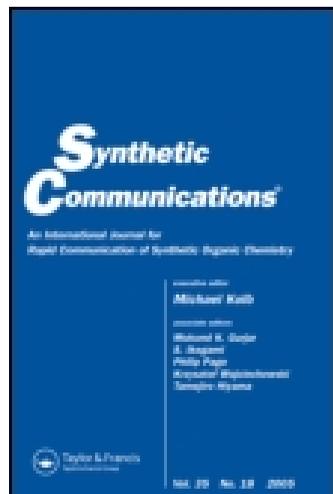


This article was downloaded by: [University of Tasmania]

On: 02 September 2014, At: 09:42

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

Unified Approach to the Asymmetric Synthesis of Higher Homologues of (3-Quinolyl)-alanine

Shital K. Chattopadhyay^a & Indranil Kundu^a

^a Department of Chemistry, University of Kalyani, Kalyani, West Bengal, India

Accepted author version posted online: 16 Apr 2014. Published online: 09 Jun 2014.

To cite this article: Shital K. Chattopadhyay & Indranil Kundu (2014) Unified Approach to the Asymmetric Synthesis of Higher Homologues of (3-Quinolyl)-alanine, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 44:14, 2112-2120, DOI: [10.1080/00397911.2014.882002](https://doi.org/10.1080/00397911.2014.882002)

To link to this article: <http://dx.doi.org/10.1080/00397911.2014.882002>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms &

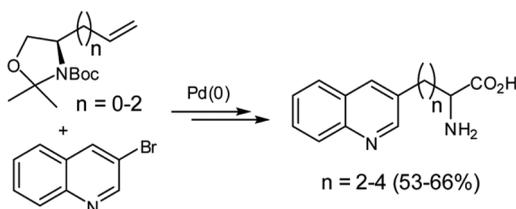
Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

UNIFIED APPROACH TO THE ASYMMETRIC SYNTHESIS OF HIGHER HOMOLOGUES OF (3-QUINOLYL)-ALANINE

Shital K. Chattopadhyay and Indranil Kundu

Department of Chemistry, University of Kalyani, Kalyani,
West Bengal, India

GRAPHICAL ABSTRACT



Abstract A new protocol based on a palladium-catalyzed Heck reaction of an amino acid-derived vinyl unit with 3-bromoquinoline has been developed to access the title compounds in good yield and optical purity.

Keywords α -Amino acids; Heck reaction; quinoline

INTRODUCTION

Synthesis of noncoded amino acids has remained an important activity over the decades because of amino acids' known importance in chemistry and biology.^[1] A class of such nonproteinogenic amino acids are the heterocycle appended α -amino acids, which are important because of their wide range of chemical and biological applications as components of natural products and peptide nucleic acids, building blocks in therapeutic lead generation, and conformational control elements in the design and synthesis of modified peptides.^[2] Thus, synthesis of α -amino acids accommodating a side-chain heterocyclic units such as pyrrole,^[3] furan,^[4] pyridine,^[5] pyrimidine,^[6] purine,^[7] pyrazole,^[8] thiophene,^[9] indole,^[10] quinoline,^[11] and quinoxaline^[12] have been described. Although superb catalytic methods are being continuously developed,^[13] conversion of simple coded amino acids into desired targets have remained important. Thus, several amino-acid-based building blocks have emerged.^[14] One such type of building block is the α , β -, or γ -vinylic oxazolidines

Received September 4, 2013.

Address correspondence to Shital K. Chattopadhyay, Department of Chemistry, University of Kalyani, Kalyani 741235, West Bengal, India. E-mail: skchatto@yahoo.com

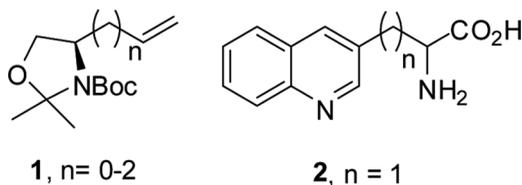


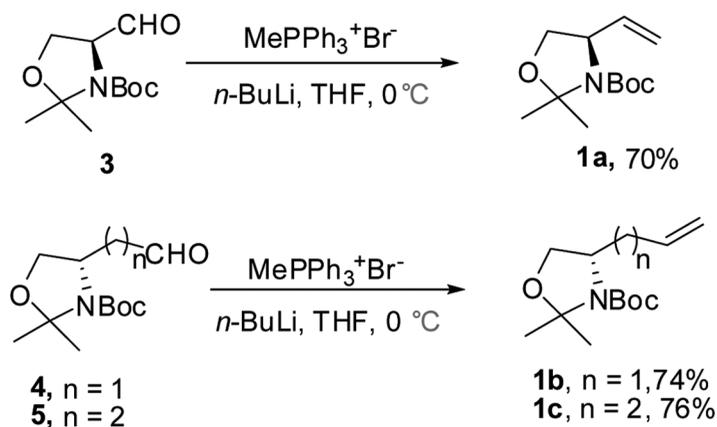
Figure 1. Background of the work.

of the general structure **1** (Fig. 1). Heck-type arylation of **1** and derivatives thereof continue to be useful in the synthesis of aryl amino acids.^[15] However, its utility for the preparation of heteroaromatic amino acids has remained comparatively less explored.

