

Note

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# Copper-Catalyzed Enantioselective Hetero-Diels-Alder

## Reaction of Danishefsky's Diene with Glyoxals

Yanan Li, Yanbin Hu, Sheng Zhang, Jianan Sun, Lijun Li, Zhenggen Zha, Zhiyong Wang\*

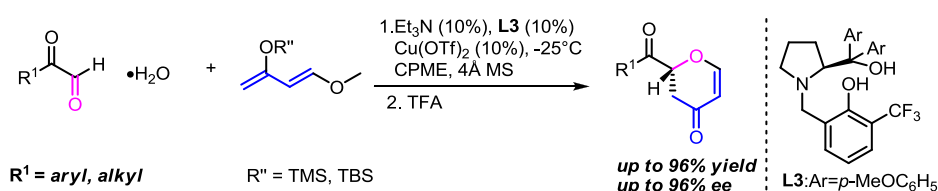
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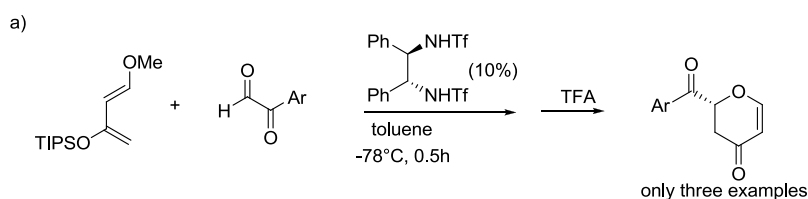


**ABSTRACT:** The highly enantioselective hetero-Diels-Alder reaction of Danishefsky's diene with glyoxals was developed by virtue of a readily accessible chiral copper catalyst. This efficient transformation provided a facile and scalable access to a wide range of biologically active dihydropyrones with high level of enantioselectivities. Moreover, the substrate scope of this reaction could be extended to isatins with this catalytic system. More importantly, the mechanism involved in this reaction was proposed on the basis of the unambiguous structures of intermediates.

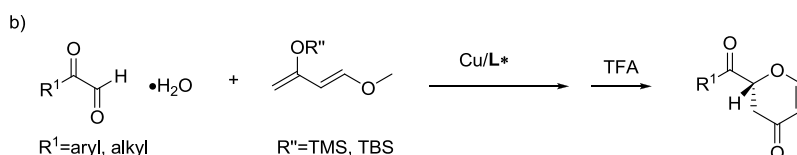
The asymmetric hetero-Diels-Alder (HDA) reaction of Danishefsky's diene with aldehydes or  $\alpha$ -ketoesters is a very important method to synthesize optically active dihydropyrones,<sup>1</sup> which are one of the most privileged substructures in natural products and medicinal compounds.<sup>2</sup> In this area, various catalytic systems, including Lewis acid catalysts,<sup>3</sup> Brønsted acid catalysts,<sup>4</sup> have been extensively exploited to obtain a diversity of chiral dihydropyrones. The Jørgensen group

reported the first highly enantioselective HDA reaction of Danishefsky's diene with ketones catalyzed by chiral copper (II) complex.<sup>5</sup> After that, the HDA reaction of  $\alpha$ -carbonyl esters was further developed by Loh.<sup>6</sup> Recently, a chiral 3,3-dibromo-BINOL-Zn complex was found to be an efficient catalyst in various HDA reactions by Ding et al.<sup>7</sup> Moreover, an impressive work in the HDA reaction of substituted Danishefsky's diene with aldehydes had been achieved in Feng group by using a chiral N,N-Dioxide/In(OTf)<sub>3</sub> catalyst.<sup>8</sup> However, to the best of our knowledge, the HDA reaction of Danishefsky's diene with glyoxals has received far less attention. In the previous work of Mikami, a chiral bis-trifluoromethanesulfonylamide was first employed to catalyze this HDA reaction of glyoxals (Scheme 1a).<sup>9</sup> Nevertheless, there remains a great challenge to develop a general method with broader substrate scope. In this context, we envisioned that our previously used catalyst, which showed high performance in the HDA reaction of  $\beta,\gamma$ -unsaturated  $\alpha$ -ketoesters,<sup>3h</sup> might be applied in this reaction. Indeed, various of dihydropyrones were provided with excellent enantioselectivities (up to 96% *ee*) and high yields (up to 96%) in the presence of this efficient catalyst.

#### Mikami's work



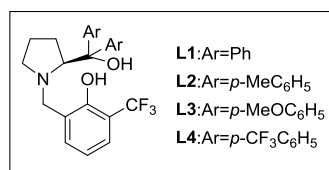
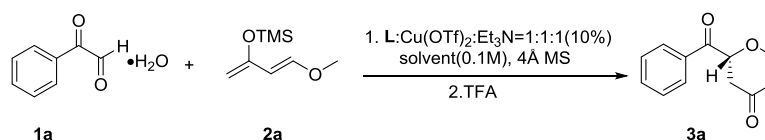
#### This work



**Scheme 1. Previous study and this work on Hetero-Diels-Alder reaction of Danishefsky's diene with glyoxals.**

First of all, we chose the 2-oxo-2-phenylacetaldehyde hydrate **1a** and Danishefsky's diene **2a** as the model substrate. Inspired by the previous work in our laboratory,<sup>3h,10</sup> the chiral **L1**-Cu complex was utilized to optimize this reaction (Table 1). Firstly, different solvents were screened for this reaction. The results showed that ethers favored this reaction (entries 1-5). Then, different ethers were further optimized and CPME proved to be the optimal solvent in terms of yield and enantioselectivity (entry 8). Afterwards, different reaction temperature was examined (entries 8-12). When the reaction was performed under -25 °C, a moderate yield and excellent enantioselectivity were obtained. Lowering the temperature to -35 °C resulted in a significantly decrease in the reaction yield, albeit a slightly improved enantiomeric excess was observed (entries 11-12). The substitutions of aromatic ring in the chiral ligand were next explored. It was found that the ligand bearing an electron-enriched aryl ring could give a better performance compared with the electron-deficient one (entries 13-15). However, a satisfied result still could not be obtained. Considering the detrimental effect of the crystal water on the stereoselectivity of this reaction, different water scavengers were screened (entries 16-18). To our delight, not only the enantioselectivity was significantly improved to 95% *ee* but also the yield was remarkably increased to 86% when 4Å molecular sieve was added to the reaction system (entry 16). As a result, the optimal reaction condition was determined as below: CPME as the reaction solvent, 4Å molecular sieve as the additive, the HDA reaction being carried out under catalysis of a facile copper complex at -25 °C.

**Table 1. Optimization of the reaction conditions<sup>a</sup>**



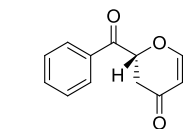
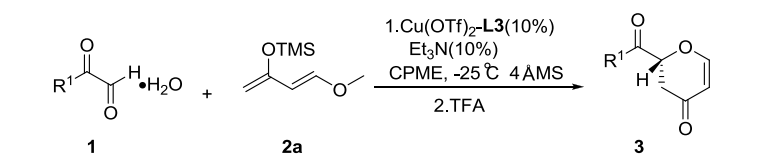
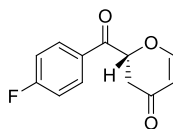
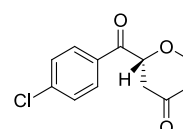
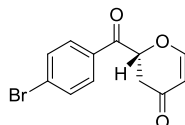
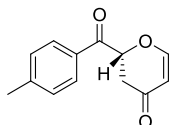
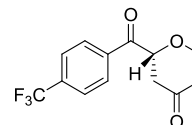
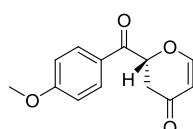
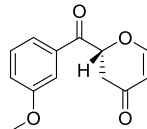
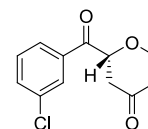
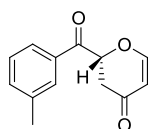
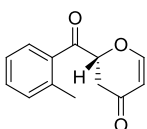
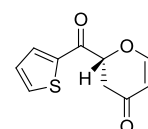
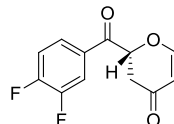
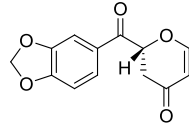
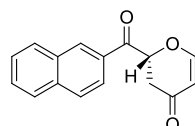
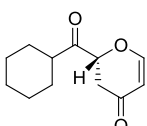
entry	ligand	solvent	T (°C)	yield <sup>b</sup> (%)	ee <sup>c</sup> (%)
1	<b>L1</b>	EtOAc	-15	79	73
2	<b>L1</b>	Toluene	-15	64	77
3	<b>L1</b>	CH <sub>2</sub> Cl <sub>2</sub>	-15	89	69
4	<b>L1</b>	CHCl <sub>3</sub>	-15	74	73
5	<b>L1</b>	MTBE	-15	69	77
6	<b>L1</b>	THF	-15	59	65
7	<b>L1</b>	Et <sub>2</sub> O	-15	50	75
8	<b>L1</b>	CPME	-15	89	78
9	<b>L1</b>	CPME	-20	80	78
10	<b>L1</b>	CPME	-25	75	79
11	<b>L1</b>	CPME	-30	63	80
12	<b>L1</b>	CPME	-35	60	81
13	<b>L2</b>	CPME	-25	75	80
14	<b>L4</b>	CPME	-25	74	75
15	<b>L3</b>	CPME	-25	78	81
16 <sup>d</sup>	<b>L3</b>	CPME	-25	86	95
17 <sup>e</sup>	<b>L3</b>	CPME	-25	81	91
18 <sup>f</sup>	<b>L3</b>	CPME	-25	83	91

