



Synthesis of 5-arylamino-1-arylpyrazoles from 5-aminopyrazoles with arylhalides via CuI catalyzed Ullman coupling reaction



En-Chiuan Chang ^b, Chun-Yen Chen ^{b,1}, Li-Ya Wang ^c, Yu-Ying Huang ^a, Mou-Yung Yeh ^{b,*}, Fung Fuh Wong ^{a,*}

^a Graduate Institute of Pharmaceutical Chemistry, China Medical University, No. 91, Hsueh-Shih Rd., Taichung 40402, Taiwan, ROC

^b Department of Chemistry, National Cheng Kung University, No. 1, Ta Hsueh Rd., Tainan 70101, Taiwan, ROC

^c The Ph.D. Program for Cancer Biology and Drug Discovery, China Medical University, No. 91, Hsueh-Shih Rd., Taichung 40402, Taiwan, ROC

ARTICLE INFO

Article history:

Received 6 September 2012

Received in revised form 30 October 2012

Accepted 6 November 2012

Available online 14 November 2012

ABSTRACT

An efficient method for synthesis of 5-arylamino-1-arylpyrazoles was developed by Ullman coupling reaction of 5-aminopyrazoles and arylhalides in the presence of cuprous iodide (CuI) as the catalyst. The corresponding 5-N-arylamino pyrazole products were obtained in good yields ($\geq 65\%$).

© 2012 Elsevier Ltd. All rights reserved.

Keywords:

Cross-coupling reaction

5-Amino-pyrazoles

5-Mono-N-substituted aminopyrazoles

Amination

Copper iodide

1. Introduction

Numerous compounds containing 1-N-arylpyrazole moiety have been shown to exhibit antihyperglycemic, analgesic, anti-inflammatory sedative, and hypnotic activities.^{1–3} 5-Alkyl/arylamino substituted pyrazoles have been exploited in the design of pharmaceuticals and agrochemical agents exhibiting a range of biological activities.^{2,4–6} Several previous literature were reported for the synthesis of 1-aryl-5-N,N-disubstituted aminopyrazoles by means of the cyclization of hydrazines with β -ketoamides,^{5d,6,7} the defined starting precursors containing the hydrazones with an electron withdrawing group,^{5b,8} or the pre-generated 2-oxocarbothioamides (β -ketothioamides).⁹ However, a few methods could provide the satisfactory reaction conditions for the preparation of the 5-arylamino-1-arylpyrazoles. Herein, we reported an efficient method for the synthesis of 5-arylamino-1-arylpyrazoles derivatives, of which analogues are known selective COX-2 inhibitors,^{2,10} via CuI-catalyzed Ullman coupling reaction.

Introduction of nitrogen functionalities on aromatic rings has been one of the main topics in organic synthesis due to compounds

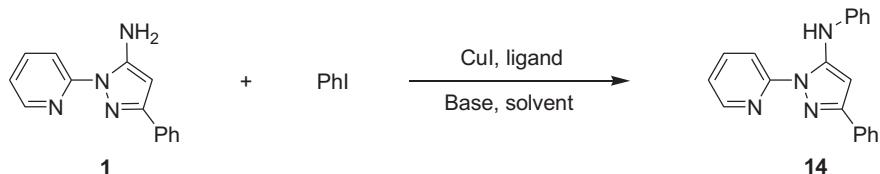
containing aromatic C(sp²)–N bonds possess various important pharmacologically activities.¹¹ Many metal-catalyzed aryl amination methods were reported by using palladium catalyst and phosphine ligand^{11–13} or copper iodide and salicylic amide derivatives or proline.^{14–16} However, most of methods were performed in the amination of aryl compounds. That we provided the new development of a novel 5-amino-pyrazoles amination catalyzed by CuI, K₂CO₃, and 1,10-phenanthroline for efficient synthesis of 5-arylamino-1-arylpyrazole derivatives in this study was due to the copper-mediated reactions are significantly superior to the conventional palladium-mediated reactions.¹⁷

2. Result and discussion

The 5-aminopyrazoles **1–13** were prepared as the starting substrates by following our previous published procedure via cyclodehydration and dehydrogenation in the neat condition.¹⁸ In this newly developed method, we chose 3-phenyl-1-(pyridin-2-yl)-1H-pyrazol-5-amine (**1**)¹⁸ as the model for the selection of best condition for the CuI-catalyzed Ullman coupling reaction (see Scheme 1). To search for the optimum conditions and establish a reproducible procedure, we firstly investigated the ligand effect. Compound **1** was allowed to react with phenyl iodide (PhI), K₂CO₃, and the various ligands including L-proline, DMEDA, pyrazole-2-COOH, N-methylglycine, and 1,10-phenanthroline in the presence

* Corresponding authors. E-mail address: wongfungfu@yahoo.com.tw (F.F. Wong).

¹ The contribution equal to first author.



Scheme 1.

of the catalytic amount of CuI (0.10 equiv) at 70 °C for 1.0 h in DMF. Following the experimental results, we found that the use of 0.10 equiv of 1,10-phenanthroline as the ligand in DMF provided the best result to give the corresponding product **14** in 84% yield (see entries 1–5 of Table 1). Consequently, we explored the cross-coupling reaction by modifying the polarity of aprotic polar solvents, such as DMF, DMSO, and DMAC. Based on the result, the 0.10 equiv of 1,10-phenanthroline in DMF solution can be considered as the best suitable ligand and solvent system (see entry 5–7 of Table 1).

