Copper-Mediated Amidation of Alkenylzirconocenes with Acyl Azides: Formation of Enamides

LETTERS XXXX Vol. XX, No. XX 000–000

ORGANIC

Hailan Liu,^{†,‡} Yiqing Zhou,[†] Xiaoyu Yan,[†] Chao Chen,[†] Qingbin Liu,^{*,‡} and Chanjuan Xi^{*,†}

Key Laboratory of Bioorganic Phosphorus Chemistry & Chemical Biology (Ministry of Education), Department of Chemistry, Tsinghua University, Beijing 100084, China, and Institute of Chemistry and Chemical Engineering, Hebei Normal University, Shijiazhuang 050091, China

cjxi@tsinghua.edu.cn

Received August 5, 2013



Copper-mediated amidation of alkenylzirconocenes generated in situ from alkynes and zirconocenes with acyl azides is accomplished under mild conditions. The reaction can be used to prepare various enamides.

Enamides contain an electron-withdrawing acyl group on the nitrogen atom, and they display an exceptionally fine balance of stability and reactivity of nitrogensubstituted alkenes. Therefore, the enamides provide various insertions into organic molecules as nitrogen-based functional compounds and are widely useful in organic synthesis.^{1,2} The traditional synthetic route to enamides is the thermal decomposition of alkylidenebisamides.³

[†] Tsinghua University.

(4) Examples for Pd-catalyzed the reactions of alkenyl halides with amides, see: (a) Xu, J.; Fu, Y.; Xiao, B.; Gong, T.; Guo, Q. *Tetrahedron Lett.* **2010**, *51*, 5476. (b) Klapars, A.; Campos, K. R.; Chen, C.; Volante, R. P. Org. Lett. **2005**, *7*, 1185. (c) Pan, X.; Cai, Q.; Ma, D. Org. Lett. **2004**, *6*, 1809. (d) Jiang, L.; Job, G. E.; Klapars, A.; Buchwald, S. L. Org. Lett. **2003**, *5*, 3667.

(5) Examples for copper-catalyzed the reactions of alkenyl halides with amides, see: (a) Lu, H.; Yuan, X.; Zhu, S.; Sun, C.; Li, C. J. Org. Chem. **2008**, 73, 8665. (b) Cesati, R. R., III; Dwyer, G.; Jones, R. C.; Hayes, M. P.; Yalamanchili, P.; Casebier, D. S. Org. Lett. **2007**, 9, 5617. (c) He, G.; Wang, J.; Ma, D. Org. Lett. **2007**, 9, 1367. (d) Toumi, M.; Couty, F.; Evano, G. Angew. Chem., Int. Ed. **2007**, 46, 572. (e) Dehli, J. R.; Bolm, C. Adv. Synth. Catal. **2005**, 347, 239. (f) Martn, R.; Rivero, M. R.; Buchwald, L. Angew. Chem., Int. Ed. **2006**, 45, 7079.

However, the synthetic method involves rather harsh reaction conditions such as high temperature and/or the use of strong acids and bases. In recent years, metal-promoted olefinic C–N bond formation reactions provided a powerful and convenient approach to the nitrogen-substituted alkenes.^{4–7} Among them, Pd- or Cu-catalyzed amidation of alkenyl halides with amides provided a straightforward method for the formation of enamides.^{4,5} Meanwhile, copper-promoted amidation of alkenylmetal-loids with amides has been also received attention.⁸

^{*}Hebei Normal University.

⁽¹⁾ For recent reviews on enamides, see: (a) Gopalaiah, K.; Kagan, H. B. *Chem. Rev.* **2011**, *111*, 4599. (b) Matsubara, R.; Kobayashi, S. *Acc. Chem. Res.* **2008**, *41*, 292.

^{(2) (}a) Lei, C.; Wang, D.; Zhao, L.; Zhu, J.; Wang, M. J. Am. Chem. Soc. 2013, 135, 4708. (b) Hesp, K. D.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2011, 133, 11430. (c) Carbery, D. R. Org. Biomol. Chem. 2008, 6, 3455.

⁽³⁾ Dov, B.-I.; Reuven, G. Tetrahedron Lett. 1965, 50, 4523.

⁽⁶⁾ Metal-catalyzed olefinic C-N bond formation, see: (a) Barluenga, J.; Fernandez, M. A.; Aznar, F.; Valdes, C. *Chem. Commun.* 2002, 2362.
(b) Reddy, C. R. V.; Urgaonkar, S.; Verkade, J. G. *Org. Lett.* 2005, *7*, 4427. (c) Dehli, J. R.; Legros, J.; Bolm, C. *Chem. Commun.* 2005, 973. (d) Wang, Y.; Liao, Q.; Xi, C. *Org. Lett.* 2010, *12*, 2951. (e) Kitamura, M.; Suga, T.; Chiba, S.; Narasaka, K. *Org. Lett.* 2004, *6*, 4619. (f) Liu, S.; Yu, Y.; Liebeskind, L. S. *Org. Lett.* 2007, *9*, 1947.

