMR (100 MHz, CDCl₃): δ = 168.5, 167.0, 137.7, 134.0, 133.8, 133.3, 131.7, 131.7, 131.4, 129.0, 128.7, 128.1, 127.8, 127.6, 126.6, 123.6, 123.2, 123.1, 122.9, 101.2, 82.9, 74.7, 72.6, 70.9, 70.4, 70.1, 69.9, 69.5, 50.7, 50.5. HRMS (ESI-TOF): m/z calcd for C₃₄H₂₉NO₆S: 602.1608 [M+Na]⁺; found: 602.1606.

Phenyl

3-*O*-benzyl-2-deoxy-4,6-*O*-(4-nitrobenzylidene)-2-phthalimide-1-thio-β-D-galactopyranosid e (7). To a solution of 18 (200 mg, 406 μmol) in DMF (810 μL) were added 4-nitrobenzaldehyde (307 mg, 2.03 mmol) and CSA (63 mg, 0.203 mmol) at rt. After stirring for 2 h at 100 °C as the reaction was monitored by TLC (1:1 EtOAc–*n*-hexane), the reaction mixture was evaporated. The resulting residue was purified by silica gel column chromatography (2:3 EtOAc–*n*-hexane) to give 7 (238 mg, 94%) as a colorless foamy compound. [α]_D +43.0 (*c* 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 8.24–7.01 (m, 18 H, Ar), 5.58 (d, $J_{1,2}$ = 10.4 Hz, 1 H, 1-H), 5.56 (s, 1 H, ArCH<), 4.70 (t, $J_{2,3}$ = 10.5 Hz, 1 H, 2-H), 4.58 (d, J_{gem} = 12.7 Hz, 1 H, PhC H_2), 4.44–4.40 (m, 3 H, 3-H, 6a-H, PhC H_2), 4.27 (d, $J_{3,4}$ = 3.2 Hz, 1 H, 4-H), 4.05 (dd, $J_{5,6b}$ = 1.5 Hz, J_{gem} = 12.2 Hz, 1 H, 6b-H), 3.64 (s, 1 H, 5-H); ¹³C NMR (125 MHz, CDCl₃): δ = 168.5, 167.1, 148.3, 144.0, 137.5, 134.1, 133.9, 133.6, 131.7, 131.2, 128.7, 128.2, 128.1, 127.8, 127.7, 123.7, 123.4, 123.2, 99.6, 83.0, 74.6, 72.7, 71.2, 69.9, 69.6, 50.7. HRMS (ESI-TOF): m/z calcd for $C_{34}H_{28}N_{2}O_{8}S$: 647.1459 [M+Na]⁺; found: 647.1459.

Phenyl 4,6-*O***-anisylidene-3-***O***-benzyl-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranoside (8).** To a solution of 18 (300 mg, 610 μmol) in MeCN (1.2 mL) were added anisaldehyde dimethyl acetal (153 μL, 915 μmol) and CSA (14 mg, 61.0 μmol) at rt. After stirring for 4 h at rt as the reaction was monitored by TLC (1:1 EtOAc–*n*-hexane), the reaction was quenched by the addition of triethylamine at 0 °C. The reaction mixture was then evaporated. The resulting residue was purified by silica gel column chromatography (1:100 acetone–CHCl₃) to give 8 (331 mg, 89%) as a colorless foamy compound. [α]_D +63.3 (*c* 4.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.87–6.89 (m, 18 H, Ar), 5.58 (d, $J_{1,2}$ = 10.4 Hz, 1 H, 1-H), 5.46 (s, 1 H, ArCH<), 4.76 (t, $J_{2,3}$ = 10.5 Hz, 1 H, 2-H), 4.58 (d, J_{gem} = 12.7 Hz, 1 H, PhCH₂), 4.40–4.36 (m, 3 H, 3-H, 6a-H, PhCH₂), 4.25 (d, $J_{3,4}$ = 3.2 Hz, 1 H, 4-H), 4.18 (dd, $J_{5,6b}$ = 1.5 Hz, J_{gem} = 12.2 Hz, 1 H, 6b-H), 3.84 (s, 3 H, OMe), 3.59 (d, 1 H, 5-H); ¹³C NMR (125 MHz, CDCl₃): δ = 168.6, 167.1, 160.2, 137.8, 134.0, 133.8, 133.4, 131.8, 131.7, 131.5, 130.5, 128.7, 128.2, 128.0, 127.9, 127.6, 123.6, 123.1, 113.6, 101.2, 83.0, 74.8, 72.7, 71.0, 70.2, 69.5, 55.4, 50.7. HRMS (ESI-TOF): m/z calcd for C₃₅H₃₁NO₇S: 632.1713 [M+Nal⁺; found: 632.1714.

Phenyl

3-*O*-benzyl-4,6-*O*-(2-chloroethylidene)-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranosid e (9). To a solution of **18** (49 mg, 100 μmol) in MeCN (500 μL) were added 2-chloroacetaldehyde diethyl acetal (23 μL, 150 μmol) and CSA (5 mg, 20.0 μmol) at rt. After stirring for 20 h at rt as the reaction was monitored by TLC (1:2 EtOAc–*n*-hexane, 1:15 MeOH–CHCl₃), the reaction was warmed to 70 °C and the stirring was continued for 7 h at the same temperature. The reaction was quenched by the addition of triethylamine at 0 °C and the reaction mixture was then evaporated. The resulting residue was purified by silica gel column

chromatography (2:3 EtOAc–n-hexane) to give **9** (41 mg, 74%) as a white foamy compound. [α]_D +0.1 (c 1.3, CHCl₃); 1 H NMR (500 MHz, CDCl₃): δ = 7.89–7.07 (m, 14 H, NPhth, 2 Ph), 5.53 (d, $J_{1,2}$ = 10.4 Hz, 1 H, 1-H), 4.74 (dd, $J_{\text{CICH2CH}<,\text{CICH2CH}<}$ = 3.7, 6.0 Hz, 1 H, CICH₂CH<), 4.65 (t, $J_{2,3}$ = 10.4 Hz, 1 H, 2-H), 4.57 (d, J_{gem} = 12.6 Hz, 1 H, PhC H_2), 4.40 (d, 1 H, PhC H_2), 4.36 (dd, $J_{3,4}$ = 3.4 Hz, 1 H, 3-H), 4.30 (dd, $J_{5,6a}$ = 1.4 Hz, J_{gem} = 12.3 Hz, 1 H, 6a-H), 4.09 (d, 1 H, 4-H), 3.87 (dd, $J_{5,6b}$ = 1.6 Hz, 1 H, 6b-H), 3.66, 3.60 (2 dd, J_{gem} = 11.5 Hz, 2 H, CICH₂CH<), 3.53 (d, 1 H, 5-H); 13 C NMR (125 MHz, CDCl₃): δ = 168.5, 167.1, 137.5, 134.1, 133.9, 133.1, 131.8, 131.7, 131.7, 128.7, 128.2, 128.0, 127.8, 127.7, 123.7, 123.2, 100.4, 83.5, 74.5, 72.4, 71.1, 69.9, 69.2, 50.6, 44.0. HRMS (ESI-TOF): m/z calcd for $C_{29}H_{26}CINO_6S$: 574.1062 [M+Na]⁺; found: 574.1065.

Phenyl

3-*O*-benzyl-2-deoxy-2-phthalimide-4,6-*O*-1',1',3',3'-tetraisopropyldisiloxanylidene-1-thio-β -**D**-galactopyranoside (10). To a solution of 18 (301 mg, 0.612 mmol) in DMF (6.1 mL) were added 1,3-dichloro-1',1',3',3'-tetraisopropyldisiloxane (215 μL, 0.671 mmol), imidazole (183 mg, 2.68 mmol) at 0 °C. After stirring for 1 h at rt as the reaction was monitored by TLC (2:7 EtOAc–*n*-hexane), the reaction was quenched by the addition of dry methanol at 0 °C. The reaction mixture was co-evaporated with toluene and then diluted with CHCl₃, washed with satd aq NaHCO₃ and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:10 EtOAc–*n*-hexane, 1:6 diethyl ether–*n*-hexane) to give 10 (380 mg, 88%) as a colorless viscous compound. [α]_D +53.8 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.71–6.81 (m, 14 H, NPhth, 2 Ph), 5.45 (d, $J_{1,2}$ = 10.1 Hz, 1 H, 1-H), 4.62 (t, $J_{2,3}$ = 10.5 Hz, 1 H, 2-H), 4.46 (d, J_{gem} =

11.9 Hz, 1 H, PhC H_2), 4.33 (br d, 1 H, 4-H), 4.18 (br dd, 1 H, 3-H), 4.12 (d, 1 H, PhC H_2), 3.82–3.77 (m, 2 H, 6a-H, 6b-H), 3.66 (m, 1 H, 5-H), 1.02–0.90 (m, 28 H, 4 i Pr); 13 C NMR (100 MHz, CDCl₃): δ = 168.4, 167.2, 137.5, 133.8, 133.6, 132.2, 132.1, 131.7, 128.6, 128.0, 127.7, 127.5, 127.4, 123.3, 123.0, 83.2, 77.8, 71.9, 65.0, 59.3, 50.8, 17.5, 17.4, 17.3, 17.2, 17.2, 17.1, 16.9, 14.0, 13.2, 12.7, 12.5. HRMS (ESI-TOF): m/z calcd for $C_{39}H_{51}NO_7SSi_2$: 756.2817 [M+Na] $^+$; found: 756.2817.

Phenyl

3-O-benzyl-4,6-di-O-tert-butyldimethylsilyl-2-deoxy-2-phthalimide-1-thio-β-D-galactopyra noside (11). To a solution of 18 (201 mg, 0.409 mmol) in DMF (2.7 mL) were added tert-butyldimethylsilyl trifluoromethanesulfonate (560 µL, 2.44 mmol), imidazole (222 mg, 3.26 mmol), and 2,6-di-tert-butyl-4-methylpyridine (251 mg, 1.22 mmol) at 0 °C. After stirring for 135 min at rt as the reaction was monitored by TLC (1:3 EtOAc-n-hexane), the reaction was quenched by the addition of dry methanol at 0 °C. The reaction mixture was diluted with EtOAc, washed with H₂O, satd aq NaHCO₃, and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:10 EtOAc-n-hexane) to give 11 (281 mg, 96%) as a colorless viscous compound. $[\alpha]_D$ +37.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.71-6.80$ (m, 14 H, NPhth, 2 Ph), 5.45 (d, $J_{1,2} = 10.1$ Hz, 1 H, 1-H), 4.58 (t, $J_{2,3} = 10.6$ Hz, 1 H, 2-H), 4.50 (d, $J_{gem} =$ 11.9 Hz, 1 H, PhC H_2), 4.16 (d, $J_{3,4} = 1.8$ Hz, 1 H, 4-H), 4.11 (d, 1 H, PhC H_2), 4.07 (dd, 1 H, 3-H), 3.71–3.70 (m, 2 H, 6a-H, 6b-H), 3.50 (br t, $J_{5.6b} = 6.4$ Hz, $J_{5.6a} = 6.9$ Hz, 1 H, 5-H), 0.83 (s, 18 H, 2 ^tBu), 0.03–0.00 (m, 12 H, 2 SiMe₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 168.5$, 167.1, 137.4, 133.8, 133.6, 132.4, 131.9, 131.7, 128.6, 128.0, 127.6, 127.4, 127.4, 123.3, 123.1, 82.9,

79.9, 78.0, 72.1, 66.9, 61.6, 50.7, 26.0, 25.9, 18.5, 18.2, -4.3, -5.1, -5.3, -5.3. HRMS (ESI-TOF): *m/z* calcd for C₃₉H₅₃NO₆SSi₂: 742.3024 [*M*+Na]⁺; found: 742.3024.

3-O-benzyl-2-deoxy-2-phthalimide-1-thio-4,6-O-xylylene-β-D-galactopyranoside Phenyl To a solution of 18 (100 mg, 0.203 mmol) in DMF (2.0 mL) was added (12).α,α'-dibromo-o-xylene (134 mg, 0.509 mmol) at rt. After cooling to 0 °C, NaH (60% in oil; 20 mg, 0.505 mmol) was added to the mixture at 0 °C. After stirring for 1 h at rt as the reaction was monitored by TLC (2:3 EtOAc-n-hexane), the reaction was quenched by the addition of triethylamine and satd aq NH₄Cl at 0 °C. The reaction mixture was co-evaporated with toluene. The residue was diluted with EtOAc and washed with brine. The organic layer was dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:3 EtOAc-n-hexane) to give 12 (46 mg, 38%) as a colorless viscous compound. $[\alpha]_D$ +32.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.88-7.63$ (m, 4 H, Phth), 7.34–7.00 (m, 14 H, 3 Ar), 5.48 (d, $J_{1,2}$ = 10.5 Hz, 1 H, 1-H), 5.20 (d, J_{gem} = 12.5 Hz, 1 H, $ArCH_2$), 5.03 (d, 1 H, $ArCH_2$), 4.92 (d, $J_{gem} = 11.5$ Hz, 1 H, $ArCH_2$), 4.84 (d, 1 H, $ArCH_2$), 4.76 $(t, J_{2,3} = 10.5 \text{ Hz}, 1 \text{ H}, 2\text{-H}), 4.61 (d, J_{gem} = 12.5 \text{ Hz}, 1 \text{ H}, ArCH_2), 4.28 (dd, J_{3,4} = 3.0 \text{ Hz}, 1 \text{ H}, ArCH_2)$ 3-H), 4.23 (d, 1 H, ArC H_2), 4.11 (d, 1 H, 4-H), 4.09 (dd, $J_{5,6a} = 6.0$ Hz, $J_{gem} = 11.0$ Hz, 1 H, 6a-H), 3.78 (t, $J_{5,6b}$ = 6.5 Hz, 1 H, 5-H), 3.72 (dd, 1 H, 6b-H); ¹³C NMR (125 MHz, CDCl₃): $\delta = 168.3, 167.5, 137.5, 137.0, 136.5, 133.9, 133.8, 132.9, 132.1, 131.7, 130.8, 130.5, 128.7,$ 128.6, 128.5, 128.2, 127.6, 127.5, 123.5, 123.1, 84.2, 77.6, 76.7, 75.6, 74.6, 71.1, 71.0, 70.5, 51.8. HRMS (ESI-TOF): m/z calcd for $C_{35}H_{31}NO_6S$: 616.1764 $[M+Na]^+$; found: 616.1765.