Table 1
The result of optimization studies

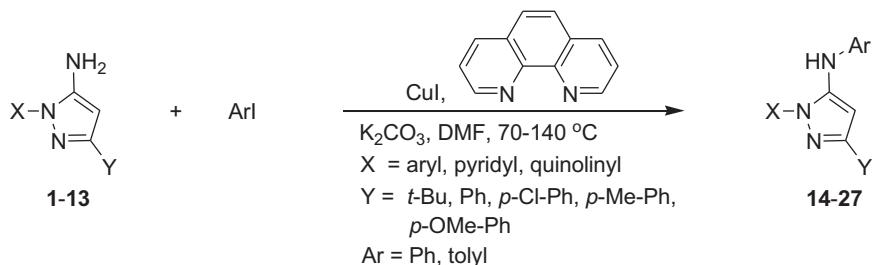
Entry	Ligand	Solvent	Base	Yields ^{a,b} (%) of compound 14
1	L-Proline	DMF	K ₂ CO ₃	62
2	DMEDA	DMF	K ₂ CO ₃	77
3	Pyrazole-2-COOH	DMF	K ₂ CO ₃	74
4	N-Methylglycine	DMF	K ₂ CO ₃	16
5	1,10-Phenanthroline	DMF	K ₂ CO ₃	84
6	1,10-Phenanthroline	DMSO	K ₂ CO ₃	52
7	1,10-Phenanthroline	DMAC	K ₂ CO ₃	34
8	1,10-Phenanthroline	DMF	Li ₂ CO ₃	34
9	1,10-Phenanthroline	DMF	Na ₂ CO ₃	39
10	1,10-Phenanthroline	DMF	Cs ₂ CO ₃	43
11	1,10-Phenanthroline	DMF	t-BuOK	60
12	1,10-Phenanthroline	DMF	NaOMe	55
13	1,10-Phenanthroline	DMF	NaHCO ₃	54
14	1,10-Phenanthroline	DMF	K ₃ PO ₄	72
15	1,10-Phenanthroline	DMF	—	Non-detectable

^a Compound **1** (1.0 mmol, 1.0 equiv), PhI (1.0 mmol, 1.0 equiv), and various bases (1.2 mmol, 1.2 equiv) and ligands (0.10 mmol, 0.10 equiv) with catalytic amount CuI (0.10 mmol, 0.10 equiv) in aprotic polar solvents at 70 °C for 1.0 h under nitrogen gas.

^b The isolated yields were determined by column chromatography on silica gel.

with column chromatography on silica gel, the desired 5-N-monarylaminopyrazole **14** was isolated as solids (entries 5 and 8–12 of Table 1). Among of bases, we found that the use of potassium phosphate (K₃PO₄) and potassium carbonate (K₂CO₃) gave the better results in 72% and 84% yields, respectively (see entries 5 and 14 of Table 1). Yields of the desired 5-arylamino-1-arylpypyrazole **14** after 1.0 h were substantially decreased when using potassium, sodium, and lithium carbonate salts (entries 5 and 8–9 of Table 1). However, low yield was also observed with Cs₂CO₃ (entry 10 of Table 1). Conditions using other bases, such as sodium bicarbonate (NaHCO₃), potassium *tert*-butoxide (*t*-BuOK), or without bases resulted in only poor result (entries 11 and 13–15 of Table 1). Based on these optimizations, we finally decided to choose the combination of CuI, 1,10-phenanthroline, and K₂CO₃ in DMF as the standard conditions for the metal catalyzed cross-coupling reaction of 5-arylamino-1-arylpypyrazoles in the following studies.

In this reaction, the reliable procedure involved the treatment of 5-amino-pyrazole substrates, aryl halide, and K₂CO₃ using 0.10 equiv of catalytic amount CuI and 1,10-phenanthroline ligand in anhydride DMF at 70 or 140 °C for more than 1.0 h (see Scheme 2). For investigation of the reactivity of different phenyl halide substrates, the amination reaction was performed with phenyl chloride, bromide, and iodide, and tolyl iodide. After comparing the results, we found that the use of phenyl and tolyl iodides were obviously better than the use of phenyl bromide and chloride (entries 1–4 of Table 2). For phenyl chloride, reaction was incomplete even after 7.0 h at 140 °C (entry 1 of Table 2). These results were consistent with the literature amination order.¹⁸ Application of same cross-coupling reaction conditions to 3-*p*-methylphenyl-1-(pyridin-2-yl)-1*H*-pyrazol-5-amine **2**, 3-*p*-methoxyphenyl-1-(pyridin-2-yl)-1*H*-pyrazol-5-amine **3**, and 3-*tert*-butyl-1-(pyridin-2-yl)-1*H*-pyrazol-5-amine **4** with various of aryl iodine, including PhI, tolyl iodine, and *p*-methoxyphenyl iodine also gave the corresponding products **16–18** in good yields (80–93%, entries 5–7 of Table 2).



Scheme 2.

We knew that the appropriate choice of base might be important for the cross-coupling reaction. Thus, we treated an anhydrous DMF solution of 3-phenyl-1-(pyridin-2-yl)-1*H*-pyrazol-5-amine (**1**) and PhI with catalytic amount of CuI and 1,10-phenanthroline in the presence of 1.2 equiv of different bases, including lithium, sodium, cesium and potassium carbonate (Li₂CO₃, Na₂CO₃, Cs₂CO₃ and K₂CO₃), potassium *tert*-butoxide (*t*-BuOK), sodium methoxide (NaOMe), sodium bicarbonate (NaHCO₃) and potassium phosphate (K₃PO₄) at 70 °C for 1.0 h. After normal work-up and purification

The synthetic strategy was applicable to 3-*p*-tolyl-1-(quinolin-2-yl)-1*H*-pyrazol-5-amine **5** with phenyl chloride, bromide, and iodide to afford the desired amination products **18** in 23%, 77%, and 96% yields, respectively (see entries 8–10 of Table 2). Although phenyl bromide and chloride were also successfully coupled by using the CuI/L catalyst system, higher temperatures (>100 °C) and long reaction times were required. For further demonstration of the reactivity of phenyl halides, 3-(4-methoxyphenyl)-1-(quinolin-2-yl)-1*H*-pyrazol-5-amine **6**, 3-*tert*-butyl-1-(quinolin-2-yl)-1*H*-

Table 2

The result of CuI catalyzed Ullman coupling reaction of 5-aminopyrazoles with arylhalides