⁽⁷⁾ Gooβen, L. J.; Rauhaus, J. E.; Deng, G. Angew. Chem. 2005, 117, 4110.

^{(8) (}a) Bolshan, Y.; Batey, R. Angew. Chem., Int. Ed. 2008, 47, 2109.
(b) Bolshan, Y.; Batey, R. Tetrahedron 2010, 66, 5283. (c) Arsenyan, P.; Petrenko, A.; Belyakov, S. Tetrahedron Lett. 2008, 49, 5255.

⁽⁹⁾ For examples, see: (a) Cheng, D.; Ling, F.; Li, Z.; Yao, W.; Ma, C. Org. Lett. 2012, 14, 3146. (b) Bae, I.; Han, H.; Chang, S. J. Am. Chem. Soc. 2005, 127, 2038. (c) Himo, F.; Lovell, T.; Hilgraf, R.; Rostovtsev, V. V.; Noodleman, L.; Sharpless, K. B.; Fokin, V. V. J. Am. Chem. Soc. 2005, 127, 210. (d) Cho, S. H.; Yoo, E. J.; Bae, I.; Chang, S. J. Am. Chem. Soc. 2005, 127, 16046. (e) Saxon, E.; Bertozzi, C. R. Science 2000, 287, 2007. (f) Cassidy, M. P.; Raushel, J.; Fokin, V. V. Angew. Chem., Int. Ed. 2006, 45, 3154. (g) Meldal, M.; Tornøe, C. W. Chem. Rev. 2008, 108, 2952.

In addition to the use of amides in amidation reactions, other N-atom reagents have been rarely reported in the reaction. Acyl azides are readily accessible and have participated in a wide range of reactions that constructed new carbon–nitrogen bonds.⁹ Moreover, molecular nitrogen is released as a byproduct from reactions that is also very attractive from an environmental perspective. However, acyl azides as nitrenoid precursors in the amidation reactions have been far less attention. Our group has succeeded in the copper-mediated aminations and iminations of alkenylzirconocenes to obtain a wide range of enamines¹⁰ and enimines,¹¹ respectively. As part of our ongoing project on alkenylzirconocene chemistry,^{12–14}

Scheme 1. Cu-Mediated Amidation of Alkenylzirconocenes with Acyl Azides

$$\begin{array}{c} O \\ R \\ \end{array} + \begin{array}{c} R^{3} \\ Cp_{2}Zr \\ OR' \end{array} + \begin{array}{c} R^{4} \\ R^{5} \\ \end{array} \begin{array}{c} CuCl (1.2 equiv) \\ THF, 50 \ ^{\circ}C \\ R \end{array} O \\ \end{array} \begin{array}{c} R^{3} \\ O \\ R \\ \end{array} + \begin{array}{c} R^{4} \\ R^{5} \\ R^{5} \\ \end{array}$$

we envisioned that use of acyl azides as nitrenoid precursors could be applied to the amidation reaction, which would provide a useful method for the synthesis of enamides. Herein, we describe a copper-mediated amidation of alkenylzirconocenes with acyl azides to afford a wide range of enamides under mild conditions (Scheme 1).

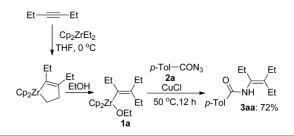
As an example, alkenylzirconocene 1a,¹⁵ which was generated by ethanolysis of the corresponding diethylzirconacyclopentene prepared from 3-hexyne and Cp₂ZrEt₂ in THF,^{13c} reacted with 4-methylbenzoyl azide **2a** in the presence of CuCl to afford enamide **3aa** in 72% isolated yield (Scheme 2).

The extensive amidation reaction with 2a was conducted with variously fresh alkenylzirconocenes $1a-l^{13c,f}$ and individually to form their corresponding enamides 3aa-lain the isolated yields between 45% to 88%, indicating the

(14) Takahashi, T.; Kotora, M.; Fischer, R.; Nishihara, Y.; Nakajima, K. J. Am. Chem. Soc. 1995, 117, 11039.

(15) Compound 1a (generated in situ) was obtained in 95% yield, which was detected by ¹H NMR using mesitylene as an internal standard.

Scheme 2. Amidation of 1a with 2a



wide scope of various substituents as alkyl, aryl, allyl, and TMS. The representative results are summarized in Table 1. In all cases, one product was solely observed. To confirm their molecular structures, single crystals of the representative **3da** were obtained by the recrystallization in *n*-hexane. Its structure revealed the *cis*-fashion of its phenyl and methyl groups (see the Supporting Information), which was consistent with its substance of the alkenylzirconocene **1d**, maintaining the configuration of the double bond during the amidation reaction. Therefore, such a reaction would be potentially used for a highly substituted enamides.