Phenyl 3,4,6-tri-O-benzyl-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranoside (13). To a solution of 18 (200 mg, 0.407 mmol) in DMF (2.0 mL) was added BnBr (194 μL, 1.63 mmol) at

rt. After cooling to 0 °C, NaH (60% in oil; 49 mg, 1.22 mmol) was added to the mixture at 0 °C. After stirring for 2 h at rt as the reaction was monitored by TLC (2:3 EtOAc-n-hexane), the reaction was quenched by the addition of dry MeOH at 0 °C. The reaction mixture was co-evaporated with toluene. The residue was diluted with EtOAc and washed with H₂O and brine. The organic layer was dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:5 EtOAc-n-hexane) to give 13 (233 mg, 85%) as a colorless viscous compound. $[\alpha]_D$ +86.8 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.84-7.65$ (m, 4 H, Phth), 7.39–6.96 (m, 20 H, 4 Ph), 5.56 (d, $J_{1,2} = 10.5$ Hz, 1 H, 1-H), 4.96 (d, $J_{gem} = 12.5$ Hz, 1 H, PhC H_2), 4.84 (t, $J_{2,3} = 10.5$ Hz, 1 H, 2-H), 4.59 (d, $J_{gem} = 1.00$ 11.5 Hz, 1 H, PhC H_2), 4.57 (d, 1 H, PhC H_2), 4.48 (d, $J_{gem} = 12.0$ Hz, 1 H, PhC H_2), 4.43 (d, 1 H, $PhCH_2$), 4.35 (dd, $J_{3.4} = 2.5$ Hz, 1 H, 3-H), 4.29 (d, 1 H, $PhCH_2$), 4.07 (d, 1 H, 4-H), 3.81 (t, $J_{5.6a}$ = $J_{5.6b}$ = 6.5 Hz, 1 H, 5-H), 3.69 (d, 2 H, 6a-H, 6b-H); ¹³C NMR (125 MHz, CDCl₃): δ = 168.3, 167.4, 140.9, 138.5, 137.9, 137.5, 133.9, 133.7, 132.9, 131.8, 131.7, 128.8, 128.6, 128.5, 128.5, 128.3, 128.3, 128.1, 128.1, 128.0, 127.9, 127.8, 127.7, 127.6, 127.5, 127.4, 127.3, 126.9, 123.4, 123.1, 84.0, 77.6, 77.5, 74.4, 73.5, 72.2, 71.4, 68.8, 65.2, 51.6. HRMS (ESI-TOF): m/z calcd for $C_{41}H_{37}NO_6S: 694.2234 [M+Na]^+$; found: 694.2233.

Phenyl 3-*O*-benzyl-2-deoxy-4,6-di-*O*-methyl-2-phthalimide-1-thio-β-D-galactopyranoside (14). To a solution of 18 (300 mg, 0.610 mmol) in DMF (6.1 mL) were added iodomethane (191 μL, 3.05 mmol), NaH (60% in oil; 97 mg, 2.44 mmol) at 0 °C. After stirring for 1 h at rt as the reaction was monitored by TLC (1:1 EtOAc–*n*-hexane), the reaction was quenched by the addition of satd aq NH₄Cl and triethylamine at 0 °C. The reaction mixture was co-evaporated with toluene and then diluted with CHCl₃, washed with H₂O, satd aq NaHCO₃, and brine. The

organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:2 EtOAc–n-hexane) to give **14** (224 mg, 68%) as a colorless foamy compound. [α]_D +109.8 (c 2.8, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.86–6.97 (m, 14 H, NPhth, 2 Ph), 5.51 (d, $J_{1,2}$ = 10.6 Hz, 1 H, 1-H), 4.71 (t, $J_{2,3}$ = 10.6 Hz, 1 H, 2-H), 4.65 (d, J_{gem} = 12.2 Hz, 1 H, PhC H_2), 4.35–4.29 (m, 2 H, 3-H, PhC H_2), 3.81 (d, $J_{3,4}$ = 2.7 Hz, 1 H, 4-H), 3.74–3.66 (m, 2 H, 5-H, 6a-H), 3.62 (s, 3 H, OMe), 3.58 (dd, $J_{5,6b}$ = 5.4 Hz, J_{gem} = 9.2 Hz, 1 H, 6b-H), 3.56 (s, 3 H, OMe); ¹³C NMR (125 MHz, CDCl₃): δ = 168.3, 167.4, 137.5, 134.0, 133.8, 133.3, 131.9, 131.7, 128.7, 128.2, 127.7, 127.6, 127.4, 123.5, 123.1, 84.4, 74.4, 71.3, 70.7, 61.4, 59.3, 51.7. HRMS (ESI-TOF): m/z calcd for $C_{29}H_{29}NO_6S$: 542.1608 [M+Na]⁺; found: 542.1612.

Phenyl 4,6-di-*O*-acetyl-3-*O*-benzyl-2-deoxy-2-phthalimide-1-thio-β-D-galactopyranoside (15). To a solution of **18** (150 mg, 0.305 mmol) in pyridine (1.5 mL) were added acetic anhydride (288 μL, 3.05 mmol), DMAP (3.7 mg, 30.5 μmol) at 0 °C. After stirring for 30 min at rt as the reaction was monitored by TLC (2:3 EtOAc–n-hexane), the reaction was quenched by the addition of dry methanol at 0 °C. The reaction mixture was co-evaporated with toluene and then diluted with CHCl₃, washed with 2M HCl, H₂O, satd aq NaHCO₃, and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:3 EtOAc–n-hexane) to give **15** (176 mg, quant) as a colorless viscous compound. [α]_D +130.5 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.86–6.89 (m, 14 H, NPhth, 2 Ph), 5.61 (d, $J_{3,4}$ = 3.0 Hz, 1 H, 4-H), 5.55 (d, $J_{1,2}$ = 10.5 Hz, 1 H, 1-H), 4.56 (d, J_{gem} = 12.0 Hz, 1 H, PhC H_2), 4.52 (t, $J_{2,3}$ = 10.5 Hz, 1 H, 2-H), 4.34 (dd, 1 H, 3-H), 4.25–4.21 (m, 3 H, 6a-H, 6b-H, PhC H_2), 4.02 (t, $J_{5,6a}$ = $J_{5,6b}$ = 6.5 Hz, 1 H, 5-H), 2.18,

2.06 (2 s, 6 H, 2 Ac); ¹³C NMR (125 MHz, CDCl₃): δ = 170.3, 170.3, 167.8, 167.2, 137.0, 134.0, 133.8, 132.3, 132.2, 131.5, 131.4, 128.6, 128.0, 127.9, 127.7, 127.6, 123.4, 123.2, 83.9, 74.7, 73.4, 71.0, 65.6, 62.3, 51.3, 20.7, 20.6. HRMS (ESI-TOF): m/z calcd for C₃₁H₂₉NO₈S: 598.1506 [M+Na]⁺; found: 598.1507.

Phenyl

3-O-benzyl-2-deoxy-4,6-O-di-tert-butylsilylene-2-phthalimide-1-thio-β-D-glucopyranoside (16). Compound 19 (283 mg, 0.488 mmol) was dissolved in 80% ag AcOH (4.9 mL) and the solution was stirred for 2 h at 60 °C as the reaction was monitored by TLC (1:1 EtOAcn-hexane). The reaction mixture was evaporated, then diluted with CHCl₃, washed with H₂O, satd aq Na₂CO₃, and brine. The organic layer was dried over Na₂SO₄, filtered, and concentrated. The resulting residue was roughly purified by silica gel column chromatography (2:1 EtOAcn-hexane) to give crude mixture mainly containing 20. The residue was then dissolved in pyridine (4.9 mL). To the solution was added DTBS(OTf)₂ (174 μL, 0.537 mmol) at 0 °C. After stirring for 1 h as the reaction was monitored by TLC (1:2 EtOAc-n-hexane), the reaction was quenched by the addition of dry methanol at 0 °C. The reaction mixture was co-evaporated with toluene and then diluted with CHCl₃, washed with 2M HCl, H₂O, satd aq NaHCO₃, and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:5 EtOAc-n-hexane) to give 16 (293 mg, 86%) as a colorless foamy compound. $[\alpha]_D +52.2$ (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.87 - 6.86$ (m, 14 H, NPhth, 2 Ph), 5.59 (d, $J_{1.2} = 10.1$ Hz, 1 H, 1-H), 4.83 (d, $J_{gem} =$ 12.3 Hz, 1 H, PhCH₂), 4.60 (d, 1 H, PhCH₂), 4.26–4.13 (m, 3 H, 2-H, 6a-H, 6b-H), 4.05 (dd, J_{2,3} = 8.3 Hz, $J_{3.4}$ = 10.0 Hz, 1 H, 3-H), 3.97 (t, $J_{4.5}$ = 10.3 Hz, 1 H, 4-H), 3.65–3.60 (m, 1 H, 5-H),

1.09, 1.06 (2 s, 18 H, 2 ${}^{t}Bu$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 167.9$, 167.2, 138.0, 134.0, 133.8, 132.7, 131.9, 131.7, 128.9, 128.2, 128.1, 128.0, 127.4, 123.6, 123.3, 84.0, 79.1, 78.1, 74.6, 74.3, 66.4, 54.4, 27.5, 27.1, 22.7, 20.0. HRMS (ESI-TOF): m/z calcd for $C_{35}H_{41}NO_{6}SSi$: 654.2316 [M+Na]⁺; found: 654.2317.

2-(Trimethylsilyl)ethyl 2,3-di-O-acetyl-4,6-O-benzylidene-β-D-glucopyranoside (22). solution of 2-(trimethylsilyl)ethyl 4,6-O-benzylidene-β-D-glucopyranoside (21) (4.69 g, 12.7 mmol) in THF (63.5 mL) was added Ac₂O (3.6 mL, 38.1 mmol) at rt. After cooling to 0 °C, DMAP (160 mg, 1.31 mmol) was added to the mixture at the same temperature. After stirring for 1 h at rt as the reaction was monitored by TLC (2:3 EtOAc-n-hexane), the reaction was quenched by the addition of dry MeOH at 0 °C. The reaction mixture was diluted with CHCl₃ and washed with 2M HCl, satd aq NaHCO₃, and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was crystallized from EtOAc-n-hexane system to give 22 (5.75 g, 98%) as a white crystalline solid. M.p. = 133 to 135 °C. $[\alpha]_D$ –81.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.45–7.26 (m, 5 H, Ph), 5.50 (s, 1 H, PhCH<), 5.31 (t, $J_{2,3} = J_{3,4} = 9.6$ Hz, 1 H, 3-H), 4.99 (t, $J_{1,2} = 8.7$ Hz, 1 H, 2-H), 4.59 (d, 1 H, 1-H), 4.37 (dd, $J_{5.6a} = 5.0$ Hz, $J_{gem} = 10.5$ Hz, 1 H, 6a-H), 3.98 (m, 1 H, Me₃SiCH₂CH₂), 3.81 (t, $J_{5,6b} = 5.0$ Hz, 1 H, 6b-H), 3.70 (t, $J_{4,5} = 10.3$ Hz, 1 H, 4-H), 3.61–3.45 (m, 2 H, 5-H, Me₃SiCH₂CH₂), 2.06 (s, 3 H, Ac), 2.05 (s, 3 H, Ac), 1.02–0.86 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2, 169.5, 136.8, 129.1, 128.2, 126.1,$ 101.5, 100.9, 78.4, 72.4, 72.0, 68.6, 67.8, 66.3, 20.8, 20.8, 18.0, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{22}H_{32}O_8Si: 475.1759 [M+Na]^+$; found: 475.1759.