Entry	5-Aminopyrazoles	ArX	Reaction time & temperature		Products	Yields ^{a,b} (%)		
			h	°C				
1		1	PhCl	48	140		14	—
2		1	PhBr	7	70		14	75
3		1	PhI	4	70		14	91
4		1	Tolyl-I	3	70		15	84
5		2	Tolyl-I	2	70		16	93
6		3	p-OMe-Ph-I	1	70		17	80
7		3	p-NO ₂ -Ph-I	>3	70	— ^c	—	— ^c
8		3	p-NO ₂ -Ph-Br	>7	70	— ^c	—	— ^c
9		4	PhI	3	70		18	86
10		5	PhCl	12	140		19	23
11		5	PhBr	5	70		19	77
12		5	PhI	3	70		19	96
13		6	PhCl	22	140		20	21

^a Isolated yield.^b Determined by ¹H NMR.^c Not detected.

Table 2 (continued)

Entry	5-Aminopyrazoles	ArX	Reaction time & temperature		Products	Yields ^{a,b} (%)	
			h	°C			
14		6	PhBr	3	70		20 68
15		6	PhI	3	70		20 94
16		7	PhCl	48	140		21 17
17		7	PhBr	24	70		21 73
18		7	PhI	22	70		21 95
19		8	Tolyl-I	3	70		22 90
20		9	Tolyl-I	3	70		23 86
21		10	PhI	30	70		24 78
22		11	PhI	36	70		25 72
23		12	PhI	2	70		26 88
24		13	Tolyl-I	36	70		27 65

^a 5-Amino-pyrazole substrates **1–13** (1.0 mmol, 1.0 equiv), arylhalides (PhCl, PhBr, PhI, *p*-MeO-PhI, or tolyl iodide, 1.0 mmol, 1.0 equiv), and K₂CO₃ (1.2 mmol, 1.2 equiv) with catalytic amount CuI (0.10 mmol, 0.10 equiv) and 1,10-phenanthroline (0.10 mmol, 0.10 equiv) in DMF at 70 or 140 °C within 1–48 h under nitrogen gas.

^b The isolated yields were performed by column chromatography on silica gel.

^c Non-detectable.

pyrazol-5-amine **7** were allowed to react with phenyl halides (PhCl, PhBr, and PhI) under the same reaction conditions. Following the experimental results, the similar reactive tendency of phenyl halides (PhI>PhBr>>PhCl) were detected in the newly developed Ullman coupling amination (see the entries 11–16 of Table 2). When employing 3-phenyl-1-(quinolin-2-yl)-1*H*-pyrazol-5-amine **8** and 3-*p*-chlorophenyl-1-(quinolin-2-yl)-1*H*-pyrazol-5-amine **9** as the reactant to treat with tolyl iodine, a good to excellent yields were also achieved (90% and 86%, entry 17–18 of Table 2).

On the other hand, we extended the new developed coupling amination to 1-aryl-3-phenyl-5-amino-1*H*-pyrazoles **10–13**, bearing various *N*-1 substituents including Ph, *m*-Cl–Ph, *m*-NO₂Ph, and *p*-OMePh. The corresponding 5-arylamino-1-aryl-3-phenyl-pyrazole products **24–27** were obtained in good yields (65–88%, entries 19–22 in Table 2). All 5-mono-*N*-substituted aminopyrazole Products **14–27** were fully characterized by spectroscopic methods and consistent with the literature data. Served as an example, compound **14** presented a peak at δ 6.41 ppm for ¹H–C= in C-4 position of pyrazole ring in ¹H NMR. In ¹³C NMR spectrum, compound **14** possessed characterization absorptions at δ 87.30 ppm for sp² carbon ¹³C–H and at δ 146.80 ppm for imine carbon N=¹³C in pyrazole ring. The IR absorptions of **14** showed peaks at 3215 cm^{−1} for the stretching of the –NH group.

3. Conclusion

In conclusion, we have successfully developed a new cross-coupling method for the synthesis of 5-arylamino-1-arylpypyrazoles using 1,10-phenanthroline as the ligand and CuI as catalyst. This methodology can applicable to the wide range of *N*-1 substituted aryl, pyridin-2-yl, isoquinolinyl or C-3 substituted aryl and *tert*-butyl of 5-amino-pyrazole substrates to give the corresponding 5-arylamino-1-arylpypyrazole products in good yields.

4. Experimental section

4.1. General procedure for the CuI catalyzed cross-coupling reaction of primary 5-amino-pyrazoles

A 5-amino-pyrazole substrates (1.0 mmol, 1.0 equiv), arylhalides (PhCl, PhBr, PhI, *p*-MeO–PhI, or tolyl iodide, 1.0 mmol, 1.0 equiv), and K₂CO₃ (1.2 mmol, 1.2 equiv) with catalytic amount CuI (0.1 mmol, 0.1 equiv) and 1,10-phenanthroline (0.1 mmol, 0.1 equiv) in DMF at 70 or 140 °C within 1–48 h under nitrogen gas. After 5-amino-pyrazoles were completely consumed, the reaction mixture was work-up, extracted with CH₂Cl₂ (20 mL×2), and concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to give the corresponding 5-arylamino-1-arylpypyrazole products in 17–96%.

4.1.1. 5-Anilino-3-phenyl-1-(2-pyridyl)-1*H*-pyrazole (14). Yellow powder; mp 98–99 °C; ¹H NMR (CDCl₃, 300 MHz) δ 6.41 (s, 1H, Py), 7.01 (t, 1H, J =7.1 Hz, ArH), 7.13 (t, 1H, J =6.4 Hz, ArH), 7.28–7.45 (m, 7H, ArH), 7.81–7.92 (m, 3H, ArH), 8.19 (d, 1H, J =8.4 Hz, ArH), 8.38 (d, 1H, J =4.3 Hz, ArH), 10.59 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 87.30, 114.24, 118.20, 119.71, 121.76, 125.91, 128.31, 128.52, 129.38, 133.11, 138.83, 141.27, 146.03, 146.80, 152.25, 154.96; IR (diffuse reflectance) 3215 (m), 2927 (m), 2862 (w), 1682 (m), 1651 (s), 1557 (s), 1538 (m), 1505 (m), 1455 (m), 1252 (m), 1140 (m), 1075 (m) cm^{−1}; MS (ESI) *m/z*: 313 (M+H)⁺. HRMS (ESI) calcd for C₂₀H₁₇N₄ (M+H)⁺ 313.1453, found 313.1454.

