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## COMMUNICATION

## Ruthenium-catalyzed regioselective oxidative coupling of aromatic and heteroaromatic esters with alkenes under an open atmosphere<sup>†</sup>

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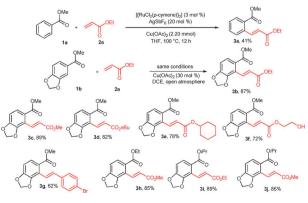
A ruthenium-catalyzed oxidative coupling of substituted aromatic and heteroaromatic esters with alkenes in the presence of catalytic amounts of  $AgSbF_6$  and  $Cu(OAc)_2$  to provide highly substituted alkene derivatives in good to excellent yields under an open atmosphere is described.

Chelation-assisted ortho C-H bond activation of directing group substituted aromatics and subsequent alkenylation with alkenes catalyzed by transition metal complexes is an efficient method for synthesizing substituted olefins in a highly regio- and stereoselective manner.<sup>1</sup> These reactions are environmentally friendly and highly atom-economical when compared with the other conventional methods such as carbon-halide functionalization.<sup>2</sup> In 1968, Fujiwara et al. demonstrated a palladium-catalyzed alkenylation of electron-rich aromatics with alkenes by C-H bond activation.3 In 1986, Lewis's group showed a ruthenium-catalyzed ortho ethylation of phenol with ethylene.<sup>4a</sup> In 1993, Murai's group described a ruthenium-catalyzed heteroatom-directed ortho C-H bond activation of substituted aromatics and subsequent addition with alkenes leading to substituted alkane derivatives.<sup>4b,c</sup> Later, palladium-catalyzed heteroatom-directed ortho-alkenvlation of substituted aromatics with alkenes via C-H bond activation has been demonstrated by the groups of Yu, Miura and others.<sup>5,6</sup> A meta-selective alkenylation of aromatics with alkenes catalyzed by a palladium pyridine complex has been illustrated by Yu and Sanford et al.<sup>7</sup> Subsequently, this type of heteroatom-directed ortho-alkenylation of substituted aromatics with alkenes in the presence of a rhodium catalyst has been successfully extended by the groups of Miura, Glorius and others.<sup>8</sup> Recently, a less expensive ruthenium complex has gained much attention in this type of reaction.<sup>9,10</sup> In this reaction,  $Cu(OAc)_2$  has been used as an internal oxidant. This oxidant provides the acetate source to the ruthenium species which facilitates ortho-metalation by a concerted deprotonation metalation pathway.1f

Various directing groups such as amines, oximes, amides, pyridyl, COOH, phenol, OH and carbonyl can be used in the metal-catalyzed *ortho* alkenylation reactions. In the presence of strong directing groups, C–H bond activation reactions are facile.

But, activation in the presence of weak directing groups such as aldehydes, esters, cyano and ketones are still a challenging task.<sup>10</sup> Herein, we would like to report *ortho*-alkenylation of aromatic esters with alkenes in the presence of catalytic amounts of [{RuCl<sub>2</sub>(*p*-cymene)}<sub>2</sub>], AgSbF<sub>6</sub> and Cu(OAc)<sub>2</sub>, affording highly substituted alkene derivatives in a highly regio- and stereoselective manner. The catalytic reaction is also compatible with various heteroaromatic esters. Interestingly, the present ruthenium-catalyzed alkenylation reaction is carried out under an open atmosphere and only a catalytic amount of Cu(OAc)<sub>2</sub> has been used as a terminal oxidant, the reduced copper source being reoxidized by air.<sup>9/,10</sup>