<sup>a</sup>Unless otherwise noted, all reactions were performed with **1a** (0.1 mmol), **2a** (0.2 mmol), **L** (10 mol%), Et<sub>3</sub>N (10 mol%) and Cu(OTf)<sub>2</sub> (10 mol%). <sup>b</sup>Isolated yield. <sup>c</sup>Determined by chiral HPLC analysis. <sup>d</sup>Addition of 50 mg of powder 4Å molecular sieve. <sup>e</sup>Addition of 50 mg of Na<sub>2</sub>SO<sub>4</sub>. <sup>f</sup>Addition of 50 mg of MgSO<sub>4</sub>. Tf = trifluoromethanesulfonyl. MTBE = Methyl *tert*-butyl ether. CPME = Cyclopentyl methyl ether.

With the optimal conditions in hand, the substrate scope of 2-oxo-2-phenylacetaldehyde hydrate was examined, as shown in Table 2. Firstly, the electronic effect of the substrates was investigated.

When the substrates bearing the strong electron-donating groups on the *para*-position of the phenyl ring were employed, the HDA reactions could be carried out smoothly to give the desired products with high yields and excellent enantioselectivities (**3e**, **3g**). Similarly, the electron-withdrawing halogen groups were compatible with the reaction system (**3b**, **3c**, **3d**). The HDA adduct **3f** was obtained with good yield and excellent enantioselectivity when the strong electron-withdrawing trifluoromethyl group was on the *para*-position of the phenyl ring. It was obvious that the substrates with the electron-donating groups could give higher yields than those with the electron-withdrawing groups. Then, the steric effect of the reaction was examined. The results showed that substitutions at different positions on the phenyl group had little influence on the reaction yields and enantioselectivities (**3e**, **3j**, **3k**). In the case of heterocyclic product **3l**, 2-thienyl-dihydropyrone, a satisfactory result could be obtained by simply lowering the reaction temperature to -35 °C. The substrates bearing 2-naphthyl group and multisubstituted group could also give the corresponding products with high yields and enantioselectivities, respectively (**3m-3o**). Notably, when the R<sup>1</sup> group was changed to the aliphatic group, the desired product **3p** was still obtained in moderate yield and excellent enantioselectivity. The absolute configuration of the product **3d** was confirmed by X-ray crystal diffraction.<sup>11</sup>

**Table 2. Scope of glyoxals<sup>a,b,c</sup>**

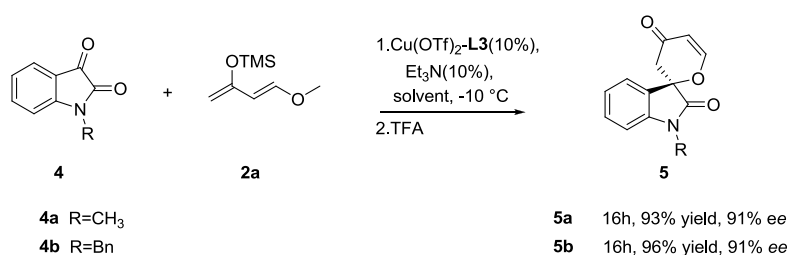
**3a** 16h, 86% yield, 95% ee**3b** 20h, 77% yield, 91% ee**3c** 20h, 84% yield, 94% ee**3d** 20h, 85% yield, 94% ee**3e** 16h, 84% yield, 96% ee**3f** 21h, 71% yield, 90% ee**3g** 16h, 96% yield, 94% ee**3h** 16h, 90% yield, 92% ee**3i** 19h, 87% yield, 91% ee**3j** 17h, 92% yield, 94% ee**3k** 24h, 85% yield, 90% ee<sup>d</sup>**3l** 19h, 80% yield, 90% ee**3m** 20h, 85% yield, 92% ee**3n** 19h, 65% yield, 93% ee**3o** 19h, 92% yield, 93% ee**3p** 24h, 61% yield, 90% ee<sup>d</sup>

<sup>a</sup>Unless otherwise noted, the reaction of **1** (0.3 mmol) and **2a** (0.6 mmol) was performed in the presence of **L3** (10 mol%), Et<sub>3</sub>N (10 mol%) and Cu(OTf)<sub>2</sub> (10 mol%) in CPME (3.0 mL) with 150 mg of powder 4 Å molecular sieve as additive at -25 °C. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by chiral HPLC analysis. <sup>d</sup>In MTBE (3.0 mL) at -35 °C.

A variety of dihydropyrones were obtained with good yields and enantioselectivities through the catalysis of the copper complex. Next, we wondered whether the catalytic system could promote

the asymmetric HDA reaction of isatins. Pleasingly, when isatins were evaluated in this reaction, the desired products **5a** and **5b** could be obtained with high yields and excellent enantioselectivities under this catalytic system (Scheme 2). And **5b** was a known compound, the absolute configuration of **5b** was determined by comparison with corresponding specific rotation data reported in literature.<sup>8e</sup>

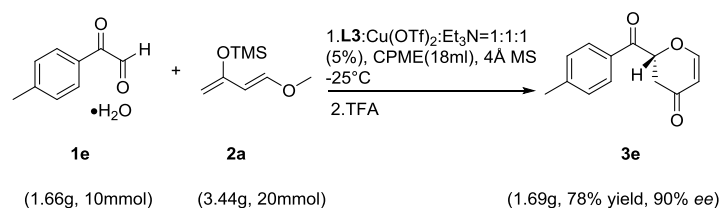
### Scheme 2. Scope of Isatins<sup>a,b,c</sup>



<sup>a</sup>Unless otherwise noted, the reactions of **4** (0.3mmol) and **2a** (0.6mmol) were performed in the presence of **L3** (10 mol%),  $\text{Et}_3\text{N}$  (10 mol%),  $\text{Cu}(\text{OTf})_2$  (10 mol%), R = CH<sub>3</sub> in MTBE (3.0 mL) and R = Bn in Toluene (3.0 mL) at  $-10^\circ\text{C}$ . <sup>b</sup>Isolated yields. <sup>c</sup>Determined by chiral HPLC analysis.

To further demonstrate the robust nature and operational simplicity of this methodology, a gram scale experiment was carried out (Scheme 3). The copper complex of 5% catalytic loading could catalyze the HDA reaction of **1e** and Danishefsky's diene to generate the desired product **3e** with 1.69g quantity, 78% yield and 90% *ee*.