(3-Quinolyl)-alanine (**2**) has been extensively utilized as a replacement of histidine residue in important peptides for the preparation of somatostatin analogs,^[16] gonadotropin-releasing-hormone-receptor antagonists,^[17] and gramicidin channel receptor agonists.^[18] However, higher homologues of (3-quinolyl)-alanine are less known. Herein, we report synthesis of three homologues of **2** having variation in the chain length ($n = 2-4$), involving Heck-type arylation of 3-bromoquinoline with the olefins **1** as the key step.

RESULTS AND DISCUSSION

Our synthesis started with the preparation of the olefins **1a-c** (Scheme 1). Wittig-type methylenation of Garner's aldehyde **3**^[19] has been found to be problematic and prone to racemization as well as poor yielding when KH or *n*-BuLi was used as base.^[20] However, use of potassium hexamethyldisilazide (KHMDS) as base circumvented both the problems.^[21] Other conditions such as use of Me₃Al/Zn/CH₂I₂ combination for racemization-free olefination of **1a** are also known.^[20a,21] However, we have found that a good yield conversion of **3** to **1a** could simply be achieved if the reaction is carried out at 0 °C instead of -78 °C in the presence of

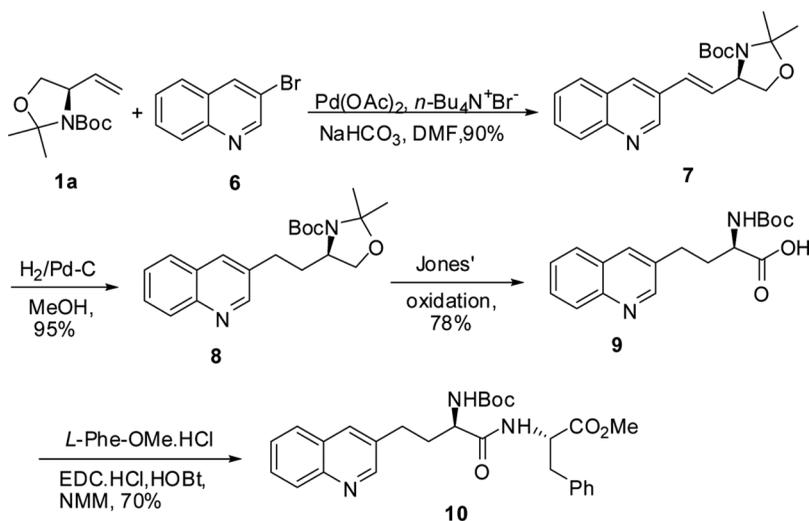


Scheme 1. Preparation of starting materials.

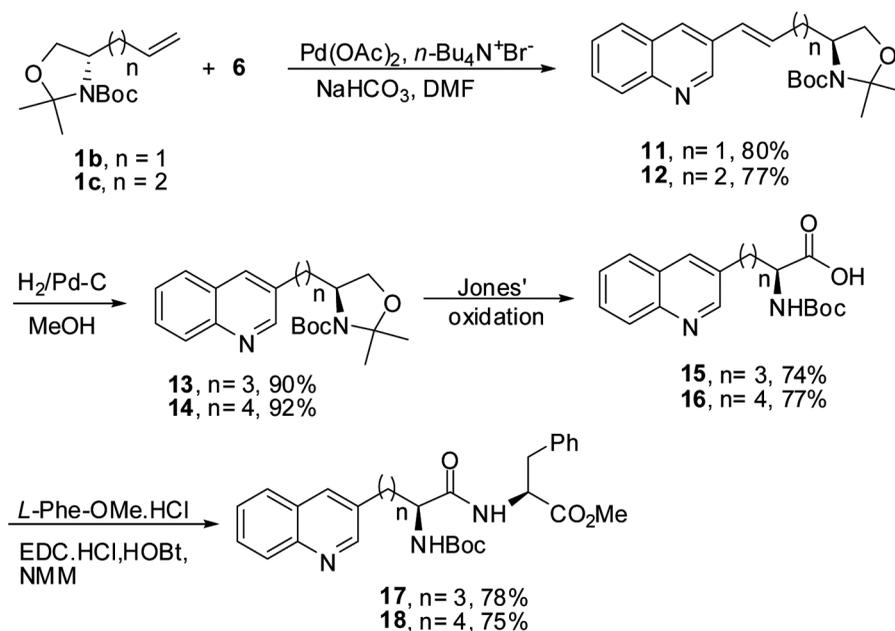
n-BuLi as base. Moreover, the optical purity of the product also proved to be unaffected because the product **1a** displayed a specific rotation value, $[\alpha]_D + 17.2$ (*c* 1.20, CHCl₃), in close agreement to those reported $[[\alpha]_D + 15.6$ (*c* 2.0, CHCl₃); $[[\alpha]_D + 15$ (*c* 2.5, CHCl₃)] for optically pure compound.^[19,20] The aspartic-acid-derived aldehyde **4**^[22,4e] and the glutamic acid-derived aldehyde **5**^[23] were similarly methylenated under the developed conditions to yield the corresponding olefins **1b** and **1c** respectively, in good yields.