To establish the full scope of the amidation reaction, we further explored a range of acyl azide substrates (Figure 1) with the alkenylzirconocenes. Aryl acyl azides 2b, 2c, and 2d derived from benzoyl chloride, p-anisoyl chloride, and *p*-chlorobenzoyl chloride¹⁶ were also used in the reaction to give the corresponding enamides (Scheme 3, 3ab, 3cc, 3jc, 3ad, and 3cd). If the substituent group was electronwithdrawing group or electron-donating on the benzene ring, the reactions could afford the desired products in good yields. When heteroaryl acylazides 2e-j derived from the corresponding acyl chloride, such as 2-pyridinecarbonyl chloride,¹⁷ 2-thiophenecarbonyl chloride,¹⁸ and 2-furancarbonyl chloride,¹⁹ were used, the corresponding products **3ae-cg** were obtained in moderate yields. When alkylacyl azide 2h and benzylacyl azide 2i derived from heptanovl chloride¹⁹ and phenylacetyl chloride¹⁹ were employed in the reaction, the expected enamides were not obtained and starting materials remained. Cinnamoyl

⁽¹⁰⁾ Yan, X.; Chen, C.; Zhou, Y.; Xi, C. Org. Lett. 2012, 14, 4750.
(11) Liu, H.; Yan, X.; Chen, C.; Liu, Q.; Xi, C. Chem. Commun. 2013, 49, 5513.

^{(12) (}a) Okukado, N.; Horn, D. E. V.; Klima, W. L.; Negishi, E. *Tetrahedron Lett.* **1978**, *12*, 1027. (b) Vincent, P.; Beaucourt, J.-P.; Pichat, L. *Tetrahedron Lett.* **1982**, *23*, 63. (c) Negishi, E.; Takahashi, T.; Baba, S.; Horn, D. E. V.; Okukado, N. J. Am. Chem. Soc. **1987**, *109*, 2393. (d) Miyaji, T.; Xi, Z.; Ogasawara, M.; Nakajima, K.; Takahashi, T. J. Org. Chem. **2007**, *72*, 8737.

^{(13) (}a) Hart, D. W.; Schwartz, J. J. Am. Chem. Soc. 1974, 96, 8115.
(b) Lipshutz, B. H.; Pfeiffer, S. S.; Noson, K.; Tomioka, T. In *Titanium and Zirconium in Organic Synthesis*; Marek, I., Ed.; Wiley-VCH: Weinheim, 2002; Chapter 4, p 110. (c) Takahashi, T.; Aoyagi, K.; Hara, R.; Suzuki, N. J. Chem. Soc., Chem. Commun. 1993, 1042. (d) Takahashi, T.; Kondakov, D. Y.; Suzuki, N. *Tetrahedron Lett.* 1993, 34, 6571. (e) Takahashi, T.; Suzuki, N.; Kageyama, M.; Kondakov, D. Y.; Hara, R. *Tetrahedron Lett.* 1993, 34, 4811. (f) Suzuki, N.; Kondakov, D. Y.; Kageyama, M.; Kondakov, D. Y.; Kageyama, M.; Kondakov, D. Y.; Kageyama, M.; Kotora, M.; Hara, R.; Takahashi, T. *Tetrahedron 1995*, 51, 4519. (g) Hara, R.; Nishihara, Y.; Landré, P. D.; Takahashi, T. *Tetrahedron Lett.* 1997, 38, 447. (h) Takahashi, T.; Xi, C.; Ura, Y.; Nakajima, K. J. Am. Chem. Soc. 2000, 112, 3228.

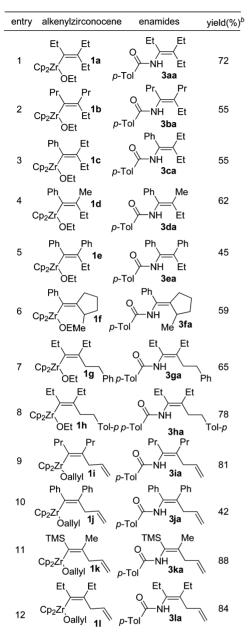
⁽¹⁶⁾ Barrett, E. W.; Porter, C. W. J. Am. Chem. Soc. 1941, 63, 3434.(17) Nettekoven, M. Synlett 2001, 1917.

⁽¹⁸⁾ Padwa, A.; Crawford, K. R.; Rashatasakhon, P.; Rose, M. J. Org. Chem. 2003, 68, 2609.

⁽¹⁹⁾ Swetha, M.; Ramana, P. V.; Shirodkar, S. G. Org. Prep. Proced. Int. 2011, 43, 348.