Treatment of methyl benzoate (1a) with ethyl acrylate (2a) in the presence of catalytic amounts of  $[{RuCl_2(p-cymene)}_2]$ (3 mol%), AgSbF<sub>6</sub> (20 mol%) and Cu(OAc)<sub>2</sub> (2.20 mmol) in THF at 100 °C for 12 h gave an alkene derivative 3a in 41% isolated yield (Scheme 1). To improve the yield of the reaction, in addition to an ester directing group, a very weak directing group such as 1,3-dioxol (-O-CH2-O-) was introduced at 3 and 4 positions of methyl benzoate 1a in order to further activate the aromatic C-H bond efficiently. Interestingly, in the reaction of methyl piperonate (1b) with ethyl acrylate (2a) under similar reaction conditions, the corresponding alkenylated compound 3b was obtained in excellent 85% isolated yield with a high E-stereoselectivity. The catalytic reaction is also regioselective and alkenylation takes place selectively at the sterically hindered C-H bond of 1b whereby both ester and 1,3-dioxol substituents act as directing groups. Thus, C-H bond activation takes place selectively at the sterically hindered C-H bond. Next, the same



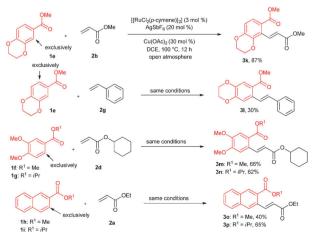
Scheme 1 Scope of alkenes 2a-f with piperonates 1b-d.

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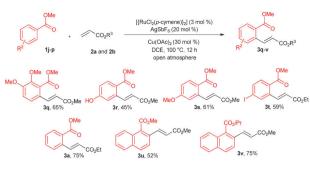
coupling reaction was carried out under an open atmosphere in the presence of  $Cu(OAc)_2$  (0.30 mmol, 30 mol%) under similar reaction conditions. Interestingly, the corresponding alkene derivative **3b** was observed in 86% yield. The catalytic reaction was also tested with various solvents such as CH<sub>3</sub>CN, toluene, MeOH, *tert*-BuOH, DMF, DME, DMA and DMSO. Among them, DCE (1,2-dichloroethane) solvent was equally effective for the reaction similar to that with THF, giving the coupling product **3b** in 87% yield under an open atmosphere in the presence of Cu(OAc)<sub>2</sub> (0.30 mmol, 30 mol%). Remaining solvents were totally ineffective for the reaction the absence of either a copper source or silver salt.

Under similar reaction conditions, various acrylates 2b-e were effectively coupled with substituted piperonates 1b-d (Scheme 1). Thus, the reaction of methyl piperonate (1b) with methyl acrylate (2b), n-butyl acrylate (2c) and cyclohexyl acrylate (2d) afforded the corresponding alkenylated products 3c-e in 89%, 82% and 78% yields, respectively. Very interestingly, 2-hydroxyethyl acrylate (2e) is also efficiently involved in the coupling reaction giving product 3f in 72% yield. The catalytic reaction was also tested with styrenes. Thus, 4-bromostyrene (2f) reacted smoothly with 1b yielding an alkenylated product 3g in 62% yield. Next, the effect of replacing the methyl group in methyl piperonate (1b) by other substituents such as ethyl and isopropyl groups was investigated. Thus, ethyl piperonate (1c) reacted nicely with methyl acrylate (2b) affording compound 3h in 85% yield. Similarly, isopropyl piperonate (1d) underwent coupling with ethyl and methyl acrylates 2a and 2b vielding coupling products 3i and 3i in 89% and 86% yields, respectively. These reactions were also highly regio- and stereoselective similar to that with 3b.

In addition to **1b–d**, the regioselectivities of other unsymmetrical aromatic esters **1e–i** were also examined under the same reaction conditions (Scheme 2). The reaction of methyl 1,4-benzodioxane-6-carboxylate (**1e**) with methyl acrylate (**2b**) provided substituted alkene derivative **3k** in 87% yield in a highly regio- and stereo-selective manner. Similar to that of **1b–d**, coupling reaction takes place at the sterically hindered C–H bond of **1e**. In contrast, **1e** reacted with styrene (**2g**) giving substituted *E*-stilbene derivative **3l**, albeit in only 30% yield in an opposite regiochemistry. In the reaction, the remaining amount of 70% of starting material **1e** was recovered. In the reaction, C–H bond of **1e**. Similarly, methyl



Scheme 2 Regioselective studies of substituted aromatic esters.



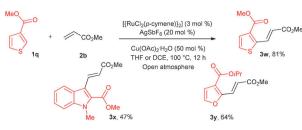
Scheme 3 Scope of aromatic esters.