2-(Trimethylsilyl)ethyl 2,3-di-O-acetyl-4-O-benzyl-β-D-glucopyranoside (23). To a

solution of 22 (3.00 g, 6.63 mmol) in CH₂Cl₂ (66.0 mL) was added 4Å MS (3.00 g). After cooling to -78 °C and stirring for 1 h at the same temperature, dichlorophenylborane (2.0 mL, 15.4 mmol) and triethylsilane (3.3 mL, 20.7 mmol) were added to the mixture at -78 °C. After stirring for 35 min as the reaction was monitored by TLC (1:2 EtOAc-n-hexane), the reaction was quenched by the addition of triethylamine at -78 °C. The reaction mixture was filtered through Celite and washed with CHCl₃. The filtrate was washed with satd aq NaHCO₃ and brine. The organic layer was subsequently dried over Na₂SO₄, filtered, and concentrated. The resulting residue was purified by silica gel column chromatography (1:4 EtOAc-n-hexane) to give 23 (2.45 g, 81%) as a white powder. $[\alpha]_D$ –47.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.36–7.25 (m, 5 H, Ph), 5.23 (t, $J_{2,3} = J_{3,4} = 9.6$ Hz, 1 H, 3-H), 4.85 (t, $J_{1,2} = 8.0$ Hz, 1 H, 2-H), 4.64 (d, $J_{gem} = 11.4$ Hz, 1 H, PhC H_2), 4.60 (d, 1 H, PhC H_2), 4.51 (d, 1 H, 1-H), 3.99-3.88 (m, 2 H, Me₃SiCH₂CH₂, 6a-H), 3.78–3.70 (m, 2 H, 6a-H, 4-H), 3.54 (m, 1 H, Me₃SiCH₂CH₂), 3.43 (m, 1 H, 5-H), 2.02 (s, 3 H, Ac), 1.93 (m, 4 H, Ac, OH), 0.97–0.84 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2$, 169.6, 137.5, 128.5, 128.0, 128.0, 100.2, 75.5, 75.1, 75.0, 74.8, 72.1, 67.7, 61.5, 20.8, 20.7, 18.0, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{22}H_{34}O_8Si$: 477.1915 [M+Na]⁺; found: 477.1911.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-4,6-*O*-di-*tert*-butylsilylene-2-phthalimide-α-D-galactopyranosyl)-(1→6)-2,3-di-*O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (24). The glycosidation of 1 (75 mg, 0.119 mmol) with 23 (67 mg, 0.147 mmol) in CH₂Cl₂ (2.0 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:4 EtOAc–*n*-hexane) to give 24 (80 mg, 94%) as a colorless viscous

compound. [α]_D +79.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 8.04–7.28 (m, 14 H, NPhth, 2 Ph), 5.16 (dd, $J_{3,4}$ = 2.7 Hz, $J_{2,3}$ = 11.5 Hz, 1 H, 3-H^{GalN}), 5.10 (m, 2 H, 1-H^{GalN}, 3-H^{Glc}), 4.96 (dd, $J_{1,2}$ = 3.2 Hz, 1 H, 2-H^{GalN}), 4.78–4.69 (m, 4 H, PhC H_2), 4.59 (d, 1 H, 4-H^{GalN}), 4.28–4.24 (m, 2 H, 1-H^{Glc}, 2-H^{Glc}), 4.20–4.13 (m, 2 H, 6a-H^{GalN}, 6b-H^{GalN}), 3.98 (dd, $J_{5,6a}$ = 2.3 Hz, J_{gem} = 11.9 Hz, 1 H, 6a-H^{Glc}), 3.81 (t, $J_{3,4}$ = $J_{4,5}$ = 9.6 Hz, 1 H, 4-H^{Glc}), 3.66 (dd, 1 H, H-6b^{Glc}), 3.61 (d, 1 H, 5-H^{GalN}), 3.58 (m, 1 H, Me₃SiCH₂CH₂), 3.32–3.24 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 2.00 (s, 3 H, Ac), 1.98 (s, 3 H, Ac), 1.19, 1.11 (2 s, 18 H, 2 'Bu), 0.77–0.61 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): δ = 170.2, 169.2, 169.0, 168.6, 138.6, 138.3, 134.3, 133.7, 132.5, 131.1, 128.3, 127.8, 127.7, 127.6, 127.5, 123.4, 122.9, 99.9, 99.8, 77.2, 75.8, 75.2, 74.8, 74.1, 72.2, 72.1, 70.5, 70.0, 67.9, 67.3, 66.5, 50.9, 29.7, 27.6, 27.3, 23.4, 20.8, 20.8, 17.6, -1.4. HRMS (ESI-TOF): m/z calcd for C₅₁H₆₉NO₁₄Si₂: 998.4149 [M+Na]*; found: 998.4150.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-4,6-*O*-diisopropylsilylene-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (25). The glycosidation of **2** (135 mg, 0.224 mmol) with **23** (67 mg, 0.147 mmol) in CH₂Cl₂ (3.7 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (2:9 EtOAc–*n*-hexane) to give **25α** (111 mg, 79%) and **25β** (29 mg, 21%) as a colorless viscous compound. **25α:** [α]_D +79.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 8.04–7.28 (m, 14 H, NPhth, 2 Ph), 5.16 (dd, $J_{3,4}$ = 2.8 Hz, $J_{2,3}$ = 11.4 Hz, 1 H, 3-H^{GalN}), 5.12 (d, $J_{1,2}$ = 3.6 Hz, 1 H, 1-H^{GalN}), 5.08 (t, $J_{2,3}$ = $J_{3,4}$ = 9.2 Hz, 1 H, 3-H^{Glc}), 4.94 (dd, 1 H, 2-H^{GalN}), 4.76–4.69 (m, 4 H, PhC H_2), 4.46 (d, 1 H, 4-H^{GalN}), 4.26 (d, $J_{1,2}$ = 7.8 Hz, 1 H, 1-H^{Glc}), 4.22 (t, 1

H, 2-H^{Glc}), 4.15–4.08 (m, 2 H, 6a-H^{GalN}, 6b-H^{GalN}), 3.97 (dd, $J_{5,6a}$ = 2.7 Hz, J_{gem} = 12.4 Hz, 1 H, 6a-H^{Glc}), 3.79 (t, $J_{4.5} = 9.6$ Hz, 1 H, 4-H^{Glc}), 3.67 (dd, 1 H, 6b-H^{Glc}), 3.61 (s, 1 H, 5-H^{GalN}), 3.59 $(m, 1 H, Me_3SiCH_2CH_2), 3.31-3.24 (m, 2 H, 5-H^{Glc}, Me_3SiCH_2CH_2), 2.00 (s, 3 H, Ac), 1.98 (s, 3 H,$ 3 H, Ac), 1.22–1.12 (m, 14 H, 2 i Pr), 0.78–0.62 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2$, 169.2, 168.9, 168.6, 138.6, 138.2, 134.3, 133.7, 132.5, 132.3, 131.1, 129.9, 129.3, 128.4, 128.2, 127.7, 127.7, 127.5, 123.4, 123.0, 99.9, 99.7, 77.2, 75.7, 75.1, 74.7, 74.0, 72.5, 72.0, 71.0, 70.0, 67.9, 66.6, 66.5, 50.8, 29.6, 28.5, 20.8, 20.7, 17.5, 17.1, 17.0, 16.6, 13.9, 12.4, -1.5. HRMS (ESI-TOF): m/z calcd for $C_{49}H_{65}NO_{14}Si_2$: 970.3836 [*M*+Na]⁺; found: 970.3838. **25β:** [α]_D +14.0 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.83 - 7.10$ (m, 14 H, NPhth, 2 Ph), 5.28 (d, $J_{1,2} = 8.7$ Hz, 1 H, 1-H^{GalN}), 5.13 (t, $J_{2,3} = J_{3,4} = 9.6$ Hz, 1 H, 3-H^{Glc}), 4.80–4.73 (m, 2 H, 2-H^{Glc}, 2-H^{GalN}), 4.65 (d, $J_{gem} = 12.4$ Hz, 1 H, PhC H_2), 4.48–4.38 (m, 4 H, 4-H^{GalN}, 3 PhC H_2), 4.37 (d, $J_{1,2} = 7.8$ Hz, 1 H, 1-H^{Glc}), 4.30-4.23 (m, 3 H, $6a-H^{GalN}$, $6b-H^{GalN}$, $3-H^{GalN}$), 4.02 (dd, $J_{gem} = 10.5$ Hz, 1 H, $6a-H^{Glc}$), $3.86-H^{GalN}$ 3.78 (m, 2 H, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.53 (t, $J_{4.5} = 9.6$ Hz, 1 H, 4-H^{Glc}), 3.50–3.40 (m, 3 H, $5-H^{Glc}$, $5-H^{GalN}$, $Me_3SiCH_2CH_2$), 2.01 (s, 3 H, Ac), 1.91 (s, 3 H, Ac), 1.28–1.10 (m, 14 H, 2 Pr), 0.96–0.72 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.1, 169.6, 168.6, 167.5, 138.0, 137.5, 133.9, 133.7, 131.8, 131.7, 128.3, 128.2, 127.8,$ 127.6, 123.4, 123.0, 99.7, 98.2, 77.2, 76.3, 75.1, 75.0, 74.7, 74.6, 72.1, 71.6, 70.6, 69.2, 66.8, 66.7, 51.4, 31.9, 29.7, 29.3, 22.7, 20.8, 17.7, 17.1, 17.0, 16.9, 14.1, 12.7, 1.0, 0, -1.3. HRMS (ESI-TOF): m/z calcd for C₄₉H₆₅NO₁₄Si₂: 970.3836 [M+Na]⁺; found: 970.3840.

2-(Trimethylsilyl)ethyl

 $(3-O-benzyl-4,6-O-tert-butylmethylidene-2-deoxy-2-phthalimide-D-galactopyranosyl)-(1 \rightarrow$

6)-2,3-di-O-acetyl-4-O-benzyl-β-D-glucopyranoside (26). The glycosidation of 3 (92 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:10 \rightarrow 1:5 EtOAc-*n*-hexane) to give **26** α (33 mg, 33%) and **26** β (60 mg, 60%) as a colorless viscous compound. **26a:** $[\alpha]_D$ +57.7 (c 1.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 8.01-7.28$ (m, 14 H, NPhth, 2 Ph), 5.30 (dd, $J_{3,4} = 3.3$ Hz, $J_{2,3} = 11.7$ Hz, 1 H, 3-H^{GalN}), 5.17 $(d, J_{1,2} = 3.4 \text{ Hz}, 1 \text{ H}, 1 - H^{GalN}), 5.05 (t, J_{2,3} = J_{3,4} = 9.5 \text{ Hz}, 1 \text{ H}, 3 - H^{Glc}), 4.83 (dd, 1 \text{ H}, 2 - H^{GalN}),$ 4.74-4.67 (m, 4 H, 2 PhC H_2), 4.25 (d, $J_{1,2} = 8.0$ Hz, 1 H, 1-H^{Glc}), 4.14-4.10 (m, 3 H, 4-H^{GalN}, 6a-H^{GalN}, 2-H^{Glc}), 3.98 (d, $J_{5.6a}$ = 3.0 Hz, 1 H, 5-H^{GalN}), 3.95 (dd, $J_{5.6a}$ = 2.4 Hz, J_{gem} = 12.2 Hz, 1 H, 6a-H^{Glc}), 3.76–3.71 (m, 2 H, 6b-H^{GalN}, 4-H^{Glc}), 3.68–3.61 (m, 2 H, 6b-H^{Glc}, Me₃SiCH₂C H_2), 3.43 (s, 1 H, 'BuCH<), 3.30–3.25 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.98 (s, 3 H, Ac), 1.96 (s, 3 H, Ac), 0.98 (s, 9 H, ^tBu), 0.79–0.65 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): δ = 171.6, 170.6, 170.2, 170.1, 140.1, 139.7, 135.6, 135.1, 133.8, 132.5, 132.2, 130.1, 129.8, 129.6, 129.1, 129.0, 128.9, 124.6, 124.5, 108.5, 101.5, 101.1, 77.1, 76.6, 76.2, 75.5, 73.8, 73.5, 72.4, 72.3, 70.6, 69.5, 68.0, 67.7, 64.7, 53.0, 40.1, 36.5, 31.7, 31.0, 30.3, 26.0, 25.1, 24.3, 22.2, 22.1, 18.9, 15.4, 12.3, 1.3. HRMS (ESI-TOF): m/z calcd for $C_{48}H_{61}NO_{14}Si: 926.3754 [M+Na]^+$; found: 926.3758. **268:** $[\alpha]_D$ -5.0(*c* 1.0, CHCl₃); ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3): \delta = 7.84-7.14 \text{ (m, 14 H, NPhth, 2 Ph)}, 5.34 \text{ (d, } J_{1,2} = 8.7 \text{ Hz, 1 H, 1-H}^{GalN}),$ 5.13 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.79–4.71 (m, 2 H, 2-H^{GalN}, 2-H^{Glc}), 4.60–4.57 (m, 2 H, PhC H_2), 4.47–4.38 (m, 4 H, 3-H^{GalN}, 1-H^{Glc}, PhC H_2), 4.21 (dd, $J_{5.6a} = 2.0$ Hz, $J_{gem} = 12.2$ Hz, 1 H, 6a-H^{GalN}), 4.18 (s, 1 H, ^tBuCH<), 4.06 (dd, $J_{5,6b} = 1.2$ Hz, 1 H, 6b-H^{GalN}), 4.01 (d, $J_{3,4} = 3.3$ Hz, 1 H, 4-H^{GalN}), 3.92 (dd, $J_{5.6a} = 5.1$ Hz, $J_{gem} = 11.2$ Hz, 1 H, 6a-H^{Glc}), 3.85–3.81 (m, 2 H,