Having access to the required olefins **1a–c**, we then focused on their projected Heck-type coupling with 3-bromoquinoline. Palladium-catalyzed Mizoroki–Heck olefination^[24] is usually more regio- and stereoselective with electron-deficient olefins and often been found to be problematic with nonactivated olefins, with poor regioselectivity and stereoselectivity being major concerns. After some experimentation, we have found that olefination of 3-bromoquinoline (**6**) with olefin **1a** proceeds better under Jeffery's two-phase protocol^[25] involving use of *n*-Bu₄N⁺Br[−] as an additive, and the coupled product **7** (Scheme 2) is obtained as the only isolable product in good yield.

The *E*-geometry of the newly formed double bond was easily secured from its ¹H NMR spectrum, a 16-Hz coupling between the olefinic protons being diagnostic. Moreover, characteristic rotamerism^[26] of the oxazolidine moiety was reflected in doubling of some signals in its ¹³C NMR spectrum. Rerecording the ¹³C NMR spectrum at higher temperature removed this commonly observed complexity and a single set of signals was observed. We also observed signal broadening in the ¹H NMR spectrum recorded in CDCl₃. A better resolution of the peaks was observed when the spectrum was recorded in DMSO-*d*₆[see Supporting Information]. Catalytic hydrogenation of the compound **7** led to the corresponding saturated compound **8** and a one-pot deprotection of the oxazolidine unit in the latter followed by in situ oxidation of the resulting primary alcohol using Jones's oxidation reaction proceeded smoothly to provide the α -amino acid derivative **9** in an overall yield of 66% over three steps. Compound **9** was then subjected to peptide bond formation with *L*-Phe-OMe.HCl



Scheme 2. Preparation of (3-quinoly)-homoalanine.



Scheme 3. Synthesis of higher homologues of (3-quinolyl)-alanine.

and the resulting dipeptide **10** was analyzed by NMR and high-performance liquid chromatography (HPLC) for its stereochemical purity. Pleasingly, compound **10** was found to be homogeneous, thereby indicating little racemization during the sequence of reactions depicted in Scheme 2.

We then focused on the projected arylation of the olefins **1b** and **1c**. Separate treatment of each of these olefins with 3-bromoquinoline under the developed conditions furnished the corresponding coupled products **11** and **12** (Scheme 3) in good yields in a regio- and stereoselective manner. Saturation of the olefinic unit in each of these compounds proceeded uneventfully to provide the compounds **13** and **14** respectively. Repetition of the one-pot deprotection–oxidation sequence on these compounds as detailed for the conversion **8** → **9** then led to the desired α -amino acid derivatives **15** and **16** in good overall yields of 53–55% over three steps. Each of these α -amino acids were then converted to the corresponding dipeptides **17** and **18** under conventional conditions. The latter were also found to be optically homogeneous.

In short, we have developed a three-step synthetic protocol for the access of three homologues of the important α -amino acid (3-quinolyl)-alanine (*viz.* compounds **9**, **15**, and **16**) in good overall yields and optical purity. Easy access to the starting olefins **1a–c** has also been developed. The synthetic protocols may allow access to related compounds of interest and hence may find application.

EXPERIMENTAL

Melting points were recorded in open capillaries and are uncorrected. Infrared (IR) spectra were recorded on a Perkin-Elmer Spectrum-1 spectrophotometer. Proton and carbon NMR spectra were recorded on a Bruker Avance-400 spectrometer

(purchased through DST-FIST Grant). Chemical shifts are recorded relative to residual solvent peak. Data for rotamers are given within parentheses. Mass spectra were recorded on a Jeol-JMS 600 instrument from I.I.C.B., Kolkata or IACS, Kolkata. Petroleum ether refers to the fraction boiling in the range 60–80 °C. Silica gel (60–120, 200–230 mesh) for column chromatography was purchased from Spectrochem, India.