3,4-dimethoxybenzoate (1f) and isopropyl 3,4-dimethoxybenzoate (1g) also regioselectively coupled with cyclohexyl acrylate (2d) providing coupling products 3m and 3n in 66% and 62% yields, respectively, in a highly regio- and stereoselective manner. In this case also, C–H bond activation takes place at the sterically less hindered C–H bond of 1f and 1g. A similar type of regioselective product 3o in moderate 40% yield was observed in the reaction of methyl 2-naphthoate (1h) with ethyl acrylate (2a) under similar reaction conditions. Interestingly, isopropyl 2-naphthoate (1i) reacted efficiently with 2a yielding the corresponding alkene derivative 3p in good 65% yield.

The present alkenvlation reaction was tested with various substituted aromatic esters 1 and ethyl or methyl acrylates (2a and 2b) under the optimized reaction conditions (Scheme 3). The reaction of methyl 2,3-dimethoxybenzoate (1) with methyl acrylate (2b) afforded coupling product 3q in 65% yield. Methyl 4-hydroxybenzoate (1k) reacted with 2b providing 3r albeit in moderate 45% yield. Whereas, methyl 4-methoxybenzoate (11) reacted with 2b affording 3s in 61% yield. Methyl 4-iodobenzoate (1m) also efficiently participated in the reaction giving the corresponding alkene derivative 3t in 59% yield. Whereas, methyl 2-iodobenzoate (1n) reacted with 2a providing coupling product 3a in 75% yield, in which the ortho I group of 1n participates in the coupling reaction *via* oxidative addition.<sup>2</sup> Methyl 1-napthoate (10) reacted with 2b affording the corresponding alkenylated derivative **3u** only in 52% yield. The yield of coupling product **3v** can be improved up to 75% by replacing the methyl group by an isopropyl group in 1-naphthoate (1p). The catalytic reaction was also tested with various electron-withdrawing group substituted aromatic esters such as methyl 4-nitrobenzoate and methyl 4-trifluorobenzoate under similar reaction conditions. However, in the reaction, no coupling product was observed and only starting material was recovered. It seems that electron-withdrawing group substituted aromatic ester was not a compatible substrate for the present coupling reaction.

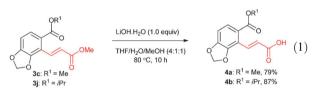
The present catalytic reaction was also examined with various heterocyclic esters (Scheme 4). Thus, methyl thiophene-3-carboxylate (1q) underwent coupling reaction with methyl acrylate (2b) providing an alkene derivative 3w in 81% yield. Substituted indole (1r) and isopropyl furan-3-carboxylate (1s) also efficiently reacted with methyl acrylate (2b) giving the corresponding alkenylated products 3x and 3y in 47% and 64% yields, respectively. In the substrate 1r, C–H bond activation takes place nicely at electron-rich C3–H carbon.

Next, the selective de-esterification of aromatic group substituted ester of compounds 3c and 3j was tried in the presence of LiOH·H<sub>2</sub>O (1.0 equiv.) in THF:H<sub>2</sub>O:MeOH

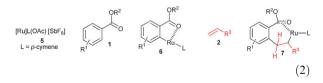


Scheme 4 Scope of heteroaromatic esters.

(4:1:1) at 80 °C for 12 h. Surprisingly, in the reaction, de-esterification takes place very selectively at unsaturated ester providing substituted carboxylic acids **4a** and **4b** in 79% and 87% yields, respectively (eqn (1)). An ester group connected with aromatic moieties of **3c** and **3j** remains intact. For selective de-esterification of aromatic group substituted ester, the reaction was tried with various bases such as NaOH and KOH under the same reaction conditions. In these reactions also, de-esterification of methyl piperonate (**1b**) with acrylic acid (**2h**) was tried under the optimized reaction conditions. However, in the reaction, coupling product **4a** was not observed and only acrylic acid dimerization was observed. By using the present de-esterification reaction, substituted acrylic acid derivatives can be synthesized in excellent yields.