4.1.2. 3-Phenyl-1-(2-pyridyl)-5-*p*-toluidine-1*H*-pyrazole (15). Yellow powder; mp 77–78 °C; ¹H NMR (CDCl₃, 300 MHz) δ 2.34 (s, 3H, CH₃), 6.32 (s, 1H, Py), 7.10–7.22 (m, 5H, ArH), 7.34–7.44 (m, 3H, ArH), 7.80–7.86 (m, 1H, ArH), 7.90 (d, 2H, J =7.4 Hz, ArH), 8.18 (d, 1H,

J =8.3 Hz, ArH), 8.37 (d, 1H, J =4.5 Hz, ArH), 10.43 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 20.72, 86.75, 114.15, 117.65, 118.77, 119.59, 125.89, 128.25, 128.49, 129.87, 131.50, 133.18, 138.76, 146.03, 147.47, 152.27, 154.99; IR (diffuse reflectance) 3282 (m), 3056 (w), 2924 (m), 2855 (m), 1683 (m), 1651 (s), 1557 (m), 1505 (s), 1455 (s), 1361 (m), 1262 (m), 1044 (m), 955 (m) cm^{−1}; MS (ESI) *m/z*: 327 (M+H)⁺. HRMS (ESI) calcd for C₂₁H₁₉N₄ (M+H)⁺ 327.1610, found 327.1609.

4.1.3. 1-(2-Pyridyl)-5-*p*-toluidine-3-*p*-tolyl-1*H*-pyrazole (16). Yellow powder; mp 86–87 °C; ¹H NMR (CDCl₃, 300 MHz) δ 2.33 (s, 3H, CH₃), 2.38 (s, 3H, CH₃), 6.29 (s, 1H, Py), 7.09–7.23 (m, 7H, ArH), 7.76–7.85 (m, 3H, ArH), 8.16 (d, 1H, J =8.3 Hz, ArH), 8.36 (dd, 1H, J =4.8, 1.1 Hz, ArH), 10.42 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 20.72, 21.32, 86.64, 114.12, 118.73, 119.48, 125.79, 129.19, 129.86, 130.37, 131.41, 138.09, 138.72, 138.80, 146.01, 147.38, 152.34, 155.01; IR (diffuse reflectance) 3253 (m), 2923 (m), 2852 (m), 1594 (s), 1557 (s), 1539 (m), 1468 (m), 1442 (m), 1362 (m), 1248 (m), 948 (m) cm^{−1}; MS (ESI) *m/z*: 341 (M+H)⁺. HRMS (ESI) calcd for C₂₂H₂₁N₄ (M+H)⁺ 341.1766, found 341.1764.

4.1.4. 5-Anisidino-3-(4-methoxyphenyl)-1-(2-pyridyl)-1*H*-pyrazole (17). Yellow powder; mp 110–111 °C; ¹H NMR (CDCl₃, 300 MHz) δ 3.81 (s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 6.12 (s, 1H, Py), 6.89–7.26 (m, 7H, ArH), 7.79–7.81 (m, 3H, ArH), 8.14 (d, 1H, J =8.3 Hz, ArH), 8.34 (d, 1H, J =3.4 Hz, ArH), 10.22 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 55.27, 55.57, 85.64, 113.86, 113.91, 114.63, 119.34, 121.30, 125.95, 127.15, 134.65, 138.69, 146.03, 148.43, 152.10, 154.99, 155.28, 159.77; IR (diffuse reflectance) 3243 (m), 2923 (m), 2852 (m), 1593 (s), 1556 (s), 1531 (s), 1470 (s), 1441 (s), 1361 (m), 1295 (m), 1246 (s), 1173 (m), 1033 (m), 948 (m) cm^{−1}; MS (ESI) *m/z*: 373 (M+H)⁺. HRMS (ESI) calcd for C₂₂H₂₁N₄O₂ (M+H)⁺ 373.1665, found 373.1667.

4.1.5. 5-Anilino-3-*tert*-butyl-1-(2-pyridyl)-1*H*-pyrazole (18). ¹H NMR (CDCl₃, 300 MHz) δ 1.34 (s, 9H, 3×CH₃), 5.95 (s, 1H, Py), 6.96 (t, 1H, J =7.3 Hz, ArH), 7.06 (t, 1H, J =6.2 Hz, ArH), 7.21 (d, 2H, J =8.2 Hz, ArH), 7.29–7.34 (m, 2H, ArH), 7.77 (td, 1H, J =8.1, 1.4 Hz, ArH), 8.04 (d, 1H, J =8.4 Hz, ArH), 8.32 (d, 1H, J =4.9 Hz, ArH), 10.50 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 30.04, 32.51, 86.89, 114.13, 117.81, 119.12, 121.28, 129.27, 138.62, 141.51, 145.81, 145.93, 155.10, 163.42; IR (diffuse reflectance) 3247 (m), 2949 (m), 2923 (m), 2854 (m), 1592 (s), 1556 (s), 1470 (s), 1441 (s), 1365 (m), 1255 (m), 1132 (m), 1086 (m), 981 (m) cm^{−1}; MS (ESI) *m/z*: 293 (M+H)⁺. HRMS (ESI) calcd for C₁₈H₂₁N₄ (M+H)⁺ 293.1766, found 293.1767.