General Procedure for the Synthesis of the Olefins 1a–c

(R)-tert-Butyl 2,2-dimethyl-4-vinyl-oxazolidine-3-carboxylate (1a). *n*-BuLi (2 M in hexane, 1.7 mL, 3.45 mmol, 1.5 eqv.) was added to a stirred suspension of methyltriphenylphosphonium bromide (1.23 g, 3.45 mmol, 1.5 eqv.) in dry THF (10 mL) under argon at 0 °C dropwise over 5 min, and the resulting solution was allowed to stir for 20 min at 0 °C. A solution of the aldehyde **3** (528 mg, 2.30 mmol) in THF (5 mL) was then added dropwise over 5 min with stirring at the same temperature. After 20 min, the solution was allowed to come to room temperature and was stirred for another 4 h. The reaction mixture was quenched by addition of aqueous NH₄Cl solution (3 mL) and then extracted with ethyl acetate (2 × 50 mL). The combined organic extract was washed successively with H₂O (40 mL) and brine solution (40 mL) and then dried over MgSO₄. It was then filtered, and the filtrate was concentrated under reduced pressure to leave the crude product, which was purified by column chromatography over silica gel (EtOAc–PE, 1:19) to give the olefin **1a** as a colorless liquid; yield: 366 mg (70%); [α]_D + 17.2 (*c* 1.20, CHCl₃); Lit.^[21a] [[α]_D + 15.6 (*c* 2.0, CHCl₃); lit.^[20a] [[α]_D + 15 (*c* 2.5, CHCl₃)].

IR (CHCl₃): 2981, 1699, 1479, 1456, 1385, 1366, 1254 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ 5.76 (s, 1H), 5.20–5.09 (m, 2H), 4.35–4.22 (br m, 1H), 3.99 (td, *J* = 6.4, 2.8 Hz, 1H), 3.70 (dd, *J* = 9.2, 2.4 Hz, 1H), 1.56 (s, 3H), 1.46–1.38 (m, 12H). ¹H NMR (DMSO-*d*₆, 400 MHz): δ 5.78 (ddd, *J* = 6.8, 10.0, 16.8 Hz, 1H), 5.20–5.09 (m, 2H), 4.35–4.22 (br m, 1H), 4.00 (dd, *J* = 6.4, 8.8 Hz, 1H), 3.70 (dd, *J* = 9.2, 2.0 Hz, 1H), 1.56 (s, 3H), 1.46–1.38 (two singlets, 12H). ¹³C NMR (CDCl₃, 100 MHz): δ 151.9, 137.3, 115.7, 93.9, 79.5, 68.1, 59.6, 28.3, 26.5, 23.6. MS (TOF, ES⁺): *m/z* (%) = 250 (100) [M + Na].

General Procedure for Heck-Type Olefination with 1a–c: (R,E)-tert-Butyl 2,2-Dimethyl-4-(2-(quinolin-3-yl)vinyl)oxazolidine-3-carboxylate (7)

A solution of the olefin **1a** (170 mg, 0.75 mmol), 3-bromoquinoline (171 mg, 0.82 mmol, 1.1 eqv.), palladium acetate (8 mg, 0.04 mmol, 5 mol %), NaHCO₃ (94 mg, 1.12 mmol, 1.5 eqv.), and *n*-Bu₄N⁺Br⁻ (723 mg, 2.24 mmol, 3 eqv.) in DMF (5 mL) was taken in a sealed tube and stirred at 90 °C for 18 h. The reaction mixture was allowed to come to rt and then diluted with H₂O (60 mL). It was then extracted with EtOAc (2 × 50 mL). The combined organic extract was washed successively with H₂O (2 × 50 mL) and brine (50 mL), and then dried (MgSO₄). It was filtered and then concentrated in vacuo to leave a residue, which was purified over silica gel (EtOAc–PE, 3:7) to provide the coupled product **7** as a colorless solid (238 mg, 90%). Mp 120–121 °C. [α]_D – 98.7 (*c* 0.42, CHCl₃). IR (CHCl₃): 2991, 2869, 1680,