### Scheme 3. Asymmetric Hetero-Diels-Alder Reaction on a gram scale<sup>a,b,c</sup>



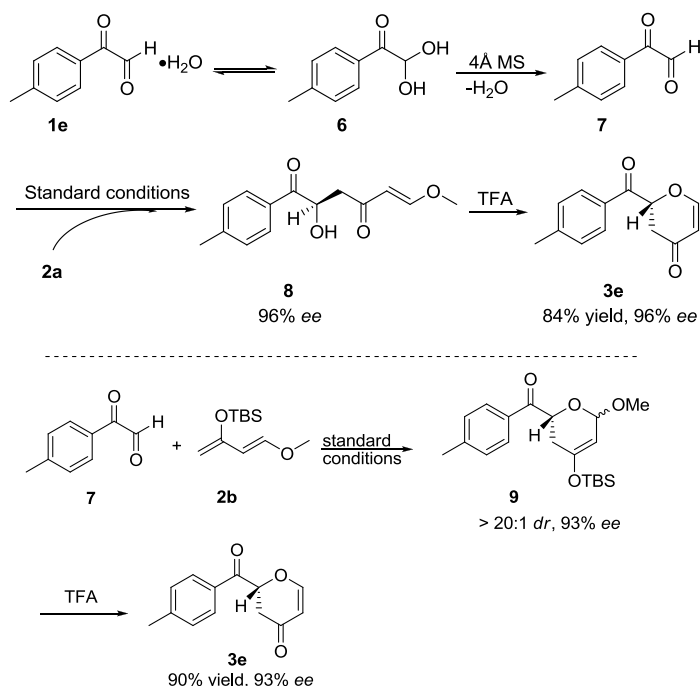


<sup>a</sup>The reaction of **1e** (10 mmol) and **2a** (20 mmol) was performed in the presence of **L3** (5 mol%), Et<sub>3</sub>N (5 mol%) and Cu(OTf)<sub>2</sub> (5 mol%) in CPME (18 mL) with 2.5g of powder 4Å molecular sieve as additive at -25 °C. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by chiral HPLC analysis.

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Finally, the reaction mechanism was studied. Generally, two possible reaction pathways, Diels-Alder cycloaddition<sup>12</sup> and stepwise Mukaiyama-aldol condensation,<sup>13</sup> were involved in this transformation. As shown in Scheme 4, in this reaction, 2-oxo-2-phenylacetaldehyde hydrate gradually released the phenylglyoxal **7** when 4Å molecular sieve was added in the reaction system.<sup>14</sup> With this strategy, the *in situ* generated phenylglyoxal immediately reacted with Danishefsky's diene in the presence of the chiral copper catalyst.

In order to get insight into the reaction mechanism, several control experiments were carried out (Scheme 4). For the HDA reaction of Danishefsky's diene **2a** with **1e**, the intermediate **8** was isolated before the treatment of TFA. The NMR spectral experiment of the intermediate **8** showed that the Mukaiyama-aldol adduct was produced under the reaction condition, corresponding to the stepwise pathway. As expected, the *ee* value of the Mukaiyama-aldol adduct was well consistent with that of final product, which indicated that TFA acidification had no influence on the stereocontrol. However, when the TMS group of Danishefsky's diene was changed to a TBS group, only the cycloaddition adduct **9**, which was confirmed by NMR, was isolated by silica gel chromatography. This cycloaddition intermediate could be easily transferred into the final product by the treatment of TFA. Similarly, the *ee* value was maintained after TFA acidification. In view of the experimental results above, the different reaction pathways involved in this reaction were greatly dependent on the substrate dienes.

**Scheme 4. The mechanism pathway of the Hetero-Diels-Alder Reaction**

In summary, a copper catalyzed asymmetric hetero-Diels-Alder reaction of Danishefsky's diene with glyoxals was developed under mild reaction conditions. With this efficient methodology, an unprecedented substrate scope was achieved and a variety of dihydropyranones were provided with excellent enantioselectivities and high yields. Moreover, a detailed reaction mechanism study showed that a substrate-dependent pathway, stepwise Mukaiyama-aldol condensation and concerted cycloaddition, was involved in this reaction.

## Experiment Section

**General Information:** <sup>1</sup>H NMR and <sup>13</sup>C NMR were recorded on a 400MHz Nuclear Magnetic Resonance Spectrometer (<sup>1</sup>H NMR: 400MHz, <sup>13</sup>C NMR: 100MHz) using TMS as internal reference. The chemical shifts (δ) and coupling constants (*J*) were expressed in ppm and Hz, respectively. UV-Vis Spectrophotometry was carried out on infrared spectrometer. HPLC analysis was carried out on HPLC with a multiple wavelength detector by commercial chiral columns. Optical rotations were measured on a Polarimeter. HRMS (ESI) were recorded on a Q-TOF

Premier. Commercially available compounds were used without further purification. Solvents were purified according to the standard procedures unless otherwise noted. Ligands<sup>10a, 10c</sup>, various glyoxals<sup>14a</sup>, *N*-methyl isatins<sup>10d</sup>, *N*-benzyl isatins<sup>15</sup>, and Danishefsky's diene<sup>16</sup> were prepared according to literature procedures.

**General procedures of Hetero-Diels-Alder reaction:** A mixture of Ligand (**L3**, 10 mol %), Cu(OTf)<sub>2</sub> (10 %, 10.8 mg) and Et<sub>3</sub>N (10 %, 4.17  $\mu$ L) in corresponding solvent (3.0 mL) was stirred for 1h at ambient atmosphere, corresponding glyoxals (0.3 mmol) and 150 mg of powder 4Å molecular sieve as additive were then added. The resulting mixture was cooled to -25 °C. After 30min, corresponding diene was added slowly by syringe. After reactions were finished (monitored by TLC), 5.0 equiv. TFA was added to quench the reaction. The system was quenched by saturated sodium bicarbonate after 2h and then extracted by ethyl acetate. The organic phase was dried with sodium sulfate, evaporated in vacuo. Purification by column chromatograph afforded HDA adducts.

#### Experimental data of HDA adducts

##### (**R**)-2-benzoyl-2H-pyran-4(3H)-one (**3a**):

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 86% yield; mp = 127-129 °C;  $[\alpha]_D^{20}$  -156.7 (c = 0.84, CHCl<sub>3</sub>, 95% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 0.8 mL/min, T = 23 °C, UV = 254 nm,  $t_R$  = 10.56 min (major),  $t_R$  = 29.09 min (minor); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.96 (d, *J* = 7.3 Hz, 2H), 7.63 (t, *J* = 7.4 Hz, 1H), 7.50 (t, *J* = 7.8 Hz, 2H), 7.40 (d, *J* = 6.1 Hz, 1H), 5.81 (dd, *J* = 11.4 Hz, 4.6 Hz, 1H), 5.49 (d, *J* = 6.04 Hz, 1H), 2.96 (dd, *J* = 17.1 Hz, 11.5 Hz, 1H), 2.80 (dd,

$J = 17.0$  Hz,  $4.5$  Hz,  $1\text{H}$ );  $^{13}\text{C}$  NMR ( $100$  MHz,  $\text{CDCl}_3$ ):  $\delta$   $193.1$ ,  $190.0$ ,  $161.7$ ,  $134.3$ ,  $133.7$ ,  $129.00$ ,  $128.99$ ,  $107.9$ ,  $78.9$ ,  $38.0$ ; IR (film,  $\text{v}/\text{cm}^{-1}$ ):  $3687$ ,  $2923$ ,  $2856$ ,  $2364$ ,  $1959$ ,  $1751$ ,  $1461$ ,  $1263$ ,  $1068$ ,  $802$ ,  $682$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_{10}\text{O}_3$   $[\text{M}+\text{Na}]^+$   $225.0528$ , found  $225.0523$ .