6b-H^{Glc}, Me₃SiCH₂CH₂), 3.54 (t, $J_{4,5} = 9.5$ Hz, 1 H, 4-H^{Glc}), 3.50–3.43 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 3.40 (s, 1 H, 5-H^{GalN}), 2.00 (s, 3 H, Ac), 1.90 (s, 3 H, Ac), 1.04 (s, 9 H, 'Bu), 0.84–0.77 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 171.4$, 170.9, 139.4, 138.9, 135.2, 135.1, 133.1, 129.7, 129.5, 129.4, 129.3, 129.2, 128.8, 128.6, 124.7, 124.4, 108.2, 101.1, 99.0, 77.5, 76.4, 76.1, 75.9, 75.3, 73.4, 73.1, 71.8, 70.2, 68.3, 68.3, 66.5, 52.7, 36.5, 26.0, 25.8, 24.0, 22.1, 19.0, 18.8, 15.4, 1.3. HRMS (ESI-TOF): m/z calcd for C₄₈H₆₁NO₁₄Si: 926.3754 [M+Na]⁺; found: 926.3755.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-4,6-*O*-ethylidene-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (27). The glycosidation of **4** (85 mg, 0.165 mmol) with **23** (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:3 EtOAc–*n*-hexane) to give **27**α (23 mg, 24%) and **27**β (59 mg, 61%) as a colorless viscous compound. **27α:** [α]_D +56.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 8.00–7.28 (m, 14 H, NPhth, 2 Ph), 5.30 (dd, $J_{3,4}$ = 3.9 Hz, $J_{2,3}$ = 11.7 Hz, 1 H, 3-H^{GalN}), 5.17 (d, $J_{1,2}$ = 3.5 Hz, 1 H, 1-H^{GalN}), 5.05 (t, $J_{2,3}$ = $J_{3,4}$ = 9.5 Hz, 1 H, 3-H^{Gilc}), 4.89 (dd, 1 H, 2-H^{GalN}), 4.77–4.65 (m, 5 H, PhC*H*₂, CH₃CH<), 4.25 (d, $J_{1,2}$ = 8.0 Hz, 1 H, 1-H^{Gilc}), 4.15–4.06 (m, 2 H, 6a-H^{GalN}, 2-H^{Gilc}), 3.96–3.93 (m, 2 H, 4-H^{GalN}, 6a-H^{Gilc}), 3.77–3.72 (m, 2 H, 6b-H^{Gilc}, 4-H^{Gilc}), 3.67–3.60 (m, 2 H, 6b-H^{GalN}, Me₃SiCH₂CH₂), 3.47 (s, 1 H, 5-H^{GalN}), 3.29–3.25 (m, 2 H, 5-H^{Gilc}, Me₃SiCH₂CH₂), 1.98 (s, 3 H, Ac), 1.96 (s, 3 H, Ac), 1.57 (d, $J_{CH3CH<CH3CH<C}$ = 5.0 Hz, 3 H, CH₃CH<), 0.76–0.66 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): δ = 171.6, 170.6, 170.2, 170.1, 140.0, 139.6, 135.7, 135.2, 133.8, 130.7, 130.4, 129.8, 129.5, 129.1, 129.0,

128.9, 124.7, 124.6, 101.6, 101.1, 100.4, 77.1, 76.6, 76.1, 75.5, 74.1, 73.5, 72.7, 72.4, 70.4, 68.0, 67.8, 64.3, 52.9, 31.0, 29.9, 22.4, 22.2, 22.1, 18.9, 1.3. HRMS (ESI-TOF): m/z calcd for $C_{45}H_{55}NO_{14}Si$: 884.3284 [M+Na]⁺; found: 884.3281. **27β**: [α]_D +2.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.85–7.04 (m, 14 H, NPhth, 2 Ph), 5.31 (d, $J_{1,2}$ = 8.5 Hz, 1 H, 1-H^{GalN}), 5.10 (t, $J_{2,3} = J_{3,4} = 9.2$ Hz, 1 H, 3-H^{Glc}), 4.76–4.61 (m, 4 H, 2-H^{GalN}, CH₃CH<, 2-H^{Glc}, PhCH₂), 4.47–4.38 (m, 5 H, 3-H^{GalN}, 1-H^{Glc}, PhCH₂), 4.21 (d, J_{gem} = 11.8 Hz, 1 H, 6a-H^{GalN}), 4.05–4.02 (m, 2 H, 4-H^{GalN}, 6a-H^{Glc}), 3.88 (dd, $J_{5,6b}$ = 1.4 Hz, 1 H, 6b-H^{GalN}), 3.81 (m, 1 H, Me₃SiCH₂CH₂), 3.76–3.73 (m, 1 H, 5-H^{Glc}), 3.46–3.40 (m, 3 H, 4-H^{Glc}, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.38 (s, 1 H, 5-H^{GalN}), 1.99 (s, 3 H, Ac), 1.88 (s, 3 H, Ac), 1.49 (d, $J_{CH3CH<CH3CH<C}$ = 5.0 Hz, 3 H, CH₃CH<), 0.86–0.74 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): δ = 171.4, 170.9, 169.9, 168.7, 139.1, 138.7, 135.2, 135.0, 133.1, 133.0, 129.7, 129.4, 129.2, 129.2, 128.9, 128.9, 124.8, 124.3, 101.0, 100.5, 99.4, 78.6, 77.7, 76.4, 76.2, 75.9, 75.3, 73.5, 73.4, 72.4, 70.1, 68.4, 68.1, 68.0, 53.3, 22.3, 22.1, 19.0, 1.3. HRMS (ESI-TOF): m/z calcd for $C_{45}H_{55}NO_{14}Si$: 884.3284 [M+Na]⁺; found: 884.3284.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-4,6-*O*-methylidene-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (28). The glycosidation of **5** (82 mg, 0.165 mmol) and **23** (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:3 EtOAc–*n*-hexane) to give **28**α (11 mg, 12%) and **28**β (58 mg, 62%) as a colorless viscous compound. **28**α: [α]_D +60.0 (*c* 1.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 8.00–7.25 (m, 14 H, NPhth, 2 Ph), 5.32–5.28 (m, 2 H, 3-H^{GalN}, CH₂O₂), 5.18 (d, $J_{1,2}$ = 3.5 Hz, 1 H, 1-H^{GalN}), 5.06

 $(t, J_{2,3} = J_{3,4} = 9.5 \text{ Hz}, 1 \text{ H}, 3 \cdot \text{H}^{Glc}), 4.91 \text{ (dd}, J_{2,3} = 11.7 \text{ Hz}, 1 \text{ H}, 2 \cdot \text{H}^{GalN}), 4.76 - 4.69 \text{ (m, 5 H, 1)}$ CH_2O_2 , PhC H_2), 4.25 (d, $J_{1,2} = 8.0$ Hz, 1 H, 1-H^{Glc}), 4.16 (t, 1 H, 2-H^{Glc}), 4.10 (d, $J_{gem} = 12.3$ Hz, 1 H, $6a-H^{GalN}$), 3.97-3.93 (m, 2 H, $4-H^{GalN}$, $6a-H^{Glc}$), 3.78-3.59 (m, 4 H, $6b-H^{GalN}$, $4-H^{Glc}$, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.52 (s, 1 H, 5-H^{GalN}), 3.30–3.24 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.99 (s, 3 H, Ac), 1.97 (s, 3 H, Ac), 0.78–0.63 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): δ = 171.6, 170.6, 170.2, 170.0, 139.7, 136.7, 135.7, 135.2, 133.8, 132.4, 129.8, 129.8, 129.6, 129.1, 129.0, 124.7, 124.6, 101.7, 101.2, 94.8, 77.1, 76.6, 76.1, 75.5, 74.1, 73.5, 72.9, 72.3, 70.4, 68.0, 67.8, 65.2, 52.8, 22.2, 22.1, 18.9, 1.3. HRMS (ESI-TOF): *m/z* calcd for $C_{44}H_{53}NO_{14}Si: 870.3128 [M+Na]^+$; found: 870.3127. **28β:** $[\alpha]_D + 4.8 (c \ 0.8, CHCl_3)$; 1H NMR (500 MHz, CDCl₃): $\delta = 7.83-7.01$ (m, 14 H, NPhth, 2 Ph), 5.30 (d, $J_{1,2} = 8.6$ Hz, 1 H, $1-H^{GalN}$), 5.28 (d, $J_{gem} = 6.4 \text{ Hz}$, 1 H, CH_2O_2), 5.09 (t, $J_{2,3} = J_{3,4} = 9.4 \text{ Hz}$, 1 H, $3-H^{Glc}$), 4.76–4.63 $(m, 4 H, 2-H^{GalN}, CH_2O_2, 2-H^{Glc}, PhCH_2), 4.46-4.34 (m, 4 H, 1-H^{Glc}, PhCH_2), 4.29 (dd, J_{34} = 3.5)$ Hz, $J_{2,3} = 11.1$ Hz, 1 H, 3-H^{GalN}), 4.20 (d, $J_{gem} = 12.2$ Hz, 1 H, 6a-H^{GalN}), 4.01–3.99 (m, 2 H, 4-H^{GalN}, 6a-H^{Glc}), 3.85-3.72 (m, 3 H, 6b-H^{GalN}, 5-H^{Glc}, Me₃SiCH₂CH₂), 3.46-3.36 (m, 4 H, 5-H^{GalN}, 4-H^{Glc}, 6b-H^{Glc}, Me₃SiCH₂CH₂), 1.98 (s, 3 H, Ac), 1.87 (s, 3 H, Ac), 0.84–0.73 (m, 2 H, $Me_3SiCH_2CH_2$), 0.00 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 171.4$, 170.9, 169.9, 168.6, 138.8, 138.8, 135.2, 135.0, 133.1, 133.0, 129.7, 129.5, 129.2, 129.0, 128.9, 124.8, 124.3, 101.0, 99.5, 94.7, 77.7, 76.4, 76.4, 76.2, 75.8, 75.4, 73.6, 73.4, 72.8, 70.1, 68.9, 68.4, 53.3, 22.1, 19.7, 1.3. HRMS (ESI-TOF): m/z calcd for $C_{44}H_{53}NO_{14}Si$: 870.3128 [M+Na]⁺; found: 870.3126.

2-(Trimethylsilyl)ethyl

 $(3-O-benzyl-4,6-O-benzylidene-2-deoxy-2-phthalimide-D-galactopyranosyl)-(1<math>\rightarrow$ 6)-2,3-di-

O-acetyl-4-O-benzyl-β-D-glucopyranoside (29). The glycosidation of 6 (101 mg, 0.174 mmol) and 23 (53 mg, 0.117 mmol) in CH₂Cl₂ (2.9 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:3 EtOAc-n-hexane) to give 29α (14 mg, 13%) and 29β (91 mg, 84%) as a colorless viscous compound. **29a:** $[\alpha]_D$ +68.5 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 8.01–7.27 (m, 19 H, NPhth, 3 Ph), 5.46 (s, 1 H, PhCH<), 5.39 (dd, $J_{3,4} = 3.2$ Hz, $J_{2,3} = 11.9$ Hz, 1 H, 3-H^{GalN}), 5.02 (d, $J_{1,2} = 3.2 \text{ Hz}$, 1 H, 1-H^{GalN}), 5.07 (t, $J_{3,4} = 9.4 \text{ Hz}$, $J_{2,3} = 9.7 \text{ Hz}$, 1 H, 3-H^{Glc}), 4.94 (dd, 1 H, 2-H^{GalN}), 4.81–4.72 (m, 4 H, 2 PhC H_2), 4.25 (m, 2 H, 1-H^{Glc}, 6a-H^{GalN}), 4.18–4.12 (m, 2 H, $4-H^{GalN}$, $2-H^{Glc}$), 3.99-3.94 (m, 2 H, $6a-H^{Glc}$, $6b-H^{GalN}$), 3.76 (t, $J_{4.5} = 9.6$ Hz, 1 H, $4-H^{Glc}$), 3.68(br d, 1 H, 6b-H^{Glc}), 3.63 (m, 1 H, Me₃SiCH₂CH₂), 3.56 (s, 1 H, 5-H^{GalN}), 3.31–3.24 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.98 (s, 3 H, Ac), 1.96 (s, 3 H, Ac), 0.80–0.63 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2$, 169.2, 168.8, 138.8, 138.3, 137.7, 134.3, 133.8, 132.5, 131.1, 128.9, 128.6, 128.5, 128.3, 128.2, 128.2, 128.1, 127.9, 127.8, 127.7, 127.6, 127.6, 127.5, 126.3, 123.3, 123.2, 100.9, 100.2, 99.8, 85.8, 75.8, 75.2, 74.8, 74.5, 74.1, 73.2, 72.1, 71.3, 71.1, 69.6, 66.6, 66.4, 63.1, 51.6, 29.7, 20.8, 20.8, 17.6, 0, -1.4, -2.1. HRMS (ESI-TOF): m/z calcd for $C_{50}H_{57}NO_{14}Si$: 946.3441 [M+Na]⁺; found: 946.3441. **296**: [α]_D +21.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.85-7.06$ (m, 19 H, NPhth, 3 Ph), 5.54 (s, 1 H, PhCH<), 5.36 (d, $J_{1,2} = 8.7$ Hz, 1 H, 1-H^{GalN}), 5.11 (m, 1 H, 3-H^{Glc}), 4.80–4.74 (m, 2 H, $2-H^{GalN}$, $2-H^{Glc}$), 4.63 (d, $J_{gem} = 12.4$ Hz, 1 H, PhC H_2), 4.50 (d, $J_{gem} = 13.2$ Hz, 1 H, PhC H_2), 4.47 (d, 1 H, PhC H_2), 4.45–4.36 (m, 3 H, 3-H^{GalN}, 6a-H^{GalN}, PhC H_2), 4.35 (d, $J_{1,2} = 7.8$ Hz, 1 H, $1-H^{Glc}$), 4.24 (d, $J_{3.4} = 2.8$ Hz, 1 H, $4-H^{GalN}$), 4.11–4.04 (m, 2 H, 6b-H^{GalN}, 6a-H^{Glc}), 3.86–3.80 (m, 2 H, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.49–3.42 (m, 4 H, 4-H^{Glc}, 5-H^{Glc}, 5-H^{GalN}, Me₃SiCH₂CH₂),

2.00 (s, 3 H, Ac), 1.88 (s, 3 H, Ac), 0.96–0.72 (m, 2 H, Me₃SiC H_2 CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): δ = 170.1, 169.6, 168.6, 167.4, 137.9, 137.7, 137.4, 133.9, 133.7, 131.8, 131.7, 128.9, 128.4, 128.2, 128.1, 127.9, 127.8, 127.6, 127.5, 126.4, 123.5, 123.0, 101.1, 99.7, 98.1, 76.3, 75.0, 74.8, 74.6, 74.1, 72.6, 72.0, 71.0, 69.3, 67.1, 66.8, 66.5, 51.9, 36.3, 20.7, 17.7, -1.3. HRMS (ESI-TOF): m/z calcd for C₅₀H₅₇NO₁₄Si: 946.3441 [M+Na]⁺; found: 946.3439.