4.1.6. 5-Anilino-1-(2-quinolinyl)-3-*p*-tolyl-1*H*-pyrazole (19). Yellow powder; mp 160–161 °C; ¹H NMR (CDCl₃, 300 MHz) δ 2.40 (s, 3H, CH₃), 6.44 (s, 1H, Py), 7.03–7.06 (m, 1H, ArH), 7.27–7.51 (m, 7H, ArH), 7.69–7.85 (m, 4H, ArH), 7.95 (d, 1H, J =8.5 Hz, ArH), 8.24 (d, 1H, J =9.0 Hz, ArH), 8.40 (d, 1H, J =8.9 Hz, ArH); ¹³C NMR (CDCl₃, 75 MHz) δ 21.35, 87.21, 114.08, 117.84, 121.66, 125.65, 125.73, 125.88, 127.34, 127.74, 129.25, 129.47, 130.20, 130.26, 138.30, 138.80, 141.24, 145.00, 147.09, 152.57, 154.00; IR (diffuse reflectance) 3210 (m), 3057 (m), 2922 (m), 2855 (m), 1602 (s), 1589 (s), 1556 (s), 1504 (s), 1455 (m), 1430 (s), 1371 (s), 1252 (m), 1119 (m), 1042 (m), 947 (m) cm^{−1}; MS (ESI) *m/z*: 377 (M+H)⁺. HRMS (ESI) calcd for C₂₅H₂₁N₄ (M+H)⁺ 377.1766, found 377.1763.

4.1.7. 5-Anilino-3-(4-methoxyphenyl)-1-(2-quinolinyl)-1*H*-pyrazole (20). Yellow powder; mp 183–184 °C; ¹H NMR (CDCl₃, 300 MHz) δ 3.86 (s, 3H, OCH₃), 6.40 (s, 1H, Py), 6.96–7.05 (m, 3H, ArH), 7.33–7.51 (m, 5H, ArH), 7.69–7.96 (m, 5H, ArH), 8.24 (d, 1H, J =9.0 Hz, ArH), 8.38 (d, 1H, J =9.2 Hz, ArH), 11.40 (br, 1H, NH); ¹³C NMR (CDCl₃, 75 MHz) δ 55.30, 87.00, 113.95, 114.04, 117.86, 121.66,

125.62, 125.70, 125.77, 127.28, 127.32, 127.74, 129.47, 130.26, 138.79, 141.24, 145.01, 147.11, 152.34, 153.99, 159.93; IR (diffuse reflectance) 3195 (w), 3072 (m), 2920 (m), 2847 (m), 1591 (s), 1556 (s), 1504 (s), 1455 (m), 1428 (s), 1366 (m), 1245 (s), 1174 (m), 1025 (m), 944 (m) cm^{-1} ; MS (ESI) m/z : 393 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{21}\text{N}_4\text{O}$ ($\text{M}+\text{H}$) $^+$ 393.1715, found 393.1716.

4.1.8. 5-Anilino-3-tert-butyl-1-(2-quinolinyl)-1*H*-pyrazole (21). Yellow powder; mp 119–120 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 1.39 (s, 9H, 3 \times CH_3), 6.03 (s, 1H, Py), 7.00 (t, 1H, $J=7.0$ Hz, ArH), 7.27–7.49 (m, 5H, ArH), 7.67–7.72 (m, 1H, ArH), 7.78 (d, 1H, $J=8.0$ Hz, ArH), 7.92 (d, 1H, $J=8.4$ Hz, ArH), 8.19 (d, 1H, $J=9.1$ Hz, ArH), 8.29 (d, 1H, $J=9.1$ Hz, ArH), 11.32 (br, 1H, NH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 30.04, 32.62, 86.98, 114.18, 117.56, 121.29, 125.39, 125.54, 127.29, 127.70, 129.40, 130.11, 138.58, 141.47, 145.07, 146.28, 154.16, 163.69; IR (diffuse reflectance) 3456 (m), 3223 (m), 3063 (w), 2959 (m), 2862 (m), 1586 (s), 1556 (s), 1504 (s), 1430 (s), 1371 (s), 1225 (m), 1134 (m), 1044 (m), 981 (m) cm^{-1} ; MS (ESI) m/z : 343 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{22}\text{H}_{23}\text{N}_4$ ($\text{M}+\text{H}$) $^+$ 343.1923, found 343.1922.

4.1.9. 3-Phenyl-1-(2-quinolinyl)-5-*p*-toluidine-1*H*-pyrazole (22). Yellow powder; mp 174–175 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 2.36 (s, 3H, CH_3), 6.39 (s, 1H, Py), 7.18–7.52 (m, 8H, ArH), 7.72 (td, 1H, $J=7.5, 1.0$ Hz, ArH), 7.81 (d, 1H, $J=7.9$ Hz, ArH), 7.93–7.95 (m, 3H, ArH), 8.25 (d, 1H, $J=9.0$ Hz, ArH), 8.40 (d, 1H, $J=9.0$ Hz, ArH), 11.22 (br, 1H, NH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 20.74, 86.78, 114.04, 118.41, 125.65, 125.75, 125.98, 127.38, 127.73, 128.39, 128.53, 129.97, 130.25, 131.45, 133.09, 138.69, 138.79, 145.04, 147.81, 152.53, 154.01; IR (diffuse reflectance) 3210 (m), 3057 (m), 2922 (m), 2855 (w), 1621 (m), 1592 (s), 1556 (s), 1504 (s), 1430 (m), 1367 (m), 1249 (m), 1042 (m), 946 (m) cm^{-1} ; MS (ESI) m/z : 377 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{21}\text{N}_4$ ($\text{M}+\text{H}$) $^+$ 377.1766, found 377.1768.