**(R)-2-(4-fluorobenzoyl)-2H-pyran-4(3H)-one (3b):**

The title compound was prepared according to the general working procedure (20h) and purified by column chromatography (petroleum ether / ethyl acetate =  $3:1$ ) to give the product as a light yellow solid:  $77\%$  yield; mp =  $146\text{--}148$  °C;  $[\alpha]_{\text{D}}^{20}$   $-172.9$  ( $c = 0.84$ ,  $\text{CHCl}_3$ ,  $91\%$  ee); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol =  $70:30$ , flow rate =  $1.0$  mL/min,  $T = 23$  °C, UV =  $254$  nm,  $t_{\text{R}} = 10.12$  min (minor),  $t_{\text{R}} = 11.75$  min (major);  $^1\text{H}$  NMR ( $400$  MHz,  $\text{CDCl}_3$ ):  $\delta$   $8.02$  (dd,  $J = 8.6$  Hz,  $5.4$  Hz,  $2\text{H}$ ),  $7.38$  (d,  $J = 6.0$  Hz,  $1\text{H}$ ),  $7.18$  (t,  $J = 8.5$  Hz,  $2\text{H}$ ),  $5.75$  (dd,  $J = 11.2$  Hz,  $4.4$  Hz,  $1\text{H}$ ),  $5.50$  (d,  $J = 6.0$  Hz,  $1\text{H}$ ),  $2.98$  (dd,  $J = 17.0$  Hz,  $11.3$  Hz,  $1\text{H}$ ),  $2.79$  (dd,  $J = 17.0$  Hz,  $4.4$  Hz,  $1\text{H}$ );  $^{13}\text{C}$  NMR ( $100$  MHz,  $\text{CDCl}_3$ ):  $\delta$   $191.6$ ,  $190.0$ ,  $166.3$  ( $^1J_{\text{CF}} = 255.9$  Hz,  $1\text{C}$ ),  $161.4$ ,  $131.9$  ( $^3J_{\text{CF}} = 9.5$  Hz,  $2\text{C}$ ),  $130.2$  ( $^4J_{\text{CF}} = 2.9$  Hz,  $1\text{C}$ ),  $116.2$  ( $^2J_{\text{CF}} = 21.9$  Hz,  $2\text{C}$ ),  $108.0$ ,  $78.8$ ,  $37.8$ ; IR (film,  $\text{v}/\text{cm}^{-1}$ ):  $3781$ ,  $3706$ ,  $3461$ ,  $2923$ ,  $2856$ ,  $1689$ ,  $1514$ ,  $1461$ ,  $1375$ ,  $1270$ ,  $1222$ ,  $1031$ ,  $859$ ,  $743$ ,  $601$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_9\text{FO}_3$   $[\text{M}+\text{Na}]^+$   $243.0433$ , found  $243.0432$ .

**(R)-2-(4-chlorobenzoyl)-2H-pyran-4(3H)-one (3c):**

The title compound was prepared according to the general working procedure (20h) and purified by column chromatography (petroleum ether / ethyl acetate =  $3:1$ ) to give the product as a light yellow solid:  $84\%$  yield; mp =  $96\text{--}98$  °C;  $[\alpha]_{\text{D}}^{20}$   $-149.9$  ( $c = 1.02$ ,  $\text{CHCl}_3$ ,  $94\%$  ee); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol =  $70:30$ , flow rate =  $1.0$  mL/min,  $T = 23$  °C, UV =  $230$  nm,  $t_{\text{R}} = 10.59$  min (minor),  $t_{\text{R}} = 12.28$  min (major);  $^1\text{H}$  NMR ( $400$  MHz,  $\text{CDCl}_3$ ):  $\delta$   $7.91$  (d,  $J = 8.5$  Hz,  $2\text{H}$ ),  $7.47$  (d,  $J = 8.5$  Hz,  $2\text{H}$ ),  $7.37$  (d,  $J = 6.0$  Hz,  $1\text{H}$ ),  $5.73$  (dd,  $J = 11.1$  Hz,  $4.6$  Hz,  $1\text{H}$ ),  $5.49$  (d,

$J = 6.0$  Hz, 1H), 2.96 (dd,  $J = 17.0$  Hz, 11.2 Hz, 1H), 2.79 (dd,  $J = 17.0$  Hz, 4.5 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  191.0, 188.8, 160.3, 139.8, 131.0, 129.5, 128.3, 107.0, 77.8, 36.7; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3766, 3704, 3459, 2923, 2856, 2370, 1679, 1592, 1452, 1398, 1085, 802, 609; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_9\text{ClO}_3$   $[\text{M}+\text{Na}]^+$  259.0138, found 259.0132.

**(R)-2-(4-bromobenzoyl)-2H-pyran-4(3H)-one (3d):**

The title compound was prepared according to the general working procedure (20h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 85% yield; mp = 107-109 °C;  $[\alpha]_{\text{D}}^{20}$  -129.2 ( $c = 0.52$ ,  $\text{CHCl}_3$ , 94% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 0.8 mL/min,  $T = 23$  °C, UV = 254 nm,  $t_{\text{R}} = 11.71$  min (major),  $t_{\text{R}} = 24.21$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.84 (d,  $J = 8.6$  Hz, 2H), 7.65 (d,  $J = 8.5$  Hz, 2H), 7.38 (d,  $J = 6.0$  Hz, 1H), 5.73 (dd,  $J = 11.1$  Hz, 4.6 Hz, 1H), 5.50 (d,  $J = 6.0$  Hz, 1H), 2.97 (dd,  $J = 17.0$  Hz, 11.1 Hz, 1H), 2.80 (dd,  $J = 17.0$  Hz, 4.5 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  192.3, 189.9, 161.4, 132.4, 132.3, 130.5, 129.7, 108.1, 78.8, 37.7; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3714, 2921, 2856, 1677, 1591, 1459, 1398, 1068, 935, 802, 732, 615; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_9\text{BrO}_3$   $[\text{M}+\text{Na}]^+$  302.9633, found 302.9629.

**(R)-2-(4-methylbenzoyl)-2H-pyran-4(3H)-one (3e):**

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 84% yield; mp = 89-91 °C;  $[\alpha]_{\text{D}}^{20}$  -123.3 ( $c = 0.94$ ,  $\text{CHCl}_3$ , 96% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 0.8 mL/min,  $T = 23$  °C, UV = 254 nm,  $t_{\text{R}} = 11.05$  min (major),  $t_{\text{R}} = 19.61$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.85 (d,  $J = 8.3$  Hz, 2H), 7.39 (d,  $J = 6.0$  Hz, 1H), 7.29 (d,  $J = 8.0$  Hz, 2H), 5.78 (dd,  $J = 11.6$  Hz, 4.6 Hz, 1H), 5.48

(dd,  $J = 6.0$  Hz,  $0.8$  Hz,  $1H$ ),  $2.95$  (dd,  $J = 17.1$  Hz,  $11.7$  Hz,  $1H$ ),  $2.78$  (ddd,  $J = 17.0$  Hz,  $4.5$  Hz,  $0.9$  Hz,  $1H$ ),  $2.42$  (s,  $3H$ );  $^{13}C$  NMR ( $100$  MHz,  $CDCl_3$ ):  $\delta$   $192.6$ ,  $190.2$ ,  $161.7$ ,  $145.5$ ,  $131.2$ ,  $129.7$ ,  $129.1$ ,  $107.8$ ,  $78.9$ ,  $38.1$ ,  $21.8$ ; IR (film,  $\nu/cm^{-1}$ ):  $3781$ ,  $2923$ ,  $2856$ ,  $1679$ ,  $1596$ ,  $1452$ ,  $1403$ ,  $1277$ ,  $1037$ ,  $935$ ,  $798$ ,  $599$ ; HRMS (ESI)  $m/z$  calcd for  $C_{13}H_{12}O_3$   $[M+Na]^+$   $239.0684$ , found  $239.0684$ .

**(R)-2-(4-(trifluoromethyl)benzoyl)-2H-pyran-4(3H)-one (3f):**

The title compound was prepared according to the general working procedure (21h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 71% yield;  $[\alpha]_D^{20}$   $-109.3$  ( $c = 0.68$ ,  $CHCl_3$ , 90% *ee*); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol = 70:30, flow rate =  $1.0$  mL/min,  $T = 23$  °C, UV =  $230$  nm,  $t_R = 9.24$  min (minor),  $t_R = 10.40$  min (major);  $^1H$  NMR ( $400$  MHz,  $CDCl_3$ ):  $\delta$   $8.10$  (d,  $J = 8.2$  Hz,  $2H$ ),  $7.79$  (d,  $J = 8.2$  Hz,  $2H$ ),  $7.38$  (d,  $J = 6.1$  Hz,  $1H$ ),  $5.77$  (dd,  $J = 10.8$  Hz,  $4.6$  Hz,  $1H$ ),  $5.53$  (d,  $J = 6.0$  Hz,  $1H$ ),  $3.01$  (dd,  $J = 17.0$  Hz,  $10.8$  Hz,  $1H$ ),  $2.85$  (dd,  $J = 17.1$  Hz,  $4.5$  Hz,  $1H$ );  $^{13}C$  NMR ( $100$  MHz,  $CDCl_3$ ):  $\delta$   $191.6$ ,  $188.7$ ,  $160.2$ ,  $135.5$ ,  $134.3$  (q,  $^2J_{CF} = 32.8$  Hz,  $1C$ ),  $128.5$ ,  $125.0$  (q,  $^4J_{CF} = 3.6$  Hz,  $2C$ ),  $122.3$  (q,  $^1J_{CF} = 271.2$  Hz,  $1C$ ),  $107.2$ ,  $77.9$ ,  $36.5$ ; IR (film,  $\nu/cm^{-1}$ ):  $3704$ ,  $3363$ ,  $2923$ ,  $2856$ ,  $2724$ ,  $2279$ ,  $1936$ ,  $1697$ ,  $1598$ ,  $1457$ ,  $1332$ ,  $1267$ ,  $1135$ ,  $1068$ ,  $1025$ ,  $966$ ,  $856$ ,  $804$ ,  $727$ ,  $597$ ; HRMS (ESI)  $m/z$  calcd for  $C_{13}H_9F_3O_3$   $[M+Na]^+$   $293.0401$ , found  $293.0401$ .