2-(Trimethylsilyl)ethyl

[3-O-benzyl-2-deoxy-4,6-O-(4-nitrobenzylidene)-2-phthalimide-D-galactopyranosyl]-(1→6) -2,3-di-O-acetyl-4-O-benzyl-β-D-glucopyranoside (30). The glycosidation of 7 (103 mg, 0.165 mmol) and 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (2:3 EtOAc-n-hexane) to give 30α (9 mg, 10%) and 30β (56 mg, 55%) as a colorless viscous compound. **30a:** $[\alpha]_D$ +92.5 (c 1.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 8.28–7.24 (m, 18 H, NPhth, Ar), 5.48 (s, 1 H, ArCH<), 5.39 (dd, $J_{3,4} = 3.4$ Hz, $J_{2,3} = 11.7$ Hz, 1 H, 3-H^{GalN}), 5.16 (d, $J_{1,2} = 3.5$ Hz, 1 H, 1-H^{GalN}), 5.04 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.86 (dd, 1 H, 2-H^{GalN}), 4.80–4.69 (m, 4 H, PhC H_2), 4.26–4.22 (m, 2 H, 6a-H^{GalN}, 1-H^{Glc}), 4.15 (d, 1 H, 4-H^{GalN}), 4.10 (dd, $J_{1,2} = 8.0$ Hz, 1 H, 2-H^{Glc}), 3.97–3.93 (m, 2 H, 6a-H^{Glc}, 6b-H^{GalN}), 3.73 (t, $J_{4.5} = 9.7 \text{ Hz}$, 1 H, 4-H^{Glc}), 3.66 (dd, $J_{5.6b} = 1.9 \text{ Hz}$, $J_{gem} = 12.2 \text{ Hz}$, 1 H, 6b-H^{Glc}), 3.61 (m, 1 H, Me₃SiCH₂CH₂), 3.57 (s, 1 H, 5-H^{GalN}), 3.29–3.22 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.96 (s, 3 H, Ac), 1.92 (s, 3 H, Ac), 0.77–0.61 (m, 2 H, Me₃SiC H_2 CH₂), -0.02 (s, 9 H, Me_3 SiC H_2 CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.3$, 169.3, 168.8, 148.3, 143.1, 138.7, 138.3, 134.4, 134.0, 132.4, 131.1, 128.5, 128.3, 127.8, 127.8, 127.7, 127.6, 127.4, 123.4, 123.3, 100.2, 99.9, 99.2,

75.8, 75.3, 74.8, 74.1, 73.4, 72.1, 71.5, 71.1, 69.7, 66.7, 66.5, 62.9, 51.5, 29.7, 20.8, 20.8, 17.6, -1.3. HRMS (ESI-TOF): m/z calcd for $C_{50}H_{56}N_2O_{16}Si$: 991.3291 [M+Na] $^+$; found: 991.3287. **30β**: [α]_D +29.0 (c 0.5, CHCl₃); 1 H NMR (500 MHz, CDCl₃): δ = 8.26–7.06 (m, 18 H, NPhth, Ar), 5.60 (s, 1 H, ArCH<), 5.37 (d, $J_{1,2}$ = 8.5 Hz, 1 H, 1-H GalN), 5.11 (m, 1 H, 3-H Gic), 4.77–4.70 (m, 2 H, 2-H GalN , 2-H Gic), 4.63 (d, J_{gem} = 12.6 Hz, 1 H, PhC H_2), 4.50–4.36 (m, 6 H, 3-H GalN , 6a-H GalN , 1-H Gic , PhC H_2), 4.27 (d, $J_{3,4}$ = 3.4 Hz, 1 H, 4-H GalN), 4.12 (dd, $J_{5,6b}$ = 1.3 Hz, J_{gem} = 12.2 Hz, 1 H, 6b-H GalN), 4.04 (d, J_{gem} = 11.1 Hz, 1 H, 6a-H Gic), 3.85–3.77 (m, 2 H, 6b-H Gic), Me₃SiCH₂CH₂), 3.54 (s, 1 H, 5-H GalN), 3.50–3.42 (m, 3 H, 4-H Gic , 5-H Gic , Me₃SiCH₂CH₂), 2.00 (s, 3 H, Ac), 1.88 (s, 3 H, Ac), 0.86–0.74 (m, 2 H, Me₃SiCH₂CH₂), -0.01 (s, 9 H, Me_3 SiCH₂CH₂); 13 C NMR (125 MHz, CDCl₃): δ = 170.1, 169.6, 168.6, 167.5, 160.1, 148.3, 144.1, 137.7, 137.4, 134.0, 133.9, 131.7, 131.6, 128.4, 128.2, 127.9, 127.9, 127.8, 127.6, 127.5, 123.6, 123.4, 123.1, 99.8, 99.4, 98.2, 76.3, 75.0, 74.8, 74.6, 74.0, 72.8, 72.0, 71.3, 69.4, 67.1, 67.0, 66.6, 52.0, 20.8, 17.8, -1.3. HRMS (ESI-TOF): m/z calcd for $C_{50}H_{56}N_2O_{16}Si$: 991.3291 [M+Na] $^+$; found: 991.3295.

2-(Trimethylsilyl)ethyl

(4,6-*O*-anisylidene-3-*O*-benzyl-2-deoxy-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O* -acetyl-4-*O*-benzyl-β-D-glucopyranoside (31). The glycosidation of 8 (101 mg, 0.165 mmol) and 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (2:3 EtOAc–*n*-hexane) to give 31α (26 mg, 29%) and 31β (56 mg, 63%) as a colorless viscous compound. 31α: [α]_D +98.8 (*c* 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.96–6.90 (m, 18 H, NPhth, Ar), 5.37 (s, 1 H, ArC*H*<), 5.33 (dd, $J_{3,4}$ = 3.0 Hz, $J_{2,3}$ = 11.5 Hz, 1 H, 3-H^{GalN}), 5.15

 $(d, J_{1,2} = 3.5 \text{ Hz}, 1 \text{ H}, 1 - H^{GalN}), 5.02 (t, J_{2,3} = J_{3,4} = 9.5 \text{ Hz}, 1 \text{ H}, 3 - H^{Glc}), 4.89 (dd, 1 \text{ H}, 2 - H^{GalN}),$ $4.74 \text{ (d, } J_{\text{gem}} = 12.0 \text{ Hz, } 1 \text{ H, PhC}H_2), 4.69 \text{ (t, } J_{\text{gem}} = 11.5 \text{ Hz, } 2 \text{ H, PhC}H_2), 4.68 \text{ (d, } 1 \text{ H, PhC}H_2),$ 4.20 (m, $J_{1,2} = 8.0$ Hz, 1 H, 1-H^{Glc}), 4.17 (dd, $J_{5,6a} = 1.5$ Hz, $J_{gem} = 12.5$ Hz, 1 H, 6a-H^{GalN}), 4.11-4.07 (m, 2 H, $4-H^{GalN}$, $2-H^{Glc}$), 3.92 (dd, $J_{5.6a} = 2.5$ Hz, $J_{gem} = 12.0$ Hz, 1 H, $6a-H^{Glc}$), 3.88(dd, $J_{5,6b} = 1.5 \text{ Hz}$, 1 H, 6b-H^{GalN}), 3.81 (s, 3 H, OMe), 3.72 (t, $J_{4.5} = 9.5 \text{ Hz}$, 1 H, 4-H^{Glc}), 3.63 (dd, $J_{5,6b} = 2.0 \text{ Hz}$, 1 H, 6b-H^{Glc}), 3.59 (m, 1 H, Me₃SiCH₂CH₂), 3.51 (s, 1 H, 5-H^{GalN}), 3.26– 3.20 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.93 (s, 3 H, Ac), 1.91 (s, 3 H, Ac), 0.75–0.59 (m, 2 H, $Me_3SiCH_2CH_2$), -0.05 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.2$, 169.2, 168.8, 160.0, 138.8, 138.3, 134.3, 133.8, 132.4, 131.1, 130.3, 128.4, 128.1, 127.7, 127.7, 127.6, 127.5, 123.3, 123.1, 113.5, 100.8, 100.2, 99.8, 75.7, 75.2, 74.8, 74.1, 73.1, 72.1, 71.2, 71.1, 69.5, 66.6, 66.4, 63.1, 55.3, 51.6, 20.8, 20.7, 17.5, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{51}H_{59}NO_{15}Si: 976.3546 [M+Na]^+$; found: 976.3545. **31β:** $[\alpha]_D +13.8 (c \ 1.0, CHCl_3)$; ¹H NMR (500 MHz, CDCl₃): $\delta = 7.83-6.89$ (m, 18 H, NPhth, Ar), 5.48 (s, 1 H, ArCH<), 5.33 (d, $J_{1,2} =$ 8.5 Hz, 1 H, 1-H^{Glc}), 5.09 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.77–4.72 (m, 2 H, 2-H^{GalN}, $2-H^{Glc}$), 4.60 (d, $J_{gem} = 12.5 \text{ Hz}$, 1 H, PhC H_2), 4.48 (d, $J_{gem} = 11.0 \text{ Hz}$, 1 H, PhC H_2), 4.44 (d, 1 H, PhC H_2), 4.40 (dd, $J_{3,4} = 3.5$ Hz, $J_{2,3} = 10.0$ Hz, 1 H, 3-H^{GalN}), 4.38 (d, 1 H, PhC H_2), 4.35 (d, $J_{1,2}$ = 8.0 Hz, 1 H, 1-H^{GalN}), 4.31 (dd, $J_{5,6a}$ = 1.0 Hz, J_{gem} = 12.0 Hz, 1 H, 6a-H^{GalN}), 4.20 (d, 1 H, $4-H^{GalN}$), 4.04 (dd, $J_{5,6b} = 1.5$ Hz, 1 H, 6b-H^{GalN}), 4.03 (d, $J_{gem} = 10.5$ Hz, 1 H, 6a-H^{Glc}), 3.84-3.78 (m, 5 H, 6b-H^{Glc}, Me₃SiCH₂CH₂, OMe), 3.49–3.40 (m, 4 H, 5-H^{GalN}, 4-H^{Glc}, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.97 (s, 3 H, Ac), 1.86 (s, 3 H, Ac), 0.85–0.71 (m, 2 H, Me₃SiCH₂CH₂), -0.02 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.0$, 169.6, 168.6, 167.4, 160.0, 137.9, 137.5, 133.9, 133.7, 131.8, 131.7, 130.4, 128.4, 128.1, 127.9, 127.8, 127.8, 127.5, 127.5,

123.5, 123.0, 113.5, 101.0, 99.8, 98.1, 76.3, 75.1, 74.8, 74.5, 74.1, 72.6, 72.0, 71.0, 69.2, 67.0, 66.8, 66.5, 55.3, 51.9, 20.7, 17.7, -1.3. HRMS (ESI-TOF): m/z calcd for $C_{51}H_{59}NO_{15}Si$: 976.3546 [M+Na] $^+$; found: 976.3544.

2-(Trimethylsilyl)ethyl

[3-O-benzyl-4,6-O-(2-chloroethylidene)-2-deoxy-2-phthalimide-D-galactopyranosyl]- $(1\rightarrow 6)$ -2,3-di-O-acetyl-4-O-benzyl-β-D-glucopyranoside (32). The glycosidation of 9 (91 mg, 0.165 mmol) and 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:2 EtOAc-n-hexane) to give 32α (18 mg, 17%) and 32β (74 mg, 73%) as a colorless viscous compound. **32a:** $[\alpha]_D$ +12.5 (c 1.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.96–7.22 (m, 14 H, NPhth, 2 Ph), 5.28 (dd, $J_{3,4} = 3.4$ Hz, $J_{2,3} = 11.7$ Hz, 1 H, 3-H^{GalN}), 5.10 (d, $J_{1,2} = 3.5 \text{ Hz}, 1 \text{ H}, 1 \text{-H}^{GalN}$, 5.01 (t, $J_{2,3} = J_{3,4} = 9.5 \text{ Hz}, 1 \text{ H}, 3 \text{-H}^{Glc}$), 4.78 (dd, 1 H, 2-H^{GalN}), 4.73-4.62 (m, 5 H, PhC H_2 , ClCH₂CH<), 4.20 (d, $J_{1,2} = 8.0$ Hz, 1 H, 1-H^{Glc}), 4.10-4.06 (m, 2 H, 6a-H^{GalN}, 2-H^{Glc}), 3.92-3.88 (m, 2 H, 4-H^{GalN}, 6a-H^{Glc}), 3.73-3.55 (m, 6 H, 6b-H^{Glc}, ClCH₂, $4-H^{\mathit{Glc}}, \;\; 6b-H^{\mathit{GalN}}, \;\; Me_{3}SiCH_{2}C\mathit{H}_{2}), \;\; 3.42 \;\; (s, \;\; 1 \;\; H, \;\; 5-H^{\mathit{GalN}}), \;\; 3.25-3.20 \;\; (m, \;\; 2 \;\; H, \;\; 5-H^{\mathit{Glc}}, \;\; (m, \;\; 2 \;\; H, \;\; 3-H^{\mathit{Glc}}, \;\; (m, \;\; 3-H^{\mathsf{Glc}}, \;\; (m, \; 3-H^{\mathsf{Glc}}, \;\; (m, \; 3-H^{\mathsf{Glc}}, \;\; (m, \;\; 3-H^{\mathsf{Glc$ Me₃SiCH₂CH₂), 1.93 (s, 3 H, Ac), 1.92 (s, 3 H, Ac), 0.74–0.59 (m, 2 H, Me₃SiCH₂CH₂), -0.05 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.2$, 169.2, 168.8, 138.6, 138.3, 134.4, 133.9, 132.5, 131.1, 128.5, 128.3, 128.2, 127.9, 127.8, 127.8, 127.7, 127.7, 127.6, 123.4, 123.3, 100.6, 100.2, 99.9, 75.8, 75.3, 74.8, 74.1, 73.0, 72.1, 71.5, 70.8, 69.3, 66.7, 66.5, 62.9, 51.4, 44.0, 20.8, 20.8, 17.6, -1.3. HRMS (ESI-TOF): m/z calcd for C₄₅H₅₄ClNO₁₄Si: 918.2894 $[M+Na]^+$; found: 918.2895. **32β:** $[\alpha]_D$ +43.0 (c 1.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta =$ 7.84–7.06 (m, 14 H, NPhth, 2 Ph), 5.28 (d, $J_{1,2} = 8.6$ Hz, 1 H, 1-H^{GalN}), 5.08 (t, $J_{2,3} = J_{3,4} = 9.3$