The catalytic reaction proceeds *via* reaction of the [{RuCl<sub>2</sub>(*p*-cymene)}<sub>2</sub>] complex with AgSbF<sub>6</sub> giving cationic ruthenium complex **5** (eqn (2)). Coordination of the carbonyl oxygen of aromatic ester **1** to the ruthenium cationic species **5** followed by *ortho*-metalation affords ruthenacycle intermediate **6** (eqn (2)).<sup>1/</sup> Insertion of alkene **2** into the Ru–carbon bond of intermediate **6** provides a seven-membered ruthenacycle intermediate **7** (eqn (2)).  $\beta$ -Hydride elimination of intermediate **7** yields coupling product **3** and regenerates the active ruthenium species for the next catalytic cycle. The remaining amount of the active Cu(OAc)<sub>2</sub> source is regenerated under an atmosphere from the reduced copper source such as CuOAc.



In conclusion, we have developed a ruthenium-catalyzed highly regioselective *ortho*-alkenylation of aromatic and heteroaromatic esters with alkenes giving substituted alkene derivatives in a highly stereoselective manner. Further extension of the C–H bond activation of other directing group substituted aromatics and functionalization with other  $\pi$ -components and the detailed mechanistic investigation are in progress.

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## Notes and references

- Selected reviews: (a) X. Chen, K. M. Engle, D.-H. Wang and J.-Q. Yu, Angew. Chem., Int. Ed., 2009, 48, 5094; (b) T. W. Lyons and M. S. Sanford, Chem. Rev., 2010, 110, 1147; (c) J. L. Bras and J. Muzart, Chem. Rev., 2011, 111, 1170; (d) C. S. Yeung and V. M. Dong, Chem. Rev., 2011, 111, 1215; (e) H. M. L. Davies, J. D. Bois and J.-Q. Yu, Chem. Soc. Rev., 2011, 40, 1855, references therein; (f) L. Ackermann, Chem. Rev., 2011, 111, 1351.
- 2 Selected papers: (a) S. A. Shahzad and T. Wirth, Angew. Chem., Int. Ed., 2009, 48, 2588; (b) S. A. Shahzad, C. Venin and T. Wirth, Eur. J. Org. Chem., 2010, 3465.
- 3 Selected papers: (a) Y. Fujiwara, I. Moritani, M. Matsuda and S. Teranishi, *Tetrahedron Lett.*, 1968, 3863; (b) Y. Fujiwara, I. Moritani, S. Danno, R. Asano and S. Teranishi, *J. Am. Chem. Soc.*, 1969, **91**, 7166.
- 4 (a) L. N. Lewis and J. F. Smith, J. Am. Chem. Soc., 1986, 108, 2728; (b) S. Murai, F. Kakiuchi, S. Sekine, Y. Tanaka, A. Kamatani, M. Sonoda and N. Chatani, Nature, 1993, 366, 529; (c) F. Kakiuchi and S. Murai, Acc. Chem. Res., 2002, 35, 826.
- 5 Selected references: (a) S. E. Diamond, A. Szalkiewicz and F. Mares, J. Am. Chem. Soc., 1979, 101, 490; (b) L.-C. Kao and A. Sen, J. Chem. Soc., Chem. Commun., 1991, 1242; (c) M. Miura, T. Tsuda, T. Satoh, S. Pivsa-Art and M. Nomura, J. Org. Chem., 1998, 63, 5211; (d) K. Li, L. N. Foresee and J. A. Tunge, J. Org. Chem., 2005, 70, 2881; (e) E. M. Beck, N. P. Grimster, R. Hatley and M. J. Gaunt, J. Am. Chem. Soc., 2006, 128, 2528; (f) T. Nishikata and B. H. Lipshutz, Org. Lett., 2010, 12, 1972; (g) P. Gandeepan, K. Parthasarathy and C.-H. Cheng, J. Am. Chem. Soc., 2010, 132, 8569; (h) J. Karthikeyan and C.-H. Cheng, Angew. Chem., Int. Ed., 2011, 50, 9880.