**(R)-2-(4-methoxybenzoyl)-2H-pyran-4(3H)-one (3g):**

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 96% yield; mp =  $122-124$  °C;  $[\alpha]_D^{20}$   $-116.9$  ( $c = 1.10$ ,  $CHCl_3$ , 94% *ee*); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol = 70:30, flow rate =  $1.0$  mL/min,  $T = 23$  °C, UV =  $254$  nm,  $t_R = 12.99$  min (minor),  $t_R = 14.51$  min (major);  $^1H$  NMR ( $400$  MHz,  $CDCl_3$ ):  $\delta$   $7.95$  (d,  $J$

= 9.2 Hz, 2H), 7.39 (d,  $J$  = 6.0 Hz, 1H), 6.96 (d,  $J$  = 8.8 Hz, 2H), 5.76 (dd,  $J$  = 11.8 Hz, 4.4 Hz, 1H), 5.48 (d,  $J$  = 6.0 Hz, 1H), 3.88 (s, 3H), 2.96 (dd,  $J$  = 17.0 Hz, 11.8 Hz, 1H), 2.76 (dd,  $J$  = 17.0 Hz, 4.3 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  191.3, 190.3, 164.4, 161.7, 131.4, 126.6, 114.2, 107.8, 78.8, 55.6, 38.2; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 2923, 2856, 2724, 1747, 1681, 1598, 1457, 1375, 1261, 1168, 1091, 1027, 804, 599; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{12}\text{O}_4$   $[\text{M}+\text{Na}]^+$  255.0633, found 255.0639.

**(R)-2-(3-methoxybenzoyl)-2H-pyran-4(3H)-one (3h):**

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 90% yield; mp = 71-73 °C;  $[\alpha]_{\text{D}}^{20}$  -115.3 ( $c$  = 0.86,  $\text{CHCl}_3$ , 92% *ee*); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T$  = 23 °C, UV = 240 nm,  $t_{\text{R}}$  = 12.03 min (major),  $t_{\text{R}}$  = 13.77 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.51-7.38 (m, 4H), 7.17-7.15 (m, 1H), 5.79 (dd,  $J$  = 11.4 Hz, 4.6 Hz, 1H), 5.48 (d,  $J$  = 6.0 Hz, 1H), 3.84 (s, 3H), 2.94 (dd,  $J$  = 17.0 Hz, 11.4 Hz, 1H), 2.79 (dd,  $J$  = 17.0 Hz, 4.6 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  191.9, 189.0, 160.7, 159.0, 133.9, 128.9, 120.4, 119.7, 112.3, 106.8, 77.9, 54.5, 37.0; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3704, 3363, 2923, 2856, 2726, 2364, 2067, 1936, 1687, 1594, 1457, 1375, 1265, 1203, 1033, 873, 798; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{12}\text{O}_4$   $[\text{M}+\text{Na}]^+$  255.0633, found 255.0634.

**(R)-2-(3-chlorobenzoyl)-2H-pyran-4(3H)-one (3i):**

The title compound was prepared according to the general working procedure (19h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 87% yield; mp = 115-117 °C;  $[\alpha]_{\text{D}}^{20}$  -118.6 ( $c$  = 1.02,  $\text{CHCl}_3$ , 91% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T$  = 23 °C, UV = 230 nm,  $t_{\text{R}}$  = 8.28 min (major),  $t_{\text{R}}$  = 16.44 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.93 (s, 1H), 7.84 (d,  $J$  = 7.8 Hz, 1H), 7.61-7.59 (m, 1H), 7.45 (t,  $J$  = 7.9 Hz, 1H), 7.39 (d,  $J$  = 6.0 Hz, 1H), 5.74 (dd,  $J$  = 11.0 Hz, 4.6 Hz, 1H), 5.50 (d,  $J$  = 6.0 Hz, 1H), 2.96 (dd,  $J$  = 17.0 Hz, 11.0 Hz, 1H), 2.81 (dd,  $J$  =

17.0 Hz, 4.5 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  192.1, 189.9, 161.5, 135.31, 135.25, 134.2, 130.3, 129.1, 127.2, 108.1, 78.8, 37.7; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3704, 2923, 2856, 2726, 1687, 1596, 1459, 1267, 1085, 946, 798, 728, 630; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_9\text{ClO}_3$   $[\text{M}+\text{Na}]^+$  259.0138, found 259.0144.

**(R)-2-(3-methylbenzoyl)-2H-pyran-4(3H)-one (3j):**

The title compound was prepared according to the general working procedure (17h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 92% yield;  $[\alpha]_{\text{D}}^{20}$  -124.4 ( $c$  = 0.98,  $\text{CHCl}_3$ , 94% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T$  = 23  $^{\circ}\text{C}$ , UV = 230 nm,  $t_{\text{R}}$  = 8.04 min (major),  $t_{\text{R}}$  = 17.75 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.74 (d,  $J$  = 8.4 Hz, 2H), 7.45-7.37 (m, 3H), 5.81 (dd,  $J$  = 11.6 Hz, 4.6 Hz, 1H), 5.50 (d,  $J$  = 6.0 Hz, 1H), 2.95 (dd,  $J$  = 17.0 Hz, 11.6 Hz, 1H), 2.79 (dd,  $J$  = 17.0 Hz, 4.4 Hz, 1H), 2.42(s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  193.3, 190.0, 161.7, 138.9, 135.1, 133.8, 129.4, 128.8, 126.1, 107.8, 78.9, 38.1, 21.3; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3704, 3627, 3459, 3359, 3187, 2923, 2856, 2732, 2630, 2443, 2258, 2069, 1903, 1685, 1594, 1457, 1376, 1160, 1033, 952, 875, 796, 736; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{12}\text{O}_3$   $[\text{M}+\text{Na}]^+$  239.0684, found 239.0687.

**(R)-2-(2-methylbenzoyl)-2H-pyran-4(3H)-one (3k):**

The title compound was prepared according to the general working procedure (24h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 85% yield; mp = 95-97  $^{\circ}\text{C}$ ;  $[\alpha]_{\text{D}}^{20}$  -155.1 ( $c$  = 0.82,  $\text{CHCl}_3$ , 90% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T$  = 23  $^{\circ}\text{C}$ , UV = 230 nm,  $t_{\text{R}}$  = 8.65 min (major),  $t_{\text{R}}$  = 19.02 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.56 (d,  $J$  = 7.9 Hz,



1H), 7.45-7.40 (m, 1H), 7.35 (d,  $J = 6.1$  Hz, 1H), 7.30 (s, 1H), 7.28 (s, 1H), 5.67 (dd,  $J = 10.5$  Hz, 5.0 Hz, 1H), 5.46 (dd,  $J = 6.1$  Hz, 0.4 Hz, 1H), 2.89 (dd,  $J = 17.0$  Hz, 10.4 Hz, 1H), 2.78 (ddd,  $J = 17.0$  Hz, 5.0 Hz, 0.6 Hz, 1H), 2.45(s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  197.4, 189.9, 161.7, 139.3, 134.3, 132.3, 132.2, 128.5, 125.7, 108.0, 80.1, 37.7, 20.9; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3704, 3459, 2923, 2856, 1679, 1594, 1454, 1403, 1272, 1066, 804, 736, 607; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{12}\text{O}_3$   $[\text{M}+\text{Na}]^+$  239.0684, found 239.0682.