Hz, 1 H, 3-H^{Glc}), 4.74–4.71 (m, 2 H, 2-H^{Glc}, CICH₂CH<), 4.64 (dd, $J_{2,3} = 11.1$ Hz, 1 H, 2-H^{GalN}), 4.58 (d, $J_{gem} = 12.5$ Hz, 1 H, PhC H_2), 4.44–4.33 (m, 5 H, 3-H^{GalN}, 1-H^{Glc}, PhC H_2), 4.25 (dd, $J_{5,6a} = 0.9$ Hz, $J_{gem} = 11.9$ Hz, 1 H, 6a-H^{GalN}), 4.03–4.00 (m, 2 H, 4-H^{GalN}, 6a-H^{Glc}), 3.89 (dd, $J_{5,6b} = 1.5$ Hz, 1 H, 6b-H^{GalN}), 3.75 (m, 1 H, Me₃SiCH₂CH₂), 3.75–3.72 (m, 1 H, 6b-H^{Glc}), 3.70–3.63 (m, 2 H, CICH₂), 3.46–3.40 (m, 4 H, 5-H^{GalN}, 4-H^{Glc}, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.97 (s, 3 H, Ac), 1.86 (s, 3 H, Ac), 0.90–0.71 (m, 2 H, Me₃SiCH₂CH₂), -0.01 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.1$, 169.6, 168.6, 167.4, 137.7, 137.4, 134.0, 133.8, 131.8, 131.7, 128.4, 128.3, 128.2, 127.9, 127.7, 127.6, 123.6, 123.0, 100.6, 99.8, 98.2, 76.4, 75.1, 74.9, 74.6, 73.9, 72.5, 72.1, 71.2, 69.0, 67.1, 66.9, 66.7, 51.9, 44.0, 20.8, 17.8, -1.3. HRMS (ESI-TOF): m/z calcd for C₄₅H₅₄ClNO₁₄Si: 918.2894 [M+Na]⁺; found: 918.2893.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-2-phthalimide-4,6-*O*-1',1',3',3'-tetraisopropyldisiloxanylidene-D-galac topyranosyl)-(1→6)-2,3-di-*O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (33). The glycosidation of 10 (211 mg, 0.287 mmol) with 23 (87 mg, 0.191 mmol) in CH₂Cl₂ (4.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:9 EtOAc–*n*-hexane) to give 33α (44 mg, 21%) and 33β (161 mg, 78%) as a colorless viscous compound. 33α: [α]_D +98.0 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.97–7.17 (m, 14 H, NPhth, 2 Ph), 5.20 (dd, $J_{3,4}$ = 2.3 Hz, $J_{2,3}$ = 11.5 Hz, 1 H, 3-H^{GalN}), 5.06 (d, $J_{1,2}$ = 3.7 Hz, 1 H, 1-H^{GalN}), 5.04 (t, $J_{2,3}$ = $J_{3,4}$ = 9.6 Hz, 1 H, 3-H^{Glc}), 4.90 (dd, 1 H, 2-H^{GalN}), 4.78–4.65 (m, 4 H, PhC*H*₂), 4.58 (br d, 1 H, 4-H^{GalN}), 4.24 (d, $J_{1,2}$ = 7.8 Hz, 1 H, 1-H^{Glc}), 4.10 (t, 1 H, 2-H^{Glc}), 4.03–3.80 (m, 4 H, 5-H^{GalN}, 6a-H^{GalN}, 6b-H^{GalN}, 6a-H^{Glc}), 3.75 (t, $J_{4,5}$ = 9.6 Hz, 1 H, 4-H^{Glc}), 3.66–3.61 (m, 2 H, 6b-H^{Glc}, Me₃SiCH₂C*H*₂), 3.31–3.25 (m, 2 H, 4.55)

5-H^{Glc}, Me₃SiCH₂CH₂), 1.97 (s, 3 H, Ac), 1.92 (s, 3 H, Ac), 1.31–0.90 (m, 28 H, 4 ⁱPr), 0.80– 0.63 (m, 2 H, Me₃SiC H_2 CH₂), 0.00 (s, 9 H, Me_3 SiC H_2 CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.3, 169.2, 169.1, 168.8, 138.1, 138.0, 134.3, 133.9, 133.7, 132.4, 131.2, 128.7, 128.5,$ 128.0, 127.8, 127.8, 127.3, 123.2, 123.0, 100.0, 99.7, 77.7, 77.6, 75.5, 75.2, 74.9, 74.3, 73.3, 72.1, 71.7, 70.5, 66.5, 66.2, 65.1, 59.5, 51.5, 29.7, 20.8, 17.6, 17.5, 17.3, 17.2, 14.3, 13.3, 12.8, 12.6, 1.0, 0, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{55}H_{79}NO_{15}Si_3$: 1100.4650 $[M+Na]^+$; found: 1100.4651. **33β:** $[\alpha]_D$ +18.5 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.78–6.96 (m, 14) H, NPhth, 2 Ph), 5.30 (d, $J_{1,2} = 8.7$ Hz, 1 H, 1-H^{GalN}), 5.11 (t, $J_{2,3} = J_{3,4} = 9.2$ Hz, 1 H, 3-H^{Glc}), 4.77 (dd, $J_{1,2} = 7.8$ Hz, 1 H, 2-H^{Glc}), 4.71 (dd, $J_{2,3} = 11.0$ Hz, 1 H, 2-H^{GalN}), 4.60 (d, $J_{gem} = 11.9$ Hz, 1 H, PhC H_2), 4.51 (d, $J_{gem} = 11.0$ Hz, 1 H, PhC H_2), 4.40 (d, $J_{3,4} = 2.3$ Hz, 1 H, 4-H^{GalN}), 4.39 (d, 1 H, PhC H_2), 4.27 (dd, 1 H, 3-H^{GalN}), 4.24 (d, 1 H, PhC H_2), 3.99 (br dd, $J_{gem} = 10.1$ Hz, 1 H, 6a-H^{Glc}), 3.89–3.75 (m, 4 H, 6a-H^{GalN}, 5-H^{GalN}, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.67 (dd, $J_{5.6b}$ = 5.5 Hz, 1 H, 6b-H^{GalN}), 3.48 (t, 1 H, 4-H^{Glc}), 3.46–3.41 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂C H_2), 1.99 (s, 3 H, Ac), 1.89 (s, 3 H, Ac), 1.28–0.72 (m, 30 H, 4 ⁱPr, Me₃SiCH₂CH₂), 0.00 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.1$, 169.6, 168.6, 167.7, 137.6, 137.5, 133.6, 133.6, 131.8, 128.3, 128.0, 127.8, 127.7, 127.4, 123.2, 123.0, 99.7, 97.8, 76.2, 76.1, 75.1, 74.5, 74.0, 72.1, 71.9, 66.9, 65.5, 65.4, 64.8, 59.3, 51.5, 20.7, 17.7, 17.6, 17.4, 17.3, 17.3, 17.2, 17.2, 17.2, 17.1, 16.9, 14.1, 13.2, 12.7, 12.5, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{55}H_{79}NO_{15}Si_3$: 1100.4650 [M+Na]⁺; found: 1100.4654.

2-(Trimethylsilyl)ethyl

(3-O-benzyl-4,6-di-O-tert-butyldimethylsilyl-2-deoxy-2-phthalimide-D-galactopyranosyl)-(1 \rightarrow 6)-2,3-di-O-acetyl-4-O-benzyl- β -D-glucopyranoside (34). The glycosidation of 11 (101

mg, 0.140 mmol) with 23 (43 mg, 94.0 μmol) in CH₂Cl₂ (2.3 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:5 EtOAc-n-hexane) to give 34α (27 mg, 27%) and 34β (70 mg, 70%) as a colorless viscous compound. 34a: $[\alpha]_D$ +56.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ = 7.97–7.17 (m, 14 H, NPhth, 2 Ph), 5.11 (dd, $J_{34} = 1.8$ Hz, $J_{23} = 11.9$ Hz, 1 H, 3-H^{GalN}), 5.09 (d, $J_{1,2} = 3.7 \text{ Hz}, 1 \text{ H}, 1 - \text{H}^{GalN}$), 5.05 (t, $J_{2,3} = J_{3,4} = 9.2 \text{ Hz}, 1 \text{ H}, 3 - \text{H}^{Glc}$), 4.93 (dd, 1 H, 2-H^{GalN}), 4.79, 4.71 (2 d, $J_{gem} = 11.4$ Hz, 2 H, PhC H_2), 4.78, 4.66 (2 d, $J_{gem} = 11.0$ Hz, 2 H, PhC H_2), 4.36(br s, 1 H, 4-H^{GalN}), 4.24 (d, $J_{1,2} = 8.2$ Hz, 1 H, 1-H^{Glc}), 4.19 (t, 1 H, 2-H^{Glc}), 4.06 (dd, $J_{5,6a} = 0.9$ Hz, $J_{gem} = 13.3$ Hz, 1 H, 6a-H^{Glc}), 3.92 (br t, $J_{5.6b} = 6.4$ Hz, $J_{5.6a} = 6.8$ Hz, 1 H, 5-H^{GalN}), 3.84– 3.79 (m, 3 H, 4-H^{Glc}, 6a-H^{GalN}, 6b-H^{GalN}), 3.67 (dd, $J_{5.6b} = 1.8$ Hz, $J_{gem} = 12.8$ Hz, 1 H, 6b-H^{Glc}), 3.60 (m, 1 H, Me₃SiCH₂CH₂), 3.29–3.22 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.99 (s, 3 H, Ac), 1.92 (s, 3 H, Ac), 0.96 (s, 18 H, 2 t Bu), 0.76–0.58 (m, 2 H, Me₃SiC H_2 CH₂), 0.13–0.00 (m, 21 H, 2 SiMe₂, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.2$, 169.2, 169.1, 168.8, 138.2, 134.3, 133.6, 132.5, 131.1, 128.5, 128.1, 128.0, 127.8, 127.4, 127.2, 123.4, 123.0, 100.3, 99.7, 77.2, 75.5, 75.0, 74.9, 74.3, 74.2, 72.8, 72.1, 71.7, 67.5, 66.5, 66.4, 62.3, 51.4, 26.0, 25.9, 20.8, 20.8, 18.6, 18.2, 17.6, -1.4, -4.1, -4.9, -5.2, -5.3. HRMS (ESI-TOF): m/z calcd for $C_{55}H_{81}NO_{14}Si_3$: 1086.4857 [M+Na]⁺; found: 1086.4856. **34β**: [α]_D +4.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.79 - 6.94$ (m, 14 H, NPhth, 2 Ph), 5.32 (d, $J_{1,2} = 8.7$ Hz, 1 H, 1-H^{GalN}), 5.10 (t, $J_{2,3} = J_{3,4} = 9.2$ Hz, 1 H, 3-H^{Glc}), 4.77 (dd, $J_{1,2} = 7.8$ Hz, $J_{2,3} = 9.2$ Hz, 1 H, 2-H^{Glc}), 4.71 $(dd, J_{2,3} = 11.0 \text{ Hz}, 1 \text{ H}, 2\text{-H}^{GalN}), 4.65 (d, J_{gem} = 11.9 \text{ Hz}, 1 \text{ H}, PhCH_2), 4.51 (d, J_{gem} = 11.0 \text{ Hz}, 1 \text{ H}, PhCH_2)$ 1 H, PhC H_2), 4.37 (d, $J_{1,2} = 7.8$ Hz, 1 H, 1-H^{Glc}), 4.36 (d, 1 H, PhC H_2), 4.27 (br d, 1 H, 4-H^{GalN}), 4.24 (d, 1 H, PhC H_2), 4.18 (dd, $J_{3,4} = 2.3$ Hz, 1 H, 3-H^{GalN}), 4.04 (dd, $J_{5,6a} = 1.8$ Hz, $J_{gem} = 11.5$