4.1.10. 3-(4-Chlorophenyl)-1-(2-quinolinyl)-5-*p*-toluidine-1*H*-pyrazole (23). Yellow powder; mp 180–181 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 2.36 (s, 3H, CH_3), 6.31 (s, 1H, Py), 7.18–7.22 (m, 4H, ArH), 7.38 (d, 2H, $J=8.3$ Hz, ArH), 7.48 (t, 1H, $J=7.5$ Hz, ArH), 7.70 (t, 1H, $J=7.8$ Hz, ArH), 7.77–7.93 (m, 4H, ArH), 8.22 (d, 1H, $J=9.0$ Hz, ArH), 8.34 (d, 1H, $J=9.0$ Hz, ArH), 11.21 (br, 1H, NH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 20.75, 86.50, 113.88, 118.46, 125.72, 125.75, 127.20, 127.37, 127.72, 128.67, 129.98, 130.28, 131.58, 131.62, 134.08, 138.55, 138.81, 144.97, 147.94, 151.31, 153.87; IR (diffuse reflectance) 3217 (m), 3065 (m), 2921 (m), 2855 (w), 1602 (s), 1591 (s), 1556 (s), 1504 (s), 1465 (m), 1431 (s), 1367 (s), 1249 (m), 1090 (m), 1043 (m), 1013 (m), 946 (m) cm^{-1} ; MS (ESI) m/z : 413 ($\text{M}+3$) $^+$, 411 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{20}\text{ClN}_4$ ($\text{M}+\text{H}$) $^+$ 411.1376, found 411.1380.

4.1.11. 5-Anilino-1,3-diphenyl-1*H*-pyrazole (24). Yellow powder; mp 152–153 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 5.55 (br, 1H, NH), 6.52 (s, 1H, Py), 6.99 (d, 2H, $J=7.8$ Hz, ArH), 7.18–7.50 (m, 9H, ArH), 7.63 (d, 2H, $J=7.7$ Hz, ArH), 7.86 (d, 2H, $J=7.2$ Hz, ArH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 93.72, 115.91, 120.94, 124.38, 125.61, 127.67, 127.93, 128.50, 129.34, 129.44, 133.18, 138.33, 142.17, 142.98, 151.47; IR (diffuse reflectance) 3233 (m), 3004 (m), 2935 (m), 1596 (s), 1494 (m), 1452 (m), 1361 (m), 1301 (m), 1255 (w), 1165 (m) cm^{-1} ; MS (ESI) m/z : 312 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{18}\text{N}_3$ ($\text{M}+\text{H}$) $^+$ 312.1501, found 312.1503.

4.1.12. 5-Anilino-1-(3-chlorophenyl)-3-phenyl-1*H*-pyrazole (25). Yellow powder; mp 128–129 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 5.49 (br, 1H, NH), 6.51 (s, 1H, Py), 6.97 (d, 2H, $J=8.4$ Hz, ArH), 7.28–7.43 (m, 7H, ArH), 7.56 (d, 1H, $J=7.7$ Hz, ArH), 7.73 (s, 1H, ArH), 7.85 (d, 2H, $J=7.4$ Hz, ArH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 94.54, 116.10, 121.32, 121.86, 124.39, 125.60, 127.54, 128.18, 128.59, 129.55, 130.28, 132.92, 135.09, 139.60, 142.37, 142.79, 151.92; IR (diffuse reflectance) 3390 (m), 3065 (w), 2923 (m), 2855 (m), 1593 (s),

1557 (s), 1538 (m), 1362 (m), 1250 (m), 1076 (m), 952 (m) cm^{-1} ; MS (ESI) m/z : 348 ($\text{M}+3$) $^+$, 346 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{17}\text{ClN}_3$ ($\text{M}+\text{H}$) $^+$ 346.1111, found 346.1109.

4.1.13. 5-Anilino-1-(4-methoxyphenyl)-3-phenyl-1*H*-pyrazole (26). Yellow powder; mp 147–148 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 3.78 (s, 3H, OCH_3), 6.77 (s, 1H, Py), 7.04 (d, 2H, $J=8.8$ Hz, ArH), 7.32–7.45 (m, 6H, ArH), 7.63 (d, 2H, $J=8.8$ Hz, ArH), 7.70 (d, 2H, $J=7.6$ Hz, ArH), 7.90 (d, 2H, $J=7.1$ Hz, ArH), 8.76 (br, 1H, NH); IR (diffuse reflectance) 3434 (m), 2922 (s), 2852 (m), 1651 (m), 1613 (m), 1557 (s), 1539 (s), 1505 (s), 1455 (m), 1273 (m), 1252 (m), 1161 (m), 1024 (m) cm^{-1} ; MS (ESI) m/z : 342 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{22}\text{H}_{20}\text{N}_3\text{O}$ ($\text{M}+\text{H}$) $^+$ 342.1606, found 342.1605.

4.1.14. 1-(3-Nitrophenyl)-3-phenyl-5-*p*-toluidine-1*H*-pyrazole (27). Yellow powder; mp 148–149 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 2.28 (s, 3H, CH_3), 5.43 (br, 1H, NH), 6.48 (s, 1H, Py), 6.83 (d, 2H, $J=8.3$ Hz, ArH), 7.07 (d, 2H, $J=8.3$ Hz, ArH), 7.34–7.45 (m, 3H, ArH), 7.58 (t, 1H, $J=8.1$ Hz, ArH), 7.84 (d, 2H, $J=7.5$ Hz, ArH), 8.08–8.14 (m, 2H, ArH), 8.63–8.65 (m, 1H, ArH); ^{13}C NMR (CDCl_3 , 75 MHz) δ 20.55, 96.02, 116.51, 118.33, 121.41, 125.63, 128.44, 128.65, 128.77, 130.03, 130.09, 131.20, 132.61, 139.87, 140.43, 143.36, 148.70, 152.45; IR (diffuse reflectance) 3376 (m), 3036 (w), 2923 (m), 2855 (m), 1593 (m), 1556 (s), 1532 (s), 1455 (m), 1350 (s), 1247 (m), 1078 (m), 951 (m) cm^{-1} ; MS (ESI) m/z : 371 ($\text{M}+\text{H}$) $^+$. HRMS (ESI) calcd for $\text{C}_{22}\text{H}_{19}\text{N}_4\text{O}_2$ ($\text{M}+\text{H}$) $^+$ 371.1508, found 371.1510.