**(R)-2-(thiophene-2-carbonyl)-2H-pyran-4(3H)-one (3l):**

The title compound was prepared according to the general working procedure (19h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 80% yield; mp = 81-82 °C;  $[\alpha]_{\text{D}}^{20}$  -136.4 ( $c = 0.74$ ,  $\text{CHCl}_3$ , 90% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T = 23$  °C, UV = 240 nm,  $t_{\text{R}} = 9.70$  min (major),  $t_{\text{R}} = 22.76$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.91 (d,  $J = 3.6$  Hz, 1H), 7.76 (d,  $J = 4.8$  Hz, 1H), 7.42 (d,  $J = 6.0$  Hz, 1H), 7.18 (t,  $J = 4.4$  Hz, 1H), 5.55 (dd,  $J = 11.8$  Hz, 4.4 Hz, 1H), 5.50 (d,  $J = 6.0$  Hz, 1H), 2.99 (dd,  $J = 17.0$  Hz, 11.8 Hz, 1H), 2.80 (dd,  $J = 17.0$  Hz, 4.4 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  189.9, 186.2, 161.4, 140.0, 135.8, 134.2, 128.6, 108.1, 80.1, 38.3; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 2923, 2856, 2726, 2364, 1751, 1675, 1594, 1459, 1375, 1265, 1218, 908, 802, 727; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{10}\text{H}_8\text{SO}_3$   $[\text{M}+\text{Na}]^+$  231.0092, found 231.0094.

**(R)-2-(3,4-difluorobenzoyl)-2H-pyran-4(3H)-one (3m):**

The title compound was prepared according to the general working procedure (20h) and purified by column chromatography (petroleum ether / ethyl acetate = 2:1) to give the product as a light yellow oil: 85% yield;  $[\alpha]_{\text{D}}^{20}$  -140.4 ( $c = 1.06$ ,  $\text{CHCl}_3$ , 92% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T = 23$  °C, UV = 230 nm,  $t_{\text{R}} = 7.77$  min

(major),  $t_R$  = 21.33 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.87-7.77 (m, 2H), 7.37 (d,  $J$  = 6.0 Hz, 1H), 7.30 (dd,  $J$  = 17.1 Hz, 8.7 Hz, 1H), 5.70 (dd,  $J$  = 10.9 Hz, 4.6 Hz, 1H), 5.50 (d,  $J$  = 6.0 Hz, 1H), 2.98 (dd,  $J$  = 17.1 Hz, 11.0 Hz, 1H), 2.80 (dd,  $J$  = 17.0 Hz, 4.5 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  190.8, 189.8, 161.2, 154.3 (dd,  $^1J_{\text{CF}}$  = 258.0 Hz,  $^2J_{\text{CF}}$  = 13.2 Hz, 1C), 150.6 (dd,  $^1J_{\text{CF}}$  = 250.5 Hz,  $^2J_{\text{CF}}$  = 12.8 Hz, 1C), 130.8 (dd,  $^3J_{\text{CF}}$  = 4.4 Hz,  $^4J_{\text{CF}}$  = 3.7 Hz, 1C), 126.4 (dd,  $^3J_{\text{CF}}$  = 8.0 Hz,  $^4J_{\text{CF}}$  = 3.6 Hz, 1C), 118.7 (dd,  $^3J_{\text{CF}}$  = 18.2 Hz,  $^4J_{\text{CF}}$  = 2.1 Hz, 1C), 118.0 (d,  $J$  = 17.7 Hz, 1C), 108.2, 78.8, 37.6; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3706, 3461, 3072, 2923, 2856, 1683, 1598, 1513, 1459, 1270, 1033, 894, 802, 607; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_8\text{F}_2\text{O}_3$   $[\text{M}+\text{Na}]^+$  261.0339, found 261.0346.

**(R)-2-(benzo[d][1,3]dioxole-5-carbonyl)-2H-pyran-4(3H)-one (3n):**

The title compound was prepared according to the general working procedure (19h) and purified by column chromatography (petroleum ether / ethyl acetate = 2:1) to give the product as a light yellow solid: 65% yield; mp = 123-125 °C  $[\alpha]_D^{20}$  -125.5 ( $c$  = 0.58,  $\text{CHCl}_3$ , 93% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T$  = 23 °C, UV = 254 nm,  $t_R$  = 13.26 min (major),  $t_R$  = 25.28 min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.59 (dd,  $J$  = 8.2 Hz, 1.6 Hz, 1H), 7.45 (d,  $J$  = 1.6 Hz, 1H), 7.41 (d,  $J$  = 6.0 Hz, 1H), 6.90 (d,  $J$  = 8.2 Hz, 1H), 6.09 (s, 2H), 5.72 (dd,  $J$  = 11.8 Hz, 4.4 Hz, 1H), 5.51 (d,  $J$  = 6.0 Hz, 1H), 2.98 (dd,  $J$  = 17.0 Hz, 11.8 Hz, 1H), 2.77 (dd,  $J$  = 17.1 Hz, 4.4 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  190.9, 190.2, 161.6, 152.8, 148.5, 128.4, 125.7, 108.6, 108.2, 107.9, 102.2, 78.7, 38.2; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3714, 3461, 2923, 2856, 1679, 1594, 1452, 1398, 1259, 1072, 802, 613; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{10}\text{O}_5$   $[\text{M}+\text{Na}]^+$  269.0426, found 269.0429.

**(R)-2-(2-naphthoyl)-2H-pyran-4(3H)-one (3o):**

The title compound was prepared according to the general working procedure (19h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 92% yield;  $[\alpha]_D^{20}$  -69.3 ( $c = 1.00$ ,  $\text{CHCl}_3$ , 93% *ee*); HPLC: Daicel Chiralpak OD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T = 23\text{ }^\circ\text{C}$ , UV = 240 nm,  $t_R = 14.65$  min (major),  $t_R = 24.44$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.50 (s, 1H), 8.02-7.89 (m, 4H), 7.67-7.57 (m, 2H), 7.44 (d,  $J = 6.0$  Hz, 1H), 5.97 (dd,  $J = 11.6$  Hz, 4.5 Hz, 1H), 5.54 (d,  $J = 6.0$  Hz, 1H), 3.04 (dd,  $J = 17.0$  Hz, 11.6 Hz, 1H), 2.87 (dd,  $J = 17.0$  Hz, 4.3 Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  193.0, 190.2, 161.7, 136.0, 132.4, 131.1, 131.0, 129.8, 129.3, 129.0, 127.9, 127.2, 124.1, 108.0, 79.0, 38.2; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3461, 3060, 2923, 2856, 2726, 2364, 1955, 1681, 1594, 1459, 1376, 1265, 1182, 1093, 1027, 809, 873, 584; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{16}\text{H}_{12}\text{O}_3$   $[\text{M}+\text{Na}]^+$  275.0684, found 275.0685.

**(R)-2-(cyclohexanecarbonyl)-2H-pyran-4(3H)-one (3p):**

The title compound was prepared according to the general working procedure (24h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow oil: 61% yield;  $[\alpha]_D^{20}$  -41.9 ( $c = 0.44$ ,  $\text{CHCl}_3$ , 90% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T = 23\text{ }^\circ\text{C}$ , UV = 254 nm,  $t_R = 5.75$  min (major),  $t_R = 8.96$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.40 (d,  $J = 6.0$  Hz, 1H), 5.47 (d,  $J = 6.0$  Hz, 1H), 4.95 (dd,  $J = 11.0$  Hz, 5.9 Hz, 1H), 2.85-2.78 (m, 1H), 2.75-2.71 (m, 2H), 1.88-1.67 (m, 5H), 1.41-1.20 (m, 5H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  208.1, 190.2, 161.7, 108.1, 80.9, 46.3, 37.6, 28.4, 28.0, 25.6, 25.57, 25.4; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3781, 3714, 2923, 2856, 2387, 1452, 1375, 1068, 800, 721, 609; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{12}\text{H}_{16}\text{O}_3$   $[\text{M}+\text{Na}]^+$  231.0997, found 231.1000.