Hz, 1 H, 6a-H^{Glc}), 3.86–3.78 (m, 3 H, 6a-H^{GalN}, 6b-H^{Glc}, Me₃SiCH₂CH₂), 3.70 (dd, $J_{5,6b}$ = 5.5 Hz, $J_{5,6a}$ = 9.6 Hz, 1 H, 5-H^{GalN}), 3.53–3.40 (m, 4 H, 4-H^{Glc}, 5-H^{Glc}, 6b-H^{GalN}, Me₃SiCH₂CH₂), 1.99 (s, 3 H, Ac), 1.87 (s, 3 H, Ac), 0.99, 0.94 (2 s, 18 H, 2 'Bu), 0.87–0.72 (m, 2 H, Me₃SiCH₂CH₂), 0.16–0.00 (m, 21 H, 2 SiMe₂, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): δ = 170.1, 169.6, 168.7, 137.6, 137.6, 133.6, 133.5, 131.7, 128.3, 128.0, 127.7, 127.6, 127.4, 123.2, 123.0, 99.7, 97.9, 77.0, 76.3, 75.7, 75.1, 74.5, 72.2, 72.1, 66.9, 66.5, 65.6, 61.2, 51.6, 29.7, 26.2, 26.1, 26.0, 25.9, 20.7, 20.7, 18.6, 18.2, 17.7, -1.3, -4.3, -4.9, -5.3. HRMS (ESI-TOF): m/z calcd for $C_{55}H_{81}NO_{14}Si_3$: 1086.4857 [M+Na]⁺; found: 1086.4852.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-2-phthalimide-4,6-*O*-xylylene-D-galactopyranosyl)-(1→6)-2,3-di-*O*-ac etyl-4-*O*-benzyl-β-D-glucopyranoside (35). The glycosidation of 12 (98 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (2:5 EtOAc-*n*-hexane) to give 35α (7 mg, 7%) and 35β (73 mg, 70%) as a colorless viscous compound. 35α: [α]_D +79.2 (*c* 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 8.01–7.13 (m, 18 H, NPhth, Ar), 5.28 (dd, $J_{3,4}$ = 3.0 Hz, $J_{2,3}$ = 12.0 Hz, 1 H, 3-H^{GalN}), 5.20 (d, J_{gem} = 12.5 Hz, 1 H, ArC H_2), 5.08 (d, $J_{1,2}$ = 4.0 Hz, 1 H, 1-H^{GalN}), 5.05 (t, $J_{2,3}$ = $J_{3,4}$ = 9.5 Hz, 1 H, 3-H^{Glc}), 4.98 (d, 1 H, ArC H_2), 4.97 (d, J_{gem} = 12.0 Hz, 1 H, ArC H_2), 4.93 (dd, 1 H, 2-H^{GalN}), 4.91 (d, 1 H, ArC H_2), 4.73–4.68 (m, 4 H, ArC H_2), 4.24 (d, $J_{1,2}$ = 8.0 Hz, 1 H, 1-H^{Glc}), 4.21–4.17 (m, 2 H, 4-H^{GalN}, 2-H^{Glc}), 4.04 (m, 1 H, 6a-H^{GalN}), 4.00–3.97 (m, 2 H, 5-H^{GalN}, 6a-H^{Glc}), 3.77 (t, $J_{4,5}$ = 9.5 Hz, 1 H, 4-H^{Glc}), 3.71 (dd, $J_{5,6b}$ = 8.0 Hz, J_{gem} = 11.0 Hz, 1 H, 6b-H^{GalN}), 3.63 (dd, $J_{5,6b}$ = 1.5 Hz, J_{gem} = 12.0 Hz, 1 H, Me₃SiCH₂CH₂), 3.27–3.21 (m, 2 H, 5-H^{Glc})

 $Me_3SiCH_2CH_2$), 2.00 (s, 3 H, Ac), 1.95 (s, 3 H, Ac), 0.76–0.60 (m, 2 H, $Me_3SiCH_2CH_2$), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.3$, 169.2, 169.0, 168.8, 138.3, 138.0, 136.9, 136.7, 135.8, 134.3, 133.7, 132.5, 131.1, 130.8, 130.6, 129.0, 128.6, 128.5, 128.3, 128.2, 127.9, 127.9, 127.5, 127.4, 123.5, 123.1, 100.1, 99.8, 79.2, 75.5, 75.1, 75.0, 74.9, 74.2, 72.9, 72.5, 72.1, 71.0, 70.2, 70.1, 66.7, 66.5, 52.3, 20.8, 17.6, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{51}H_{59}NO_{14}Si: 960.3597 [M+Na]^+$; found: 960.3595. **35β:** $[\alpha]_D$ –46.2 (c 1.0, CHCl₃); ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: $\delta = 7.82-6.99 \text{ (m, 18 H, NPhth, Ar)}, 5.22 \text{ (d, } J_{gem} = 13.0 \text{ Hz}, 1 \text{ H, ArC}H_2)$, 5.18 (d, $J_{1,2} = 8.5$ Hz, 1 H, 1-H^{GalN}), 5.07 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 5.04 (d, 1 H, $ArCH_2$), 4.92 (d, $J_{gem} = 11.5 Hz$, 1 H, $ArCH_2$), 4.81 (d, 1 H, $ArCH_2$), 4.72 (dd, $J_{1,2} = 8.0 Hz$, 1 H, $2-H^{Glc}$), 4.67 (dd, $J_{2,3} = 11.0$ Hz, 1 H, $2-H^{GalN}$), 4.61 (d, $J_{gem} = 12.5$ Hz, 1 H, ArC H_2), 4.37 (d, $J_{\text{gem}} = 11.5 \text{ Hz}, 1 \text{ H}, \text{ArC}H_2$, 4.33 (d, 1 H, ArC H_2), 4.32 (d, 1 H, 1-H^{Glc}), 4.23 (dd, $J_{3.4} = 3.0 \text{ Hz}$, 1 H, 3-H^{GalN}), 4.22 (d, 1 H, ArC H_2), 4.10 (dd, $J_{5.6a} = 3.5$ Hz, $J_{gem} = 8.5$ Hz, 1 H, 6a-H^{GalN}), 4.06 (d, 1 H, 4-H^{GalN}), 3.94 (dd, $J_{5.6a} = 1.5$ Hz, $J_{gem} = 11.0$ Hz, 1 H, 6a-H^{Glc}), 3.78 (m, 1 H, $Me_3SiCH_2CH_2$), 3.74–3.68 (m, 2 H, 5-H^{GalN}, 6b-H^{GalN}), 3.65 (dd, $J_{5.6b} = 5.0$ Hz, 1 H, 6b-H^{Glc}), 3.44-3.36 (m, 3 H, $4-H^{Glc}$, $5-H^{Glc}$, $Me_3SiCH_2CH_2$), 1.97 (s, 3 H, Ac), 1.86 (s, 3 H, Ac), 0.84– 0.70 (m, 2 H, Me₃SiCH₂CH₂), -0.02 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.0, 169.5, 168.4, 167.6, 137.5, 137.4, 137.1, 136.4, 133.8, 133.6, 131.7, 131.6, 130.9,$ 130.5, 128.7, 128.5, 128.3, 128.2, 127.8, 127.7, 127.5, 127.4, 123.3, 123.0, 99.7, 98.2, 76.2, 76.0, 75.7, 75.1, 74.6, 74.4, 74.4, 73.7, 72.0, 71.2, 70.6, 70.5, 67.0, 66.9, 52.8, 20.7, 17.7, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{51}H_{59}NO_{14}Si: 960.3597 [M+Na]^+$; found: 960.3597.

2-(Trimethylsilyl)ethyl

(3,4,6-tri-O-benzyl-2-deoxy-2-phthalimide-D-galactopyranosyl)- $(1\rightarrow 6)$ -2,3-di-O-acetyl-4-O-

benzyl-β-D-glucopyranoside (36). The glycosidation of 13 (111 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:3 EtOAc*n*-hexane) to give 36α (10 mg, 9%) and 36β (90 mg, 80%) as a colorless viscous compound. **36a:** $[\alpha]_D$ +76.0 (c 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.94–7.19 (m, 24 H, NPhth, 4 Ph), 5.27 (dd, $J_{3,4} = 2.5$ Hz, $J_{2,3} = 11.5$ Hz, 1 H, 3-H^{GalN}), 5.07 (d, $J_{1,2} = 3.5$ Hz, 1 H, 1-H^{GalN}), 5.00 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.94–4.91 (m, 2 H, 2-H^{GalN}, PhC H_2), 4.74 (d, $J_{gem} = 1.00$ 11.0 Hz, 1 H, PhC H_2), 4.71 (d, $J_{gem} = 11.5$ Hz, 1 H, PhC H_2), 4.66–4.63 (m, 2 H, PhC H_2), 4.55 $(d, J_{gem} = 11.5 \text{ Hz}, 1 \text{ H}, PhCH_2), 4.49, 4.42 (2 d, J_{gem} = 12.0 \text{ Hz}, 2 \text{ H}, PhCH_2), 4.18 (d, J_{1.2} = 8.0)$ Hz, 1 H, 1-H^{Glc}), 4.14–4.10 (m, 3 H, 4-H^{GalN}, 2-H^{Glc}, 6a-H^{GalN}), 3.98 (dd, $J_{5,6a}$ = 2.5 Hz, J_{gem} = 12.5 Hz, 1 H, 6a-H^{Glc}), 3.70 (t, $J_{4.5} = 9.5$ Hz, 1 H, 4-H^{Glc}), 3.64–3.57 (m, 3 H, 6b-H^{Glc}, 5-H^{GalN}, 6b-H^{GalN}), 3.53 (m, 1 H, Me₃SiCH₂CH₂), 3.24–3.17 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 1.93 (s, 3 H, Ac), 1.89 (s, 3 H, Ac), 0.71–0.55 (m, 2 H, Me₃SiCH₂CH₂), -0.06 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.2$, 169.3, 169.0, 168.8, 138.5, 138.1, 138.1, 138.0, 134.3, 133.7, 132.5, 131.2, 128.5, 128.4, 128.3, 128.2, 127.9, 127.8, 127.8, 127.7, 127.6, 127.5, 123.4, 123.1, 100.0, 99.8, 77.6, 75.6, 75.1, 75.0, 74.7, 74.2, 73.9, 73.5, 72.8, 72.1, 71.3, 70.2, 69.0, 66.6, 66.5, 52.2, 20.8, 20.8, 17.6, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{57}H_{65}NO_{14}Si$: 1038.4067 $[M+Na]^+$; found: 1038.4069. **36β:** $[\alpha]_D$ +6.0 (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.83-6.99$ (m, 24 H, NPhth, 4 Ph), 5.28 (d, $J_{1,2} = 8.5$ Hz, 1 H, 1-H^{GalN}), 5.10 (m, 1 H, 3-H^{Glc}), 4.99 (d, $J_{gem} = 11.5$ Hz, 1 H, PhC H_2), 4.79–4.75 (m, 2 H, 2-H^{GalN}, 2-H^{Glc}), 4.64 (d, 1 H, PhC H_2), 4.60 (d, $J_{gem} = 12.5$ Hz, 1 H, PhC H_2), 4.50, 4.46 (2 d, $J_{gem} = 12.0$ Hz, 2 H, PhC H_2), 4.42 (d, $J_{\text{gem}} = 11.0 \text{ Hz}$, 1 H, PhC H_2), 4.36 (d, $J_{1,2} = 8.5 \text{ Hz}$, 1 H, 1-H Glc), 4.35–4.30 (m, 3 H,

3-H^{GalN}, PhC H_2), 4.08 (d, $J_{3,4} = 2.5$ Hz, 1 H, 4-H^{GalN}), 4.03 (d, $J_{gem} = 10.5$ Hz, 1 H, 6a-H^{Glc}), 3.81 (m, 1 H, Me₃SiCH₂C H_2), 3.78–3.64 (m, 4 H, 5-H^{GalN}, 6a-H^{GalN}, 6b-H^{GalN}, 6b-H^{Glc}), 3.47–3.39 (m, 3 H, 4-H^{Glc}, 5-H^{Glc}, Me₃SiCH₂C H_2), 1.99 (s, 3 H, Ac), 1.87 (s, 3 H, Ac), 0.86–0.72 (m, 2 H, Me₃SiC H_2 CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.0$, 169.5, 168.5, 167.6, 138.5, 137.9, 137.6, 137.4, 133.7, 133.6, 131.7, 128.4, 128.3, 128.2, 128.2, 128.1, 127.8, 127.7, 127.7, 127.6, 127.5, 127.5, 123.3, 123.0, 99.7, 98.3, 77.2, 76.3, 75.1, 74.5, 74.4, 73.6, 73.4, 72.1, 71.9, 71.5, 68.4, 66.8, 52.7, 20.7, 20.7, 17.6, -1.4. HRMS (ESI-TOF): m/z calcd for C₅₇H₆₅NO₁₄Si: 1038.4067 [M+Na]⁺; found: 1038.4067.

2-(Trimethylsilyl)ethyl

(3-*O*-benzyl-2-deoxy-4,6-di-*O*-methyl-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O*-a cetyl-4-*O*-benzyl-β-D-glucopyranoside (37). The glycosidation of 14 (86 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:2 EtOAc–*n*-hexane) to give 37α (7 mg, 7%) and 37β (69 mg, 72%) as a colorless viscous compound. 37α: [α]_D +5.2 (*c* 1.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 7.94–7.18 (m, 14 H, NPhth, 2 Ph), 5.23 (dd, $J_{3,4}$ = 2.6 Hz, $J_{2,3}$ = 11.7 Hz, 1 H, 3-H^{GalN}), 5.06 (d, $J_{1,2}$ = 3.7 Hz, 1 H, 1-H^{GalN}), 5.00 (t, $J_{2,3}$ = $J_{3,4}$ = 9.4 Hz, 1 H, 3-H^{GlC}), 4.82 (dd, 1 H, 2-H^{GalN}), 4.75 (d, J_{gem} = 10.9 Hz, 1 H, PhC H_2), 4.70 (d, J_{gem} = 11.4 Hz, 1 H, PhC H_2), 4.64 (d, 1 H, PhC H_2), 4.63 (d, 1 H, PhC H_2), 4.19 (d, $J_{1,2}$ = 8.0 Hz, 1 H, 1-H^{GlC}), 4.13 (t, 1 H, 2-H^{GlC}), 4.04 (br t, 1 H, 5-H^{GalN}), 3.98 (dd, $J_{5,6a}$ = 2.3 Hz, J_{gem} = 12.1 Hz, 1 H, 6a-H^{GalN}), 3.84 (d, 1 H, 4-H^{GalN}), 3.71 (t, $J_{4,5}$ = 9.7 Hz, 1 H, 4-H^{GlC}), 3.64–3.49 (m, 7 H, 6b-H^{GalN}), 6a-H^{GlC}, 6b-H^{GlC}, OMe, Me₃SiCH₂CH₂), 3.34 (s, 3 H, OMe), 3.24–3.18 (m, 2 H, 5-H^{GlC}), Me₃SiCH₂CH₂), 1.93 (s, 3 H, Ac), 1.89 (s, 3 H, Ac), 0.70–0.55 (m, 2 H,