1494, 1390, 1361, 1255 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 8.98 (s, 1H), 8.08–8.05 (m, 2H), 7.78 (s, 1H), 7.66 (s, 1H), 7.52 (s, 1H), 6.75–6.61 (br m, 1H), 6.41 (s, 1H), 4.64–4.92 (br m, 1H), 4.16 (dd, $J=8.8, 6.4$ Hz, 1H), 3.89 (dd, $J=8.8, 1.2$ Hz, 1H), 1.71 (s, 3H), 1.57 (1.49) (s, 3H), 1.42 (s, 9H). ^1H NMR (400 MHz, DMSO-d_6): δ 9.06 (s, 1H), 8.38 (s, 1H), 8.00–7.95 (m, 2H), 7.72 (t, $J=7.6$ Hz, 1H), 7.60 (dd, $J=8.4, 6.8$ Hz, 1H), 6.66 (d, $J=16$ Hz, 1H), 6.55 (dd, $J=7.2, 15.6$ Hz, 1H), 4.56–4.48 (br m, 1H), 4.13 (dd, $J=8.8, 6.4$ Hz, 1H), 3.84 (dd, $J=8.8, 1.2$ Hz, 1H), 1.48–1.35 (overlapping singlets, 15H). ^{13}C NMR (100 MHz, CDCl_3): δ 151.9, 149.2, 147.5, 132.7, 132.4, 131.2, 129.5, 129.2, 128.3, 128.0, 127.8, 127.0, 94.3 (93.8), 80.5 (79.9), 68.2, 59.5, 28.5, 27.6 (26.7), 24.7 (23.7). ^{13}C NMR (100 MHz, $\text{DMSO-d}_6, 70^\circ\text{C}$): δ 152.0, 149.8, 147.6, 132.7, 132.5, 130.2, 129.8, 129.3, 128.7, 128.3, 128.1, 127.5, 94.0, 79.7, 68.3, 59.6, 28.7, 26.4, 23.2. Anal. Calcd. for $\text{C}_{21}\text{H}_{26}\text{N}_2\text{O}_3$: C, 71.16; H, 7.39; N, 7.90; found: C, 71.31, H, 7.46; N, 7.85. MS (TOF, ES+): m/z (%) = 355 (100) [M + H].

General Procedure for Hydrogenation of the Coupled Olefins: (*R*)-*tert*-Butyl 2,2-Dimethyl-4-(2-(quinolin-3-yl)ethyl)oxazolidine-3-carboxylate (8)

Pd-C (10 mol%; 14 mg) was added to a solution of the coupled product **7** (200 mg, 0.56 mmol) in MeOH (6 mL) at rt and the heterogeneous mixture was vigorously stirred under hydrogen atmosphere for 2 h. It was then filtered through celite, and the filter cake was thoroughly washed with MeOH (10 mL). The combined filtrate was concentrated in vacuo to leave a crude mass, which was purified by chromatography over silica gel (EtOAc-PE, 3:7) to furnish the product **8** as a colorless solid (189 mg, 95%). Mp 80–81 $^\circ\text{C}$. $[\alpha]_{\text{D}} - 41.8$ (c 0.61, CHCl_3). IR (CHCl_3): 2979, 2870, 1676, 1496, 1392, 1251 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ 8.78 (s, 1H), 8.08 (d, $J=8.4$ Hz, 1H), 7.98–7.92 (br m, 1H), 7.77 (d, $J=8.0$ Hz, 1H), 7.66 (s, 1H), 7.53 (s, 1H), 4.06–3.86 (m, 3H), 2.90–2.70 (m, 2H), 2.30–1.97 (m, 2H), 1.66 (1.63) (s, 3H), 1.50 (s, 9H), 1.41 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ 151.9, 151.7, 146.9, 134.1, 129.2, 128.6, 128.1, 127.3, 126.6, 93.4, 80.2 (79.7), 66.8, 57.4 (56.8), 35.1 (34.4), 29.9, 28.5, 27.6 (26.8), 24.5 (23.2). Anal. calcd. for $\text{C}_{21}\text{H}_{28}\text{N}_2\text{O}_3$: C, 70.76; H, 7.92; N, 7.86. Found: C, 70.91, H, 7.84; N, 7.95. MS (TOF, ES+): m/z (%) = 357(100) [M + H].

General Procedure for Jones's Oxidation: (*R*)-2-(*tert*-Butoxycarbonylamino)-4-(quinolin-3-yl)butanoic Acid (9)

Jones's reagent (2.67 M, 366 μL , 0.66 mmol) was added to a solution of the appropriate oxazolidine derivative **8** (90 mg, 0.25 mmol) in acetone (3 mL) at 0 $^\circ\text{C}$. The reaction mixture was stirred for 5 h and then quenched by addition of 2-propanol (2.5 mL). The resulting reaction mixture was stirred for 10 min and then neutralized with saturated aqueous NaHCO_3 solution (pH = 4–5). It was then extracted with ethyl acetate (2 \times 25 mL), and the combined organic extract was washed successively with H_2O (20 mL) and brine solution (15 mL) and then dried over MgSO_4 . It was then filtered, and the filtrate was concentrated under reduced pressure to leave the crude product, which was purified by column chromatography over silica gel (EtOAc-PE, 7:3) to obtain compound **9** as a colorless solid (64 mg, 78%); Mp 176–178 $^\circ\text{C}$; $[\alpha]_{\text{D}} - 51.5$