**(R)-1-methylspiro[indoline-3,2'-pyran]-2,4'(3'H)-dione (5a):**

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a white solid: 93% yield; mp = 106-108 °C;  $[\alpha]_D^{20} +363.9$  (c = 1.08, CHCl<sub>3</sub>, 91% ee); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min, T = 23 °C, UV = 240 nm,  $t_R$  = 9.09 min (major),  $t_R$  = 11.95 min (minor); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.47 (ddd, *J* = 7.5 Hz, 1.2 Hz, 0.5 Hz, 1H), 7.41-7.37 (m, 2H), 7.07-7.03 (m, 1H), 6.87 (d, *J* = 7.9 Hz, 1H), 5.60 (dd, *J* = 6.2 Hz, 0.8 Hz, 1H), 3.21 (s, 3H), 3.18 (d, *J* = 16.6 Hz, 1H), 2.64 (dd, *J* = 16.6 Hz, 0.8 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 189.3, 171.9, 161.4, 143.0, 131.3, 127.5, 124.3, 123.6, 109.1, 107.0, 81.4, 41.3, 26.5; IR (film, ν/cm<sup>-1</sup>): 3781, 2923, 2856, 1727, 1679, 1604, 1461, 1375, 1093, 1029, 904, 800, 607; HRMS (ESI) *m/z* calcd for C<sub>13</sub>H<sub>11</sub>NO<sub>3</sub> [M+Na]<sup>+</sup> 252.0637, found 252.0639.

**(R)-1-benzylspiro[indoline-3,2'-pyran]-2,4'(3'H)-dione (5b):**

The title compound was prepared according to the general working procedure (16h) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a white solid: 96% yield; mp = 106-108 °C  $[\alpha]_D^{20} +186.3$  (-191.7, reported) (c = 0.30, CH<sub>2</sub>Cl<sub>2</sub>, 91% ee); HPLC: Daicel Chiralpak ID, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min, T = 23 °C, UV = 254 nm,  $t_R$  = 24.36 min (minor),  $t_R$  = 34.95 min (major), [HPLC: Daicel Chiralpak ID, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min, UV = 254 nm,  $t_{R1}$  = 21.15 min (major),  $t_{R2}$  = 29.10 min (minor), reported]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.50-7.48 (m, 1H), 7.42 (d, *J* = 6.2 Hz, 1H), 7.35-7.31 (m, 2H), 7.29-7.24 (m, 4H), 7.04-7.00 (m, 1H), 6.76 (d, *J* = 7.9 Hz, 1H), 5.64 (dd, *J* = 6.2 Hz, 0.7 Hz, 1H), 4.93 (d, *J* = 15.6 Hz, 1H), 4.85 (1H, *J* = 15.6 Hz, 1H), 3.25 (d, *J* = 16.6 Hz, 1H), 2.73 (dd, *J* = 16.6 Hz, 0.7 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 189.2, 172.2, 161.4,

142.1, 134.8, 131.2, 129.0, 128.0, 127.5, 127.2, 124.4, 123.6, 110.1, 107.0, 81.4, 44.0, 41.4; IR (film,  $\nu/\text{cm}^{-1}$ ): 3459, 2923, 2856, 2724, 1733, 1702, 1669, 1459, 1375, 1265, 1155, 802, 730, 613; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{15}\text{NO}_3$   $[\text{M}+\text{Na}]^+$  328.0950, found 328.0944.

**(R)-2-hydroxy-6-methoxy-1-p-tolylhex-5-ene-1,4-dione (8):**

The title compound was prepared according to the general working procedure and purified by column chromatography (dichloromethane / ethyl acetate = 8:1) to give the product as a colorless oil.  $[\alpha]_{\text{D}}^{20} +5.7$  ( $c = 0.33$ ,  $\text{CHCl}_3$ , 96% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 1.0 mL/min,  $T = 23\text{ }^\circ\text{C}$ , UV = 230 nm,  $t_{\text{R}} = 10.93$  min (minor),  $t_{\text{R}} = 12.01$  min (major);  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{COCD}_3$ ):  $\delta$  7.92 (d,  $J = 8.2$  Hz, 2H), 7.68 (d,  $J = 12.7$  Hz, 1H), 7.35 (d,  $J = 8.0$  Hz, 2H), 5.69 (d,  $J = 12.8$  Hz, 1H), 5.45 (dd,  $J = 7.2$  Hz, 4.4 Hz, 1H), 3.76 (s, 3H), 2.98 (dd,  $J = 16.4$  Hz, 4.4 Hz, 1H), 2.87 (dd,  $J = 16.3$  Hz, 7.2 Hz, 1H), 2.41 (s, 3H);  $^1\text{H}$  NMR (400 MHz,  $(\text{CD}_3)_2\text{SO}$ ):  $\delta$  7.87 (d,  $J = 8.2$  Hz, 2H), 7.71 (d,  $J = 12.7$  Hz, 1H), 7.33 (d,  $J = 8.0$  Hz, 2H), 5.65 (d,  $J = 12.8$  Hz, 1H), 5.52 (d,  $J = 7.2$  Hz, 1H), 5.29 (dd,  $J = 12.8$  Hz, 6.7 Hz, 1H), 3.70 (s, 3H), 2.93 (dd,  $J = 16.3$  Hz, 5.7 Hz, 1H), 2.81 (dd,  $J = 16.3$  Hz, 6.9 Hz, 1H), 2.38 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{COCD}_3$ ):  $\delta$  200.4, 196.6, 164.3, 145.0, 133.2, 130.2, 129.7, 107.1, 70.7, 58.3, 45.8, 21.6; IR (film,  $\nu/\text{cm}^{-1}$ ): 3687, 3459, 2923, 2856, 2726, 2281, 1745, 1679, 1594, 1459, 1176, 1089, 808, 728, 599, 522; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{14}\text{H}_{16}\text{O}_4$   $[\text{M}+\text{Na}]^+$  271.0946, found 271.0945.

**(R)-4-(tert-butyldimethylsilyloxy)-6-methoxy-3,6-dihydro-2H-pyran-2-yl(p-tolyl)methanone (9):**

The title compound was prepared according to the general working procedure and purified by column chromatography (dichloromethane / ethyl acetate = 10:1) to give the product as a colorless

oil.  $[\alpha]_{\text{D}}^{20}$  -7.5 ( $c = 0.78$ ,  $\text{CHCl}_3$ , 93% *ee*); HPLC: Daicel Chiralpak AD-H, hexane: 2-propanol = 70:30, flow rate = 0.7 mL/min,  $T = 23\text{ }^{\circ}\text{C}$ , UV = 230 nm,  $t_{\text{R}} = 5.61$  min (major),  $t_{\text{R}} = 6.01$  min (minor);  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{COCD}_3$ ):  $\delta$  7.95 (d,  $J = 8.2$  Hz, 2H), 7.38-7.36 (m, 2H), 5.30 (ddd,  $J = 11.2$  Hz, 3.7 Hz, 0.6 Hz, 1H), 5.07-5.06 (m, 1H), 4.95 (dd,  $J = 3.2$  Hz, 2.0 Hz, 1H), 3.45 (s, 3H), 2.63-2.55 (m, 1H), 2.42 (s, 3H), 2.10-2.04 (m, 1H), 0.95 (s, 9H), 0.23 (s, 3H), 0.22 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{COCD}_3$ ):  $\delta$  195.3, 152.2, 144.0, 133.1, 129.1, 102.5, 98.2, 68.1, 55.3, 30.7, 25.1, 20.8, 17.7, -5.1, -5.4; IR (film,  $\text{v}/\text{cm}^{-1}$ ): 3775, 3704, 3459, 3189, 2925, 2856, 2726, 1743, 1672, 1606, 1459, 1373, 1263, 1211, 1126, 1056, 950, 898, 829; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{30}\text{O}_{4\text{Si}}$   $[\text{M} + \text{Na}]^+$  385.1811, found 385.1814.

## ASSOCIATED CONTENT

### Supporting Information

$^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra for all the products; HPLC profiles and crystallographic data of compound **3d** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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### Notes

The authors declare no competing financial interest.