⁽²⁰⁾ Leading references: (a) Takahashi, T.; Kotora, M.; Kasai, K.;
Suzuki, N.; Nakajima, K. Organometallics 1994, 13, 4184. (b) Takahashi,
T.; Hara, R.; Nishihara, Y.; Kotora, M. J. Am. Chem. Soc. 1996, 118,
5154. (c) Takahashi, T.; Xi, Z.; Yamazaki, A.; Liu, Y.; Nakajima, K.;
Kotora, M. J. Am. Chem. Soc. 1998, 120, 1672. (d) Takahashi, T.; Sun,
W.-H.; Liu, Y.; Nakajima, K.; Kotora, M. Organometallics 1998, 17,
3841. (e) Kotora, M.; Xi, C.; Takahashi, T. Tetrahedron Lett. 1998, 39,
4321. (f) Xi, C.; Kotora, M.; Nakajima, K.; Takahashi, T. J. Org. Chem.
2000, 65, 945. (g) Chen, C.; Xi, C.; Jiang, Y.; Hong, X. J. Am. Chem. Soc.
2005, 127, 8024. (h) Dufková, L.; Kotora, M.; Cisafová, I. Eur. J. Org.
Chem. 2005, 2491. (i) Takahashi, T.; Nishihara, Y.; Hara, R.; Huo, S.;
Kotora, M. Chem. Commun. 1997, 1599. (j) Nishihara, Y.; Miyasaka, M.;
Okamoto, M.; Takahashi, H.; Inoue, E.; Tanemura, K.; Takagi, K. J. Am.
Chem. Soc. 2007, 129, 12634. (k) Nishihara, Y.; Okada, Y.; Jiao, J.;
Suetsugu, M.; Lan, M.; Kinoshita, M.; Iwasaki, M.; Takagi, K. Angew.
Chem., Int. Ed. 2011, 50, 8660.

Table 1. Amidation of Various Alkenylzirconocenes^a



^{*a*} Reaction conditions: alkenylzirconocene **1** prepared in situ from alkyne (0.5 mmol) with zirconocene in 3 mL of THF solution, acyl azide **2a** (0.45 mmol), CuCl (0.55 mmol), N₂, 50 °C, protected from light, 12 h. ^{*b*} Isolated yield based on acyl azide.

azide **2j** derived from cinnamoyl chloride¹⁷ was used in the reaction to afford the corresponding enamides **3cj** and **3dj** in 66% and 44% yield, respectively. Sulfonyl azide $2k^{17}$ could also offer the corresponding enamide **3ck** in 44% yield.

The plausible mechanism is proposed in Scheme 4. On the base of the known transmetalation from the C–Zr to C–Cu bond,²⁰ the alkenylzirconocene **1** is first transmetalated with CuCl to form an intermediate as the alkenylcopper **4**. Then coordination of acyl azide^{10c} to **4** forms the intermediate **5**. Rearrangement of **5** and a release of N₂ forms the intermediate **6**, which is hydrolyzed to afford product **3**.

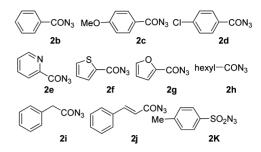
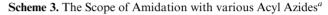
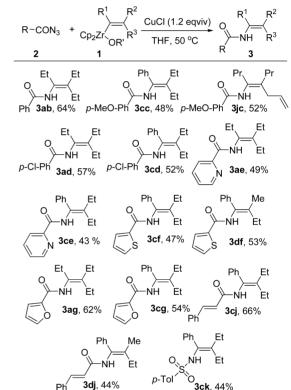


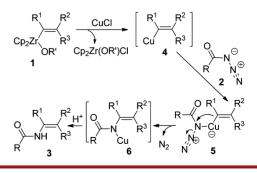
Figure 1. Range of azide substrates.





^{*a*} Reaction conditions: alkenylzirconocene **1** prepared in situ from alkyne (0.5 mmol) with zirconocene in 3 mL of THF solution, acyl azide **2** (0.45 mmol), CuCl (0.55 mmol), N₂, 50 °C, protected from light, 12 h, isolated yield based on acyl azide.

Scheme 4. Possible Reaction Pathway



In summary, a general copper-mediated amidation of alkenylzirconocenes with acyl azides has been developed. This reaction represents an interesting entry to the synthesis of enamides within a one-pot reaction from alkynes.

Acknowledgment. This work was supported by the National Key Basic Research Program of China (973 program) (2012CB933402) and the National Natural Science Foundation of China (21032004 and 21272132).

Supporting Information Available. Experimental procedures, full characterization including ¹H NMR and ¹³C NMR data and IR data for all new compounds, copies of spectra for all compounds, and the X-ray structure of product **3da** and X-ray data for **3da** (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.