(*c* 0.13, CHCl₃). IR (CHCl₃): 3402, 2976, 2479, 1716, 1584, 1521, 1364, 1282 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 13.28 (s, 1H), 8.84 (s, 1H), 8.14 (s, 1H), 8.01 (d, *J* = 8.0 Hz, 1H), 7.76 (d, *J* = 8.0 Hz, 1H), 7.59 (t, *J* = 6.8 Hz, 1H), 7.50 (t, *J* = 7.6 Hz, 1H), 5.67 (d, *J* = 7.2 Hz, 1H), 4.48 (d, *J* = 5.2 Hz, 1H), 2.98 (d, *J* = 6.8 Hz, 2H), 2.43 (d, *J* = 6.4 Hz, 1H), 2.24 (d, *J* = 6.4 Hz, 1H), 1.47 (s, 9H). ¹³C NMR (100 MHz, CDCl₃): δ 175.3, 155.6, 149.8, 143.8, 137.2, 134.6, 129.8, 128.2, 127.5, 127.4, 126.2, 79.7, 53.3, 33.9, 28.7, 28.4. HRMS (TOF, ES⁺): *m/z* [M⁺ + H] calcd. for C₁₈H₂₃N₂O₄: 331.1658; found: 331.1651.

General Procedure for Peptide Coupling: (S)-Methyl 2-((R)-2-(tert-Butoxycarbonylamino)-4-(quinolin-3-yl)butanamido)-3-phenylpropanoate (10)

N-Methylmorpholine (75 μL, 0.75 mmol) was added dropwise to a stirred suspension of *L*-Phe-OMe.HCl (50 mg, 0.3 mmol, 1 eqv.) in dry CH₂Cl₂ (4 mL) at 0 °C under argon. After 10 min, a solution of the carboxylic acid **9** (100 mg, 0.3 mmol, 1 eqv) in dry CH₂Cl₂ (4 mL) was added dropwise and stirred for 10 min. Then *N*-(3-dimethylamino-propyl)-*N'*-ethylcarbodiimide hydrochloride (EDC · HCl) (64 mg, 0.34 mmol, 1.1 eqv.) and 1-hydroxybenzotriazole (HOBT) (46 mg, 0.34 mmol, 1.1 eqv.) were added successively in 10-min intervals at the same temperature. The reaction mixture was allowed to come to room temperature and stirred for 18 h. It was then diluted with water (25 mL) and extracted with EtOAc (2 × 25 mL). The combined organic layer was washed successively with H₂O (2 × 20 mL) and brine (15 mL). It was then dried over Na₂SO₄ and filtered, and the filtrate was concentrated in vacuo to leave a pale yellow liquid, which was purified by chromatography over silica gel (EtOAc–PE, 1:1) as eluent to give the product **10** as a colorless solid (104 mg, 70%). Mp 124–126 °C. [α]_D + 36.7 (*c* 0.12, CHCl₃). IR (CHCl₃): 3337, 2991, 2858, 1736, 1681, 1661, 1537, 1522, 1435, 1281 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ 8.71 (s, 1H), 8.08 (d, *J* = 8.4 Hz, 1H), 7.87 (s, 1H), 7.75 (d, *J* = 8.0 Hz, 1H), 7.67 (dt, *J* = 7.2, 1.2 Hz, 1H), 7.53 (t, *J* = 7.6 Hz, 1H), 7.23 (d, *J* = 7.6 Hz, 2H), 7.19 (d, *J* = 6.8 Hz, 1H), 7.11 (d, *J* = 7.6 Hz, 2H), 6.67 (d, *J* = 7.2 Hz, 1H), 5.13 (d, *J* = 7.6 Hz, 1H), 4.89 (dd, *J* = 13.2, 6.8 Hz, 1H), 4.20 (d, *J* = 4.4 Hz, 1H), 3.73 (s, 3H), 3.18 (dd, *J* = 14.0, 5.6 Hz, 1H), 3.05 (dd, *J* = 14.0, 6.8 Hz, 1H), 2.74 (t, *J* = 7.6 Hz, 1H), 2.20–2.15 (m, 1H), 1.94–1.87 (m, 2H), 1.43 (s, 9H). ¹³C NMR (100 MHz, CDCl₃): δ 171.9, 171.5, 155.7, 151.6, 146.7, 135.8, 134.5, 133.7, 129.2, 129.0, 128.9, 128.6, 128.0, 127.4, 127.2, 126.7, 80.2, 53.8, 53.1, 52.4, 37.8, 33.8, 28.9, 28.3. Anal. calcd for C₂₈H₃₃N₃O₅: C, 68.41; H, 6.77; N, 8.55. Found: C, 68.64, H, 6.67; N, 8.61. MS (TOF, ES⁺): *m/z* (%) = 492 (100) [M + H].

ACKNOWLEDGMENT

This article is respectfully dedicated to the late Professor Alexander McKillop in admiration of his immense contribution to organic chemistry.