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## REFERENCES

- (1) For selected reviews or reports on the hetero-Diels-Alder reaction, see: (a) Danishefsky, S. J., *Aldrichimica Acta*. **1986**, *19*, 59; (b) Kametani, T.; Hibino, S., *Adv. Heterocycl. Chem.* **1987**, *42*, 245; (c) Danishefsky, S. J., Chemtracts: *Org. Chem.* **1989**, 273; (d) Ooi, T.; Maruoka, K. In *Comprehensive Asymmetric Catalysis I-III*; Jacobsen, E. N., Yamamoto, H., Eds.; Springer-Verlag: Berlin, 1999; Vol. 3, pp. 1237; (e) Jørgensen, K. A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3558; (f) Jørgensen, K. A., in *Cycloaddition Reactions in Organic Synthesis*; Kobayashi, S.; Jørgensen, K. A., Eds.; Wiley-VCH: Weinheim, Germany, **2002**, pp. 151; (g) Pellissier, H. *Tetrahedron* **2009**, *65*, 2839; (h) Gasperi, T.; Punzi, P.; Migliorini, A.; Tofani, D. *Curr. Org. Chem.* **2011**, *15*, 2098.
- (2) (a) Ciavatta, M. L.; Trivellone, E.; Villani, G.; Cimino, G. *Tetrahedron Lett.* **1993**, *34*, 6791; (b) Matteson, D. S.; Man, H. W. *J. Org. Chem.* **1993**, *58*, 6545; (c) Harris, J. M.; O'Doherty, G. A. *Tetrahedron Lett.* **2000**, *41*, 183; (d) Thompson, C. F.; Jamison, T. F.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2001**, *123*, 9974; (e) Dreeßen, S.; Schabbert, S.; Schaumann, E. *Eur. J. Org. Chem.* **2001**, 245; (f) Sampson, R. A.; Perkins, M. V. *Org. Lett.* **2002**, *4*, 1655; (g) Baker-Glenn, C.; Hodnett, N.; Reiter, M.; Ropp, S.; Ancliff, R.; Gouverneur, V. *J. Am. Chem. Soc.* **2005**, *127*, 1481; (h) Wetzell, S.; Bonl, R. S.; Kumar, K.; Waldmann, H. *Angew. Chem., Int. Ed.* **2011**, *50*, 10800.
- (3) For examples of chiral Lewis acid catalysts, see: (a) Danishefsky, S. J.; Deninno, M. P. *Angew. Chem., Int. Ed.* **1987**, *26*, 15; (b) Hanamoto, T.; Furuno, H.; Sugimoto, Y.; Inanaga, J. *Synlett* **1997**, 79; (c) Dossetter, A. G.; Jamison, T. F.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **1999**, *38*, 2398; (d) Quitschalle, M.; Christmann, M.; Bhatt, U.; Kalesse, M. *Tetrahedron Lett.* **2001**, *42*, 1263; (e) Gademann, K.; Chavez, D. E.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2002**, *41*, 3059; (f) Doyle,

M. P.; Valenzuela, M.; Huang, P. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, *101*, 5391; (g) Anada, M.; Washio, T.; Shimada, N.; Kitagaki, S.; Nakajima, M.; Shiro, M.; Hashimoto, S.; *Angew. Chem., Int. Ed.* **2004**, *43*, 2665; (h) Hu, Y.; Xu, K.; Zhang, S.; Guo, F.; Zha Z.; Wang, Z. *Org. Lett.* **2014**, *16*, 3564.

(4) For examples of chiral Brønsted acid catalysts, see: Schuster, T.; Bauch, M.; Durner, G.; Gobel, M. W. *Org. Lett.* **2000**, *2*, 179.

(5) (a) Johannsen, M.; Yao, S.; Jørgensen, K. A. *Chem. Commun.* **1997**, 2169; (b) Yao, S.; Johannsen, M.; Audrain, H.; Hazell, R. G.; Jørgensen, K. A. *J. Am. Chem. Soc.* **1998**, *120*, 8599.

(6) Zhao, B.; Loh, T.-P. *Org. Lett.* **2013**, *15*, 2914.

(7) (a) Du, H.; Long, J.; Hu, J.; Li, X.; Ding, K. *Org. Lett.* **2002**, *4*, 4349; (b) Yuan, Y.; Li, X.; Sun, J.; Ding, K. *J. Am. Chem. Soc.* **2002**, *124*, 14866; (c) Ji, B. M.; Yuan, Y.; Ding, K.; Meng, J. B. *Chem. Eur. J.* **2003**, *9*, 5989; (d) Du, H. F.; Ding, K. *Org. Lett.* **2003**, *5*, 1091; (e) Zhang, X.; Du, H.; Wang, Z.; Wu, Y. D.; Ding, K. *J. Org. Chem.* **2006**, *71*, 2862.

(8) (a) Fan, Q.; Lin, L.; Liu, J.; Huang, Y.; Feng, X.; Zhang, G. *Org. Lett.* **2004**, *6*, 2185; (b) Lin, L.; Fan, Q.; Qin, B.; Feng, X. *J. Org. Chem.* **2006**, *71*, 4141; (c) Yu, Z.; Liu, X.; Dong, Z.; Xie, M.; Feng, X. *Angew. Chem., Int. Ed.* **2008**, *47*, 1308; (d) Lin, L. L.; Kuang, Y. L.; Liu, X. H.; Feng, X. *Org. Lett.* **2011**, *13*, 3868; (e) Zheng, J.; Lin, L.; Fu, K.; Zhang, Y.; Liu, X.; Feng, X. *Chem. -Eur. J.* **2014**, *20*, 14493.

(9) Tono, T.; Mikami, K. *Tetrahedron Lett.* **2005**, *46*, 6355.



(10) (a) Guo, F.; Lai, G.; Xiong, S.; Wang, S.; Wang, Z. *Chem. Eur. J.* **2010**, *16*, 6438; (b) Lai, G.; Guo, F.; Zheng, Y.; Fang, Y.; Song, H.; Xu, K.; Wang, S.; Zha, Z.; Wang, Z. *Chem. Eur. J.* **2011**, *17*, 1114; (c) Zhang, S.; Xu, K.; Guo, F.; Hu, Y.; Zha, Z.; Wang, Z. *Chem. Eur. J.* **2014**, *20*, 979; (d) Li, C.; Guo, F.; Xu, K.; Zhang, S.; Hu, Y.; Zha, Z.; Wang, Z. *Org. Lett.* **2014**, *16*, 3192.

(11) Details of the crystal structure analysis are provided as Supporting Information. CCDC-1437133 contains the supplementary crystallo-graphic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif)

(12) For the [4+2] cycloaddition pathway, see: (a) Bednarski, M.; Maring, C.; Danishefsky, S. *Tetrahedron Lett.* **1983**, *24*, 3451; (b) Bednarski, M.; Danishefsky, S. *J. Am. Chem. Soc.* **1983**, *105*, 3717; (c) Motoyama, Y.; Koga, Y.; Nishiyama, H. *Tetrahedron*, **2001**, *57*, 853; (d) Yang, X.-B.; Feng, J.; Zhang, J.; Wang, N.; Wang, L.; Liu, J.-L.; Yu, X.-Q. *Org. lett.* **2008**, *10*, 1299.

(13) For the Mukaiyama-aldol pathway, see: (a) Corey, E.; Cywin, C.; Roper, T. *Tetrahedron Lett.* **1992**, *33*, 6907; (b) Yamashita, Y.; Saito, S.; Ishitani, H.; Kobayashi, S. *J. Am. Chem. Soc.* **2003**, *125*, 3793.

(14) (a) Wang, P.; Tao, W.; Sun, X.; Liao, S.; Tang, Y. *J. Am. Chem. Soc.* **2013**, *135*, 16849; (b) Wang, P.; Feng, L.; Wang, L.; Li, J.; Liao, S.; Tang, Y. *J. Am. Chem. Soc.* **2015**, *137*, 4626.

(15) Liu, Z.; Gu, P.; Shi, M.; McDowell, P.; Li, G. *Org. Lett.* **2011**, *13*, 2314.

(16) Danishefsky, S.; Kitahara, T. *J. Am. Chem. Soc.* **1974**, *96*, 7807; Nakagawa M. *et al*, *Heterocycles*. **2003**, *59*, 721.