Acknowledgements

We are grateful to the China Medical University (CMU100-ASIA-17 and CMU101-S-23), and the National Science Council of Republic of China (NSC-99-2320-B-039-014-MY3) for financial support. This study is also supported in part by Taiwan Department of Health Clinical Trial and Research Center of Excellence (DOH100-TD-B-111-004).

Supplementary data

Supplementary data related to this article can be found online at <http://dx.doi.org/10.1016/j.tet.2012.11.022>.

References and notes

- For the Reviews; (a) Elguero, J. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R., Rees, C. W., Potts, K. T., Eds.; Pergamon: Oxford, 1984; Vol. 5; p 167; (b) Elguero, J. In *Comprehensive Heterocyclic Chemistry II*; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.; Pergamon: Oxford, 1996; Vol. 3; p 1; (c) Kost, A. N.; Grandberg, I. I. In *Advances in Heterocyclic Chemistry*; Katritzky, A. R., Boulton, A. J., Eds.; Academic: New York, NY, 1966; Vol. 6; p 347; (d) Lee, K. Y.; Kim, J. M.; Kim, J. N. *Tetrahedron Lett.* **2003**, *44*, 6737–6740 and references therein.
- Sakya, S. M.; Rast, B. *Tetrahedron Lett.* **2003**, *44*, 7629–7632 and references therein.
- Huang, Y. R.; Katzenellenbogen, J. A. *Org. Lett.* **2000**, *2*, 2833–2836 and references therein.
- (a) Nishigaki, N. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 2985–2988; (b) Braibante, M. E. F.; Braibante, H. T. S.; da Roza, J. K.; Henriques, D. M. *Synthesis* **2003**, 1160–1162; (c) Schenone, S.; Bruno, O.; Ranise, A.; Bondavalli, F.; D'Amico, M.; Filippelli, W.; Falcone, G.; De Novellis, V. *Il Farmaco* **1995**, *50*, 179–182; (d) Cooper, C. B.; Helal, C. J.; Sanner, M. A.; Wagner, T. T. *PCT WO 18346 A1*, 2002.
- (a) Ansel, J. E.; El Kaim, L.; Gadras, A.; Grimaud, L.; Jana, N. K. *Tetrahedron Lett.* **2002**, *43*, 8319–8321; (b) Atlan, V.; Buron, C.; El Kaim, L. *Synlett* **2000**, 489–490; (c) Palacios, F.; Aparicio, D.; de los Santos, J. M. *Tetrahedron* **1996**, *52*, 4123–4132; (d) Moreno-Manas, M.; Sebastián, R. M.; Vallribera, A.; Carini, F. *Synthesis* **1999**, 157–161.
- Dodd, D. S.; Martinez, R. L. *Tetrahedron Lett.* **2004**, *45*, 4265–4267.
- Abdel-Rahman, R. M.; Seada, M.; Fawzy, M.; El-Baz, I. *Pharmazie* **1994**, *49*, 729–733.
- Atlan, V.; El Kaim, L.; Grimaud, L.; Jana, N. K.; Majee, A. *Synlett* **2002**, 352–354.
- (a) Tang, J.; Shewchuk, L. M.; Sato, H.; Hasegawa, M.; Washio, Y.; Nishigaki, N. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 2985–2988; (b) Cooper, C. B.; Helal, C. J.; Sanner, M. A.; Wagner, T. T. *PCT WO 02/18346 A1*, 2002.