FUNDING

Financial assistance from the Department of Science and Technology (DST), New Delhi (Grant No. SR/S1/OC-92/2012) is gratefully acknowledged. One of

the authors (I. K.) is also thankful to the University Grants Commission, New Delhi, for a fellowship. Financial assistance from DST–PURSE and the University of Kalyani is also thankfully acknowledged.

SUPPLEMENTARY CONTENT

Full experimental detail and ^1H and ^{13}C NMR spectra are included in the Supplementary Content. This material can be accessed on the publisher's website.

REFERENCES

1. (a) Williams, R. M. In *Synthesis of Optically Active α -Amino Acids*; J. E. Baldwin, P. D. Magnus (Eds.); Organic Chemistry Series; Pergamon Press: Oxford, 1989; Vol. 7; (b) Barret, G. C. In *Amino Acids, Peptides, and Proteins*; Chemical Society: London, 2001; Vol. 32.
2. Matthews, J. L. Synthesis of peptides and peptidomimetics. In *Houben-Weyl Methods in Organic Chemistry*; M. Goodman, A. Felix, L. Moroder, and C. Toniolo (Eds.); Thieme: Stuttgart, 2001; Vol. E22C, p. 552.
3. (a) Dörr, A. A.; Lubell, W. D. *Tetrahedron Lett.* **2011**, 52, 2159; (b) Sarkar, K.; Singha, S. K.; Chattopadhyay, S. K. *Tetrahedron: Asymmetry* **2009**, 20, 1719; (c) Adamczyk, M.; Johnson, D. D.; Reddy, R. E. *Tetrahedron: Asymmetry* **2000**, 11, 3063.
4. (a) Molina, L.; Moreno-Vargas, A. J.; Carmona, A. T.; Robina, I. *Synlett* **2006**, 1327; (b) Saaby, S.; Bayón, P.; Aburel, P. S.; Jørgensen, K. A. *J. Org. Chem.* **2002**, 67, 4352; (c) Reingruber, R.; Baumann, T.; Dahmen, S.; Bräse, S. *Adv. Synth. Catal.* **2009**, 351, 1019; (d) Sakai, N.; Asano, J.; Shimano, Y.; Konakahara, T. *Synlett* **2007**, 2675; (e) Chattopadhyay, S. K.; Sarkar, K.; Karmakar, S. *Synlett* **2005**, 2083.
5. (a) Dondoni, A.; Massi, A.; Minghini, E.; Sabbatini, S.; Bertolasi, V. *J. Org. Chem.* **2003**, 68, 6172; (b) Adamczyk, M.; Akireddy, S. R.; Reddy, R. E. *Org. Lett.* **2001**, 3, 3157; (c) Walker, M. A.; Kaplita, K. P.; Chen, T.; Dalton King, H. *Synlett* **1997**, 169; (d) Ye, B.; Burke, Jr., T. R. *J. Org. Chem.* **1995**, 60, 2640.
6. (a) Elmarrouni, A.; Güell, M.; Collell, C.; Heras, M. *Tetrahedron* **2010**, 66, 612; (b) Font, D.; Heras, M.; Villalgordo, J. M. *Tetrahedron* **2008**, 64, 5226; (c) Jones, R. C. F.; Berthelot, D. J. C.; Iley, J. N. *Chem. Commun.* **2000**, 2131.
7. Čapek, P.; Pohl, R.; Hocek, M. *J. Org. Chem.* **2005**, 70, 8001.
8. De Luca, L.; Giacomelli, G.; Porcheddu, A.; Spanedda, A. M.; Falorni, M. *Synthesis* **2000**, 1295.
9. Podesa, P. V.; Tosa, M. I.; Paizs, C.; Irimie, F. *Tetrahedron: Asymmetry* **2008**, 19, 500.
10. Sakai, N.; Asano, J.; Shimano, Y.; Konakahara, T. *Synlett* **2007**, 17, 2675.
11. (a) Jones, S. W.; Palmer, C. F.; Paul, J. M.; Tiffin, P. D. *Tetrahedron Lett.* **1999**, 40, 1211; (b) Cabarrocas, G.; Rafel, S.; Ventura, M.; Villalgordo, J. M. *Synlett* **2000**, 595; (c) Cabarrocas, G.; Ventura, M.; Maestro, M.; Mahía, J.; Villalgordo, J. M. *Tetrahedron: Asymmetry* **2001**, 12, 1851.
12. Staszewska, A.; Stefanowicz, P.; Szewczuk, Z. *Tetrahedron Lett.* **2005**, 46, 5525.
13. For a recent review, see Nájera, C.; Sansano, J. M. *Chem. Rev.* **2007**, 107, 4584.
14. (a) Arnold, L. D.; May, R. G.; Vederas, J. C. *J. Am. Chem. Soc.* **1988**, 110, 2237; (b) Sasaki, N. A.; Hashimoto, C.; Potier, P. *Tetrahedron Lett.* **1987**, 28, 6069; (c) Evans, D. A.; Bach, T. *Angew. Chem. Int. Ed.* **1993**, 33, 1326; (d) Jackson, R. F. W.; Wishart, N.; Wood, A.; James, K.; Wythes, M. J. *J. Org. Chem.* **1992**, 57, 3397; (e) Harwood, L. M.; Manage, A. C.; Robin, S.; Hopes, S. F. G.; Watkin, D. J.; Williams, C. E. *Synlett* **1993**,