 $Me_3SiCH_2CH_2$), -0.05 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.3$, 169.3, 169.0, 168.8, 138.1, 138.0, 134.3, 133.8, 132.5, 131.2, 128.5, 128.3, 128.0, 127.9, 127.8, 127.6, 123.4, 123.1, 100.0, 99.8, 77.6, 75.6, 75.1, 75.0, 75.0, 74.2, 73.7, 72.1, 71.3, 71.2, 70.0, 66.6, 66.5, 61.2, 59.2, 52.1, 20.8, 17.6, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{45}H_{57}NO_{14}Si$: 886.3441 $[M+Na]^+$; found: 886.3440. **37β:** $[\alpha]_D$ –12.3 (c 2.8, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.78-7.01$ (m, 14 H, NPhth, 2 Ph), 5.21 (d, $J_{1,2} = 8.6$ Hz, 1 H, 1-H^{GalN}), 5.08 (m, 1 H, 3-H^{Glc}), 4.73 (dd, $J_{1,2} = 8.0$ Hz, $J_{2,3} = 9.6$ Hz, 1 H, 2-H^{Glc}), 4.65 (d, $J_{gem} = 12.1$ Hz, 1 H, PhC H_2), 4.61 (dd, $J_{2,3} = 11.1$ Hz, 1 H, 2-H^{GalN}), 4.40 (d, $J_{gem} = 11.0$ Hz, 1 H, PhC H_2), 4.35–4.32 (m, 3 H, 1-H^{Glc}, PhC H_2), 4.26 (dd, $J_{3,4} = 2.9$ Hz, 1 H, 3-H^{GalN}), 3.99 (dd, $J_{5,6a} = 0.5$ Hz, $J_{gem} =$ 11.0 Hz, 1 H, 6a-H^{GalN}), 3.80-3.75 (m, 2 H, 4-H^{GalN}, Me₃SiCH₂CH₂), 3.70-3.66 (m, 3 H, $6b-H^{\it GalN}$, $6a-H^{\it Glc}$, $6b-H^{\it Glc}$), 3.62 (s, 3 H, OMe), 3.55 (dd, 1 H, $5-H^{\it GalN}$), 3.45-3.37 (m, 6 H, $4-H^{Glc}$, $5-H^{Glc}$, OMe, Me₃SiCH₂CH₂), 1.97 (s, 3 H, Ac), 1.86 (s, 3 H, Ac), 0.84–0.70 (m, 2 H, $Me_3SiCH_2CH_2$), -0.01 (s, 9 H, $Me_3SiCH_2CH_2$); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.1$, 169.6, 166.4, 165.9, 146.8, 137.4, 137.0, 128.5, 128.4, 128.0, 127.9, 127.8, 127.7, 99.9, 98.4, 76.1, 74.9, 74.7, 74.6, 73.3, 72.8, 71.9, 71.5, 70.3, 67.8, 67.3, 63.7, 51.5, 20.8, 20.8, 17.9, -1.3. HRMS (ESI-TOF): m/z calcd for $C_{45}H_{57}NO_{14}Si$: 886.3441 [M+Na]⁺; found: 886.3439.

2-(Trimethylsilyl)ethyl

(4,6-di-*O*-acetyl-3-*O*-benzyl-2-deoxy-2-phthalimide-D-galactopyranosyl)-(1→6)-2,3-di-*O*-ac etyl-4-*O*-benzyl-β-D-glucopyranoside (38). The glycosidation of 15 (95 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:3 EtOAc–*n*-hexane) to give 38α (5 mg, 5%) and 38β (94 mg, 92%) as a colorless viscous

compound. **38a:** $[\alpha]_D + 111.5$ (c 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.96 - 7.16$ (m, 14 H, NPhth, 2 Ph), 5.68 (d, $J_{3,4} = 3.0$ Hz, 1 H, 4-H^{GalN}), 5.32 (dd, $J_{2,3} = 11.5$ Hz, 1 H, 3-H^{GalN}), 5.10 (d, $J_{1,2} = 3.5$ Hz, 1 H, 1-H^{GalN}), 5.02 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.75 (d, $J_{gem} = 10.0$ Hz, 1 H, PhC H_2), 4.69, 4.66 (2 d, $J_{gem} = 11.5$ Hz, 2 H, PhC H_2), 4.62 (dd, 1 H, 2-H^{GalN}), 4.48 (d, 1 H, PhC H_2), 4.21 (d, $J_{1,2} = 8.0$ Hz, 1 H, 1-H Glc), 4.20–4.16 (m, 2 H, 5-H GalN , 6a-H GalN), 4.11 (dd, $J_{5,6b} = 9.0 \text{ Hz}, J_{\text{gem}} = 13.0 \text{ Hz}, 1 \text{ H}, 6\text{b-H}^{GalN}), 4.06 \text{ (dd, 1 H, 2-H}^{Glc}), 3.95 \text{ (dd, } J_{5,6a} = 2.5 \text{ Hz}, J_{\text{gem}} = 1.0 \text{ Hz},$ = 12.0 Hz, 1 H, 6a-H^{Glc}), 3.68–3.64 (m, 2 H, 4-H^{Glc}, 6b-H^{Glc}), 3.59 (m, 1 H, Me₃SiCH₂C H_2), 3.26–3.21 (m, 2 H, 5-H^{Glc}, Me₃SiCH₂CH₂), 2.16 (s, 3 H, Ac), 2.08 (s, 3 H, Ac), 1.93 (s, 3 H, Ac), 1.90 (s, 3 H, Ac), 0.76–0.60 (m, 2 H, Me₃SiCH₂CH₂), -0.04 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR CDCl₃): (125)MHz, $\delta = 170.5, 170.4, 170.2, 169.2, 168.9, 168.6, 138.1, 137.5, 134.5, 134.0, 132.3, 131.0, 128.5, 12$ 8.3, 128.2, 127.9, 127.9, 127.7, 123.4, 123.2, 99.8, 99.7, 77.6, 75.5, 75.3, 74.9, 74.2, 72.1, 71.2, 70.5, 67.4, 66.7, 66.5, 66.3, 62.5, 52.1, 20.9, 20.8, 20.8, 20.7, 17.6, -1.4. HRMS (ESI-TOF): *m/z* calcd for $C_{47}H_{57}NO_{16}Si: 942.3339 [M+Na]^+$; found: 942.3338. **38β:** $[\alpha]_D + 18.8$ (c 1.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.81-6.90$ (m, 14 H, NPhth, 2 Ph), 5.59 (d, $J_{3,4} = 3.0$ Hz, 1 H, 4-H^{GalN}), 5.28 (d, $J_{1,2}$ = 8.5 Hz, 1 H, 1-H^{GalN}), 5.10 (t, $J_{2,3}$ = $J_{3,4}$ = 9.0 Hz, 1 H, 3-H^{Glc}), 4.76 (dd, $J_{1,2} = 8.0 \text{ Hz}$, 1 H, 2-H^{Glc}), 4.59 (d, $J_{\text{gem}} = 12.0 \text{ Hz}$, 1 H, PhC H_2), 4.46 (dd, $J_{2,3} = 11.0 \text{ Hz}$, 1 H, 2-H^{GalN}), 4.38-4.29 (m, 4 H, 1-H^{Glc}, 3-H^{GalN}, PhCH₂), 4.27-4.19 (m, 3 H, 6a-H^{GalN}, 6b-H^{GalN}, PhC H_2), 4.04 (d, $J_{gem} = 11.0 \text{ Hz}$, 1 H, 6a-H^{Glc}), 3.96 (t, $J_{5,6a} = J_{5,6b} = 6.5 \text{ Hz}$, 1 H, 5-H^{GalN}), 3.83 (m, 1 H, Me₃SiCH₂CH₂), 3.69 (dd, $J_{5.6b} = 4.5$ Hz, $J_{gem} = 11.0$ Hz, 1 H, 6b-H^{Glc}), 3.46–3.41 (m, 3 H, 4-H^{Glc}, 5-H^{Glc}, Me₃SiCH₂CH₂), 2.22 (s, 3 H, Ac), 2.08 (s, 3 H, Ac), 1.98 (s, 3 H, Ac), 1.87 (s, 3 H, Ac), 0.88–0.74 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me₃SiCH₂CH₂); ¹³C NMR (125

MHz, CDCl₃): δ = 176.3, 170.4, 170.4, 170.0, 169.5, 168.1, 167.4, 137.2, 137.1, 133.9, 133.9, 133.7, 132.2, 131.5, 131.5, 129.8, 129.3, 128.3, 128.1, 127.9, 127.8, 127.8, 127.6, 127.6, 123.3, 123.1, 99.8, 98.2, 77.2, 76.2, 75.0, 74.5, 74.4, 72.9, 72.0, 71.2, 71.1, 67.4, 67.0, 65.5, 62.0, 52.6, 28.5, 20.8, 20.7, 20.6, 17.7, -1.4. HRMS (ESI-TOF): m/z calcd for $C_{47}H_{57}NO_{16}Si$: 942.3339 $[M+Na]^+$; found: 942.3340.

2-(Trimethylsilyl)ethyl

 $(3-O-benzyl-2-deoxy-4,6-O-di-tert-butylsilylene-2-phthalimide-\beta-D-glucopyranosyl)-(1\rightarrow 6)$ **2,3-di-***O*-acetyl-4-*O*-benzyl-β-D-glucopyranoside (39). The glycosidation of 16 (104 mg, 0.165 mmol) with 23 (50 mg, 0.111 mmol) in CH₂Cl₂ (2.8 mL) was performed according to the typical procedure. The glycosylated products were purified by silica gel column chromatography (1:20→1:5 EtOAc-toluene) to give 39 (105 mg, 97%) as a colorless viscous compound. $[\alpha]_D$ +2.3 (c 2.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 7.80-6.86$ (m, 14 H, Phth, 2 Ph), 5.34 (d, $J_{1,2} = 8.5$ Hz, 1 H, 1-H^{GlcN}), 5.08 (t, $J_{2,3} = J_{3,4} = 9.5$ Hz, 1 H, 3-H^{Glc}), 4.85 (d, $J_{\text{gem}} = 12.5 \text{ Hz}, 1 \text{ H}, \text{PhC}H_2$, 4.76 (dd, $J_{1,2} = 8.0 \text{ Hz}, 1 \text{ H}, 2\text{-H}^{Glc}$), 4.61 (d, 1 H, PhC H_2), 4.36 (d, 1 H, 1-H^{Glc}), 4.31 (d, J_{gem} = 11.0 Hz, 1 H, PhC H_2), 4.29 (d, 1 H, PhC H_2), 4.24 (dd, $J_{5.6a}$ = 5.0 Hz, $J_{\text{gem}} = 10.5 \text{ Hz}, 1 \text{ H}, 6a-H^{GlcN}), 4.19 \text{ (dd, } J_{2,3} = 10.0 \text{ Hz}, 1 \text{ H}, 2-H^{GlcN}), 4.13 \text{ (t, } J_{3,4} = 9.0 \text{ Hz}, 1 \text{ H},$ 3-H^{GlcN}), 4.08 (t, $J_{4,5} = 9.0$ Hz, 1 H, 4-H^{GlcN}), 4.00 (t, $J_{5,6b} = 10.0$ Hz, 1 H, 6b-H^{GlcN}), 3.96 (dd, $J_{5.6a} = 1.5 \text{ Hz}, J_{\text{gem}} = 11.5 \text{ Hz}, 1 \text{ H}, 6a\text{-H}^{Glc}), 3.84 \text{ (m, 1 H, Me}_3 \text{SiCH}_2 \text{C} H_2), 3.69 \text{ (dd, } J_{5.6b} = 5.5 \text{ (dd, } J_{5.6b} =$ Hz, 1 H, 6b-H^{Glc}), 3.58 (m, 1 H, 5-H^{GlcN}), 3.47–3.42 (m, 2 H, Me₃SiCH₂C H_2 , 4-H^{Glc}), 3.38 (m, 1 H, 5-H^{Glc}), 1.99 (s, 3 H, Ac), 1.87 (s, 3 H, Ac), 1.11, 1.08 (2 s, 18 H, 2 Bu), 0.86–0.75 (m, 2 H, Me₃SiCH₂CH₂), 0.00 (s, 9 H, Me_3 SiCH₂CH₂); ¹³C NMR (125 MHz, CDCl₃): $\delta = 170.1$, 169.6, 138.2, 137.3, 133.8, 131.6, 129.0, 128.4, 128.2, 128.1, 128.0, 127.9, 127.7, 127.3, 123.2, 99.8,

98.2, 79.3, 76.2, 75.1, 74.6, 74.5, 74.1, 72.0, 70.4, 67.3, 67.1, 66.4, 55.2, 27.5, 27.1, 22.7, 20.8, 20.7, 20.0, 17.8, -1.3. HRMS (ESI-TOF): *m/z* calcd for C₅₁H₆₉NO₁₄Si₂: 998.4149 [*M*+Na]⁺; found: 998.4150.

Supporting Information

The Supporting Information is available free of charge on the ACS Publications web site at DOI:

¹H and ¹³C NMR spectra of all new compounds, Tables of atom coordinates and absolute energies to document the theoretical calculations. (PDF)

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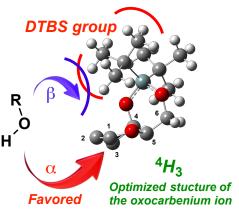
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TOC graphic



Requirements for high α -selectivity

- 1. 4,6-O-Silylene structure
- 2. Galacto-type glycosyl donor
- 3. Generation of oxocarbenium ions
- 4. Through-space electron donation
- 5. Steric hindrance from bulky substituents