10. Penning, T. D.; Talley, J. J.; Bertenshaw, S. R.; Carter, J. S.; Collins, P. W.; Docter, S.; Graneto, M. J.; Lee, L. F.; Malecha, J. W.; Miyashiro, J. M.; Rogers, R. S.; Rogier, D. J.; Yu, S. S.; Anderson, G. D.; Burton, E. G.; Cogburn, J. N.; Gregory, S. A.; Koboldt, C. M.; Perkins, W. E.; Seibert, K.; Veenhuizen, A. W.; Zhang, Y. Y.; Isakson, P. C. *J. Med. Chem.* **1997**, *40*, 1347–1365.
11. For reviews, see: Jiang, L.; Buchwald, S. L.; *Metal-Catalyzed Cross-Coupling Reactions*, 2nd ed.; Wiley-VCH: Weinheim, 2004; vol. 2, pp 699–760.
12. (a) Ullmann, F. *Ber. Dtsch. Chem. Ges.* **1903**, *36*, 2382–2384; (b) Goldberg, I. *Ber. Dtsch. Chem. Ges.* **1906**, *39*, 1691–1692 For reviews, see: (c) Lindley, J. *Tetrahedron* **1984**, *40*, 1433–1456; (d) Ley, S. V.; Thomas, A. W. *Angew. Chem., Int. Ed.* **2003**, *42*, 5400–5449.
13. (a) Wolfe, J. P.; Buchwald, S. L. *J. Org. Chem.* **2000**, *65*, 1144–1157; (b) Wolfe, J. P.; Tomori, H.; Sadighi, J. P.; Yin, J.; Buchwald, S. L. *J. Org. Chem.* **2000**, *65*, 1158–1174; (c) Yin, J.; Buchwald, S. L. *Org. Lett.* **2000**, *2*, 1101–1104; (d) Huang, X.; Buchwald, S. L. *Org. Lett.* **2001**, *3*, 3417–3419; (e) Ali, M. H.; Buchwald, S. L. *J. Org. Chem.* **2001**, *66*, 2560–2565; (f) Klapars, A.; Huang, X.; Buchwald, S. L. *J. Am. Chem. Soc.* **2002**, *124*, 7421–7428; (g) Yin, J.; Buchwald, S. L. *J. Am. Chem. Soc.* **2002**, *124*, 6043–6048; (h) Gaertzen, O.; Buchwald, S. L. *J. Org. Chem.* **2002**, *67*, 465–475; (i) Enguehard, C.; Allouchi, H.; Gueiffier, A.; Buchwald, S. L. *J. Org. Chem.* **2003**, *68*, 4367–4370; (j) Charles, M. D.; Schultz, P.; Buchwald, S. L. *Org. Lett.* **2005**, *7*, 3965–3968; (k) Klingensmith, L. M.; Strieter, E. R.; Bader, T. E.; Buchwald, S. L. *Organometallics* **2006**, *25*, 82–91; (l) Strieter, E. R.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2006**, *45*, 925–928; (m) Shekhar, S.; Ryberg, P.; Hartwig, J. F.; Mathew, J. S.; Blackmond, D. G.; Strieter, E. R.; Buchwald, S. L. *J. Am. Chem. Soc.* **2006**, *128*, 3584–3591; (n) Anderson, K. W.; Tundel, R. E.; Ikawa, T.; Altman, R. A.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2006**, *45*, 6523–6527; (o) Ikawa, T.; Bader, T. E.; Biscoe, M. R.; Buchwald, S. L. *J. Am. Chem. Soc.* **2007**, *129*, 13001–13007; (p) Biscoe, M. R.; Bader, T. E.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2007**, *46*, 7232–7235.
14. (a) Klapars, A.; Antilla, J. C.; Huang, X.; Buchwald, S. L. *J. Am. Chem. Soc.* **2001**, *123*, 7727–7729; (b) Wolter, M.; Klapars, A.; Buchwald, S. L. *Org. Lett.* **2001**, *3*, 3803–3805; (c) Antilla, J. C.; Klapars, A.; Buchwald, S. L. *J. Am. Chem. Soc.* **2002**, *124*, 11684–11688; (d) Job, G. E.; Buchwald, S. L. *Org. Lett.* **2002**, *4*, 3703–3706; (e) Kwong, F. Y.; Klapars, A.; Buchwald, S. L. *Org. Lett.* **2002**, *4*, 581–584; (f) Kwong, F. Y.; Buchwald, S. L. *Org. Lett.* **2003**, *5*, 793–796; (g) Klapars, A.; Parris, S.; Anderson, K. W.; Buchwald, S. L. *J. Am. Chem. Soc.* **2004**, *126*, 3529–3533; (h) Antilla, J. C.; Baskin, J. M.; Bader, T. E.; Buchwald, S. L. *J. Org. Chem.* **2004**, *69*, 5578–5587; (i) Strieter, E. R.; Blackmond, D. G.; Buchwald, S. L. *J. Am. Chem. Soc.* **2005**, *127*, 4120–4121; (j) Altman, R. A.; Buchwald, S. L. *Org. Lett.* **2006**, *8*, 2779–2782; (k) Martin, R.; Rivero, M. R.; Buchwald, S. L. *Angew. Chem., Int. Ed.* **2006**, *45*, 7079–7082; (l) Shafir, A.; Buchwald, S. L. *J. Am. Chem. Soc.* **2006**, *128*, 8742–8743; (m) Altman, R. A.; Buchwald, S. L. *Org. Lett.* **2007**, *9*, 643–646; (n) Martin, R.; Larsen, C. H.; Cuenca, A.; Buchwald, S. L. *Org. Lett.* **2007**, *9*, 3379–3382; (o) Jones, C. P.; Anderson, K. W.; Buchwald, S. L. *J. Org. Chem.* **2007**, *72*, 7968–7973.
15. (a) Ma, D.; Jiang, J. *Tetrahedron: Asymmetry* **1998**, *9*, 1137–1142; (b) Ma, D.; Zhang, Y.; Yao, J.; Wu, S.; Tao, F. *J. Am. Chem. Soc.* **1998**, *120*, 12459–12467; (c) Ma, D.; Xia, C. *Org. Lett.* **2001**, *3*, 2583–2586; (d) Ma, D.; Cai, Q.; Zhang, H. *Org. Lett.* **2003**, *5*, 2453–2455; (e) Ma, D.; Xia, C.; Jiang, J.; Zhang, J.; Tang, W. *J. Org. Chem.* **2003**, *68*, 442–451; (f) Ma, D.; Cai, Q. *Synlett* **2004**, 128–130; (g) Cai, Q.; Zhu, W.; Zhang, H.; Zhang, Y.; Ma, D. *Synthesis* **2005**, 496–499; (h) Zhang, H.; Cai, Q.; Ma, D. *J. Org. Chem.* **2005**, *70*, 5164–5173; (i) Zou, B.; Yuan, Q.; Dawei, M. *Angew. Chem., Int. Ed.* **2007**, *46*, 2598–2601; (j) He, G.; Wang, J.; Ma, D. *Org. Lett.* **2007**, *9*, 1367–1369.
16. (a) Yamada, K.; Kubo, T.; Tokuyama, H.; Fukuyama, T. *Synlett* **2002**, 231–234; (b) Okano, K.; Tokuyama, H.; Fukuyama, T. *Org. Lett.* **2003**, *5*, 4987–4990; (c) Yamada, K.; Kurokawa, T.; Tokuyama, H.; Fukuyama, T. *J. Am. Chem. Soc.* **2003**, *125*, 6630–6631; (d) Okano, K.; Tokuyama, H.; Fukuyama, T. *J. Am. Chem. Soc.* **2006**, *128*, 7136–7137; (e) Okano, K.; Tokuyama, H.; Fukuyama, T. *Chem.—Asian. J.* **2008**, *3*, 296–309.
17. Kubo, T.; Katoh, C.; Yamada, K.; Okano, K.; Tokuyama, H.; Tohru Fukuyama, T. *Tetrahedron* **2008**, *64*, 11230–11236.
18. Su, W.-N.; Lin, T.-P.; Cheng, K.-M.; Sung, K.-C.; Lin, S.-K.; Wong, F. F. *J. Heterocycl. Chem.* **2010**, *47*, 831–837.