- 777; (f) Dondoni, A.; Marra, A.; Massi, A. *J. Org. Chem.* **1999**, *64*, 933; (g) Collier, P. N.; Campbell, A. D.; Patel, I.; Raynham, T. M.; Taylor, R. J. K. *J. Org. Chem.* **2002**, *67*, 1802.
15. (a) Liang, H.; Zhou, Y.; Ciufolini, M. A. *Synthesis* **2010**, 2515; (b) Čapek, P., Pohl, R., Hocek, M. *J. Org. Chem.* **2005**, *70*, 8001; (c) Della Sala, G.; Izzo, I.; Spinella, A. *Synlett* **2006**, 1319; (d) Collier, P. N.; Patel, I.; Taylor, R. J. K. *Tetrahedron Lett.* **2002**, *43*, 3401; (e) Gibson, S. E.; Jones, J. O.; Kalindjian, S. B.; Knight, J. D.; Mainolfi, N.; Rudd, M.; Steed, J. W.; Tozer, M. J.; Wright, P. T. *Tetrahedron* **2004**, *60*, 6945; (f) Gurjar, M. K.; Talukdar, A. *Synthesis* **2002**, 315; (g) Huck, J.; Receveur, J.-M.; Roumestant, M.; Martinez, J. *Synlett* **2001**, 1467; (h) Møller, B.; Undheim, K. *Tetrahedron* **1998**, *54*, 5789; (i) Crisp, G. T.; Glink, P. T. *Tetrahedron* **1992**, *48*, 3541.
16. Ramón, R.; Martín-Gago, P.; Verdager, X.; Macias, M. J.; Martín-Malpartida, P.; Fernández-Carneado, J.; Gomez-Caminals, M.; Ponsati, B.; López-Ruiz, P.; Cortés, M. A.; Colás, B.; Riera, A. *ChemBioChem.* **2011**, *12*, 625.
17. (a) Huirne, J. A. F.; Lambalk, C. B. *Lancet* **2001**, *358*, 1793; (b) Yahalom, D.; Rahimpour, S.; Koch, Y.; Nurit Ben-Aroya, N.; Fridkin, M. *J. Med. Chem.* **2000**, *43*, 2831.
18. Dumas, P.; Benamar, D.; Heitz, F.; Ranjalahy-Rasoloarijao, L.; Mouden, R.; Lazaro, R.; Pullman, A. *Int. J. Pep. Prot. Res.* **1991**, *38*, 218.
19. Garner, P.; Park, J. M. *J. Org. Chem.* **1987**, *52*, 2361; *Org. Synth.* **1991**, *70*, 18.
20. (a) Moriwake, T.; Hamano, S.-i.; Saito, S.; Torii, S. *Chem. Lett.* **1987**, 2085; (b) Beaulieu, P. L.; Duceppe, J. S.; Johnson, C. *J. Org. Chem.* **1991**, *56*, 4196.
21. (a) McKillop, A.; Taylor, R. J. K.; Watson, R. J.; Lewis, N. *Synthesis* **1994**, 31; (b) Campbell, A. D.; Raynham, T. M.; Taylor, R. J. K. *Synthesis* **1998**, 1707.
22. Shimamoto, K.; Ishida, M.; Shinozaki, H.; Ohfuné, Y. *J. Org. Chem.* **1991**, *56*, 4167.
23. Truchot, C.; Wang, Q.; Sasaki, N. A. *Eur. J. Org. Chem.* **2005**, 1765.
24. (a) Beletskaya, I.; Cheprakov, A. V. *Chem. Rev.* **2000**, *100*, 3009; (b) Bräse, S.; de Meijere, A. In *Metal-Catalyzed Cross-Coupling Reactions*; F. Diederich and P. J. Stang (Eds.); Wiley-VCH: Weinheim, 2004; (c) Heck, R. F. *Org. React.* **1982**, *27*, 345.
25. Jeffery, T. *Tetrahedron Lett.* **1985**, *26*, 2667.
26. Avenoza, A.; Busto, J. H.; Corzana, F.; Jiménez-Osés, G.; Peregrina, J. M. *Tetrahedron* **2003**, *59*, 5713.