- (a) J.-J. Li, T.-S. Mei and J.-Q. Yu, Angew. Chem., Int. Ed., 2008, 47, 6452; (b) K. M. Engle, D.-H. Wang and J.-Q. Yu, J. Am. Chem. Soc., 2010, 132, 14137; (c) D.-H. Wang, K. M. Engle, B.-F. Shi and J.-Q. Yu, Science, 2010, 327, 315; (d) B.-F. Shi, Y.-H. Zhang, J. K. Lam, D.-H. Wang and J.-Q. Yu, J. Am. Chem. Soc., 2010, 132, 460; (e) Y. Lu, D.-H. Wang, K.-M. Engle and J.-Q. Yu, J. Am. Chem. Soc., 2010, 132, 5916; (f) M. Ye, G.-L. Gao and J.-Q. Yu, J. Am. Chem. Soc., 2011, 133, 6964.
- 7 (a) Y.-H. Zhang, B.-F. Shi and J.-Q. Yu, J. Am. Chem. Soc., 2009, 131, 5072; (b) A. Kubota, M. H. Emmert and M. S. Sanford, Org. Lett., 2012, 14, 1760.
- Selected references: (a) N. Umeda, K. Hirano, T. Satoh and M. Miura, J. Org. Chem., 2009, 74, 7094; (b) S. Mochida, K. Hirano, T. Satoh and M. Miura, Org. Lett., 2010, 12, 5776; (c) S. Rakshit, C. Grohmann, T. Besset and F. Glorius, J. Am. Chem. Soc., 2011, 133, 2350; (d) F. W. Patureau and F. Glorius, J. Am. Chem. Soc., 2010, 132, 9982; (e) F. W. Patureau, T. Besset and F. Glorius, Angew. Chem., Int. Ed., 2011, 50, 1064; (f) C. Feng and T.-P. Loh, Chem. Commun., 2011, 47, 10458; (g) S. Park, J. Y. Kim and S. Chang, Org. Lett., 2011, 13, 2372; (h) F. Wang, G. Song and X. Li, Org. Lett., 2010, 12, 5430; (i) T.-J. Gong, B. Xiao, Z.-J. Liu, J. Wan, J. Xu, D.-F. Luo, Y. Fu and L. Liu, Org. Lett., 2011, 13, 3235; (j) K. Muralirajan, K. Parthasarathy and C.-H. Cheng, Angew. Chem., Int. Ed., 2011, 50, 4169; (k) A. S. Tsai, M. Brasse, R. G. Bergman and J. A. Ellman, Org. Lett., 2011, 13, 540; (l) J. Jayakumar, K. Parthasarathy and C.-H. Cheng, Angew. Chem., Int. Ed., 2012, 51, 197.
- Alkylation: (a) M. Sonoda, F. Kakiuchi, A. Kamatani, N. Chatani and S. Murai, *Chem. Lett.*, 1996, 109; Alkenylation: (b) T. Ueyama, S. Mochida, T. Fukutani, K. Hirano, T. Satoh and M. Miura, *Org. Lett.*, 2011, 13, 706; (c) L. Ackermann and J. Pospech, *Org. Lett.*, 2011, 13, 4153; (d) L. Ackermann, L. Wang, R. Wolfram and A. V. Lygin, *Org. Lett.*, 2012, 14, 728; (e) B. Li, J. Ma, N. Wang, H. Feng, S. Xu and B. Wang, *Org. Lett.*, 2012, 14, 736; (f) L. Ackermann, L. Wang and A. V. Lygin, *Chem. Sci.*, 2012, 3, 177; (g) L. Ackermann, A. V. Lygin and N. Hofmann, *Angew. Chem., Int. Ed.*, 2011, 50, 6379.
- 10 (a) P. Kishor and M. Jeganmohan, Org. Lett., 2012, 14, 1134;
  (b) P. Kishor and M. Jeganmohan, Org. Lett., 2011, 13, 6144;
  (c) C. G. Ravi Kiran and M. Jeganmohan, Eur. J. Org. Chem., 2012, 417; (d) C. G. Ravi Kiran and M. Jeganmohan, Chem. Commun., 2012, 48, 2030.